Leaching- and Sintering-Resistant Hollow or Structurally Ordered Intermetallic PtFe Alloy Catalyst for Oxygen Reduction Reaction

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Figure S1 Powder X-ray diffraction patterns of PtCo (a) and PtNi (b) alloy samples obtained at different feeding Co²⁺ or Ni²⁺ contents; the standard powder diffraction data for Pt (PDF card #65-2868), intermetallic PtCo (PDF card #43-1358) and intermetallic PtNi (PDF card #65-2797) are shown for comparison.



Figure S2 TEM images, particle distribution histograms and particle morphology statistics (**e**) of PtFe/C@NC-0.5-800 (**a**), PtFe/C@NC-3-800 (**b**), PtFe/C@NC-6-800 (**c**) and PtFe/C@NC-9-800 (**d**).



Figure S3 TEM images of PtFe/C@NC-9-800 samples after treatment with NaOH (a) or HF (b).



Figure S4 XPS survey (a), N1s XPS spectra (b), HRTEM image (c) and pore size distribution from the BJH

method of PtFe/C@NC-9-800 catalyst.



Figure S5 (a) The CO stripping curves of ordered PtFe (PtFe/C@NC-3-800), disordered PtFe (PtFe/C@NC-9-800) and Pt/C catalysts recorded at room temperature in CO-saturated 0.1 M HClO₄ solution. (b) The specific electrochemically active surface area values of ordered PtFe (PtFe/C@NC-3-800), disordered PtFe (PtFe/C@NC-9-800) and Pt/C catalysts calculated from the ECSA-CO and Pt mass loaded on the RDE working electrode. (c) The ratio between ECSA values determined by ECSA-CO and ECSA-H_{upd}.



Figure S6 Pt 4f XPS spectra of ordered PtFe (PtFe/C@NC-3-800), disordered PtFe (PtFe/C@NC-9-800) and

Pt/C catalysts.



Figure S7 The Tafel curves of ordered PtFe (PtFe/C@NC-3-800), disordered PtFe (PtFe/C@NC-9-800) and

Pt/C catalysts.



Figure S8 TEM images and particle-distribution histograms of a commercial Pt/C catalyst before (**a**) and after (**b**) 30,000 CV cycles between 0.6 and 1.1 V in O₂-saturated 0.1 M HClO₄ at a scan rate of 50 mV s⁻¹.



Figure S9 TEM images and particle-distribution histograms of ordered PtFe (PtFe/C@NC-3-800) catalyst before (**a**) and after (**b**) 30,000 CV cycles between 0.6 and 1.1 V in O₂-saturated 0.1 M HClO₄ at a scan rate of 50 mV s⁻¹.



Figure S10 TEM images and particle-distribution histograms of disordered PtFe (PtFe/C@NC-9-800) catalyst before (**a**) and after (**b**) 30,000 CV cycles between 0.6 and 1.1 V in O₂-saturated 0.1 M HClO₄ at a scan rate of 50 mV s⁻¹.

Sample	2θ/degre e (111)	Lattice parameter (1/Å)	Lattice constant (%)
Pt/C JCPDS:65-2868	39.761	0.2206	
Ordered PtFe alloy	40.433	0.2261	2.50
Disordered PtFe alloy	40.315	0.2238	1.45

Table S1. The parameters of catalysts According to the XRD patterns and the Debye-Scherrer equation.

Sample	Pt loading (ug)	ECSA- CO (m²/g)	Half- wave potential (V)	Exchange currents density (A/cm ²)	Mass activity @0.9V (A/mgPt)	Specific activity @0.9V (mA/cm ²)
Pt/C	5.90	69	0.905	7.1×10 ⁻⁶	0.22	0.31
Ordered PtFe alloy	3.18	73	0.933	7.6×10 ⁻⁵	1.48	2.03
Disordered PtFe alloy	2.82	81	0.930	7.4×10 ⁻⁵	1.47	1.81

Table S2. Pt loading from ICP-MS, ECSA-CO, half-wave potential, the exchange currents and the calculated mass activity and specific activity at 0.9V of commercial Pt/C, ordered PtFe alloy (PtFe/C@NC-3-800) and disordered PtFe alloy (PtFe/C@NC-9-800) catalysts.

	Initial				After 30k CV cycles			
Sample	half- wave potenti al (V)	mass activity @0.9V (A/mg Pt)	specific activity @0.9V (mA/cm ²)	Pt/Fe molar ratio	Half- wave potenti al (V)	mass activity @0.9V (A/mg Pt)	specific activity @0.9V (mA/c m ²)	Pt/Fe molar ratio
Pt/C	0.905	0.22	0.31	No	0.864	0.082	0.185	No
Ordered PtFe alloy	0.933	1.48	2.03	76.9/23.1	0.927	1.17	1.65	78.2/21. 8
Disordered PtFe alloy	0.930	1.47	1.81	87.7/12.3	0.919	0.85	1.18	91.7/8.3

Table S3. Half-wave potential, the calculated mass activity and specific activity at 0.9V, and the Pt/Fe molar ratio

from ICP-MS before and after 30K CV cycles of commercial Pt/C, ordered PtFe alloy (PtFe/C@NC-3-800) and

disordered PtFe alloy (PtFe/C@NC-9-800) catalysts.