

**Electronic Supporting Information (ESI)**

**Electrodeposition of Ni-Co-Fe mixed sulfide ultrathin nanosheets on Ni nanocones: low-cost, durable and high performance catalyst for electrochemical water splitting**

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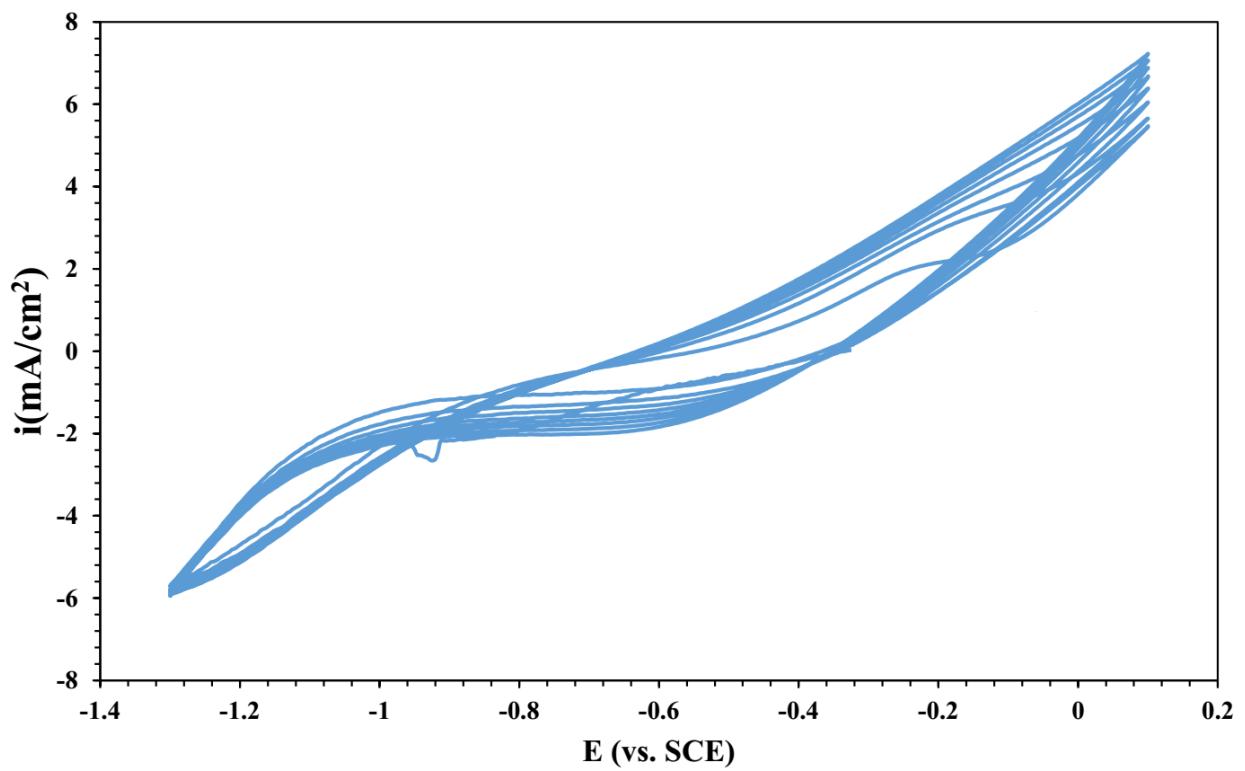


Fig. S1. CV curves during electrochemical deposition of Ni-Fe-Co-S nanosheets

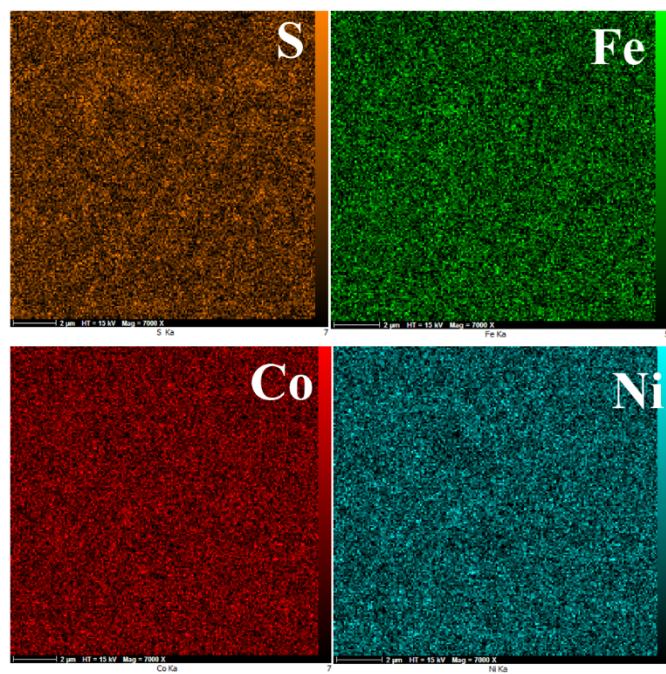


Fig .S2. Elemental mapping of Ni-Fe-Co-S electrode.

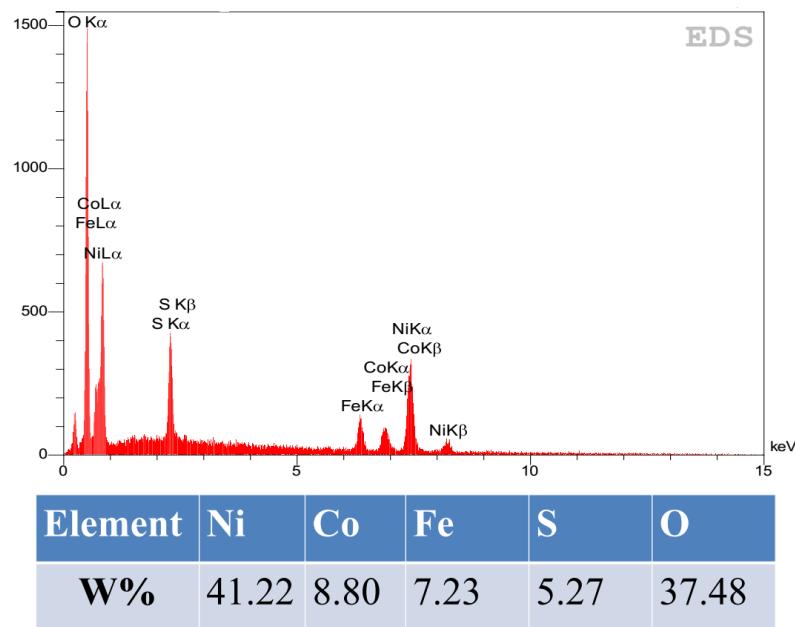


Fig. S3. EDS pattern of Ni-Fe-Co-S electrode.

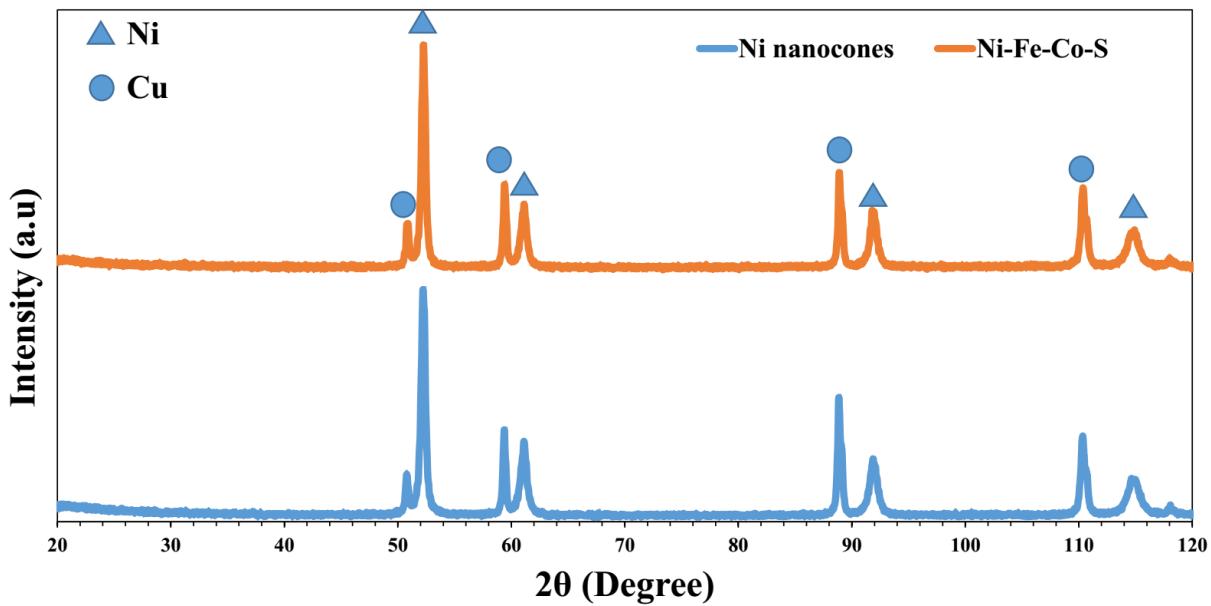


Fig. S4. XRD pattern of Ni nanocones and Ni-Fe-Co-S samples

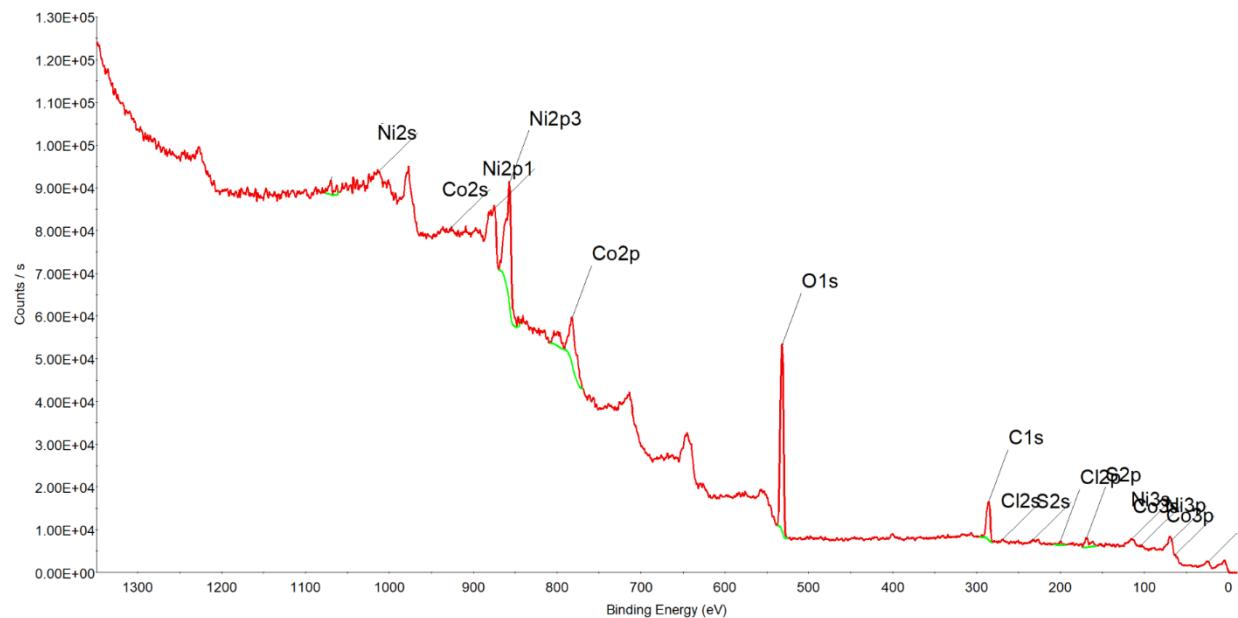


Fig. S5. XPS spectrum of Ni-Fe-Co-S nanosheets.

Table S1. Comparison of electrochemical catalytic hydrogen evolution of Ni-Fe-Co-S @NNCs with other recently reported sulfide-based catalyst in alkaline solution.

Catalyst	Electrolyte	$\eta_{10}$ (mV vs.RHE)	$\eta_{20}$ (mV vs.RHE)	$\eta_{100}$ (mV vs.RHE)	b (mV/dec)	Ref
Ni-Fe-Co-S	1.0 M KOH	106	134	215	95	<b>This work</b>
Fe <sub>0.1</sub> -NiS <sub>2</sub> NA/T	1.0 M KOH	200	243	310	108	[1]
MoO <sub>x</sub> /Ni <sub>3</sub> S <sub>2</sub> /NS	1.0 M KOH	136		310		[2]
Ni <sub>x</sub> Co <sub>3-x</sub> S <sub>4</sub> /Ni <sub>3</sub> S <sub>2</sub> /NF	1.0 M KOH	160		320	95	[3]
V-NiS <sub>3</sub> -NW	1.0 M KOH		203	350	112	[4]
Ni-Ni <sub>3</sub> S <sub>2</sub> /NF	1.0 M KOH	110	160	230		[5]
Ni <sub>3</sub> S <sub>2</sub> /AT-NF	1.0 M KOH	200			107	[6]
NiCo <sub>2</sub> S <sub>4</sub>	1.0 M KOH	210				[7]
NiS <sub>2</sub> nanosheets/CC	1.0 M KOH	149			104	[8]
Ni <sub>3</sub> S <sub>2</sub> nanocomposites /CNT	1.0 M KOH	340			167	[9]
NiS Porous/ Copper plates	1.0 M KOH	301			107	[10]
NiS-Ni <sub>9</sub> S <sub>8</sub> -NiSe-NR/ NF composite	1.0 M KOH	112	148	243	94	[11]
NiS-Ni <sub>2</sub> P <sub>2</sub> S <sub>6</sub> /NF	1.0 M KOH	140			72.8	[12]

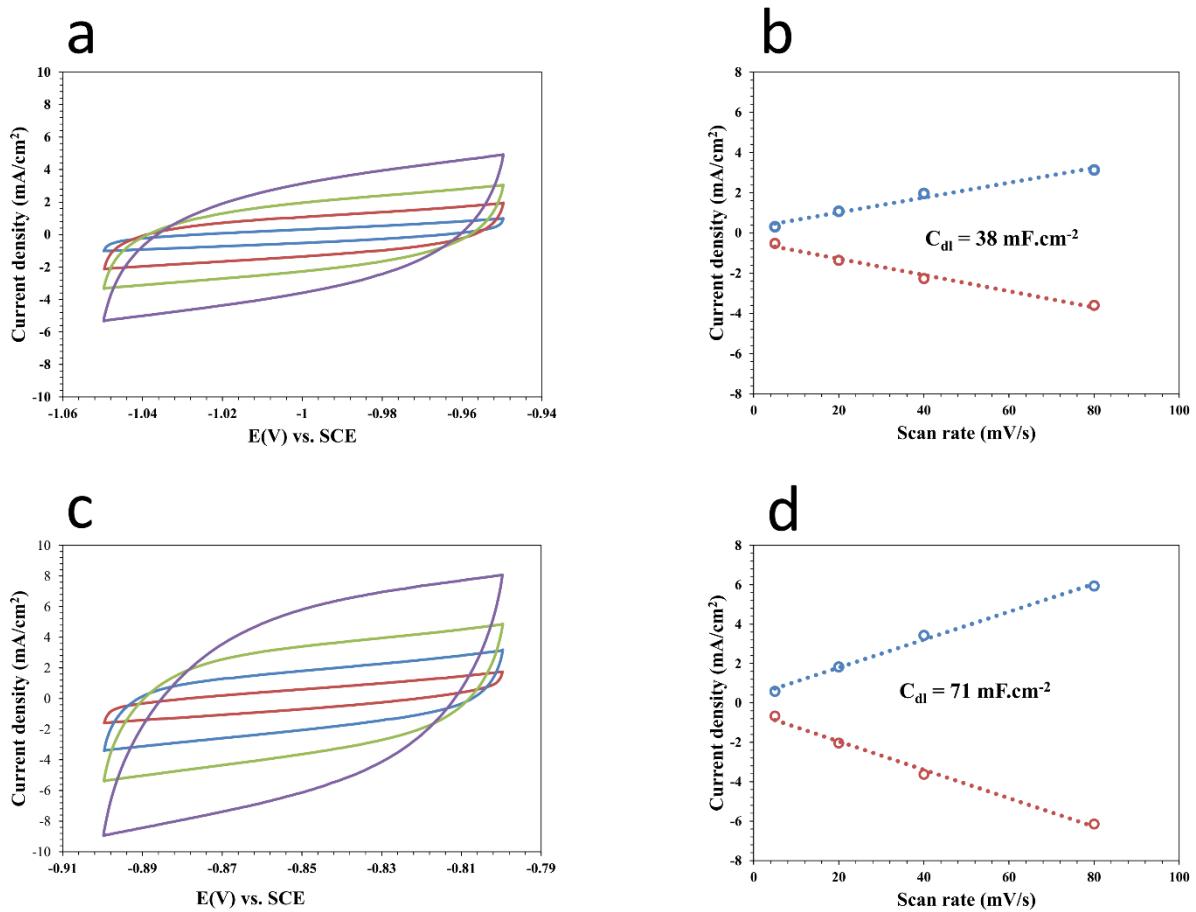


Fig. S6. CV curves of a) NNCs and c) Ni-S@NNCs together with i-scan rates curves of b) NNCs and d) Ni-S@NNCs.

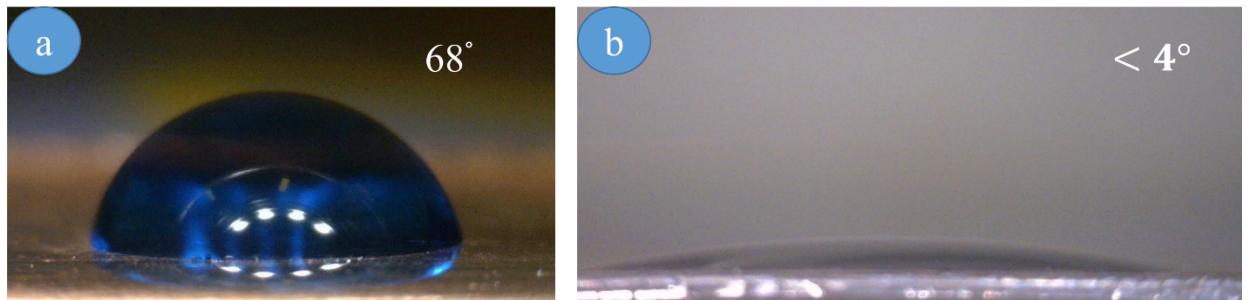


Fig. S7. Contact angle of a) flat Ni and b) Ni-S@NNCs electrodes.

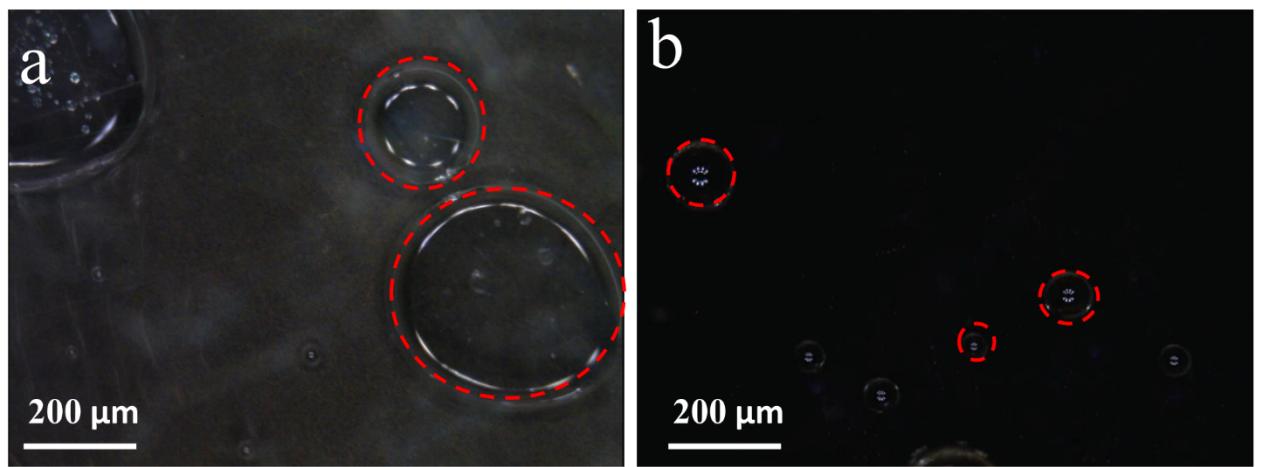


Fig. S8. Bubbles size on the a) flat Ni and b) Ni-S@NNCs electrodes during HER.

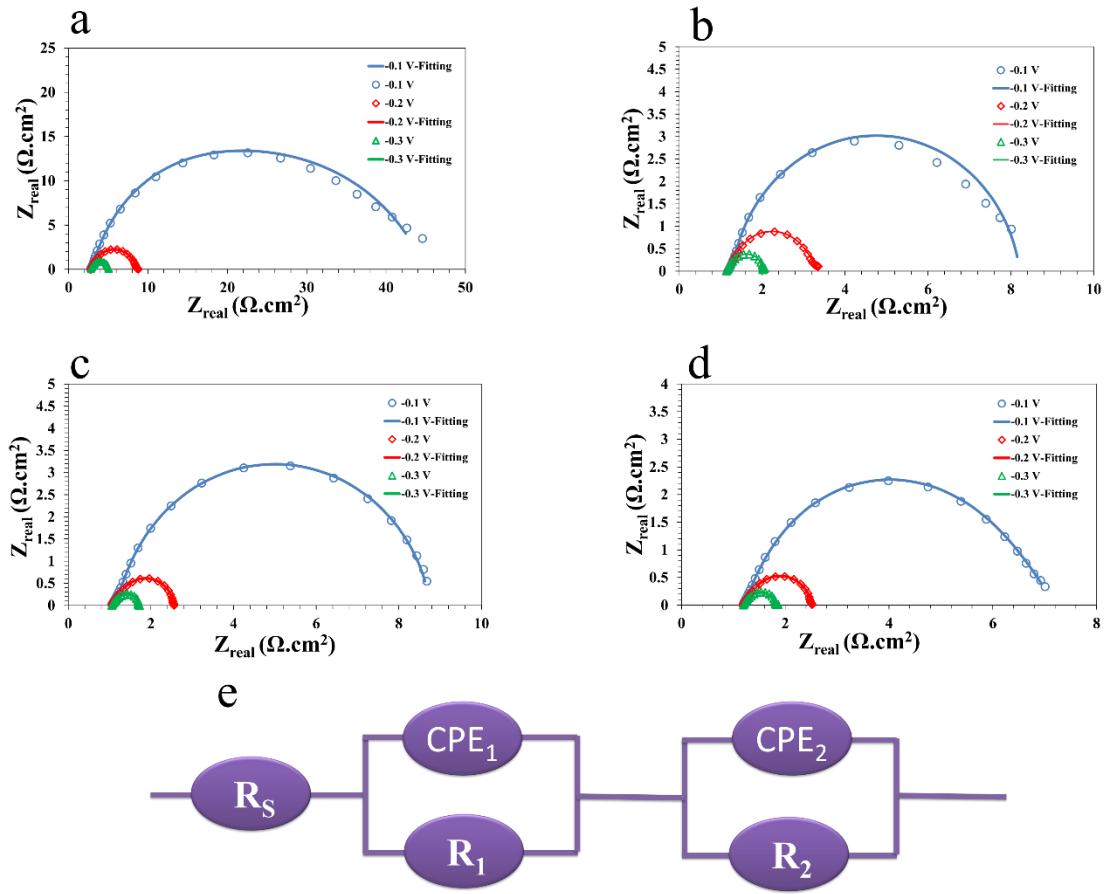


Fig.S9. Nyquist curves of different electrodes at varying overpotentials during HER on the a) Ni-S, b) Ni-Co-S and c) Ni-Fe-S d) Ni-Fe-Co-S and e) equivalent electrical circuit used for fitting the impedance data.

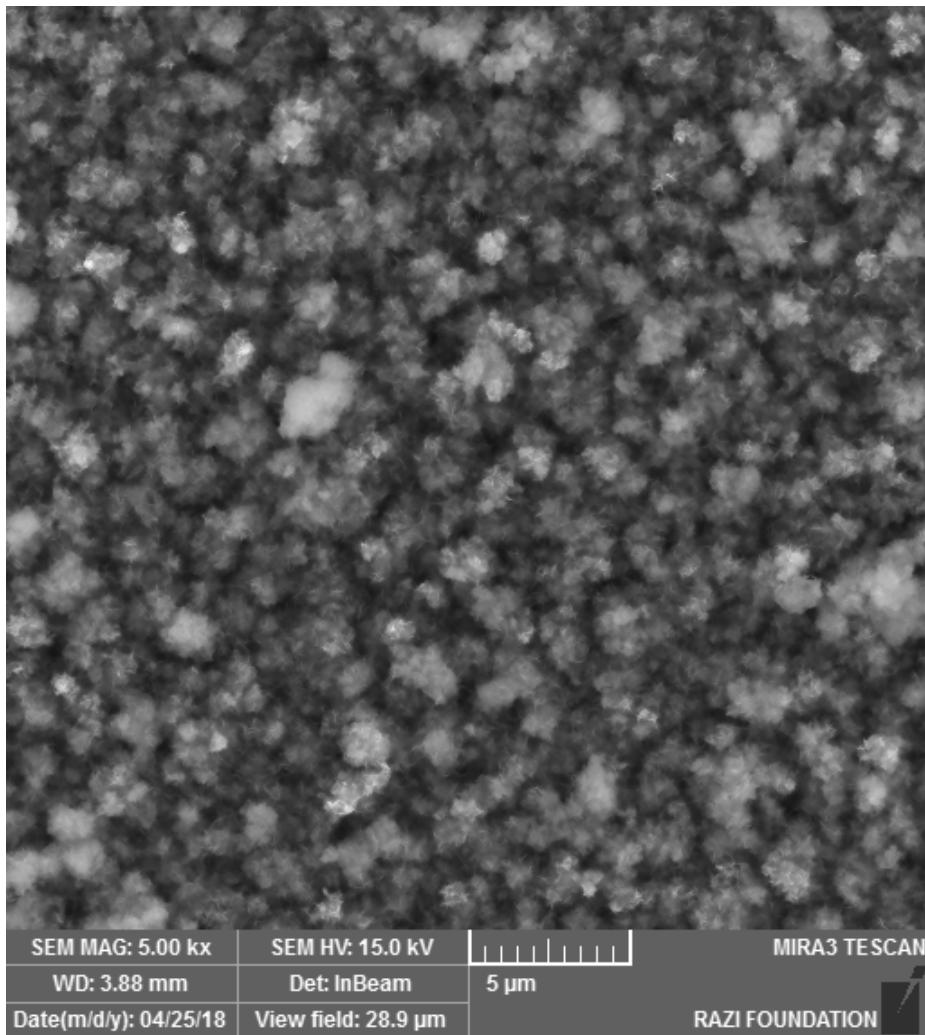


Fig. S10. FESEM image of Ni-Fe-Co-S nanosheets after long-term stability test.

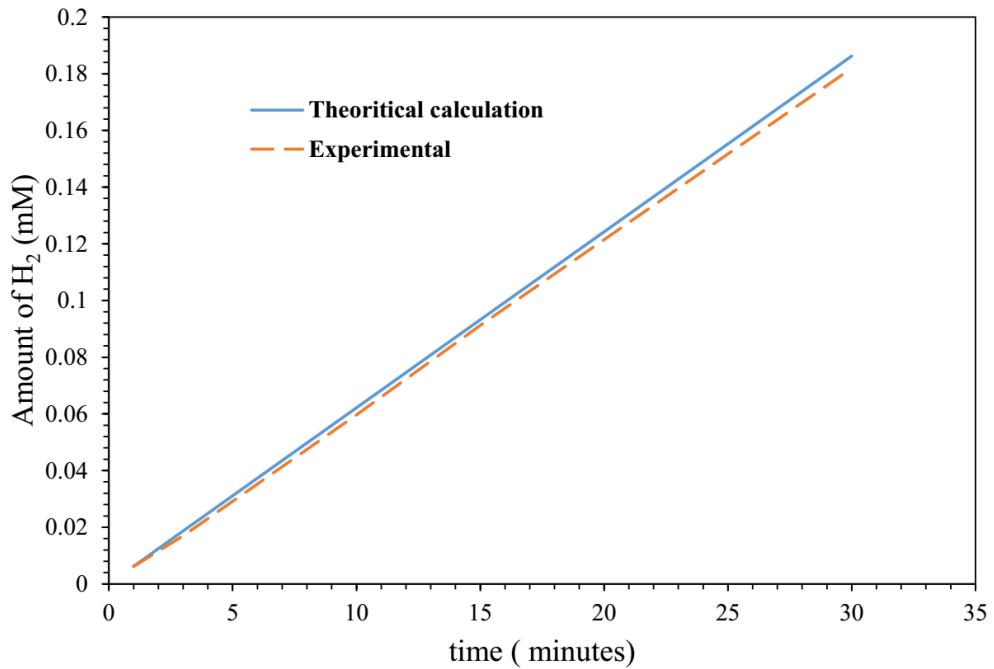


Fig. S11. Current efficiency for HER

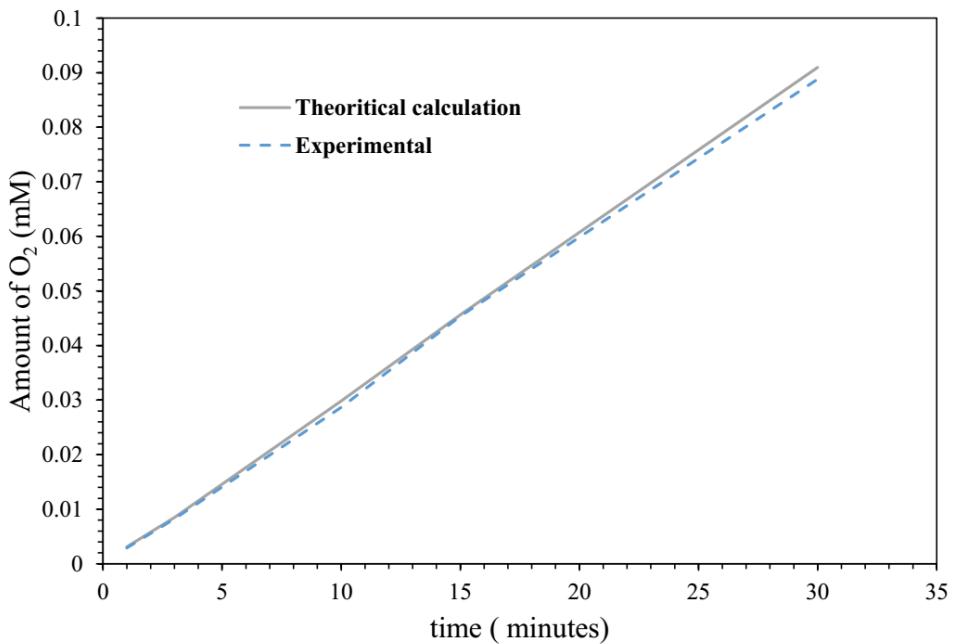


Fig. S12. Current efficiency for OER

Table.S2: Comparison of electrochemical catalytic oxygen evolution of Ni-Fe-Co-S @NNCs with other recently reported sulfide-based catalyst in alkaline solution.

Catalyst	Electrolyte	$\eta_{10}$ (mV vs.RHE)	$\eta_{20}$ (mV vs.RHE)	$\eta_{100}$ (mV vs.RHE)	b (mV/dec)	Ref
Ni-Fe-Co-S	1.0 M KOH	207	232	272	63	<b>This work</b>
Fe-Ni <sub>3</sub> S <sub>2</sub> /FeNi	1.0 M KOH	283	320		54	[13]
N-Ni <sub>3</sub> S <sub>2</sub> /NF	1.0 M KOH			340	70	[5]
FeNiS <sub>2</sub> NNs	1.0 KOH	310			46	[14]
CoMoS <sub>3</sub> nanotube	1.0 KOH	320	370			[15]
Ni/NiS	1.0 KOH		320	390	109.8	[16]
NiCo <sub>2</sub> S <sub>4</sub>	1.0 KOH	260			40	[7]
Ni <sub>3</sub> S <sub>2</sub> Nanorod	0.1 KOH	217				[6]
Zn-Ni <sub>3</sub> S <sub>2</sub> /NF	1.0 KOH			300	87	[17]
Mo-doped Ni <sub>3</sub> S <sub>2</sub>	1.0 KOH			400	45.5	[18]
CoMoS <sub>4</sub> /Ni <sub>3</sub> S <sub>2</sub>	1.0 KOH	200			63	[19]
MoS <sub>2</sub> /Ni <sub>3</sub> S <sub>2</sub>	1.0 KOH	218			88	[20]
Ni <sub>3</sub> S <sub>2</sub> /Co <sub>9</sub> S <sub>8</sub>	1.0 KOH			340	66	[21]
CoPS	1.0 KOH	308.1				[22]

## References

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