MSOT/CT/MR imaging Guided and hypoxia Maneuvered Oxygen self-sufficiency radiotherapy based on One-pot MnO₂-mSiO₂ @ Au nanoparticle

Supporting Information

Siyu Wang¹, Qing You¹, Jinping Wang¹, Yilin Song¹, Yu Cheng¹, Yidan Wang¹, Shan Yang¹,

Lifang Yang¹, Peishan Li¹, Qianglan Lu¹, Meng Yu**², Nan Li*¹

(1 Tianjin Key Laboratory of Drug Delivery & High-Efficiency, School of Pharmaceutical

Science and Technology, Tianjin University, 300072, Tianjin, PR China.)

(2 Guangdong Provincial Key Laboratory of new Drug Screening, School of Pharmaceutical

Science, Southern Medical University, 510515, Guangzhou, PR China.)

*(Li N.) Corresponding author at: School of Pharmaceutical Science and Technology, Tianjin University, 300072, Tianjin, PR China.

E-mail address: linan19850115@163.com

*(Yu M.) Corresponding author at: School of Pharmaceutical Science, Southern Medical University, 510515, Guangzhou, PR China.

E-mail address: meng4716@126.com

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Figure. S1 (a). Elements analysis of MNH_{S} (b). TEM images of MNH_{S} scale bar from up to down are 500 nm, 200nm and 20 nm. Table1. The element types O, Si, Mn and the atom weight respectively.



Figure. S2 (a). Surface area distributed by different pore width of MnO_2 -mSiO₂ nanohybrids. (b). Multi-point BET plot of MnO_2 -mSiO₂ nanohybrids. Table.2. BET data summary.



Figure. S3 (a). Cumulative surface area of MnO_2 -SiO₂ nanohydrids. (b) Cumulative pore volume distributed by different pore width of MnO_2 -SiO₂ nanohybrids. Table3. DFT method Summary.



Figure. S4 (a). ¹H-NMR spectrum of the unmodified hyaluronic acid (b) ¹H-NMR spectrum of dopamine (c). ¹H-NMR spectrum of HA-DN complex.



Figure. S5 Oxygen generation capability of different formulations.



Figure. S6 Photoacoustic imaging for tumor oxygen generation



Figure. S7. CT imaging for formulation distribution in major organs after 24 h injection. The circled sections represent the tumor sites.



Figure. S8 (a). Pharmacokinetic profiles of MnO₂-mSiO₂@Au-HA nanoparticles following intravenous administration by measureing Au concentrations. (b). Biodistribution in major organs after intravenously injection of MnO₂-mSiO₂@Au-HA nanoparticles.



Figure S9. T₂-weighted MR images of MnO_2 -mSiO₂@Au-HA at varying Mn^{2+} concentrations. Lower is the linear relationship between the T₂ relaxation rate (1/T₂) and Mn^{2+} concentrations in MnO_2 -mSiO₂@Au-HA aqueous solutions.



Figure. S10 (a). CT phantom images of Iohexol (upper panel) and MAHNPs (lower panel) at different I or Au concentrations. (b). CT values of MAHNPs and Iohexol with different Au or I concentrations.

The Calculation of the photothermal conversion efficiency was determined by the following steps according to the references[1-3].

Firstly, the aqueous solution of the MnO₂-mSiO₂@Au-HA nanoparticles (100 ppm) underwent continuous irradiation of 808 nm laser (1.5 W/cm²) until steady state temperature was reached. Then the laser was shut off, and the aqueous solution was naturally cooled to environment temperature. The temperature change of the aqueous solution was recorded (Figure. 2g). The η value was calculated as follows:

$$\eta = \frac{hs(T_{Max} - T_{Surr}) - Q_S}{I(1 - 10^{-A808})}$$
(1)

Where h is the heat transfer coefficient, S is the surface area of the container, and the value of *hs* is obtained from the Eq. (4) and Figure 2h. The maximum steady temperature (T_{max}) was 63.1 °C and environmental temperature (T_{Surr}) was 25.9 °C. The laser power *I* used in irradiation was 1.5 W/cm². The absorbance of the MnO₂-mSiO₂@Au-HA nanoparticles at 808 nm was 0.534. Q_{Dis} is heat dissipated from the light absorbed by the solvent and container. A dimensionless parameter θ is calculated as followed:

$$\theta = \frac{T - T_{Surr}}{T_{Max} - T_{Surr}}$$
(2)

A sample system time constant τ_s can be calculated as Eq. (3)

$$t = -\tau_s \ln(\theta) \tag{3}$$

According to figure 2g, τ_s was obtained to be 166.97 s.

$$hs = \frac{m_D C_D}{\tau_s} \tag{4}$$

In addition, m_D is 0.5 g and C_D is 4.2 J/g·°C in the experiment.

 Q_S is heat dissipated from the light absorbed by the container itself, and it was determined independently to be 3.5 mW using a container containing pure water. Thus, substituting according values of each parameters to Eq. 1, the 808 nm laser photothermal conversion efficiency (η) of the MnO₂-mSiO₂@Au-HA nanoparticles can be calculated to be 25.38%.

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