

Pd(II)/Zn Co-catalyzed Chemo-Selective Hydrogenation of α -Methylene- γ -Keto Carboxylic Acids

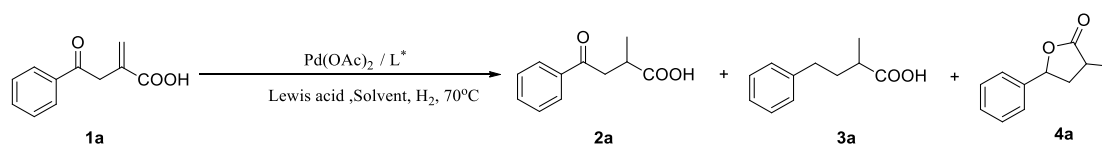
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Table of Contents

A: Table 1. Reaction condition optimizations	S2
B: Table 2. The asymmetric hydrogenation of α -methylene- γ -keto carboxylic acids using different ligands	S3
C: NMR spectra of substrates	S4
D: NMR spectra of products	S12
E: HPLC spectra of products	S51

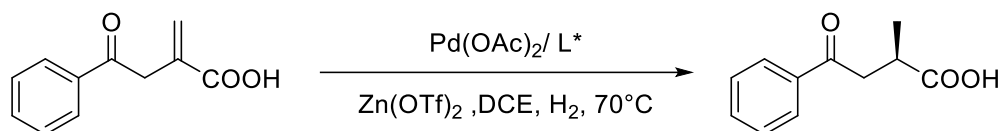
A: Table 1. Reaction condition optimizations

Entry	Ligand	Additive	Solvent	Time (h)	Yield ^b (2a)	Yield ^b (3a)	Yield ^{b,c} (4a)
1	(±)-Binap	Zn(OTf) ₂	DCE	8	96	-	trace
2	PPh ₃	Zn(OTf) ₂	DCE	24	70	-	-
3	DPPF	Zn(OTf) ₂	DCE	20	92	-	-
4	DPPP	Zn(OTf) ₂	DCE	14	97	-	-
5	DPPE	Zn(OTf) ₂	DCE	38	93	-	-
6	DPPB	Zn(OTf) ₂	DCE	15	99	-	-
7	Xantphos	Zn(OTf) ₂	DCE	36	64	-	31
8	o-Phenanthroline	Zn(OTf) ₂	DCE	24	72	-	-
9	PCy ₃	Zn(OTf) ₂	DCE	24	92	-	-
10	DPPB	Fe(OTf) ₃	DCE	7	87	-	10
11	DPPB	Fe(OTf) ₂	DCE	10	97	-	-
12	DPPB	Cu(OTf) ₂	DCE	10	50	18	22
13	DPPB	Cu(OTf)	DCE	10	76	-	20
14	DPPB	AgBF ₄	DCE	7	75	-	19
15	DPPB	AgOTf	DCE	6	82	6	5
16	DPPB	Al(OTf) ₃	DCE	7	48	19	26
17	DPPB	ZnI ₂	DCE	48	-	-	-
18	DPPB	-	DCE	60	20	-	-
19	DPPB	Zn(OTf) ₂	Toluene	10	95	-	Trace
20	DPPB	Zn(OTf) ₂	TFE	6	53	-	41
21	DPPB	Zn(OTf) ₂	1,4-dioxane	16	98	-	Trace
22	DPPB	Zn(OTf) ₂	CH ₃ CN	24	92	-	-
23 ^d	DPPB	Zn(OTf) ₂	DCE	15	99	-	-
24 ^e	DPPB	Zn(OTf) ₂	DCE	18	98	-	-
25 ^f	DPPB	Zn(OTf) ₂	DCE	45	98	-	-
26 ^g	BPA	Zn(OTf) ₂	DCE	10	-	98	-
27 ^g	BPA	-	DCE	15	-	97	-
28	(R,S)-PPF-P ^t Bu ₂	Zn(OTf) ₂	DCE	12	-	-	96
29	(R,S)-PPF-P ^t Bu ₂	-	DCE	24	68	-	-
30	HOAc	Zn(OTf) ₂	DCE	48	48	34	15
31	Diethyl phosphite	Zn(OTf) ₂	DCE	48	15	-	-

32	Triphenyl phosphite	Zn(OTf) ₂	DCE	48	19	trace	-
33 ^{g)}	H ₃ PO ₄	Zn(OTf) ₂	DCE	18	-	92	trace
34 ^{g)}	NaH ₂ PO ₄	Zn(OTf) ₂	DCE	24	-	91	-
35 ^{g)}	Na ₂ HPO ₄	Zn(OTf) ₂	DCE	22	-	90	trace
36 ^{g)}	K ₃ PO ₄	Zn(OTf) ₂	DCE	24	20	-	-

^aReaction conditions: Pd(OAc)₂ (0.01 mmol), Lewis acid (0.02 mmol), and ligand (0.012 mmol) in DCE (2 mL) was stirred at room temperature for 30 min under Argon. **1a** (0.2 mmol) were added, and the reaction mixture in 1 bar H₂ was stirred at 70 °C for indicated period of time. ^bIsolated yields. ^c*anti/syn* = 1.5/1.0. ^dThe reaction was performed at 60 °C. ^eThe reaction was performed at 50 °C. ^fThe reaction was performed at 40 °C. ^gBronsted acid

B: Table 2. The asymmetric hydrogenation of α -methylene- γ -keto carboxylic acids using different ligands

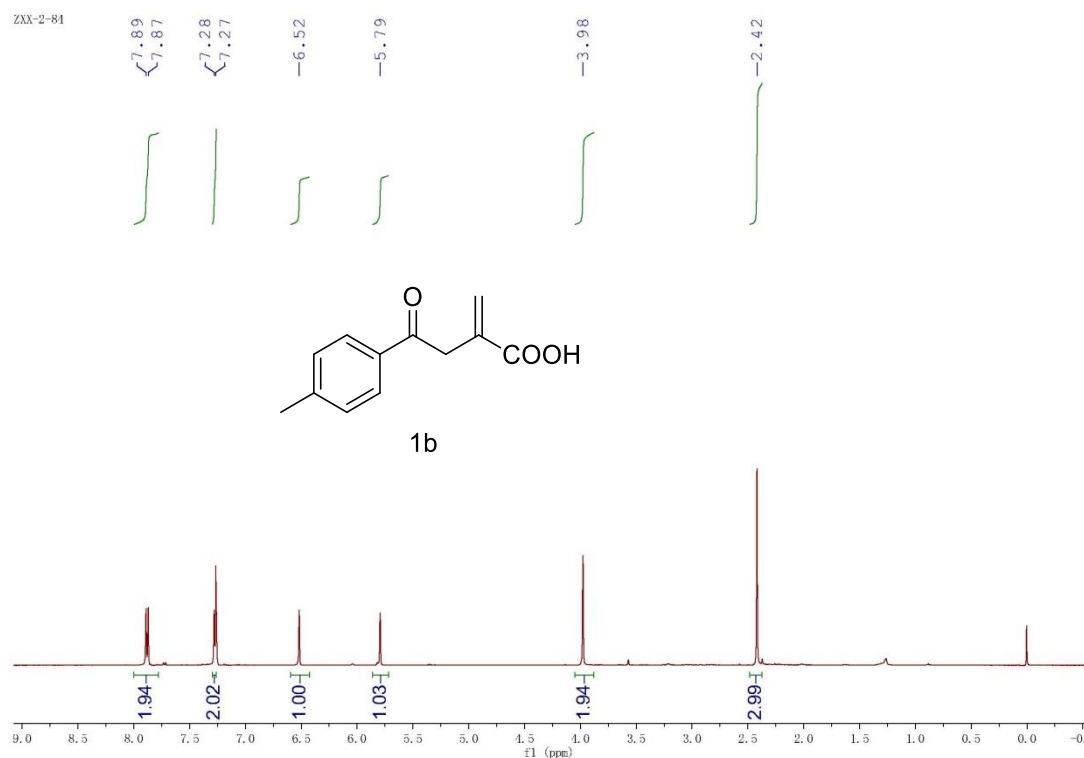
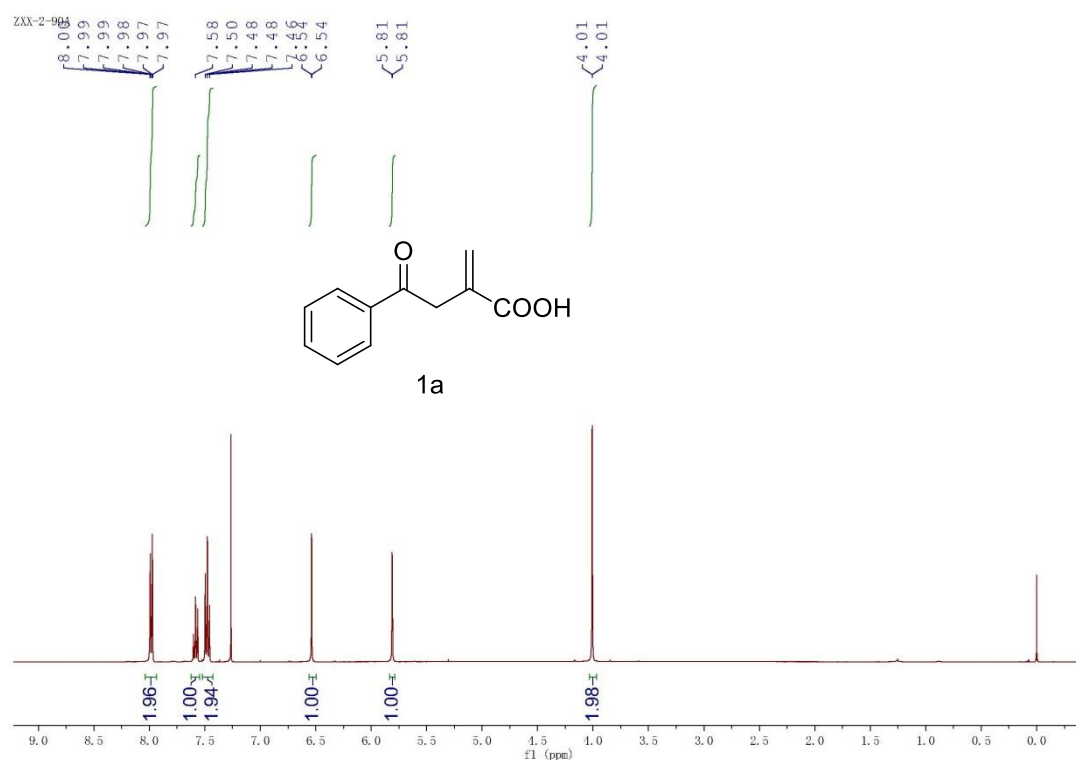


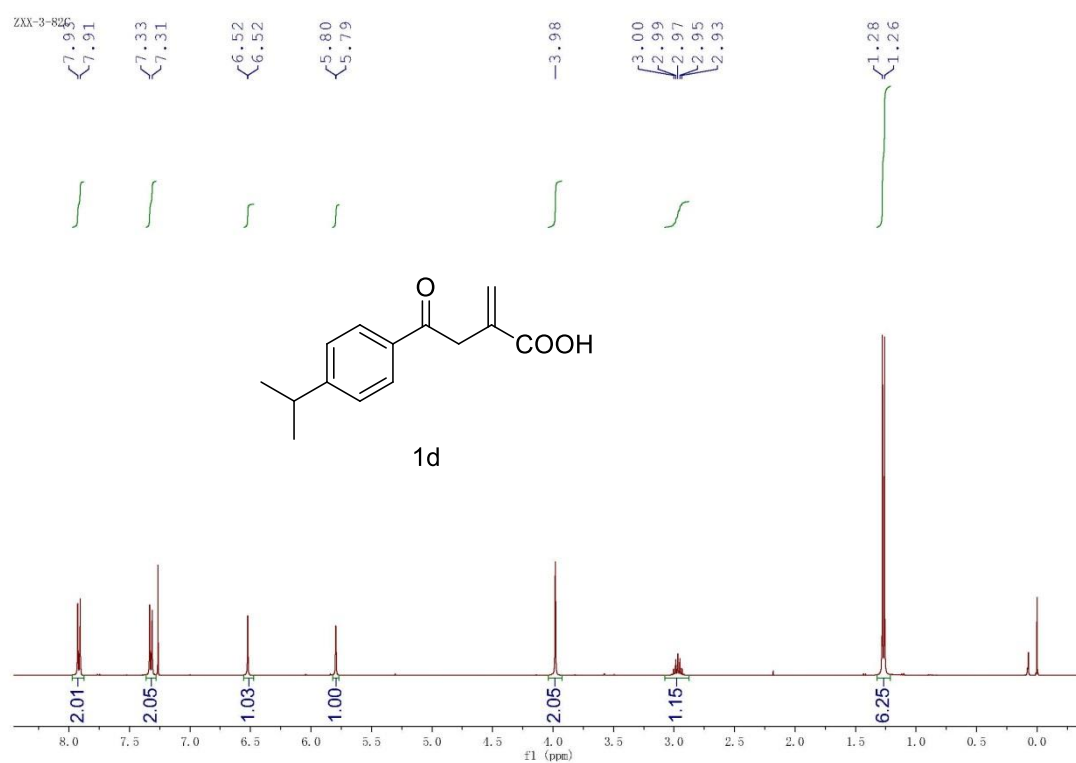
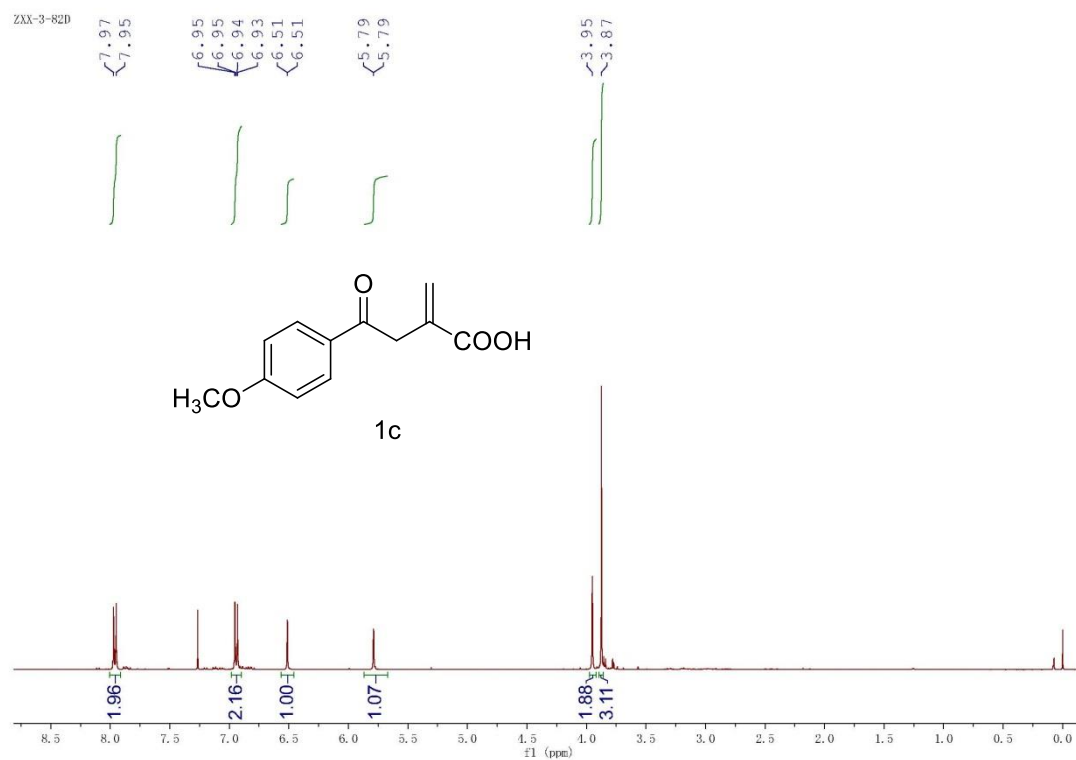
Entry	Solvent	Ligand	Time (h)	Yield(%)	Ee(%)
1	DCE	(R)-BINAP	12	98	21
2	DCE	(R)-OMe-BIPHEP	12	92	35
3	DCE	(R)-SEGPLHOS	12	89	37
4	DCE	(R)-MONOPHOS	48	75	3
5	DCE	(R)-Xyl-SDP	11	98	43
6	DCE	(R,S)-Ph-Bn-SIPHOS	30	81	10
7	DCE	(S,S)-Et-FerroTANE	11	97	19
8	DCE	(R)-DIFLUORPHOS	15	94	5
9	DCE	(R)-P-PHOS	10	98	15
10	DCE	(R)-BDPP	10	98	9
11	DCE	(S)-SYNPHOS	10	81	41
12	DCE	(R)-DIOP	30	88	3
13	DCE	(R)-An-SDP	10	98	17
14 ^a	DCE	(R)-Xyl-SDP	11	98	45

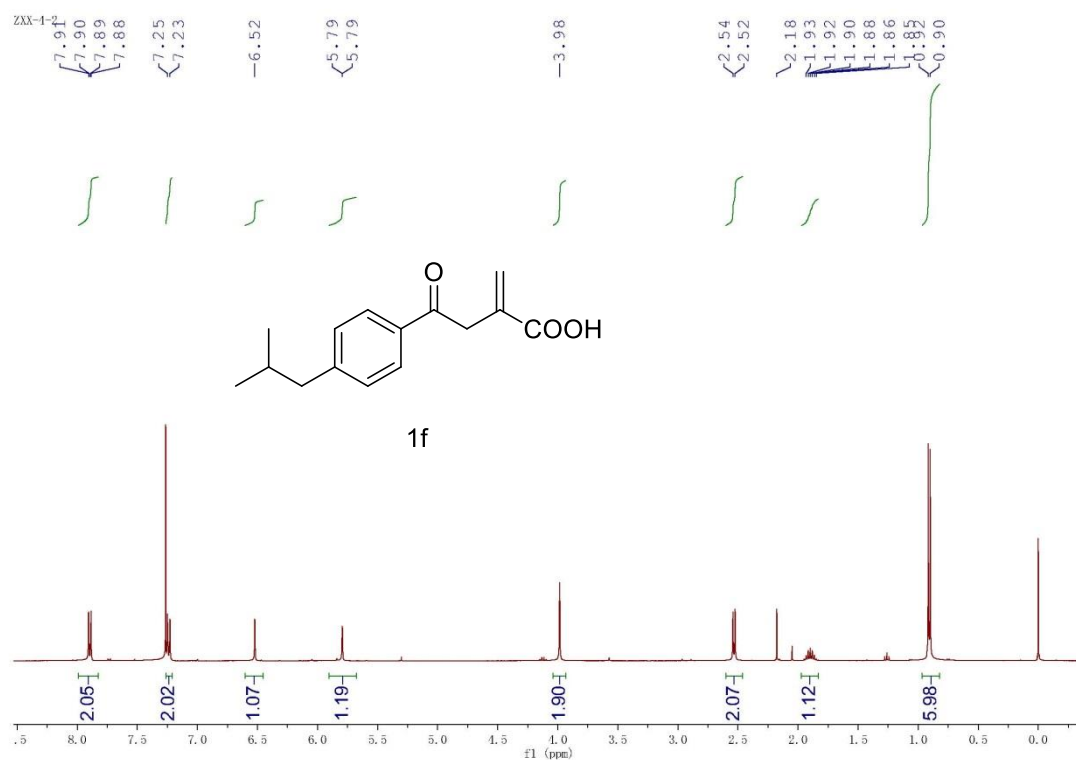
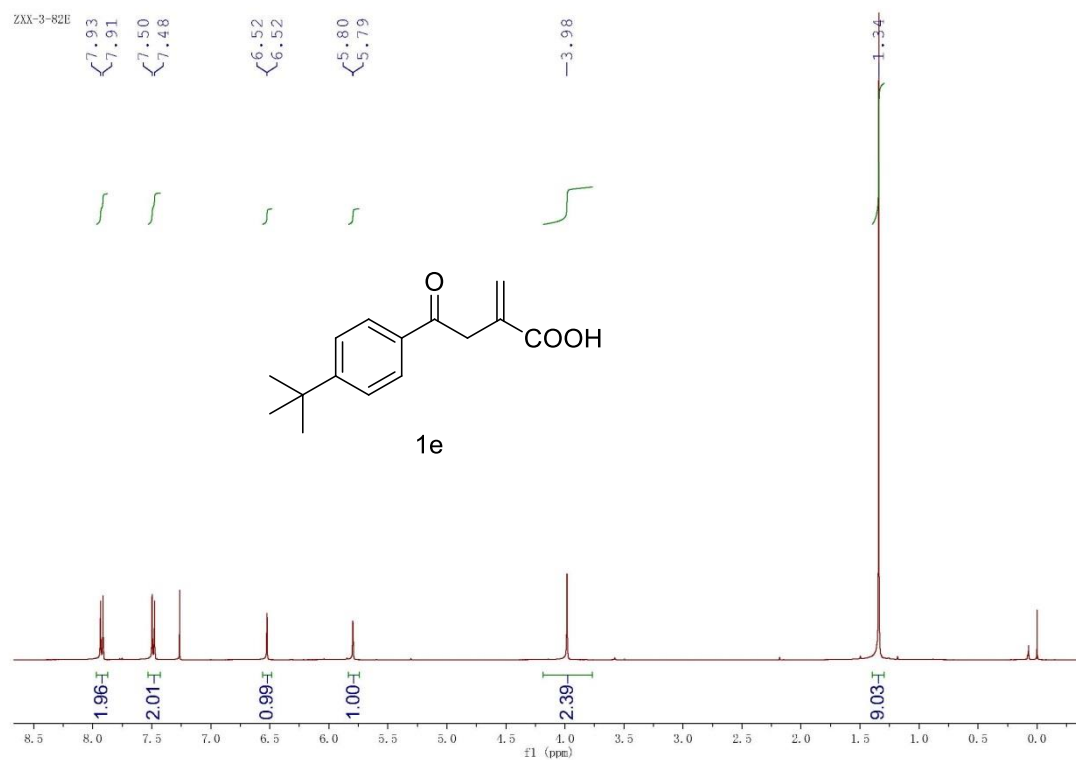
^aReaction conditions: Pd(OAc)₂ (0.01 mmol), Lewis acid (0.02 mmol), and ligand (0.012 mmol) in DCE (2 mL) was stirred at room temperature for 30 min under Argon. **1a** (0.2 mmol) were added, and the reaction mixture in 1 bar H₂ was stirred at 70 °C for indicated period of

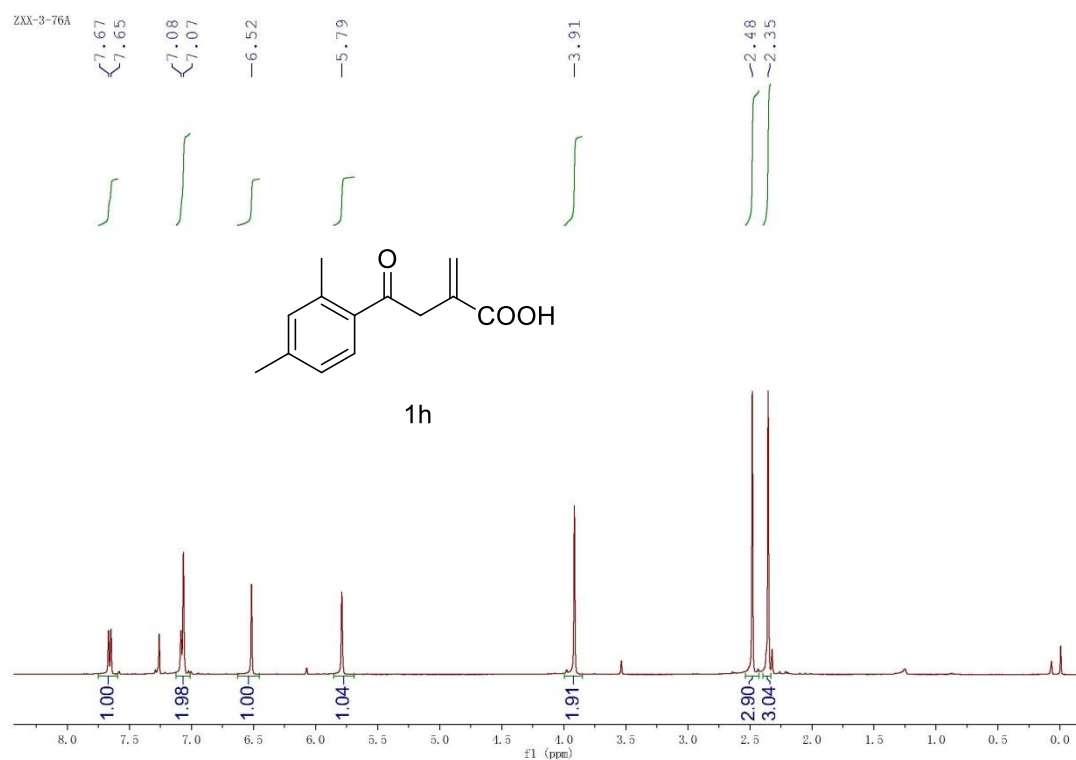
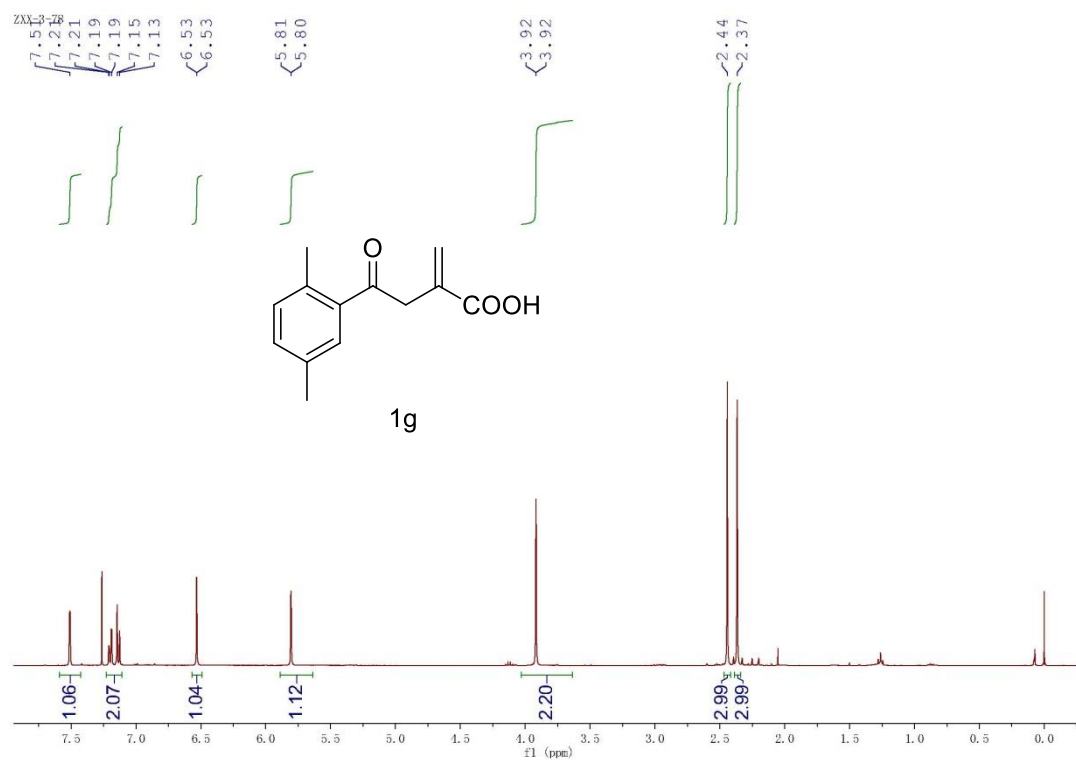
time. ^bThe reaction was performed at 60 °C.

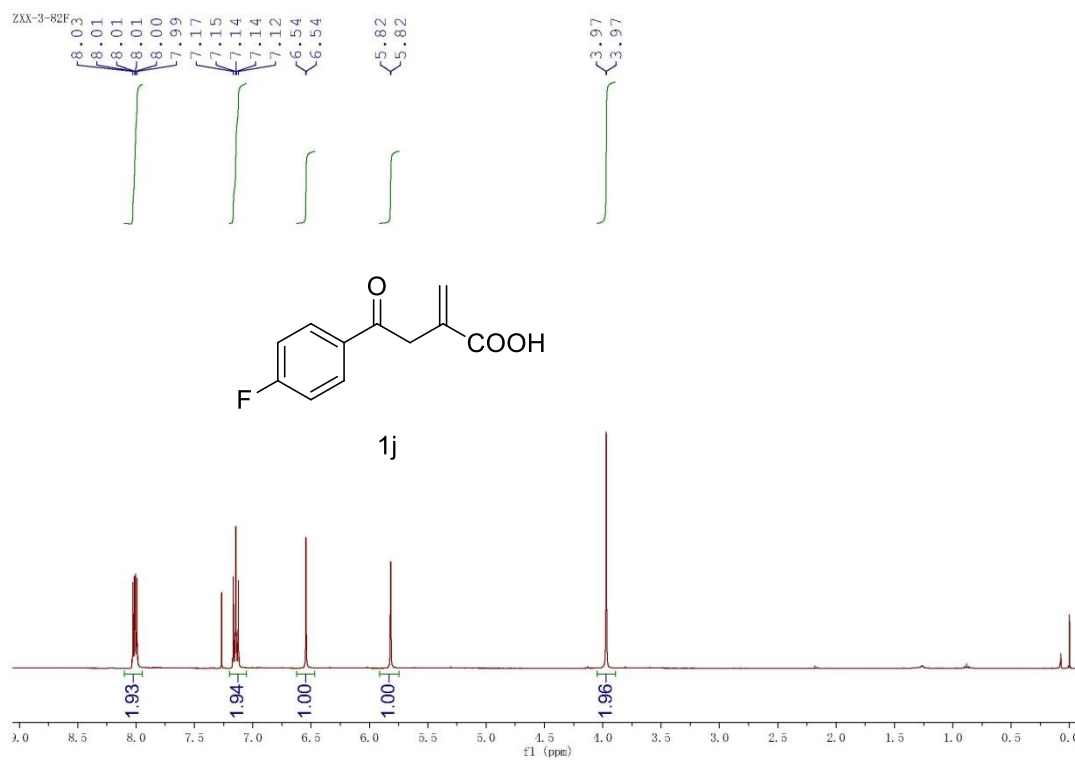
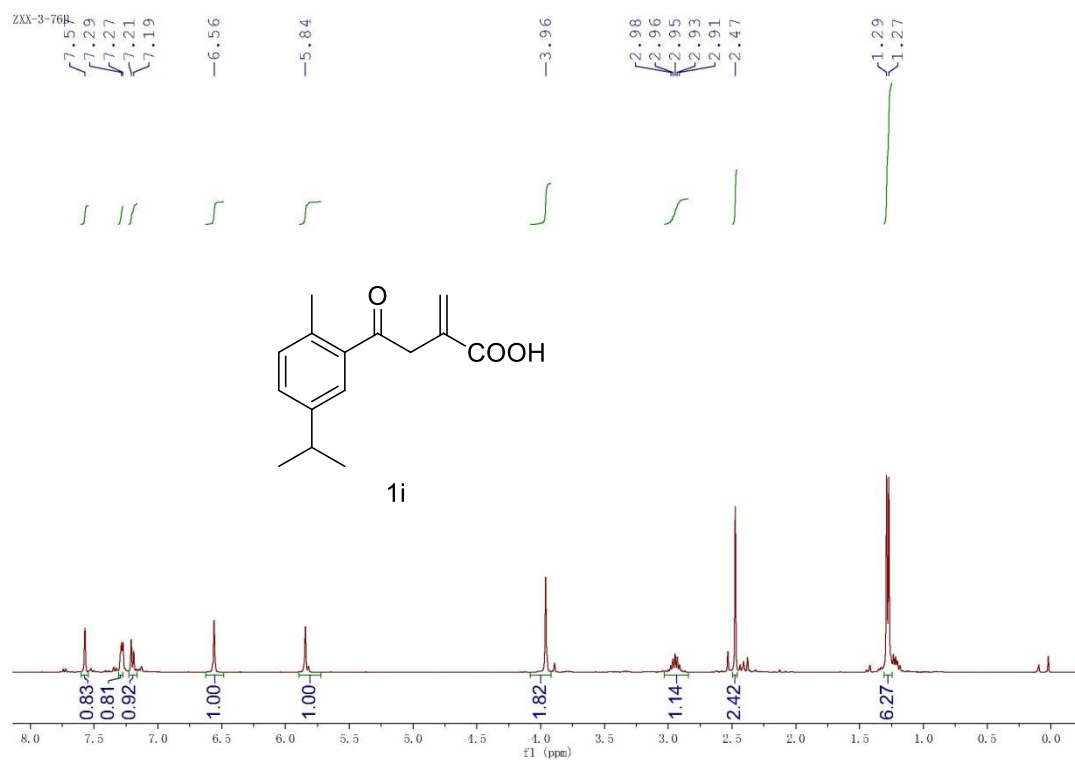
C: NMR spectra of substrates



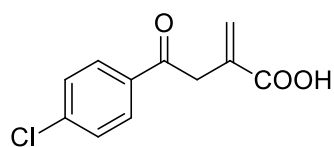
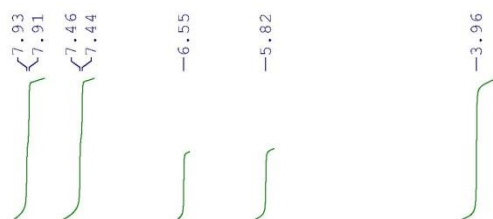




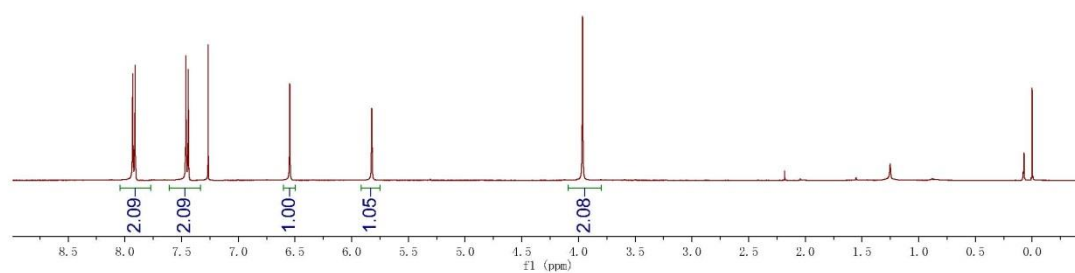




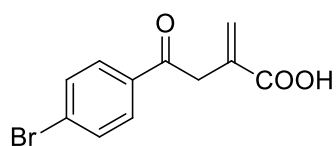
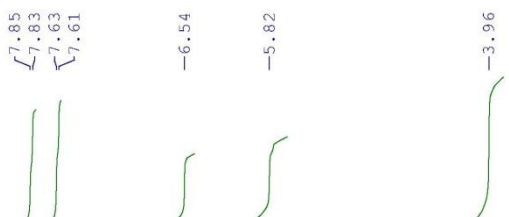
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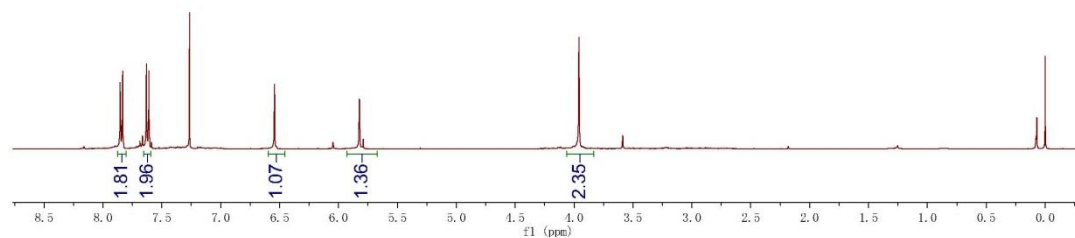
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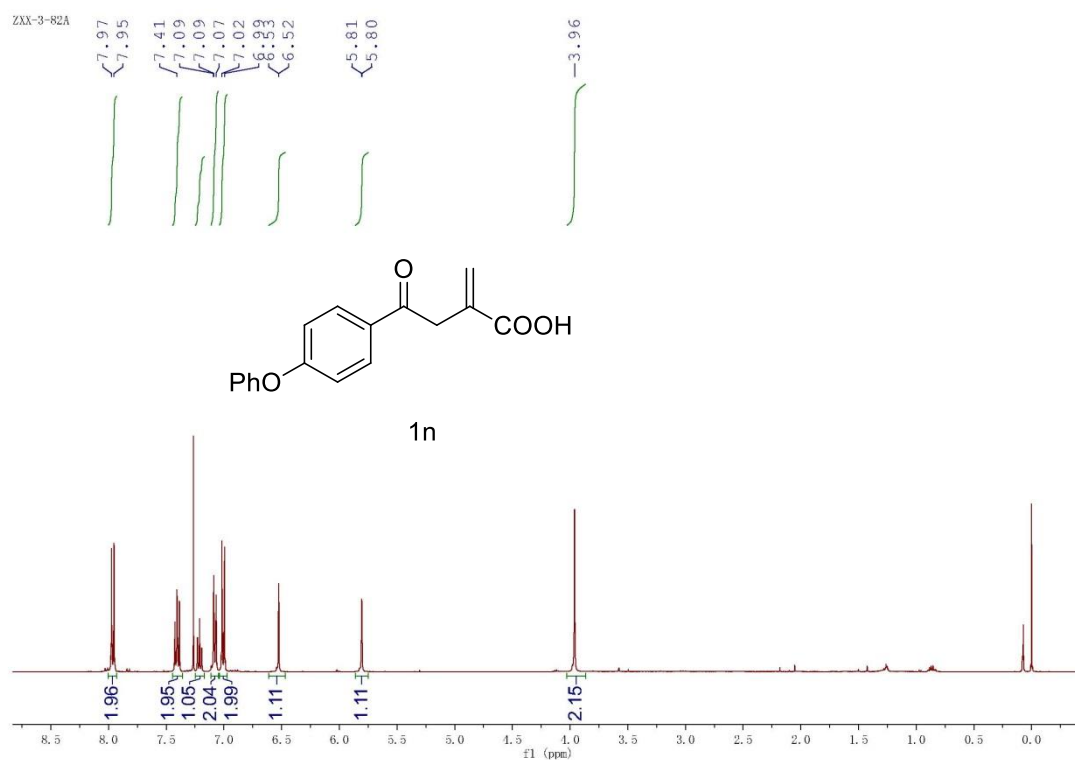
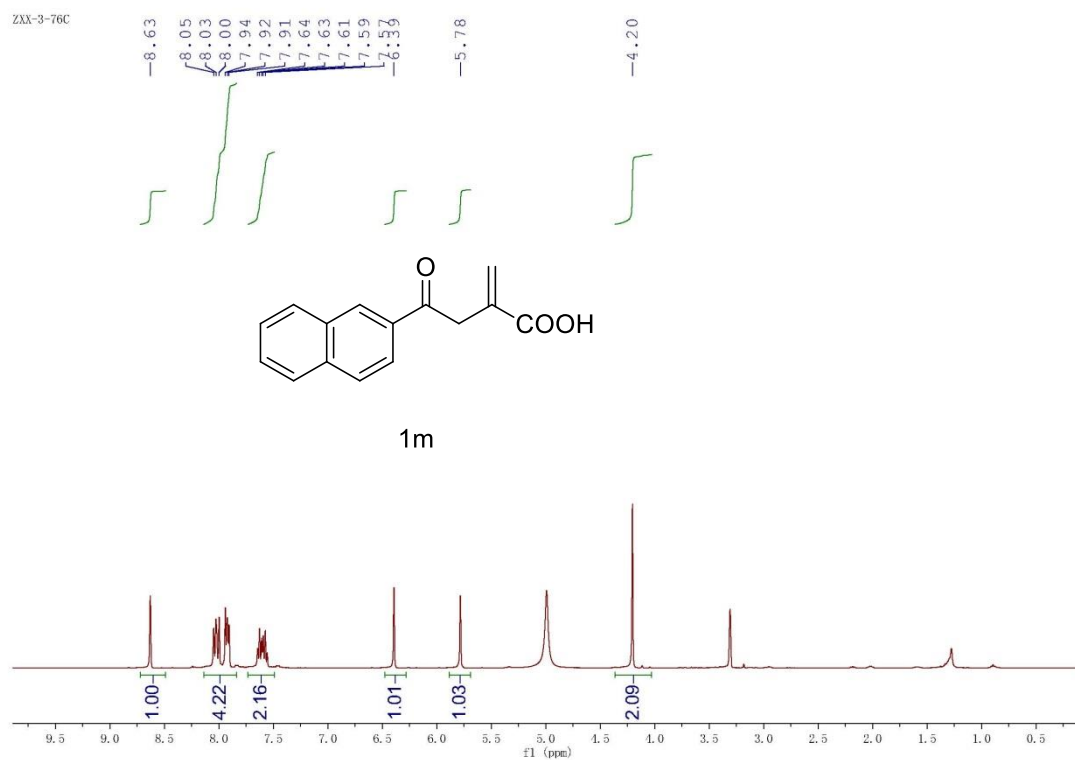


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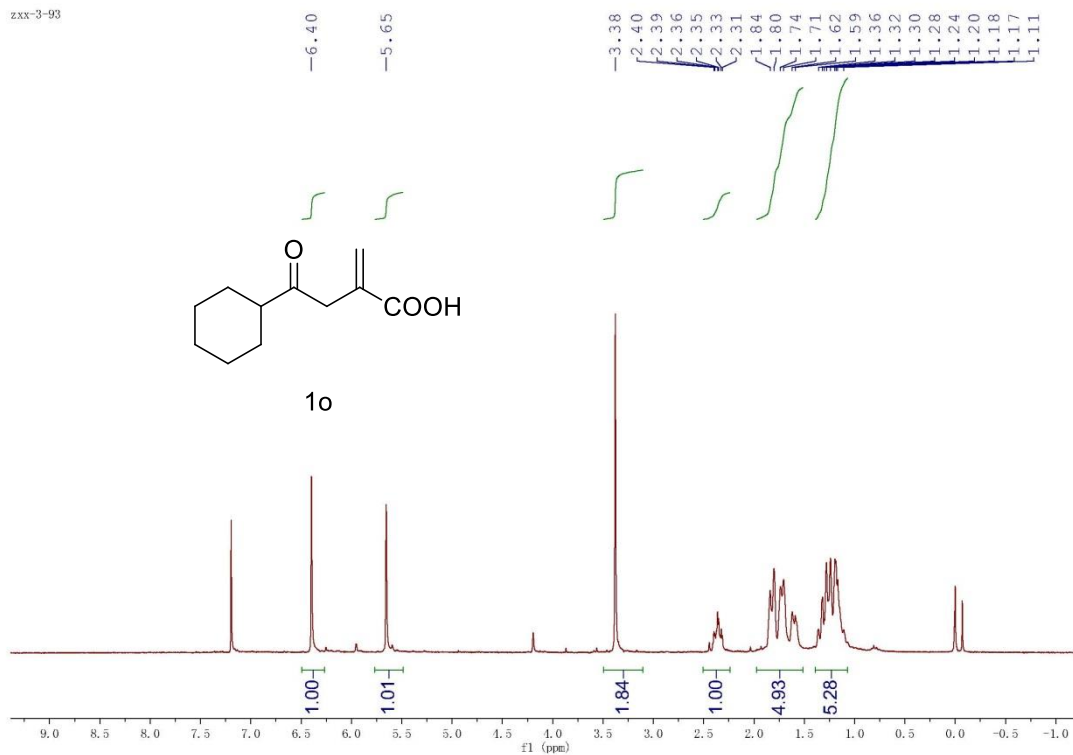


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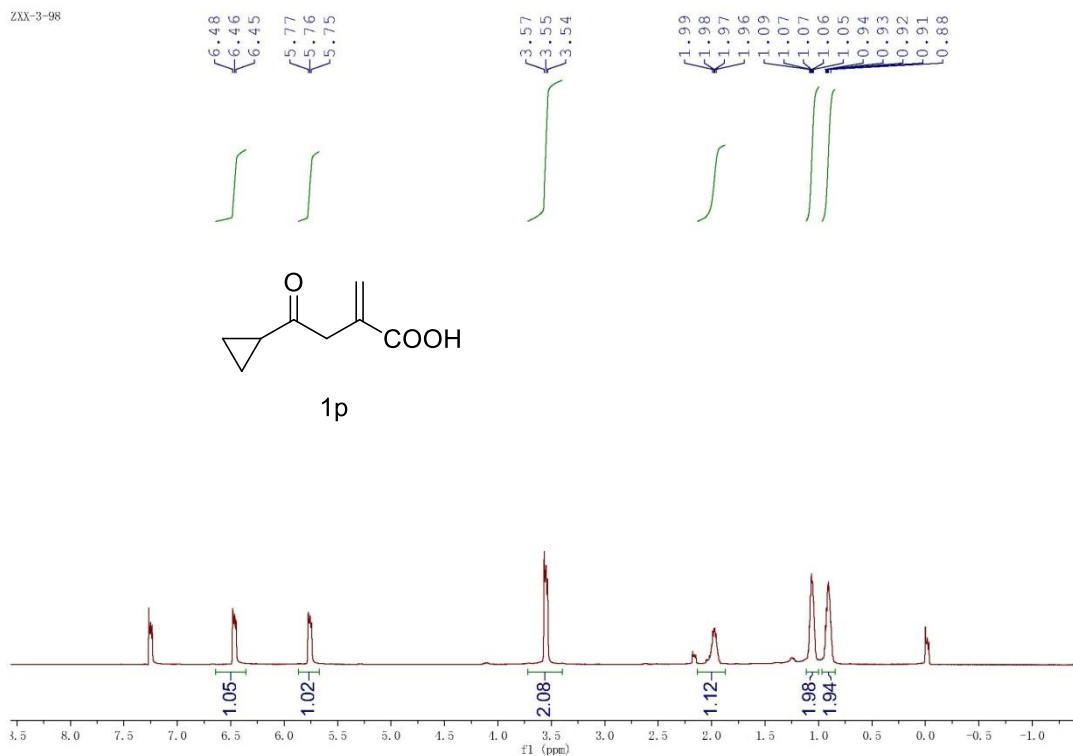


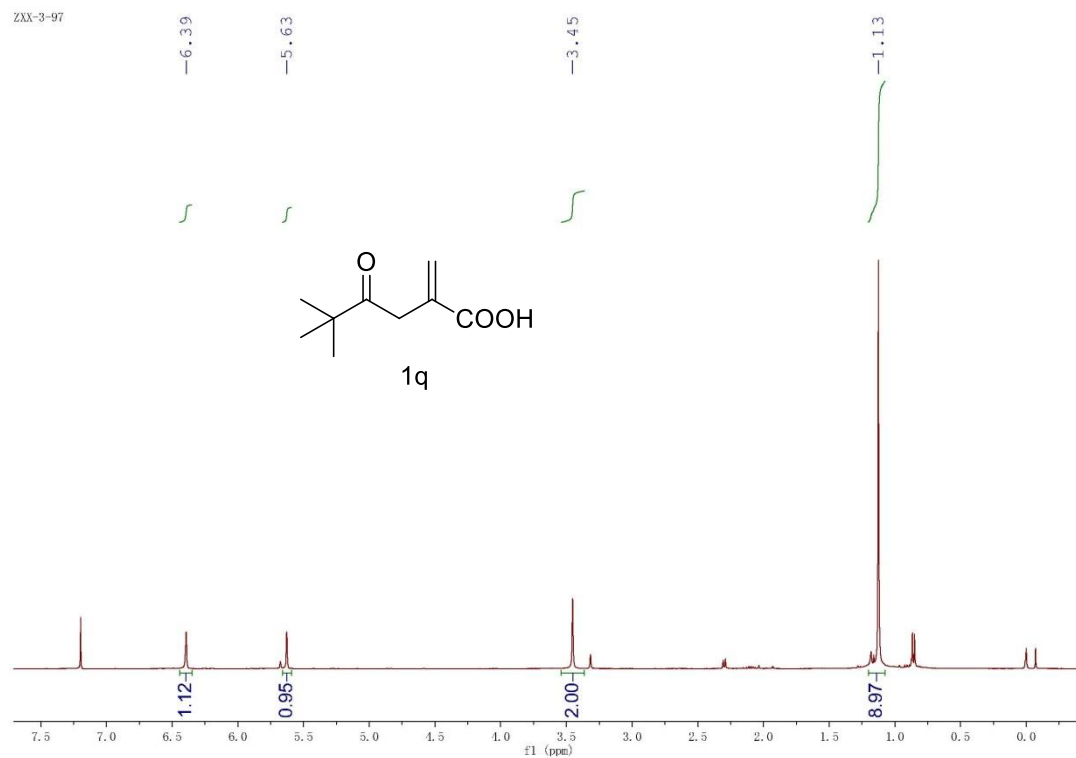


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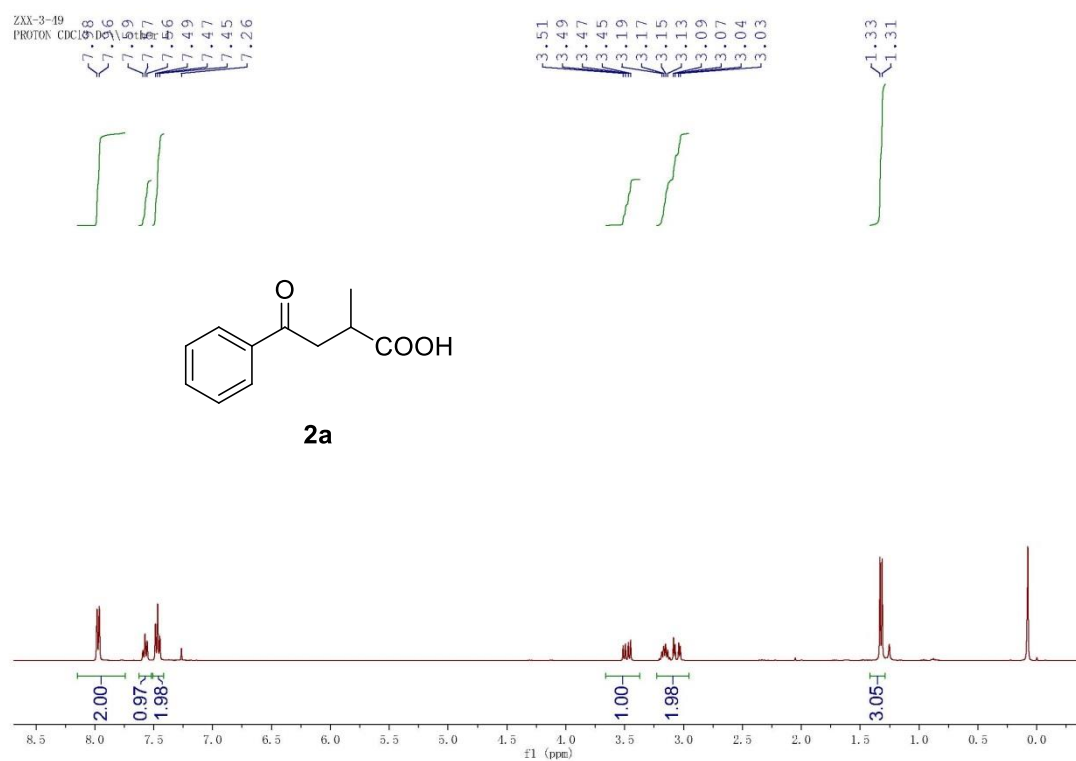


ZXX-3-98





D: NMR Spectra of Products



ZXX-3-19

-197.90

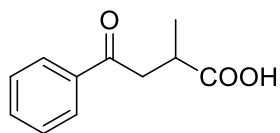
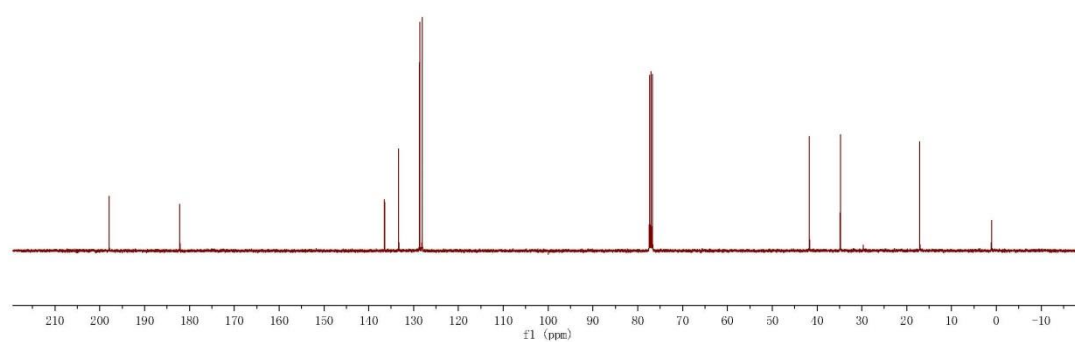
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133.34
128.65
128.08

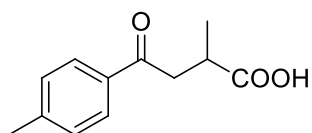
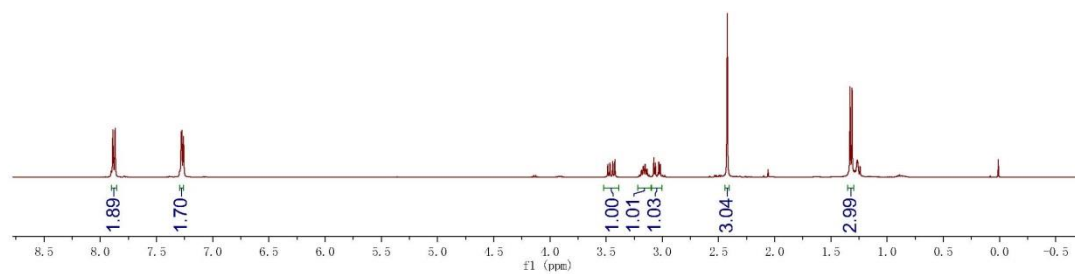
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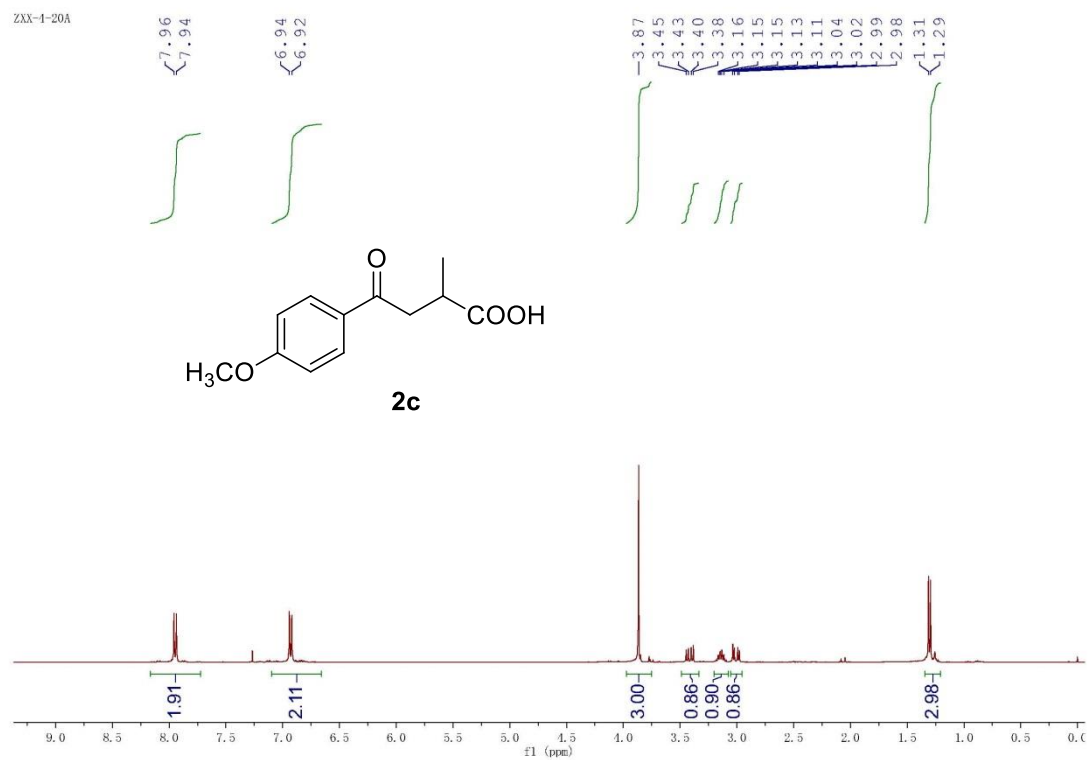
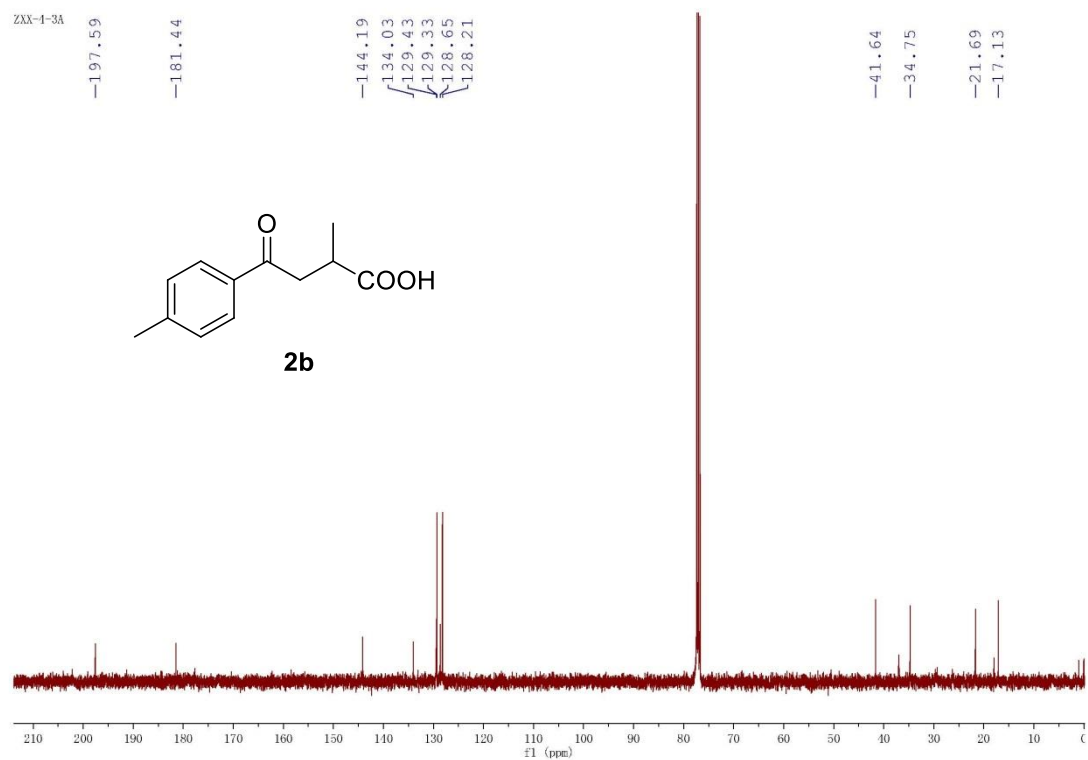
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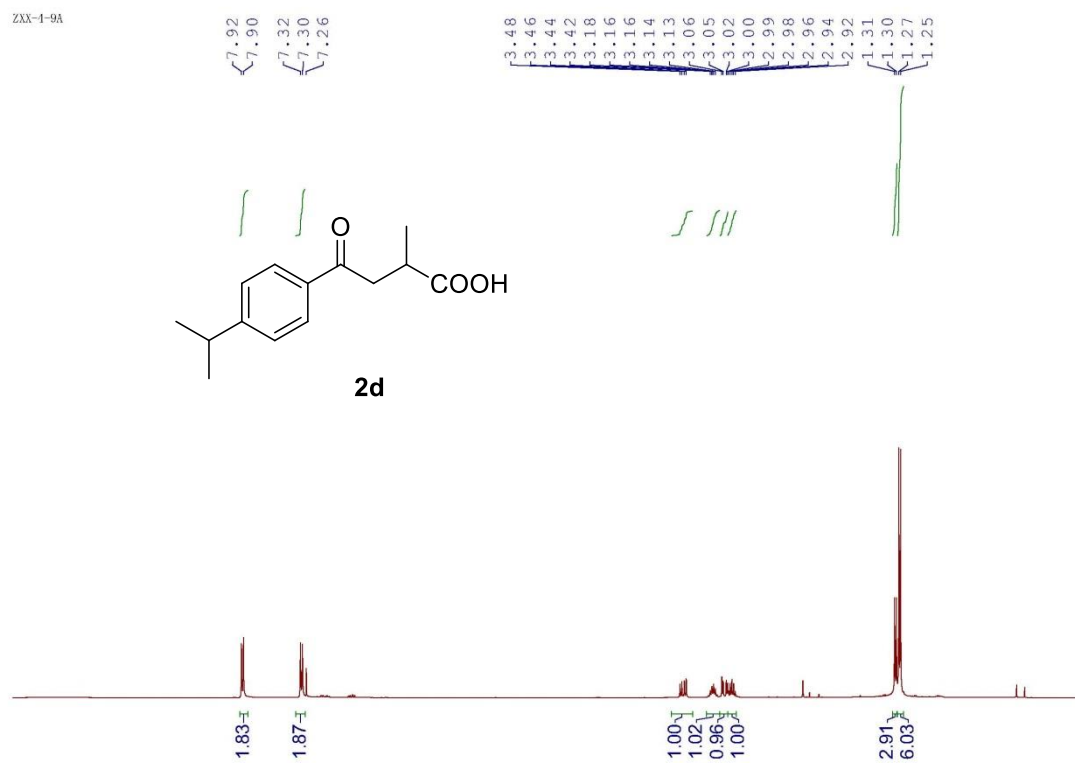
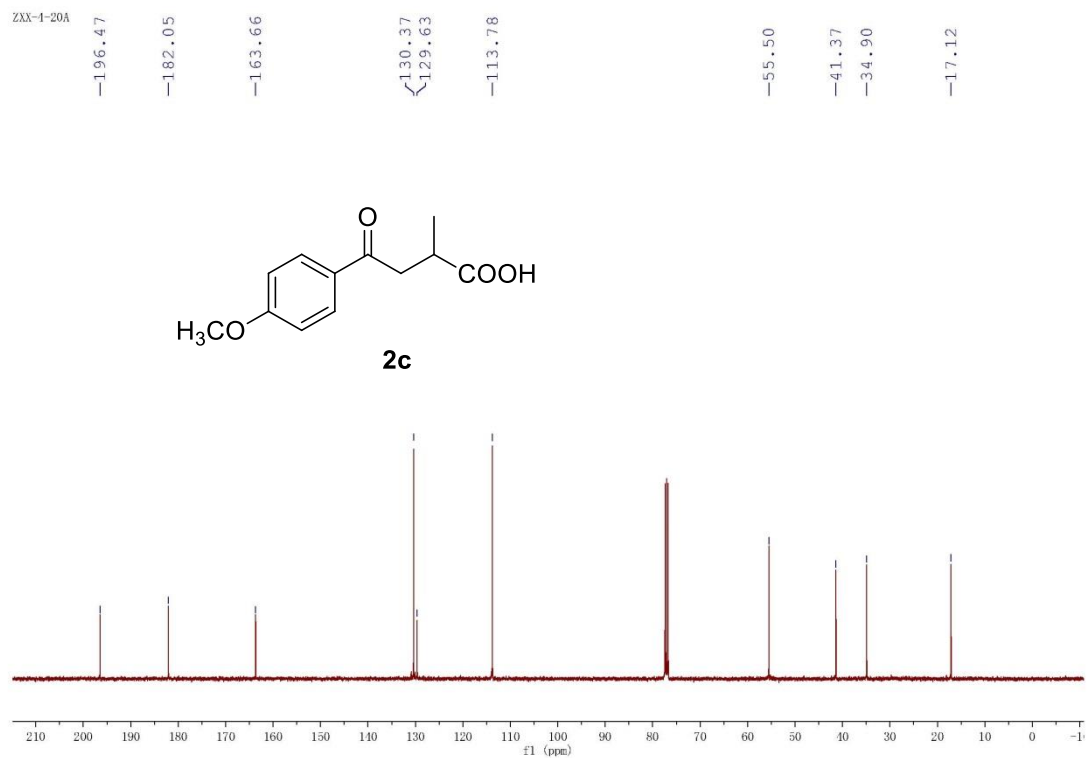
-17.12

**2a**

ZXX-4-3A

7.89
7.87
7.28
7.273.49
3.47
3.44
3.42
3.20
3.19
3.17
3.15
3.14
3.08
3.06
3.03
3.02
2.421.33
1.31**2b**





ZXX-1-9A

-197.62

-182.23

-154.88

-134.41

-128.36

-126.74

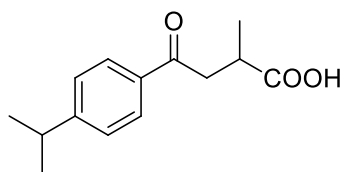
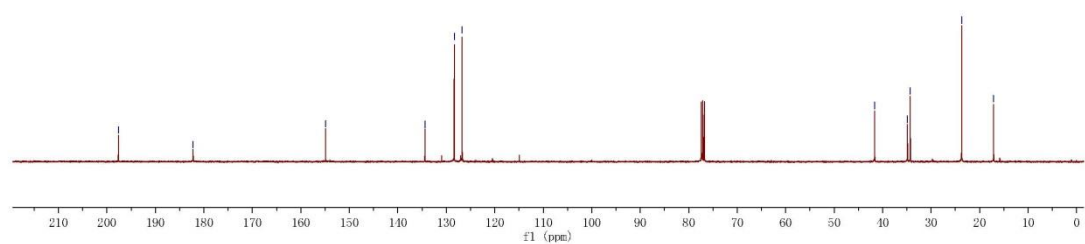
-41.64

-34.88

-34.29

-23.68

-17.10

**2d**

ZXX-1-8A

7.92

7.90

7.48

7.46

7.26

3.48

3.46

3.44

3.42

3.18

3.16

3.16

3.15

3.14

3.13

3.06

3.05

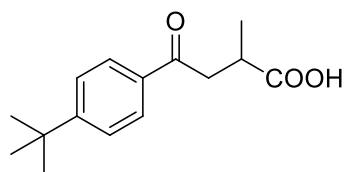
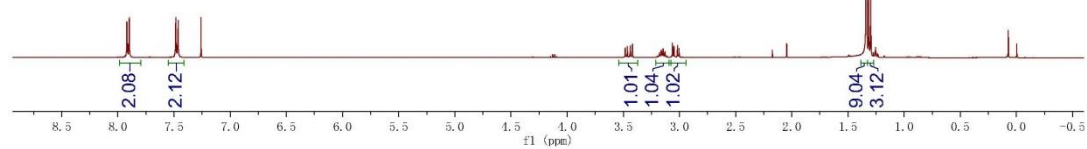
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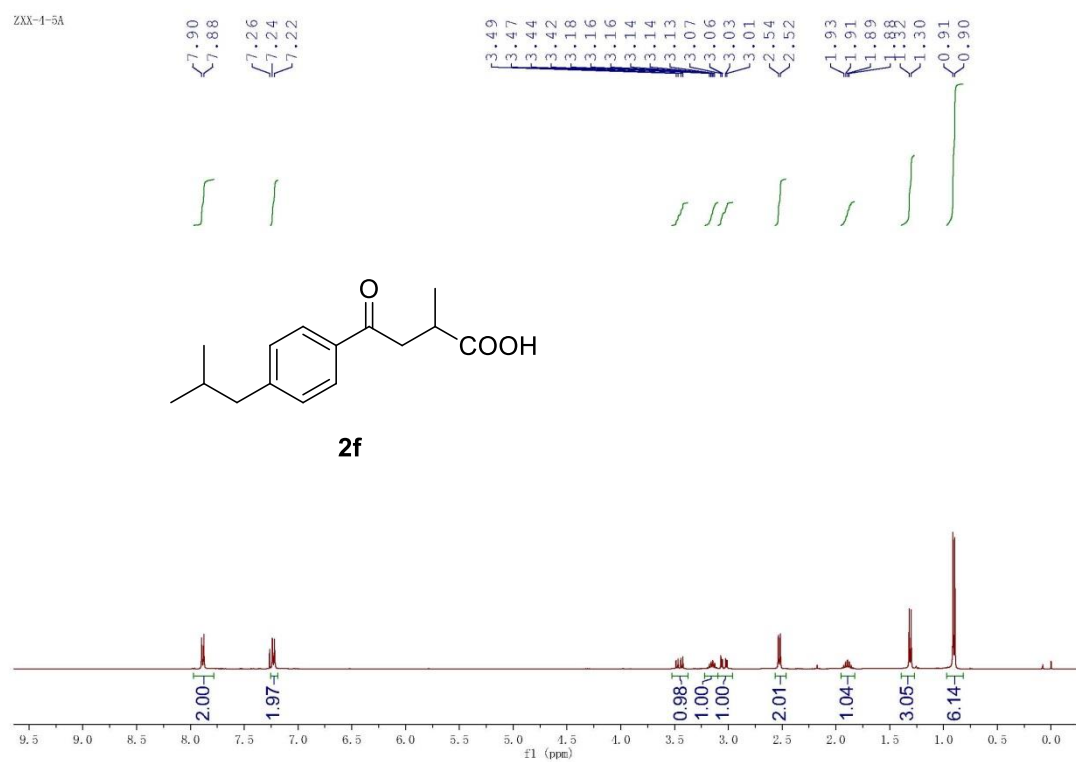
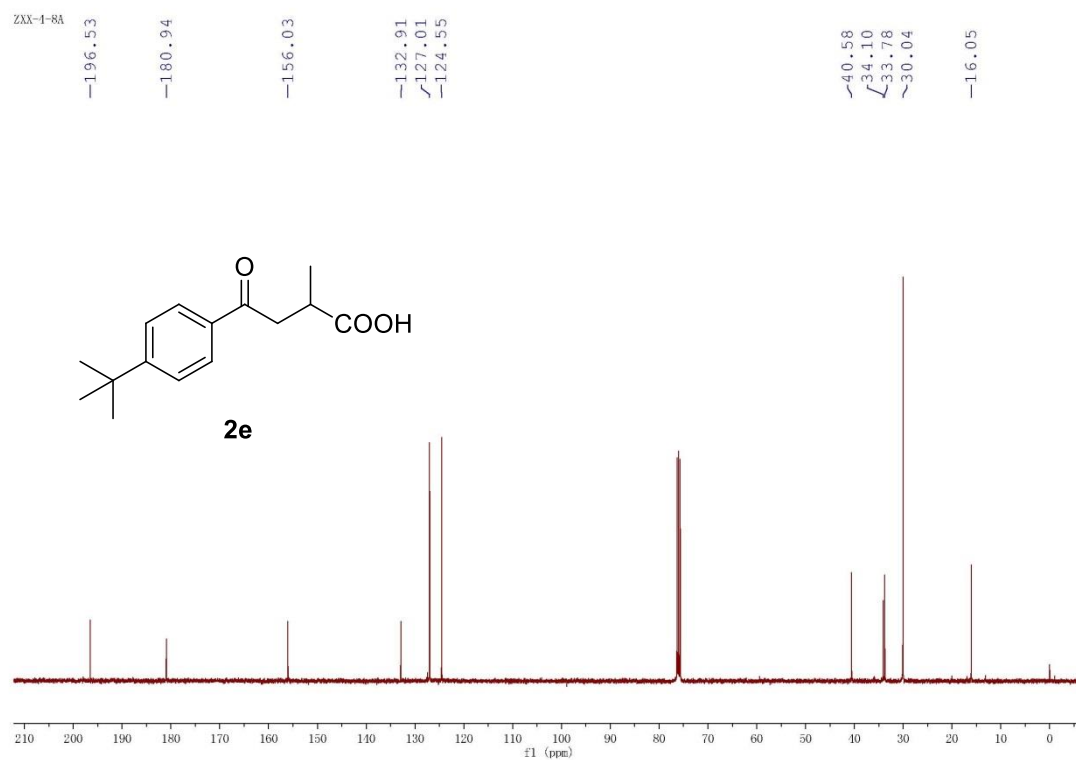
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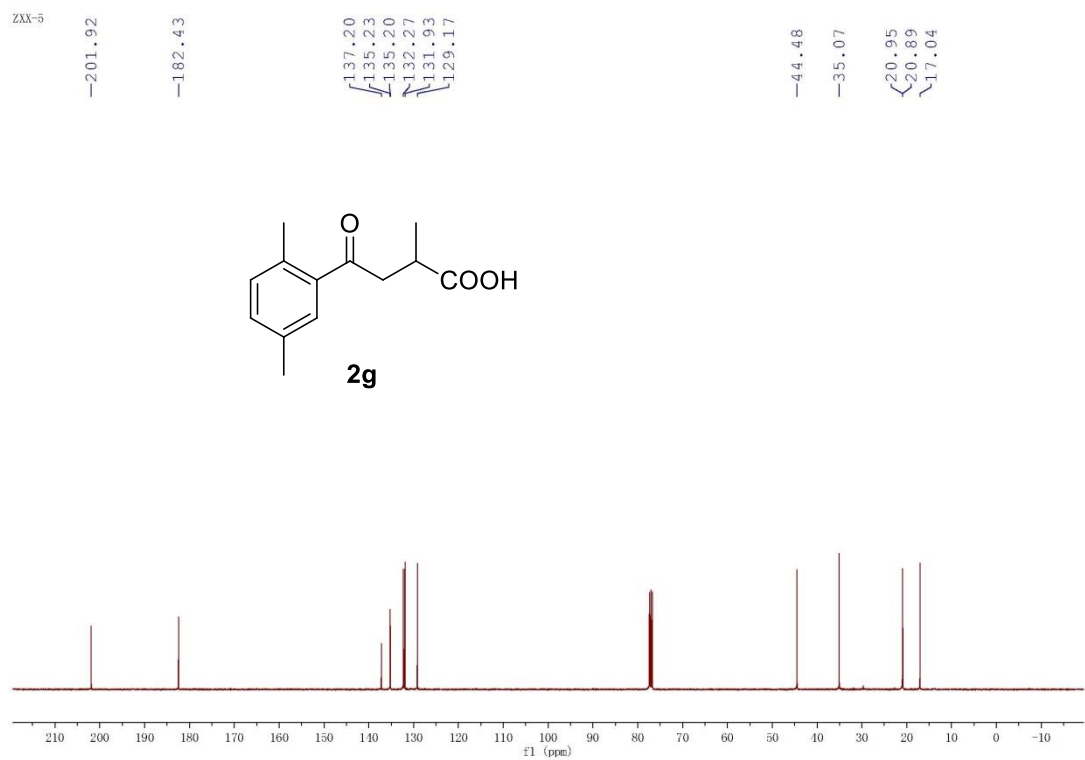
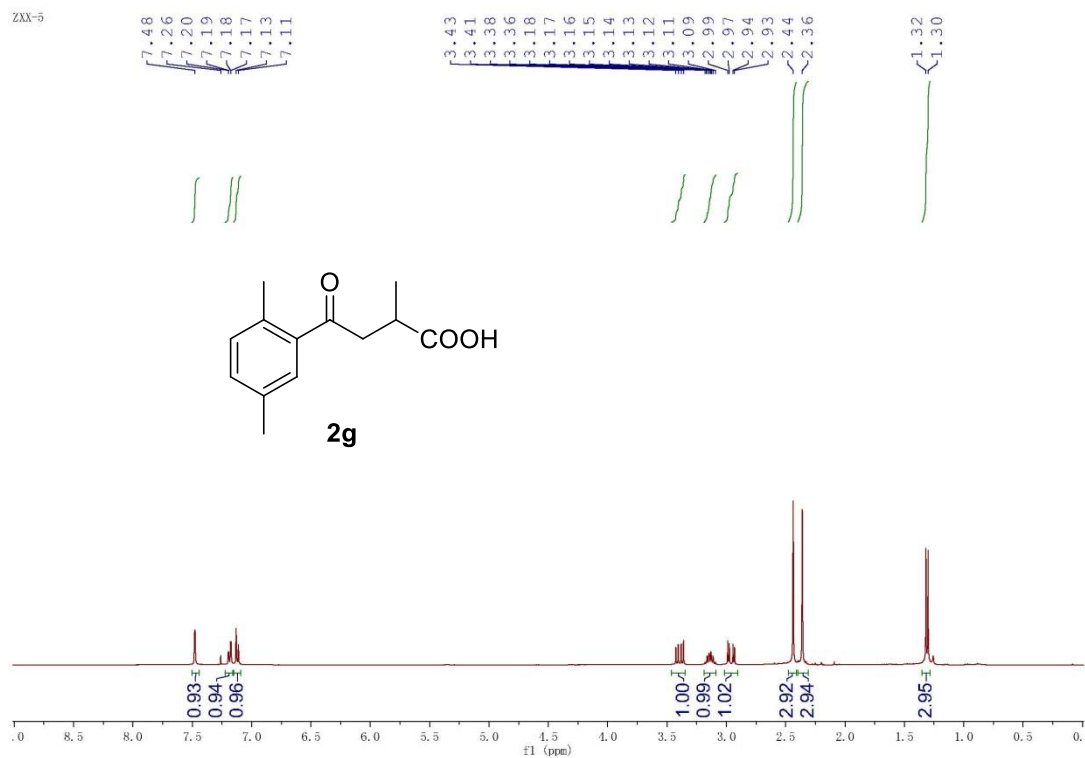
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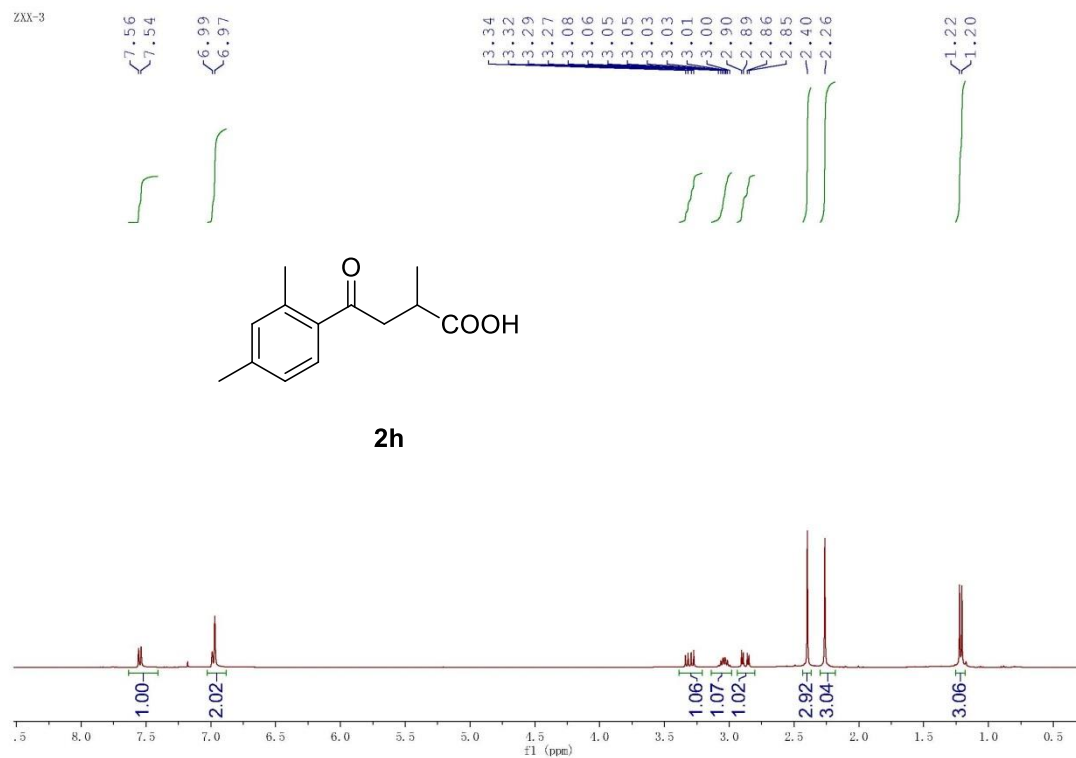
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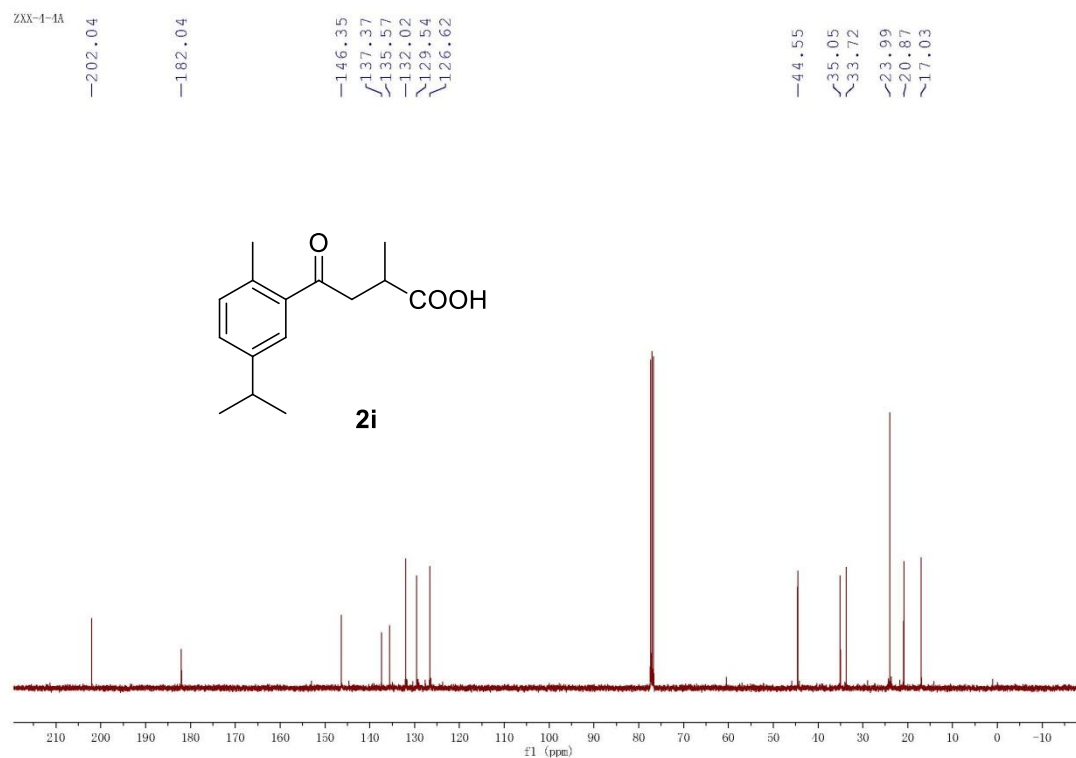
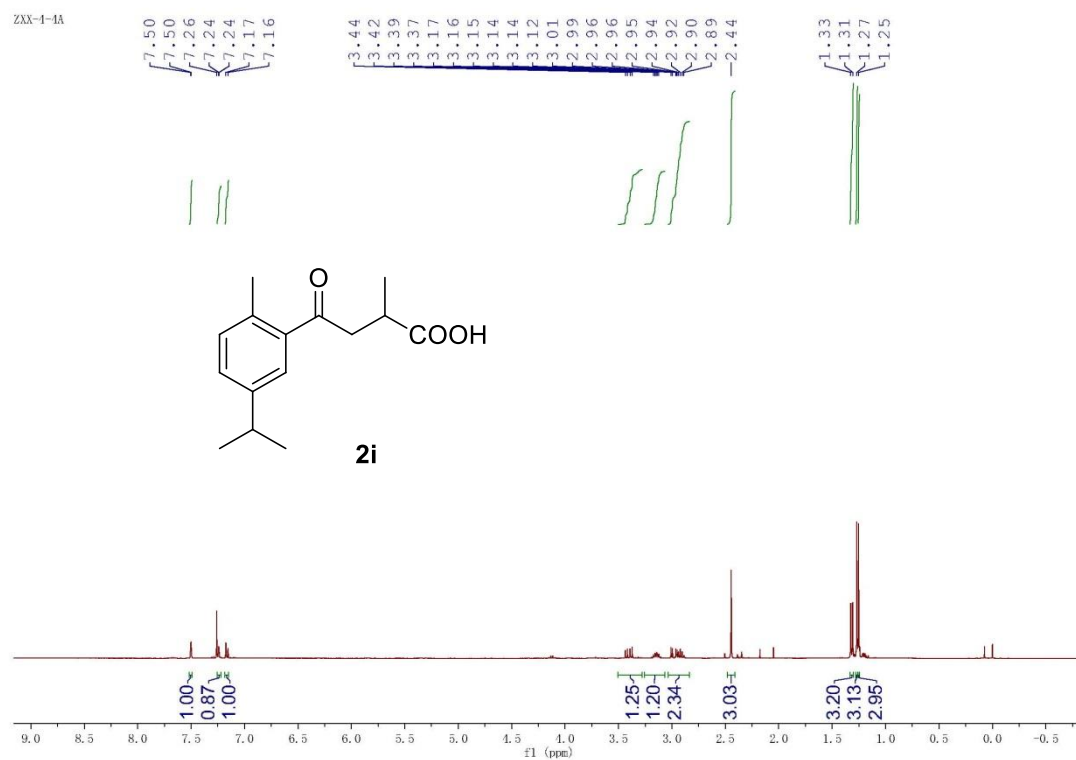
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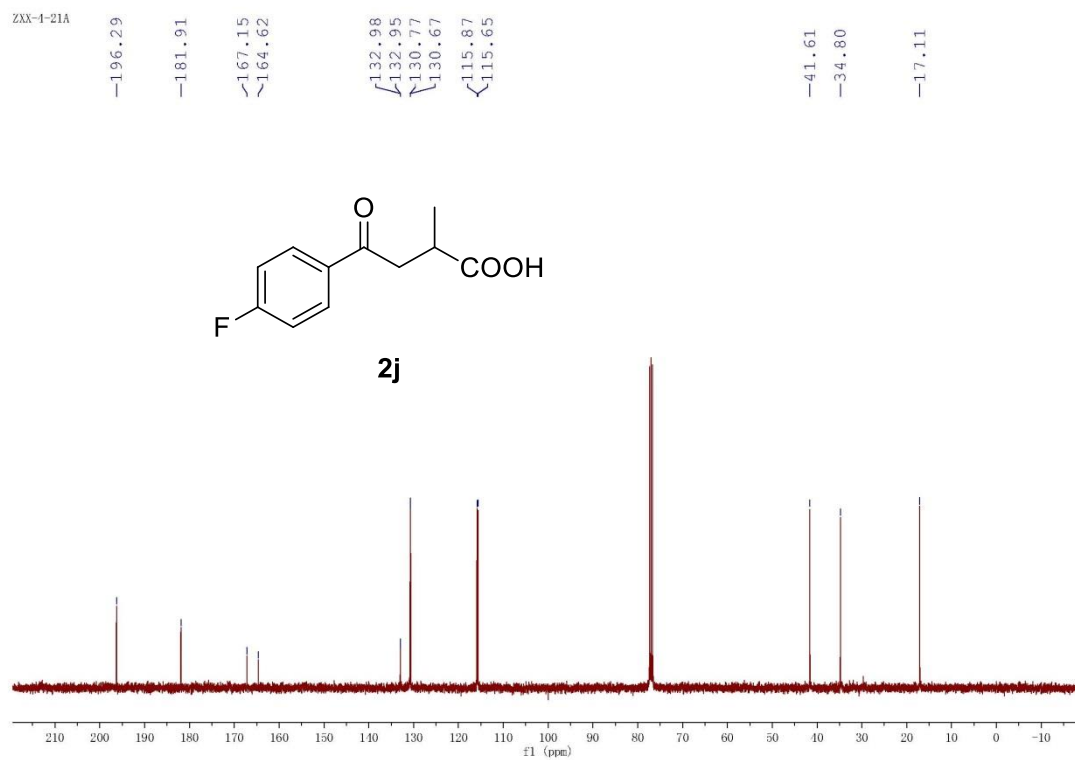
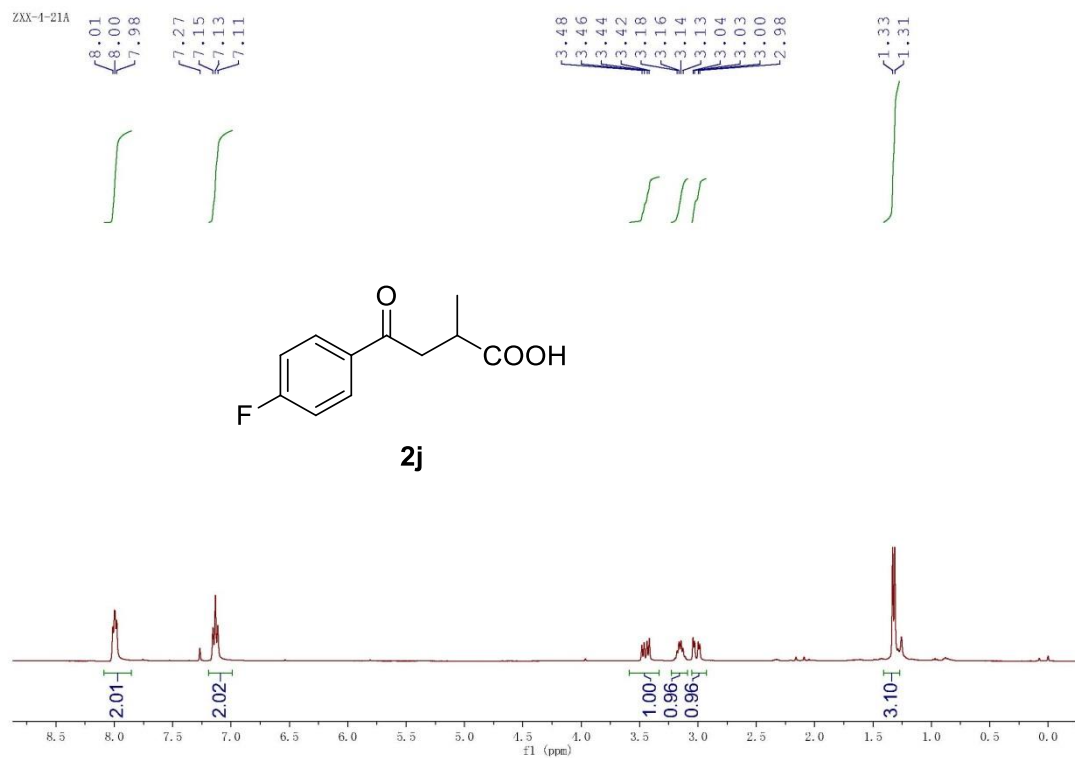
**2e**



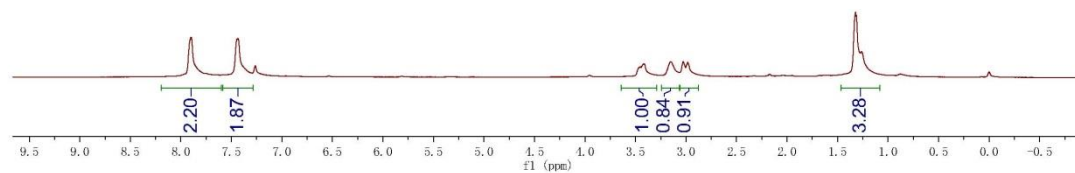
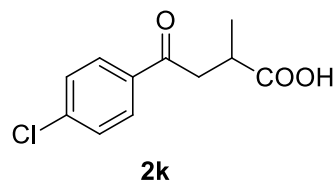




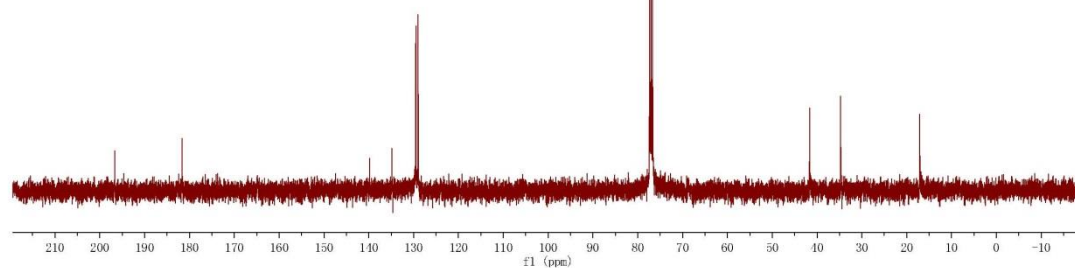
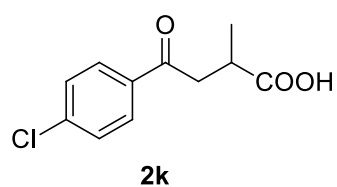




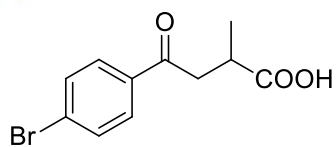
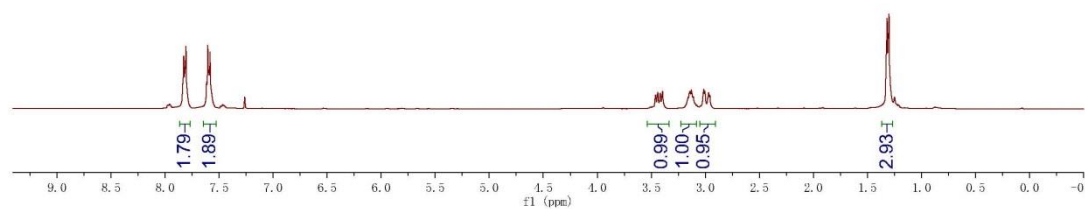
ZXX-1-21B



ZXX-1-21B



ZXX-6

7.83
7.81
7.60
7.583.46
3.44
3.42
3.40
3.16
3.15
3.14
3.13
3.11
3.09
3.01
3.00
2.97
2.961.32
1.30**2l**

ZXX-6

-196.89

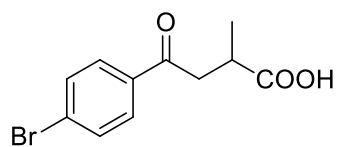
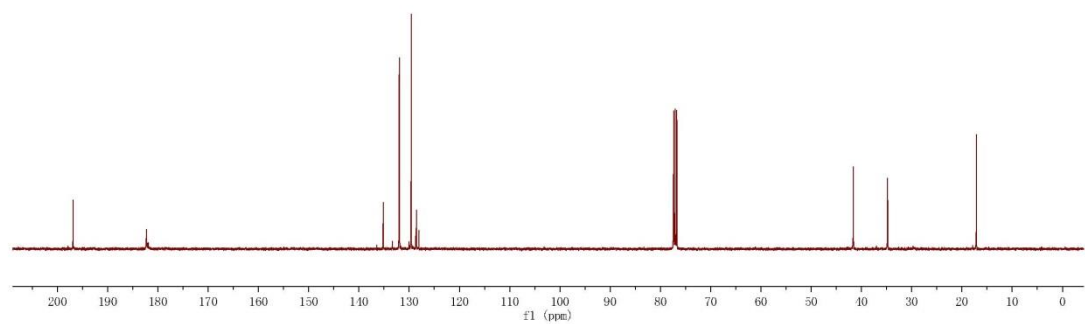
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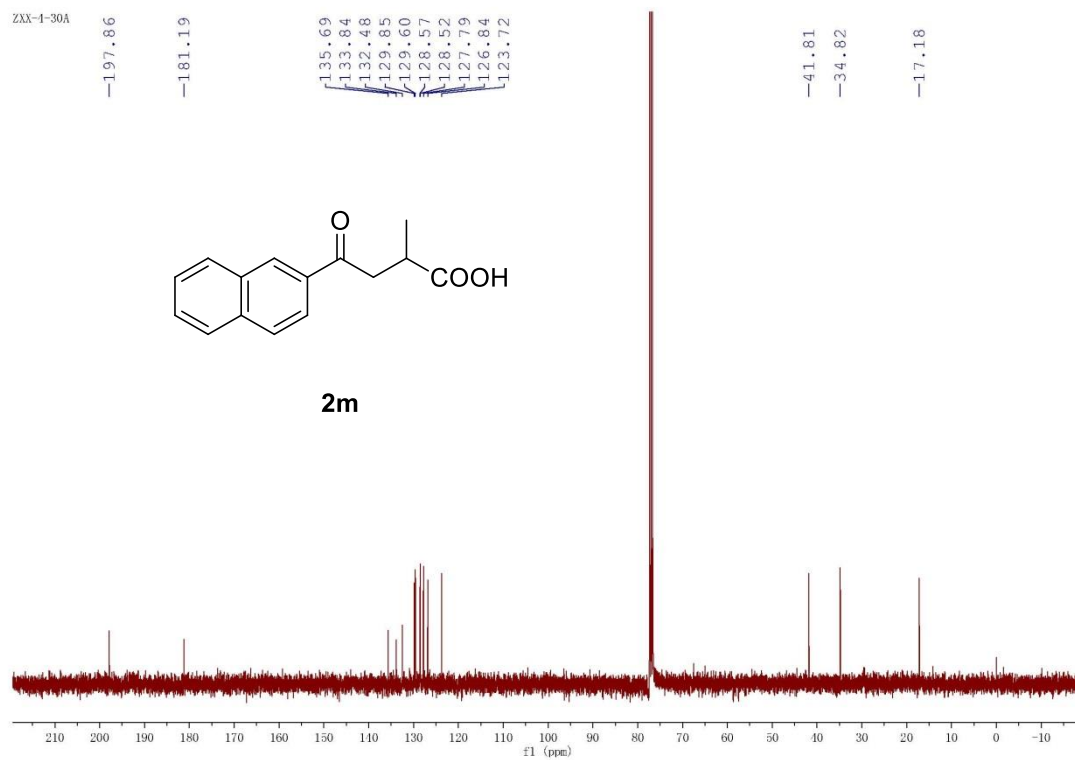
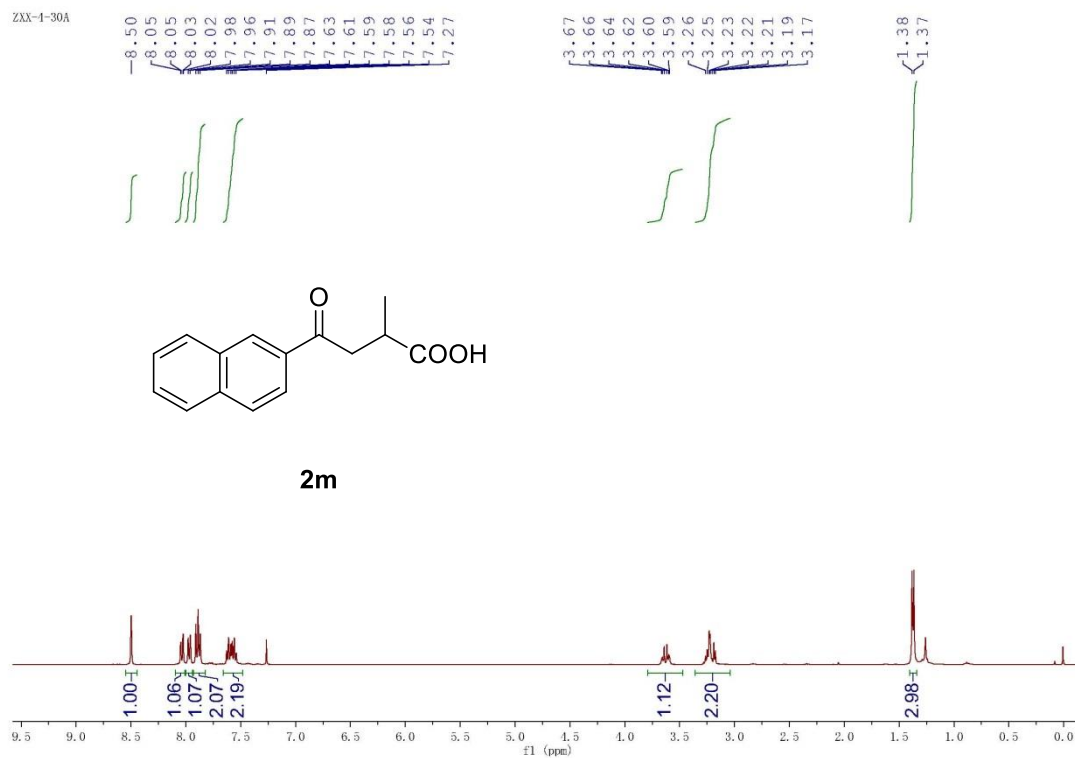
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128.54

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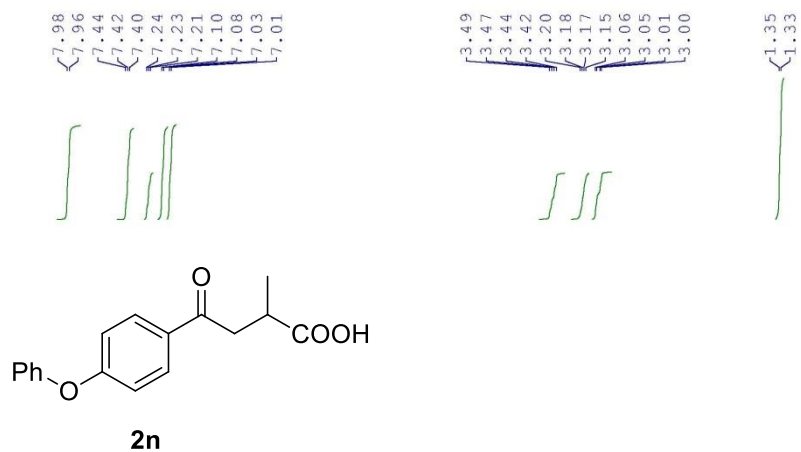
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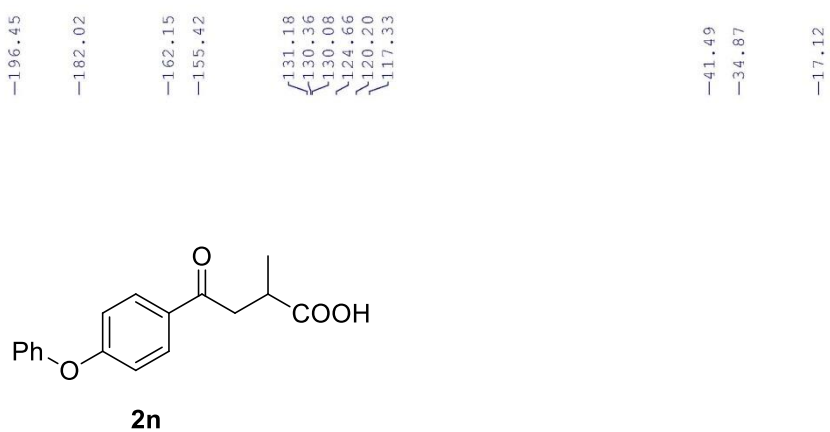
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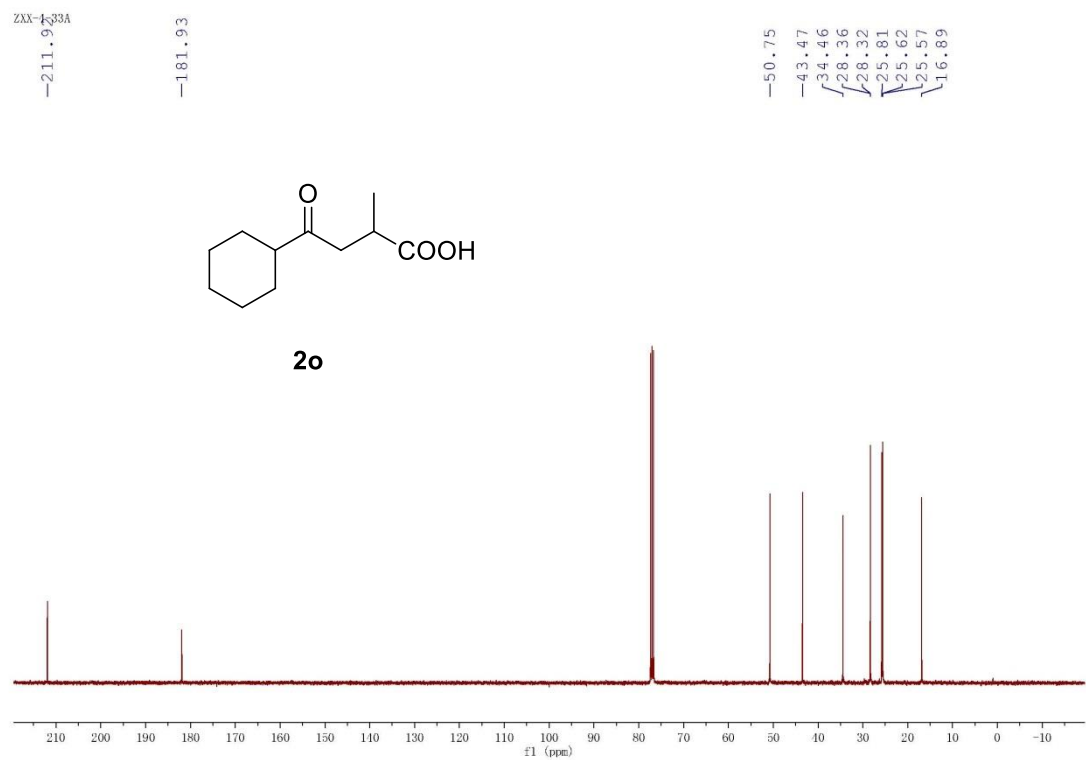
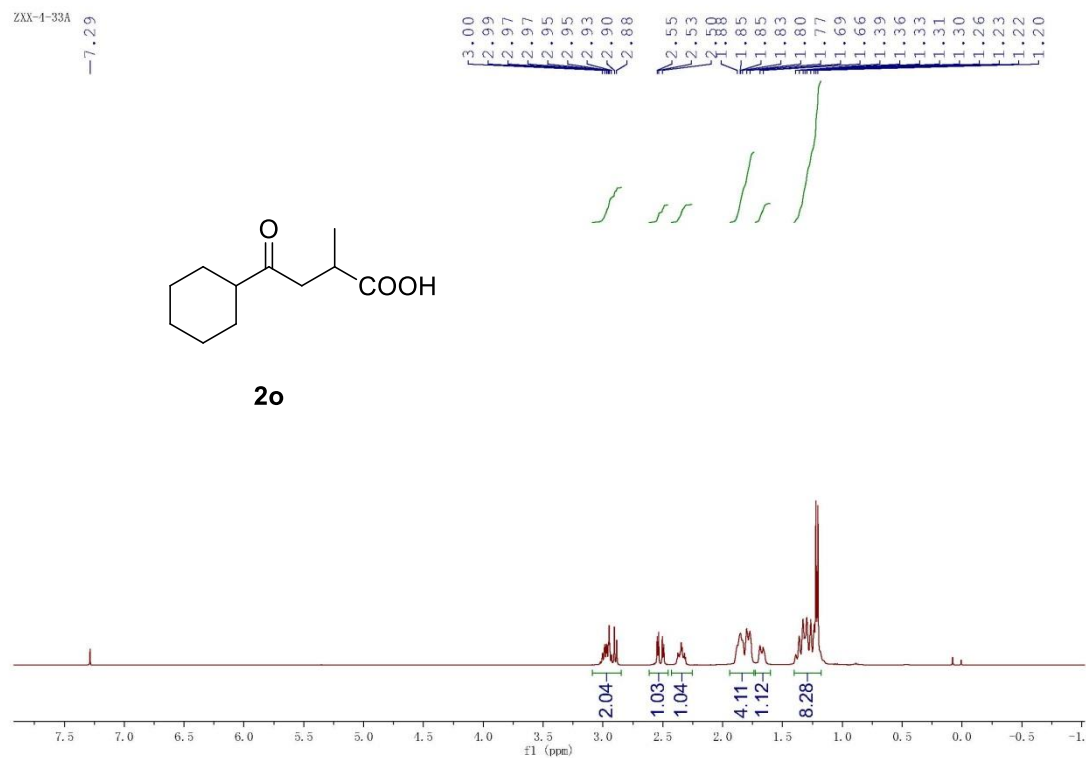


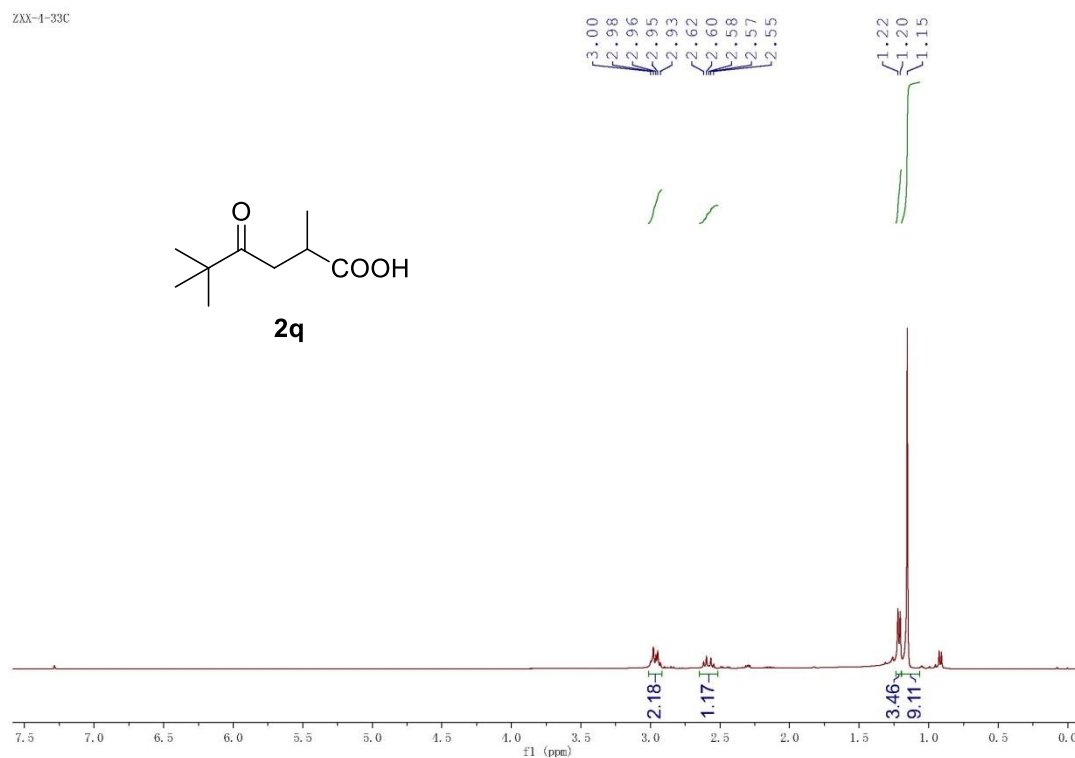
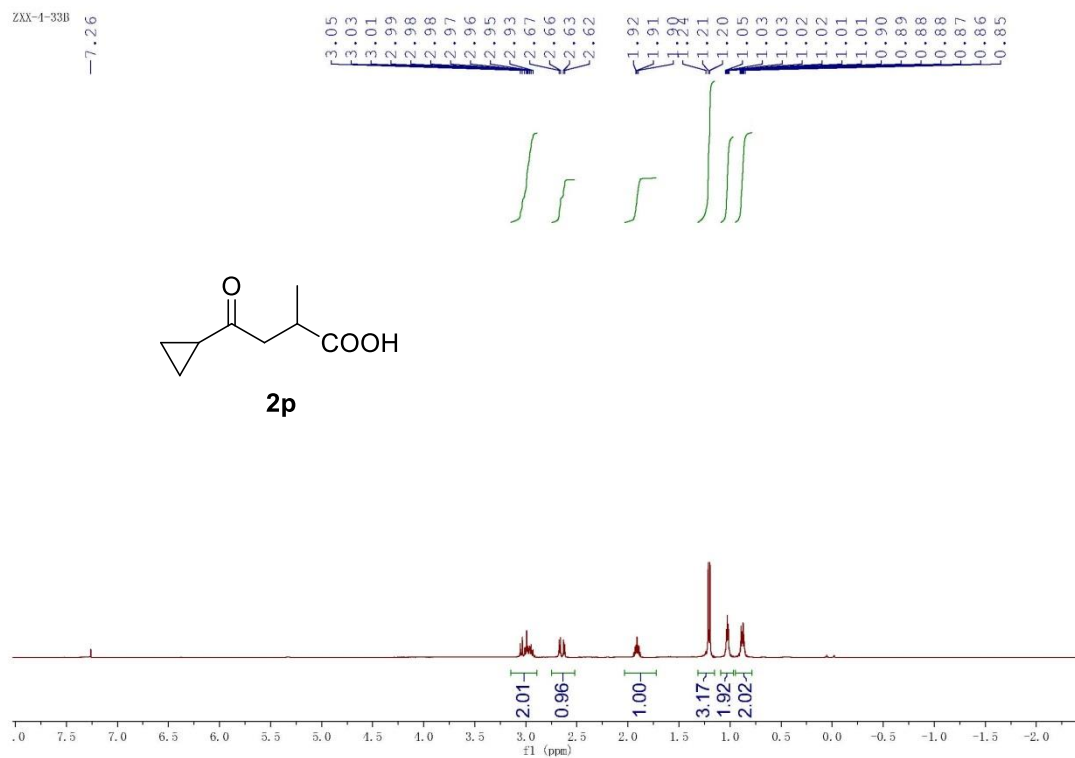
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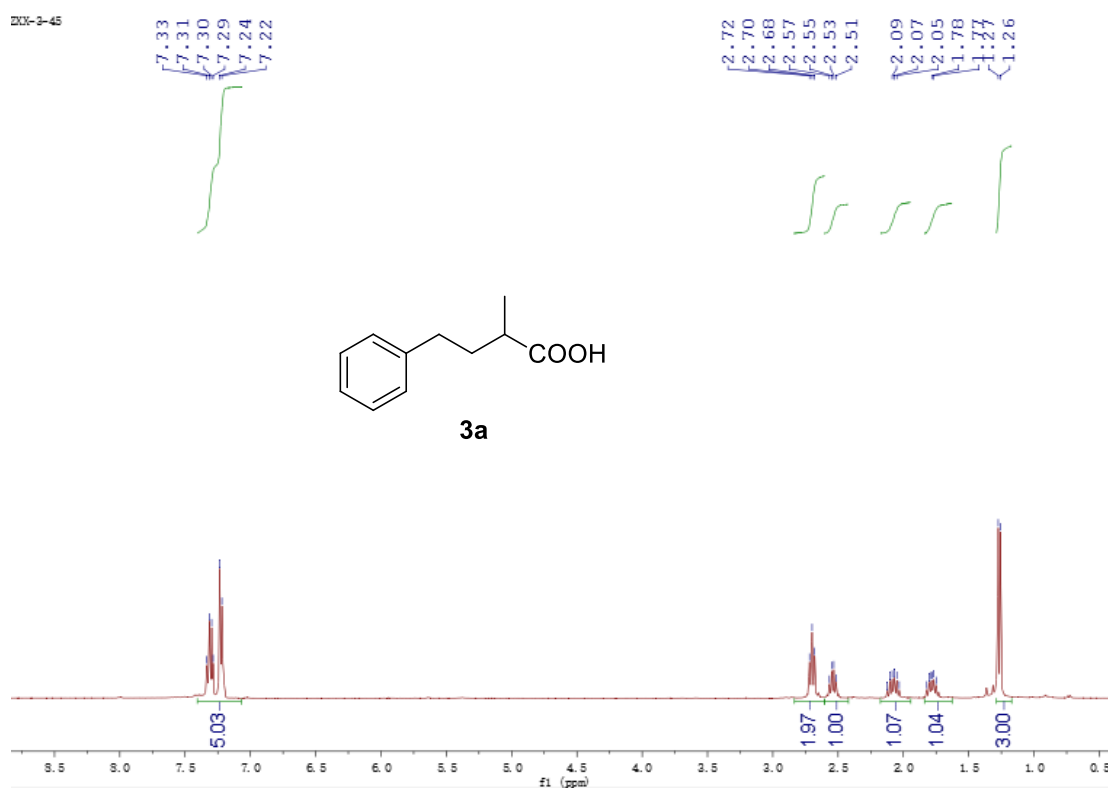
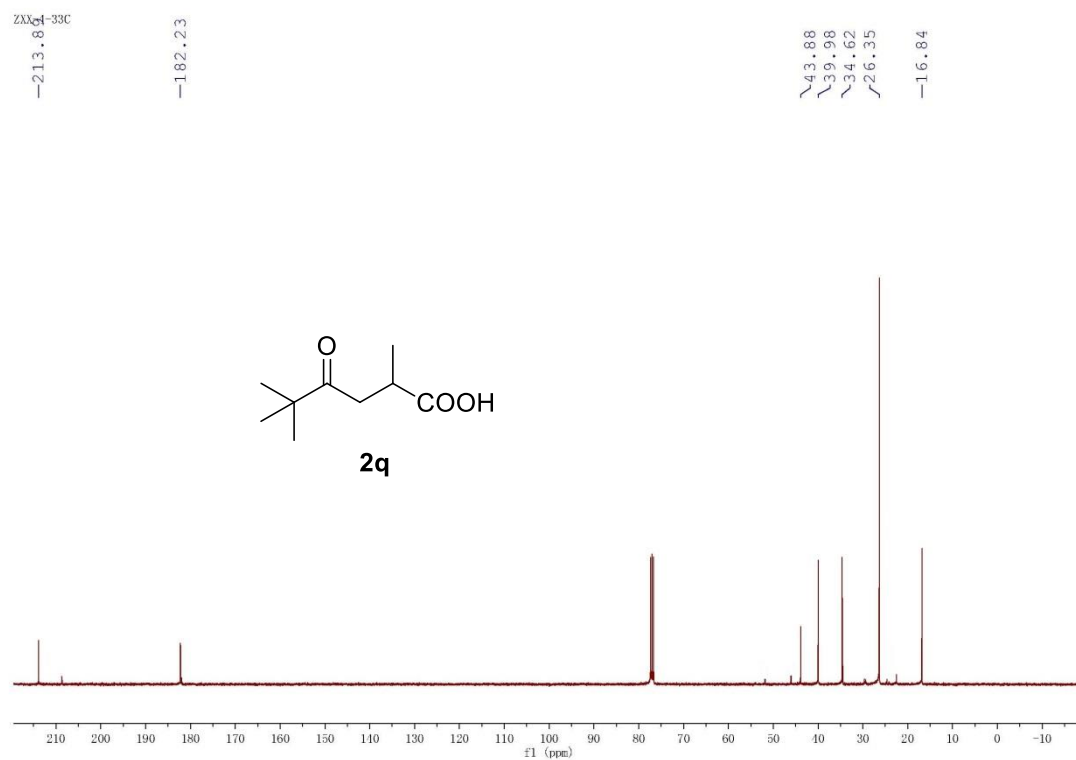


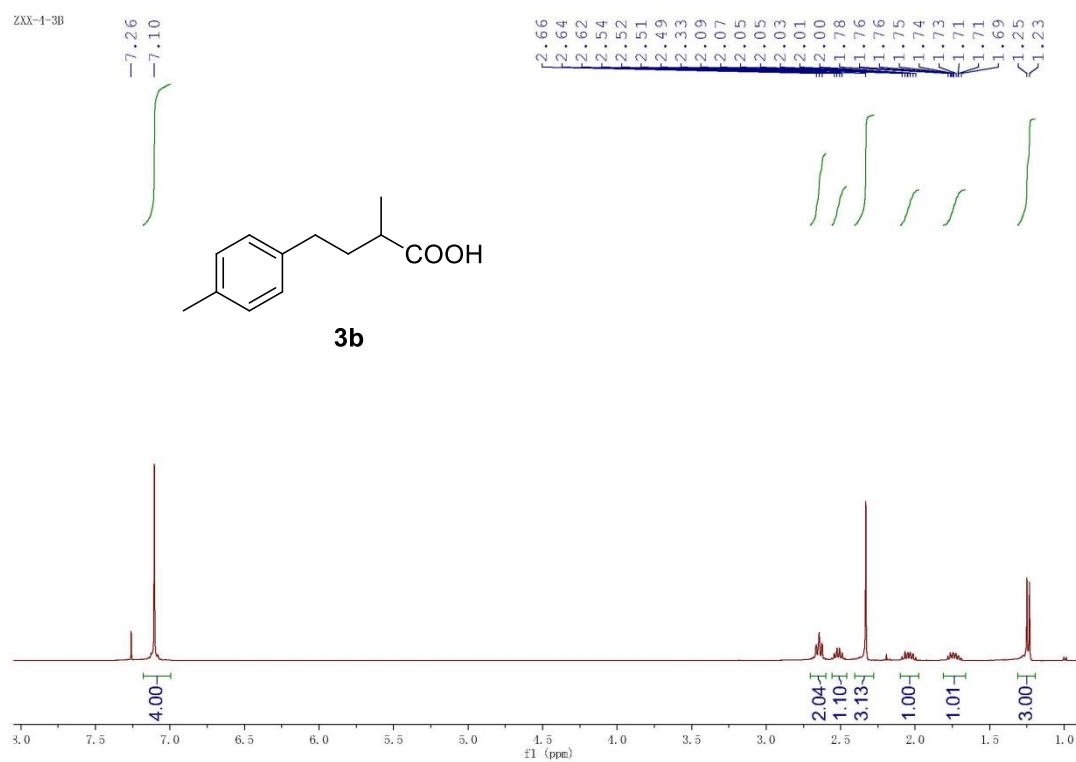
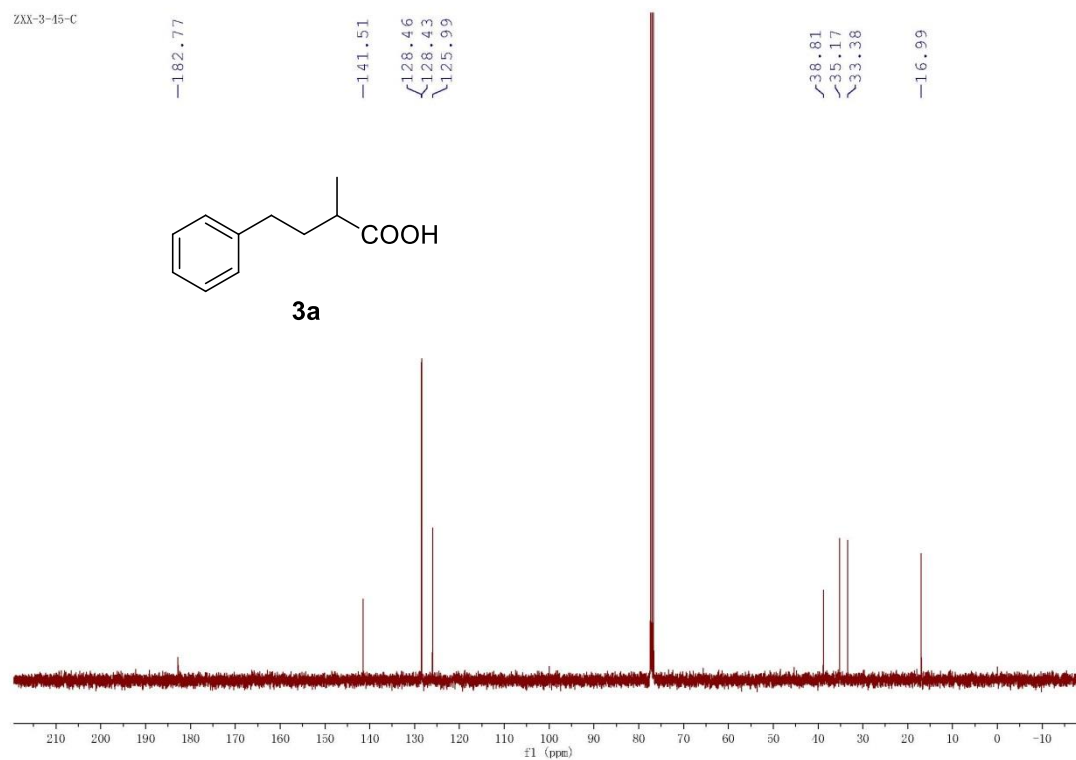
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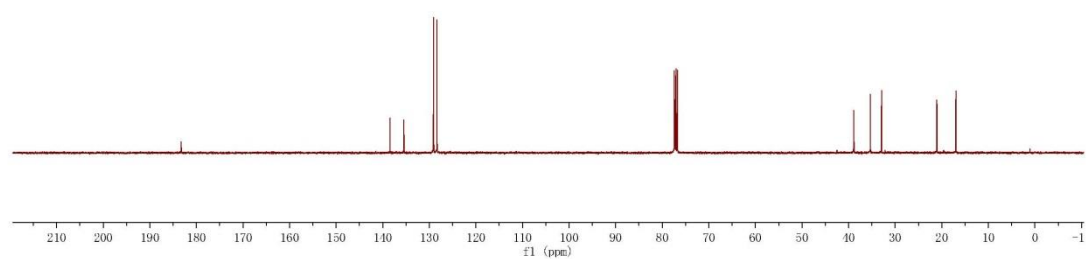
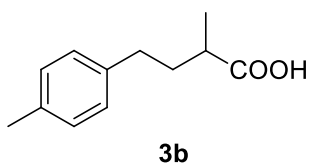






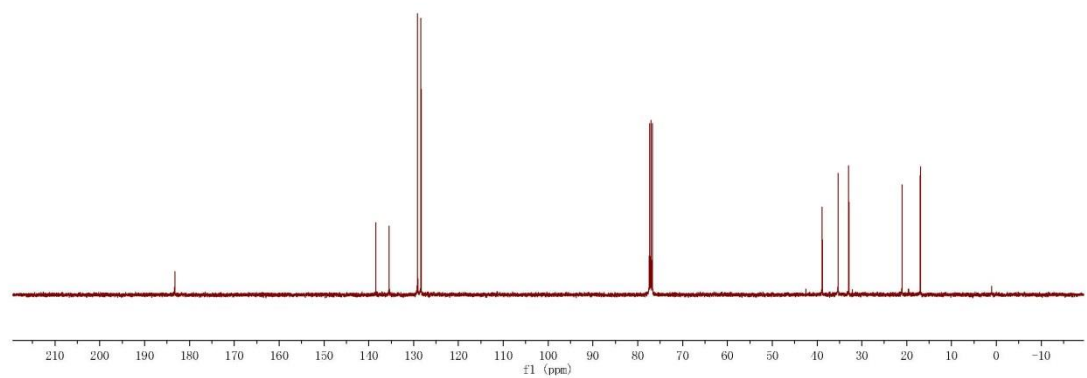
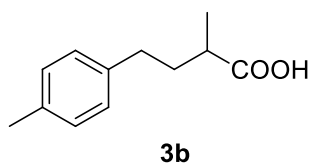
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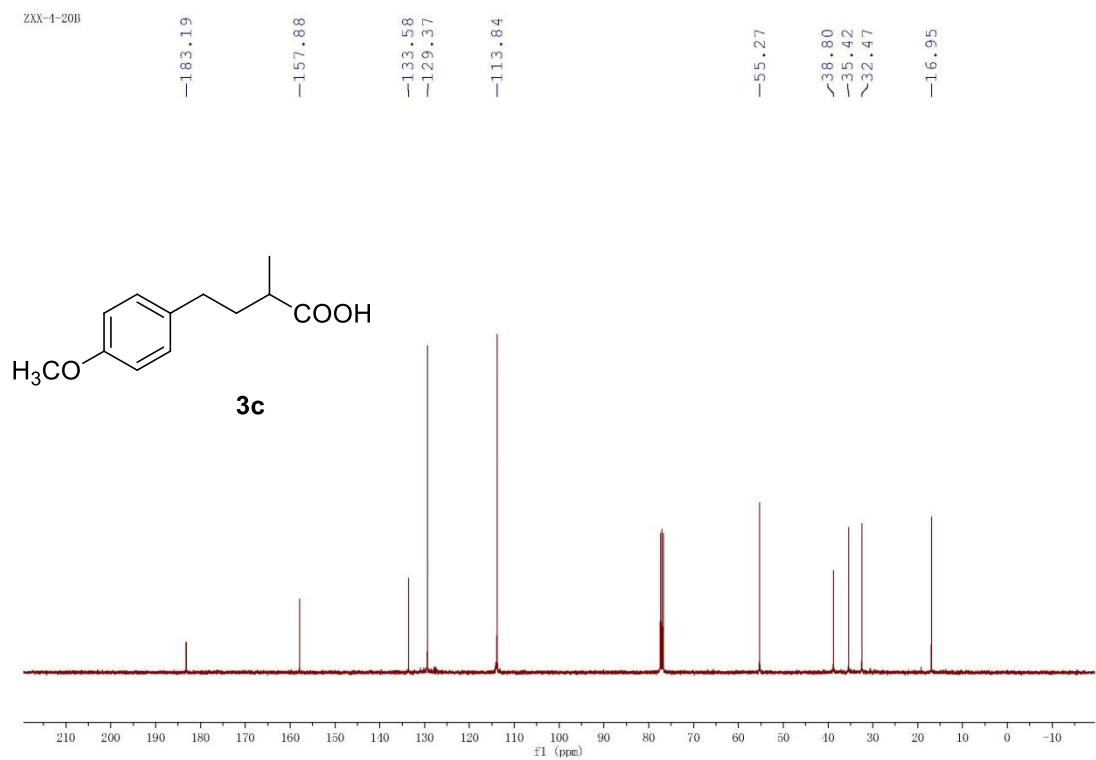
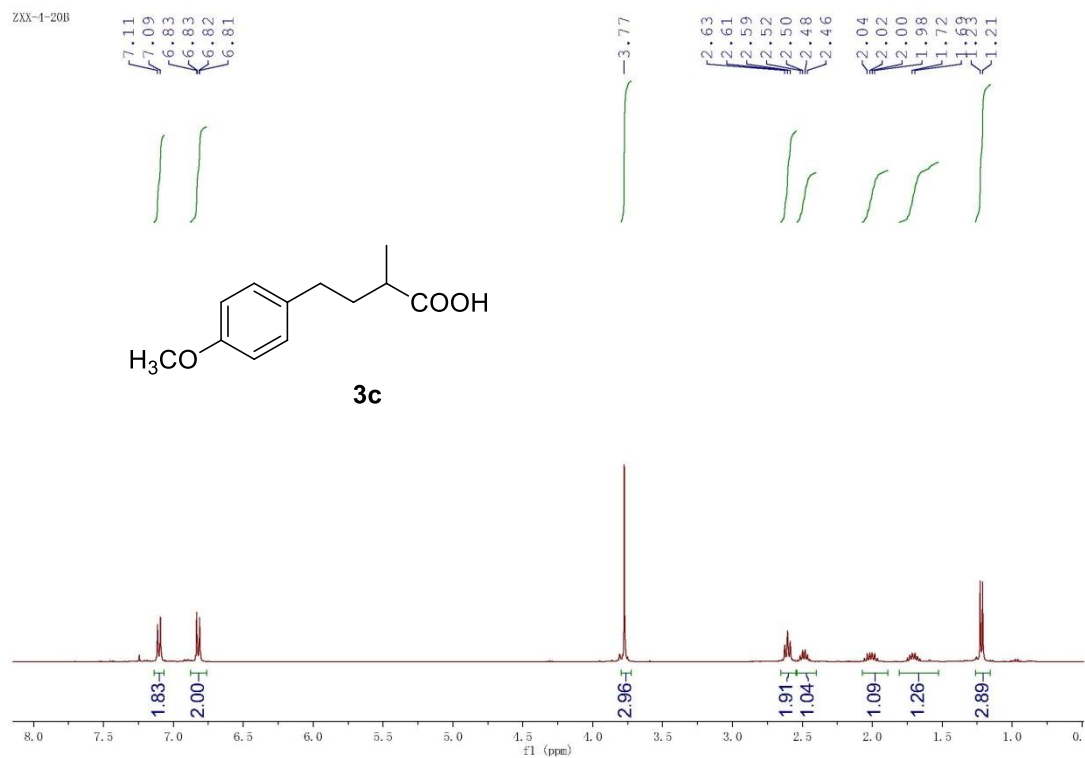
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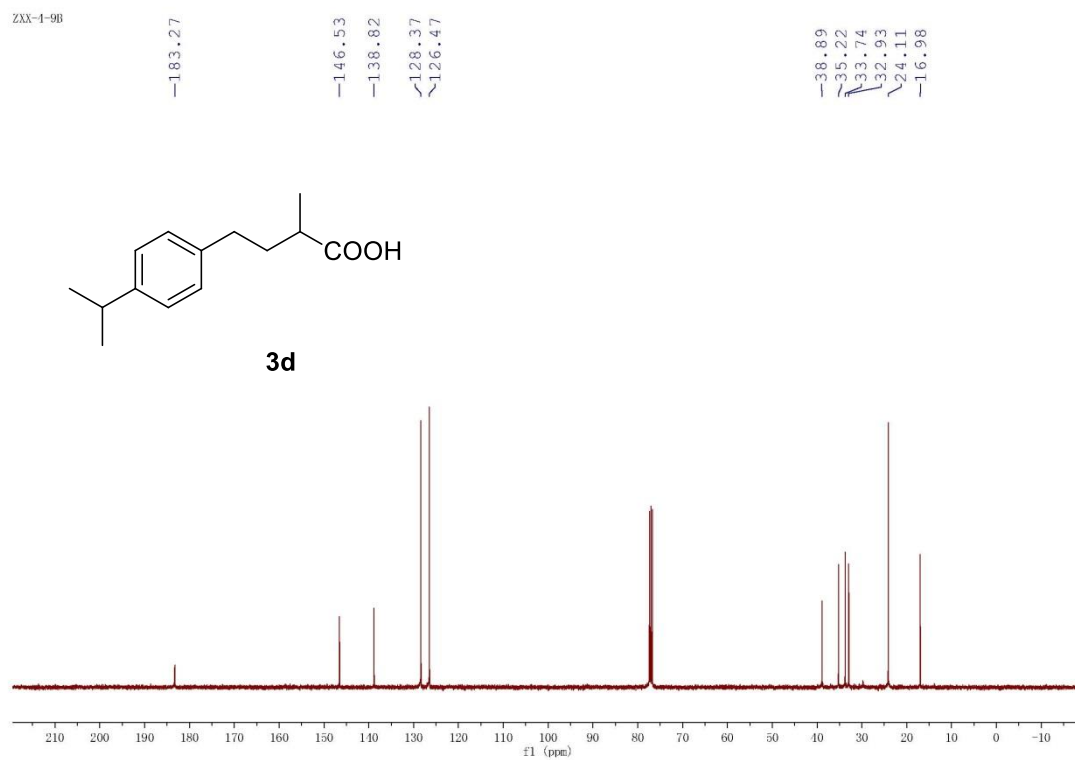
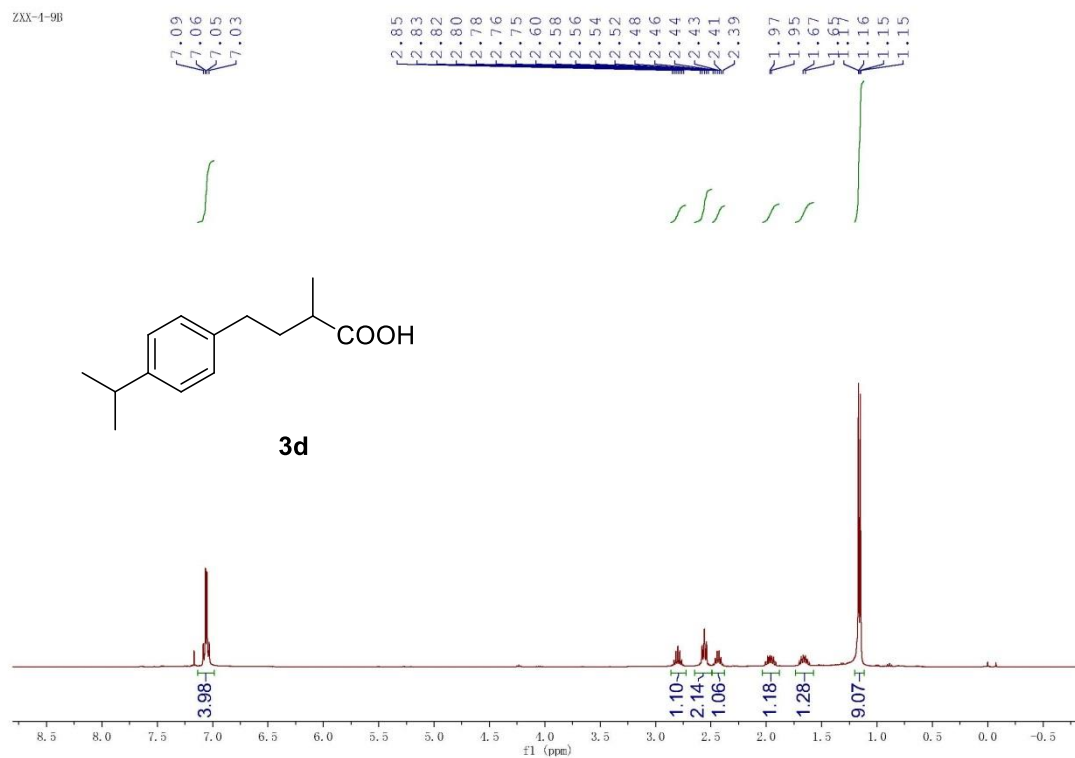
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135.45
129.12
129.01
128.3538.85
35.31
32.93
21.04
16.97

ZXX-1-3B

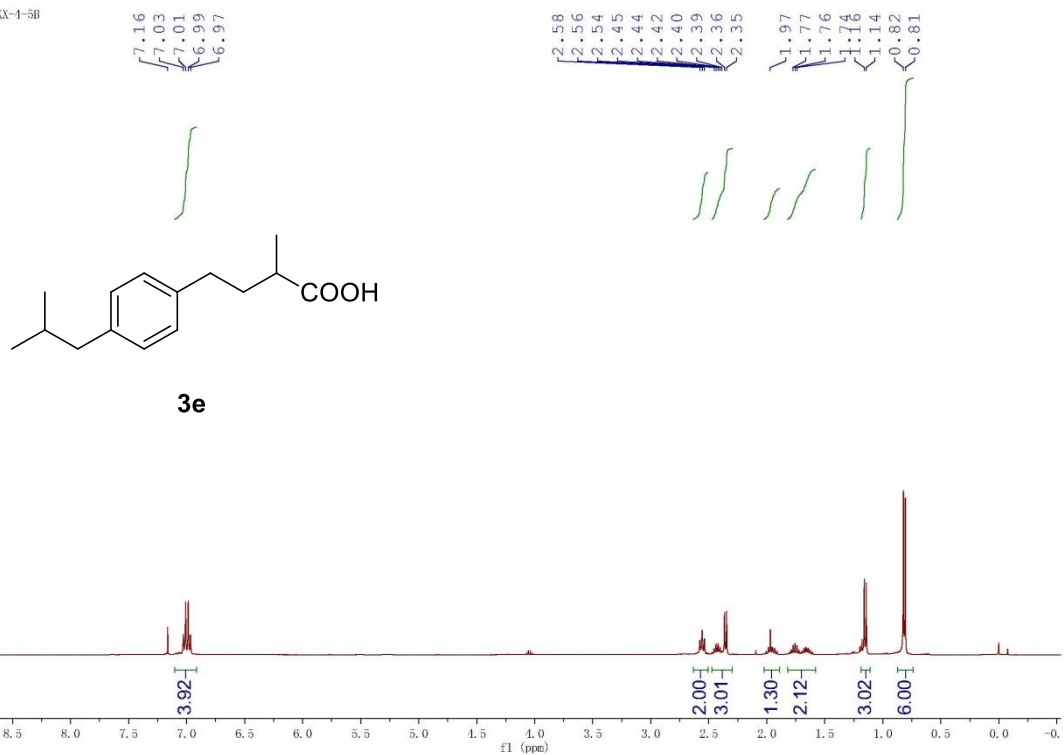
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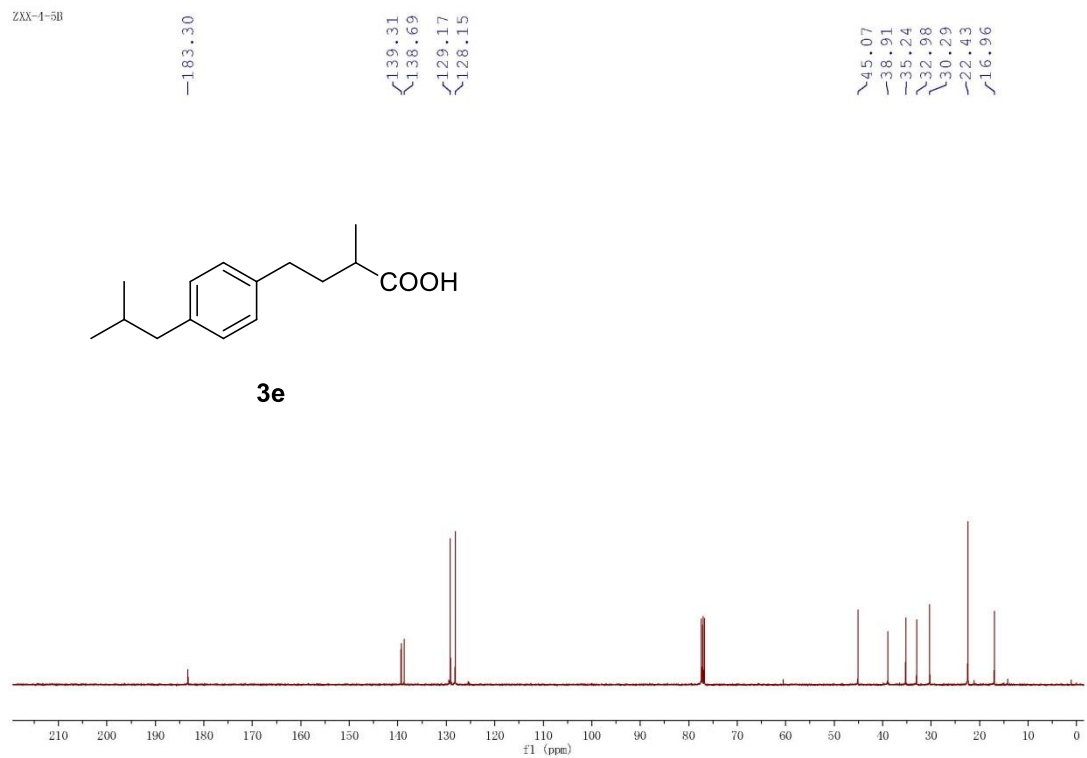


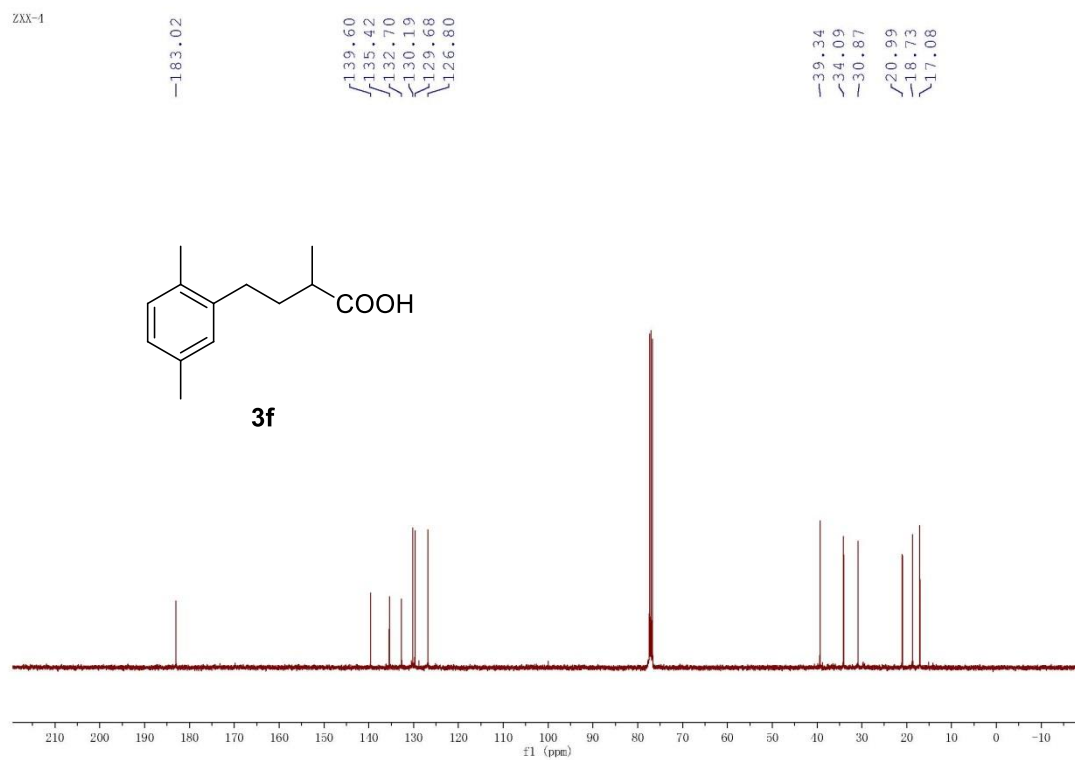
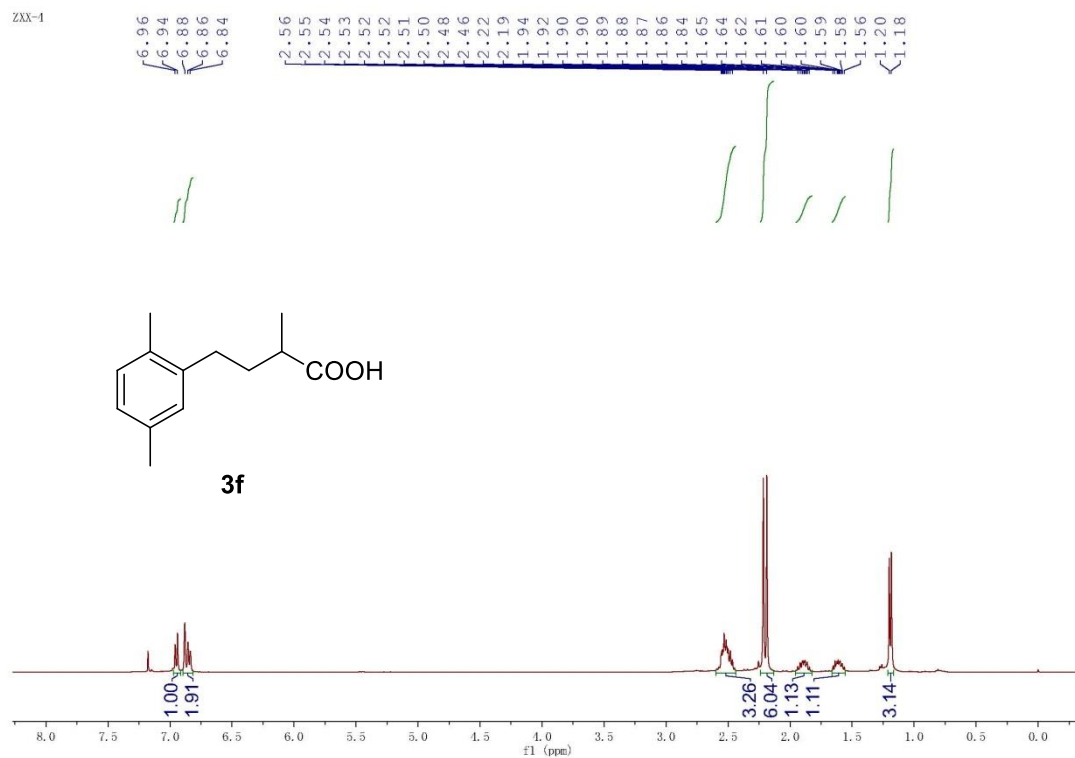


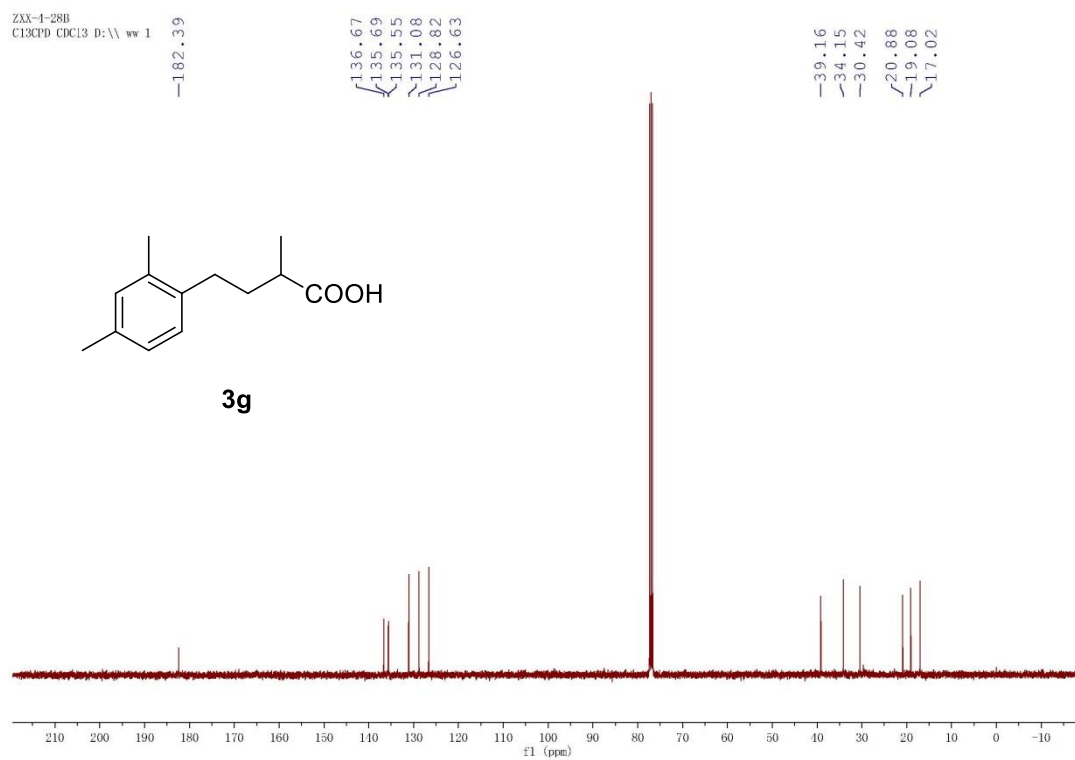
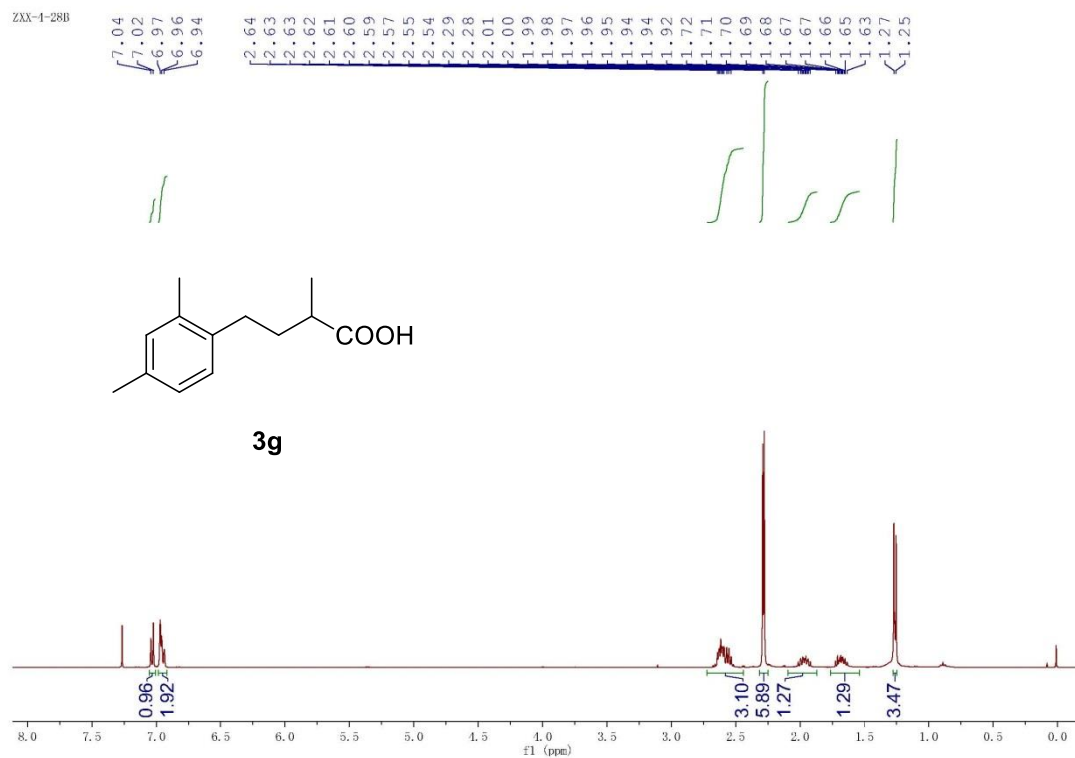
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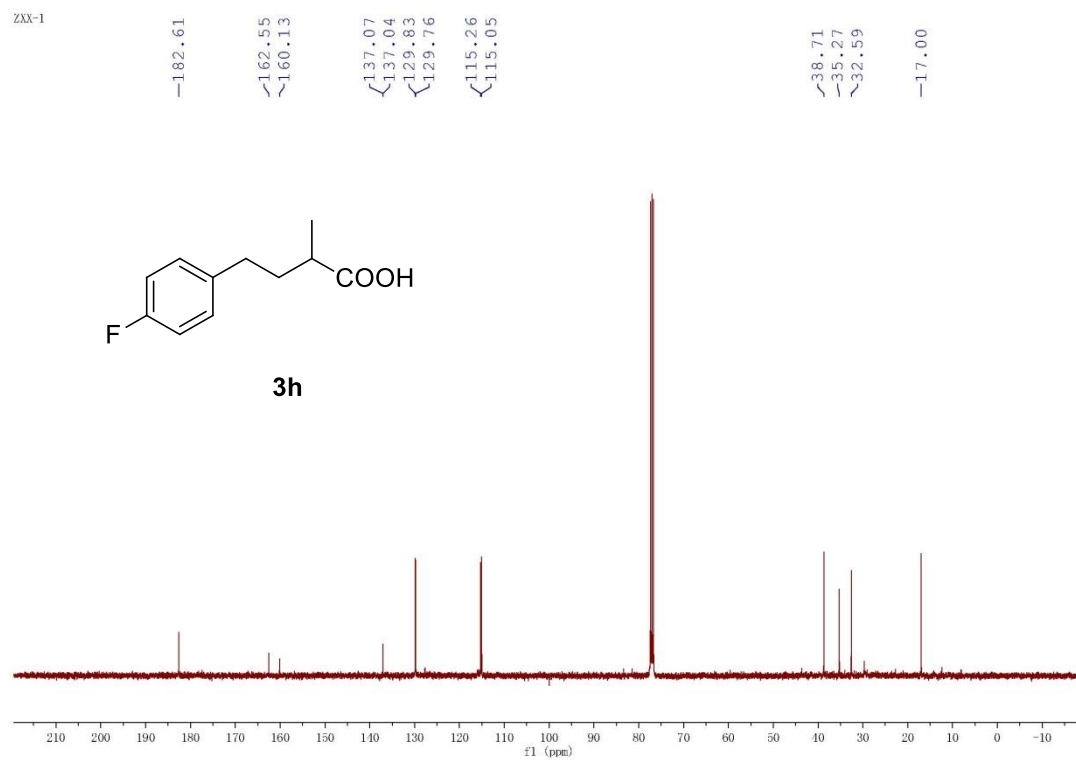
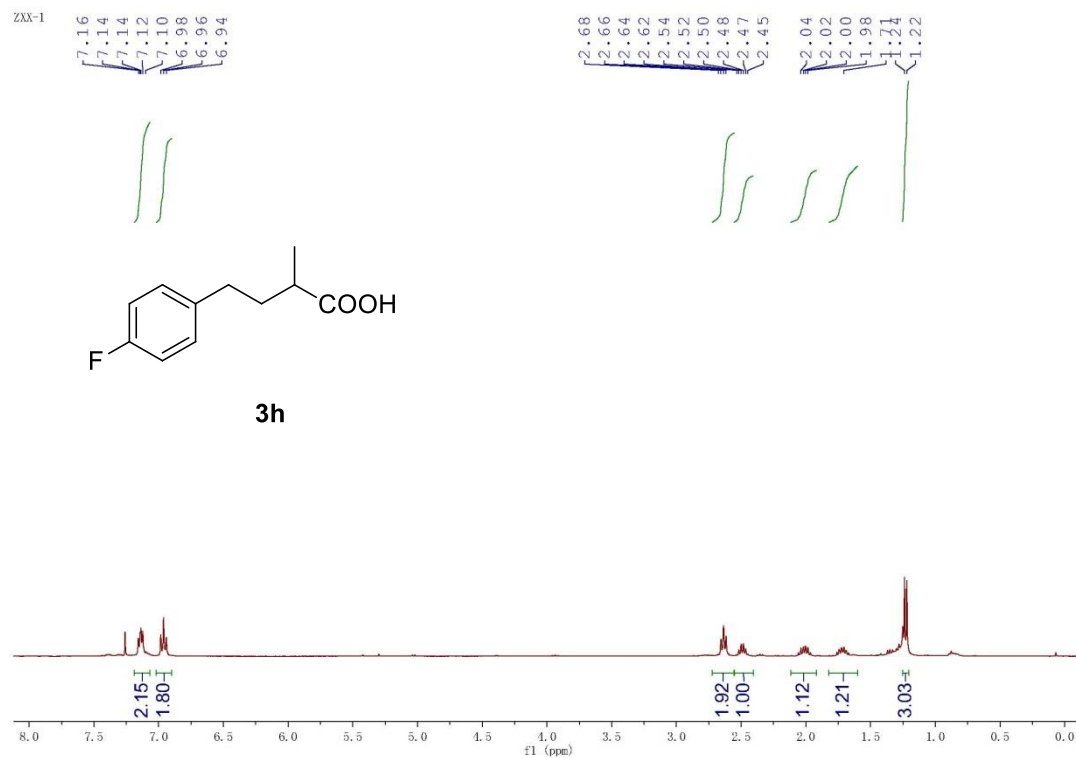


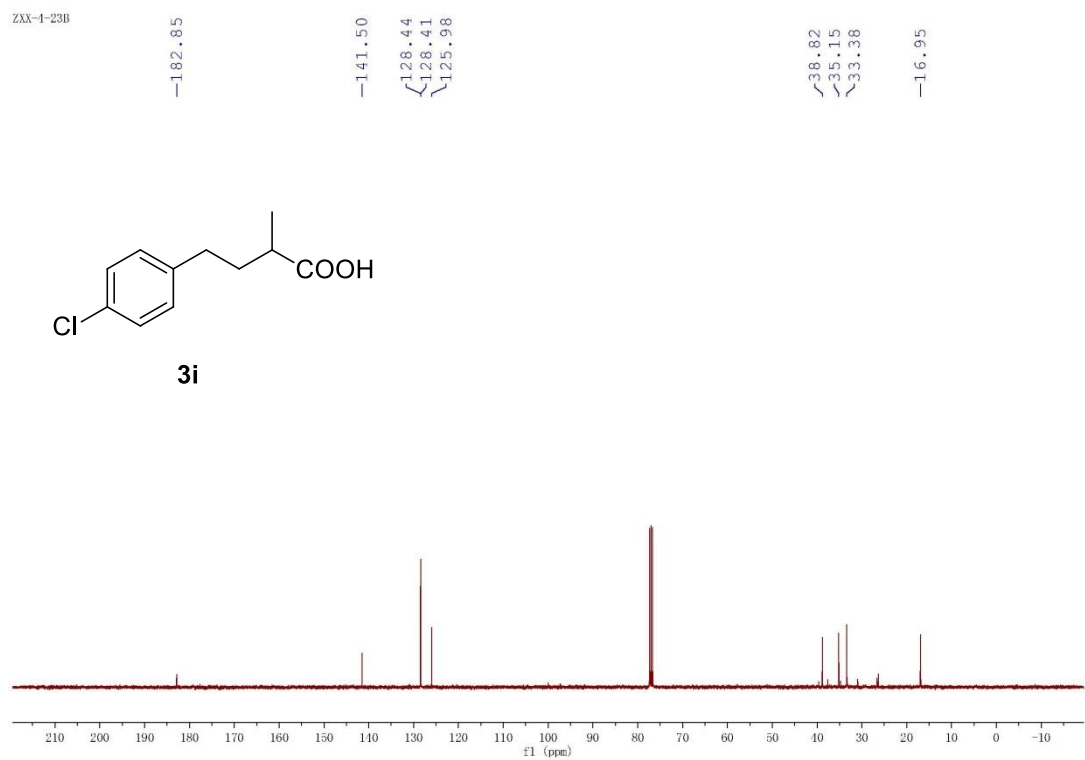
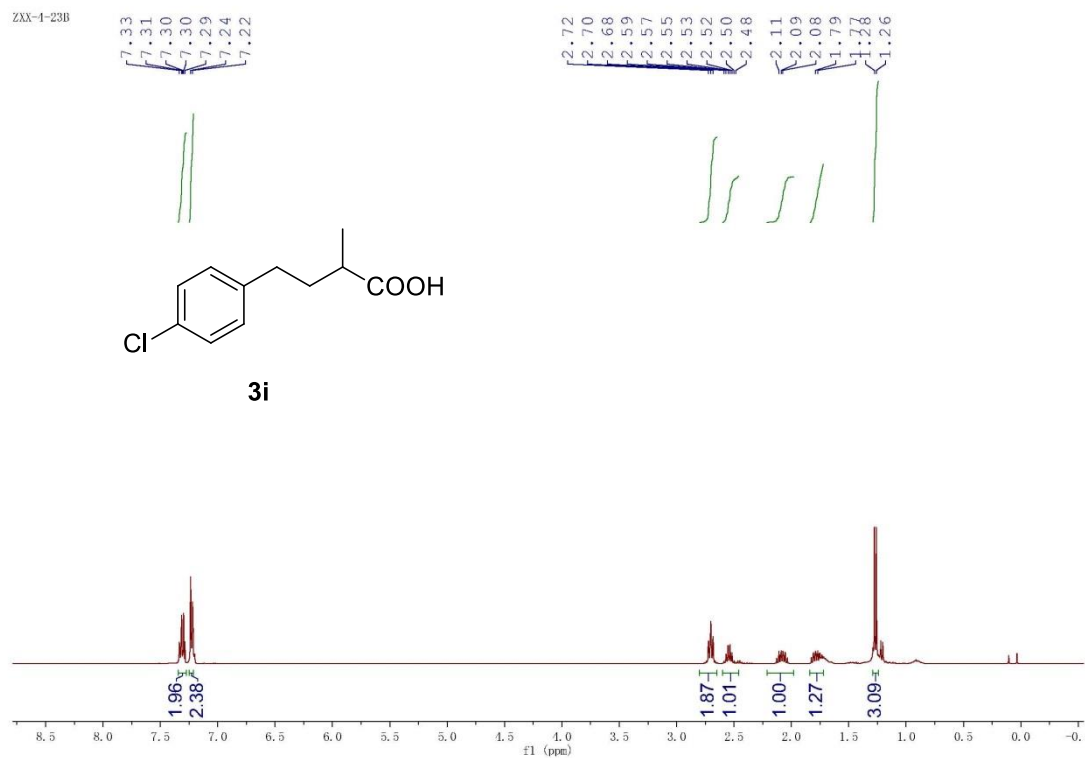
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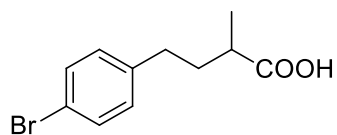
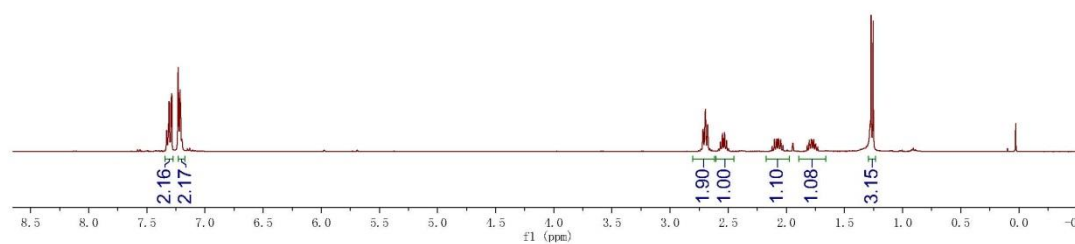




ZXX-1-29

7.33
7.31
7.30
7.29
7.29
7.23
7.21
7.20

2.72
2.70
2.68
2.58
2.57
2.55
2.53
2.51
2.50
2.10
2.08
2.07
1.79
1.29
1.25

**3j**

ZXX-1-29

-182.26

-141.50

128.43

128.41

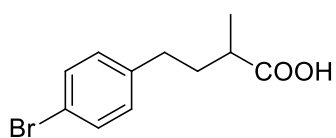
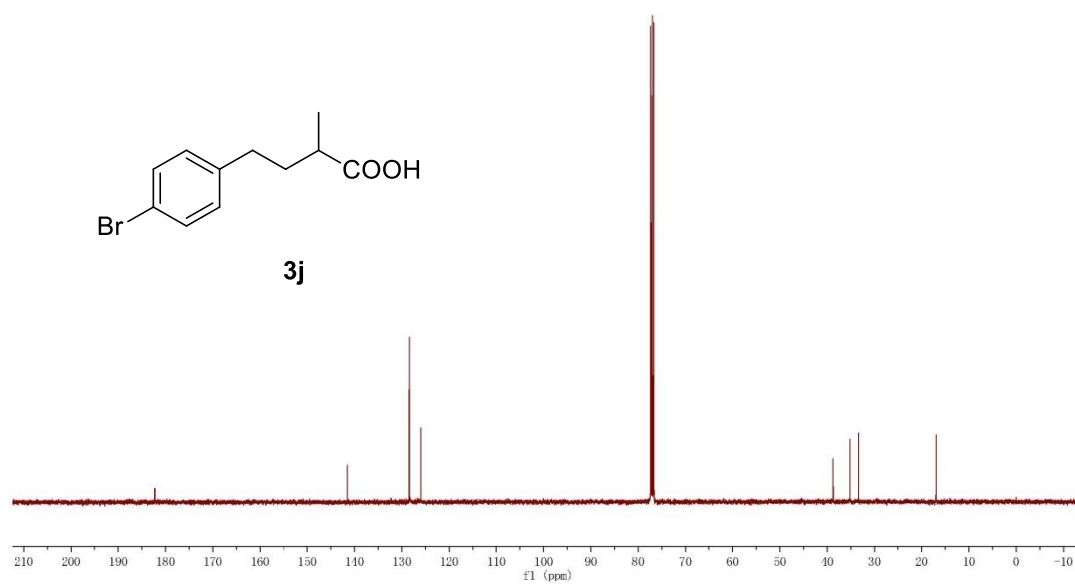
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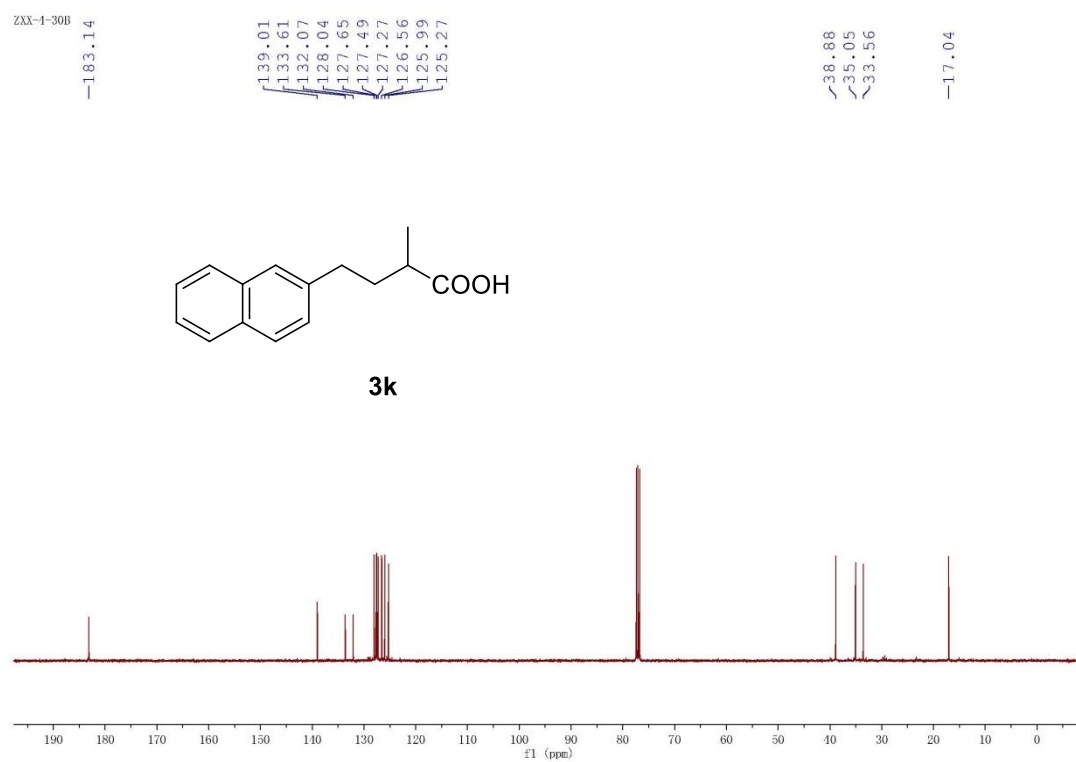
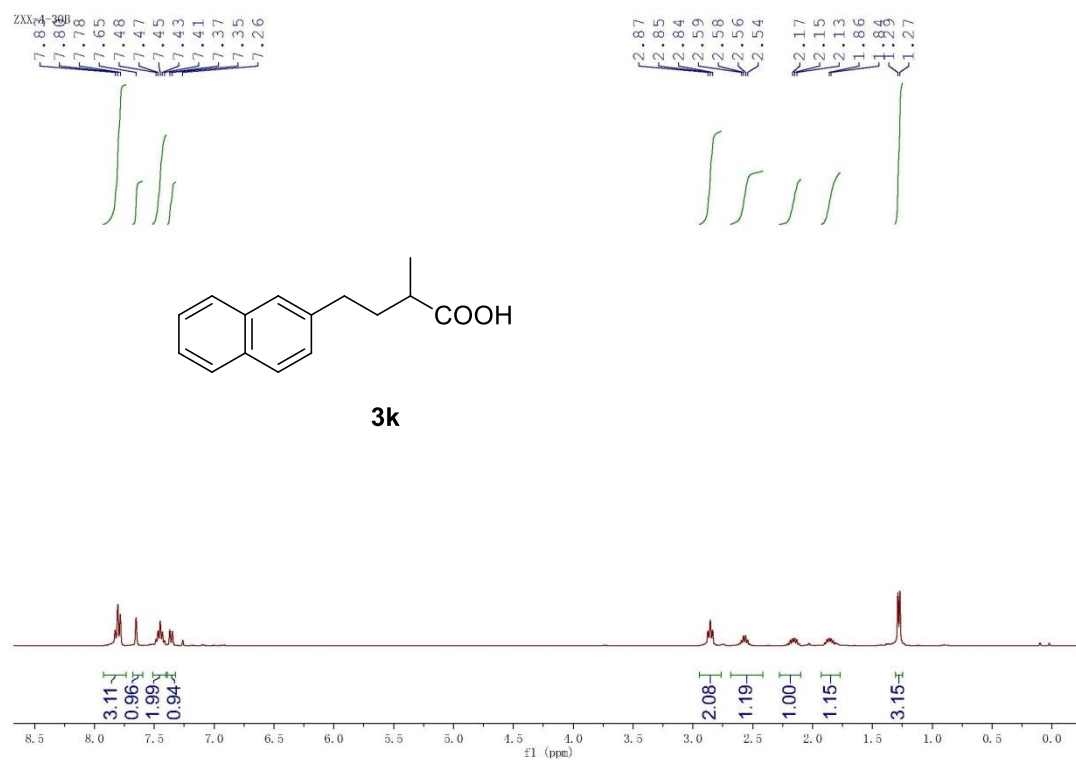
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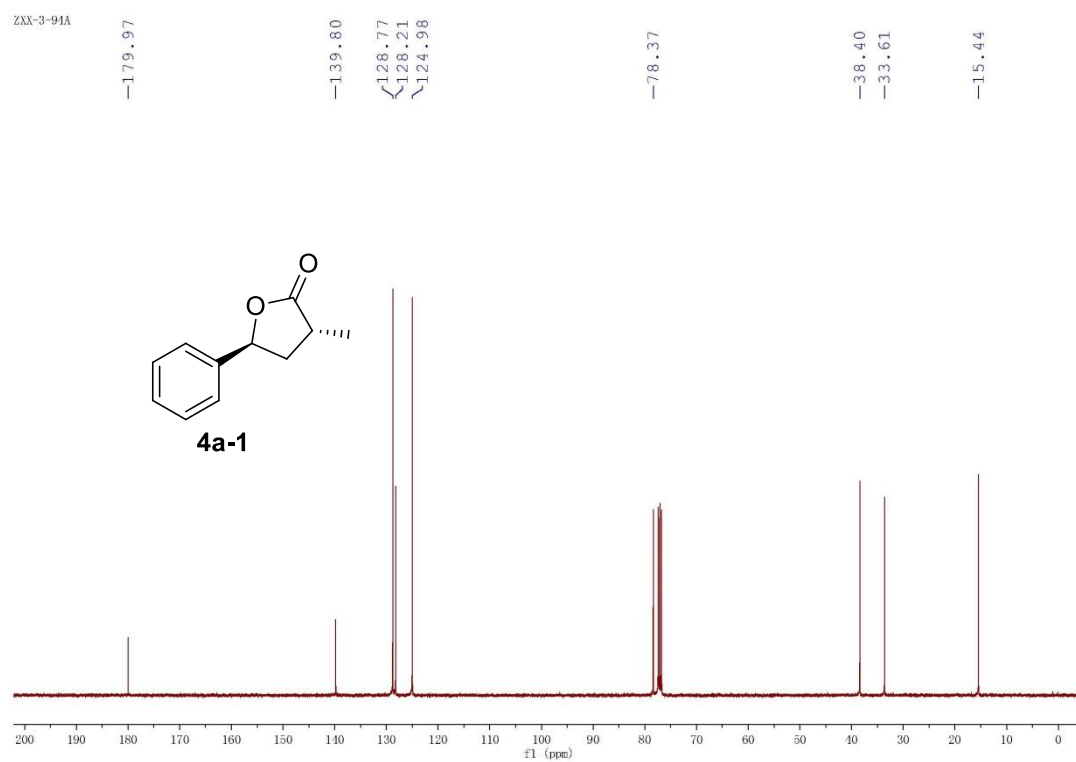
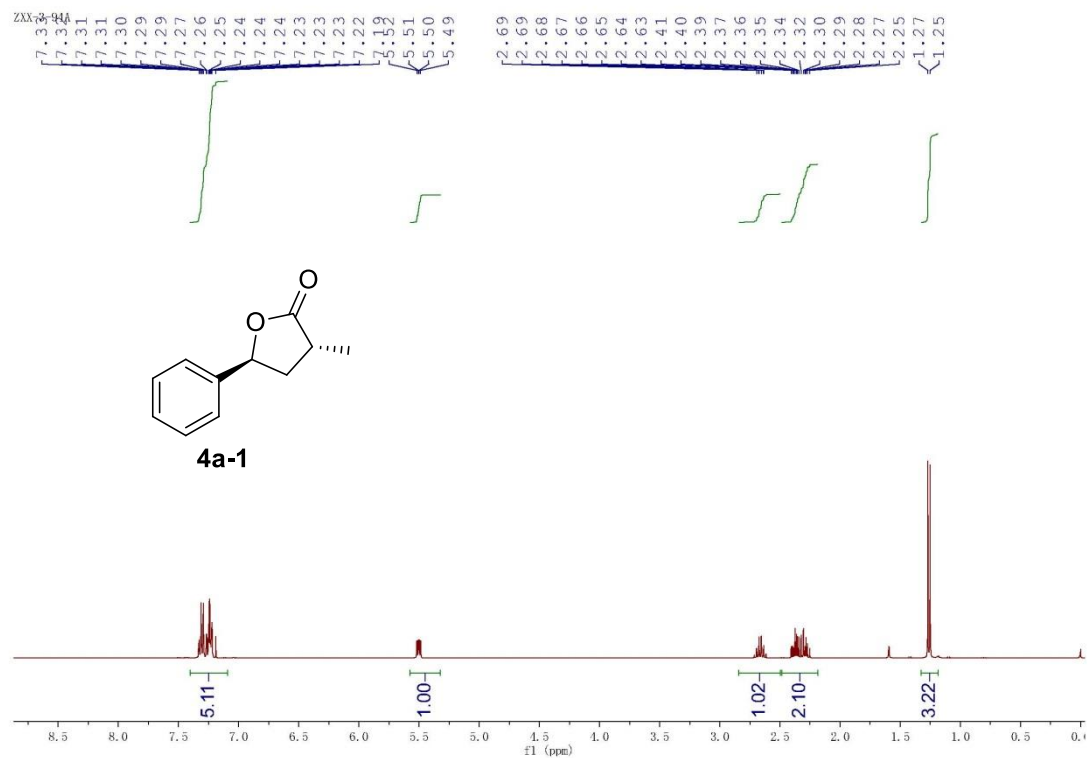
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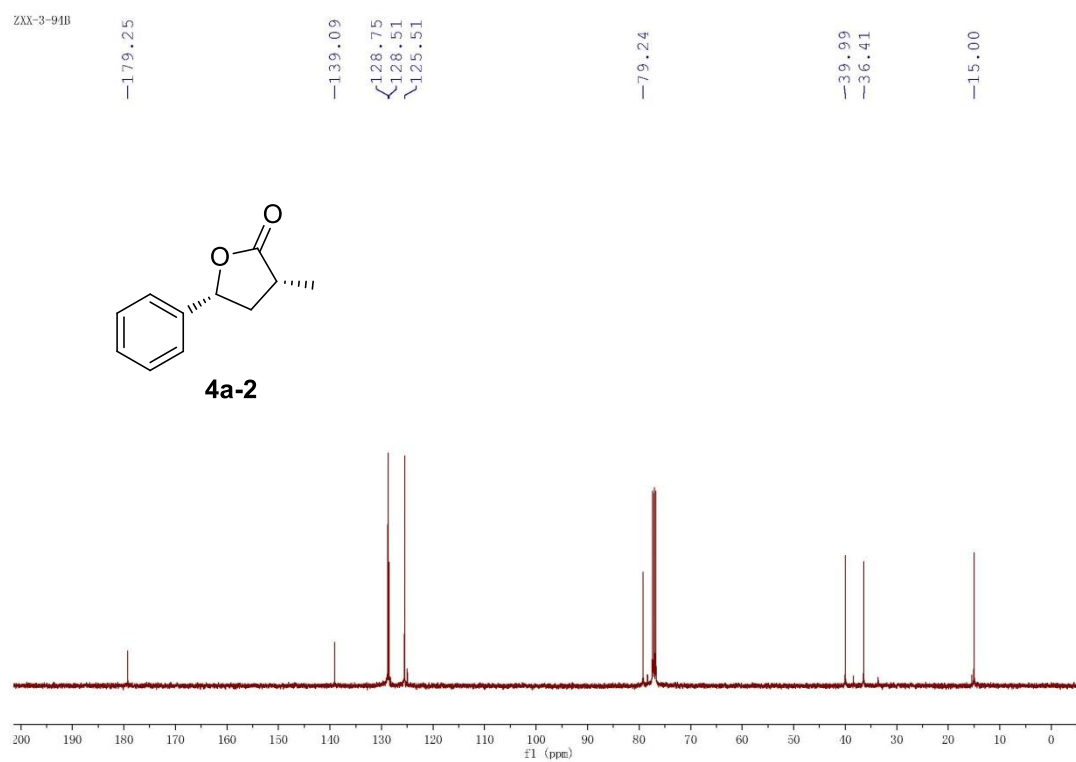
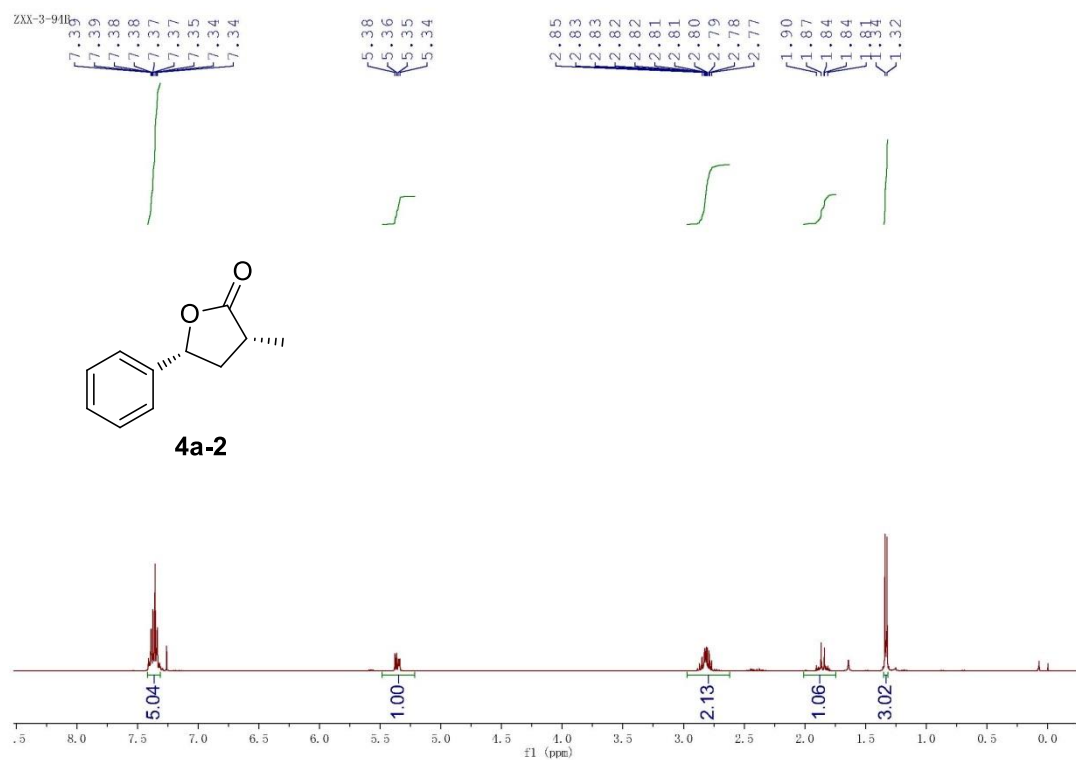
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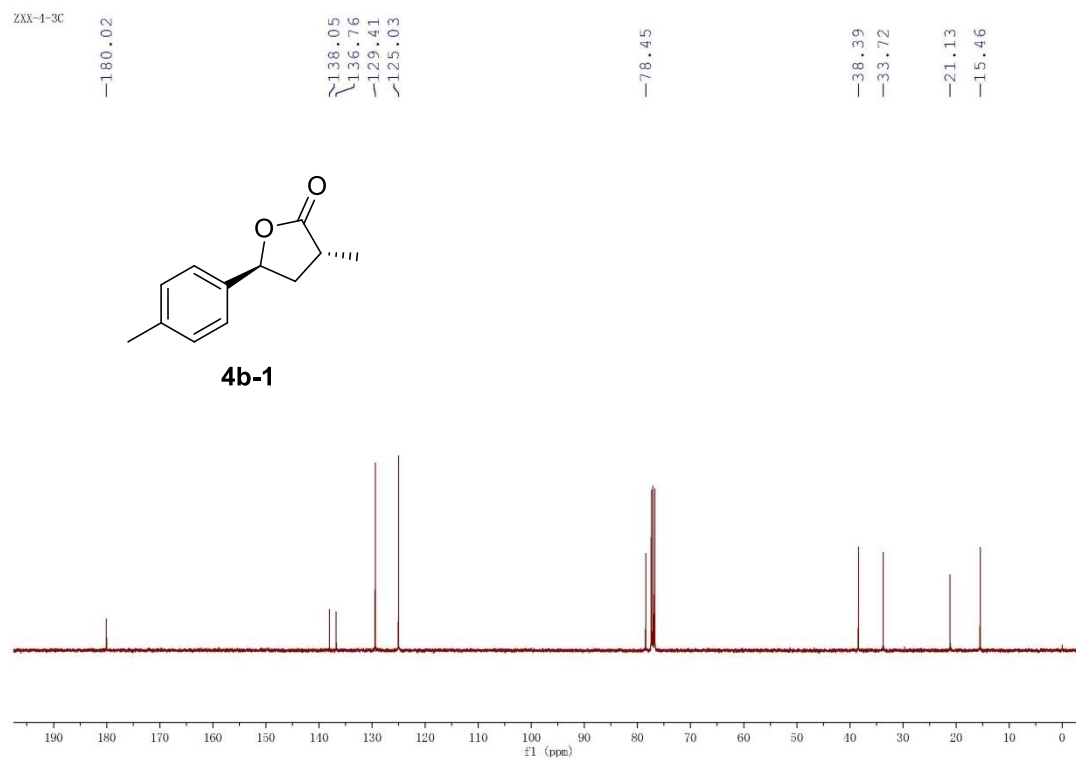
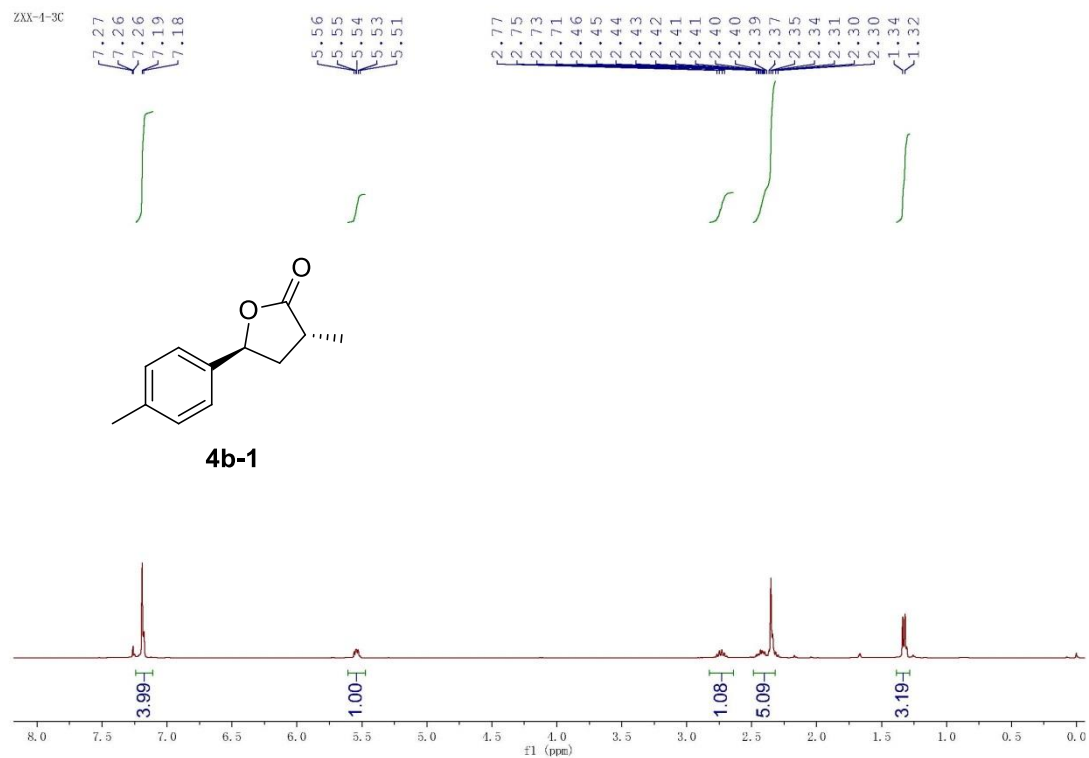
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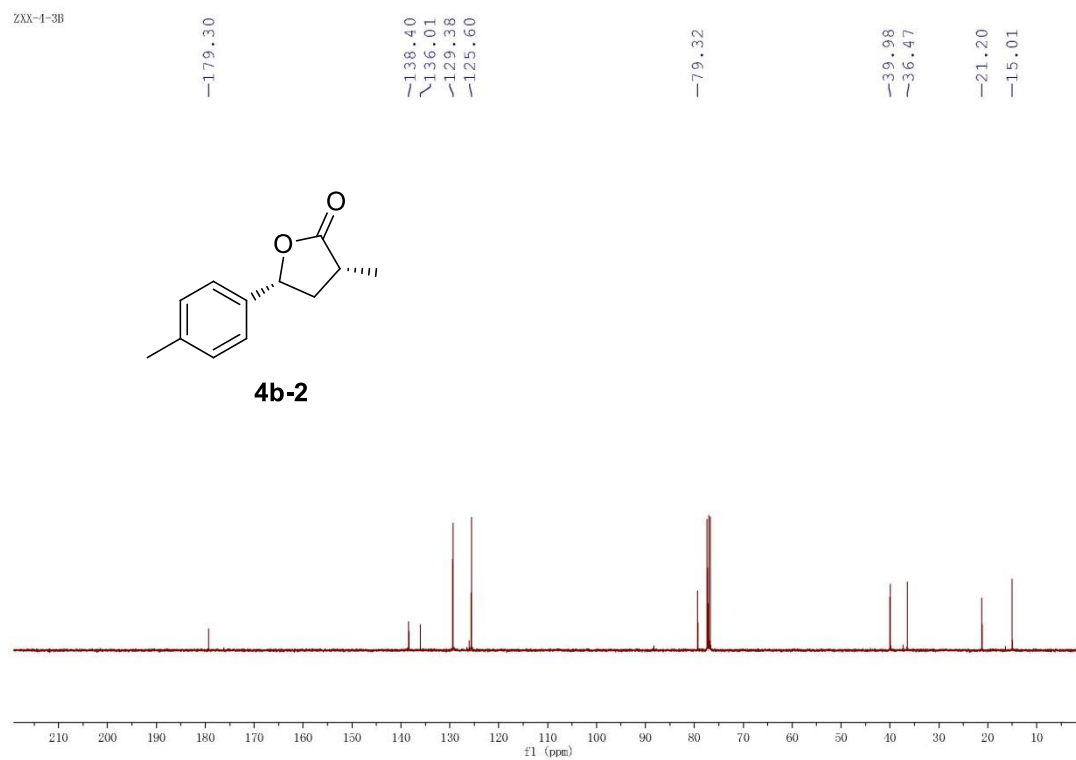
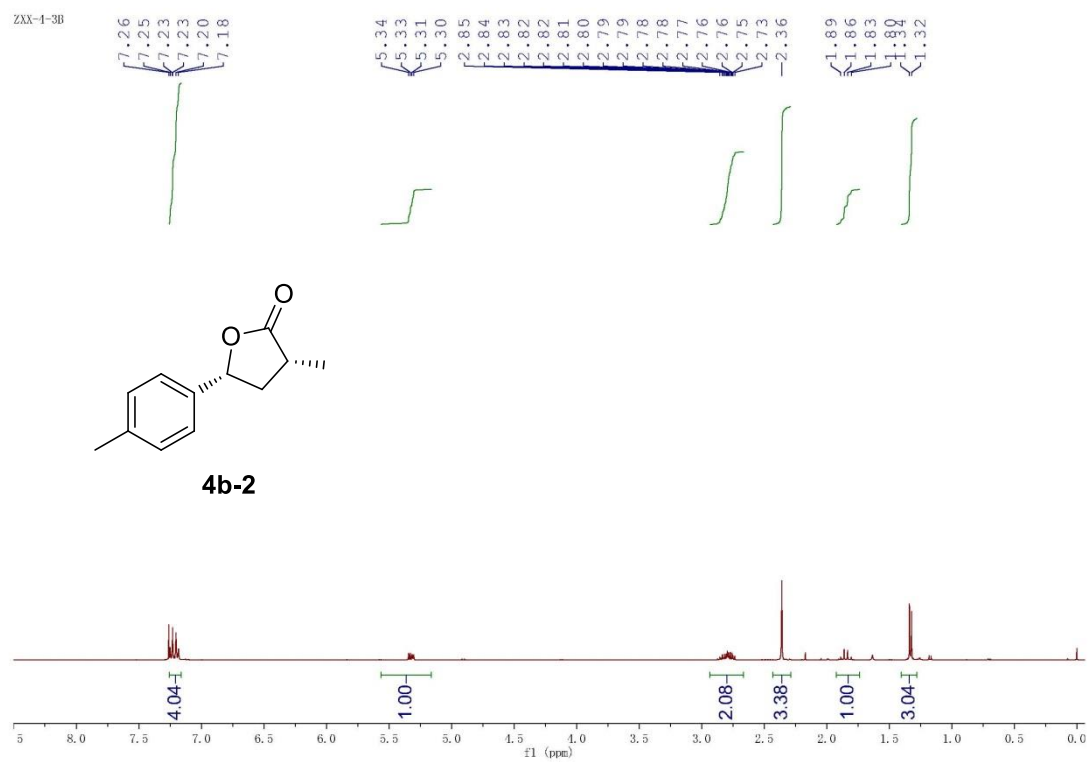
**3j**



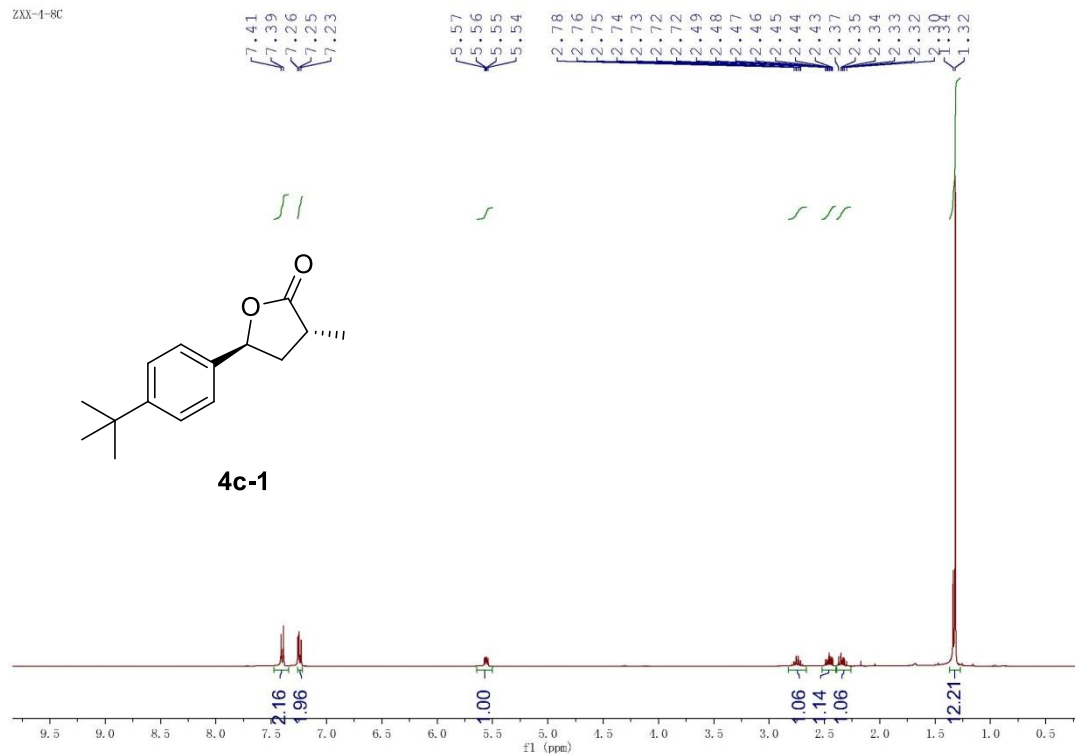




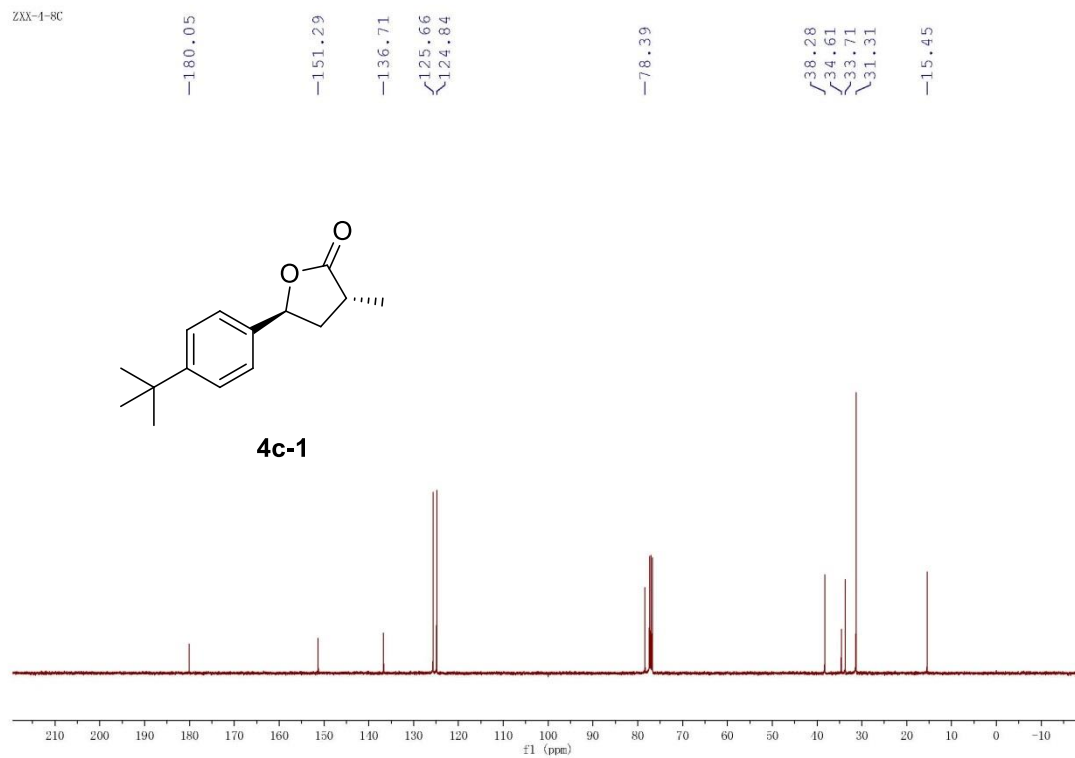


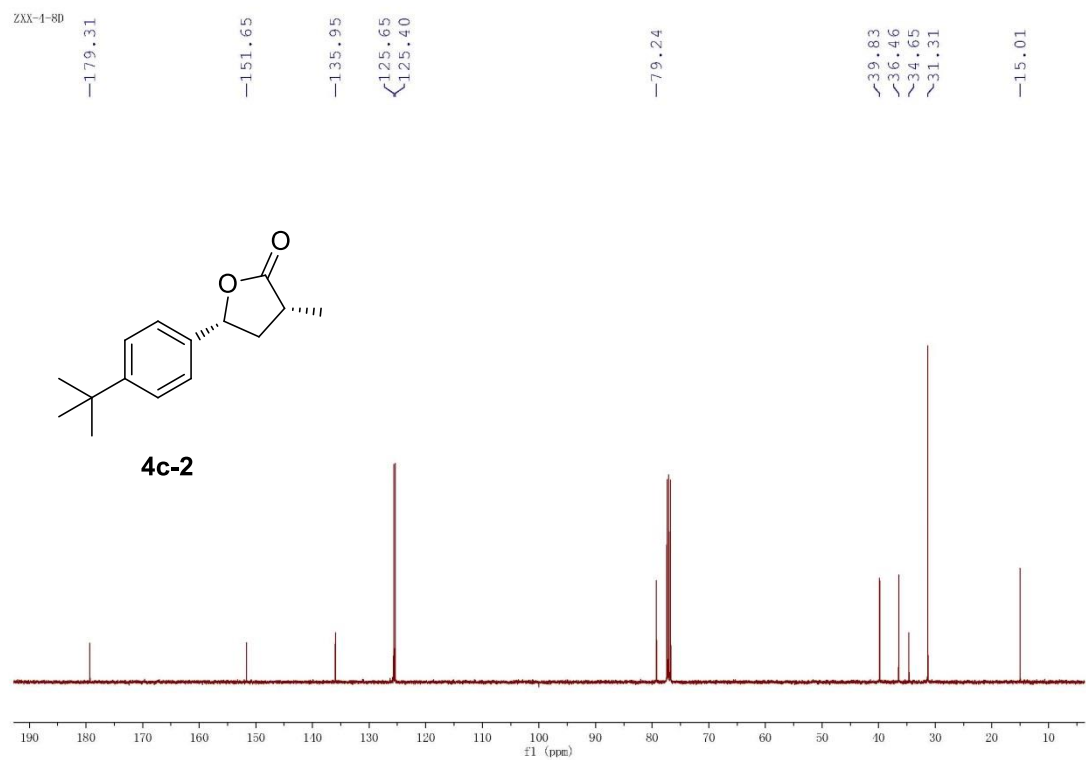
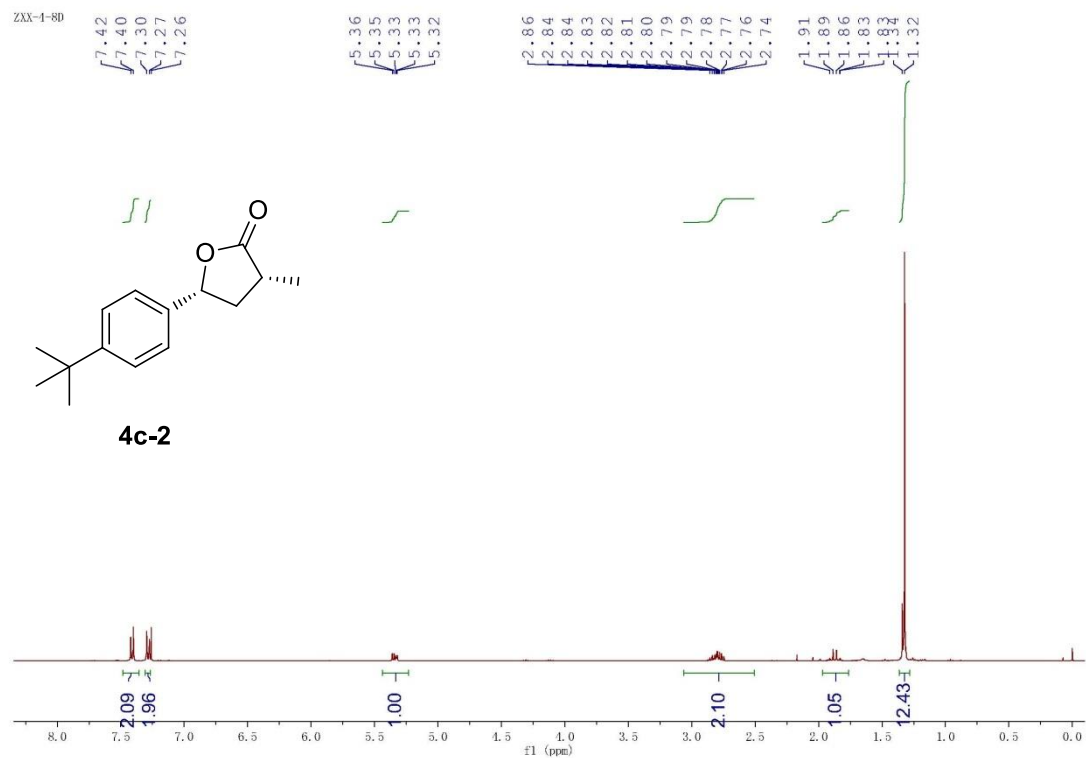


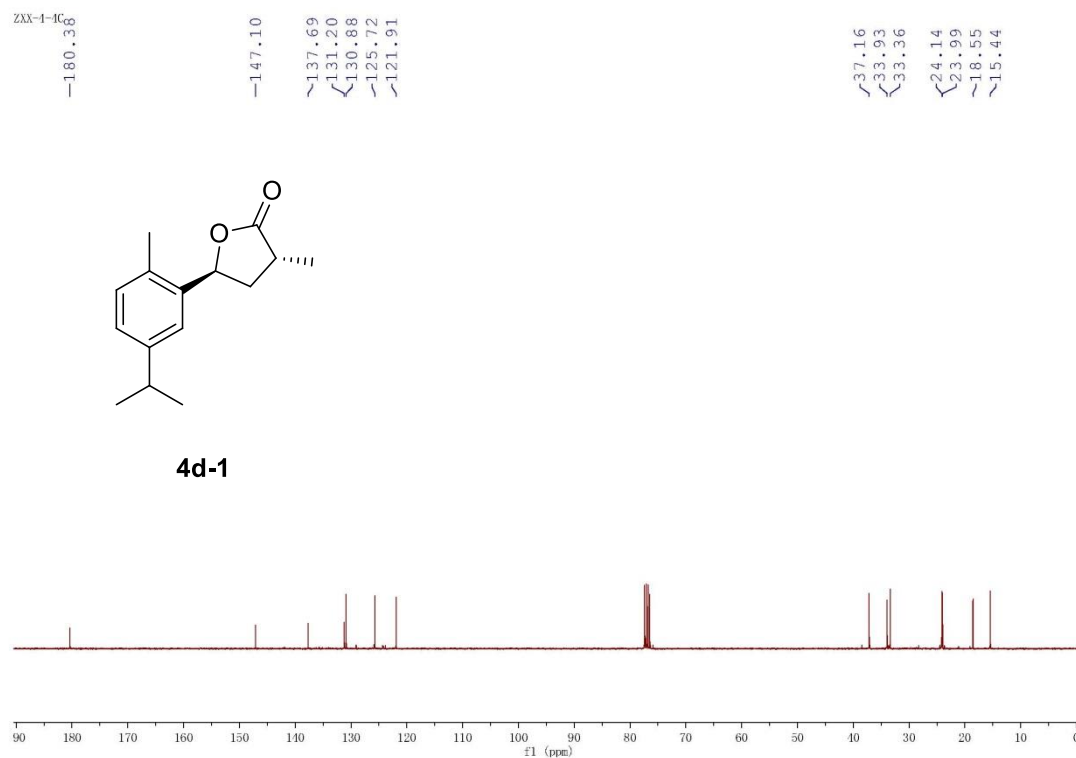
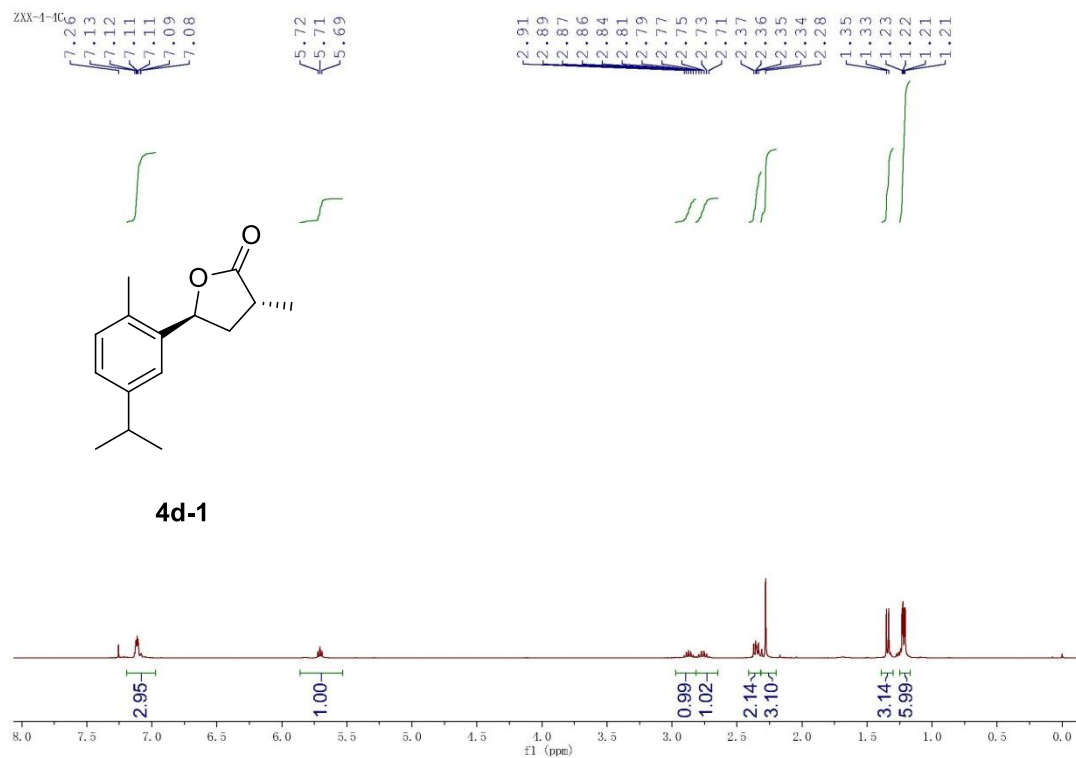
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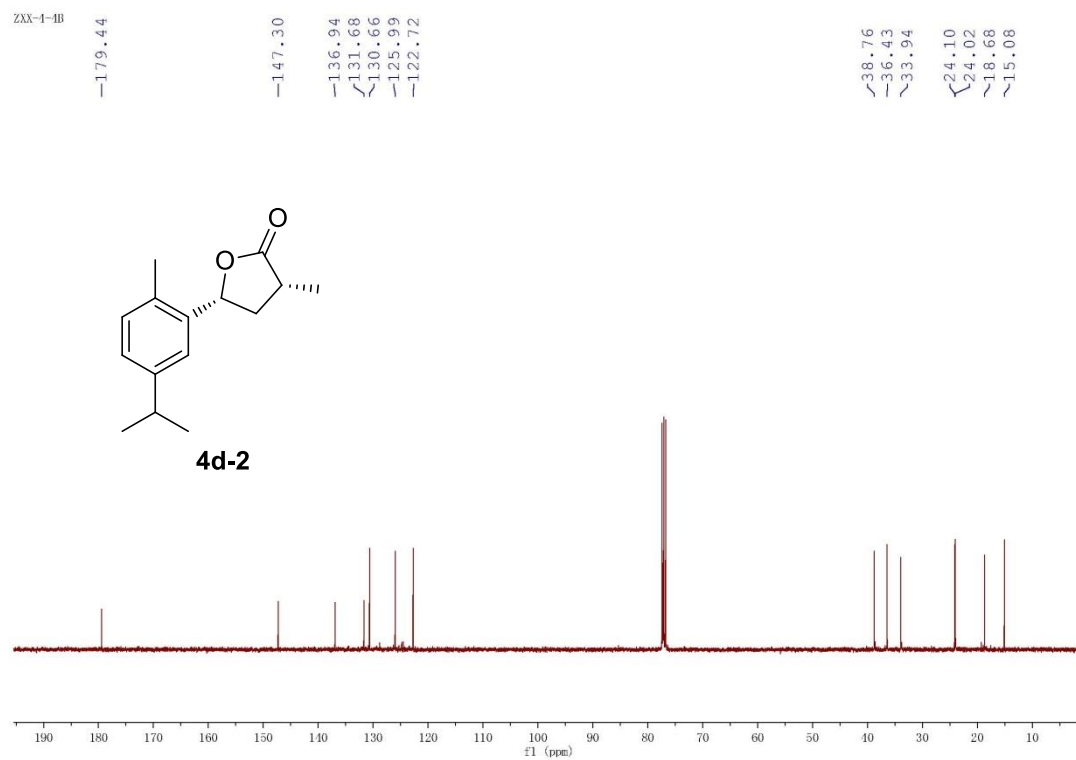
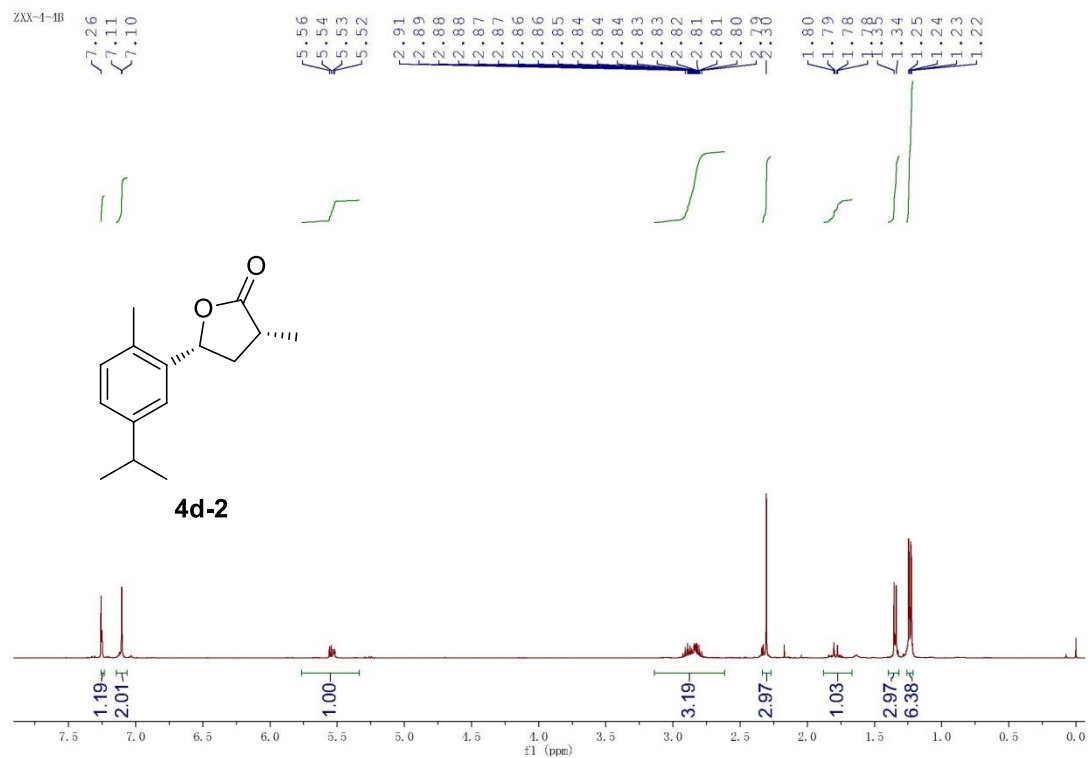


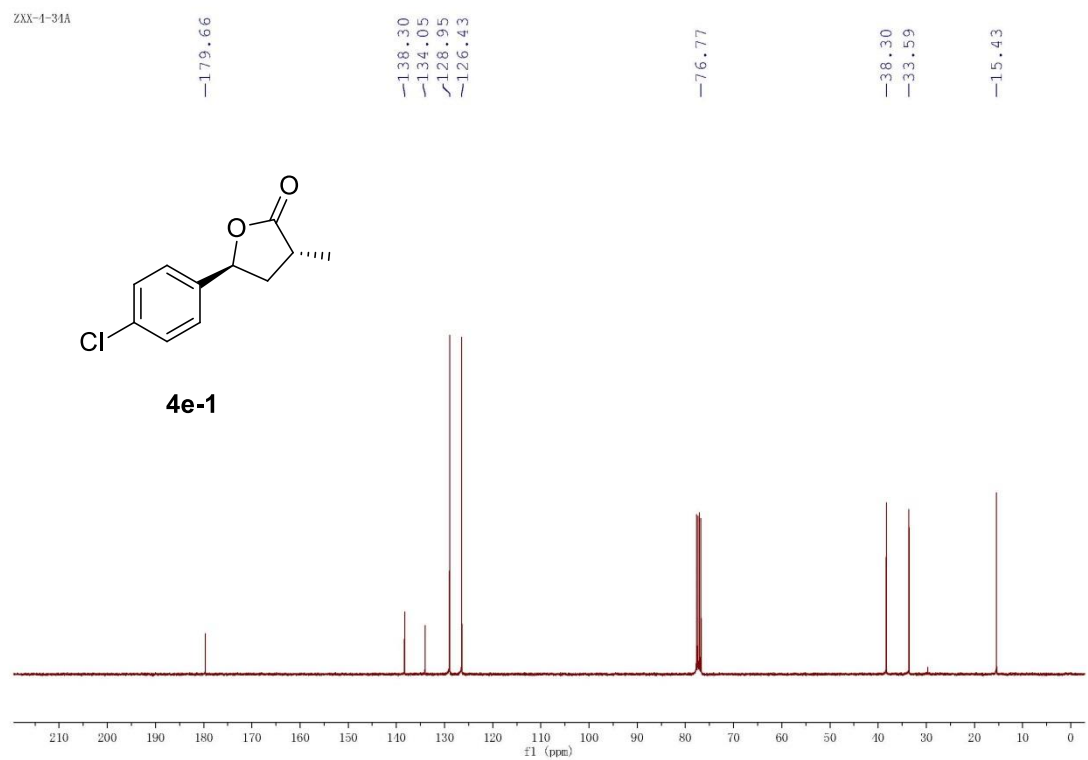
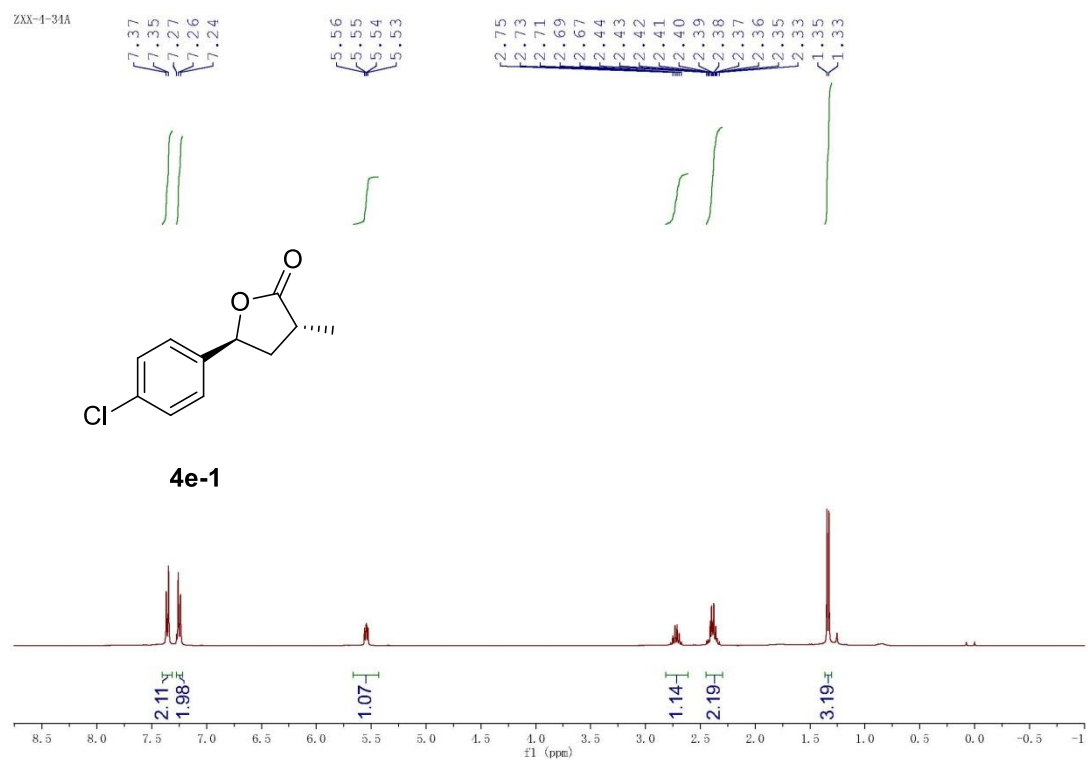
ZXX-1-8C



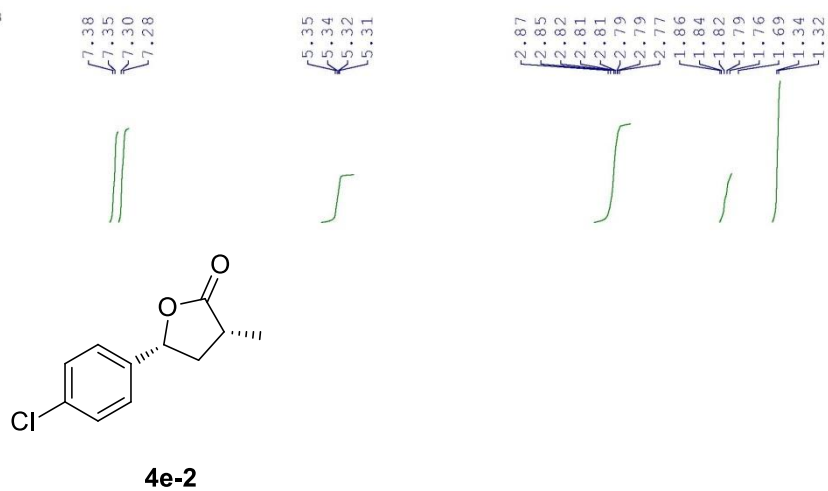




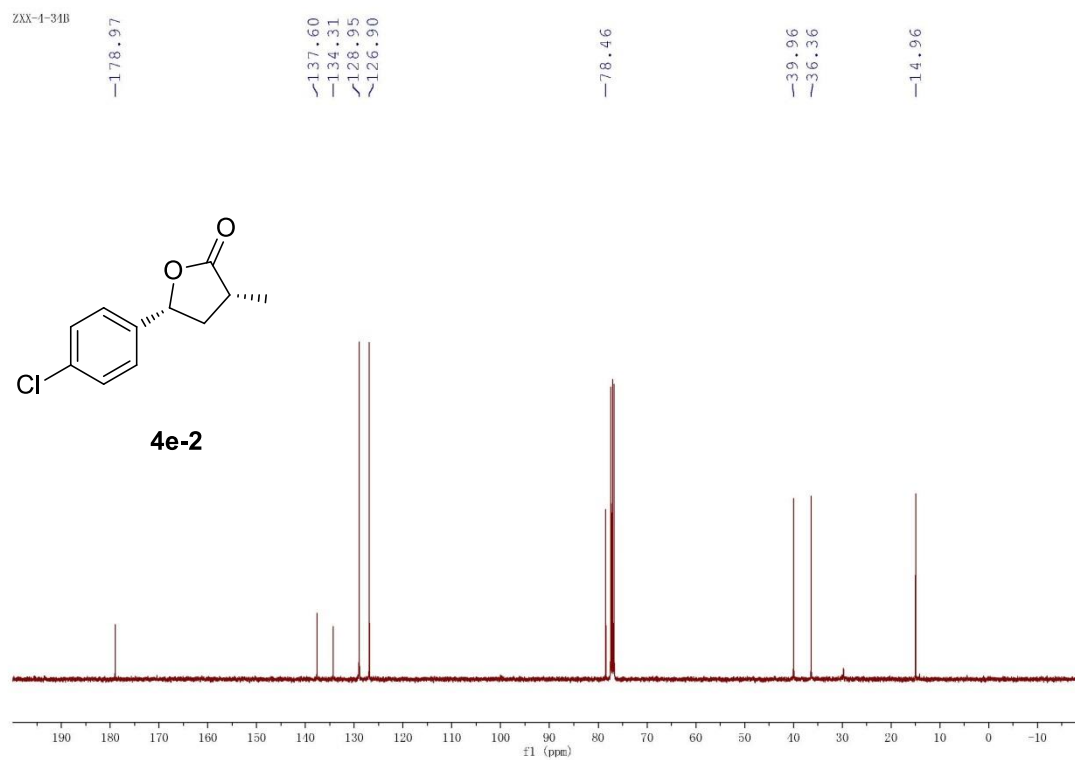


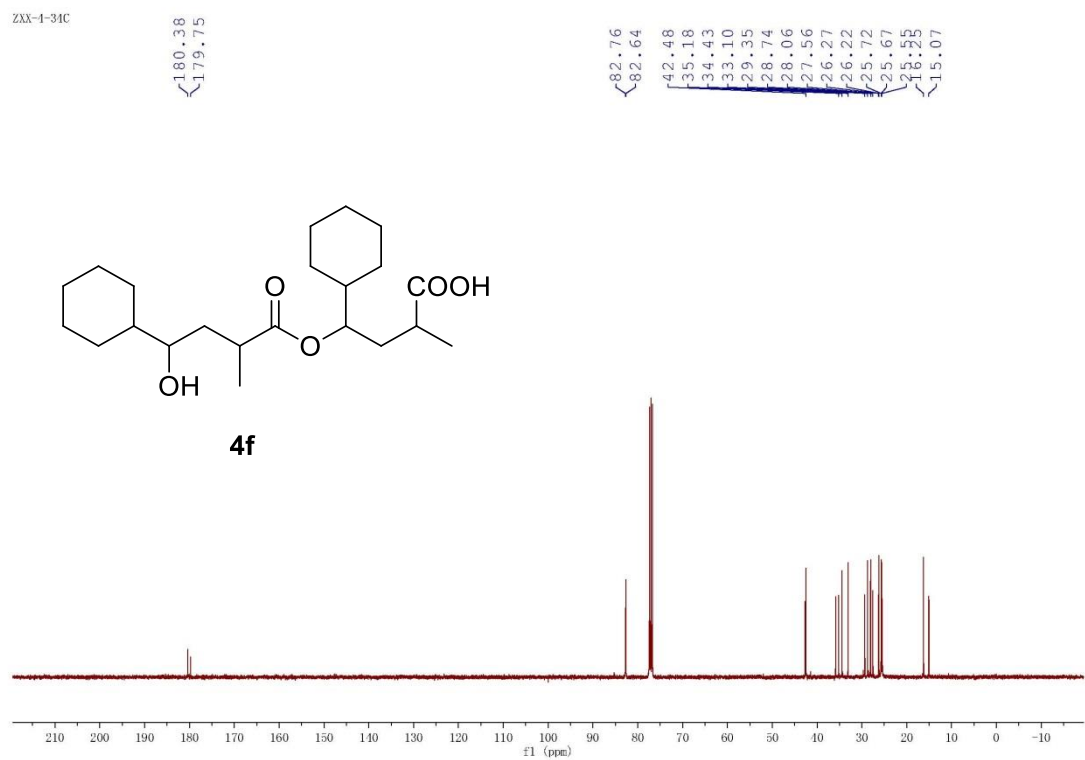
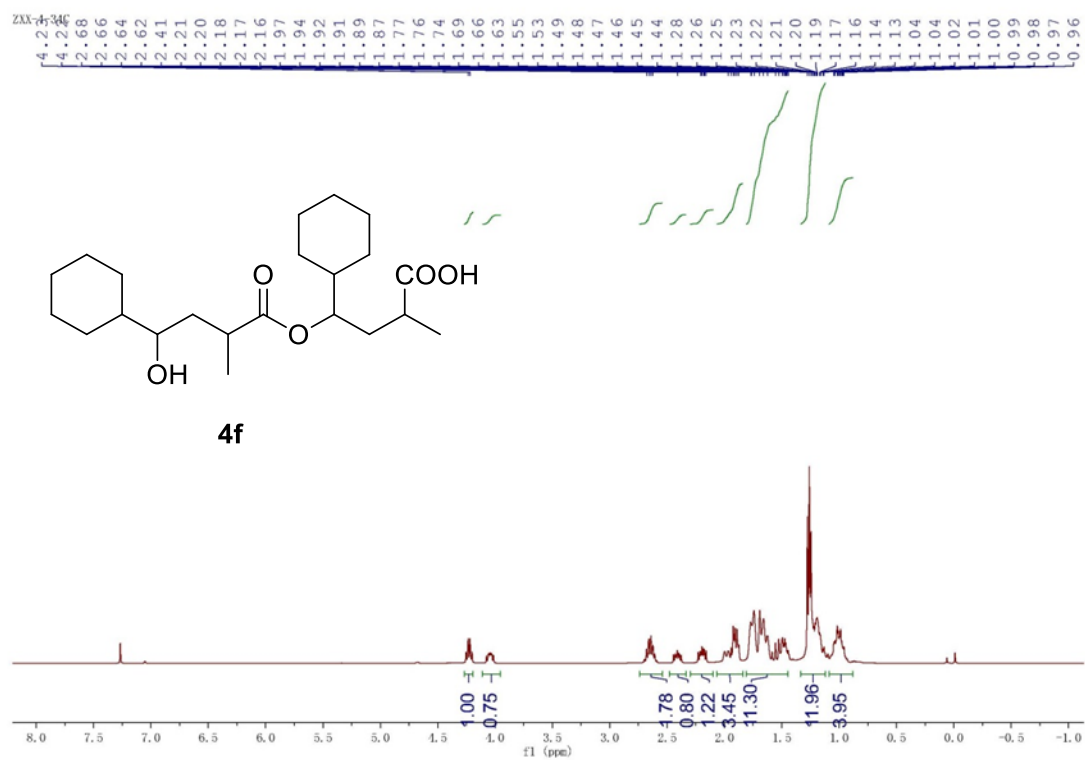


ZXX-1-34B

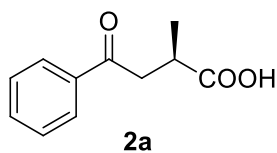


ZXX-1-34B

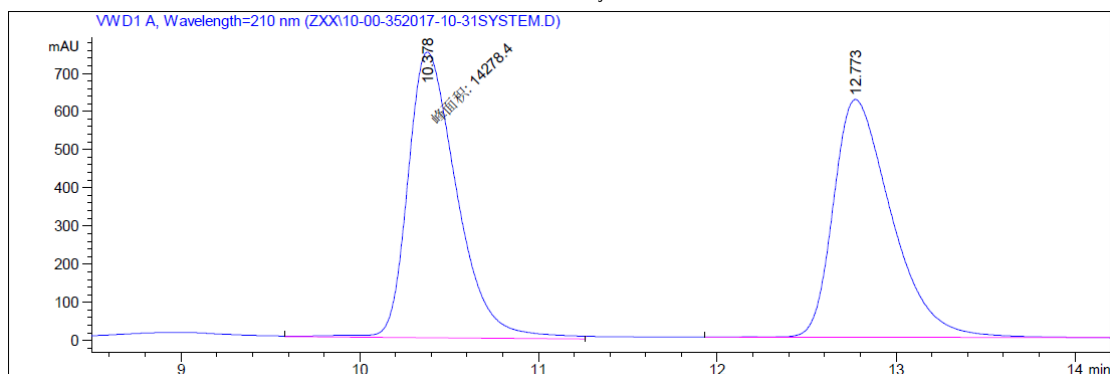




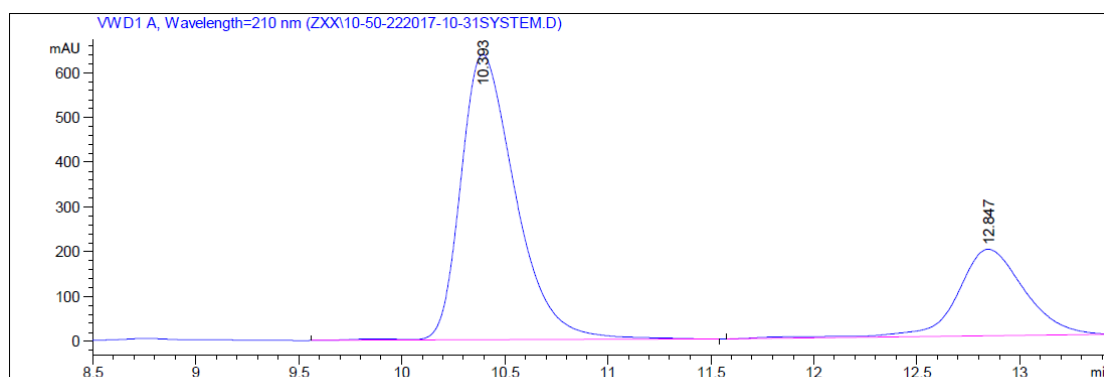
E: HPLC Spectra of Products



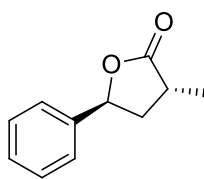
white solid, 96% yield, 45% *ee*, $[\alpha]_D^{20} = -9.0$ ($c=0.4$, CH_2Cl_2). The *ee* of **2a** was determined by HPLC analysis using Daicel Chiralcel OJ-H column (25 cm $\times$ 0.46 cm ID), conditions: *n*-hexane/*i*-PrOH = 90/10, 1.0 mL/min, 210 nm; $t_{\text{major}} = 10.393$ min, $t_{\text{minor}} = 12.847$ min.



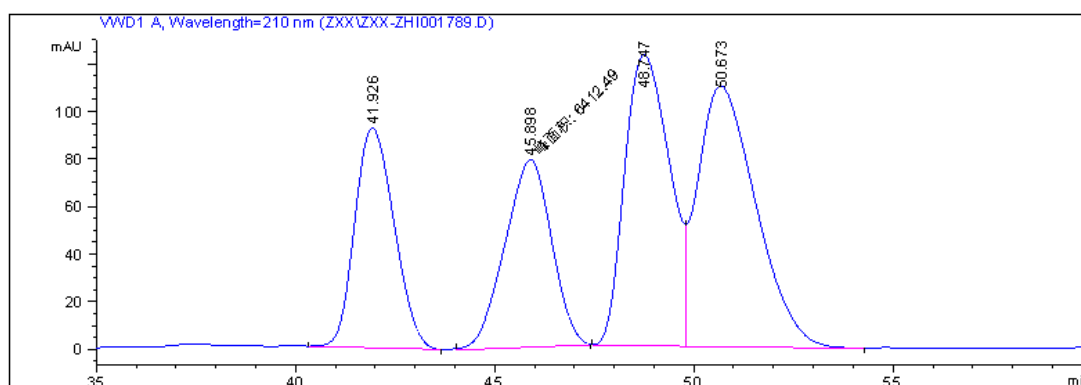
Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	10.378	MM	0.3179	1.42784e4	748.68951	50.2502
2	12.773	VBAR	0.3454	1.41362e4	623.67511	49.7498



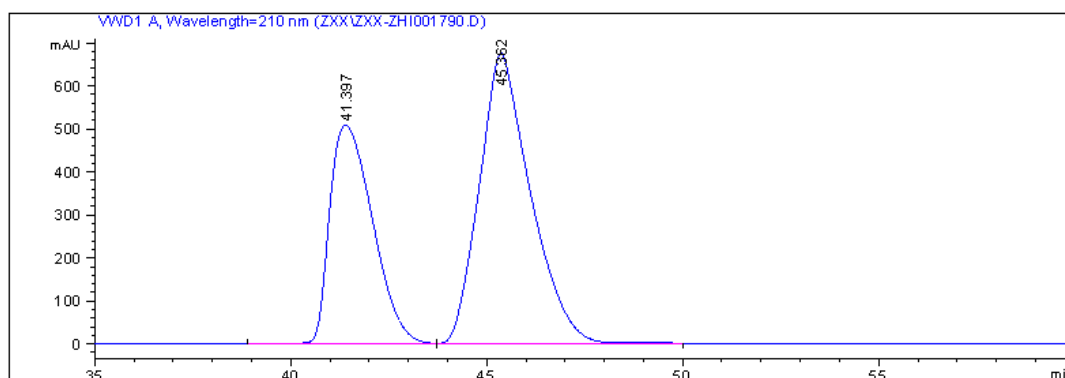
Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	10.393	VB R	0.2803	1.17759e4	638.23584	72.8186
2	12.847	BBA	0.3434	4395.64648	193.26276	27.1814

**4a-1**

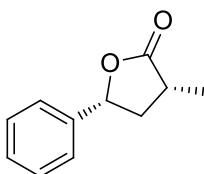
white solid, 50% yield, 22% *ee*, $[\alpha]_D^{25} = -1.0$ ($c=0.4$, CH_2Cl_2). The *ee* of **4a-1** was determined by HPLC analysis using Daicel Chiralcel OJ-H column (25 cm $\times$ 0.46 cm ID), conditions: *n*-hexane/*i*-PrOH = 95/5, 0.5 mL/min, 210 nm; $t_{\text{major}} = 45.362$ min, $t_{\text{minor}} = 41.397$ min.



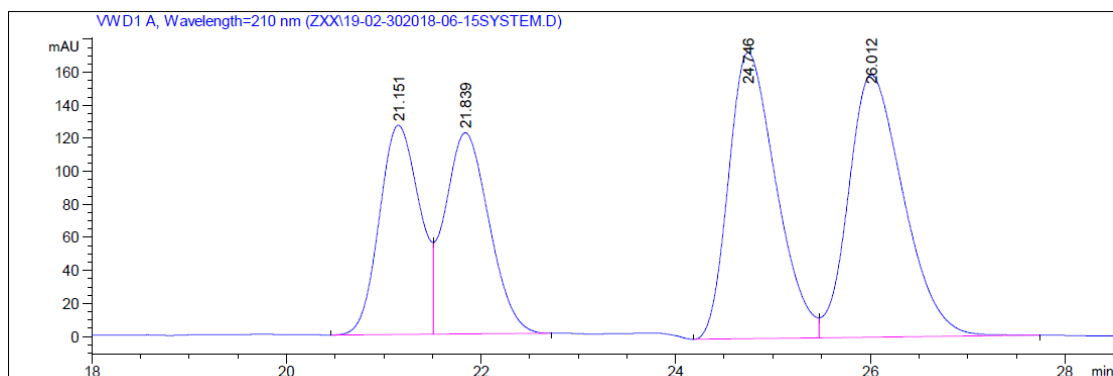
Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	41.926	BB	1.0792	6348.04639	92.43748	18.9303
2	45.898	MM	1.3546	6412.49316	78.90000	19.1224
3	48.747	BV	1.2353	9629.09180	122.33156	28.7145
4	50.673	VB	1.5388	1.11442e4	109.64556	33.2328



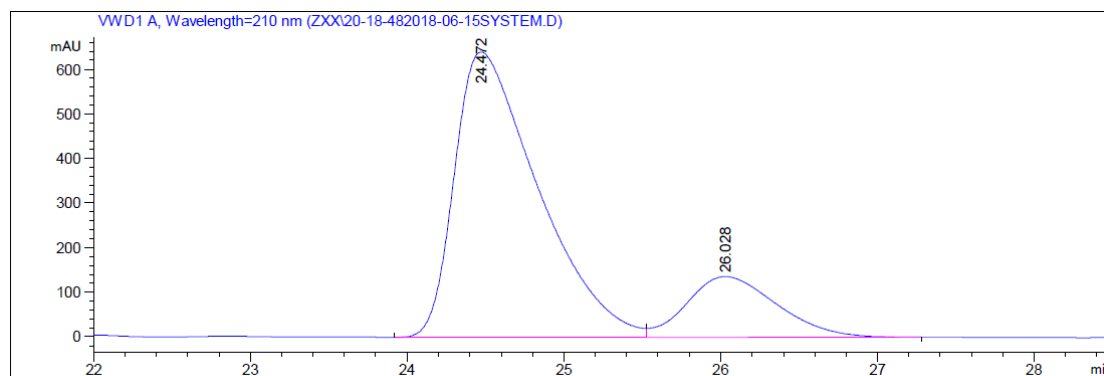
Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	41.397	BB	1.2178	3.84919e4	507.52402	38.7016
2	45.362	BB	1.2772	6.09663e4	674.71521	61.2984

**4a-2**

White solid, 46% yield, 63% *ee*, $[\alpha]_D^{25} = -24.0$ ($c=0.42$, CH_2Cl_2). The *ee* of **4a-2** was determined by HPLC analysis using Daicel Chiralcel OJ-H column (25 cm $\times$ 0.46 cm ID), conditions: *n*-heptane/*i*-PrOH = 95/5, 1 mL/min, 210 nm; $t_{\text{major}} = 24.472$ min, $t_{\text{minor}} = 26.028$ min.



Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	21.151	BV	0.4444	3651.12720	126.62154	18.6244
2	21.839	VB	0.4838	3860.59351	121.79987	19.6929
3	24.746	BV	0.5211	5858.72607	173.42030	29.8853
4	26.012	VB	0.6045	6233.56055	158.96889	31.7974



Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	24.472	BV	0.5523	2.37064e4	639.84930	81.4157
2	26.028	VB	0.6154	5411.31299	135.93820	18.5843

I: References

1. Chen, C.; Zhang, Z.; Jin, S.; Fan, X.; Geng, M.; Zhou, Y.; Wen, S.; Wang, X.; Chung, L. W.; Dong, X.-Q.; Zhang, X. *Angew. Chem. Int. Ed*, **2017**, 56, 6808.
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3. Mayer; Stamm; *Chemische Berichte.*, **1923**, 56; 1431.
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5. Gupta; Muthana; *J. Ind. Inst. Sc.*, **1953**, 35, 310.
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