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Unraveling Template-free Fabrication of Carbon Nitride Nanorods Codoped with for Efficient Electrochemical and Photoelectrochemical Carbon Monoxide Oxidation at Room Temperature

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Figure S1. Schematic shows the synthesis process of PtPd/CNs nanorods

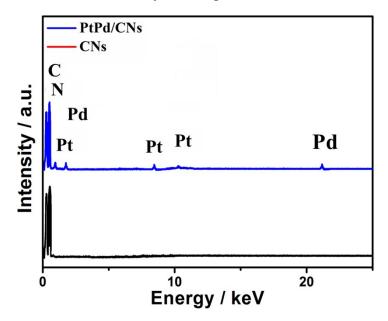


Figure S2. EDX analysis of PtPd/CNs nanorods relative to metal-free CNs nanorods.

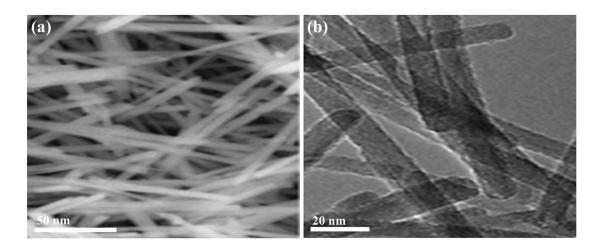


Figure S3 (a) SEM image and (b) TEM image of metal-free CNs nanorods.

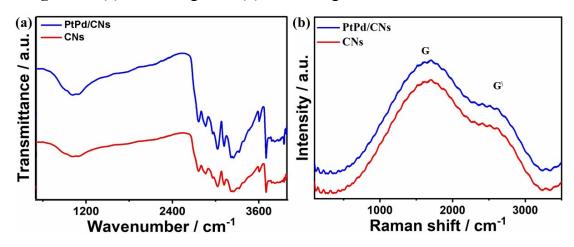


Figure S4. (a) FTIR spectra and (b) Raman spectra of PtPd/CNs nanorods relative to CNs nanorods.

Table S1. Comparison the Surface area of our developed CNs nanorods with previous reported CNs-based nanostructures

Catalyst	Morphology	BET surface area m ² g ⁻	Reference
CNs	Nanorods	155.2	Our work
CN600	Long Needles	99	1
CNs	Nanotubes Nanofibers	32.27 12.96	2
CNs	Nanosheets	46.2	3
CNs	Nanosheets	84.2	4
CNs- Triton- 0.6	Nanoporous sheets	116	51
C ₃ N ₄	Nanoporous sheets	123	6
CN-24-1.0 1=	Nanoporous sheets 24 represents size of colloidal SiO ₂	130	7
	1 represents SiO ₂ /cyanamide ratio		

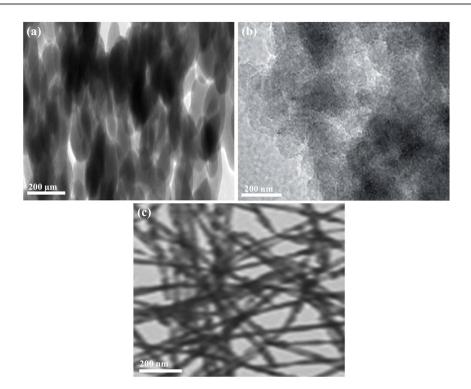


Figure S5 (a) TEM images of aggregated CNs prepared in the absence of NaNO₃ and HCL solutions, (b) CNs nanosheets obtained by the quick addition of NaNO3 and HCL solution, and (c) CNs nanowires formed using ethanol-mediated solution instead of glycol-mediated solution.

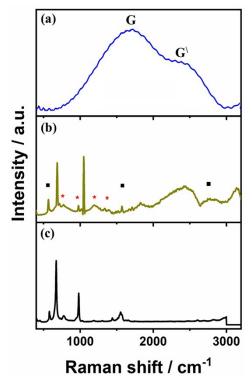


Figure S6. Raman spectra of (a) PtPd/CNs nanorods, (b) Pt/Pd/melon formed after polymerization of melamine, and (c) pure melamine. The asterisks and boxes indicate the Pt/Pd bonded to N and polycondensation of melamine.

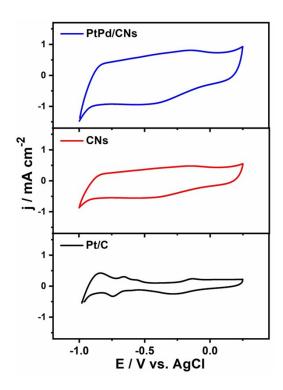


Figure S7. CVs of the PtPd/CNs and CNs nanorods compared to commercial Pt/C catalyst in N_2 -staurated aqueous solution of 0.1 M KOH at 50 mV s⁻¹ at room temperature.

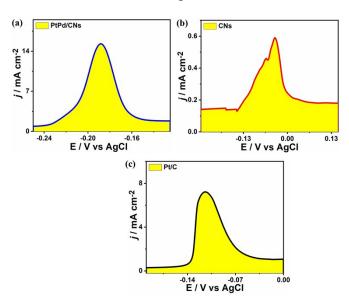


Figure S8. The Co-adsorbed amount over the as-synthesized materials.

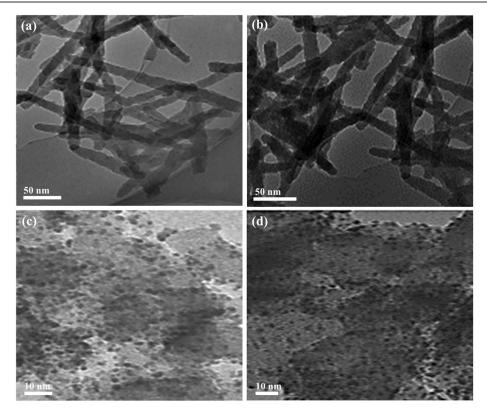


Figure S9. TEM image of (a-b) PtPd/CNs nanorods and (c-d) before and after CO-durability tests, respectively.

References

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