

Supplementary Information

Photodeposition of Pd onto TiO₂ nanowires for aqueous-phase selective hydrogenation of phenolics to cyclohexanones

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Table S1 Physiochemical properties of TiO₂ samples.

Sample	S_{BET} (m ² /g)	$V_{\text{pore}}^{\text{a}}$ (cm ³ /g)	$V_{\text{mes}}^{\text{b}}$ (cm ³ /g)	$V_{\text{mes}}/V_{\text{tol}}$ (%)	D_{pore} (nm)
TiO ₂ (P25)	58	0.28	0.26	95.1%	15.6
TNWs-450	220	0.64	0.57	90.4%	9.9
TNWs-550	102	0.85	0.82	96.5%	29.8
TNWs-650	61	0.48	0.47	96.4%	39.3
TNWs-750	40	0.20	0.19	94.4%	32.0

^a Total pore volume of all pores; ^b total volume of mesopores pores.

Table S2 Catalytic performance for cyclohexanone and phenol hydrogenation of Pd/TNWs and Pd/CeO₂.

Entry	Catalyst	Reactant	Products / %		
			Phenol	C=O	C—O
1	Pd/TNWs	Cyclohexanone	—	83.5	16.5
2	Pd/CeO ₂	Cyclohexanone	—	11.5	88.5
3	Pd/TNWs	90% Cyclohexanone+10% Phenol	5.7	94	0.3
4	Pd/CeO ₂	90% Cyclohexanone+10% Phenol	8.2	83.4	8.4

Typical conditions: 25 mg catalyst, 5.0 mL solvent with 1.0 mmol reactant, 50 °C, 5 bar, stirring at 500 rpm for 2 h.

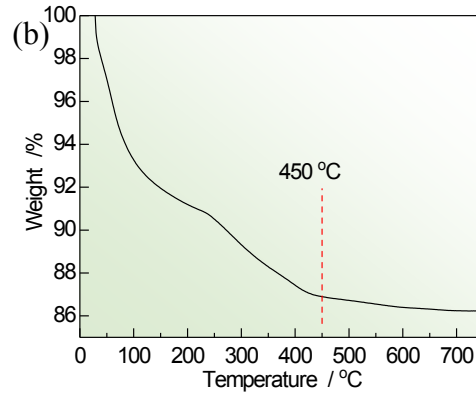
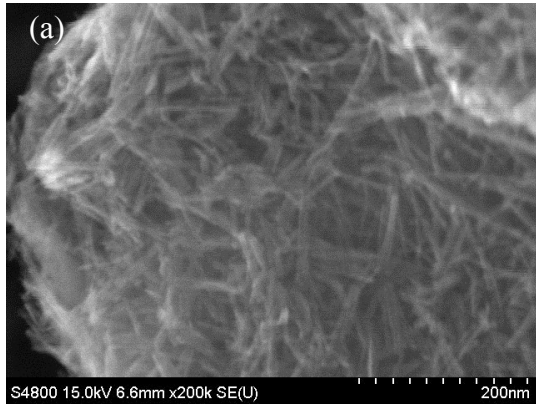


Fig. S1 (a) SEM image and (b) TGA pattern of titanic acid nanowire synthesized by hydrothermal-ion exchange method.

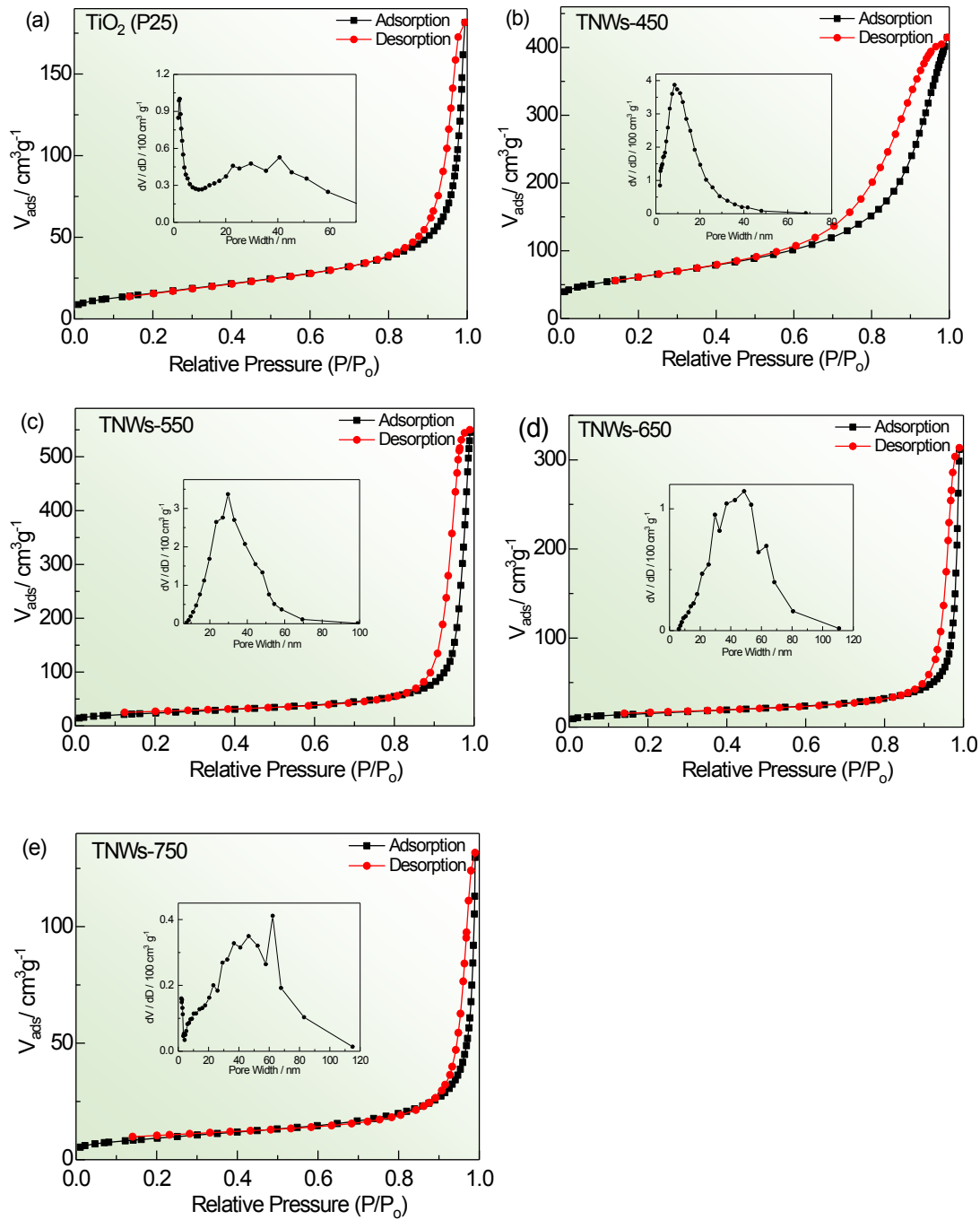


Fig. S2 N_2 adsorption/desorption isotherms and BJH pore size distributions (inserts) of (a) P25; (b) TNWs-450 °C; (c) TNWs-550 °C, (d) TNWs-650 °C, (e) TNWs-750 °C.

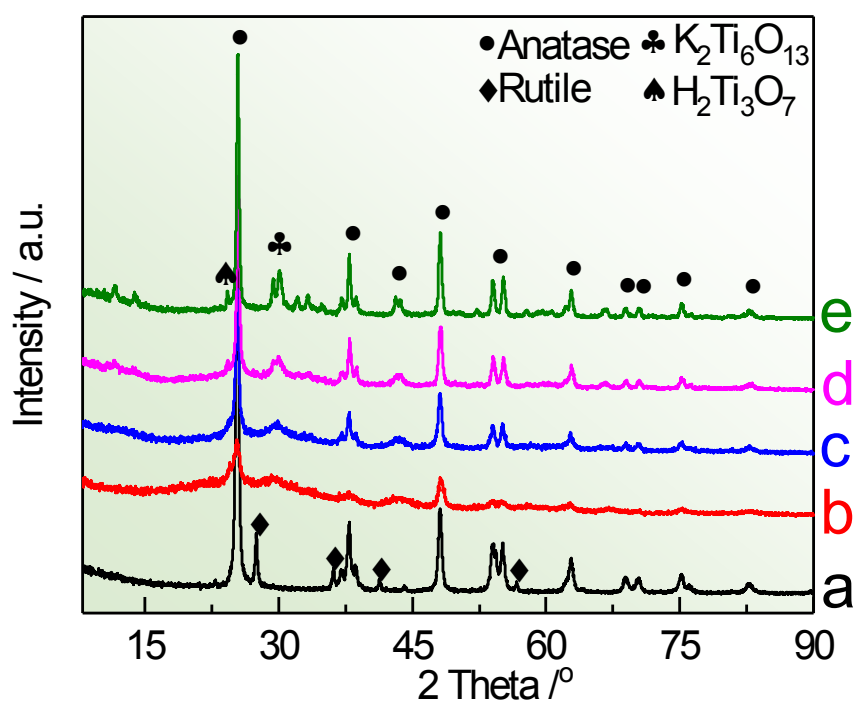


Fig. S3 XRD patterns of (a) P25, (b) TNWs-450 °C, (c) TNWs-550 °C, (d) TNWs-650 °C, and (e) TNWs-750 °C.

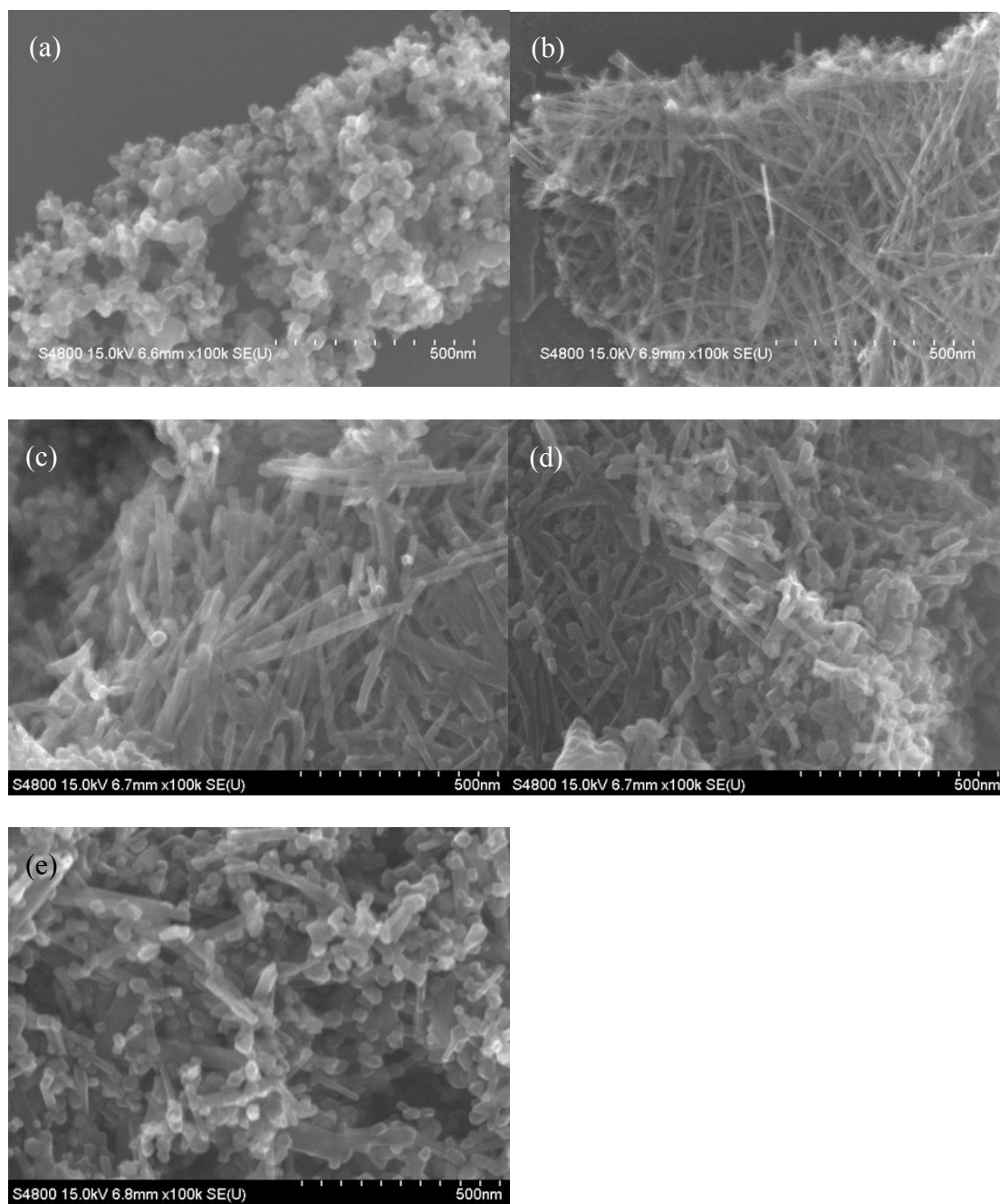


Fig. S4 SEM images of (a) P25; (b) TNWs-450 °C; (c) TNWs-550 °C, (d) TNWs-650 °C, and (e) TNWs-750 °C.

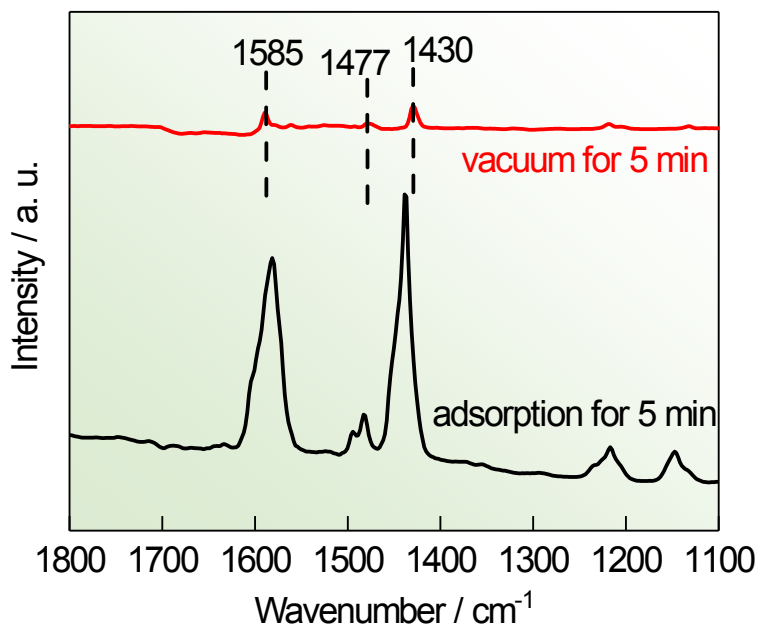


Fig. S5 In-situ FTIR spectra of TiO₂ (P25) for pyridine adsorption: adsorption for 5 min (black line); evacuation of excess probe molecules under vacuum for 5 min (red line).

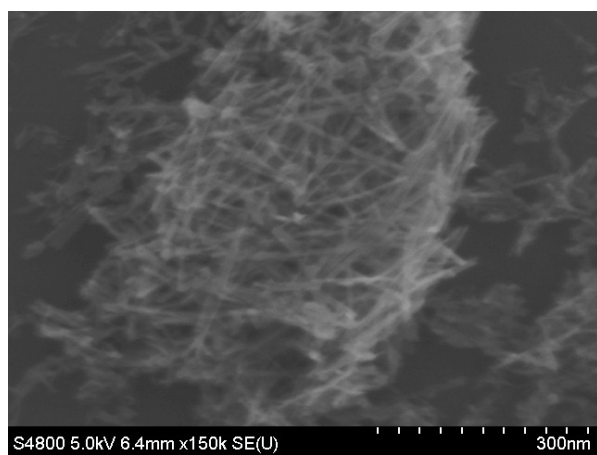


Fig. S6 SEM images of Pd/TNWs.

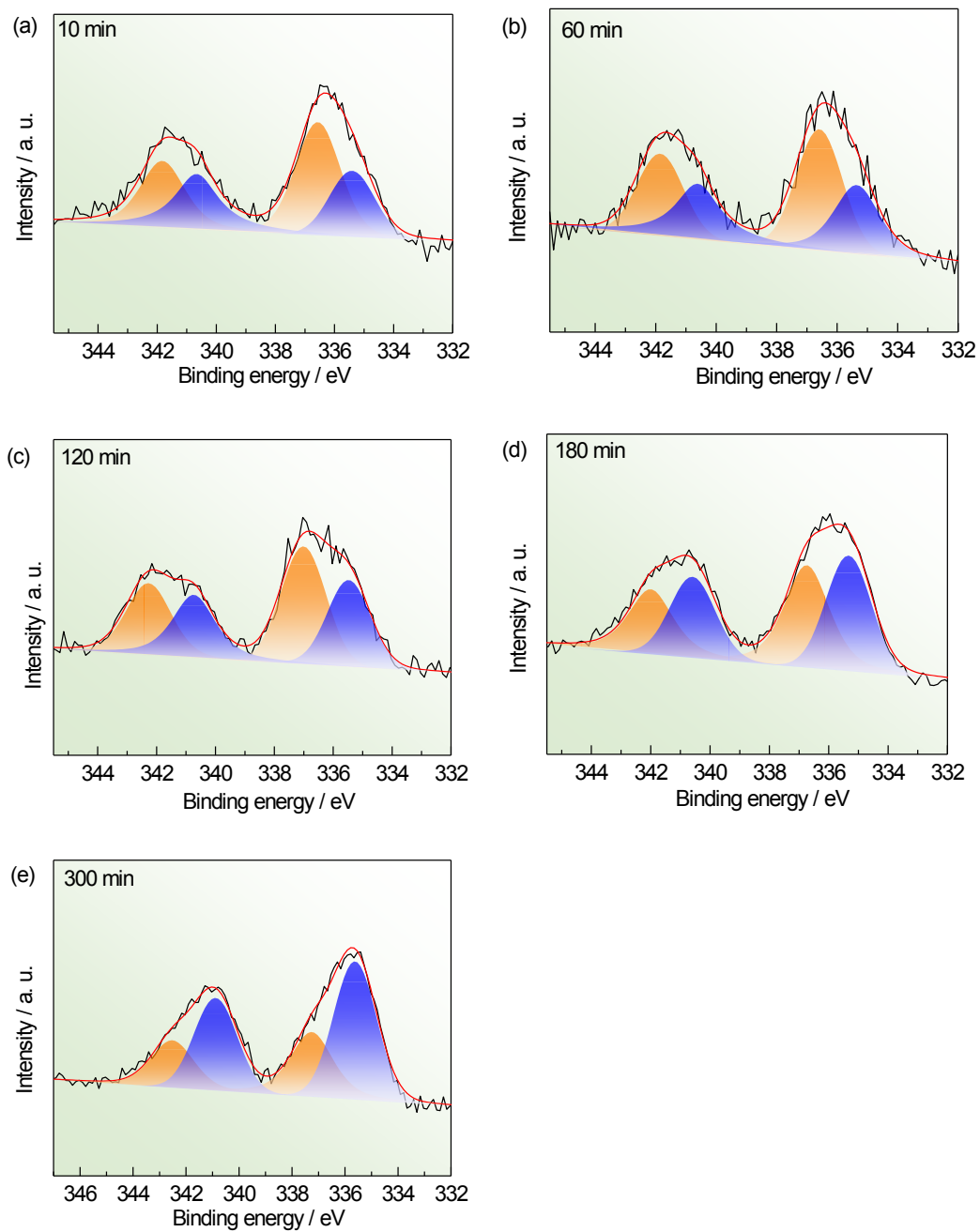


Fig. S7 XPS analyses of Pd/TNWs by photoreduction for different time: (a) 10 min, (b) 60 min, (c) 120 min, (d) 180 min, and (e) 300 min.

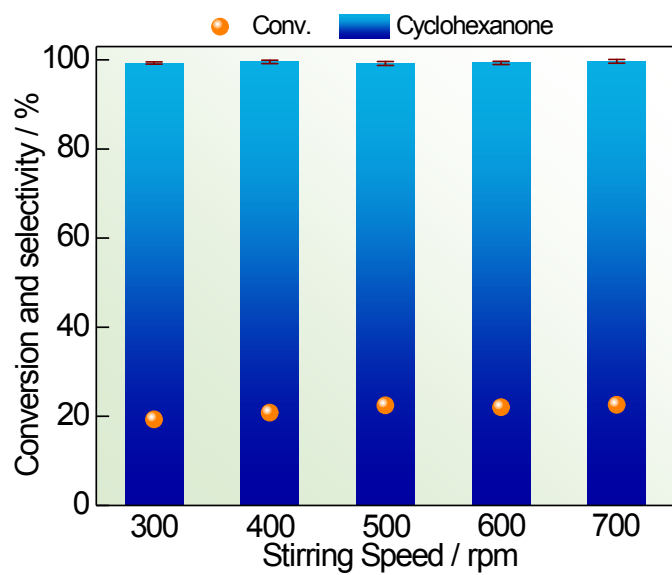


Fig. S8 Effect of stirring speed on the phenol conversion over Pd/TNWs.

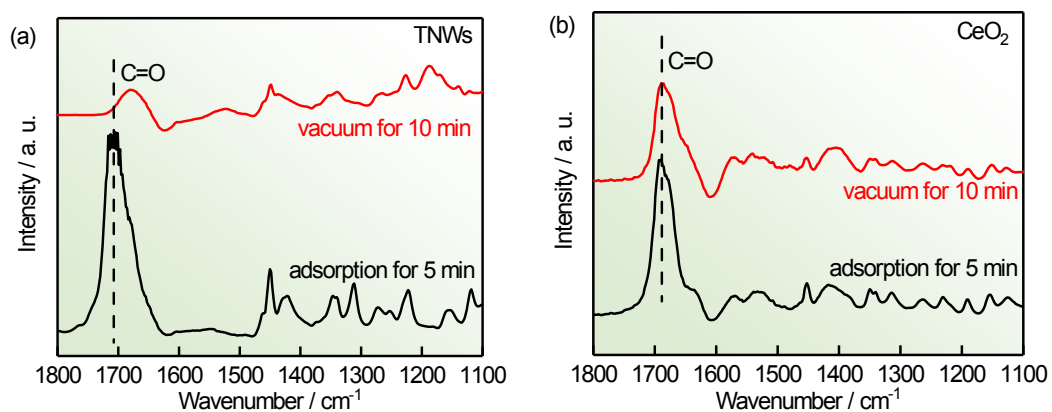


Fig. S9 In-situ FTIR spectra of TNWs (a) and CeO₂ (b) for cyclohexanone absorption: adsorption for 5 min (black line); evacuation of excess probe molecules under vacuum for 10 min (red line).

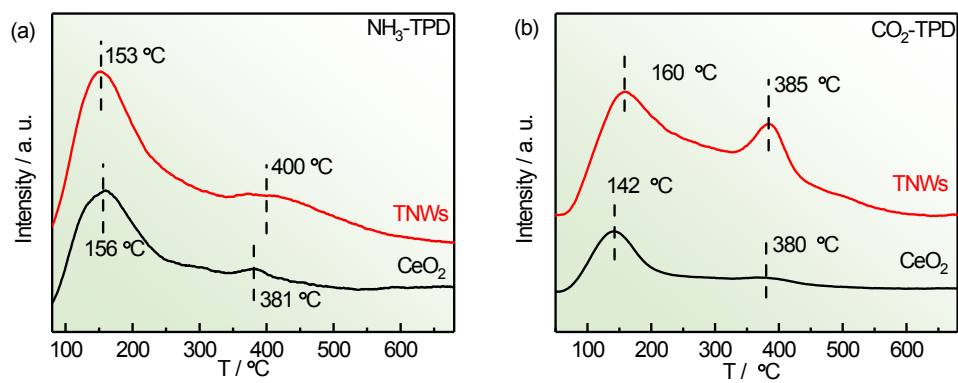


Fig. S10 (a) NH₃-TPD spectra and (b) CO₂-TPD of TNWs and CeO₂ samples.