

Supporting information

Well-defined CoSe₂@MoSe₂ hollow heterostructured nanocubes with enhanced dissociation kinetics for overall water splitting

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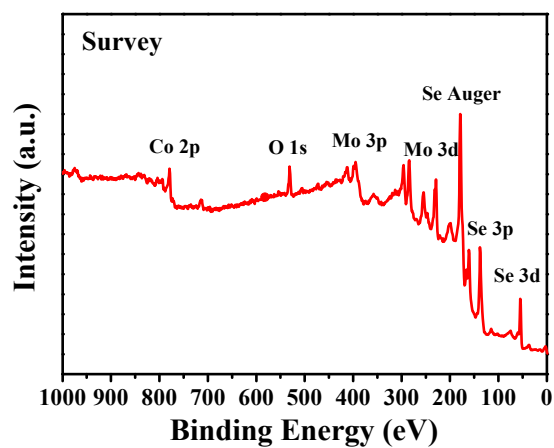


Fig. S1. The survey XPS spectrum of the as-synthesized $\text{CoSe}_2@\text{MoSe}_2$ heterostructures.

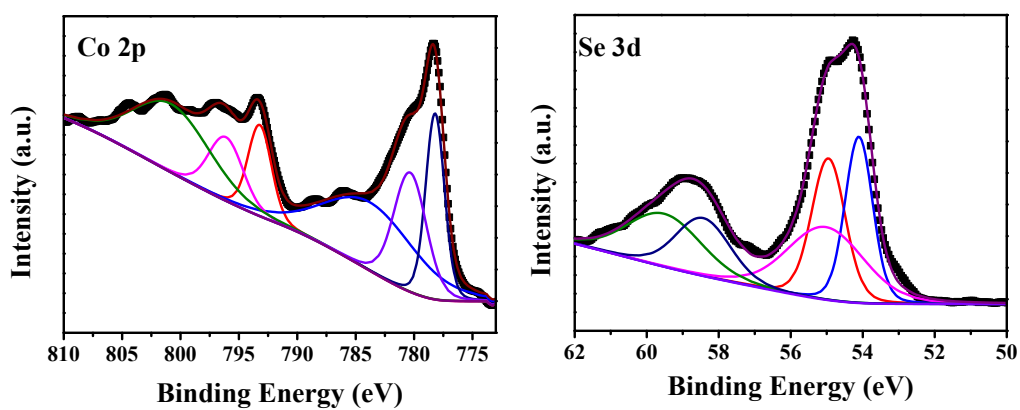


Fig. S2 XPS spectrum of the as-obtained CoSe_2 hollow nanocubes.

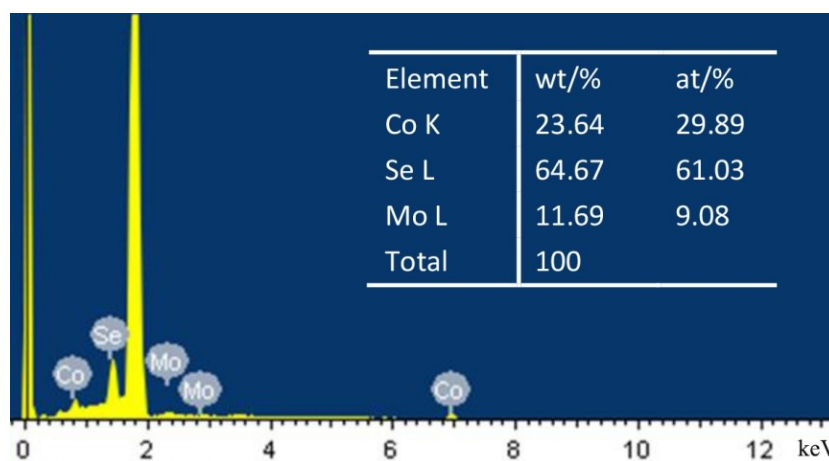


Fig. S3. The energy dispersive X-ray spectroscopy (EDX) of the as-synthesized $\text{CoSe}_2@\text{MoSe}_2$ heterostructures.

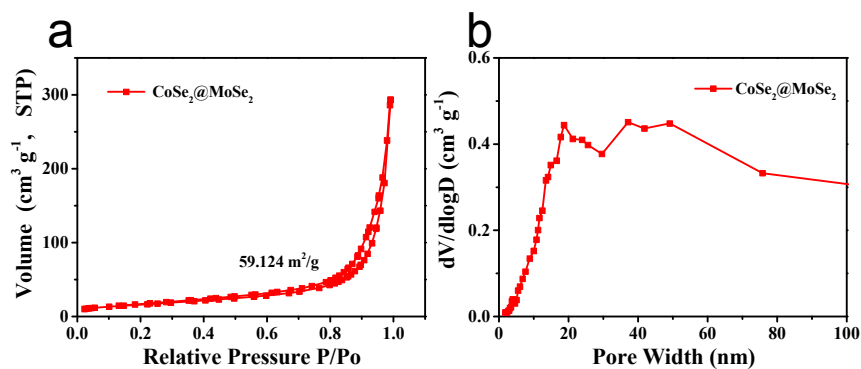


Fig. S4 (a) Nitrogen adsorption/desorption isotherms of CoSe₂@MoSe₂. (b) Pore size distribution plots of CoSe₂@MoSe₂.

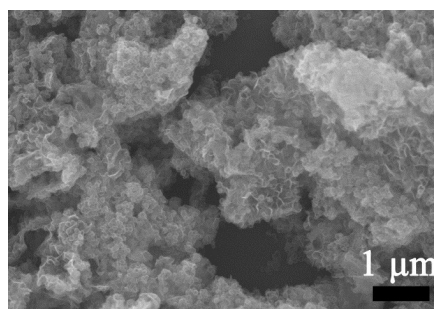


Fig S5. SEM image of the as-synthesized MoSe₂ microspheres

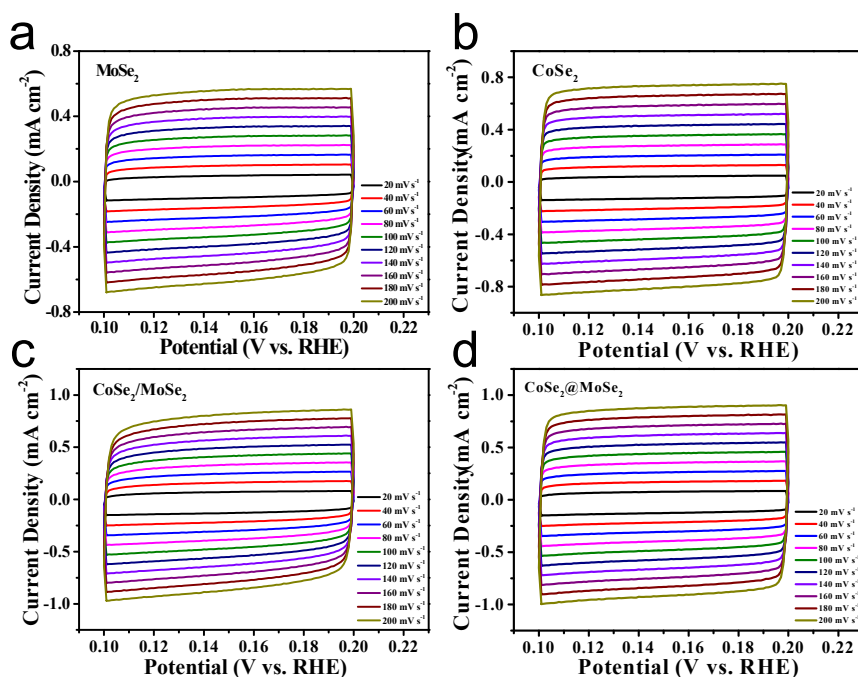


Fig S6. Cyclic voltammograms of the as-synthesized (a) CoSe₂, (b) MoSe₂, (c) CoSe₂/MoSe₂ and (d) CoSe₂@MoSe₂ heterostructures in 0.1~0.2 V vs. RHE at different scan rates from 20 mV s⁻¹ to 200 mV s⁻¹ in 0.5 M H₂SO₄ by Ni foam.

Tab S1 Comparison of HER Performance of Metal Sulfide Catalysts under Alkaline Conditions

Catalysts	electrolyte	Electrode	η_{10} (mV)	Tafel slop (mV/dec)	Reference
CoSe ₂ @MoSe ₂	1.0 M KOH	NF	183	87.69	This work
CoS ₂ @N-GN	1.0 M KOH	GCE	204	108	1
Ni ₃ S ₂ /NiS	1.0 M KOH	GCE	180	83	2
CoNi ₂ S ₄	1.0 M KOH	NF	54	129	3
Co _{0.85} Se@NC	1.0 M KOH	NF	230	125	4
NiCo ₂ Se ₄	1.0 M KOH	NF	150	122	5
Ni ₃ S ₂	1.0 M KOH	NF	223	/	6
Co ₃ S ₄	1.0 M KOH	NF	199	91	7
NiS ₂	1.0 M KOH	GS	190	80	8

GCE: Glassy carbon electrode; GS: Graphite substrate; NF: Ni foam

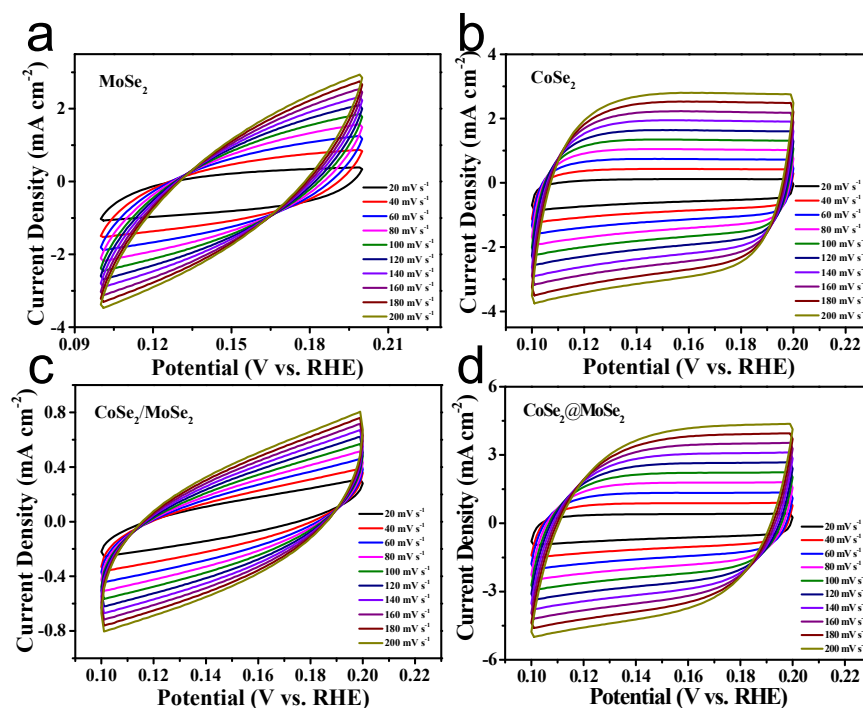


Fig S7. Cyclic voltammograms of the as-synthesized (a) CoSe₂, (b) MoSe₂, (c) CoSe₂/MoSe₂ and (d) CoSe₂@MoSe₂ heterostructures in 0.1~0.2 V vs. RHE at different scan rates from 20 mV s⁻¹ to 200 mV s⁻¹ in 1.0 M KOH.

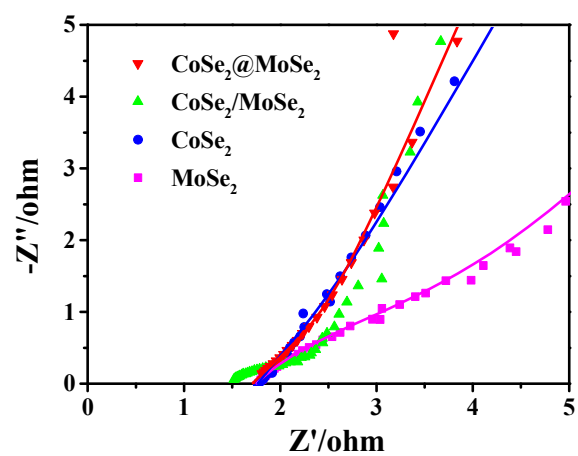


Fig S8. Nyquist plots and the corresponding simulated equivalent circuit diagram of the samples.

References

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