Supporting Information

Self-limiting electrode with double-carbon layers as walls for efficient sodium storage performance

Yinghui Wang,[a] Deyang Zhang,[a] Yangbo Wang,[a] Yingge Zhang,[a] Xianming Liu,[b] Weiwei Zhou,[*c] Jang-Kyo Kim,[d] and Yongsong Luo[*a]

[a]*Key Laboratory of Microelectronics and Energy of Henan Province, School of Physics and Electronic Engineering, Xinyang Normal University, Xinyang 464000, P. R. China.*

[b]*College of Chemistry and Chemical Engineering, Luoyang Normal University, Luoyang 471934, P. R. China.*

[c]*School of Materials Science and Engineering, Harbin Institute of Technology at Weihai, Weihai 264209, China.*

[d]*Department of Mechanical and Aerospace Engineering, Hong Kong University of Science and Technology, Clear Water Bay, Kowloon, Hong Kong, P. R. China.*

[*] Corresponding author. E-mail: ysluo@xynu.edu.cn (Y. S. Luo), Tel./fax: +86 376 6390801. zhouweiwei@hit.edu.cn (W. W. Zhou), Tel./fax: + 0631-5687157.

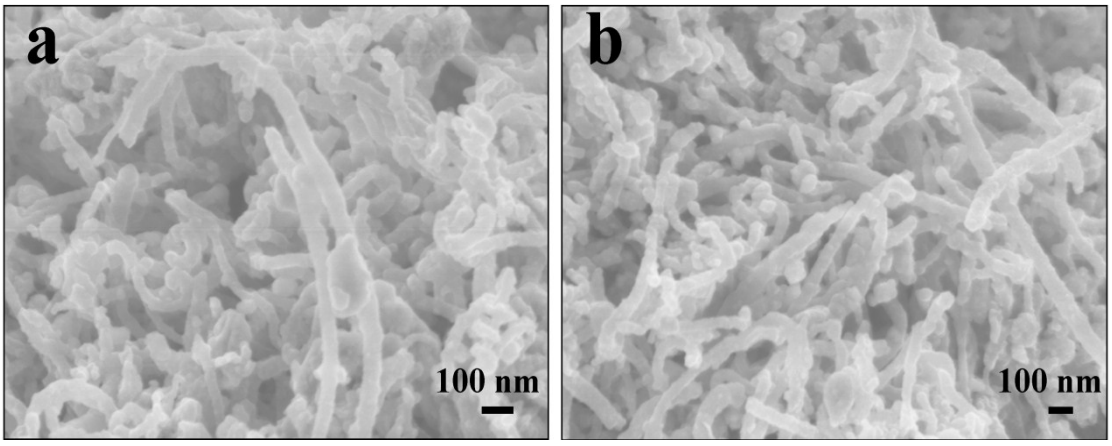

Fig. S1 (a,b) Typical FESEM image of CNT@SnO_2@NCT.

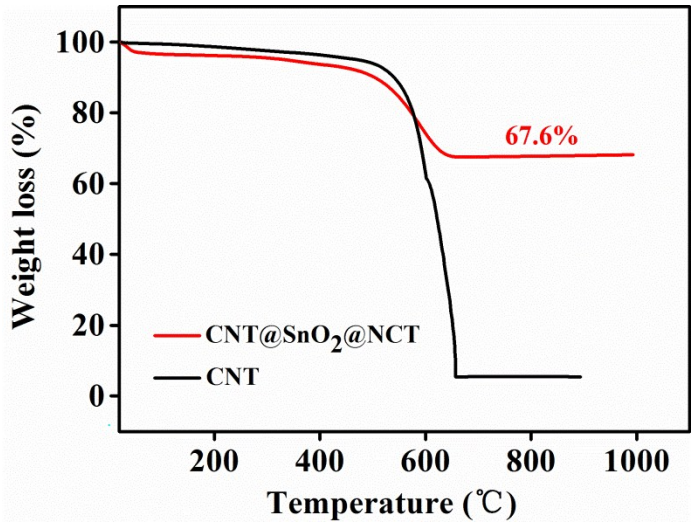

Fig. S2 TGA curves of CNT and CNT@SnO_2@NCT. This pyrolysis process was carried out in oxygen at a heating rate of 10 ^{o}C min^{-1}.

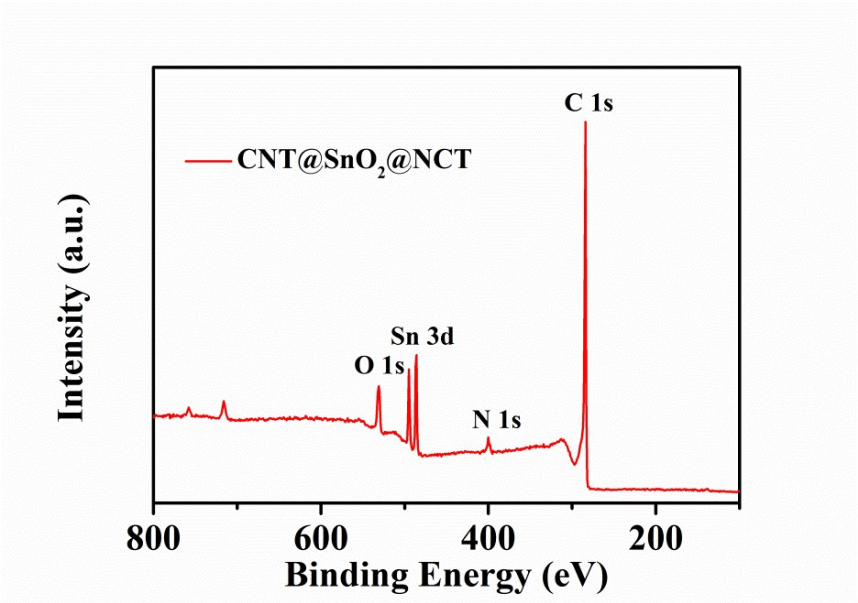

Fig. S3 The survey scan XPS spectrum of CNT@SnO_2@NCT.

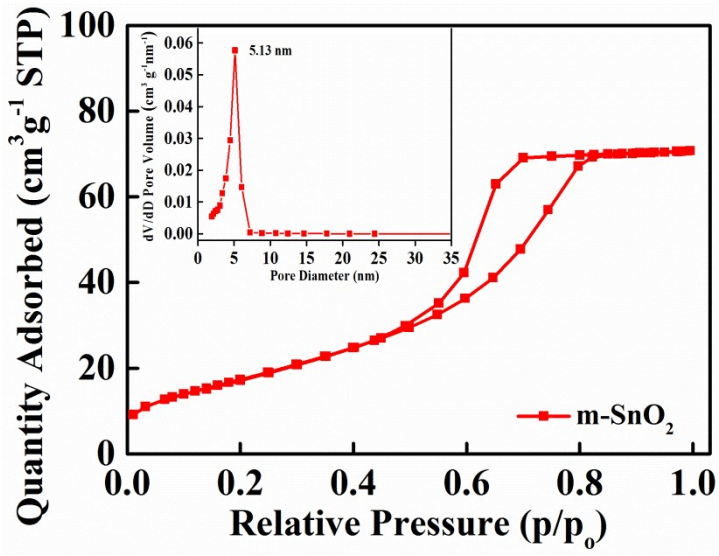

Fig. S4 Nitrogen adsorption/desorption isotherms of the m-SnO_2. Inset is the pore size distributions.

Table S1. A comparison of SnO_2-based electrode materials for SIBs.

Materials	Current Density ($A\ g^{-1}$)	Capacity ($mAh\ g^{-1}$)	Cycle Numbers	Ref.
Sn@SnO_2@C	0.02	245	100	1
CNT@SnO_2@PPy	0.1	226	100	2
SnO_2/C	0.05	240	50	3
SnO_2 nanowires	0.1	189	50	4
Sn/SnO_2/PC	0.05	265	50	5
SnO_2/GNS-$SCCO_2$	0.1	239	100	6
SnO_2@C/CF	0.5	171	100	7
SnO_2-N-GNS	0.05	280	50	8
CNT@SnO_2@G	0.2	180	300	9
SnSb/SnO_2/Sn/C	0.05	305	40	10
SnO_2 QDs/GA	0.05	319	50	11
SnO_2@RGO	0.05	236	100	12
SnO_2/NG	0.04	238	100	13
SnO_2 microfibers	0.02	240	30	14
SnO_2/C nanofibers	0.1	306	200	this work

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