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Excellent Adsorption-Photocatalysis Synergistic Activity of 3D-3D Flower-like

BiOBr/Graphene Hydrogel Composite and the Removal of PBX

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Supporting information

Fig. S2 Adsorption efficiency of PBX by BiOBr and composites (The dosage of all adsorbents was 25 mg, initial concentration of PBX was 40 mg/L, the reaction temperature was 298 K)

Table :	S1 Fi	tting	adsorption	kinetic	models	of PBX	on	75%-Bi0) Br/Gł	н
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C _o (mg/L)	$q_{e(exp)}(mg/g)$	Pseudo-first-order model			Pseudo-second-order model		
		$q_{e(cal)}(mg/g)$	$K_1(min^{-1})$	R ²	$q_{e(cal)}(mg/g)$	$K_2(min^{-1})$	R ²
40mg/L	66.70	51.73	0.13	0.97145	74.46	0.0046	0.99926

	Langmuir isc	otherm		Freundlich isotherm			
Q _{max} (mg/g)	g) b(L/mg)		R ²	$K_F (mg^{1-1/n}L^{1/n}/g)$	n	R ²	
120	0.2416	0.29204	11.53	1.5941	0.96378		

Table S2 Fitting results of Langmuir and Freundlich isotherm



Fig. S3 Removal efficiency of PBX by dark reaction- photocatalysis



Fig. S4 Comparison of adsorption, photocatalysis, and synergistic efficiency of PBX by 75%-BiOBr /GH. (The dosage of 75%-BiOBr /GH was 25 mg, the initial concentration of PBX was 40 mg/L, the volume of PBX was 100 mL)



Fig. S5 XRD patterns of 75%-BiOBr/GH before and after 6th run cycle synergistic experiments



Fig.S6 FT-IR spectra of 75%-BiOBr/GH before use and after 6 recycled