

Supporting Information

Metal-organic framework derived amorphous VO_x coated Fe₃O₄/C hierarchical nanospindle as anode material for superior lithium-ion batteries

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The supporting information includes 13 figures, 2 tables, and 17 references.

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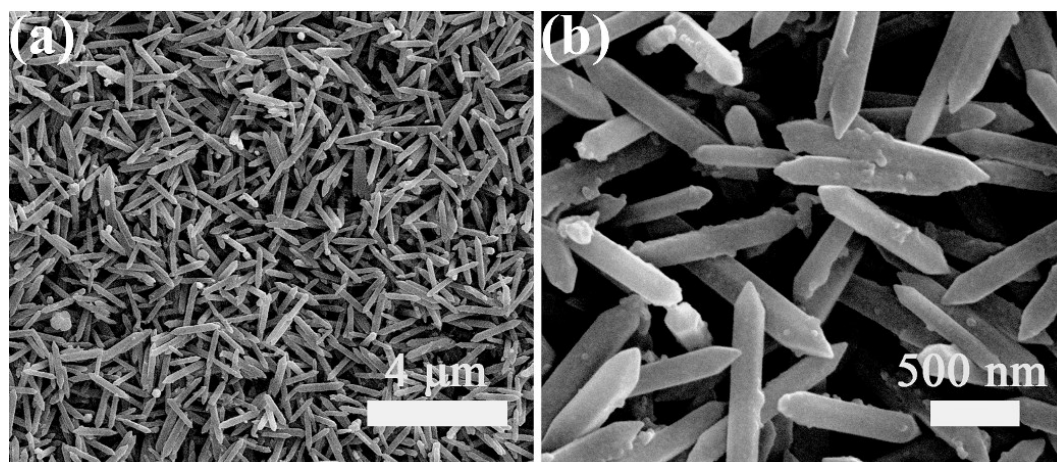


Fig. S1 SEM images of Fe-MOF nanospindles at (a) low- and (b) high-magnification.

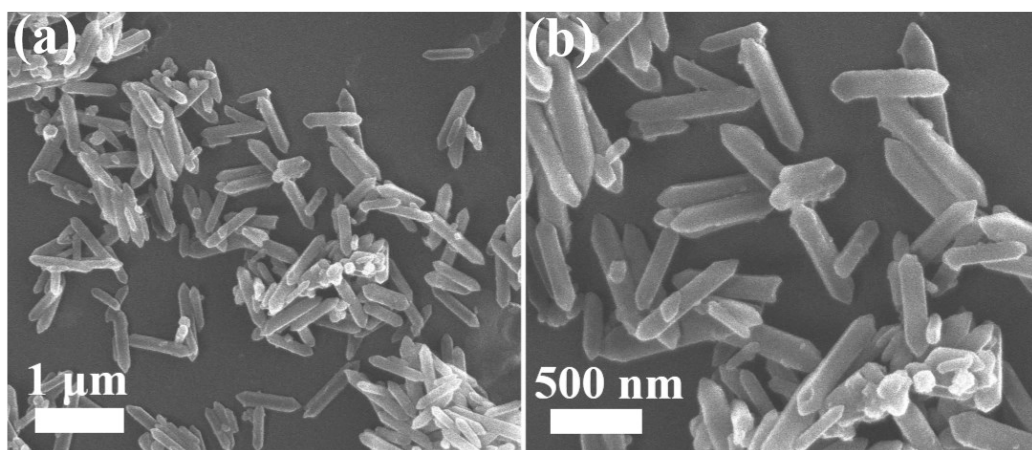


Fig. S2 SEM images of Fe-MOF@V₂O₅ nanospindles at (a) low- and (b) high-magnification.

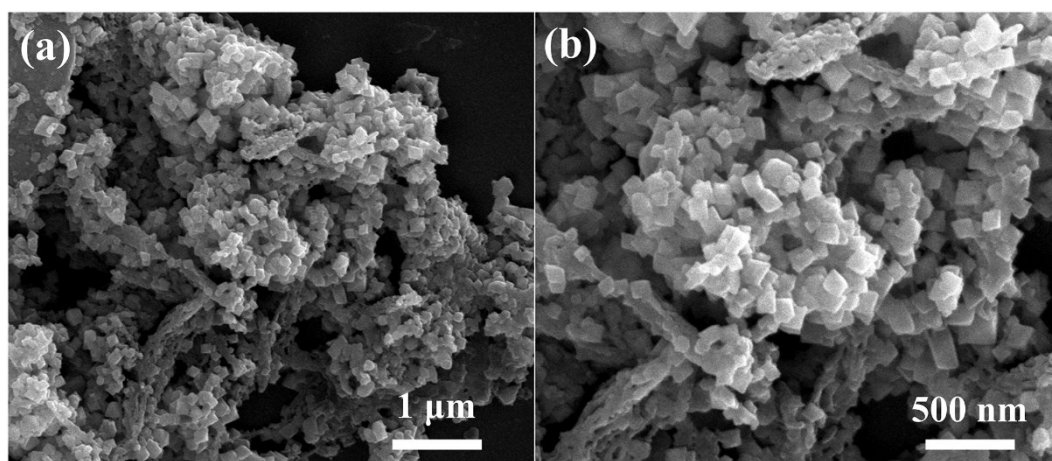


Fig. S3 SEM images of the Fe₃O₄/C sample at (a) low- and (b) high-magnification.

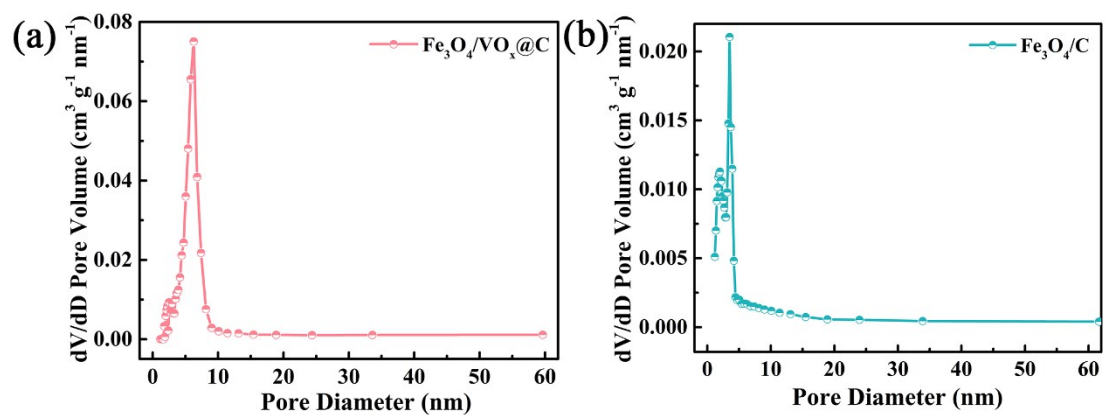


Fig. S4 The pore size distribution of (a) $\text{Fe}_3\text{O}_4/\text{C}@\text{VO}_x$ and (b) $\text{Fe}_3\text{O}_4/\text{C}$ samples.

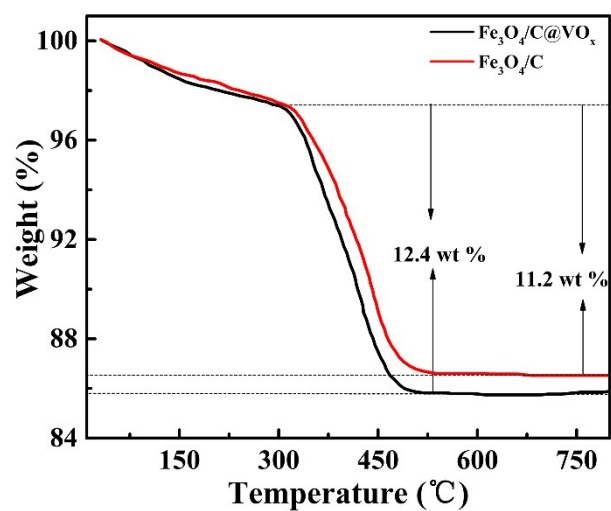


Fig. S5 Thermo gravimetric analysis (TGA) curves of $\text{Fe}_3\text{O}_4/\text{C}@/\text{VO}_x$ and $\text{Fe}_3\text{O}_4/\text{C}$ samples at a temperature ramp of $10\text{ }^\circ\text{C min}^{-1}$ in air.

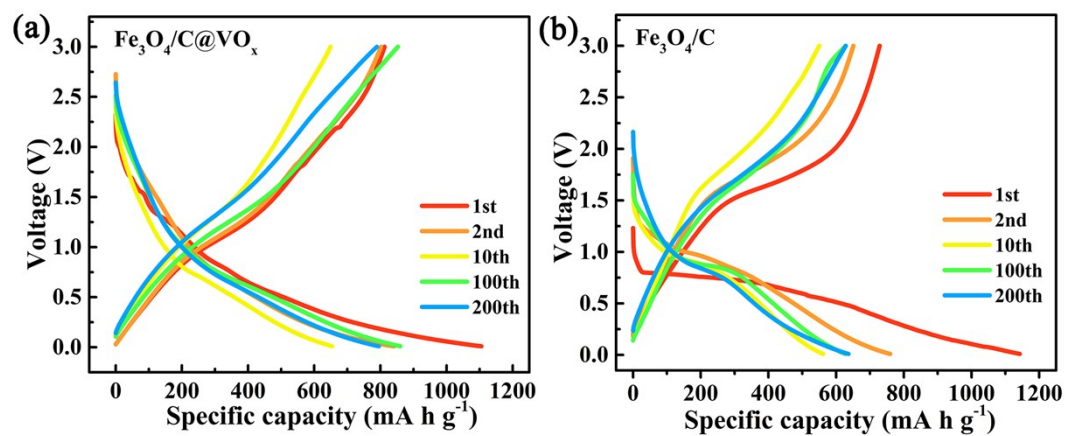


Fig. S6 Charge-discharge curves of (a) $\text{Fe}_3\text{O}_4/\text{VO}_x@\text{C}$ and (b) $\text{Fe}_3\text{O}_4/\text{C}$ at the current density of 1000 mA g^{-1} .

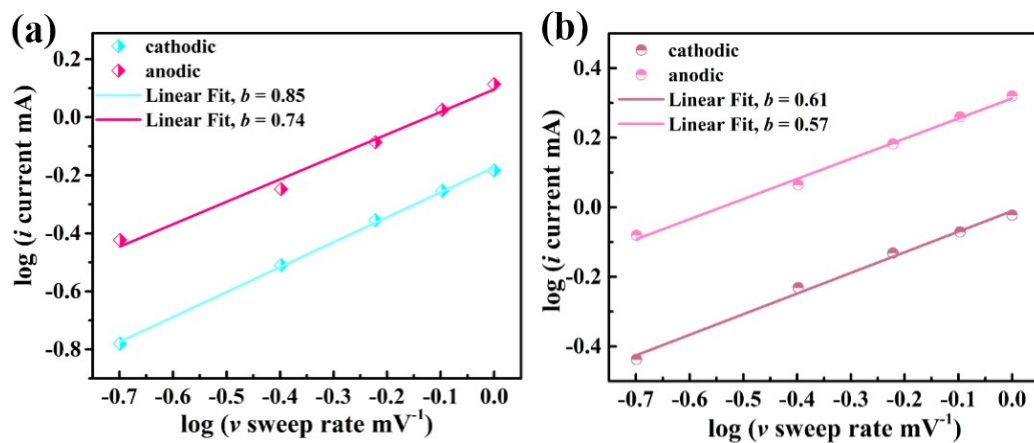


Fig. S7 b -values evaluation of (a) $\text{Fe}_3\text{O}_4/\text{C}@\text{VO}_x$ and (b) $\text{Fe}_3\text{O}_4/\text{C}$ using the relationship between peak current and scan rate.

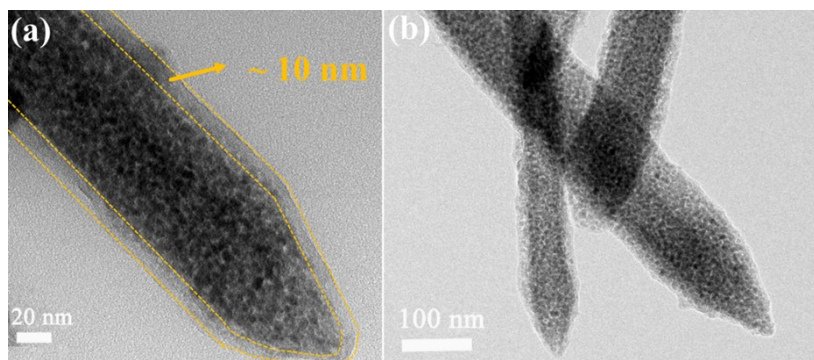


Fig. S8 TEM images of Fe₃O₄/C@VO_x-1 (the thickness of the VO_x layer is ~ 10 nm)

at (a) high- and (b) low- magnification.

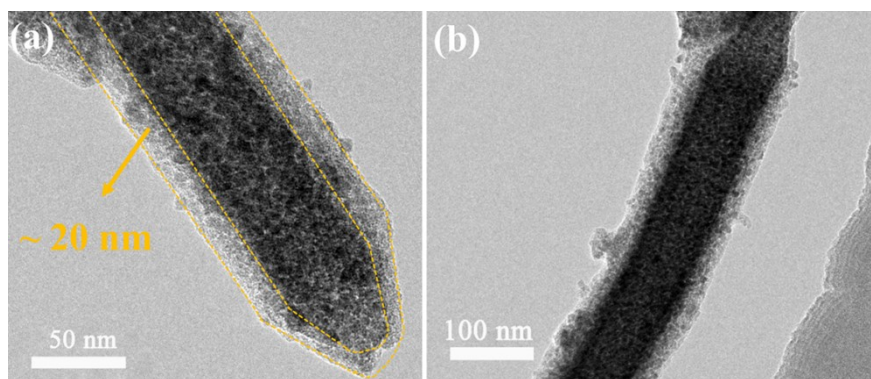


Fig. S9 TEM images of Fe₃O₄/C@VO_x-2 (the thickness of the VO_x layer is ~ 20 nm)

at (a) high- and (b) low- magnification.

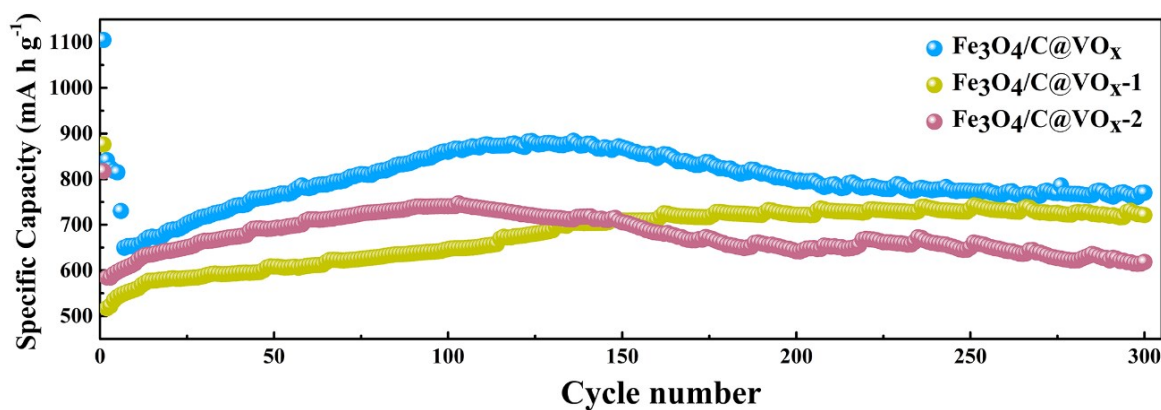


Fig. S10 Charge/discharge capacities of Fe₃O₄/C@VO_x sample with different thickness of the VO_x layer is ~ 5 nm (Fe₃O₄/C@VO_x), ~ 10 nm (Fe₃O₄/C@VO_x-1) and ~ 20 nm (Fe₃O₄/C@VO_x-2), respectively.

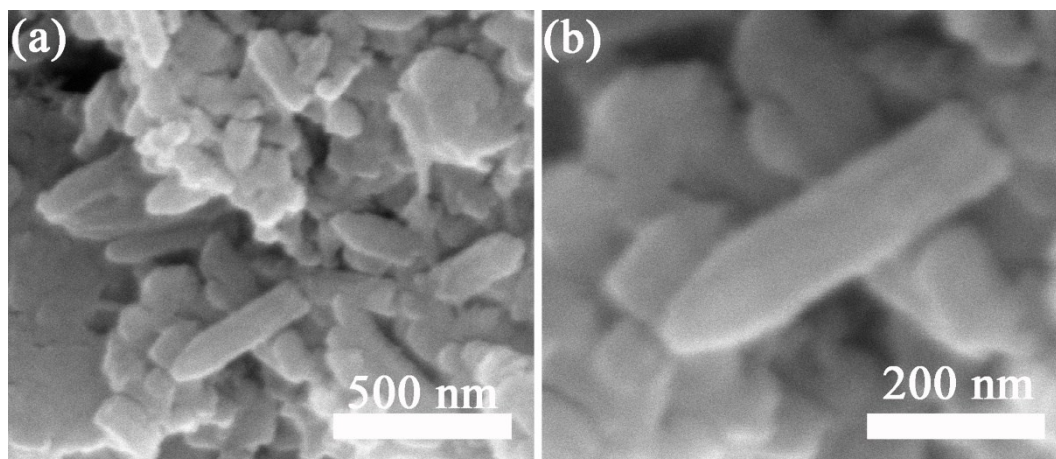


Fig. S11 SEM images of the $\text{Fe}_3\text{O}_4/\text{C}@\text{VO}_x$ electrode after 500 cycles at 1 A g^{-1} current density in (a) low- and (b) high-magnification.

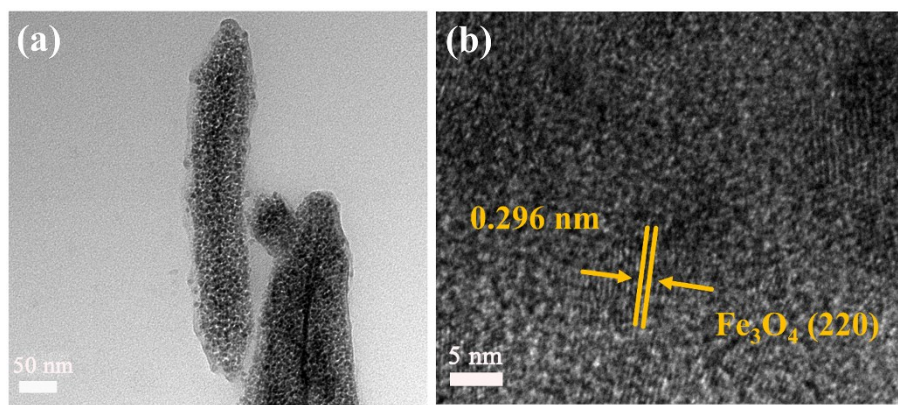


Fig. S12 The $\text{Fe}_3\text{O}_4/\text{C}@\text{VO}_x$ electrode after 500 cycles at 1000 mA g^{-1} current density in (a) TEM and (b) HRTEM images.

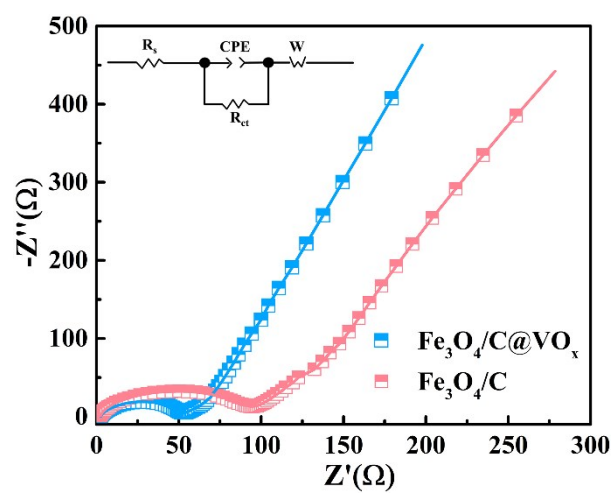


Fig. S13 Nyquist plots of fresh cells (inset: equivalent circuit for plot fitting).

Table S1 Comparison of the LIBs performance of Fe₃O₄/C@VO_x electrode material and recently reported Fe₃O₄ and Fe₃O₄/C based materials in the literature

Electrode material	Current density (mA g ⁻¹)	Reversible capacity (mA h g ⁻¹)	Cycle number (Times)	Ref.
Hollow Fe ₃ O ₄ /C spheres	200	984	70	[1]
Yolk–Shell Fe ₃ O ₄ @C Composite	1000	750	70	[2]
FeO _x @C yolk-shelled structure	200	790	100	[3]
Graphene- Encapsulated Fe ₃ O ₄ nanoparticles	100	650	100	[4]
Carbon-coated Fe ₃ O ₄ nanospindles	500	530	80	[5]
Fe ₃ O ₄ /C nanotubes	150	600	100	[6]
Fe ₃ O ₄ /C microrods	200	650	100	[7]
Porous Fe ₃ O ₄ /C Microbelts	100	710	50	[8]
3D Hierarchical Fe ₃ O ₄ /Graphene Composites	92.5	609	50	[9]
Mesoporous Fe ₃ O ₄ @C Microcapsules	100	928	50	[10]
Fe ₃ O ₄ /rGO nanorod	500	890	100	[11]
Fe ₃ O ₄ /C Filament Network	100	1278	100	[12]
Hierarchical 3D Fe ₃ O ₄ @porous carbon matrix/graphene	200	1077	100	[13]
Fe ₃ O ₄ @C yolk-shell nanorods	500	954	200	[14]
Fe ₃ O ₄ @C nanosheet	200	1232	120	[15]
Fe ₃ O ₄ /C nanofibers	500	761	300	[16]
Fe ₃ O ₄ microflowers	100	1000	50	[17]
Fe₃O₄/C@VO_x nanospindles	200	1336	300	Our work

Table S2 Electrochemical Impedance Parameters of the Fe₃O₄/C@VO_x and Fe₃O₄/C

Samples	R _s (Ω)	R _{ct} (Ω)
Fe ₃ O ₄ /C@VO _x	1.96	49.85
Fe ₃ O ₄ /C	2.41	91.50

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