

Electronic Supplementary Information

Facile sequential ion exchange strategy to synthesize CoSe₂/FeSe₂ double-shelled hollow nanocuboids for highly active and stable oxygen evolution reaction

Chunyang Xu,^{a,1} Qinghao Li,^{a,1} Junling Shen,^{a,1} Ze Yuan,^a Jiqiang Ning,^b Yijun Zhong,^a Ziyang Zhang^b and Yong Hu*^a

^aKey Laboratory of the Ministry of Education for Advanced Catalysis Materials, Department of Chemistry, Zhejiang Normal University, Jinhua 321004, China.

E-mail: yonghu@zjnu.edu.cn; yonghuzjnu@163.com

^bVacuum Interconnected Nanotech Workstation, Suzhou Institute of Nano-Tech and Nano-Bionics, Chinese Academy of Sciences, Suzhou 215123, China.

¹C. Xu, Q. Li and J. Shen contributed equally.

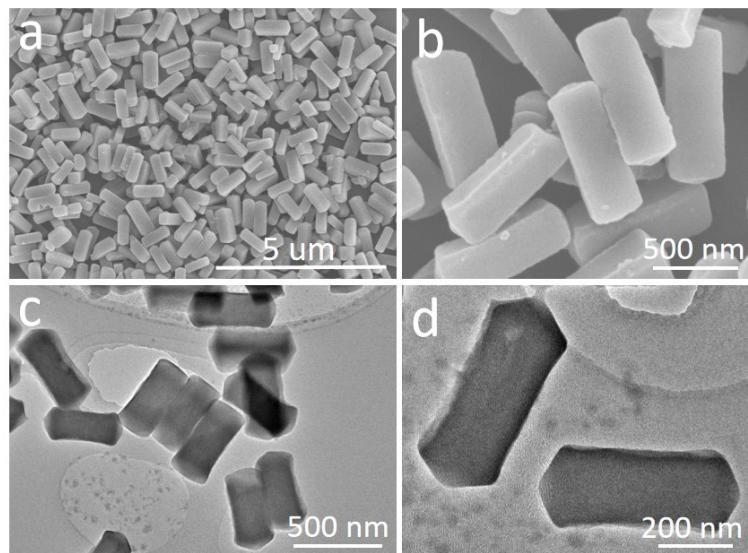


Fig. S1 (a, b) SEM images and (c, d) TEM images of the as-prepared Co-precursor NCs.

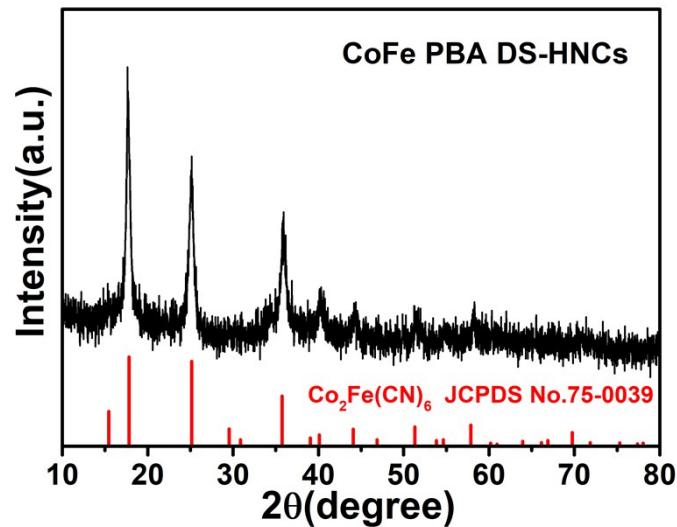


Fig. S2 XRD pattern of the as-prepared CoFe-PBA DS-HNCs.

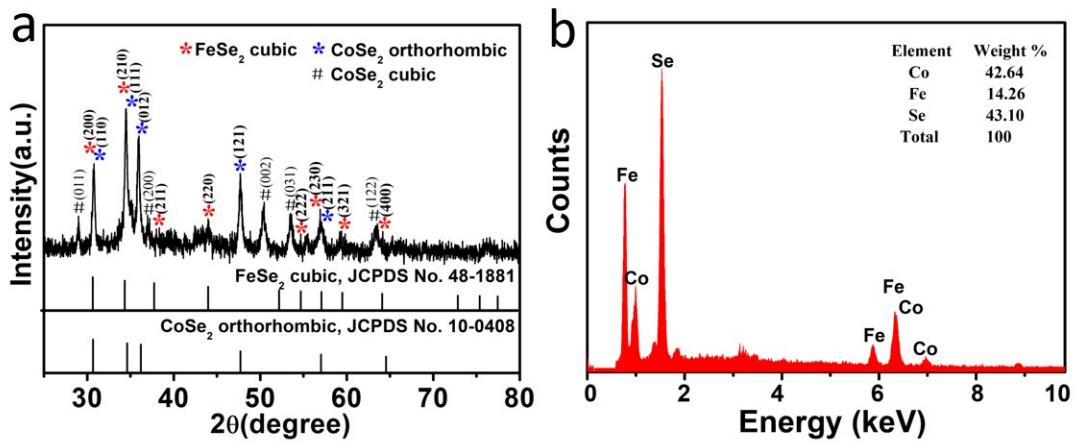


Fig. S3 (a) XRD pattern and (b) EDX spectrum of the as-prepared CoSe₂/FeSe₂ DS-HNCs.

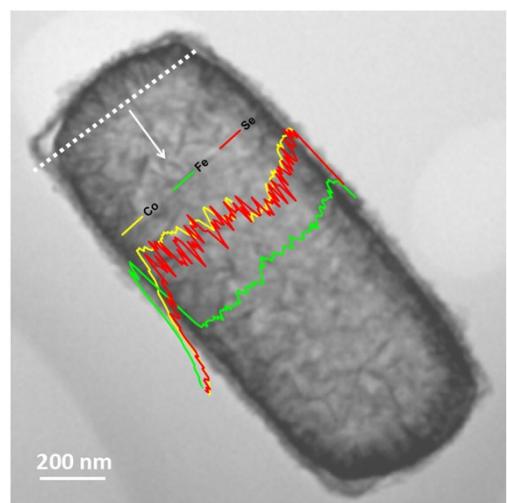


Fig. S4 STEM image and corresponding EDS line scanning of a single $\text{CoSe}_2/\text{FeSe}_2$ DS-HNC.

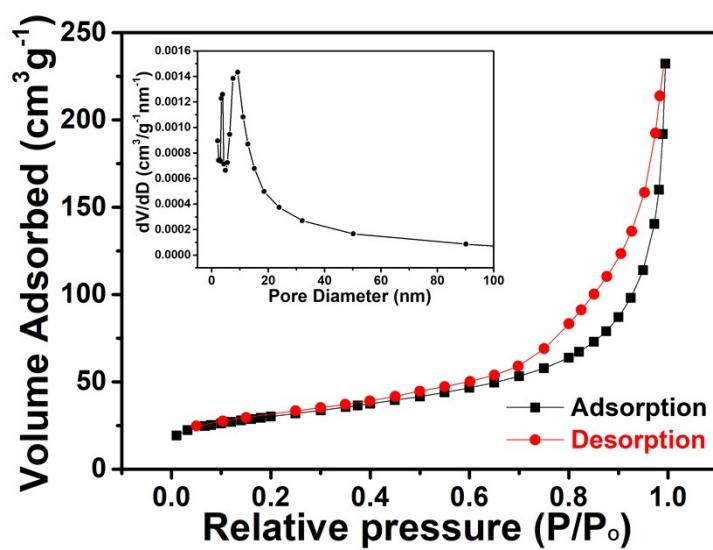


Fig. S5 N_2 adsorption-desorption isotherm and pore size distribution (inset) of the as-prepared $\text{CoSe}_2/\text{FeSe}_2$ DS-HNCs.

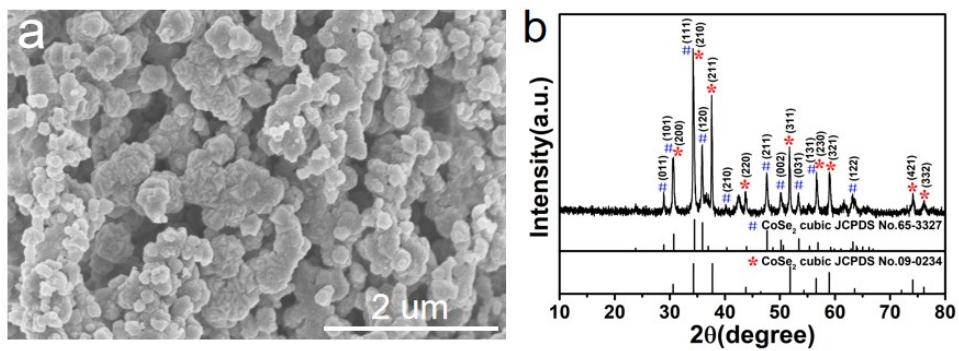


Fig. S6 (a) SEM image and (b) XRD pattern of the as-prepared CoSe₂ NPs.

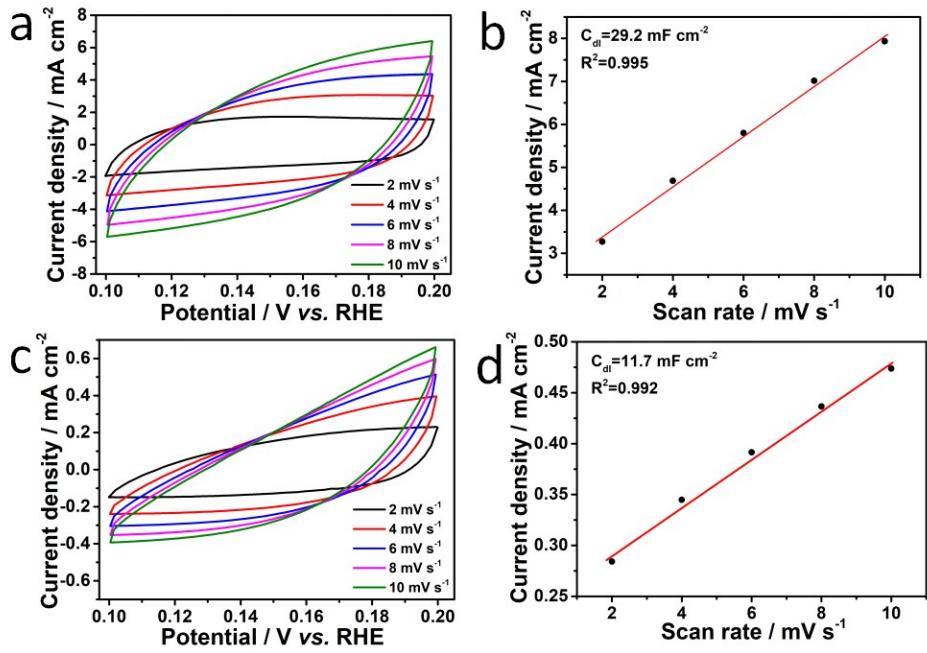


Fig. S7 Cyclic voltammogram (CV) curves of CoSe₂/FeSe₂ DS-HNCs (a) and CoSe₂ NPs (c) in the double layer region at scan rates of 2, 4, 6, 8 and 10 mV s⁻¹ in 1.0 M KOH; (b) and (d) current density as a function of scan rate derived from (a) and (c), respectively.

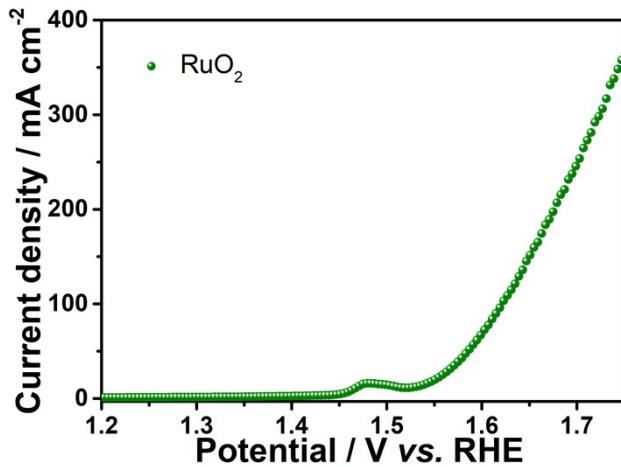


Fig. S8 LSV curve of RuO₂ for OER.

To achieve the current density of 10 mA cm⁻², RuO₂ catalyst needs an overpotential of 290 mV for OER.

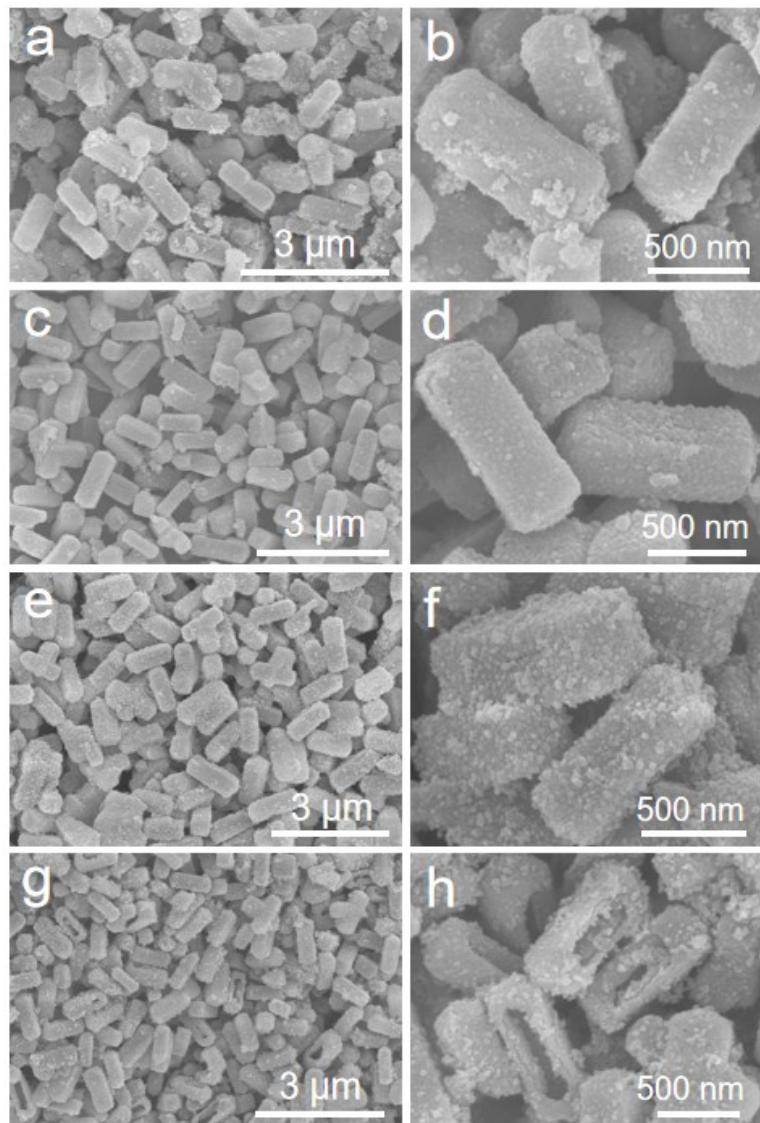


Fig. S9 SEM images of CoFe-PBA samples by using different amount of $K_3[Fe(CN)_6]$: (a, b) 20 mg, CoFe-PBA DS-HNCs-2, (c, d) 40 mg, CoFe-PBA DS-HNCs, (e, f) 60 mg, CoFe-PBA DS-HNCs-6 and (g, h) 80 mg, CoFe-PBA DS-HNCs-8.

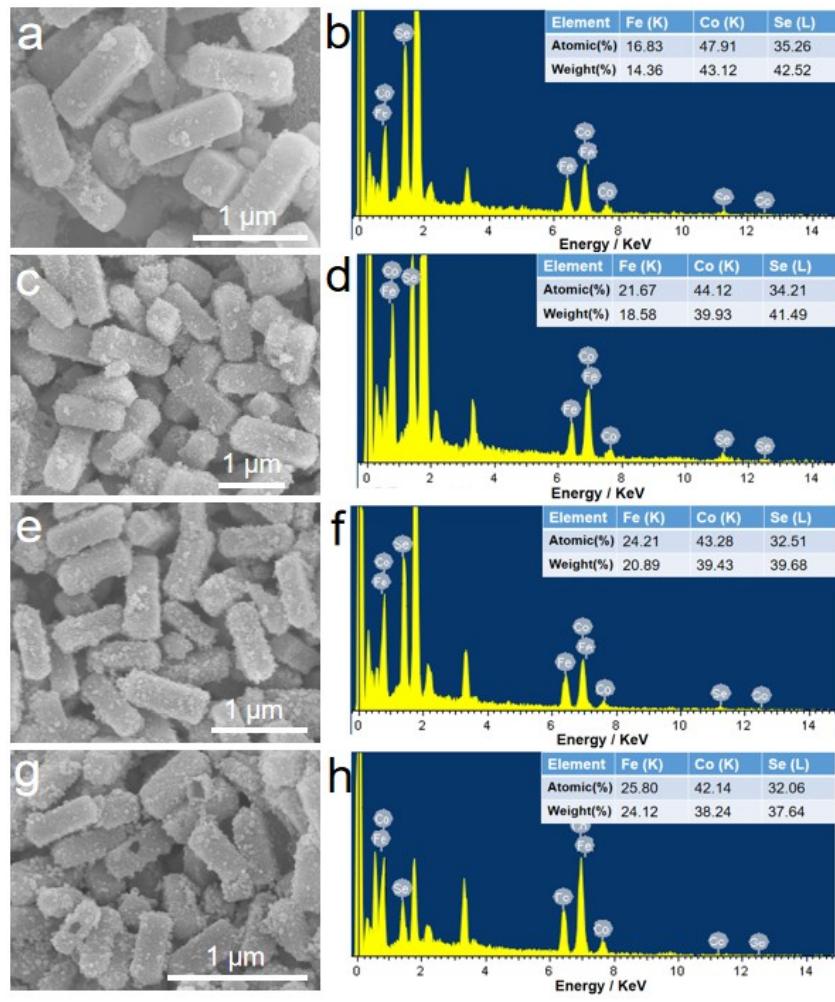


Fig. S10 SEM images, corresponding EDX patterns and the element contents (inset) of different CoSe₂/FeSe₂ DS-HNCs samples: (a, b) CoSe₂/FeSe₂ DS-HNCs-2, (c, d) CoSe₂/FeSe₂ DS-HNCs, (e, f) CoSe₂/FeSe₂ DS-HNCs-6 and (g, h) CoSe₂/FeSe₂ DS-HNCs-8.

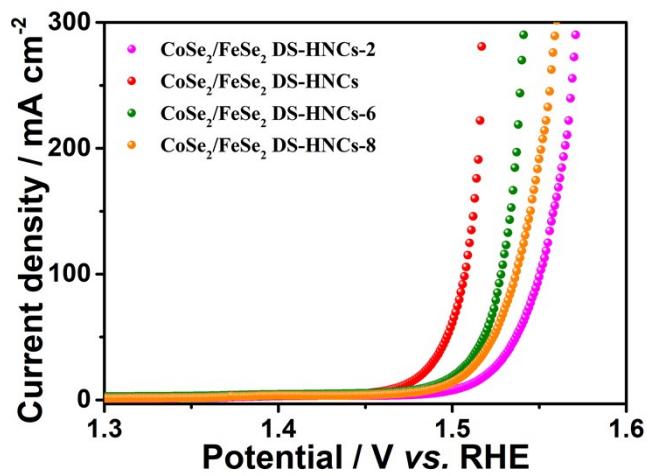


Fig. S11 LSV curves of different CoSe₂/FeSe₂ DS-HNCs samples.

Table S1. EIS data of CoSe₂/FeSe₂ DS-HNCs and CoSe₂ NPs for OER.

	CoSe ₂ /FeSe ₂ DS-	CoSe ₂
	HNCs	NPs
R _s (Ω cm ²)	1.56	1.06
R _{ct} (Ω cm ²)	2.52	7.04
CPE1-T	0.49	0.03
CPE1-P	0.97	0.71

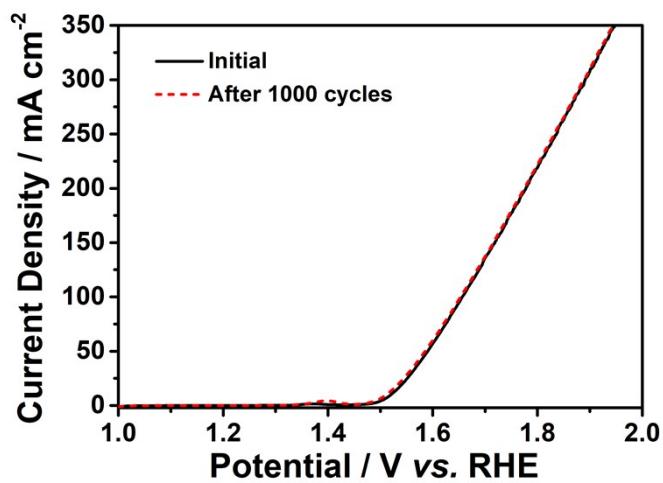


Fig. S12 The LSV curves for the CoSe₂/FeSe₂ DS-HNCs before and after 1000 cycles.

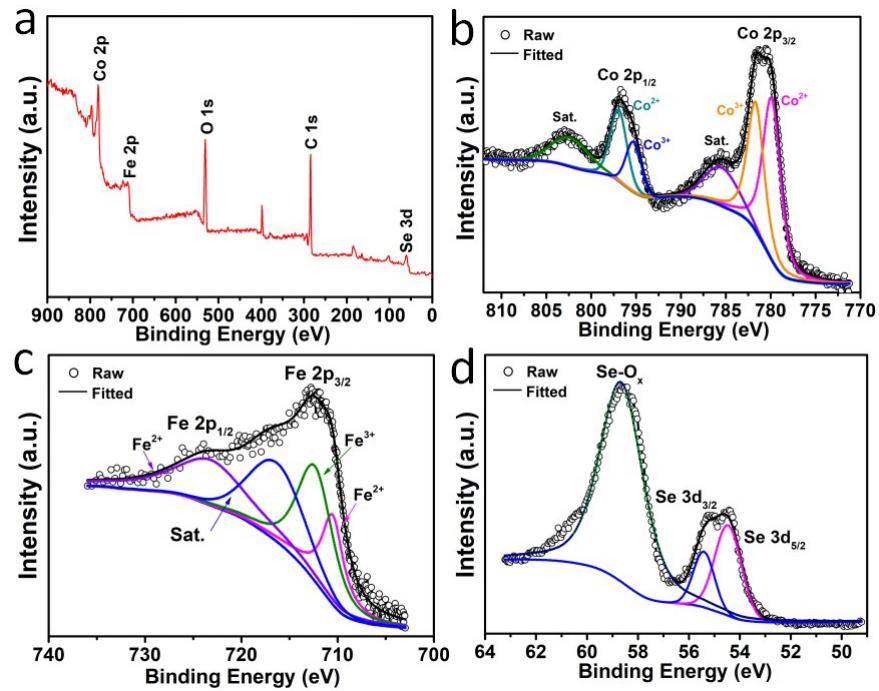


Fig. S13 XPS survey spectrum (a) and high-resolution XPS spectra of Co 2p (b), Fe 2p (c) and Se 3d (d) of the as-prepared CoSe₂/FeSe₂ DS-HNCs after OER test.

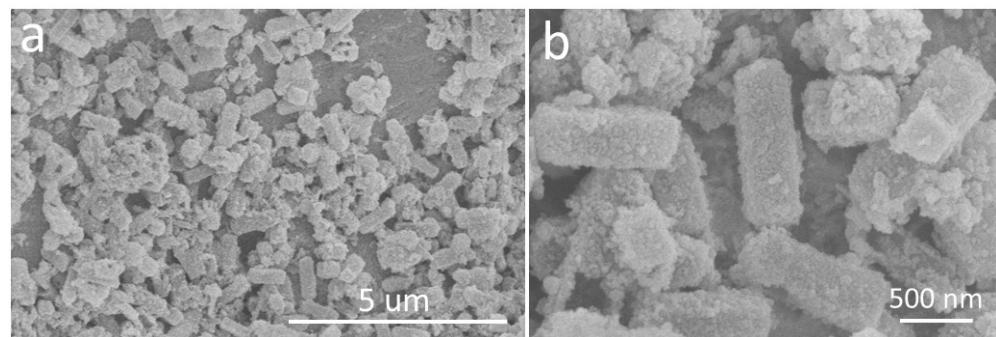


Fig. S14 SEM images of the as-prepared $\text{CoSe}_2/\text{FeSe}_2$ DS-HNCs after OER test.

Table S2. Comparison of OER performance with recently reported metal selenides catalysts in alkaline medium.

Catalysts	Overpotential (mV) at 10 mA cm ⁻²	References
CoSe ₂ nanosheets	320	<i>J. Am. Chem. Soc.</i> 2014 , <i>136</i> , 15670.
Co _{0.85} Se	324	<i>Adv. Mater.</i> 2016 , <i>28</i> , 77.
Ag-CoSe ₂ -belt	320	<i>Angew. Chem. Int. Ed.</i> 2017 , <i>56</i> , 328.
Ni _{0.88} Co _{1.22} Se ₄	320	<i>Chem. Mater.</i> 2017 , <i>29</i> , 7032.
(Ni,Co)Se-GA	250	<i>ACS Catal.</i> 2017 , <i>7</i> , 6394.
Co _{0.85} Se@NC	320	<i>J. Mater. Chem. A</i> 2017 , <i>5</i> , 7001.
NiSe	290	<i>Adv. Energy Mater.</i> 2018 , <i>8</i> , 1702704.
Fe-doped NiSe ₂	268	<i>Angew. Chem. Int. Ed.</i> 2018 , <i>57</i> , 4020.
FeSe ₂ @CoSe ₂ /rGO	260	<i>ACS Appl. Mater. Interfaces</i> 2018 , <i>10</i> , 19258.
Fe _{0.09} Co _{0.13} -NiSe ₂	251	<i>Adv. Mater.</i> 2018 , <i>30</i> , 1802121.
CoSe ₂ /FeSe ₂ DS-HNCs	240	This work