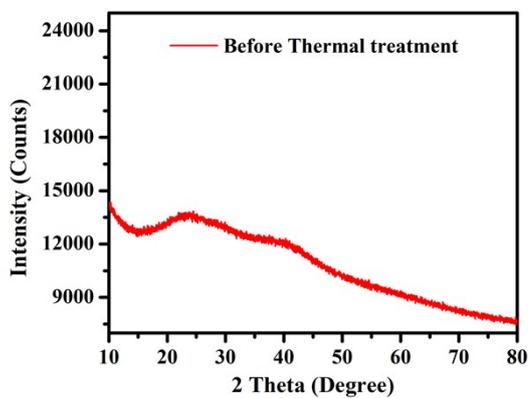
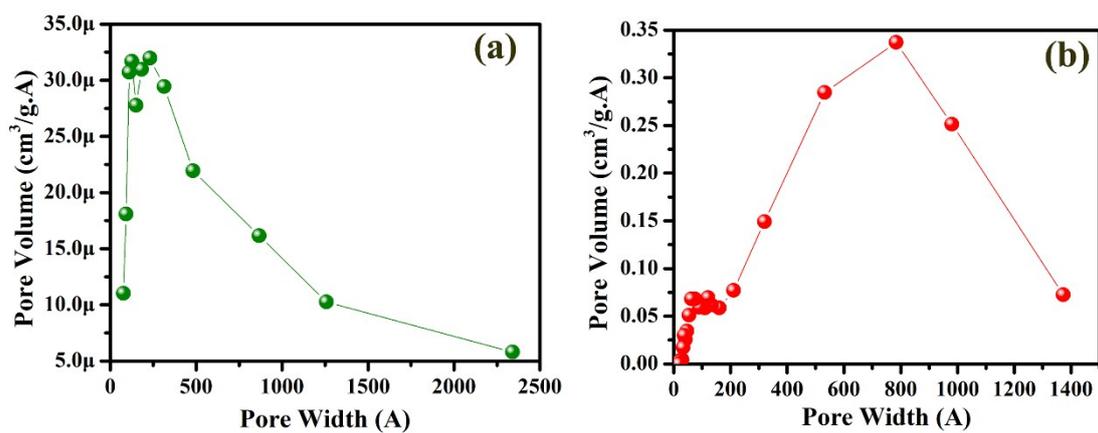


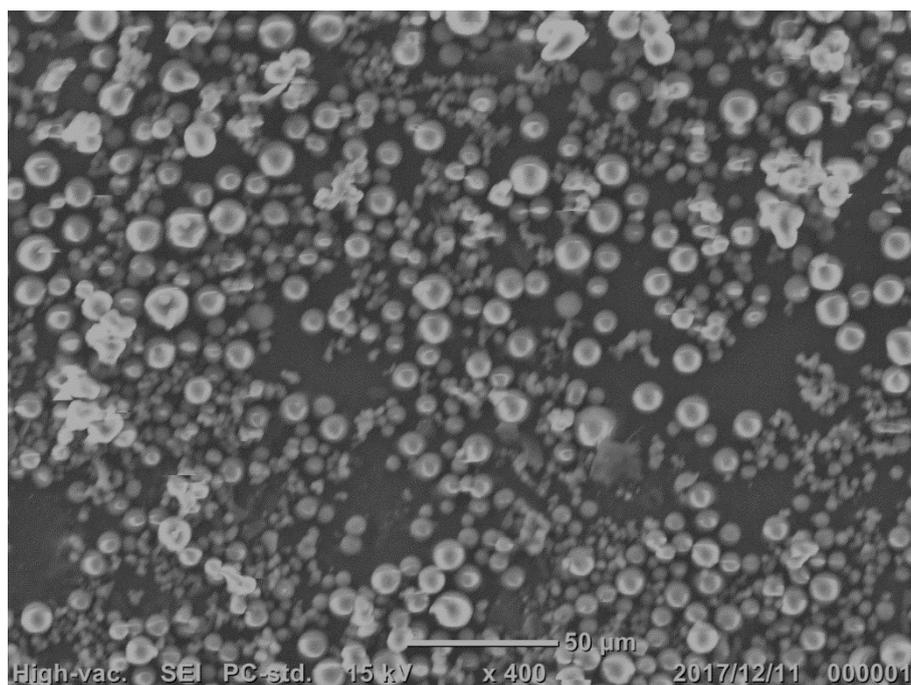
## Supporting Information



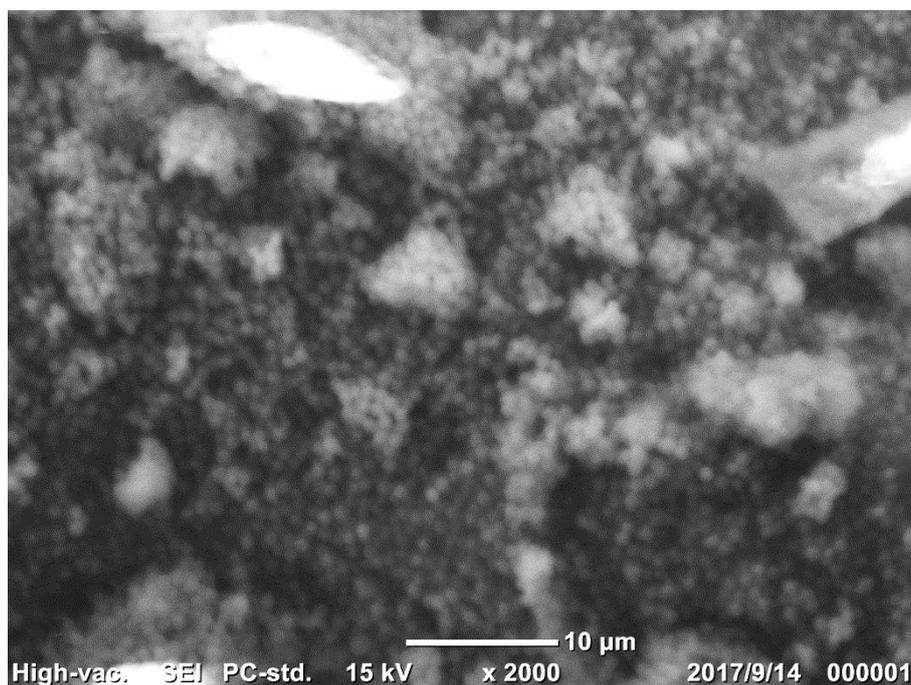
**Figure S1** XRD profile of carbonaceous microspheres with uniformly embedded metal ions obtained after hydrothermal treatment (NMC-CS).



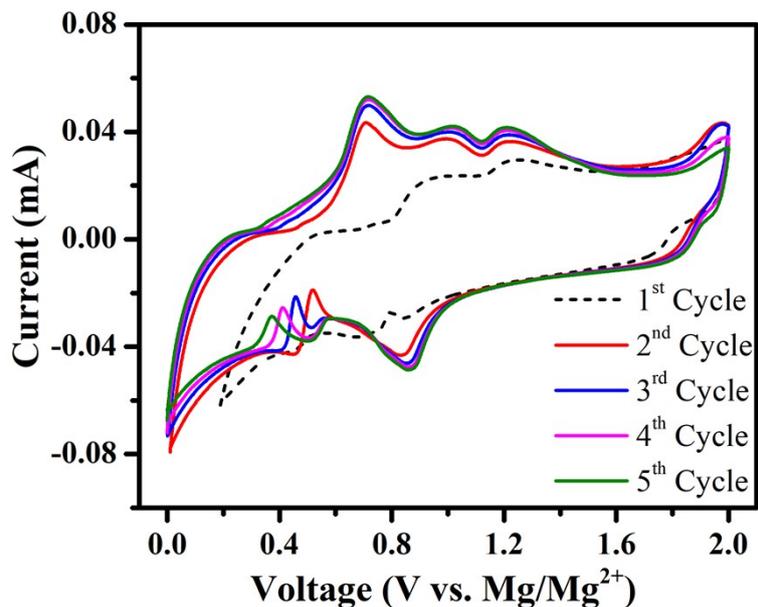
**Figure S2** Pore-size distribution curve of (a) NMC-CS, and (b) NMCO-HS.



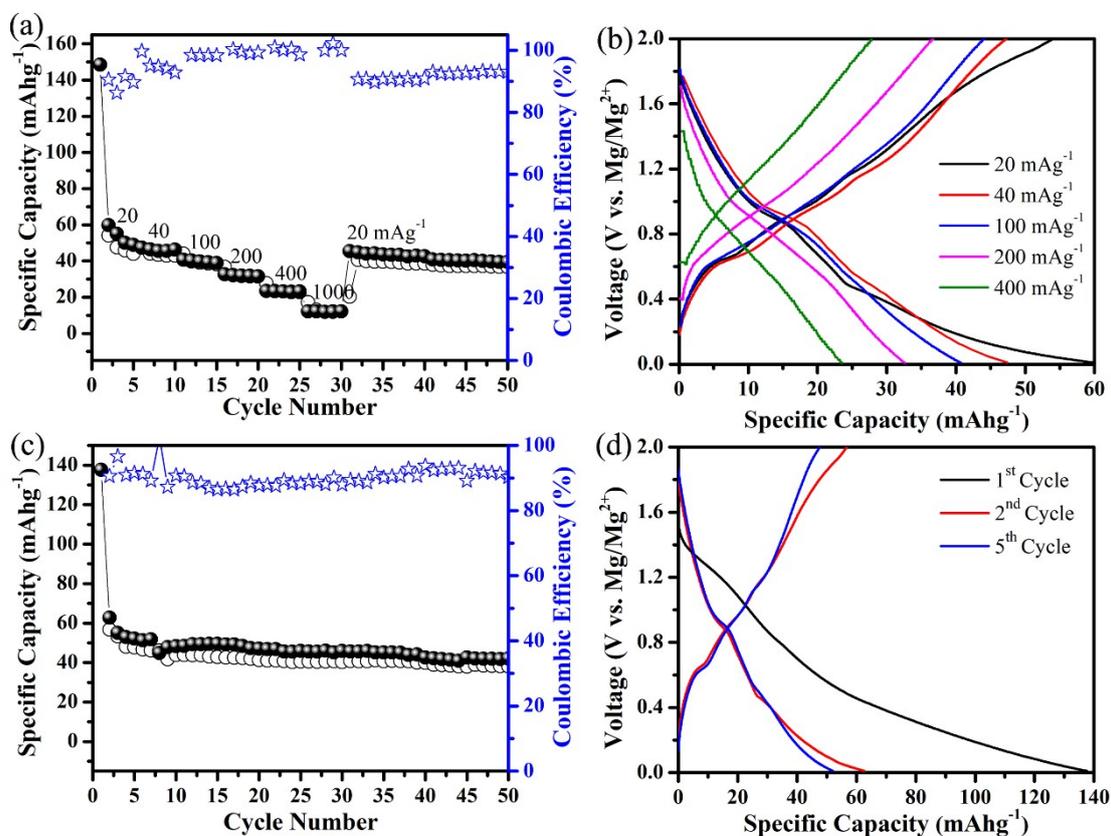
**Figure S3** SEM image of carbonaceous microspheres with uniformly embedded metal ions obtained after hydrothermal treatment.



**Figure S4** SEM image of pure carbon spheres.

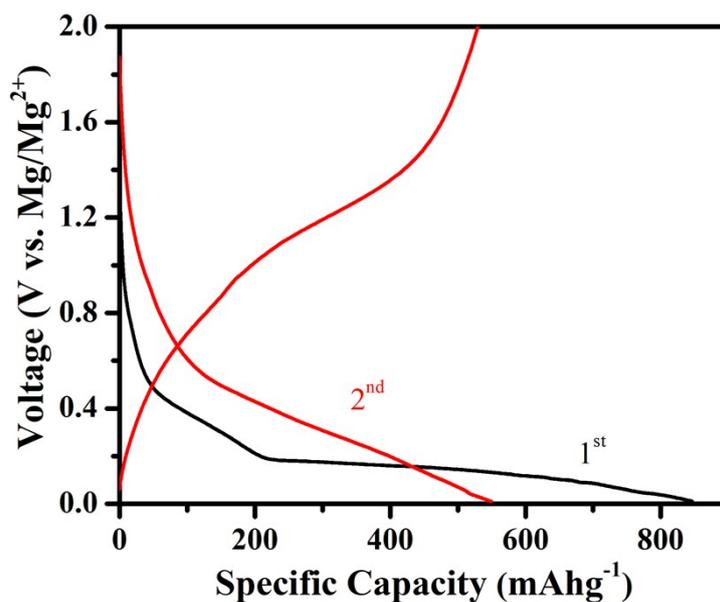


**Figure S5** Cyclic voltammety of MNC-O HS500 based MIBs at the scan rate of 0.5 mVs<sup>-1</sup>.

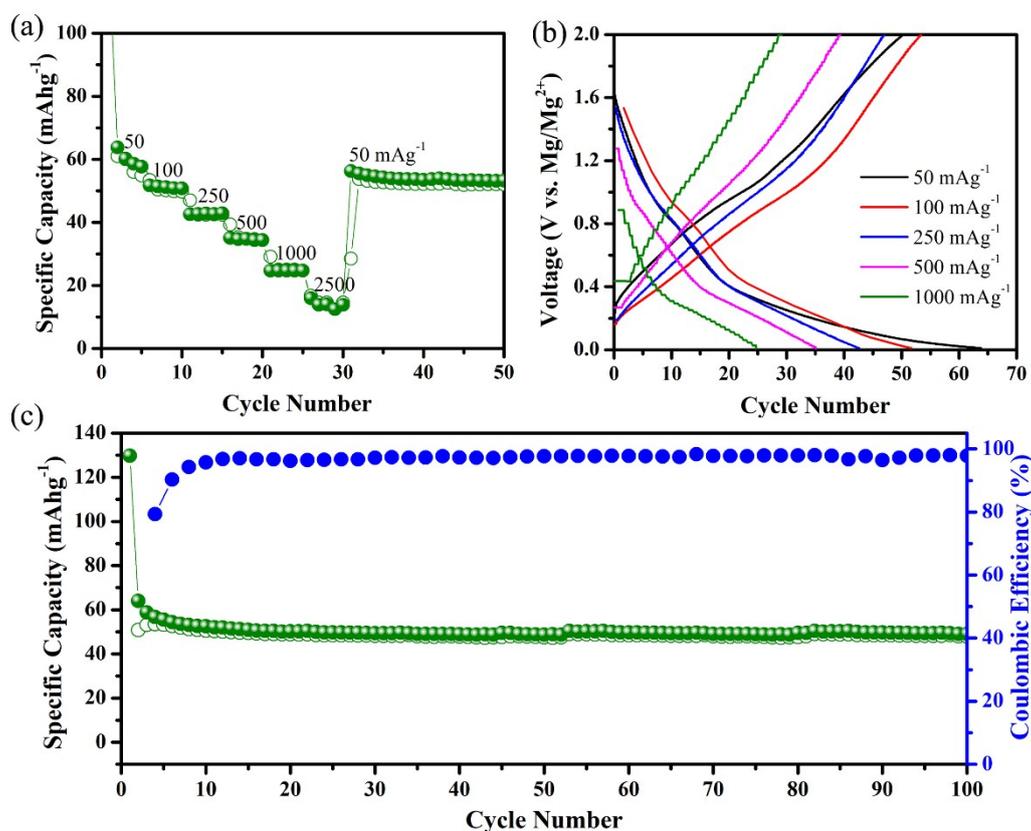


**Figure S6** Electrochemical characterizations of MNC-O HS500 based MIBs: (a) Rate performance, (b) Charging-discharging profiles at different current rates (20-

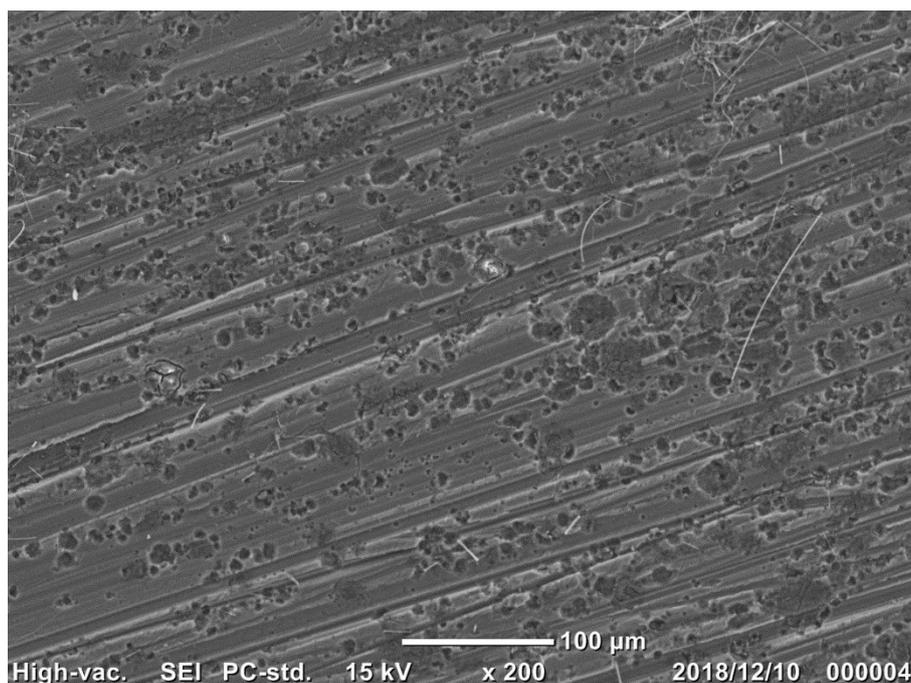
1000mA $g^{-1}$ ), (c) Cycle stability, and (d) 1<sup>st</sup>, 2<sup>nd</sup>, and 5<sup>th</sup> charge-discharge profiles at 20mA $g^{-1}$ .



**Figure S7** 1<sup>st</sup> and 2<sup>nd</sup> charge-discharge profiles of MNC-O HS500 based MLIBs at the current density of 50mA $g^{-1}$ .



**Figure S8** Electrochemical characterizations of MNC-O HS000 based MLIBs: (a) Rate performance, (b) Charging-discharging profiles at different current rates (50-1000mA $g^{-1}$ ), (c) Cycle stability at 50mA $g^{-1}$ .



**Figure S9** Surface morphology of magnesium anode after 100 cycles in 0.4APC-1.0LiCl hybrid electrolytes.

**Table S1** Inductively Coupled Plasma (ICP) mass spectrometry results

Electrodes	Lithium ( $\mu\text{g/L}$ )	Magnesium ( $\mu\text{g/L}$ )
Pristine	008.675	2745.057
Discharged (0.05V)	912.561	6734.681
Charged (2.0V)	203.173	4245.116

#### Calculation of Energy and Power Densities:

The power density ( $P$ ) was calculated using Equation 1 given below.

$$P = U * I \quad (1)$$

Where  $U$  is average working voltage of battery and  $I$  is applied current. Whereas, energy density  $E$  was calculated using average working voltage ( $U$ ), specific capacity ( $C$ ) based on the total mass of the active materials ( $m$ ) using Equation 2 given below.

$$E = U * C / m \quad (2)$$