

Supporting Information for

**Bi-metal-organic frameworks type II heterostructure for
enhanced photocatalytic styrene oxidation**

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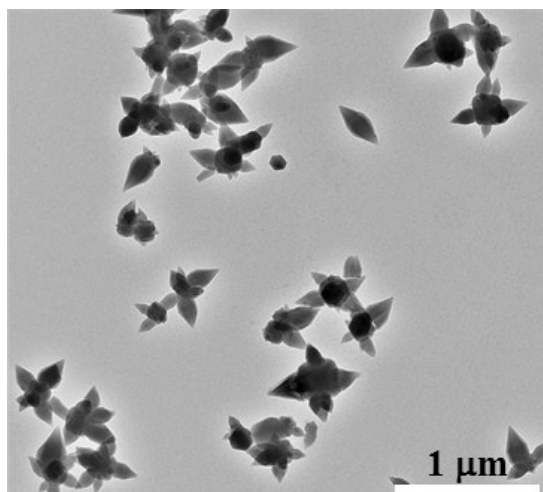


Fig.S1 TEM image of UOML-7 heterostructure.

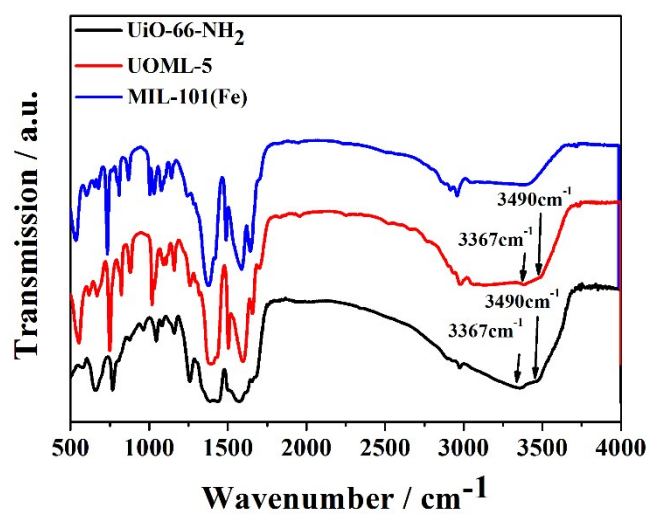


Fig.S2 FTIR spectra of pristine UiO-66-NH₂, MIL-101(Fe) and UOML-5 heterostructure.

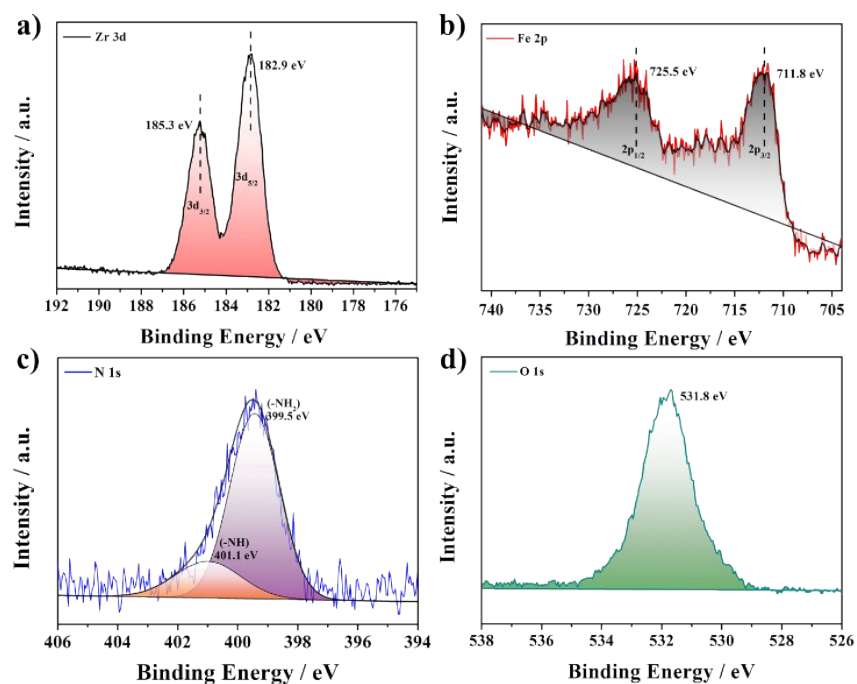


Fig.S3 XPS spectra of UOML-5: a) Zr 3d, (b) Fe 2p, (c) N 1s, and (d) O 1s.

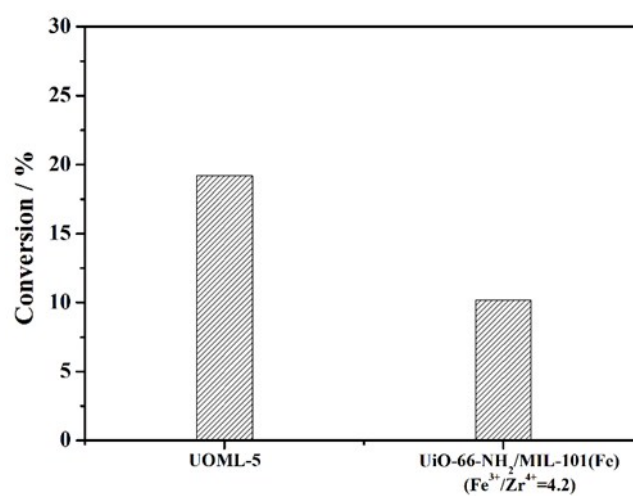


Fig.S4 Styrene conversion over UOML-5 heterostructure and physical mixing UiO-66-NH₂/MIL-101(Fe) composite (Fe³⁺/Zr⁴⁺=4.2).

Table S1. The atomic percentage of Fe and Zr determined using ICP-MS.

Material	Fe, at%	Zr, at%	Experimental Fe ³⁺ /Zr ⁴⁺
UOML-3	71.4	28.6	2.5
UOML-5	80.8	19.2	4.2
UOML-7	85.9	14.1	6.1
UOML-9	88.1	11.9	7.4

Table S2. The photocatalytic activity of UOML-5 heterostructure in different organic solvents after 12 h.

Reagent	Solvent	Conversion	Selectivity	
			Benzaldehyde	Others
Styrene	CH ₃ CN	19.2%	91.7%	8.3%
	Acetone	15.6%	93.4%	6.6%
	Toluene	12.8%	90.2%	9.8%

Table S3. Summary of photocatalytic oxidation of styrene based on various photocatalysts reported in references and obtained in the present work.

Photocatalyst	Reaction Conditions	Conversion %	Selectivity (Benzaldehyde) %	Ref.
UOML-5 heterostructure	O ₂ , visible-light	19.2	91.7	This work
UiO-66-NH ₂	O ₂ , visible-light	7.7	83.4	[1]
TiO ₂	Acid, H ₂ O ₂ , UV	12	90	[2]
V(0.5)/SBA-15	H ₂ O ₂ , UV	28.2	29.4	[3]
Cr-SiO ₂	O ₂ , visible-light	2.4	80	[4]

Table S4. Photocatalytic oxidation of toluene over UOML-5 heterostructure.

Reagent	Conversion	Selectivity	
		Benzaldehyde	Others
Toluene	3.8%	89.3%	10.7%

Reaction conditions: photocatalyst (100 mg), toluene (0.5 mmol), benzene (10 ml), room temperature, O₂, visible-light irradiation ($\lambda > 420\text{nm}$, 8h).

Table S5. Structural parameters obtained from N₂ adsorption isotherms analysis.

Sample	S _{BET}	Pore volume	Average Pore Size (nm)
	(m ² g ⁻¹)	(cm ³ g ⁻¹)	
UiO-66-NH ₂	770.1	0.4516	1.52
UOML-5	916.2	0.6223	1.77

References

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