

Synthesis and Spectroscopic Characterization of Ternary Copper(II) Complexes Containing Nitrogen and Oxygen Donors as Functional Mimics of Catechol Oxidase and Phenoxazinone Synthase

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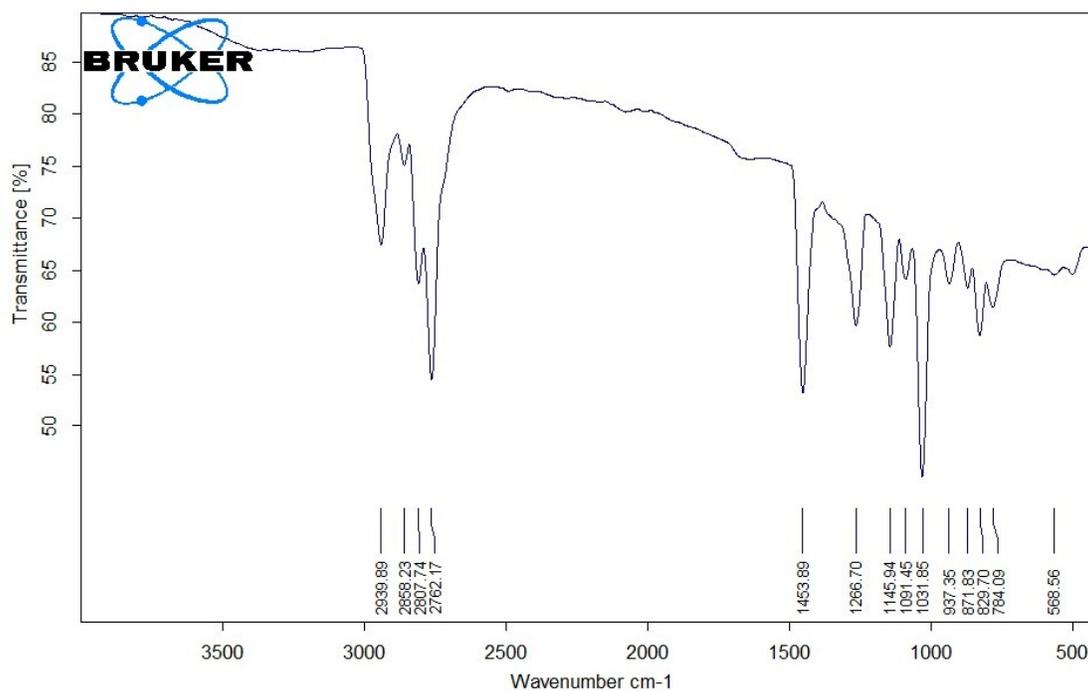
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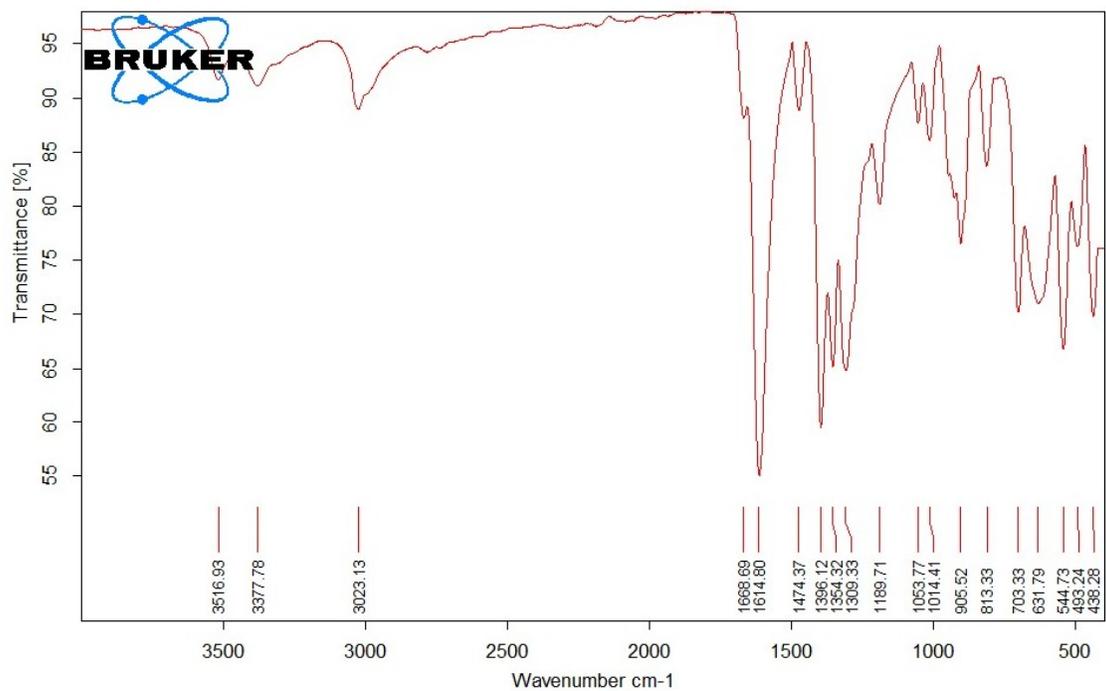
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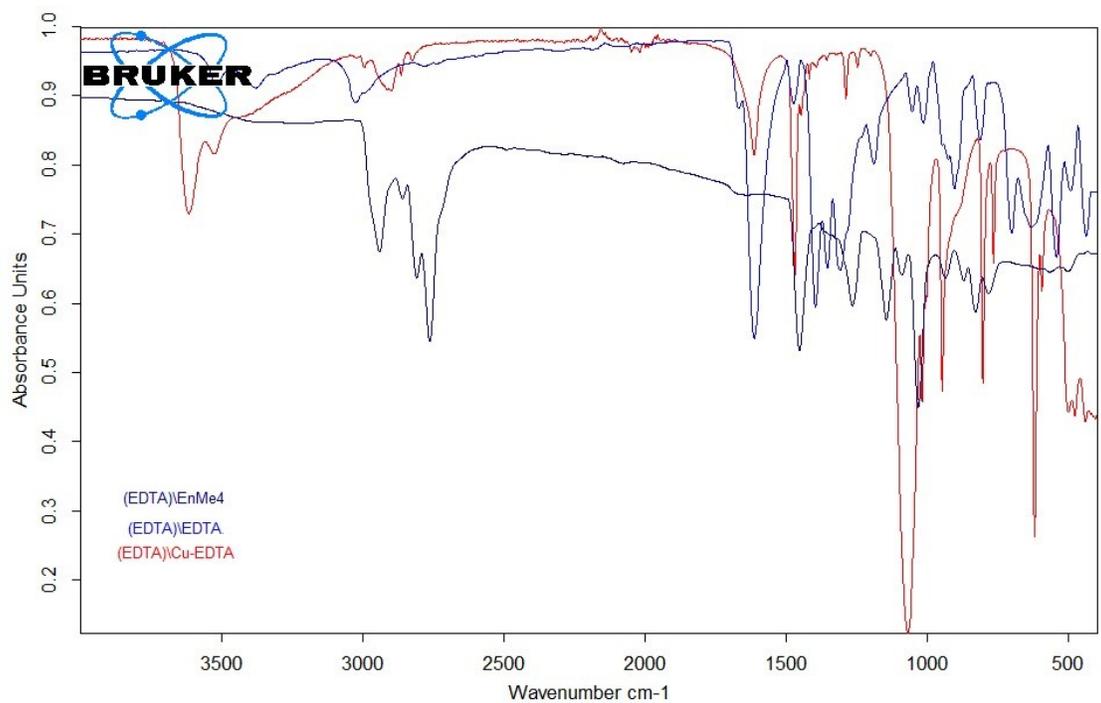
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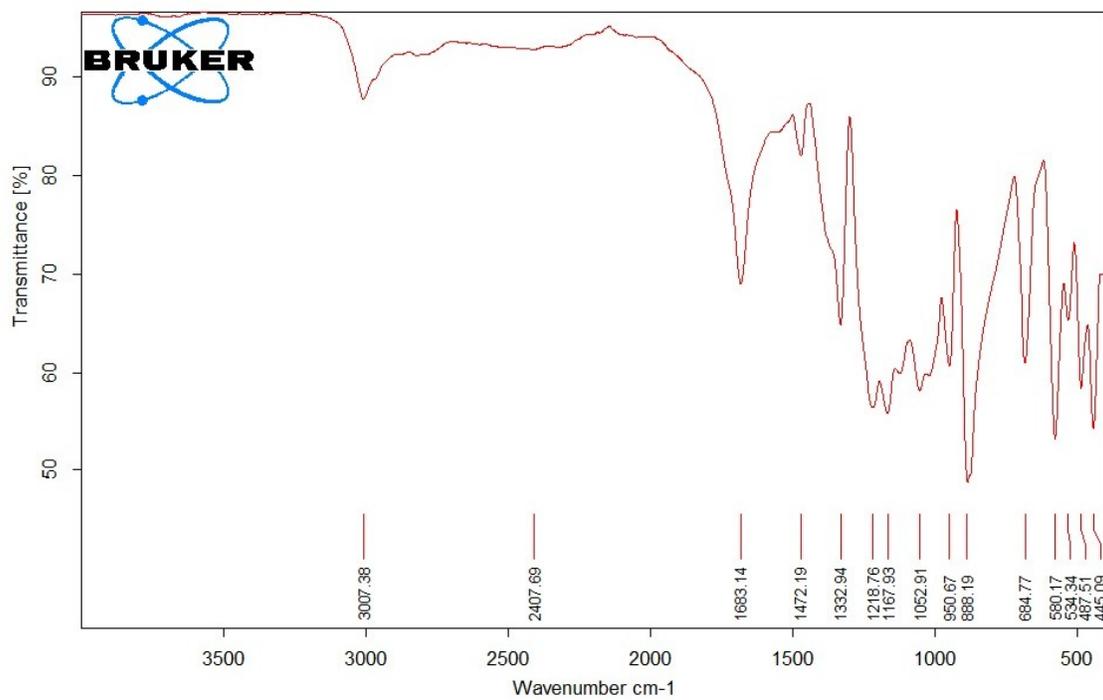
S1: FTIR spectrum of Me₄en



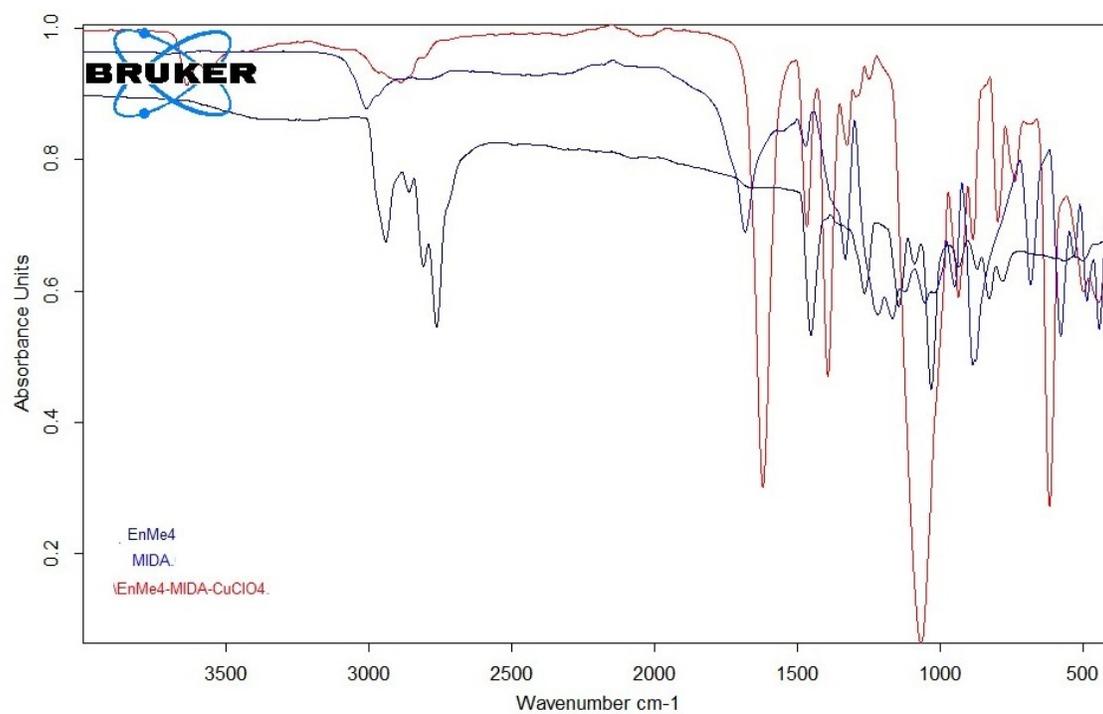
S2: FTIR spectrum of EDTA



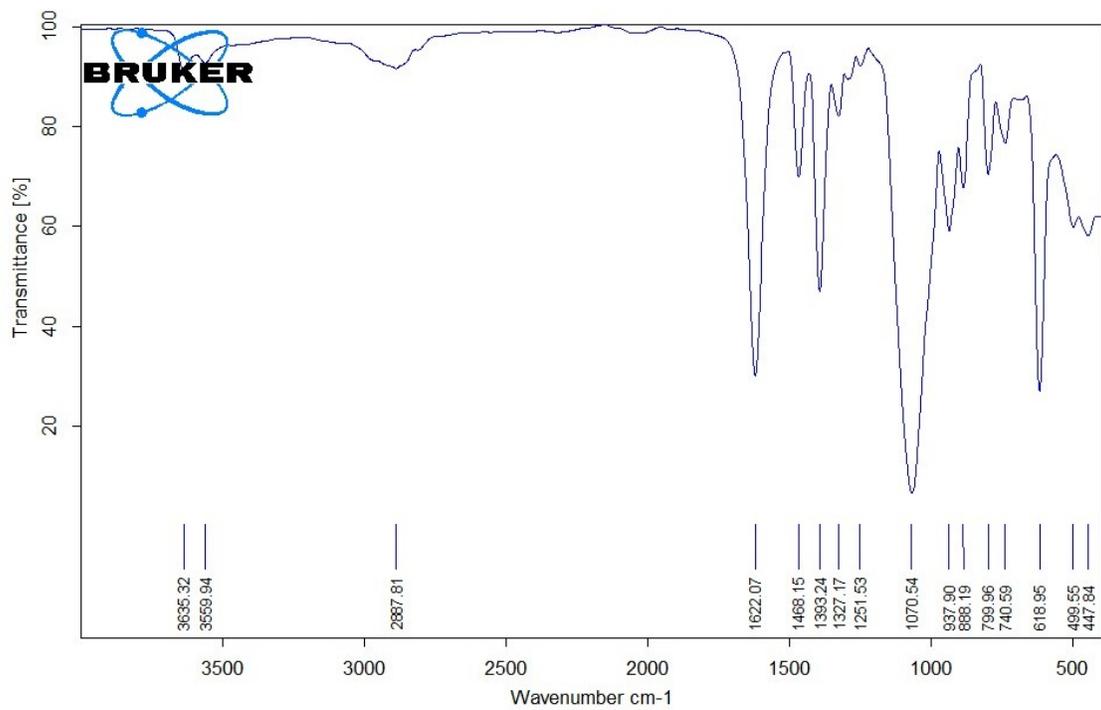
S3: Multi FTIR spectra of complex **1** and its ligand system L and L'



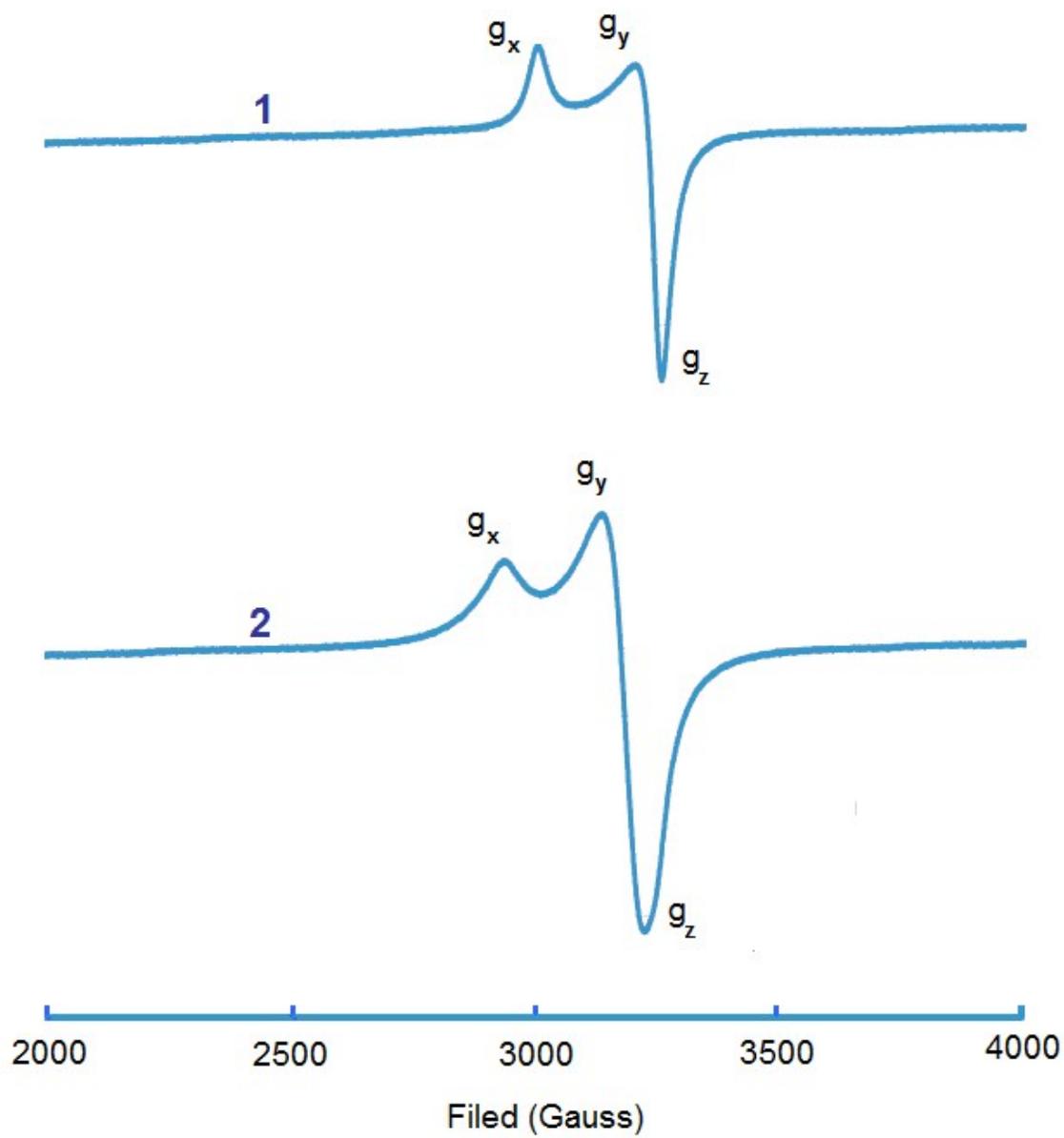
S4: FTIR spectrum of N-methyl iminodiacetic acid



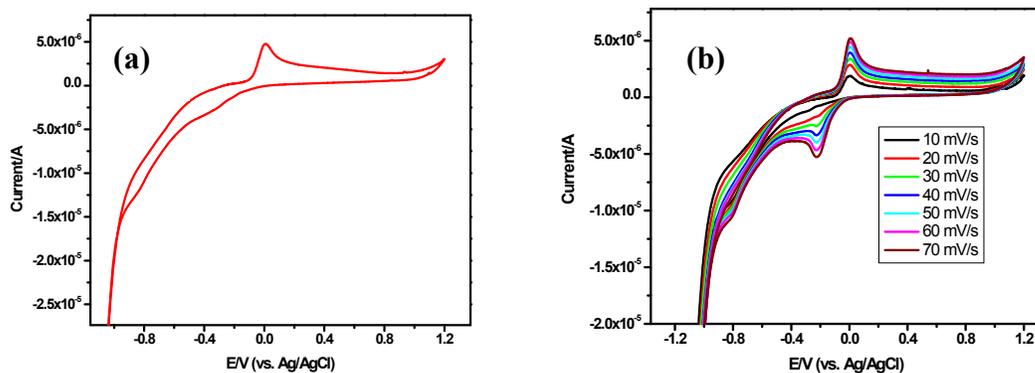
S5: Multi FTIR spectra of complex **2** and its ligand system L and L'



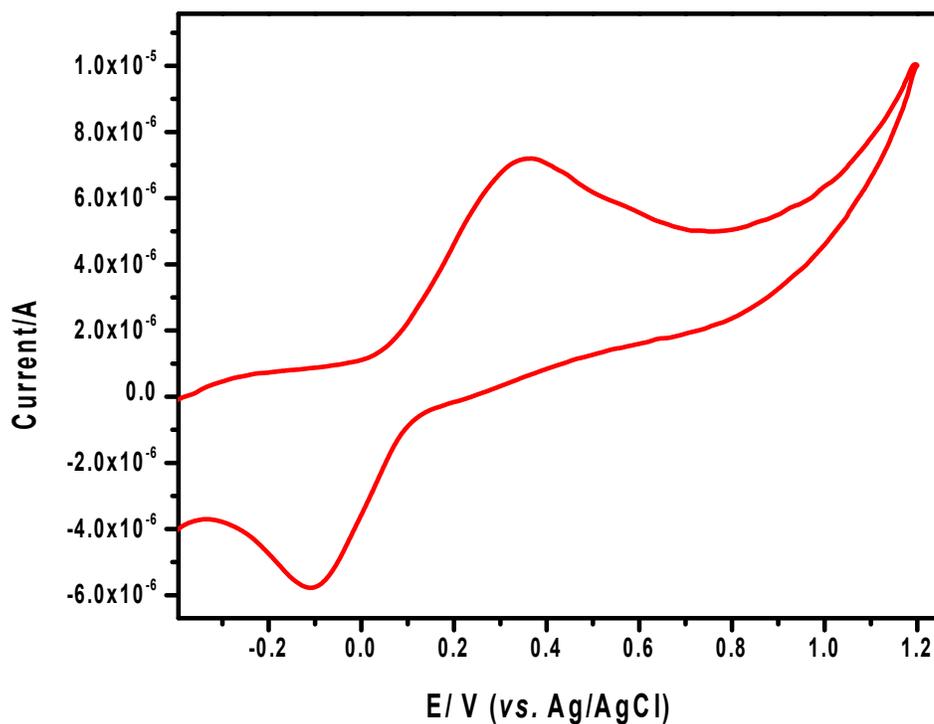
S6: FTIR spectrum of complex 2



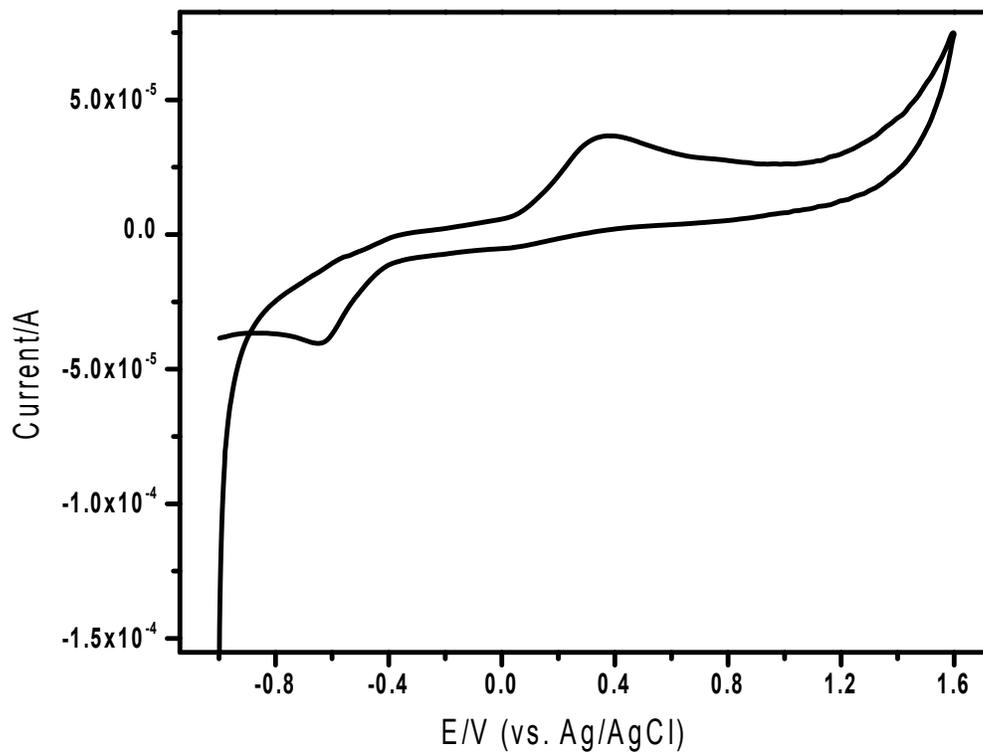
S7: ESR spectra of complexes 1 and 2



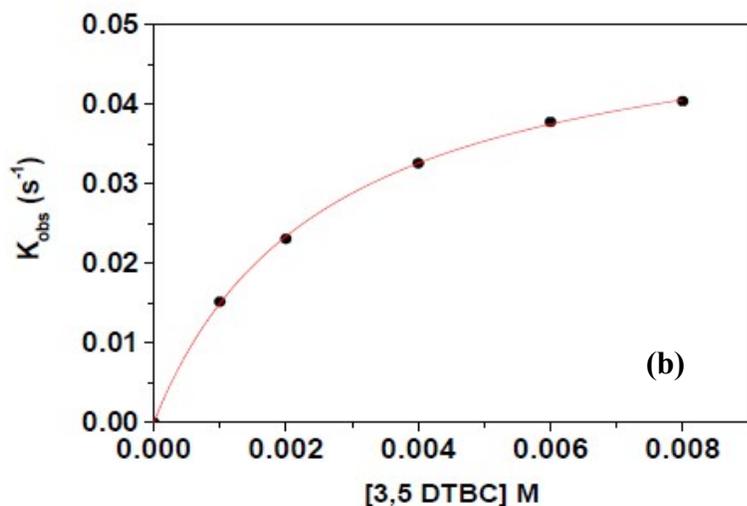
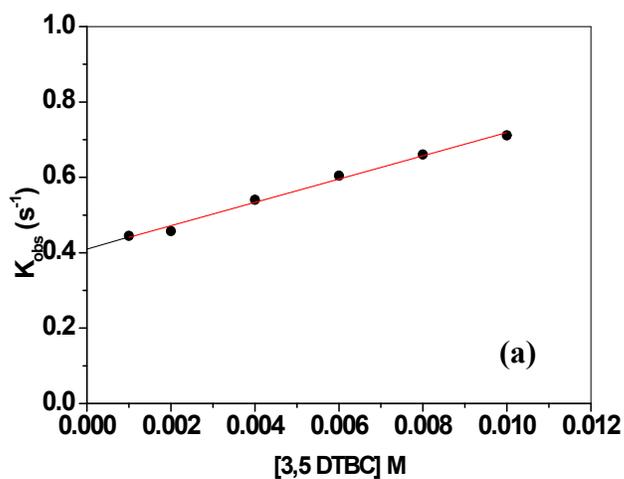
S8: (a) Cyclic voltammogram for complex **2** at CPE at phosphate buffer pH 7 using scan rate of 50 mV/s; (b) Effect of scan rate on the peak current height of 1×10^{-3} M complex **2** using cyclic voltammetry at CPE at phosphate buffer pH 7



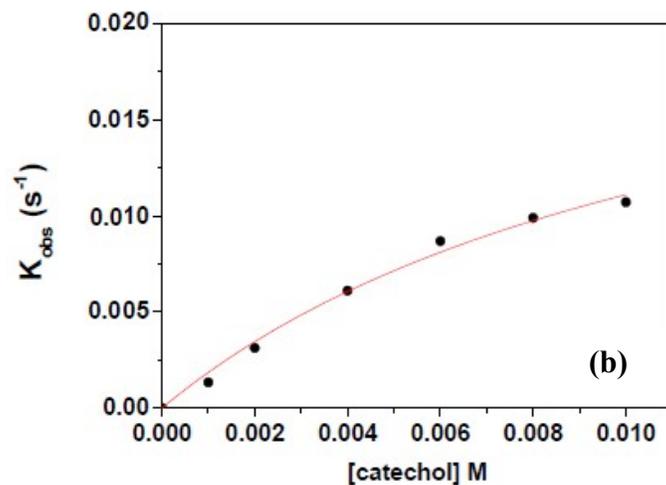
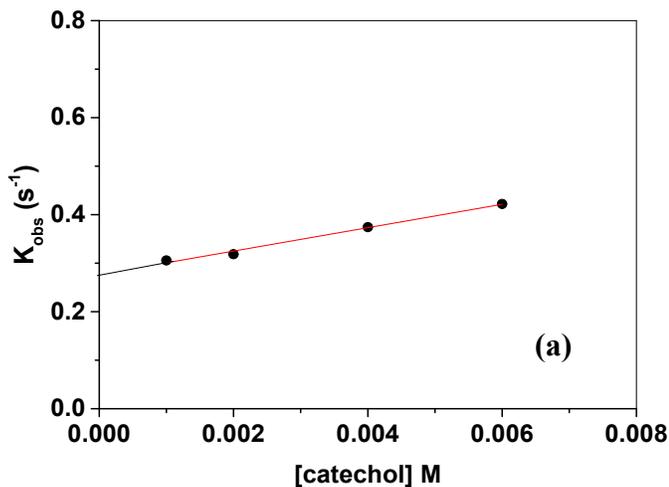
S9: Cyclic voltammogram for 1×10^{-3} M 3,5-DTBCCH₂ at CPME at phosphate buffer pH 7 using scan rate of 50 mV/s.



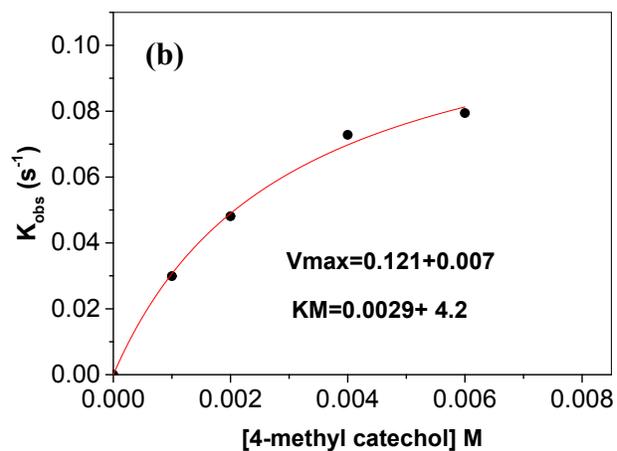
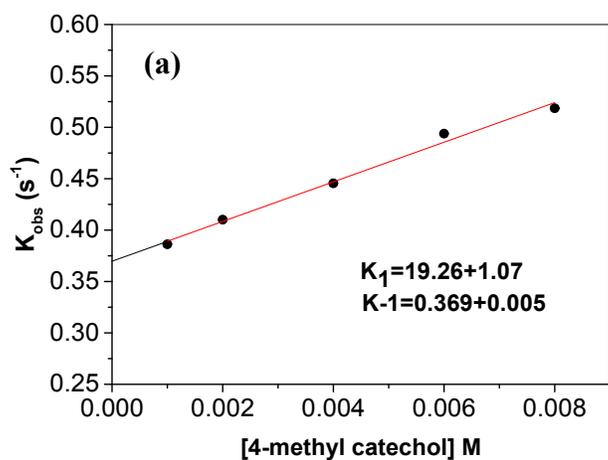
S10: Cyclic voltammogram for 1×10^{-3} M *o*-APH₃ at CPME at phosphate buffer pH 7 using scan rate of 50 mV/s



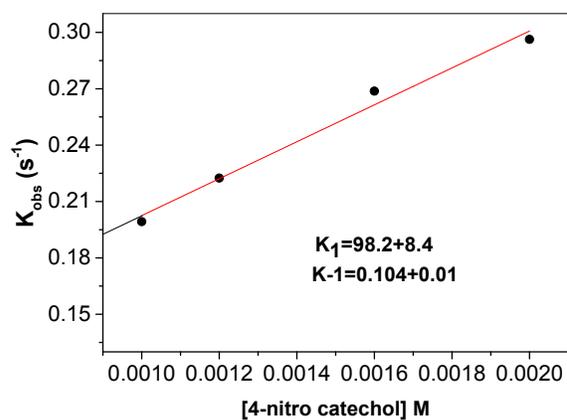
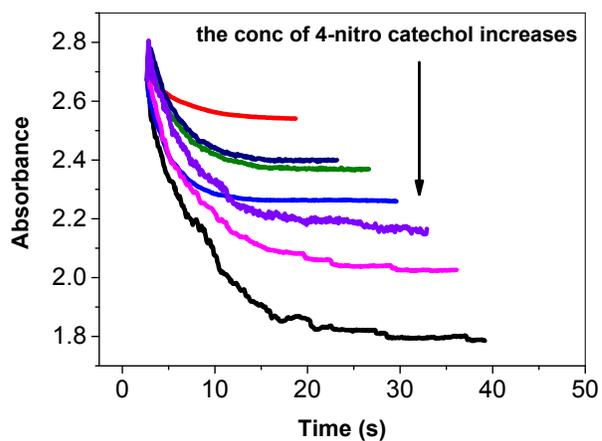
S11: Dependence of the initial rate on the concentration of the substrate for the oxidation reaction of 3,5-DTBC₂ catalyzed by complex **2** in methanol; (a) is the first step and (b) is the second step. The concentration of complex was 1.0×10^{-4} M and the reaction was followed at 400 nm



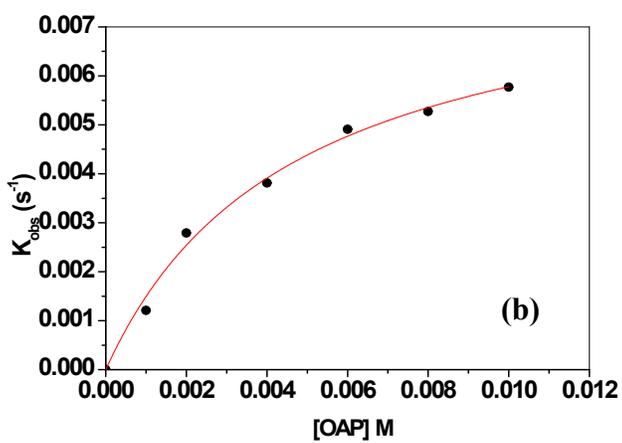
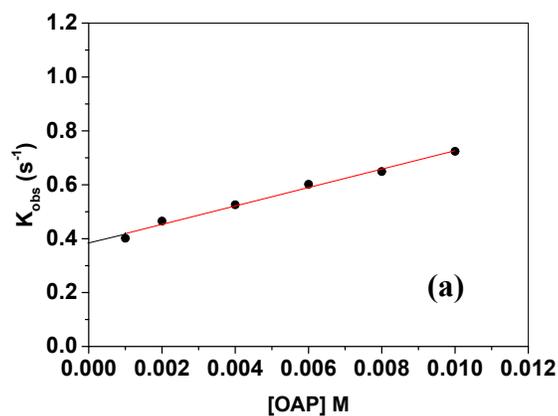
S12: Dependence of the initial rate on the concentration of the substrate for the oxidation reaction of CatH₂ catalyzed by complex **1** in methanol; (a) is the first step and (b) is the second step. The concentration of complex was 1.0×10^{-4} M and the reaction was followed at 390 nm



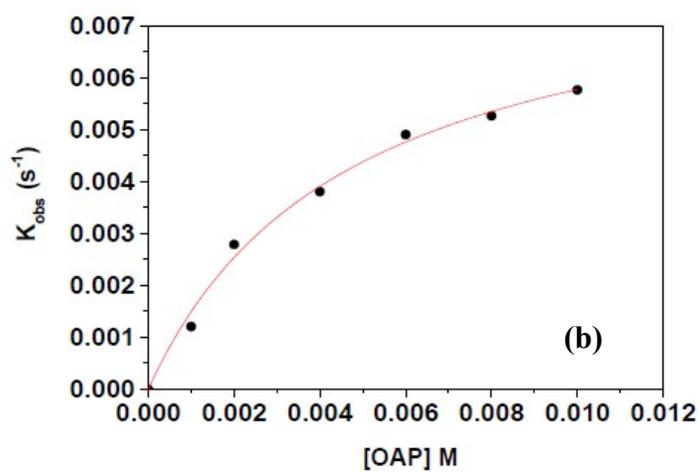
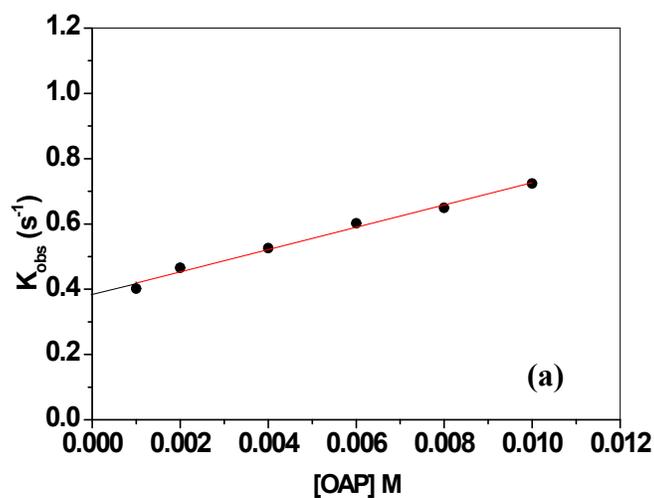
S13: Dependence of the initial rate on the concentration of the substrate for the oxidation reaction of 4-CH₃-CatH₂ catalyzed by complex **1** in methanol; (a) is the first step and (b) is the second step. The concentration of complex was 1.0×10^{-4} M and the reaction was followed at 400 nm



S14: Dependence of the initial rate on the substrate concentration for the oxidation reaction of 4- NO₂-CatH₂ catalyzed by complex **1** in methanol for the first step; the concentration of complex was 1.0×10^{-4} M and the reaction was followed at 400 nm



S15: Dependence of the initial rate for on the concentration of the substrate for the oxidation reaction of *o*-APH₃ catalyzed by complex **2** in methanol; (a) is the first step and (b) is the second step. The concentration of complex was 1.0×10^{-4} M and the reaction was followed at 433 nm



S16: Dependence of the initial rate for on the concentration of the substrate for the oxidation reaction of *o*-APH₃ catalyzed by complex **2** in methanol; (a) is the first step and (b) is the second step. The concentration of complex was 1.0×10^{-4} M and the reaction was followed at 433 nm