

Supporting Information

Metallic 1T-phase MoS₂ quantum dots/g-C₃N₄ heterojunctions for enhanced photocatalytic hydrogen evolution

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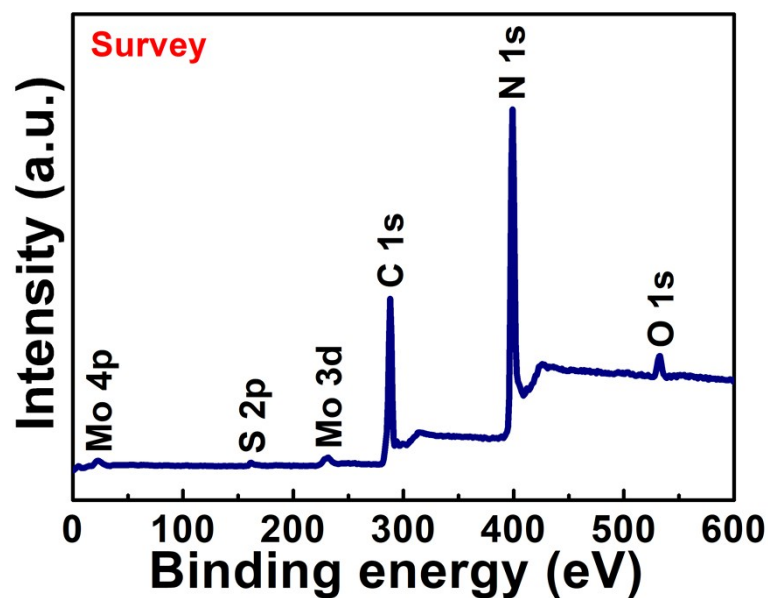


Fig. S1. XPS survey spectra of 1T-MoS₂ QDs@g-C₃N₄ composites (15 wt%).

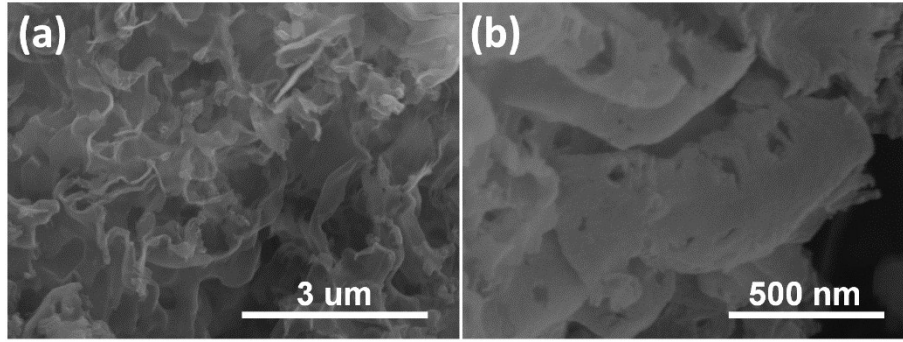


Fig. S2. SEM images of pure g-C₃N₄ NSs.

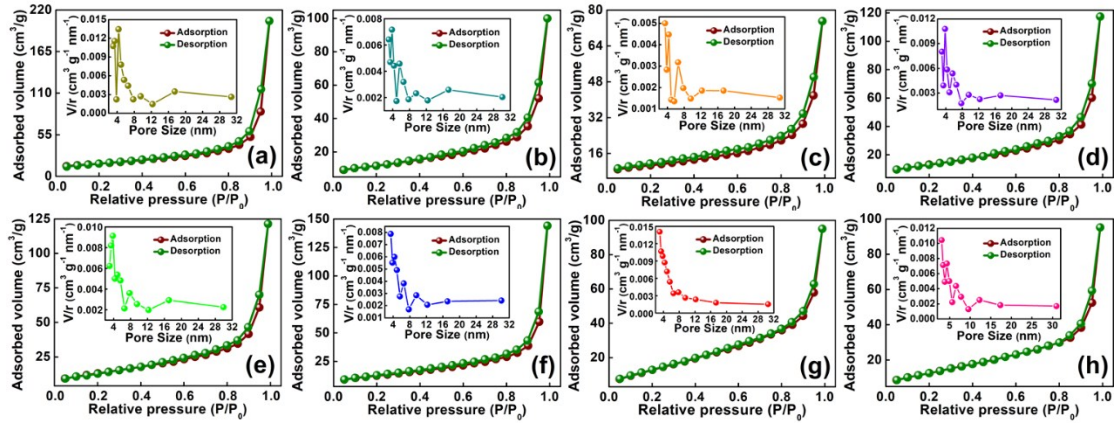


Fig. S3. Nitrogen adsorption/desorption isotherms of (a) g-C₃N₄ NSs and 1T-MoS₂ QDs@g-C₃N₄ composites containing different amounts of 1T-MoS₂ QDs: (b) 1, (c) 3, (d) 5, (e) 7, (f) 9, (g) 15 and (h) 20 wt% (inset shows the corresponding BJH pore size distribution curves).

Table S1. BET surface area of 1T-MoS₂ QDs@g-C₃N₄ composites containing different amounts of 1T-MoS₂ QDs (1, 3, 5, 7, 9, 15 and 20 wt%).

Samples	BET Surface area (m ² g ⁻¹)
1T-MoS ₂ QDs@g-C ₃ N ₄ -1 wt%	41.379
1T-MoS ₂ QDs@g-C ₃ N ₄ -3 wt%	36.004
1T-MoS ₂ QDs@g-C ₃ N ₄ -5 wt%	47.938
1T-MoS ₂ QDs@g-C ₃ N ₄ -7 wt%	48.450
1T-MoS ₂ QDs@g-C ₃ N ₄ -9 wt%	45.018
1T-MoS ₂ QDs@g-C ₃ N ₄ -15 wt%	53.643

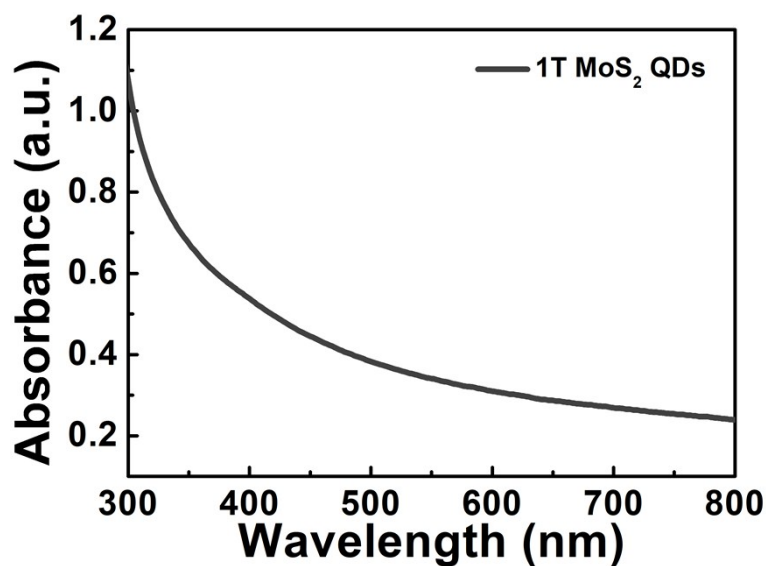


Fig. S4. UV-vis light absorption spectrum of 1T-MoS₂ QDs.

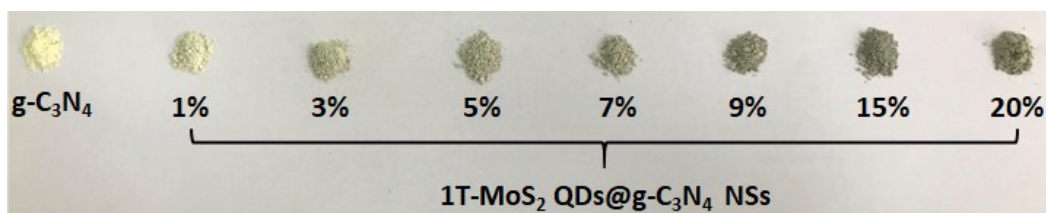


Fig. S5. Optical photographs of g-C₃N₄ NSs and 1T-MoS₂ QDs@g-C₃N₄ composites containing different amounts of 1T-MoS₂ QDs (1, 3, 5, 7, 9, 15 and 20 wt%).

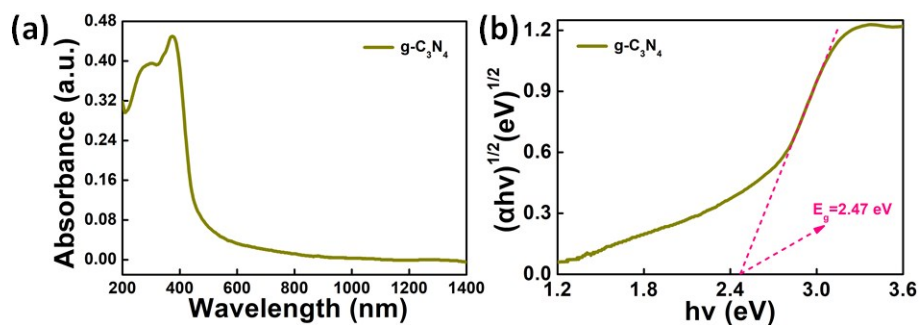


Fig. S6. (a) UV-vis-NIR diffuse reflectance spectra and (b) band gap values of pure g-C₃N₄ NSs.

Table S2. Hydrogen evolution performance of the as-prepared MoS₂@g-C₃N₄

compared with some of the typical semiconductors.

	evolution ($\mu\text{mol g}^{-1}\text{ h}^{-1}$)			
g-C ₃ N ₄ /Ag/MoS ₂	104	100 mg of photocatalysts, 100 mL of aqueous solutions with 15 vol % of TEOA	visible light irradiation, $\lambda > 420\text{ nm}$	1
g-C ₃ N ₄ /RGO/MoS ₂	317	TEOA aqueous solution	visible light irradiation, $\lambda > 420\text{ nm}$	2
2D-2D MoS ₂ /g-C ₃ N ₄	1155	50 mg of photocatalysts, 250 ml 0.1 M TEOA aqueous acetone	300 W Xeon lamp, $\lambda > 420\text{ nm}$	3
MoS ₂ /g-C ₃ N ₄ /GO	1650	4 mg of photocatalysts, 20 mL aqueous solution containing Na ₂ SO ₃ (0.25 M)	450 W xenon lamp with AM 1.5 filter	4
1T-MoS ₂ /O-g-C ₃ N ₄	1841	10 mg of photocatalysts, 100 mL TEOA/H ₂ O (1:9 by volume)	300 W Xeon arc lamp, $\lambda > 400\text{ nm}$	5
1T-MoS ₂ QDs@g-C ₃ N ₄ (this paper)	1857	20 mg of photocatalysts, 20 ml TEOA, 80 ml deionized water	300 W Xeon arc lamp with AM-1.5 filter	

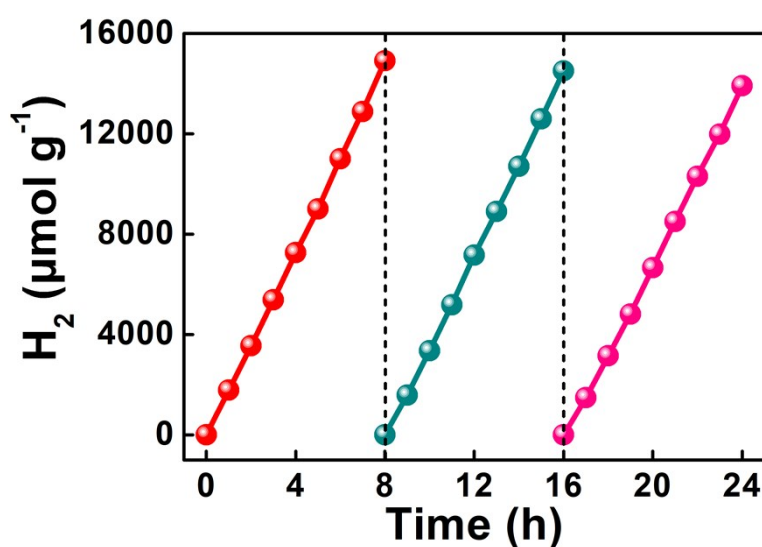


Fig. S7. Stability and recyclability of the 1T-MoS₂ QDs@g-C₃N₄ composites (15 wt%).

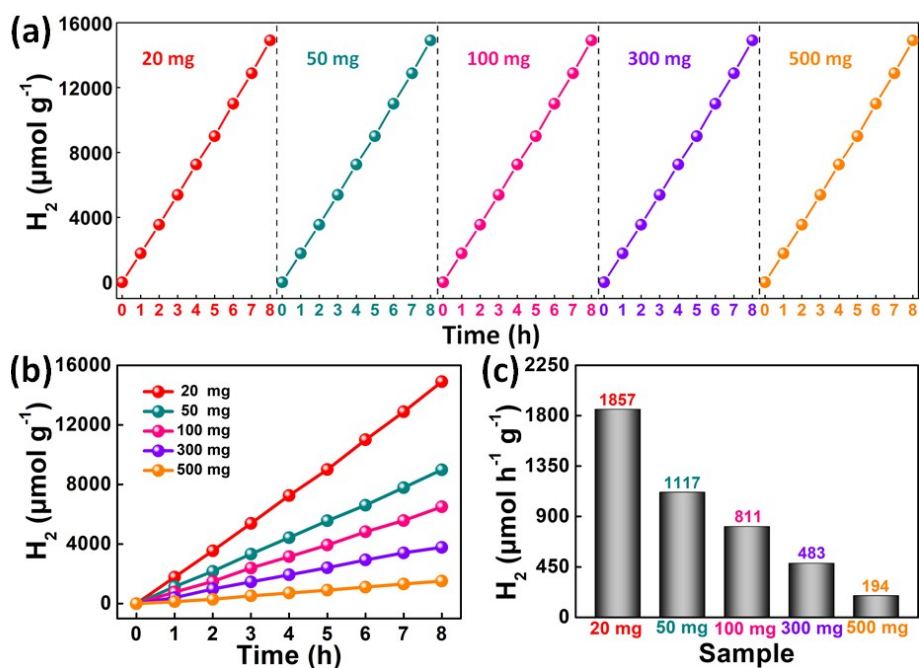


Fig. S8. (a) Cumulated evolution of 1T-MoS₂ QDs@g-C₃N₄ composites (15 wt%) of different dosages with the same concentration. (b) Cumulated evolution and (b) photocatalytic H₂ evolution rate of 1T-MoS₂ QDs@g-C₃N₄ composites (15 wt%) of different dosages and different concentration.

Table S3. Comparison of AQE values over g-C₃N₄ NSs and 11T-MoS₂ QDs@g-C₃N₄ composites containing different amounts of 1T-MoS₂ QDs (1, 3, 5, 7, 9, 15 and 20 wt%) under simulated solar light.

Samples	AQE values (%)
g-C ₃ N ₄ NSs	0.15
1T-MoS ₂ QDs@g-C ₃ N ₄ -1 wt%	1.71
1T-MoS ₂ QDs@g-C ₃ N ₄ -3 wt%	2.20
1T-MoS ₂ QDs@g-C ₃ N ₄ -5 wt%	2.78
1T-MoS ₂ QDs@g-C ₃ N ₄ -7 wt%	3.42
1T-MoS ₂ QDs@g-C ₃ N ₄ -9 wt%	4.54
1T-MoS ₂ QDs@g-C ₃ N ₄ -15 wt%	5.76
1T-MoS ₂ QDs@g-C ₃ N ₄ -20 wt%	4.33

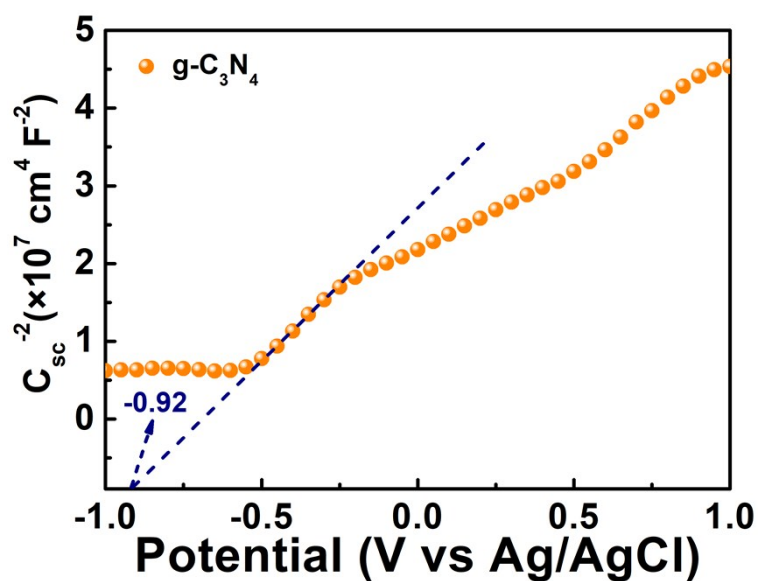


Fig. S9. Mott-Schottky plot of g-C₃N₄ NSs.

References:

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