

[Electronic Supplementary Information (ESI)]

Photocatalytic properties of hybrid materials based on multicharged polymer matrix with encored TiO_2 and noble metals (Pt, Pd or Au) nanoparticles†

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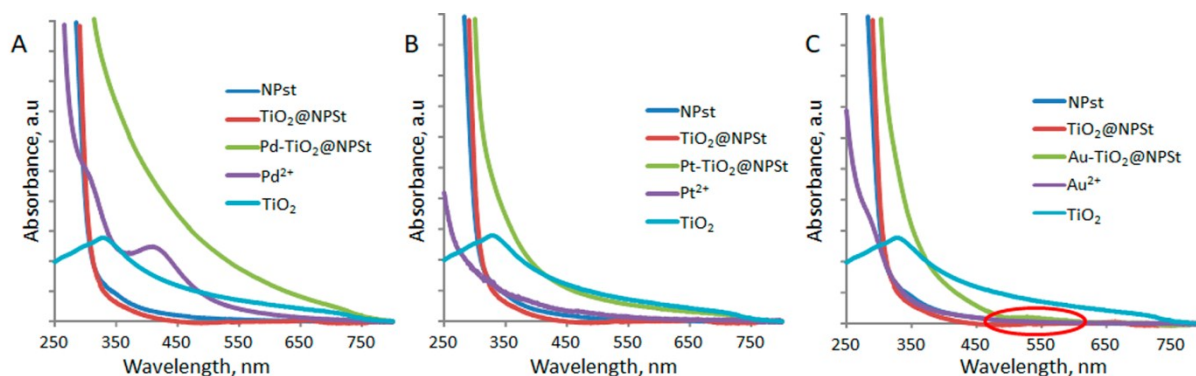


Fig. S1. UV-vis spectra of NPst (blue), TiO_2 (cyan), salts (Pd^{2+} (A), Pt^{2+} (B), Au^{2+} (C)) (purple), TiO_2 @NPst (red), of MNPs- TiO_2 @NPst (MNPs = Pd^0 , Pt^0 , Au^0) (green), H_2O , 20 °C, $l = 0.5$ cm.

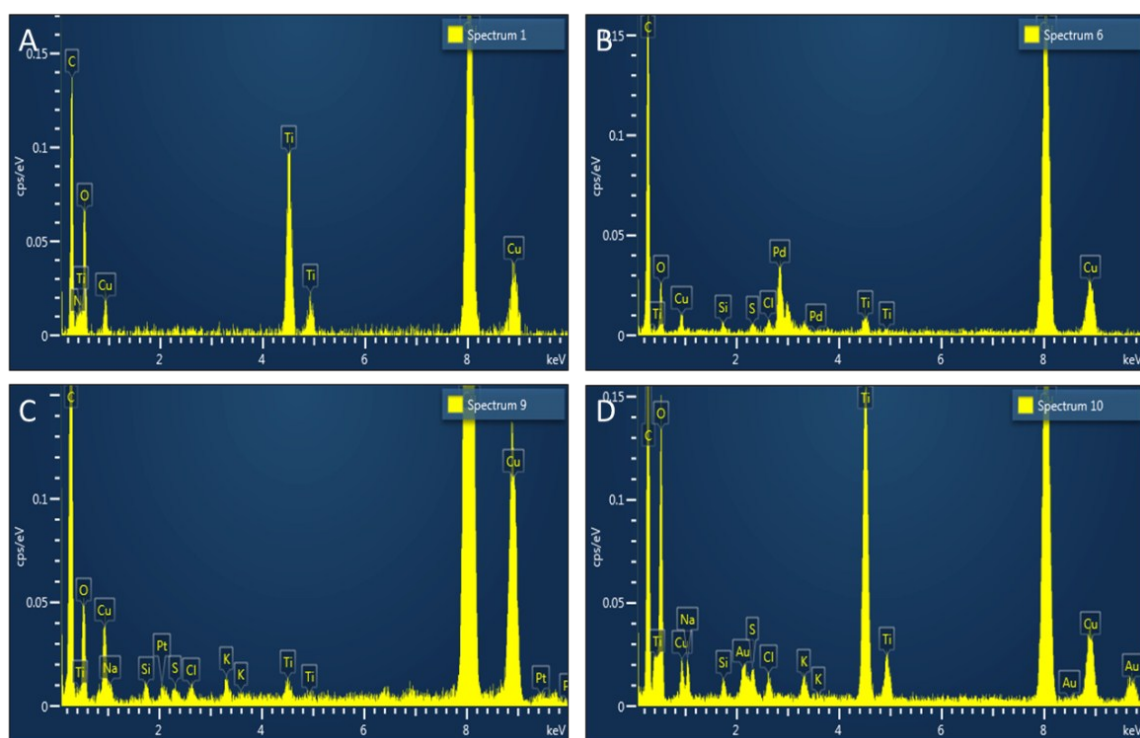


Fig S2. EDX profile of (A) NPst; (B) TiO_2 @NPst; (C) Pd- TiO_2 @NPst; (D) Pt- TiO_2 @NPst; (E) Au- TiO_2 @NPst.

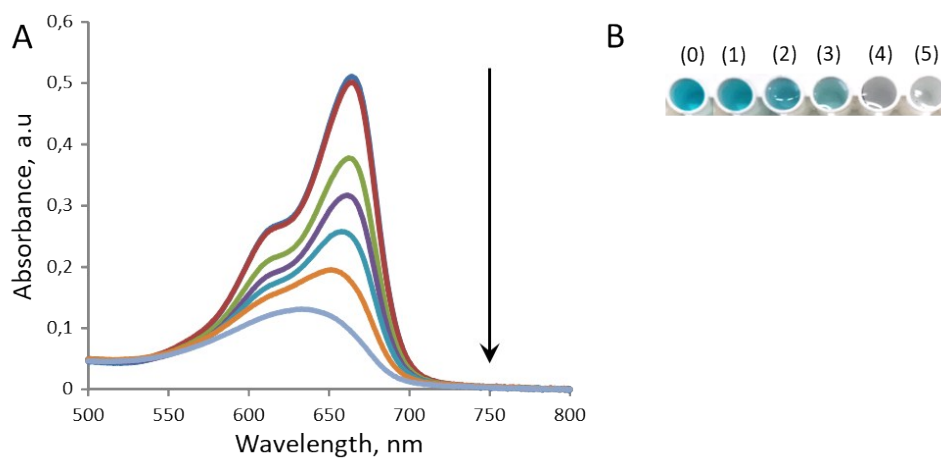


Fig. S3. (A) UV-vis spectral changes of MB aqueous solution over Pt- TiO_2 @NPst under UV irradiation (364 nm) at different time intervals; (B) different colors of MB in 130 min after UV irradiation (364 nm) over composite: (0) without composite, (1) TiO_2 , (2) TiO_2 @NPst, (3) Pd- TiO_2 @NPst, (4) Pt- TiO_2 @NPst, (5) Au- TiO_2 @NPst H_2O , 24 °C, $C(\text{MB}) = 1.6 \text{ mg/l}$, $C(\text{TiO}_2 \text{ in composites}) = 0.613 \text{ mg/l}$, $l = 1 \text{ cm}$

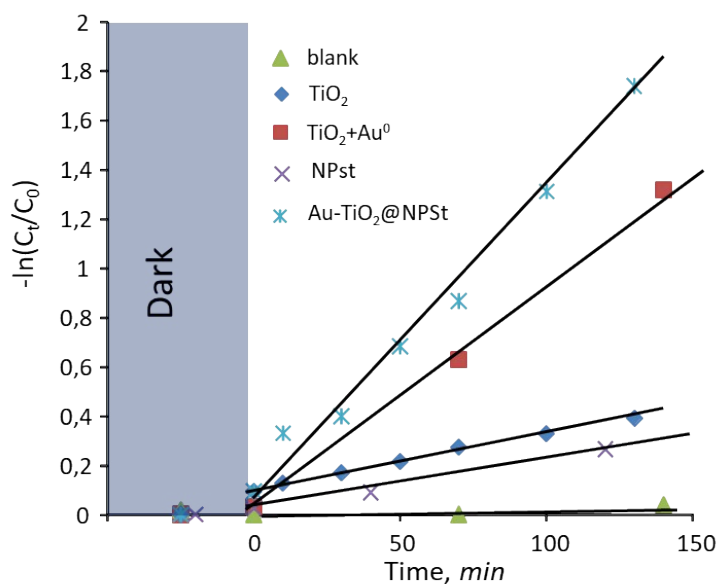


Fig. S4. Plot of $-\ln(C_t/C_0)$ vs. irradiation time under UV. $C(\text{MB}) = 1.6 \text{ mg/l}$, $C(\text{TiO}_2 \text{ in composites}) = 0.613 \text{ mg/l}$, $V = 4 \text{ ml}$.

Table S1. The photodegradation of MB by TiO_2 , TiO_2+Au^0 , NPst, $\text{Au-TiO}_2@\text{NPst}$ and without catalyst; $C(\text{MB}) = 1.6 \text{ mg/l}$, $C(\text{TiO}_2 \text{ in composites}) = 0.613 \text{ mg/l}$, $V = 4 \text{ ml}$, 24°C , H_2O .

catalyst	rate constants (min^{-1})
-	0.0001
TiO_2	0.0023
$\text{TiO}_2 + \text{Au}^0$	0.0078
NPst	0.002
$\text{Au-TiO}_2@\text{NPst}$	0.0122