

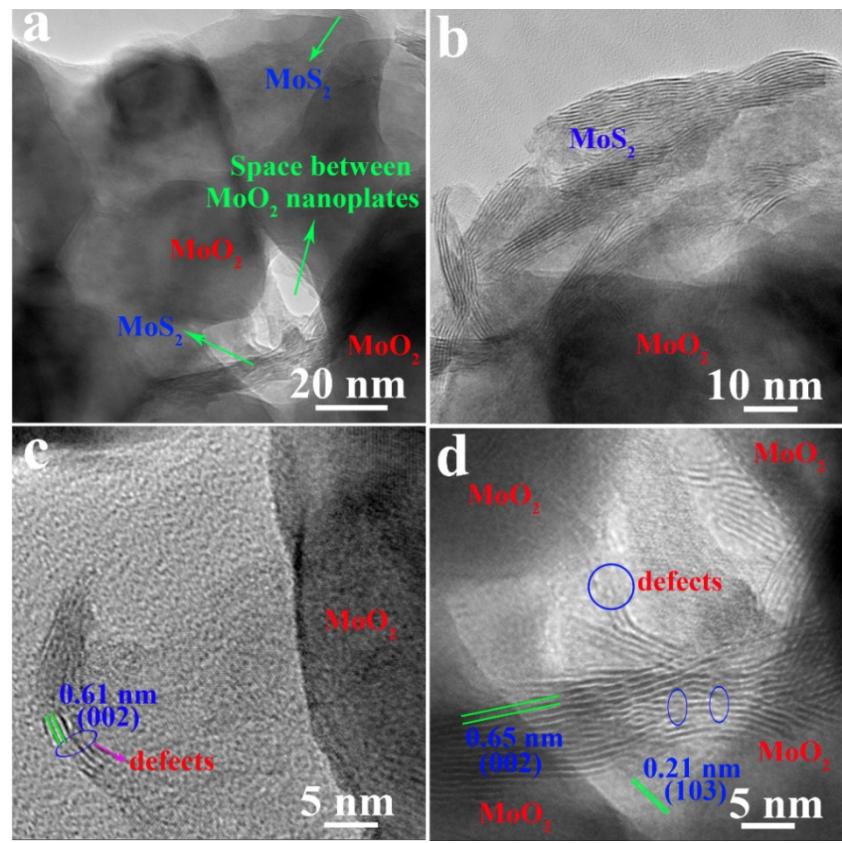
# Controllable synthesis of MoS<sub>2</sub>@MoO<sub>2</sub> nanonetwork to boost NO<sub>2</sub> room temperature sensing in air

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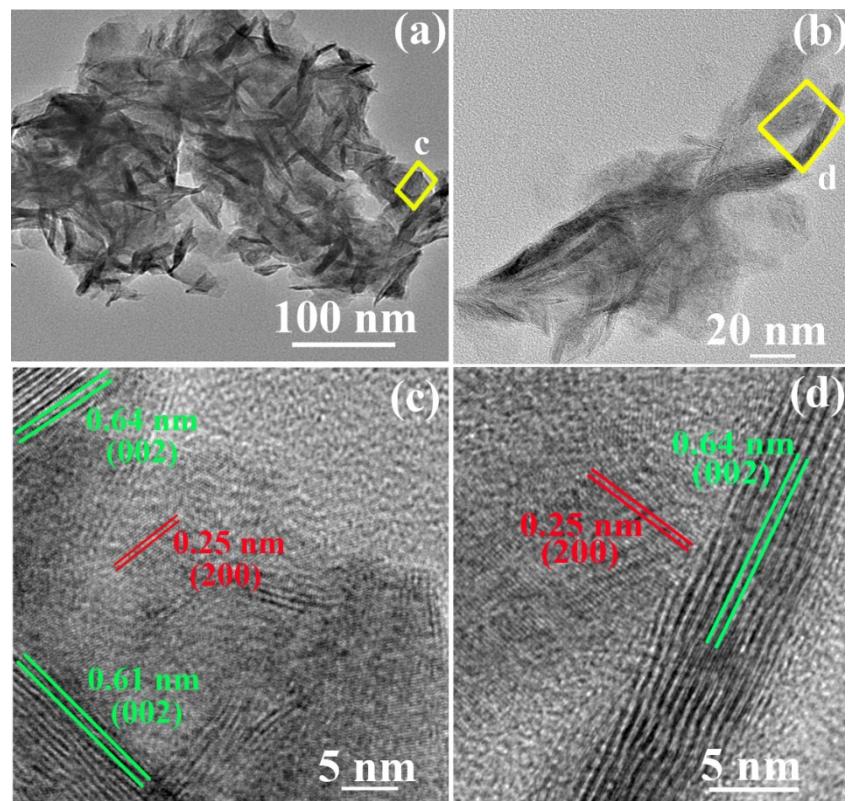
**Abstract:** the supporting information provided shows: (i) TEM and HRTEM images of the as prepared MSO-1 nanocomposite, (ii) TEM and HRTEM images of MSO-3 nanocomposite, (iii), (iv), (v) TEM and HRTEM images of the flower like MoS<sub>2</sub> NSs, thin MoO<sub>2</sub> and thin MoO<sub>3</sub> nanoplates respectively, (vi) SEM images of the flower like MoS<sub>2</sub> NSs and of the MoO<sub>3</sub> thin nanoplates, (vii) EDS spectra and mapping of the MSO-2, (viii) XRD pattern of the MoO<sub>3</sub> nanoplates, (ix) XPS analysis of the pure MoS<sub>2</sub> NSs, (x) Fitted impedance parameters of the samples (xi) stability of the MSO-2 and of the pure MoS<sub>2</sub> NSs in air, (xii) response-recovery curve of the as fabricated samples, (xiii) calibration curve of MSO-2, and response/recovery time all of the sensors, (xiv) humidity effect, (xv) table of the comparative response and response/recovery time all of the samples, (xvi) Band diagram of the MoS<sub>2</sub> and MoO<sub>2</sub> before contact, (xvii) comparison of the present study with reported work.

## Table of contents

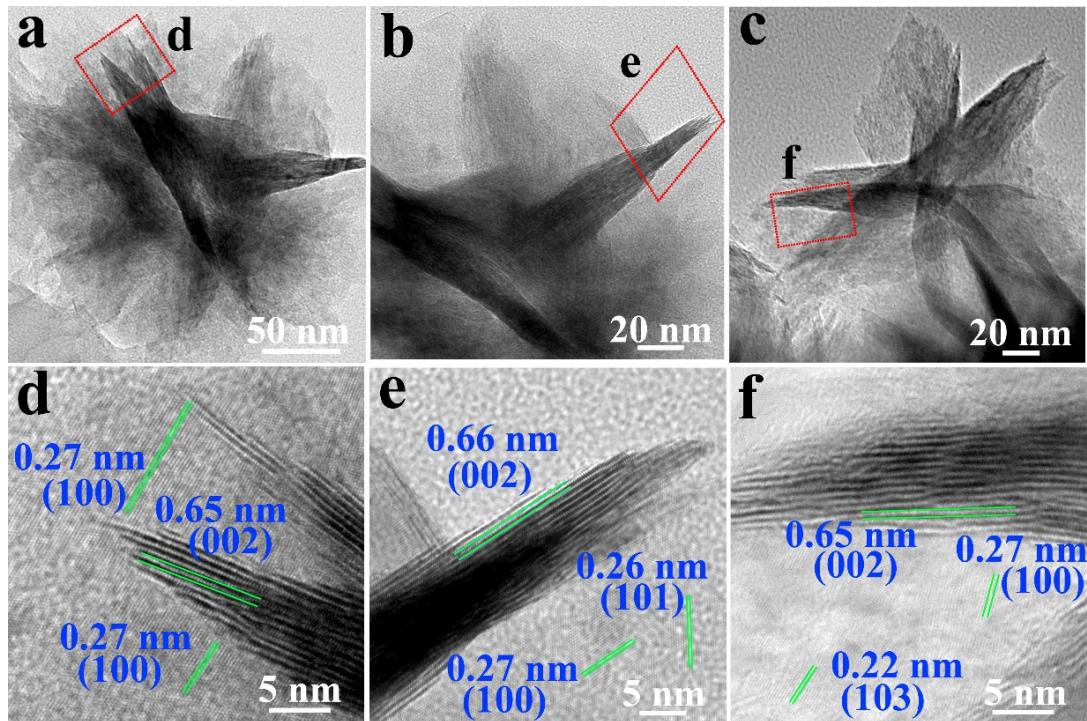
- S1 Cover page and abstract
- S2 **Table of contents**
- S3 **Fig. S1:** TEM/HRTEM images of MSO-1 to show the edge growth MoS<sub>2</sub> NSs.
- S4 **Fig. S2:** TEM/HRTEM images of MSO-1 to indicated the high concentration of MoS<sub>2</sub> NSs
- S5 **Fig. S3:** TEM/HRTEM the flower like MoS<sub>2</sub> NSs to show complete sulfurization of the MoO<sub>3</sub> nanoplates
- S6 **Fig. S4, Fig S5:** TEM/HRTEM of the thin plates MoO<sub>2</sub> and MoO<sub>3</sub>to represent high crystallinity.
- S7 **Fig. S6, Fig. S7:** SEM images of flower like MoS<sub>2</sub> NSs, and thin nanoplates of MoO<sub>3</sub>, EDS spectra and mapping of the MSO-2 nanonetworks.
- S8 **Fig. S8, Fig S9:** XRD pattern of MoO<sub>3</sub> nanoplates, XPS spectra of Mo 3d and S 2p of the pure MoS<sub>2</sub> NSs.
- S9 **Table S1, Fig S10:** Fitted impedance parameters of samples, comparative stability of MSO-2 and of the pristine MoS<sub>2</sub> NSs in air.
- S10 **Fig. S11:** Response-recovery curve to different NO<sub>2</sub> concentration at RT in air.
- S11 **Fig. S12, Fig S13:** calibration curve of MSO-2 to 0.1-100 ppm NO<sub>2</sub>, response/recovery time graphs all of the sensors.
- S12 **Table S2, Scheme S1:** response, response/recovery time values all of the sensors to NO<sub>2</sub> gas, Band diagram of MoS<sub>2</sub> and MoO<sub>2</sub> before contact.
- S13 **Table S3:** comparison of the present study gas sensing performance with reported work.
- S14 **Reference:**



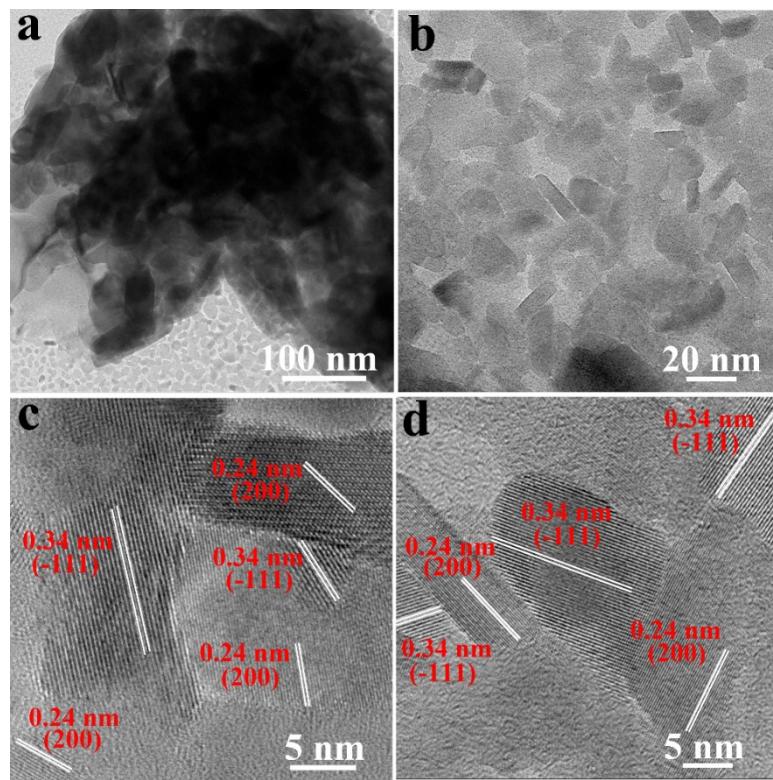
**Fig. S1** (a) TEM image; (b-d) HRTEM images of MSO-1 nanocomposite (green line:  $\text{MoS}_2$ ). HRTEM images showing defects and heterostructure interfaces).



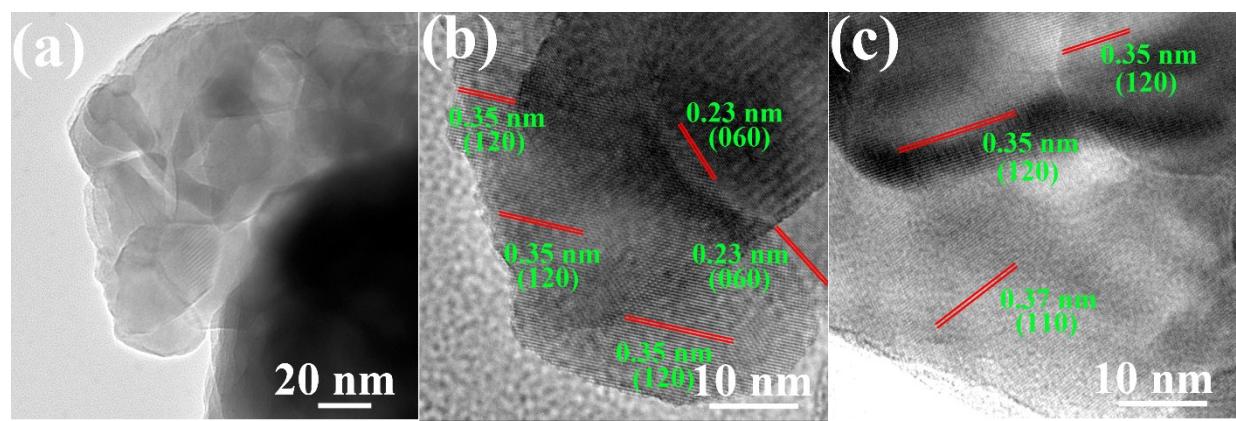
**Fig. S2** (a, b) TEM images (c, d) HRTEM images of MSO-3 nanocomposite (green line:  $\text{MoS}_2$ ; red line:  $\text{MoO}_2$ ).



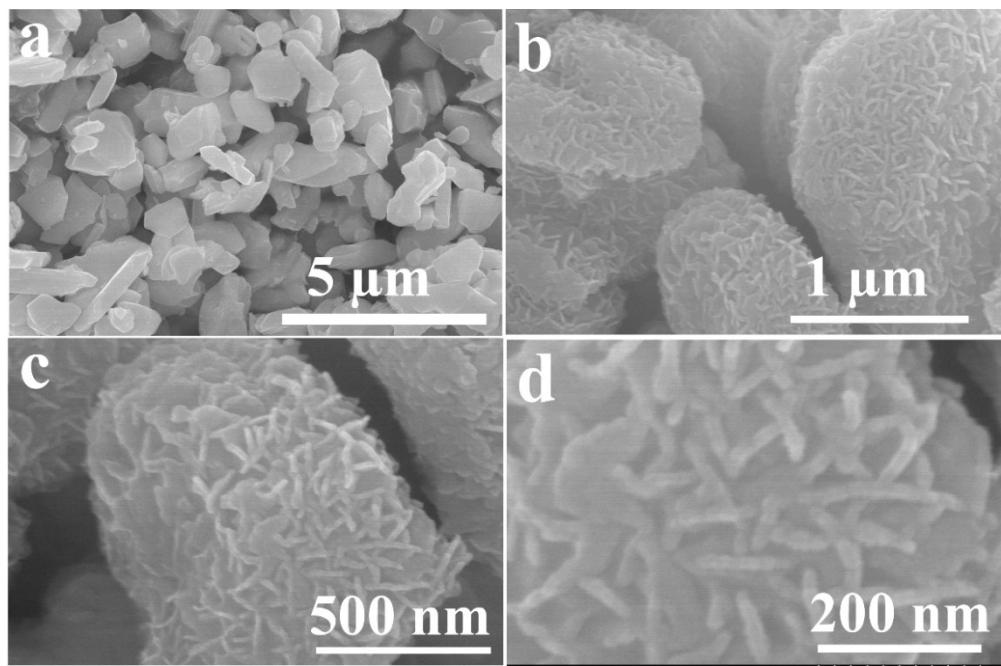
**Fig. S3** (a-c) TEM images of flower like pure MoS<sub>2</sub> NSs; (d-f) HRTEM images of (a), (b) and (c).



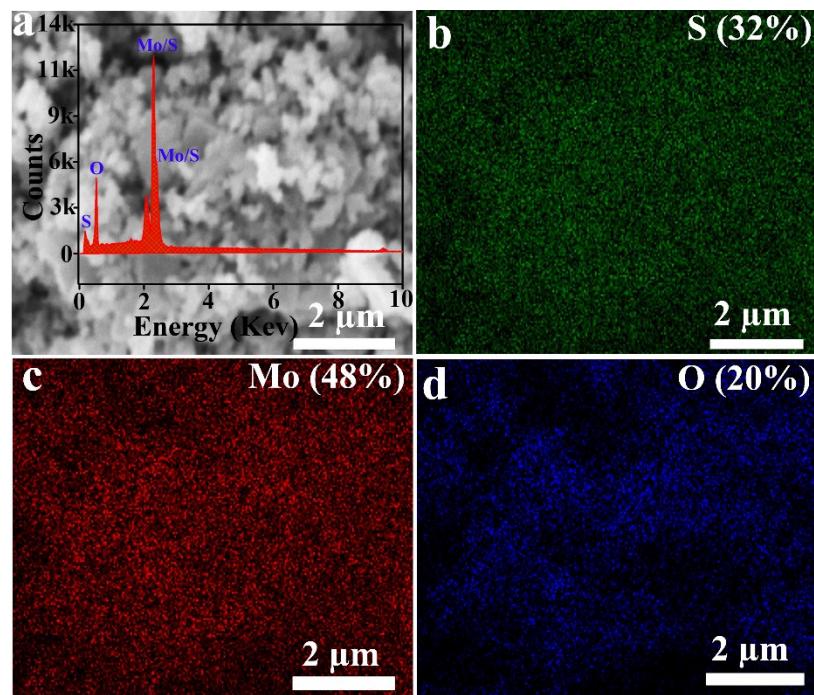
**Fig. S4** (a, b) TEM images of pristine MoO<sub>2</sub> thin nanoplates; (c, d) HRTEM images of (a) and (b).



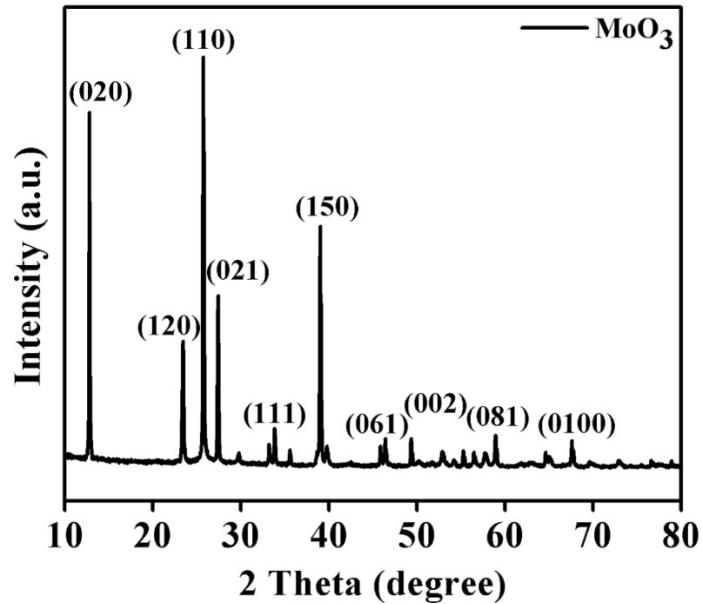
**Fig. S5** (a) TEM images of MoO<sub>3</sub> thin nanoplates; (c, d) HRTEM images.



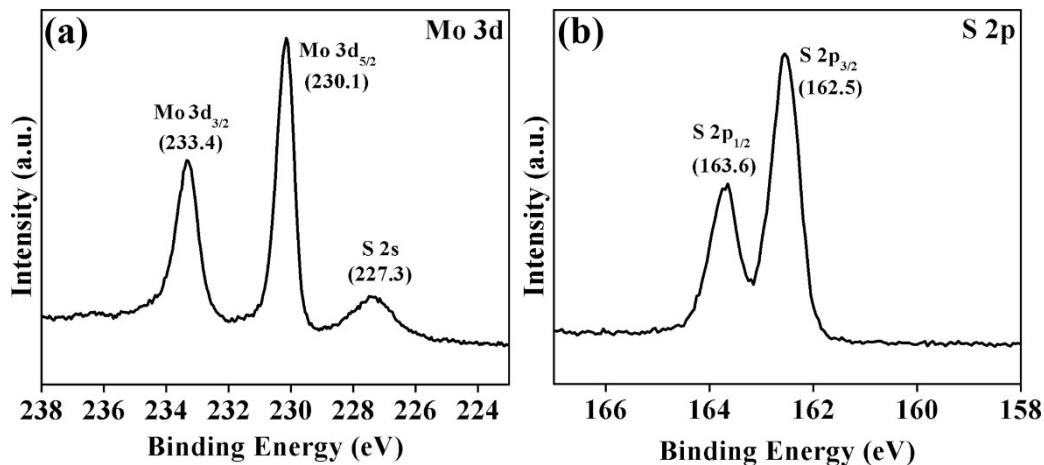
**Fig. S6** SEM images of (a) Pure  $\text{MoO}_3$ ; (b-d)  $\text{MoS}_2$  NSs (which were formed with sulfurization of  $\text{MoO}_3$  for 2 h).



**Fig. S7** (a) SEM image, EDS spectra (inset in (a)) and elemental mapping of MSO-2.



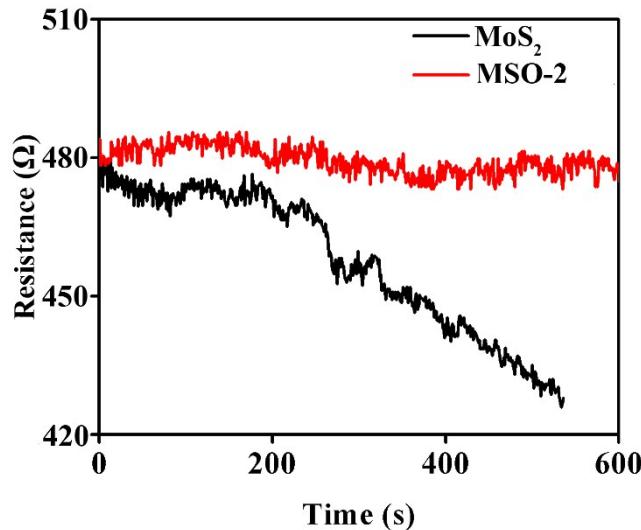
**Fig. S8** XRD diffraction pattern of  $\text{MoO}_3$ .



**Fig. S9 (a, b)** XPS spectra of Mo 3d and S 2p, respectively of the pure  $\text{MoS}_2$  NSs.

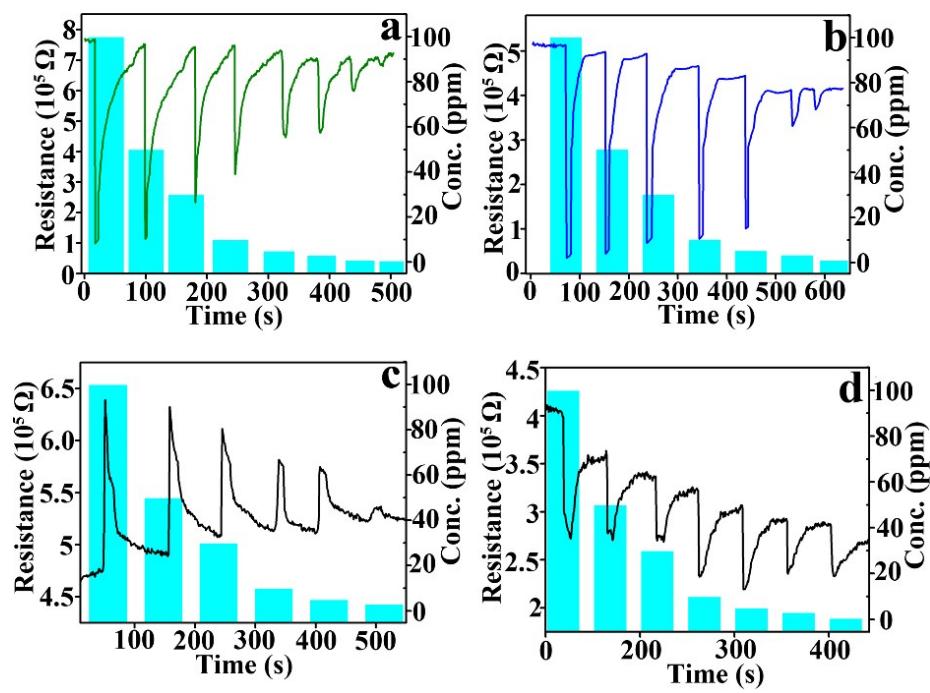
**Table S1.** Fitted impedance parameters of samples

Samples	MoS <sub>2</sub>	MoO <sub>2</sub>	MSO-2
R <sub>Ω</sub> (Ω)	74.4	69.1	45.3
R <sub>ct</sub> (Ω)	7336	5145	1041

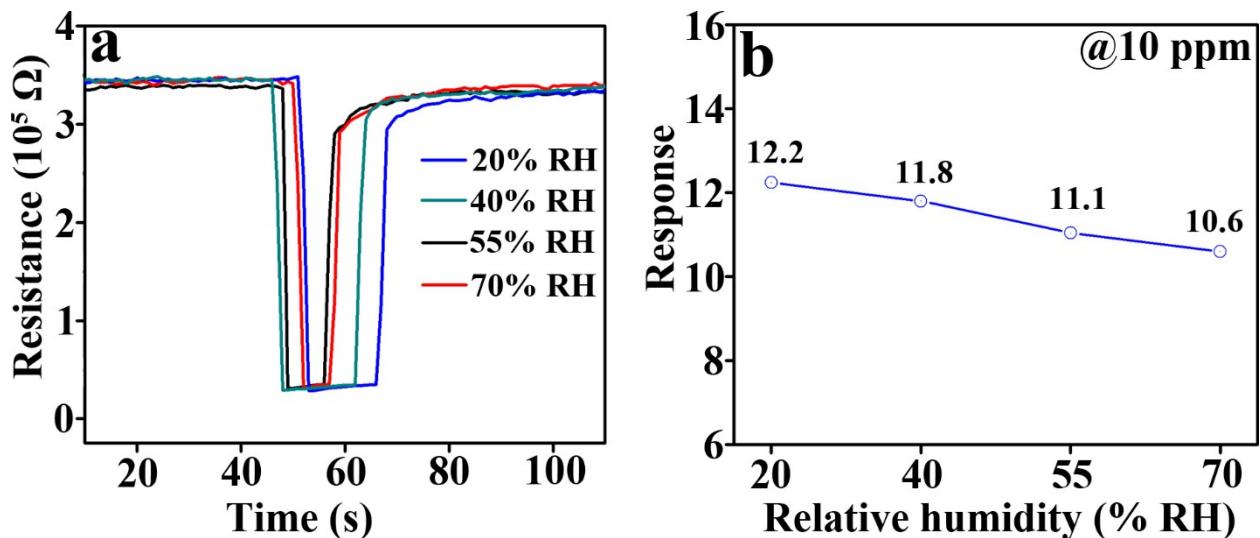
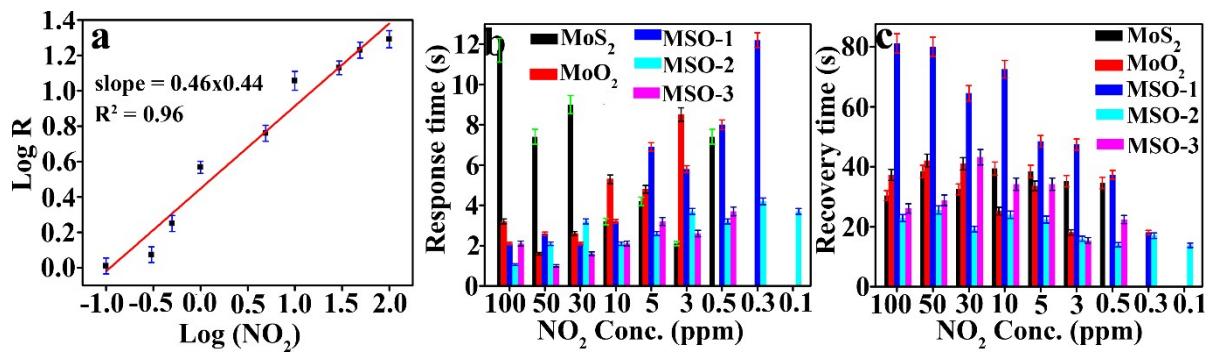


**Fig. S10** The resistance of the MSO-2 sensor compared with pure MoS<sub>2</sub> sensor in air at RT, indicating the high stability of MSO-2 nanonetworks sensor compare to pure MoS<sub>2</sub> sensor.

As shown in Fig. S10, the pure MoS<sub>2</sub> is highly unstable in air and also in NO<sub>2</sub> atmosphere at RT, and the resistance of the devices increasingly shifting to lower compared to MSO-2 nanonetworks.



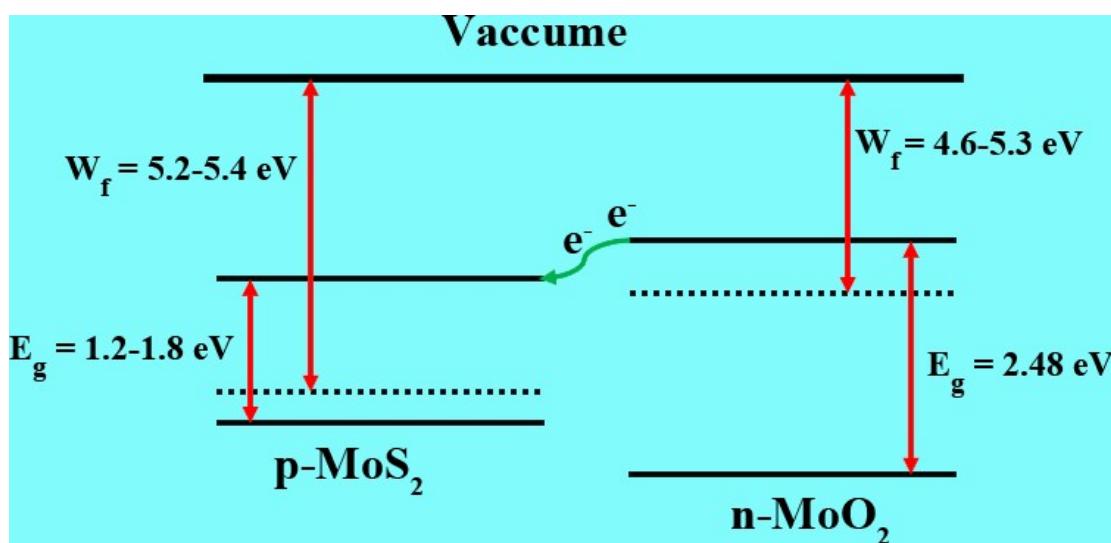
**Fig. S11** The response-recovery curve of (a) MSO-1, (b) MSO-3, (c) pure  $\text{MoO}_2$ , (d) pure  $\text{MoS}_2$  to  $\text{NO}_2$  gas at RT in air.



**Table S2.** Response, response time and recovery time of the sensors to NO<sub>2</sub> at room temperature.

Sensors	MoS <sub>2</sub>			MoO <sub>2</sub>			MSO-1			MSO-2			MSO-3		
NO <sub>2</sub> (ppm)	Response	T <sub>s</sub>	T <sub>r</sub>	Response	T <sub>s</sub>	T <sub>r</sub>	Response	T <sub>s</sub>	T <sub>r</sub>	Response	T <sub>s</sub>	T <sub>r</sub>	Response	T <sub>s</sub>	T <sub>r</sub>
100	1.47	11.7	30.4	1.32	3.2	37.3	7.79	2.1	81.1	<b>19.43</b>	<b>1.06</b>	<b>22.9</b>	14.73	2.1	26.1
50	1.34	7.4	38.4	1.28	1.6	42.1	6.44	2.6	80.0	<b>17.30</b>	<b>2.1</b>	<b>25.6</b>	11.22	1.0	28.8
30	1.25	9.0	32.5	1.19	2.6	41.0	3.18	2.1	64.5	<b>13.71</b>	<b>3.2</b>	<b>19.2</b>	7.29	1.6	43.2
10	1.39	3.2	39.4	1.12	5.3	25.2	2.29	3.2	72.5	<b>12.33</b>	<b>2.1</b>	<b>24.0</b>	6.01	2.1	34.2
5	1.40	4.2	38.4	1.11	4.8	33.6	1.63	6.9	48.5	<b>5.81</b>	<b>2.6</b>	<b>22.4</b>	4.41	3.2	34.2
3	1.24	2.1	35.2	1.02	8.5	18.1	1.52	5.8	47.4	<b>3.84</b>	<b>3.7</b>	<b>16.0</b>	1.12	2.6	15.4
0.5	1.22	7.4	34.6				1.18	8.0	37.3	<b>1.79</b>	<b>3.2</b>	<b>14.0</b>	1.09	3.7	22.4
0.3							1.04	12.2	18.1	<b>1.12</b>	<b>4.2</b>	<b>17.0</b>			
0.1										<b>1.03</b>	<b>3.7</b>	<b>13.8</b>			

T<sub>s</sub> = Response Time (s), T<sub>r</sub> = Recovery Time (s)



**Scheme S1.** Band diagram of MoS<sub>2</sub> and MoO<sub>2</sub> before contact.

**Table S3.** Comparison of the Gas-Sensing Performances of the MoS<sub>2</sub>@MoO<sub>2</sub> Nanonetwork toward NO<sub>2</sub> with Previous Works.

Sensing material	NO <sub>2</sub> conc. (ppm)	Operation temperature (°C)	Response/Recovery time (s)	Ref.
MoS <sub>2</sub> microspheres	100	150	79/225	1
MoS <sub>2</sub> flakes	100	RT(UV)	29/350	2
3D MoS <sub>2</sub> aerogel	1	200	33/107	3
2D MoS <sub>2</sub>	500	RT	180/480	4
MoS <sub>2</sub> /ZnO hetero-nanostructure	5	RT	40/40	5
MoS <sub>2</sub> –MoO <sub>3</sub>	10	RT	19/182	6
MoS <sub>2</sub> /Graphene Hybrid Aerogel	1	200	21.6/29.4	7
MoS <sub>2</sub> /SnO <sub>2</sub>	10	RT	408/162	8
RGO-MoS <sub>2</sub> -CdS nanocomposite	0.2	75	25/34	9
ZnO/rGO nanocomposite	1	RT	75/132	10
Graphene-SnO <sub>2</sub> nanocomposite	5	150	129/107	11
2D SnS <sub>2</sub>	10	120	170/140	12
SnS <sub>2</sub> /SnO <sub>2</sub> nanoheterojunctions	1	100	299/143	13
MoS <sub>2</sub> @MoO <sub>2</sub> nanonetwork	100/0.1	RT	1.06/22.9 (100 ppm) 3.7/13.8 (0.1 ppm)	This work

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