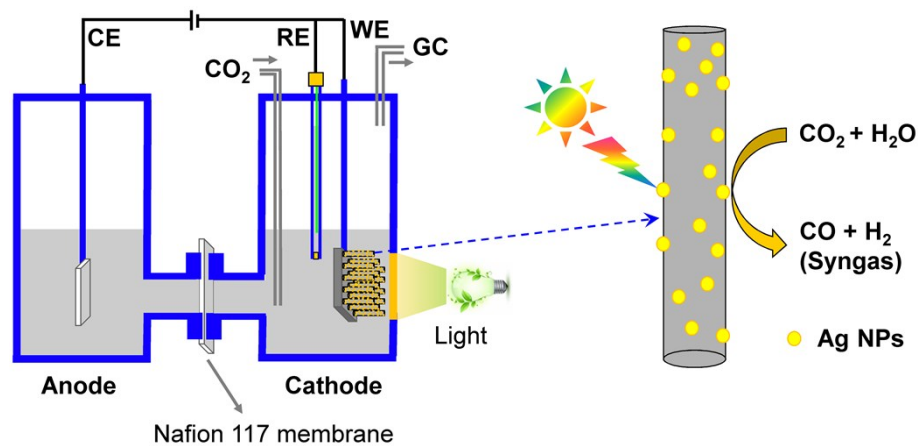


Electronic Supplementary Information (ESI)

Photoelectrocatalytic reduction of CO₂ to syngas over Ag nanoparticle modified p-Si nanowire arrays

Longfu Wei, Jinchi Lin, Shunji Xie,* Wenchao Ma, Qinghong Zhang, Zebin Shen and Ye Wang*

State Key Laboratory of Physical Chemistry of Solid Surfaces, Collaborative Innovation Center of Chemistry for Energy Materials, National Engineering Laboratory for Green Chemical Productions of Alcohols, Ethers and Esters, College of Chemistry and Chemical Engineering, Xiamen University, Xiamen 361005, China.



CE: counter electrode; RE: reference electrode; WE: working electrode

Fig. S1 Schematic illustration of the H-cell photoelectrocatalytic reaction system.

Table S1 The photoelectrocatalytic behaviours of SiNWs and metal/SiNWs in CO₂-saturated 0.1 M KHCO₃ solution at -1.0 V vs. RHE^a

Photocathode	Faradic efficiency (%)				Current density (mA cm ⁻²)	H ₂ /CO molar ratio
	H ₂	CO	HCOOH	CH ₃ OH		
planar Si	95	3.5	0	0	0.23	27
SiNWs	91	7.4	2.4	0	1.6	12
Pt/SiNWs	93	5.5	0.16	0	6.8	17
Pd/SiNWs	73	22	3.5	0	2.4	3.3
Au/SiNWs	30	59	1.9	3.6	3.2	0.51
Ag/SiNWs	55	37	2.1	1.6	2.7	1.5

^a Reaction conditions: Pt plate counter electrode; Ag/AgCl (saturated KCl) reference electrode; light source, Xe lamp equipped with AM 1.5G at 100 mW cm⁻²; reaction time, 4 h.

Table S2 The electrocatalytic behaviours of Ag-x catalysts with different Ag NPs sizes for CO₂ reduction in CO₂-saturated 0.1 M KHCO₃ solution at −1.0 V vs. RHE^a

Cathode	Formation rate (μmol cm ⁻² h ⁻¹)		Faradic efficiency (%)		Current density (mA cm ⁻²)
	H ₂	CO	H ₂	CO	
Ag-4.2	3.0	5.6	32	60	0.50
Ag-6.3	2.9	10	26	66	0.81
Ag-8.2	4.0	14	21	75	1.0
Ag-11	5.2	9.4	33	59	0.85
Ag-14	5.5	7.8	38	54	0.77
Ag-16	3.4	3.7	46	49	0.40

^a Reaction conditions: Pt plate counter electrode; Ag/AgCl (saturated KCl) reference electrode; reaction time, 2 h.

Table S3 The electrocatalytic behaviours of Ag-8.2 catalyst for CO₂ reduction in CO₂-saturated 0.1 M KHCO₃ solution at different applied potentials^a

Potential (V vs. RHE)	Formation rate (μmol cm ⁻² h ⁻¹)		Faradic efficiency (%)		Current density (mA cm ⁻²)
	H ₂	CO	H ₂	CO	
−0.8	2.7	2.0	56	41	0.26
−1.0	4.0	14	21	75	1.0
−1.2	5.0	27	15	81	1.8
−1.4	26	31	42	51	3.3
−1.6	57	37	57	37	5.4

^a Reaction conditions: Pt plate counter electrode; Ag/AgCl (saturated KCl) reference electrode; reaction time, 2 h.

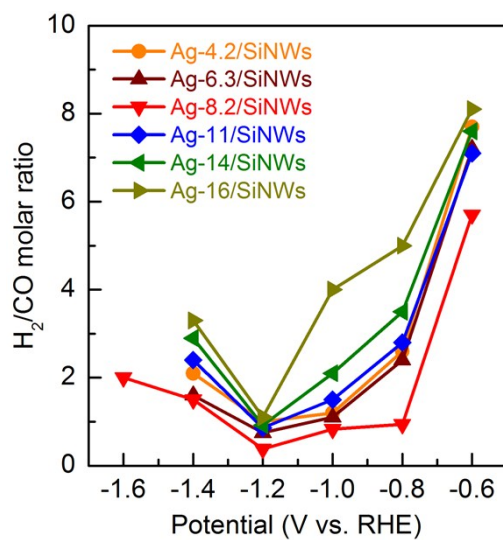


Fig. S2 The H₂/CO molar ratio of Ag-*x*/SiNWs catalysts at different applied potentials.

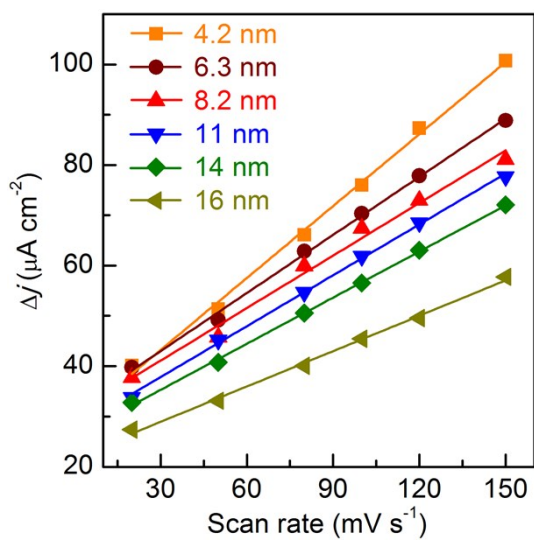


Fig. S3 Charging current density differences (Δj) of Ag NPs with different sizes plotted against scan rates.

Table S4 The electrochemical surface area (ECSA) of Ag NPs with different sizes.

Ag NPs (nm)	Slop ($\mu\text{F cm}^{-2}$)	C_{dl} ($\mu\text{F cm}^{-2}$)	R_f	ECSA (cm^2)
4.2	477	239	12	12
6.3	385	193	9.6	9.6
8.2	348	174	8.7	8.7
11	337	169	8.4	8.4
14	305	153	7.6	7.6
16	234	117	5.9	5.9

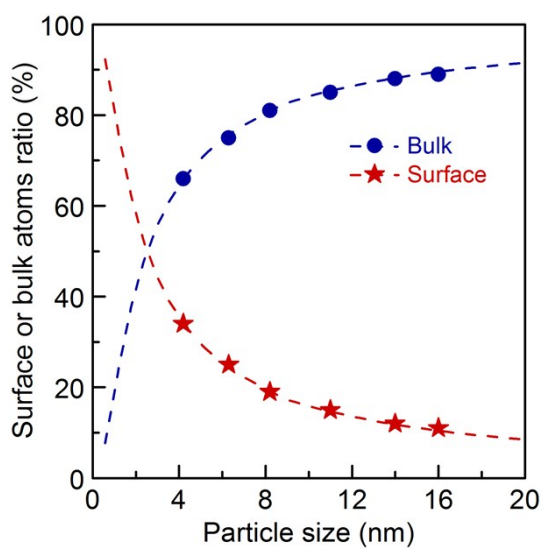


Fig. S4 The surface and bulk atoms ratio of Ag NPs with different sizes calculated from the theoretical Ag NPs model.