Supporting information for

2D MnO₂ nanosheets generated signal transduction with 0D carbon quantum dots: synthesis strategy, dual-mode behavior and glucose detection

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Figure S1 Emission spectra of obtained CDs (20 μ g/mL) in aqueous solutions in the presence of different volume of HNO₃ (0-500 μ L) (excitation wavelength at 452 nm).



Figure S2 XRD pattern of CDs.



Figure S3 Raman spectrum of CDs.



Figure S4 $N_2\,adsorption\mbox{-}desorption\mbox{-}isotherm\mbox{ of }CDs\mbox{-}MnO_2\mbox{ nanocomposite}.$



Figure S5 EDS spectra of (A) CDs, (B) MnO₂ nanosheets, (C) CDs-MnO₂ nanocomposite, respectively.



Figure S6 EDX elemental mapping of CDs-MnO₂ nanocomposite.



Figure S7 Normalized emission spectra of CDs (20 μ g/mL) in aqueous solutions at different excitation wavelengths from 380 to 460 nm.

Table S1. Summary of the quantum yields (QY) and applications of CDs prepared by using various materials.

Materials	QY/%	Fluorescence Application		Ref.
Bergamot	50.78	blue	Hg ²⁺ and Fe ³⁺ detection	1
Honey	19.8	blue	Fe ³⁺ detection and cell imaging	2
Aloe	10.37	yellow tartrazine detection		3
Lotus root	19.0	blue	Hg(II) ions detection and cell imaging	4
Bamboo leaves	7.1	blue	copper(II) ion detection	5
Prawn shells	9	blue	copper ions detection	6
Garlic	13	blue	Fe(III) ions detection and cell imaging	7
Bauhinia flower	27	blue	Fe ³⁺ and adenosine triphosphate detection	8
Tulsi leaves	9.3	green	Pb ²⁺ ions detection and live cell imaging	9
Gynostemma	5.7	blue	bioimaging and antioxidant	10
Peony	11.6	green	glucosedetection	This work

[1] J. Yu, N. Song, Y.K. Zhang, S.X. Zhong, A.Jun. Wang, J.R. Chen, Green preparation of carbon dots by Jinhua bergamot for sensitiveand selective fluorescent detection of Hg²⁺and Fe³⁺, Sensor. Actuat. B-Chem. 214 (2015) 29-35.

- [2] X.M. Yang, Y. Zhuo , S.S. Zhu, Y.W. Luo, Y.J. Feng, Y. Dou, Novel and green synthesis of high-fluorescent carbon dots originated from honey for sensing and imaging, Biosens. Bioelectron. 60 (2014) 292-298.
- [3] H. Xu, X.P. Yang, G. Li, C. Zhao, and X.J. Liao, Green synthesisoffluorescent carbon dots for selective detection of tartrazine in food samples, J. Agric. Food Chem. 63 (2015) 6707-6714.
- [4] D. Gu, S.M. Shang, Q. Yu, J. Shen, Green synthesis of nitrogen-doped carbon dots from lotus root for Hg(II) ions detection and cell imaging, Appl. Surf. Sci. 390 (2016) 38-42.
- [5] Y.S. Liu, Y.N. Zhao, Y.Y. Zhang, One-step green synthesized fluorescent carbon nanodots frombamboo leaves for copper(II) ion detection, Sensor. Actuat. B-Chem. 196 (2014) 647-652.
- [6] G. Gedda, C.Y. Lee, Y.C. Lin, H.F. Wu,Green synthesis of carbon dots from prawn shells for highly selective and sensitive detection of copper ions, Sensor. Actuat. B-Chem. 224 (2016) 396-403.
- [7] Y.F. Chen, Y.Y. Wu, B. Weng, B. Wang, C.M. Li, Facile synthesis of nitrogen and sulfur co-doped carbon dots and application for Fe(III) ions detection and cell imaging, Sensor. Actuat. B-Chem. 23 (2016) 689–696.
- [8] Q.T. Huang, Q. Li, Y.F. Chen, L.L. Tong, X.F. Lin, J.J. Zhu, Q.X. Tong, High quantum yield nitrogen-doped carbon dots: green synthesis and application as"off-on" fluorescent sensors for the determination of Fe³⁺ and adenosine triphosphate in biological samples, Sensor. Actuat. B-Chem. 276 (2018) 82-88.
- [9] A. Kumar, A.R. Chowdhuri, D. Laha, T.K. Mahto, P. Karmakar, S.K. Sahu, Green synthesis of carbon dots from ocimum sanctum for effective fluorescent sensing of Pb²⁺ ions and live cell imaging, Sensor. Actuat. B-Chem. 242 (2017) 679-686.
- [10] X.J. Wei, L. Li, J.L. Liu, L.D.Yu, H.B. Li, F. Cheng, X.T. Yi, J.M. He,B.S. Li, Green synthesis of fluorescent carbon dots from gynostemma for bioimaging and antioxidant in zebrafish, ACS Appl. Mater. Interfaces 2019, 11, 9832-9840.



Figure S8 UV-vis absorption spectra of CDs (20 μ g/mL, red line), MnO₂ (750 μ M, blue line) and CDs-MnO₂ (20 μ g/mL CDs + 750 μ M MnO₂, green line) in aqueous solution.



Figure S9 UV-visible spectra of CDs (20 μ g mL⁻¹) in aqueous solutions with the addition of various concentrations of MnO₂ nanosheets from 0 to 750 μ M. (B) Linear calibration curve between the absorbance of CDs aqueous solution and the concentration of MnO₂ nanosheets. The error bars represent the standard deviation of three replicate measurements.



Figure S10 Fluorescent intensity of CDs-MnO₂ nanocomposite (20 µg/mL CDs + 300 µM MnO₂) in aqueous solutions at various pH values (A);different concentrations of NaCl range from 10-80 mM (B); ultraviolet irradiation from 0 to 3600 s (C); different temperature ranging from 20 to 80°C(D). ($\lambda_{ex} = 452$ nm, $\lambda_{em} = 523$ nm). The error bars represent the standard deviation of three replicate measurements.



Figure S11 UV-vis absorption of MnO₂ nanosheets (750 μ M) (blue line) and optimal fluorescence excitation and emission spectra of the CDs (20 μ g/mL) in aqueous solutions (green line) ($\lambda_{ex} = 452$ nm, $\lambda_{em} = 523$ nm).



Figure S12 Lifetime measurements of CDs (blue line) and CDs-MnO₂ (red line) ($\lambda_{ex} = 452$ nm, $\lambda_{em} = 523$ nm).



Figure S13 UV-visible spectra of CDs-MnO₂ nanocomposite (20 μ g/mL CDs + 300 μ M MnO₂) in aqueous solutions with the addition of various concentrations of H₂O₂ from 0 to 45 μ M. (B) Linear calibration curve between the absorbance of CDs-MnO₂ nanocomposite and the concentration of H₂O₂. The error bars represent the standard deviation of three replicate measurements.

Materials Detection mo		Detection limit	Ref.
Lanthanum	Colorimetric	37.5 μΜ	1
incorporated			
MCM-41			
CePO ₄ :Tb/Au	Fluorometric	5.25 μΜ	2
Cu/Ni based	Electrochemical	0.06 μΜ	3
MOF			
Bodipy	Fluorometric	2.19 µM	4
derivative			
MOF nanosheet	Fluorometric	1.3 μΜ	5
Peony-derived	Fluorometric	0.18 μΜ	This
carbon dots-			work
MnO_2			

Table S2 Comparison of different materials for the determination of glucose

 S. Jabariyan, M. A. Zanjanchi, M. Arvand, S. Sohrabnezhad, Colorimetric detection of glucose using lanthanum-incorporated MCM-41, Spectrochim. Acta A 203 (2018) 294-300.

[2] G.W. Yan, Y. Zhang, W.H. Di, W.P. Qin, Synthesis of luminescent CePO₄:Tb/Au composite for glucose detection, Dyes and Pigments 159 (2018) 28-34.

[3] X.L. Xiao, S.H. Peng, C. Wang, D. Cheng, N. Li, Y.L. Dong, Q.H. Li, D.G. Wei, P. Liu,
Z.Z. Xie, D.Y. Qu, X. Li, Metal/metal oxide@carbon composites derived from bimetallic
Cu/Ni-based MOF and their electrocatalytic performance for glucose sensing, J. Electroanal.
Chem. 841 (2019) 94-100.

[4] B. Unal, M. Akarsu, R. Kilincarslan, D.O. Demirkol, S. Timur, B. Cetinkaya, Novel fluorescence assay using μ -wells coated by BODIPY dye as an enzymatic sensing platform, Measurement 135 (2019) 145-150.

[5] D. Ning, Q. Liu, Q. Wang, X.M. Du, W.J. Ruan, Y. Li, Luminescent MOF nanosheets for enzyme assisted detection of H₂O₂ and glucose and activity assay of glucose oxidase, Sensors Actuat. B-Chem. 282 (2019) 443-448.

Sample	Spiked (µM)	Measured (µM)	Recovery (%)	RSD (%)
1	0	38.05±0.1	/	/
2	50.0	85.6±0.14	95.1	2.62
3	100.0	143.9±0.17	105.8	4.08
4	150.0	187.3±0.12	99.5	2.19

Table S3 Determination of glucose in human serum samples.