

Supporting Information (SI):

**Tuning coupling interface of ultrathin Ni₃S₂@NiV-LDH
heterogeneous nanosheet electrocatalysts for improved overall
water splitting**

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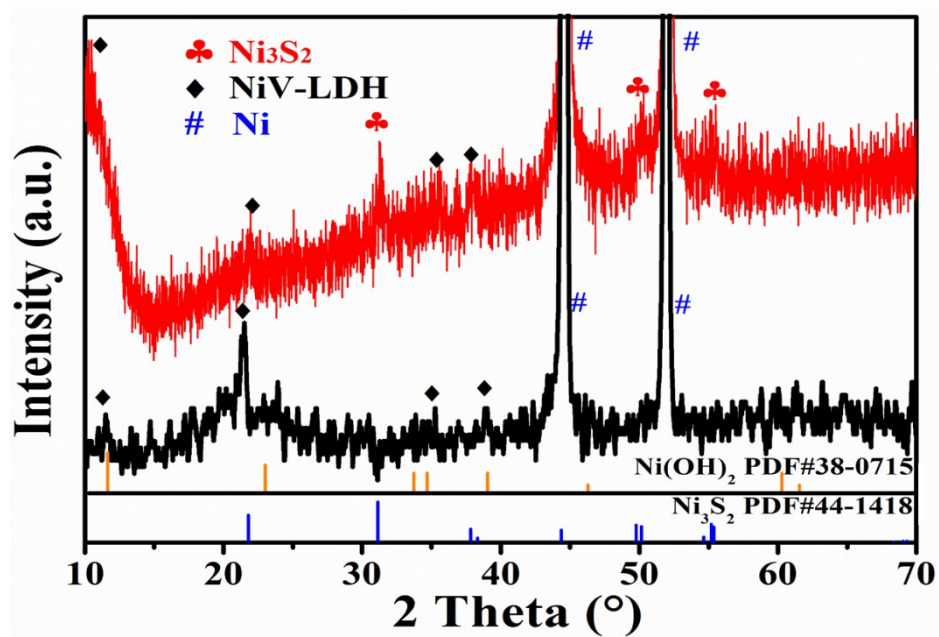


Fig. S1 XRD patterns of NiV-LDH/NF and Ni₃S₂@NiV-LDH/NF.

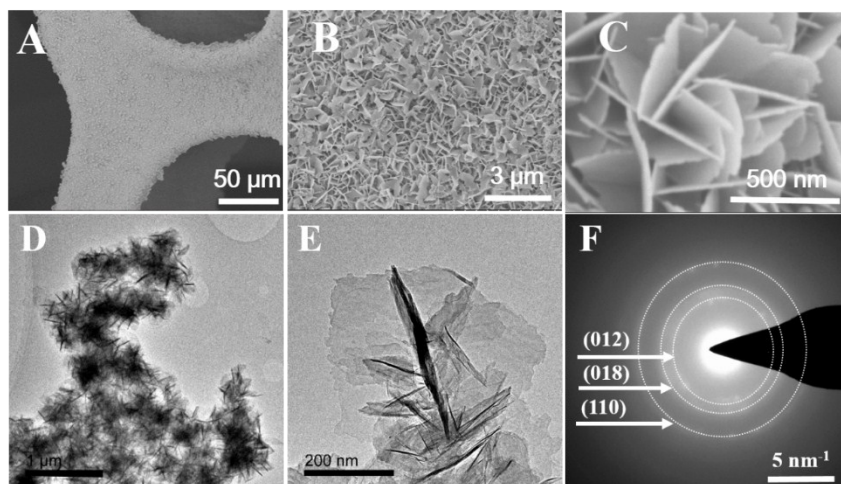


Fig. S2 (A-C) FESEM images of the NiV-LDH/NF; (D)(E) TEM images of the NiV-LDH/NF architecture; (F) Selected area electron diffraction pattern.

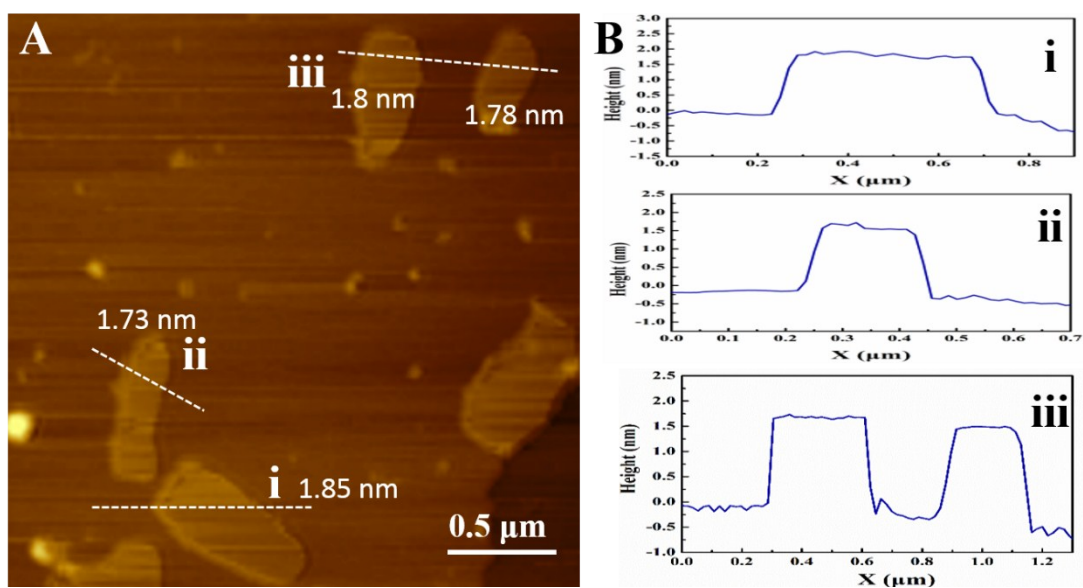


Fig. S3 AFM image and the height profile of $\text{Ni}_3\text{S}_2@\text{NiV-LDH}$ nanosheets.

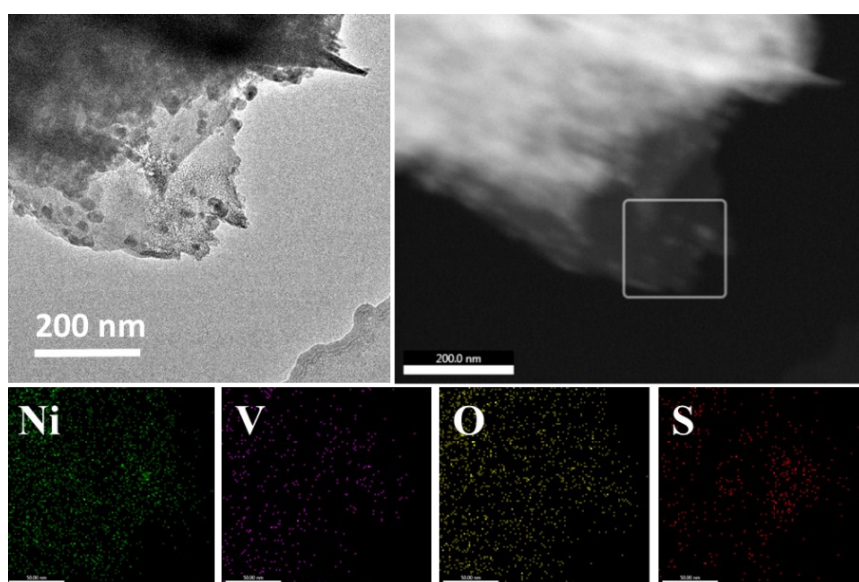


Fig. S4 HAADF-STEM image and elemental mapping of Ni, V, O and S of $\text{Ni}_3\text{S}_2@\text{NiV-LDH}$ nanosheet.

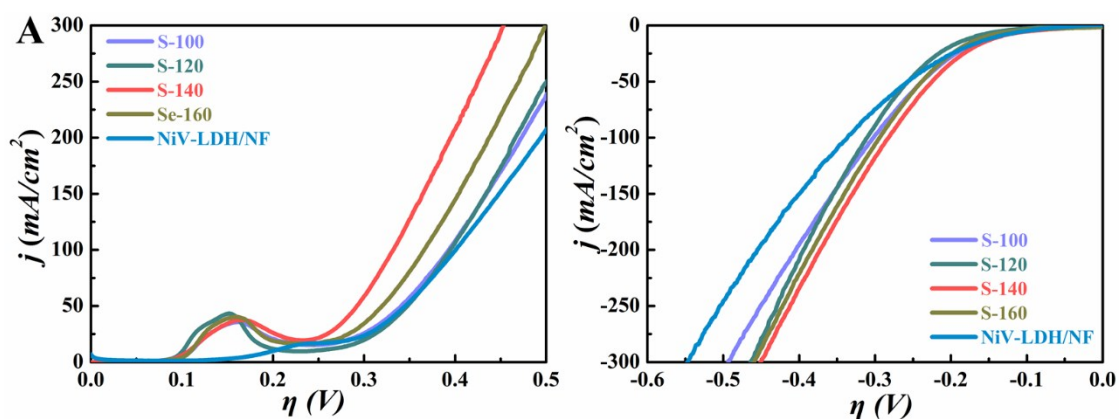


Fig. S5(A)OER and (B)HER polarization curves of NiV-LDH/NF and $\text{Ni}_3\text{S}_2@\text{NiV-LDH/NF}$ at different reaction temperatures.

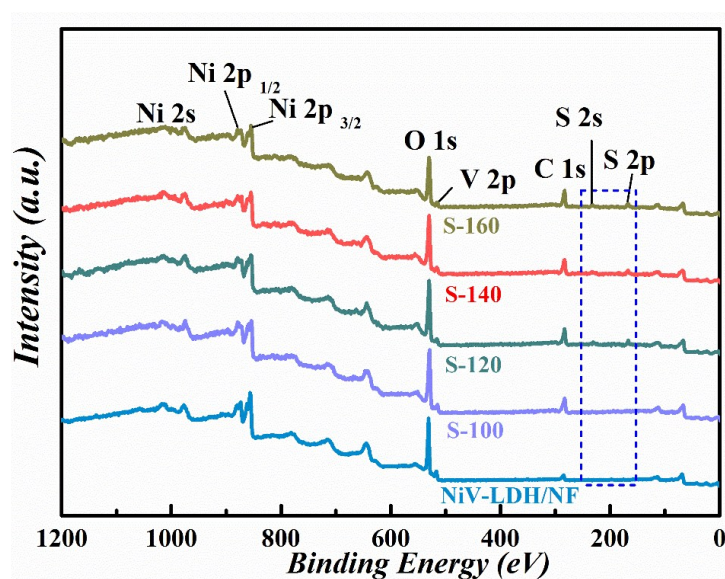


Fig. S6 The survey spectra of the five samples (S-100, S-120, S-140, S-160 and NiV-LDH/NF).

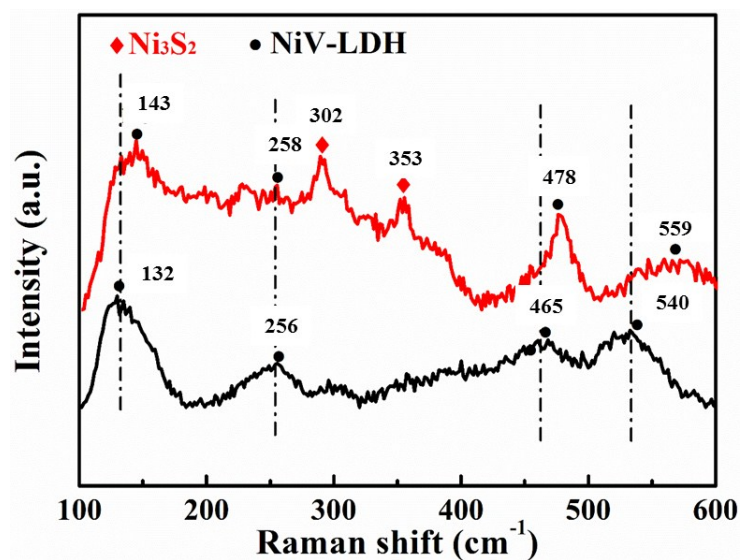


Fig. S7. Raman spectra of NiV-LDH/NF and Ni₃S₂@NiV-LDH/NF.

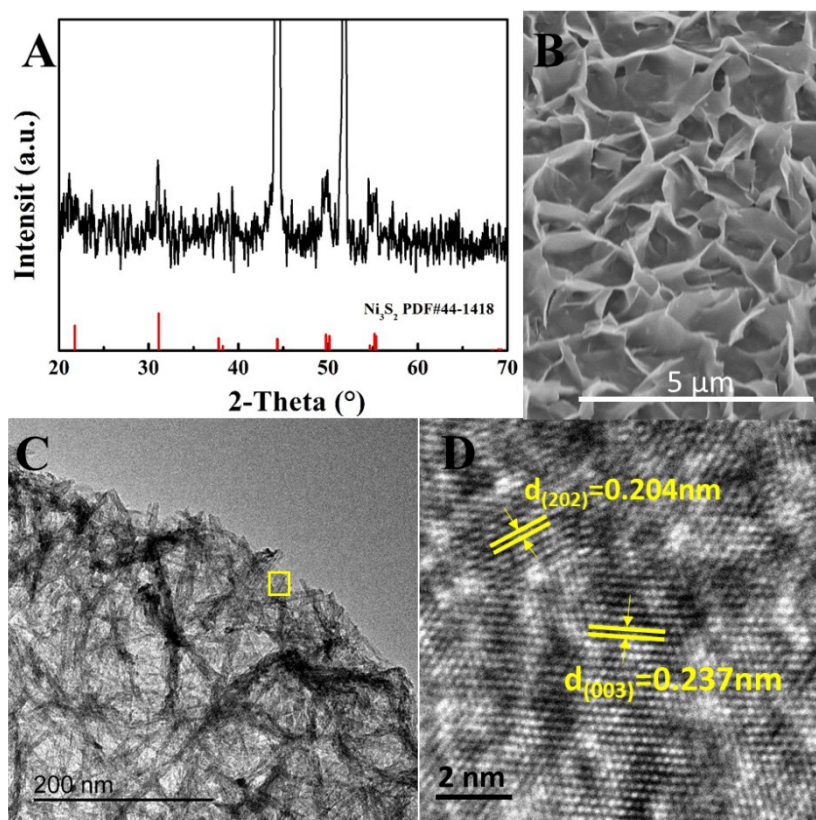


Fig. S8 (A)XRD; (B) SEM; (C) TEM and (D) the high-resolution TEM images of Ni₃S₂/NF.

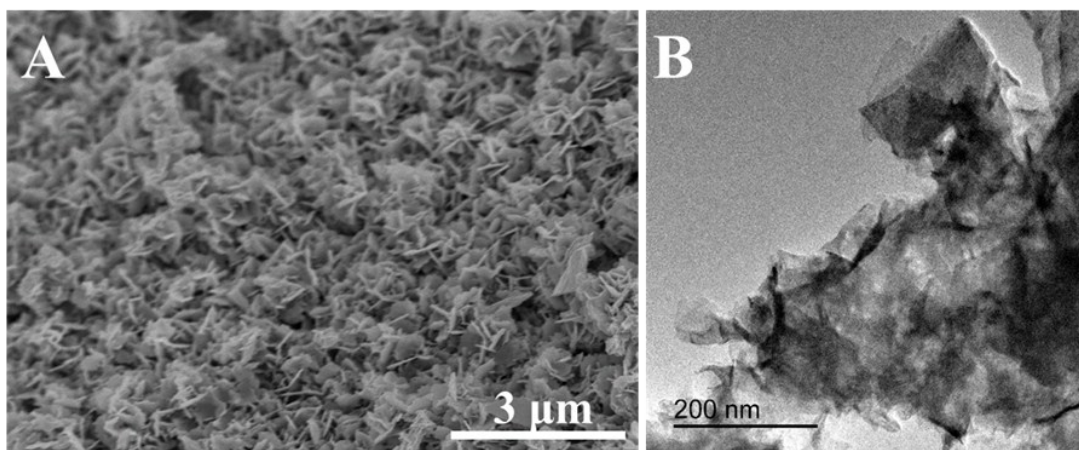


Fig. S9 (A)SEM image and (B) TEM image of $\text{Ni}_3\text{S}_2@\text{NiV-LDH}$ /NF after 100h OER electrocatalysis.

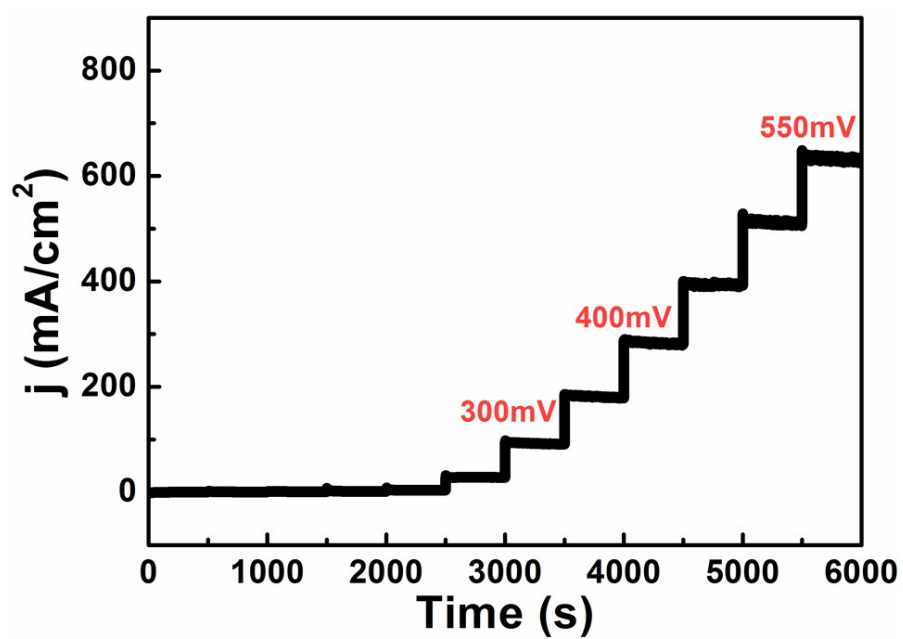


Fig. S10 Multi-step chronoamperometric curve of OER over $\text{Ni}_3\text{S}_2@\text{NiV-LDH}$ /NF in 1.0 M KOH.

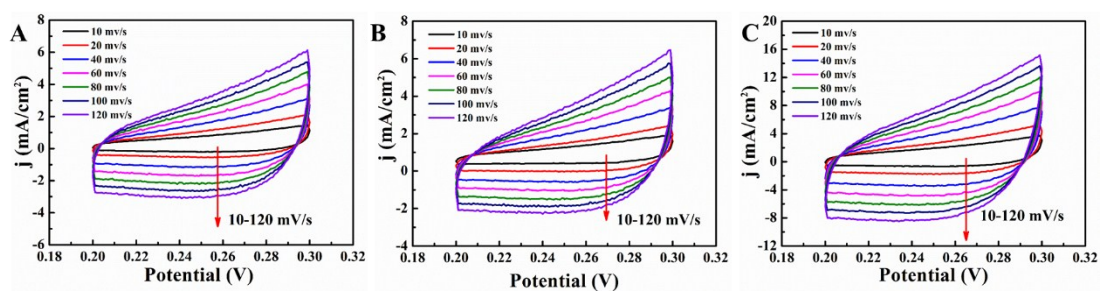


Fig. S11 CV curves of (A) NiV-LDH/NF; (B) Ni₃S₂/NF and (C) Ni₃S₂@NiV-LDH/NF at different scan rates.

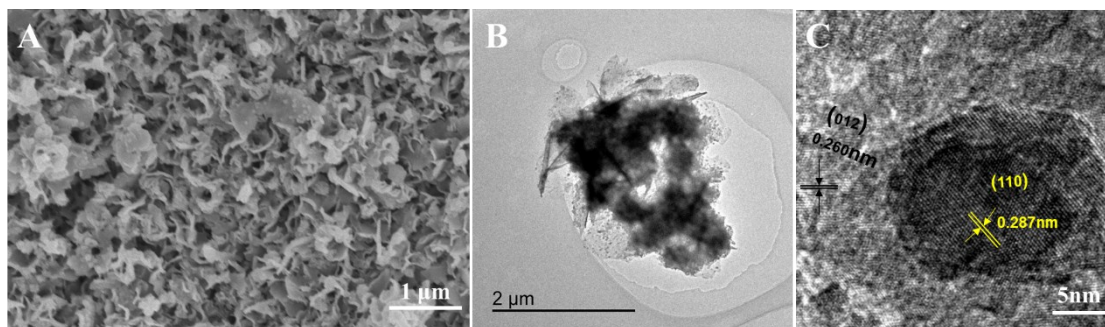


Fig. S12 (A) SEM image; (B) TEM image and (C) the high-resolution TEM image of Ni₃S₂@NiV-LDH/NF after 100h HER electrocatalysis.

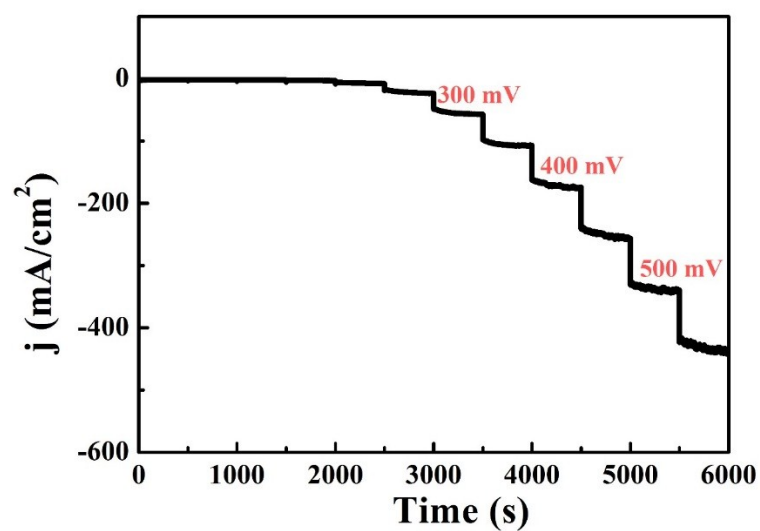


Fig. S13 Multi-step chronoamperometric curve of HER over $\text{Ni}_3\text{S}_2@\text{NiV-LDH/NF}$ in 1.0 M KOH.

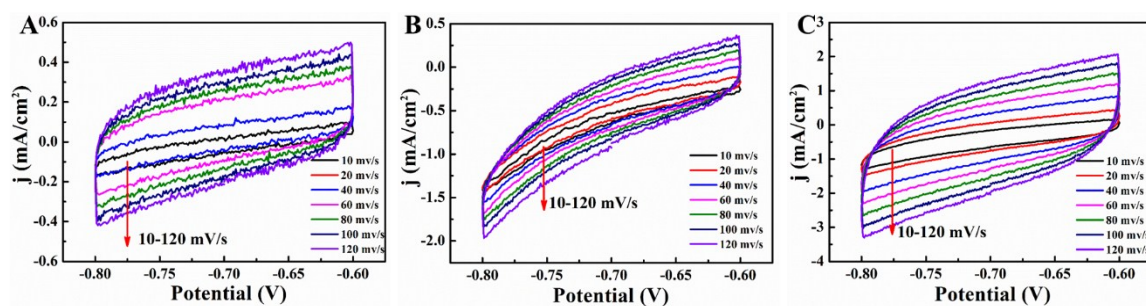


Fig. S14 CV curves of (A) NiV-LDH/NF ; (B) $\text{Ni}_3\text{S}_2/\text{NF}$ and (C) $\text{Ni}_3\text{S}_2@\text{NiV-LDH/NF}$ at different scan rates.

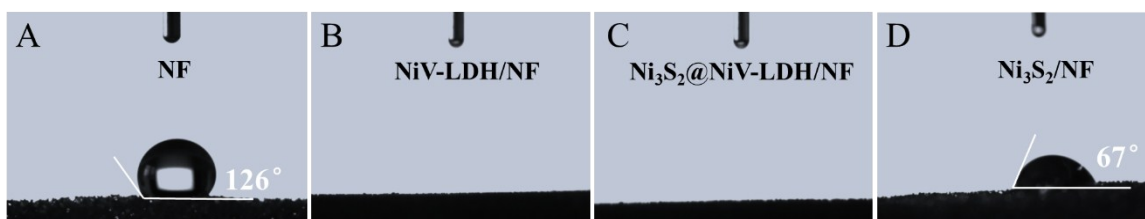


Fig. S15 Contact angle measurements of different samples. (A) Ni foam; (B) NiV-LDH/NF; (C) Ni₃S₂@NiV-LDH/NF and (D) Ni₃S₂/NF.

Table S1. Comparison of the electrocatalytic performance of Ni₃S₂@NiV-LDH/NF with other OER catalysts reported recently

Catalyst	Electrolyte solution	OER Overpotential for 100 mA/cm ² (mV)	Stability test	mass loading (mg cm ⁻²)	Reference
Ni ₃ S ₂ @NiV-LDH/NF	1 M KOH	320	100	0.71	This work
NiFe LDH@NiCoP/NF	1 M KOH	350	100	2	Adv. Funct. Mater. 2018, 28, 1706847
Co(OH) ₂ @NCNTs@NF	1 M KOH	400	>200	0.72	Nano Energy, 2018, 47, 96-104
NiCo ₂ S ₄ @NiFe LDH/NF	1 M KOH	230	10h	/	ACS Appl. Mater. Interfaces, 2017, 9, 15364-15372
Co ₁ Mn ₁ CH/NF	1 M KOH	340	18	5.6	J. Am. Chem. Soc. 2017, 139, 8320–8328
Ni(OH) ₂ /Ni ₃ S ₂ /NF	1 M KOH	490	10	/	J. Mater. Chem. A, 2018, 6, 6938-6946
NiS/NiS ₂ /CC	1 M KOH	416	24	2.4	J. Mater. Chem. A, 2018, 6, 8233-8237
Ni ₂ P@NiFe/NF	1 M KOH	230	25	1.0	Chem. Sci., 2018,9, 1375-1384
NiFe-LDH/NiCo ₂ O ₄ /NF	1 M KOH	390	10	4.9	Acs Appl. Mater. Interfaces, 2017, 9, 1488–1495
CoNiP @ NiFe-LDH/NF	1 M KOH	230	500	/	ACS Appl. Energy Mater., 2018, 1, 623–631
NiFe ₂ O ₄ /NiFe LDH/NF	1 M KOH	213	20	2.8	ACS Appl. Mater. Interfaces, 2018, 10, 26283–26292
Ni _x Co _{3-x} S ₄ /Ni ₃ S ₂ /NF	1 M KOH	330	30	0.56	Nano Energy 2017, 35, 161-170
Ni ₂ P/Ni ₃ S ₂ /NF	1 M KOH	330	24	/	Nano Energy, 2018, 51, 26-36
CoSx/Ni ₃ S ₂ @NF	1 M KOH	373	20	2.83	ACS Appl. Mater. Interfaces, 2018, 10, 27712–27722
MoS ₂ -Ni ₃ S ₂ HNRs/NF	1 M KOH	340	48	13	ACS Catal., 2017, 7, 2357–2366

Mo-W-S- 2@Ni₃S₂/NF	1 M KOH	>370	50	/	ACS Appl. Mater. Interfaces, 2017, 9, 26066–26076
NiS/NiS₂/CC	1 M KOH	416	24	2.4	J. Mater. Chem. A, 2018, 6, 8233-8237

Table S2. Comparison of the electrocatalytic performance of Ni₃S₂/NiV-LDH/NF with other HER catalysts reported recently

Catalyst	HER Overpotential for 10 mA/cm ² (mV)	Electrolyte solution	Stability test (h)	mass loading (mg/cm ²)	Reference
Ni ₃ S ₂ @NiV-LDH/NF	126	1 M KOH	100	0.71	This work
NiFe LDH@NiCoP/NF	120	1 M KOH	100	2	Adv. Funct. Mater. 2018, 28, 1706847
Co ₅ Mo _{1.0} P NSs@NF	173	0.5 M H ₂ SO ₄	20	/	Nano Energy, 2018, 45, 448-455
NiS ₂ /MoS ₂ HNW	204	1 M KOH	6	/	ACS Catal. 2017, 7, 6179–6187
Co(OH) ₂ @NCNTs@NF	170	1 M KOH	600	0.72	Nano Energy, 2018, 47, 96-104
Cu@CoFe-LDH/Cu	171	1 M KOH	48	1.8	Nano Energy, 2017, 41, 327–336
Co ₅ Mo _{1.0} P NSs@NF	173	0.5 M H ₂ SO ₄	20	/	Nano Energy, 2018, 45, 448-455
Ni ₂ P@NiFe/NF	75	1 M KOH	25	1.0	Chem. Sci., 2018,9, 1375-1384
NiCo ₂ S ₄ @NiFe LDH/NF	200	1 M KOH	10	/	ACS Appl. Mater. Interfaces 2017, 9, 15364–15372
Ni(OH) ₂ /Ni ₃ S ₂ /NF	150	1 M KOH	10	/	J. Mater. Chem. A, 2018,6, 6938-6946
NiFeLDH/NiCo ₂ O ₄ /NF	192	1 M KOH	10	4.9	ACS Appl. Mater. Interfaces, 2017, 9, 1488–1495
NiFe ₂ O ₄ /NiFe LDH/NF	101	1 M KOH	20	2.8	ACS Appl. Mater. Interfaces, 2018, 10, 26283–26292
Ni ₂ P/Ni ₃ S ₂ /NF	80	1 M KOH	24	/	Nano Energy, 2018, 51, 26-36
CoSx/Ni ₃ S ₂ @NF	204	1 M KOH	20	2.83	ACS Appl. Mater. Interfaces, 2018, 10, 27712–27722
MoS ₂ -Ni ₃ S ₂ HNRs/NF	98	1 M KOH	48	13	ACS Catal., 2017, 7, 2357–2366
Mo-W-S-2@Ni ₃ S ₂ /NF	98	1 M KOH	50	/	ACS Appl. Mater. Interfaces, 2017, 9, 26066–26076

MoS₂/Ni₃S₂/NF	110	1 M KOH	10	9.7	Angewandte Chemie, 2016, 55, 6702-6707.
Ni_xCo_{3-x}S₄/Ni₃S₂/NF	136	1 M KOH	50	0.56	Nano Energy 2017, 35, 161-170

Table S3. Comparison of the electrocatalytic water splitting performance of $\text{Ni}_3\text{S}_2@\text{NiV-LDH/NF}$ *vis-à-vis* heterostructured catalysts reported in alkaline media.

Catalyst	Electrolyte	η @10 mA cm ⁻² (V)	Stability Test(h)	Reference
$\text{Ni}_3\text{S}_2@\text{NiV-LDH/NF}$	1 M KOH	1.54	160	This work
$\text{NiFe LDH}@ \text{NiCoP/NF}$	1 M KOH	1.57	100	Adv. Funct. Mater. 2018, 28, 1706847
$\text{Cu}@ \text{CoFe-LDH/Cu}$	1 M KOH	1.68	48	Nano Energy, 2017, 41, 327–336
$\text{Ni}_2\text{P}@ \text{NiFe LDH/NF}$	1 M KOH	1.51	100	Chem. Sci., 2018, 9, 1375-1384
$\text{NiCo}_2\text{S}_4@ \text{NiFe LDH/NF}$	1 M KOH	1.60	12	Acs Appl. Mater. Interfaces, 2017, 9, 15364-15372
$\text{NiFe LDH/ NiCo}_2\text{O}_4/\text{NF}$	1 M KOH	1.60	12	ACS Appl. Mater. Interfaces 2017, 9, 1488–1495
$\text{NiFe}_2\text{O}_4/\text{NiFe LDH/NF}$	1 M KOH	~1.54	20	ACS Appl. Mater. Interfaces, 2018, 10, 26283–26292
$\text{MoS}_2/\text{Ni}_3\text{S}_2/\text{NF}$	1 M KOH	1.56	10	Angewandte Chemie, 2016, 55, 6702-6707.
$\text{Ni (OH)}_2/\text{Ni}_3\text{S}_2/\text{NF}$	1 M KOH	1.57	/	J. Mater. Chem. A, 2018, 6, 6938-6946
$\text{Ni}_x\text{Co}_{3-x}\text{S}_4/ \text{Ni}_3\text{S}_2/\text{NF}$	1 M KOH	1.53	>200	Nano Energy 2017, 35, 161-170
$\text{CoSx}/\text{Ni}_3\text{S}_2@\text{NF}$	1 M KOH	~1.57	30	ACS Appl. Mater. Interfaces, 2018, 10, 27712–27722
$\text{MoS}_2\text{-Ni}_3\text{S}_2$ HNRs/NF	1 M KOH	1.5	50	ACS Catal., 2017, 7, 2357–2366
$\text{Mo-W-S-2}@ \text{Ni}_3\text{S}_2/\text{NF}$	1 M KOH	1.62	50	ACS Appl. Mater. Interfaces, 2017, 9, 26066–26076
$\text{MoS}_2/\text{Ni}_3\text{S}_2/\text{NF}$	1 M KOH	1.56	10	Angewandte Chemie, 2016, 55, 6702-6707.