

Novel liquid crystalline organogelators based on terephthalic acid and terephthalaldehyde derivatives: Properties and promotion through the formation of halogen bonding

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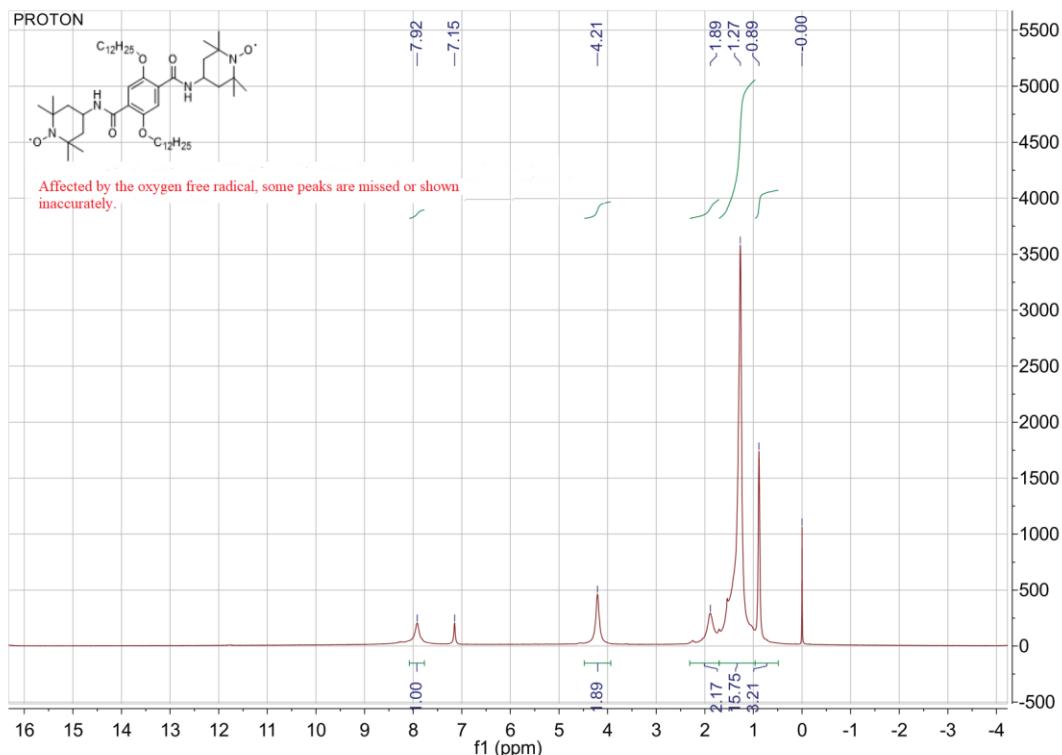


Fig. S1 ¹H NMR of A1

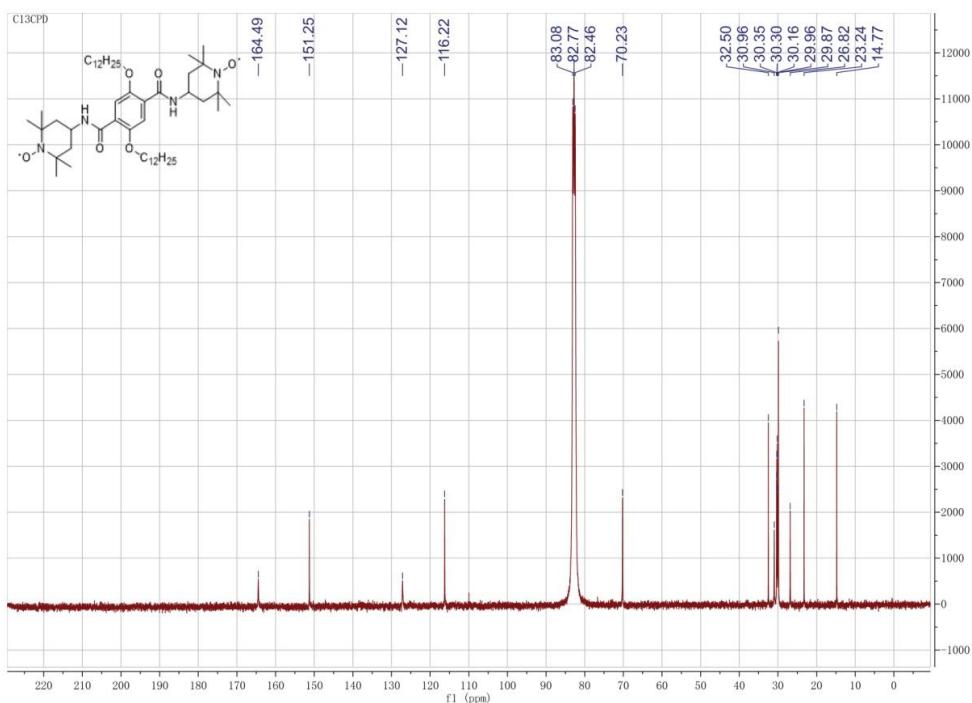


Fig. S2 ^{13}C NMR of A1

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	2.0 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	2500 m/z	Set Collision Cell RF	420.0 Vpp	Set Divert Valve	Source

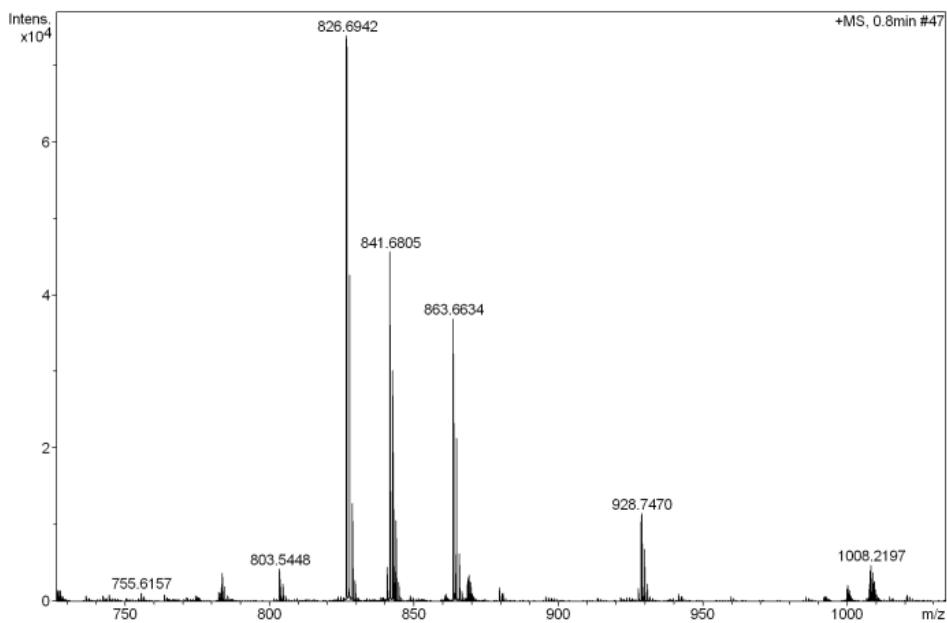


Fig. S3 HRMS of A1

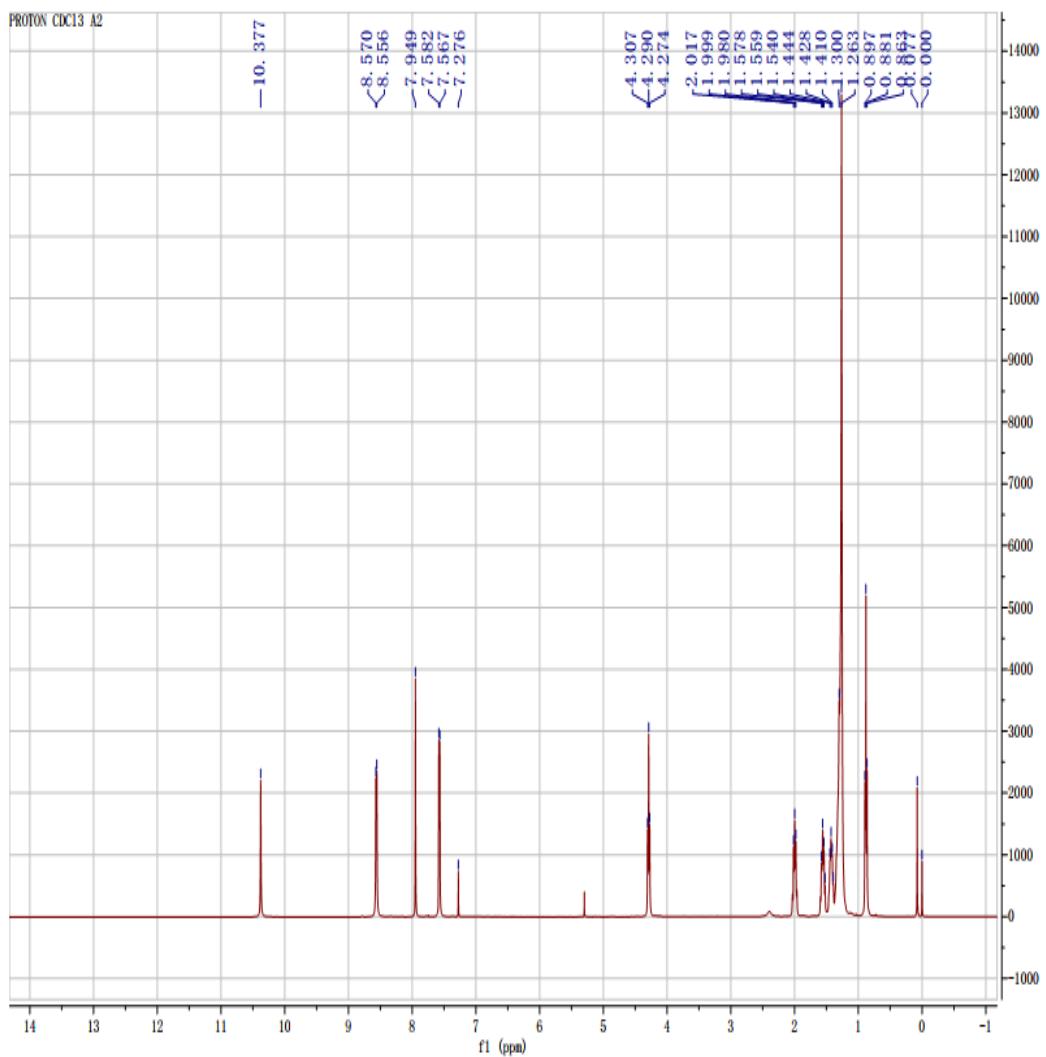


Fig. S4 ¹H NMR of A2

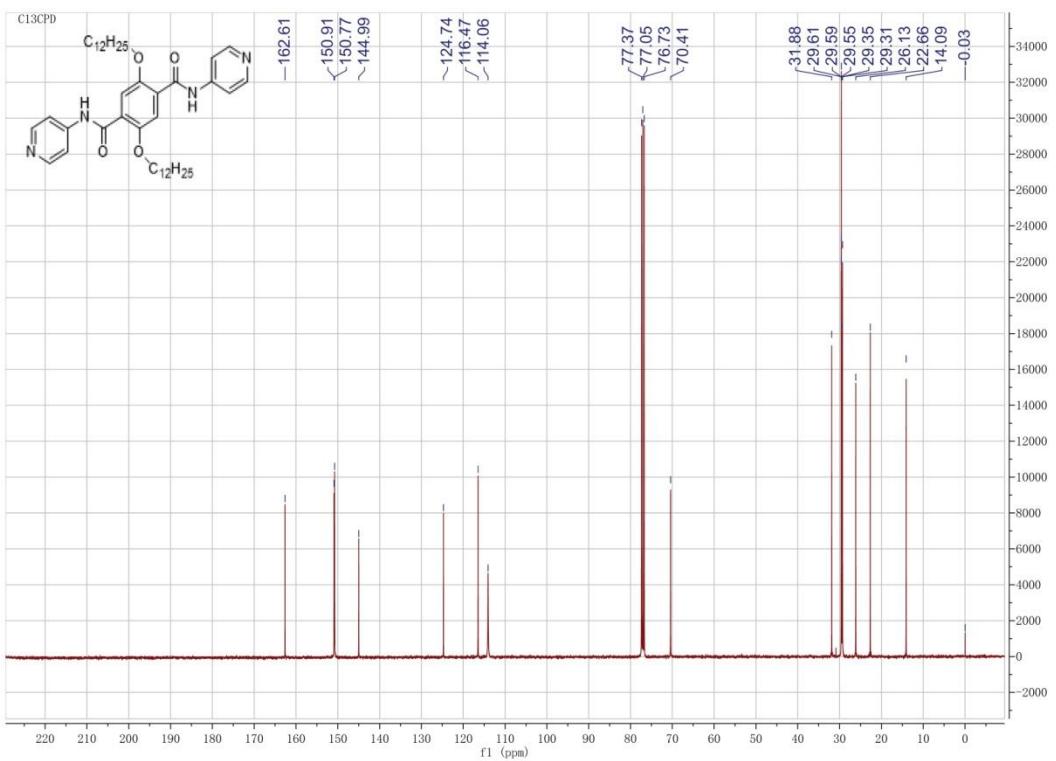


Fig. S5 ^{13}C NMR of A2

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	2.0 Bar
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Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	2500 m/z	Set Collision Cell RF	420.0 Vpp	Set Divert Valve	Source

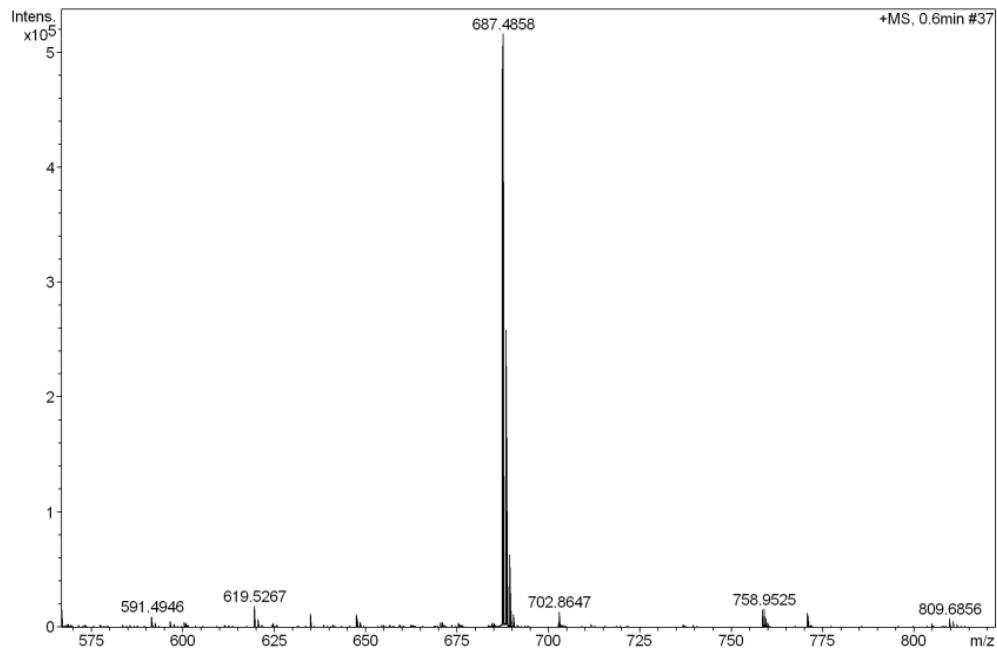


Fig. S6 HRMS of A2

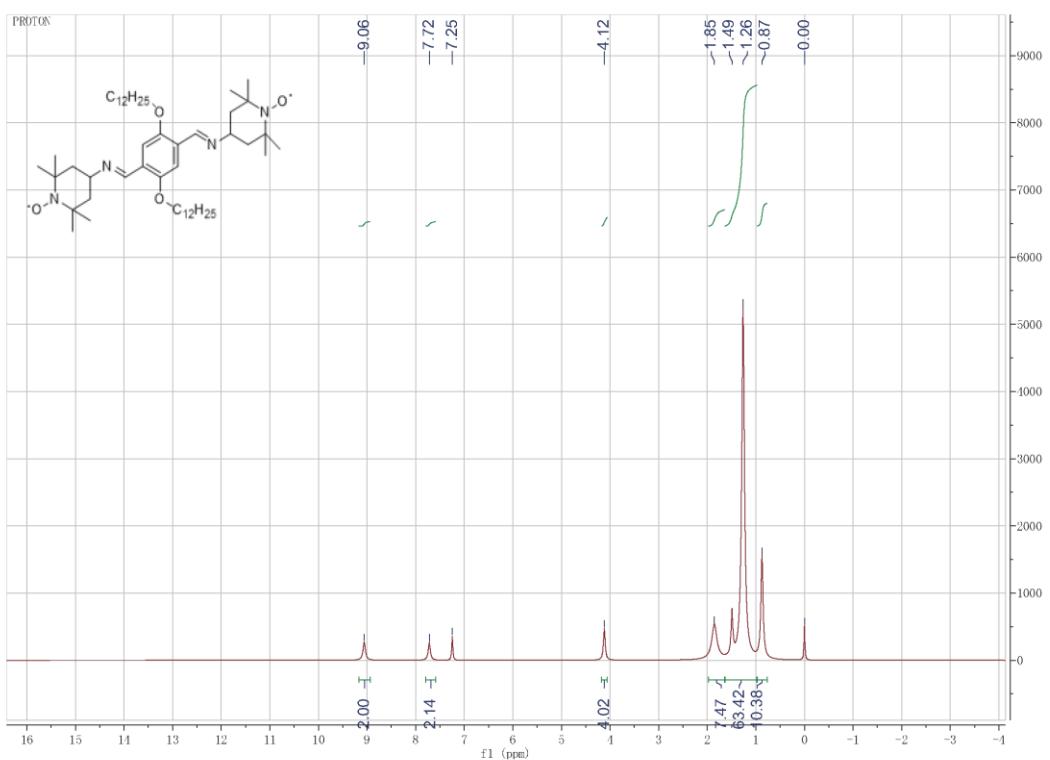


Fig. S7 ¹H NMR of SB

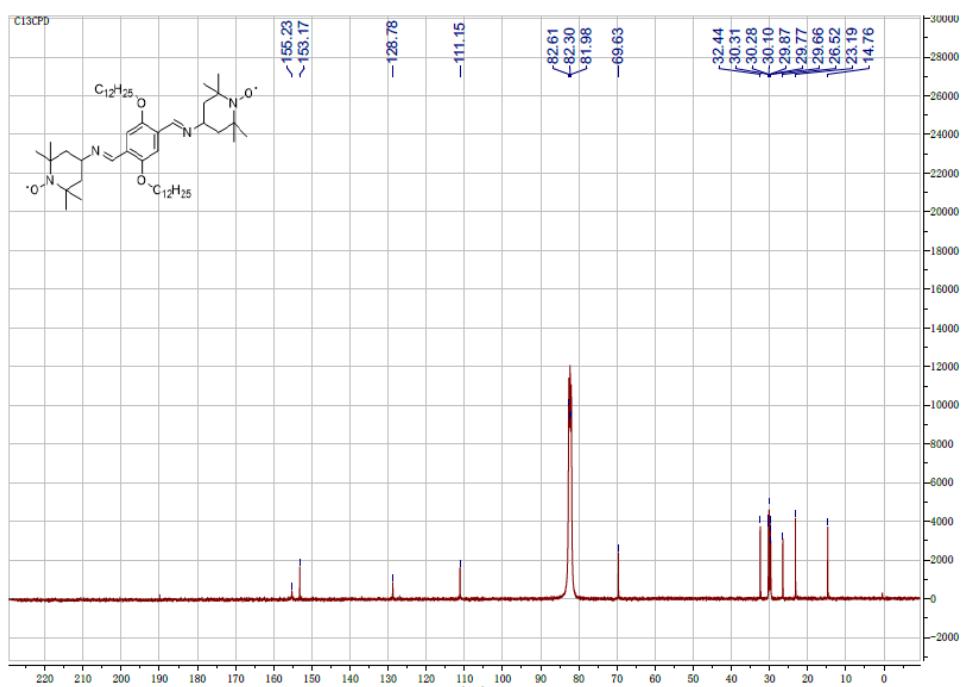
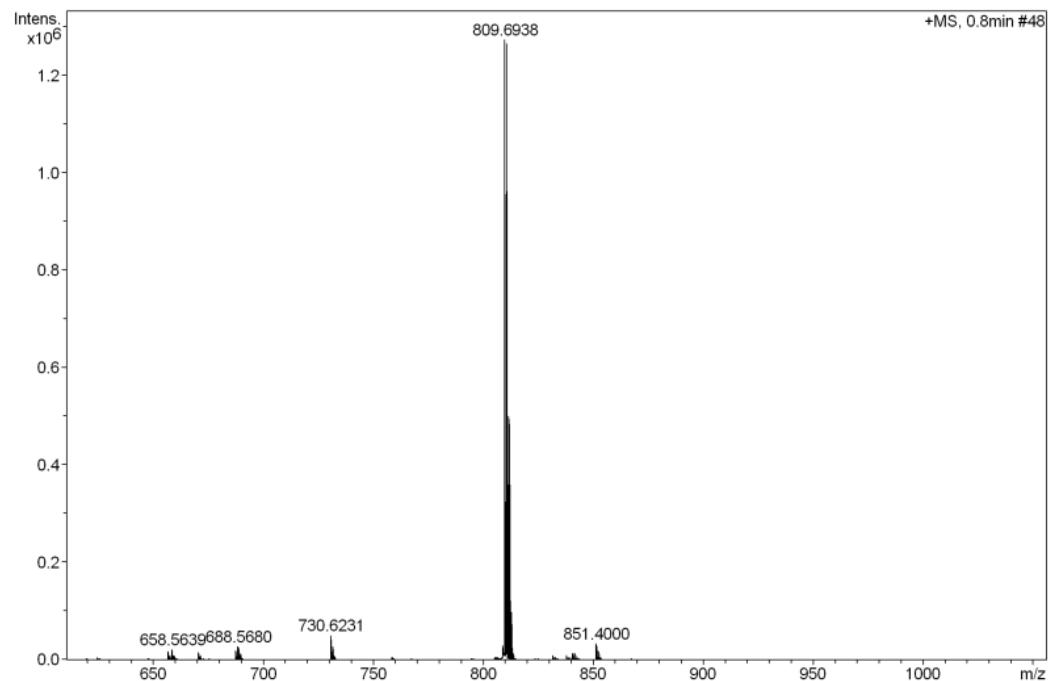
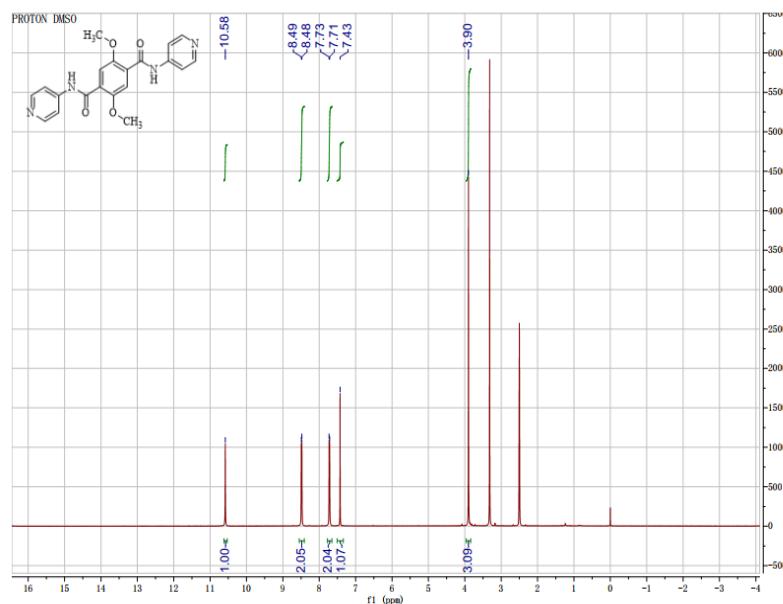


Fig. S8 ¹³C NMR of SB

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	2.0 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	100 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	2500 m/z	Set Collision Cell RF	150.0 Vpp	Set Divert Valve	Source

**Fig. S9 HRMS of SB****Fig. S10 ¹H NMR of A2-OMe**

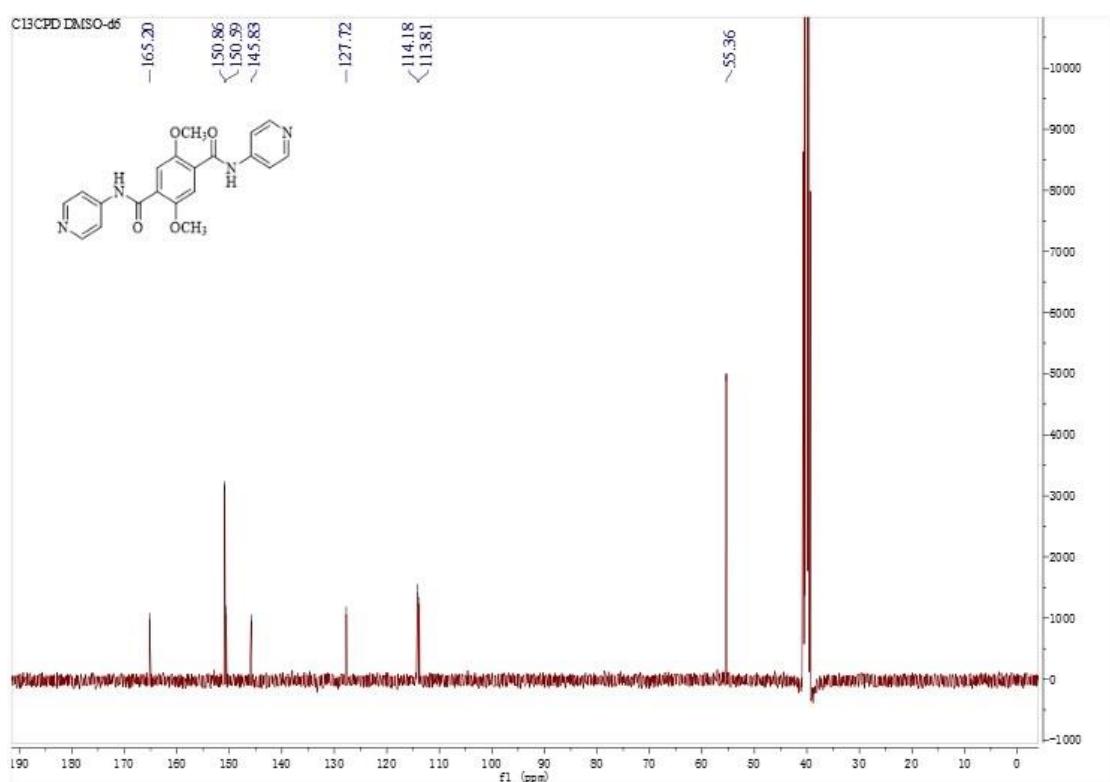


Fig. S11 ^{13}C NMR of A2-OMe

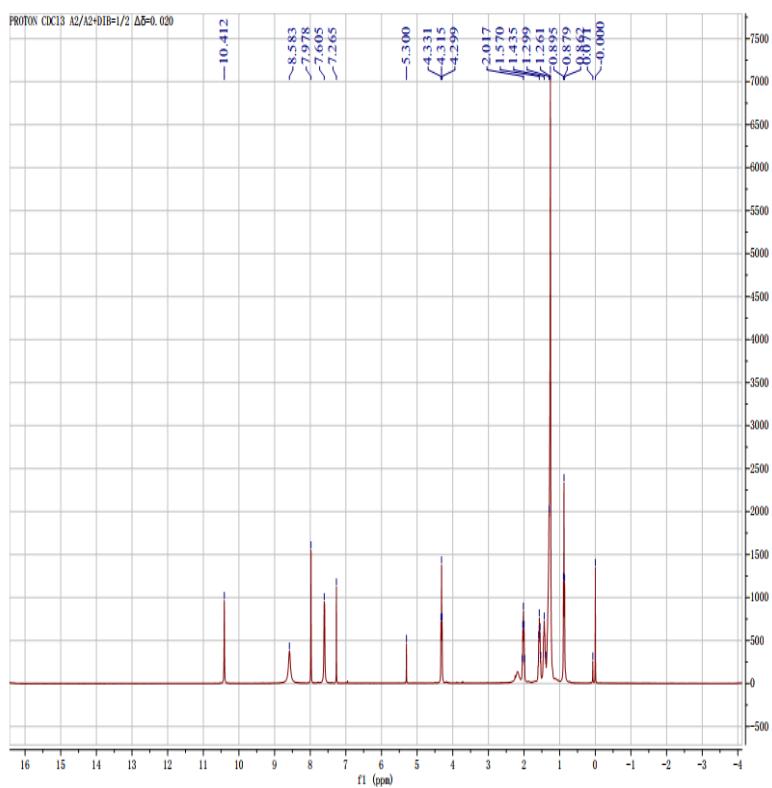


Fig. S12 ^1H NMR of A2/(A2+DIB)=1/2 $\Delta\delta$ =0.020

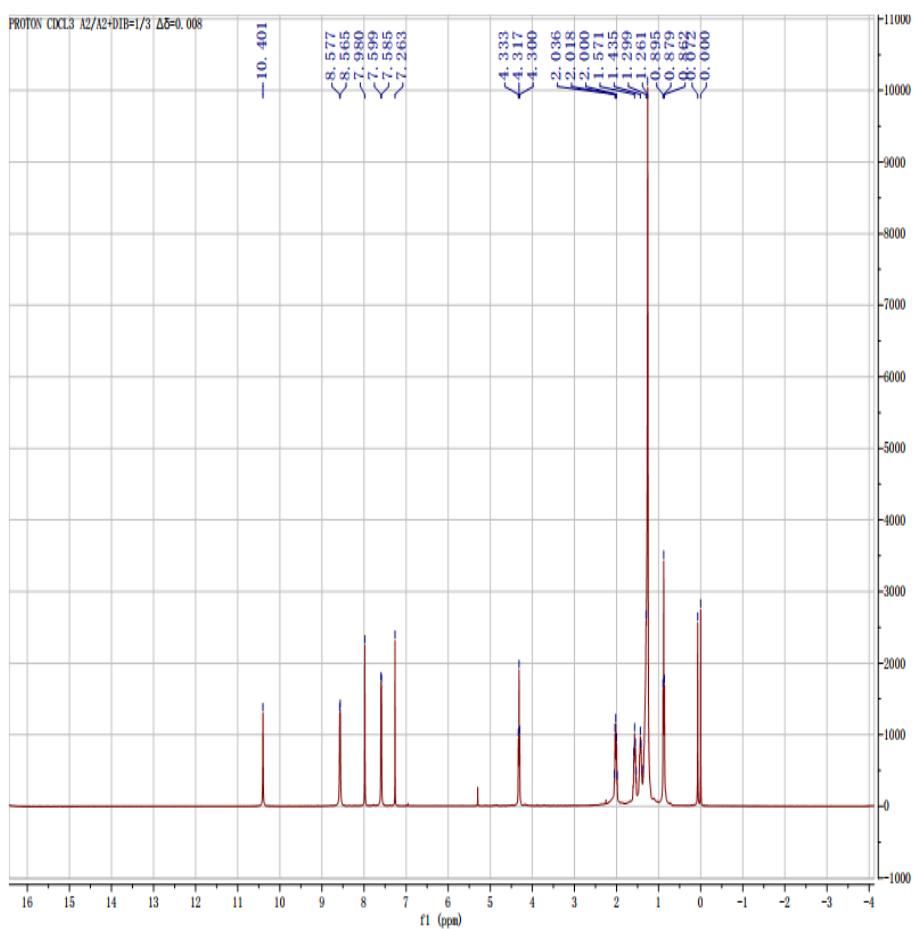


Fig. S13 ¹H NMR of A2/(A2+DIB)=1/3 Δδ=0.008

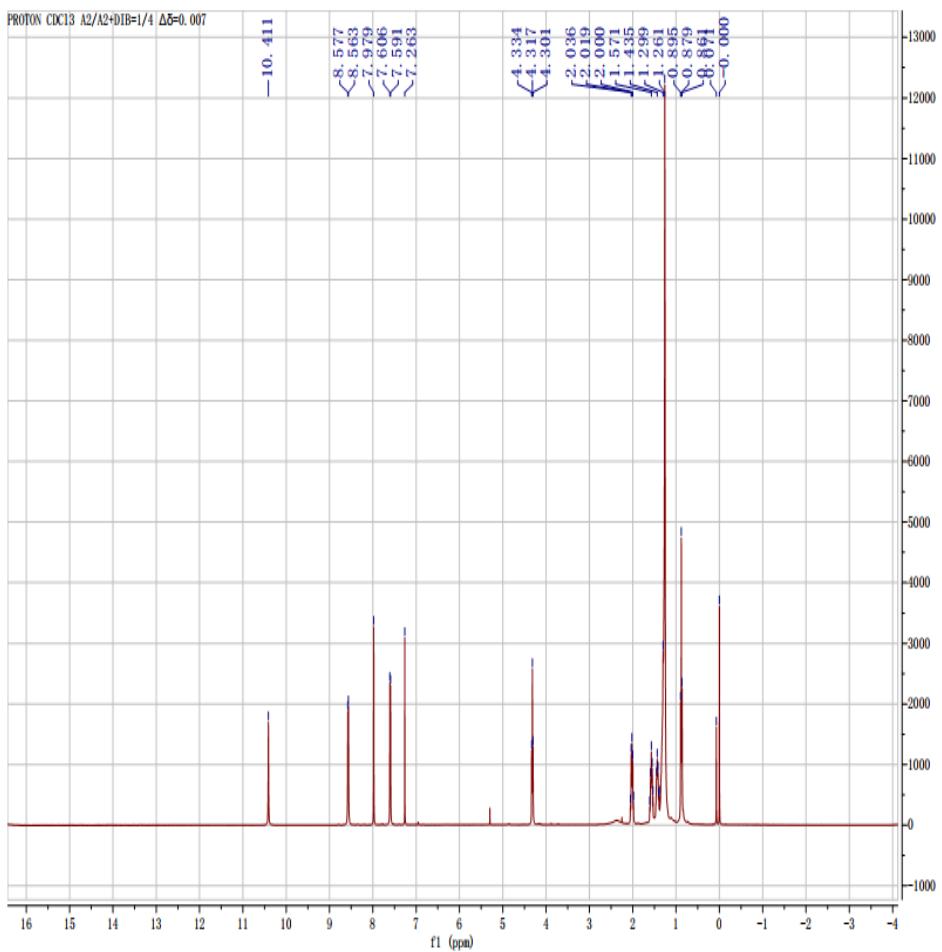


Fig. S14 ¹H NMR of A2/(A2+DIB)=1/4 Δδ=0.007

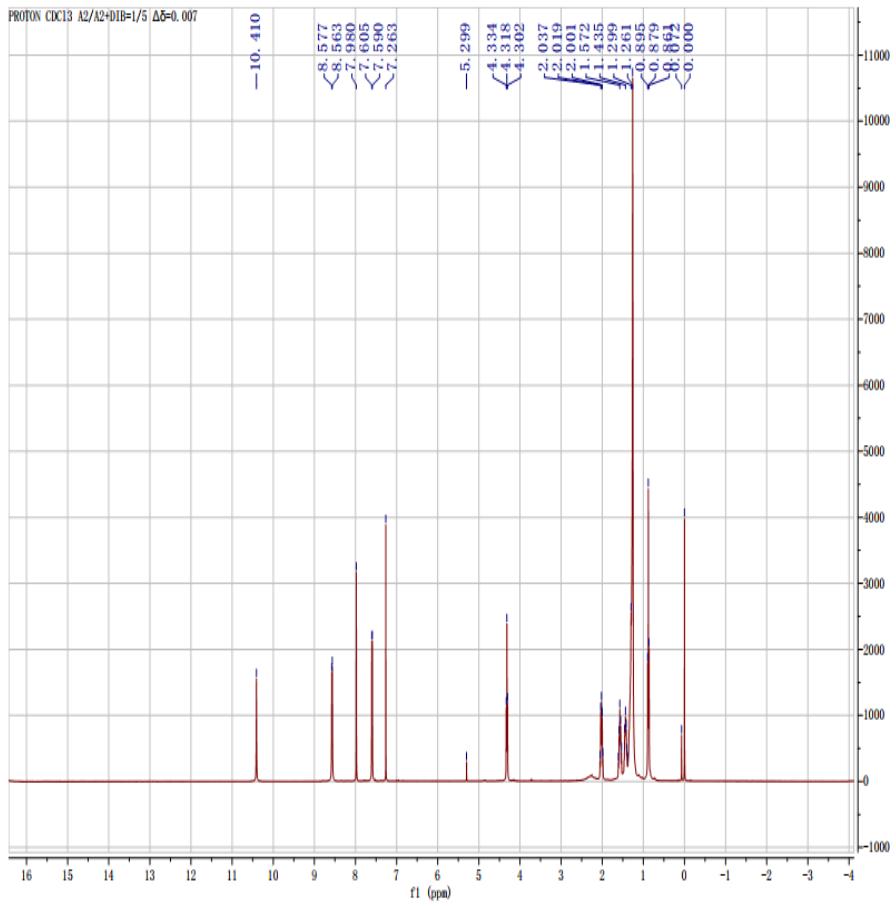


Fig. S15 ¹H NMR of A2/(A2+DIB)=1/5 Δδ=0.007

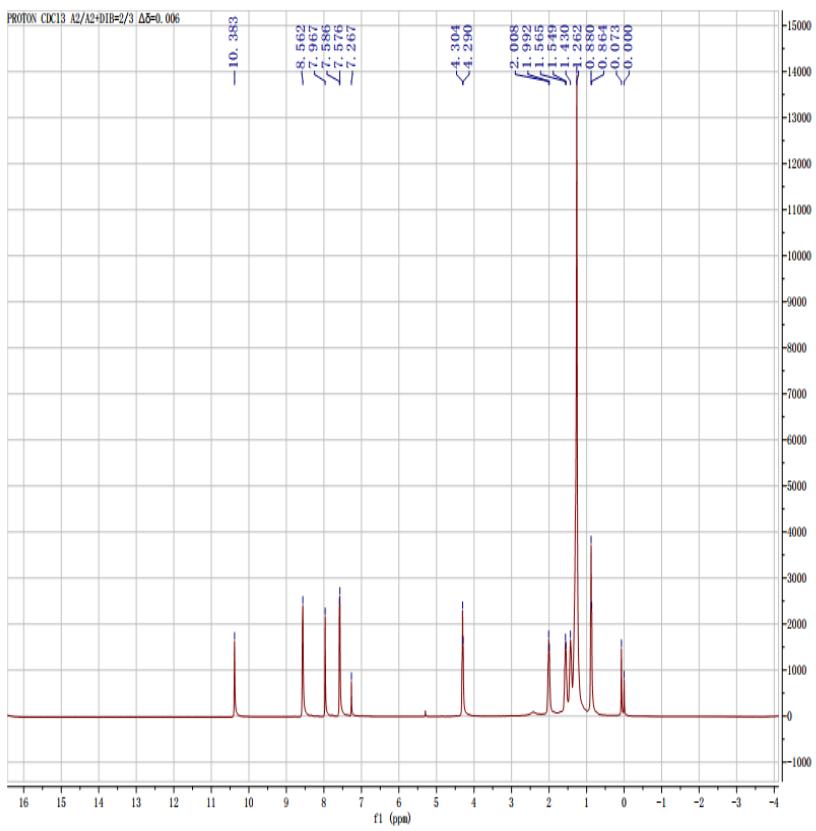


Fig. S16 ¹H NMR of A2/(A2+DIB)=2/3 Δδ=0.006

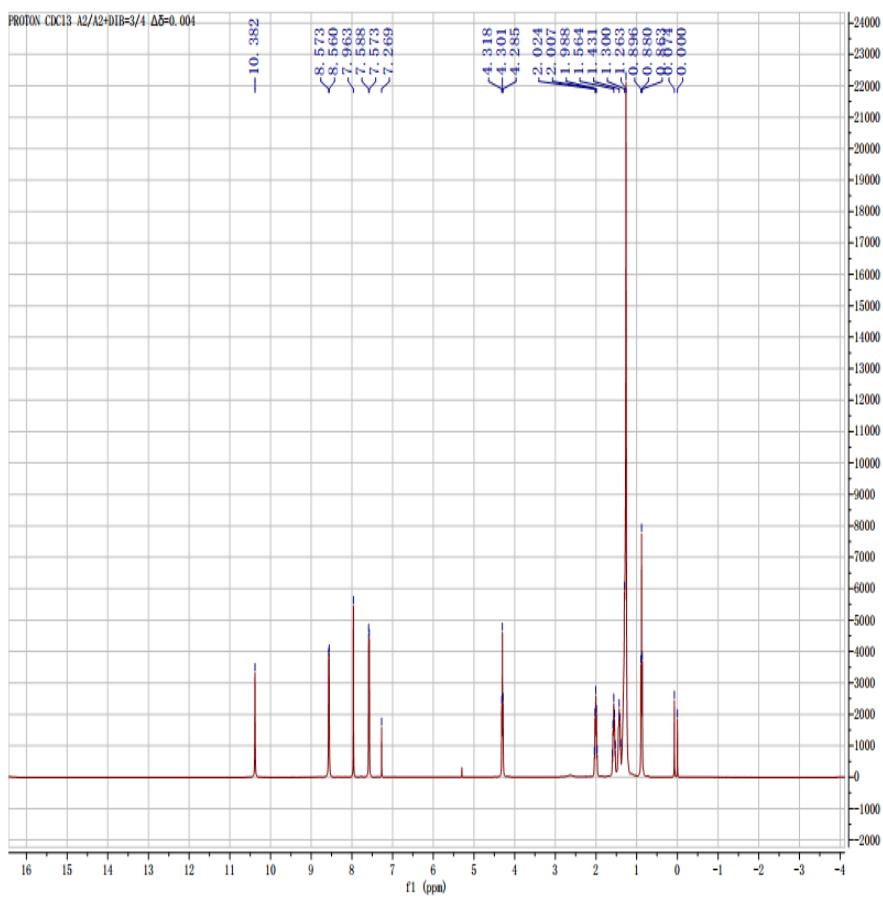


Fig. S17 ^1H NMR of A2/(A2+DIB)=3/4 Δδ=0.004

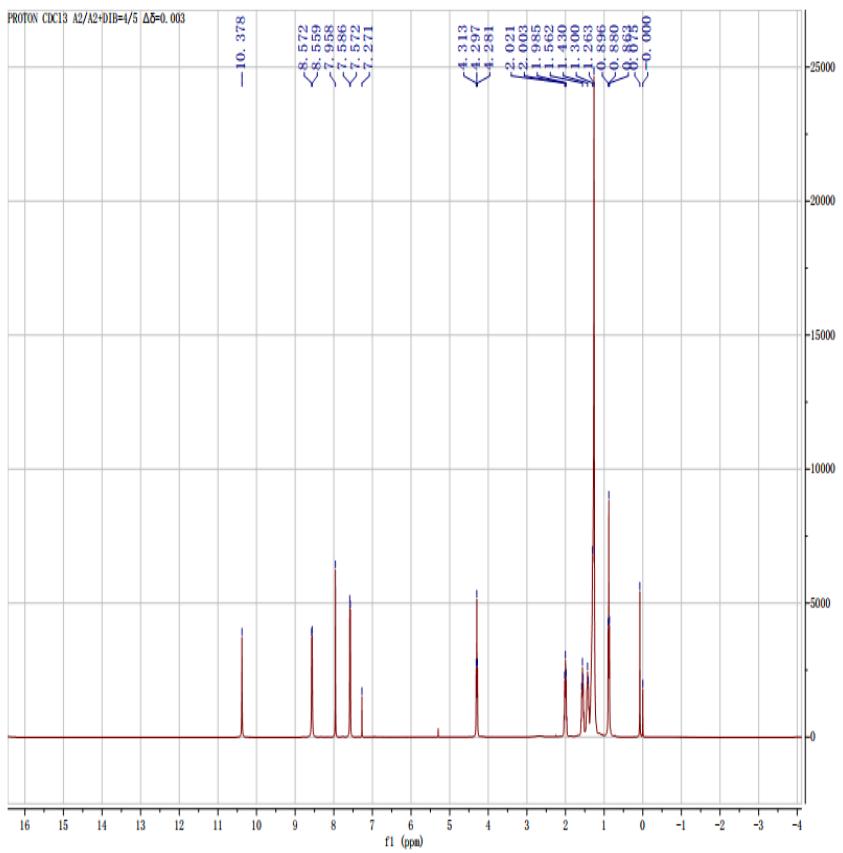


Fig. S18 ¹H NMR of A2/(A2+DIB)=1/5 Δδ=0.003

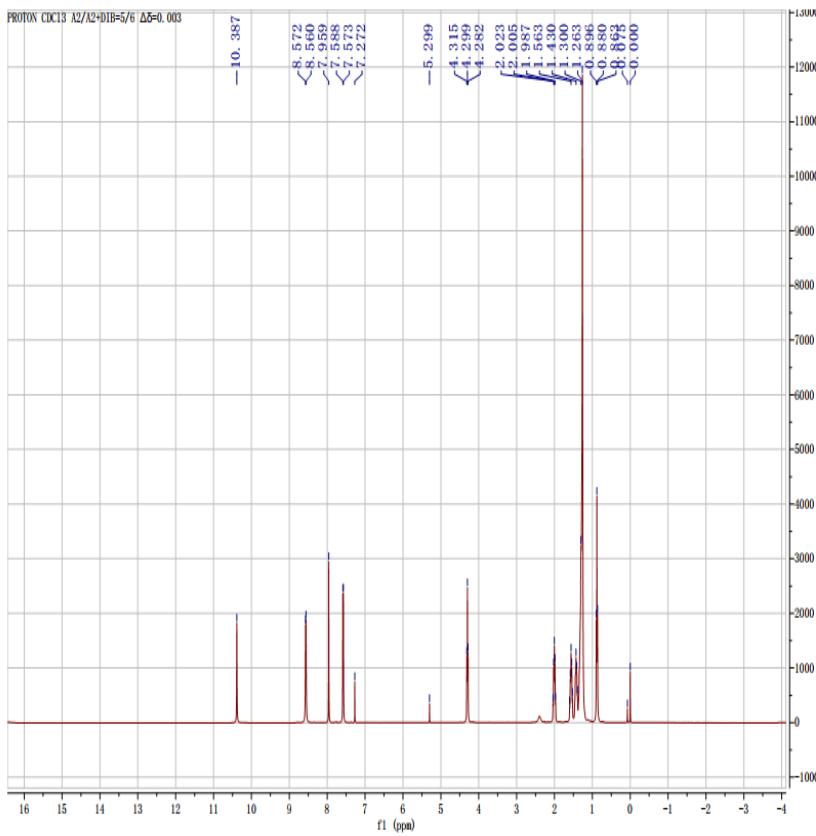


Fig. S19 ^1H NMR of A2/(A2+DIB)=5/6 $\Delta\delta=0.003$

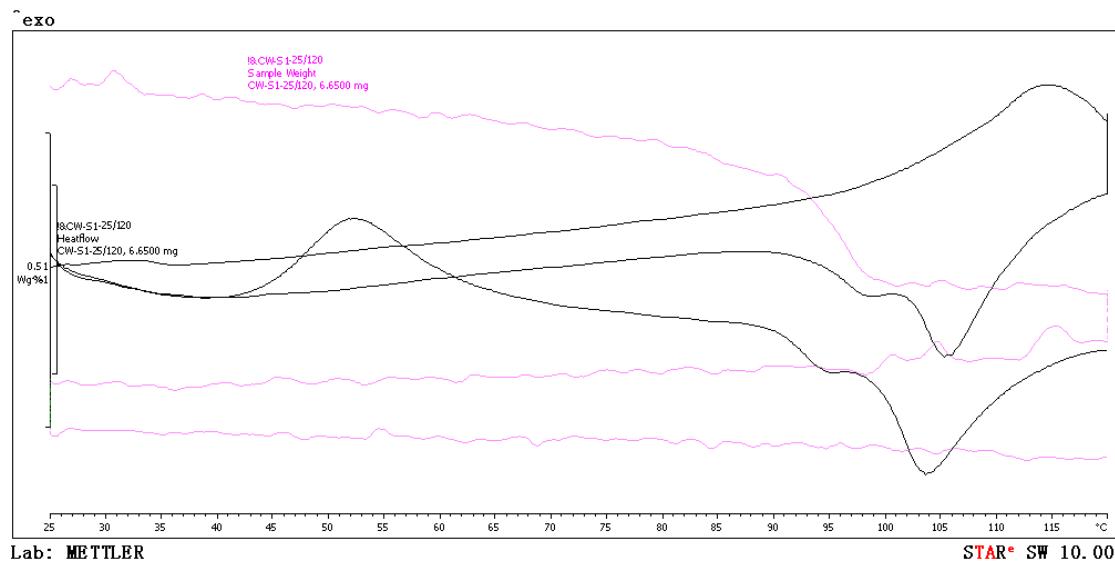


Fig. S20 TG and DSC curves of SB.

Table S1 Crystal data for complex A2-OMe-DIB

Identification code	A2-OMe-DIB
Empirical formula	C ₂₆ H ₁₈ F ₄ I ₂ N ₄ O ₄
Formula weight	780.24
Temperature/K	160.00(10)

Crystal system	triclinic
Space group	P-1
a/Å	8.3364(3)
b/Å	8.9922(2)
c/Å	9.09840(10)
$\alpha/^\circ$	82.895(2)
$\beta/^\circ$	73.539(2)
$\gamma/^\circ$	84.883(2)
Volume/Å ³	648.01(3)
Z	1
ρ_{calc} g/cm ³	1.999
μ/mm^{-1}	19.699
F(000)	376.0
Crystal size/mm ³	0.12 × 0.11 × 0.08
Radiation	CuK α ($\lambda = 1.54184$)
2 θ range for data collection/°	9.928 to 154.17
Index ranges	-10 ≤ h ≤ 10, -11 ≤ k ≤ 11, -11 ≤ l ≤ 11
Reflections collected	10921
Independent reflections	2624 [$R_{\text{int}} = 0.0373$, $R_{\text{sigma}} = 0.0232$]
Data/restraints/parameters	2624/0/182
Goodness-of-fit on F ²	1.064
Final R indexes [$I >= 2\sigma(I)$]	$R_1 = 0.0248$, $wR_2 = 0.0629$
Final R indexes [all data]	$R_1 = 0.0281$, $wR_2 = 0.0637$
Largest diff. peak/hole/e Å ⁻³	0.77/-1.01

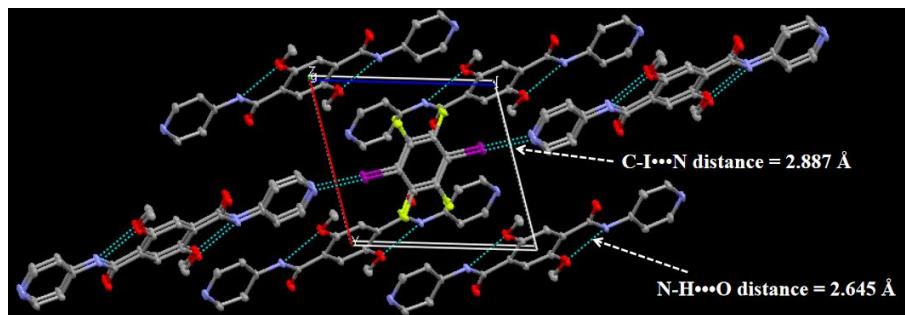


Fig. S21 Crystal-packing modes of A2-OMe-DIB exhibiting a C–I···N XB interaction with a distance of 2.887 Å and a N–H···O HB interaction with a distance of 2.645 Å