

Supporting Information

Photo-sensitized oxy-thiocyanation of terminal alkynes / 1,3-aryldienes and their one-pot conversion to 2-hydroxy 4-substitutedaryl thiazoles

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I. Fluorescence emission spectra of $\text{Acr}^+ - \text{Mes ClO}_4^- - \text{NH}_4\text{SCN}$

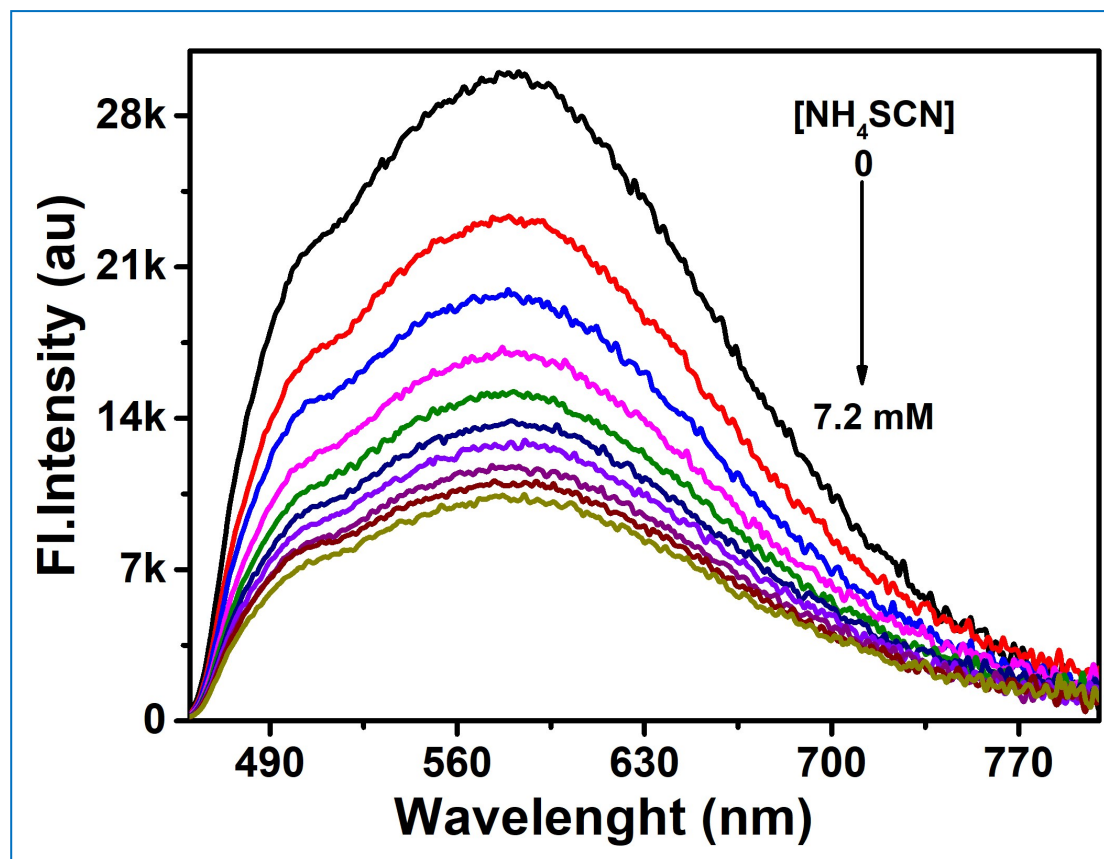


Figure 1: Fluorescence emission quenching of $\text{Acr}^+ - \text{Mes ClO}_4^-$ (10^{-3}M) in acetonitrile solvent with addition of 0.0, 20, 40, 60, 80, 100, 120, 140, 160 μL , 180 μL , of NH_4SCN (10^{-1}M), respectively. The $\text{Acr}^+ - \text{Mes ClO}_4^-$ was excited at 450nm and emission intensity monitored at 580nm.

Stern-Volmer plot

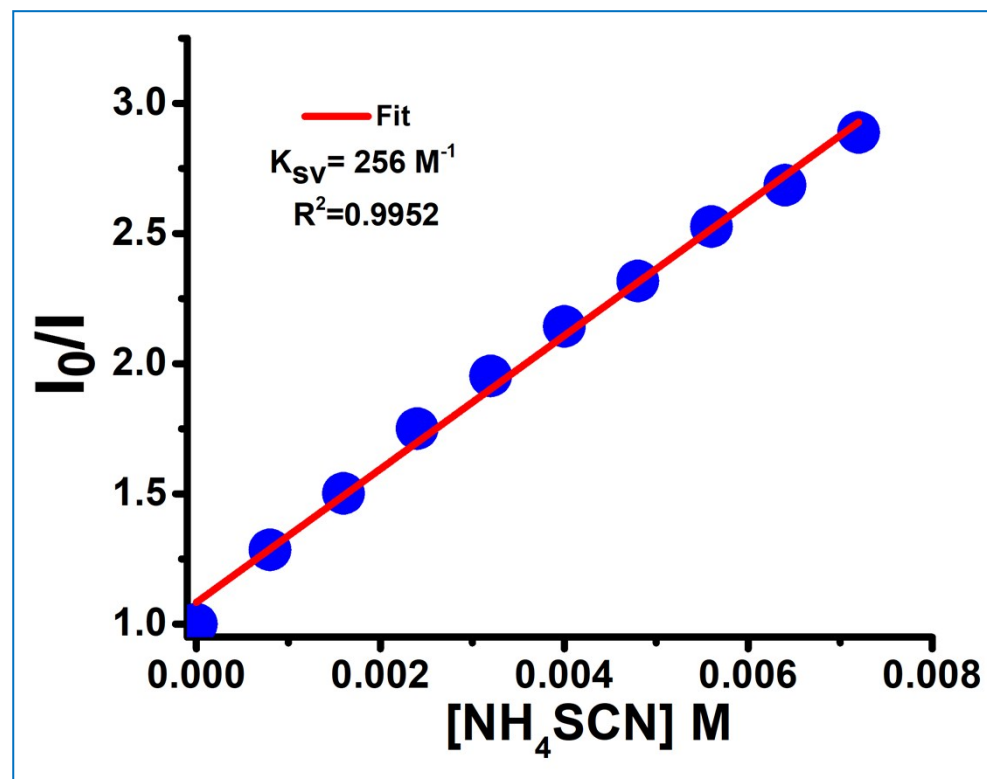


Figure 2: Stern-Volmer plot for Acr⁺-Mes ClO₄⁻ emission quenching by NH₄SCN.

II. Fluorescence emission spectra of Acr^+ -Mes ClO_4^- - 1-Phenyl-1,3-butadiene

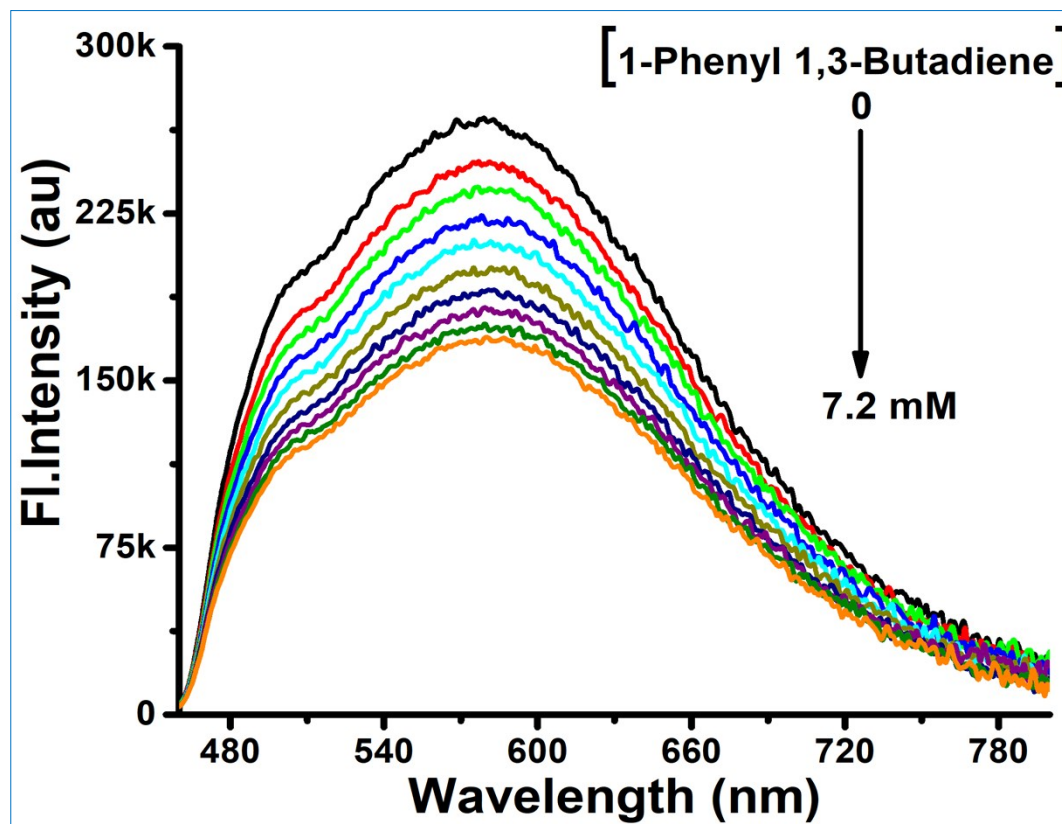


Figure 3: Fluorescence emission quenching of Acr^+ -Mes ClO_4^- (10^{-3}M) in acetonitrile solvent with addition of 0.0, 20, 40, 60, 80, 100, 120, 140, 160 μL , 180 μL of 1-Phenyl-1,3-butadiene(10^{-1}M) respectively. The Acr^+ -Mes ClO_4^- was excited at 450nm and emission intensity monitored at 580nm.

Stern-Volmer plot of $\text{Acr}^+ - \text{Mes ClO}_4^-$ - 1-Phenyl-1,3-butadiene

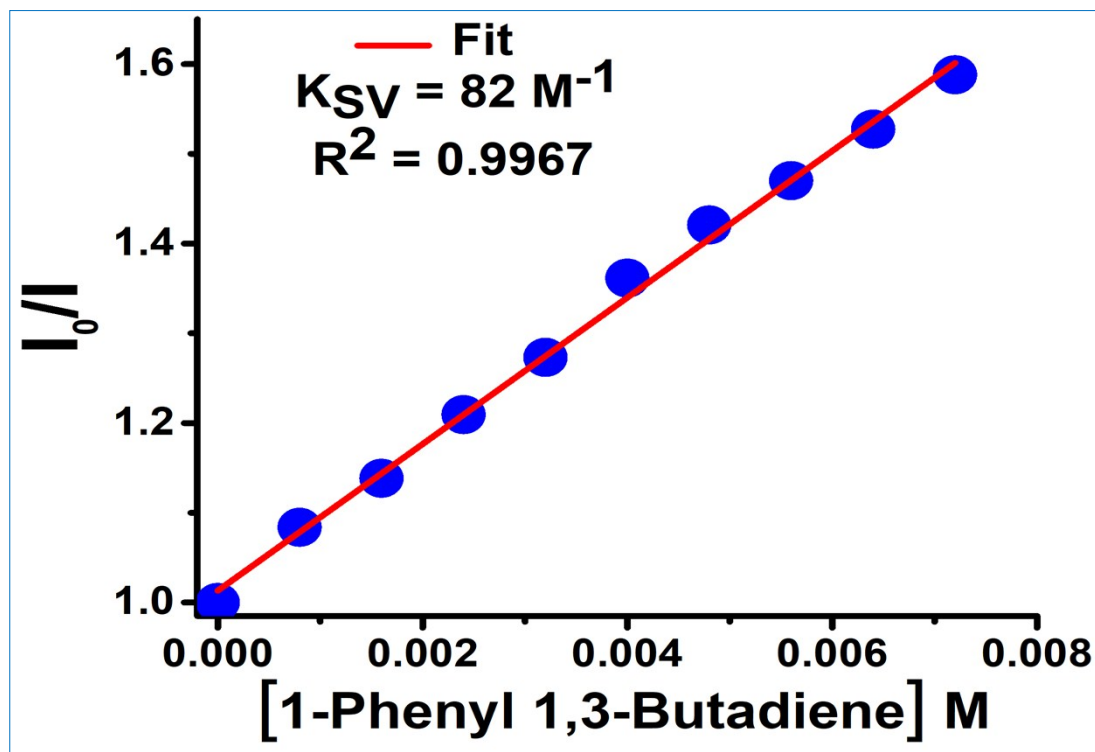


Figure 4: Stern-Volmer plot for $\text{Acr}^+ - \text{Mes ClO}_4^-$ emission quenching by 1-Phenyl-1,3-butadiene. The Stern-Volmer constant (K_{SV}) of 1-Phenyl-1,3-butadiene is three times less than NH_4SCN . Hence NH_4SCN is better quencher than 1-Phenyl -1,3-butadiene.

Cyclic Voltammogram Of 1-Phenyl-1,3-butadiene.

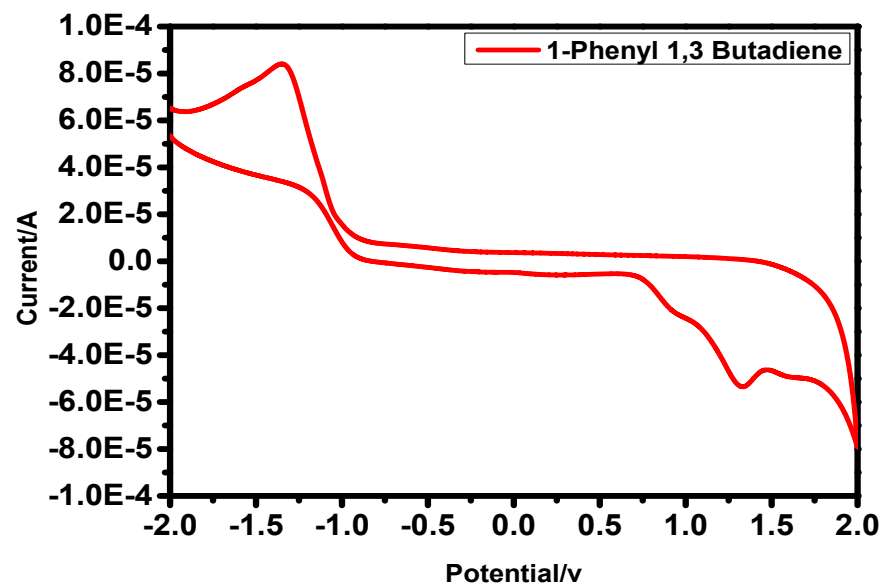


Figure 5. 1-Phenyl-1,3-butadiene oxidation potential (1.32V)in MeCN.

Electrochemical measurements were performed on a PC-controlled CH instruments model CHI 620C electrochemical analyzer. The experiments were performed in a 1 mM 1-Phenyl-1,3-butadiene solution in CAN at scan rate of 100 mV s^{-1} using 0.1 M tetrabutyl ammonium perchlorate (TBAP) as supporting electrolyte, a cyclic voltammogram (CV) had been recorded. The working electrode was glassy carbon, a standard calomel electrode (SCE) was the reference electrode and platinum wire was an auxiliary electrode.

III. Fluorescence emission spectra of Acr^+ -Mes ClO_4^- - Phenyl acetylene

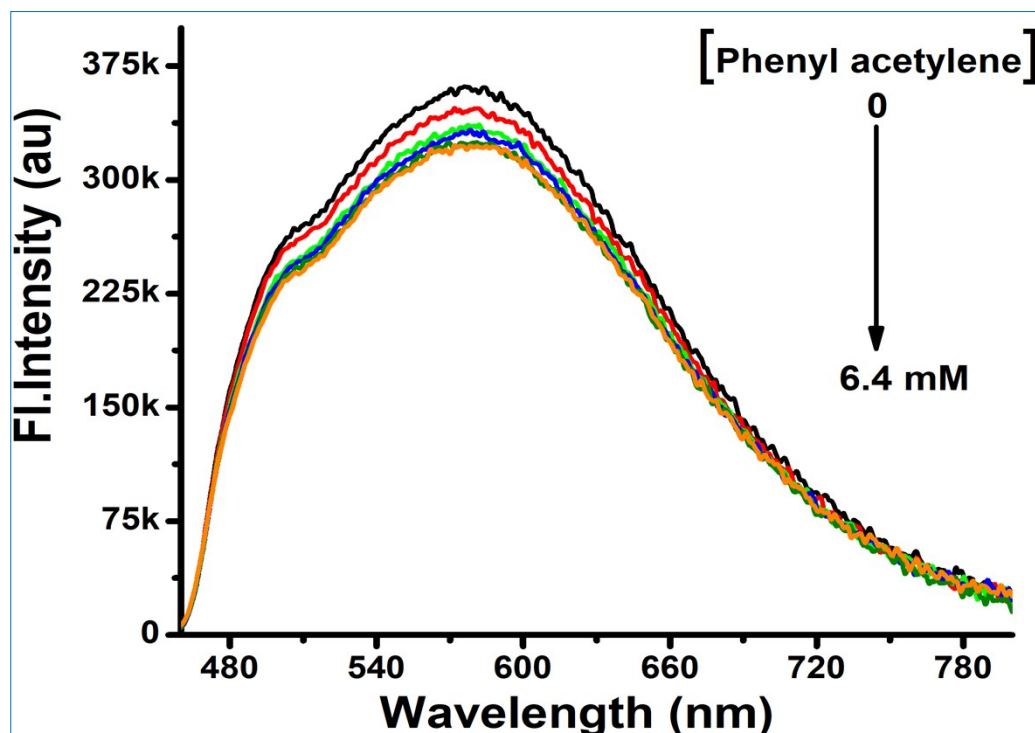


Figure 6: Fluorescence emission quenching of Acr^+ -Mes ClO_4^- (10^{-3}M) in acetonitrile solvent with addition of 0.0, 20, 40, 80, 120, 160, of Phenyl acetylene(10^{-1}M), respectively. The Acr^+ -Mes ClO_4^- was excited at 450nm and emission intensity monitored at 580nm.

Stern-Volmer plot of $\text{Acr}^+ - \text{Mes ClO}_4^-$ - Phenyl acetylene

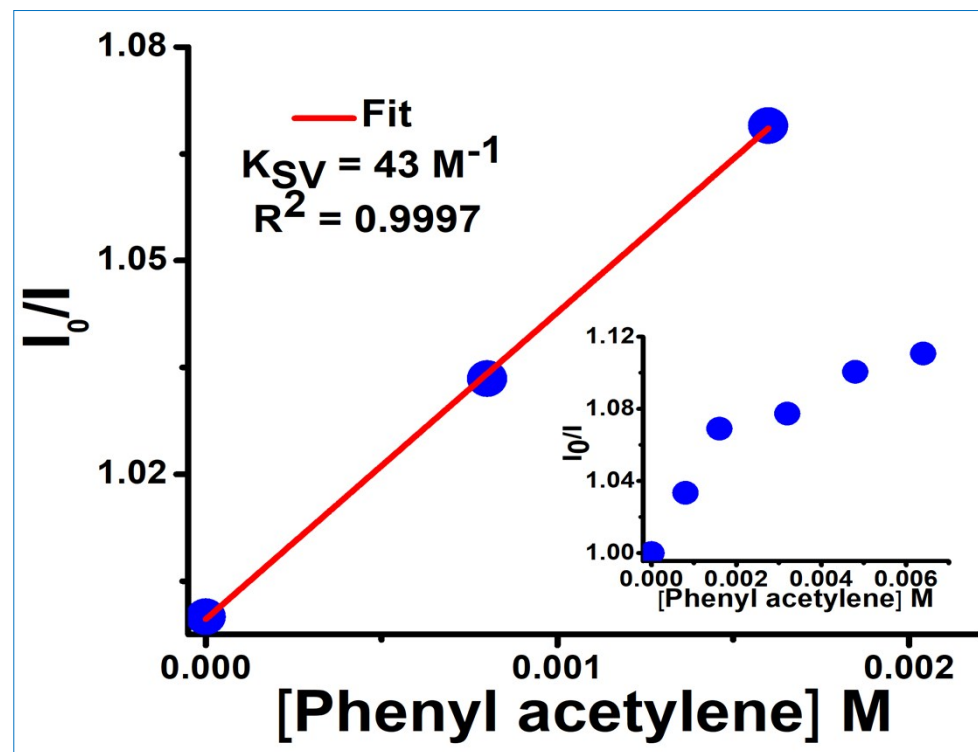
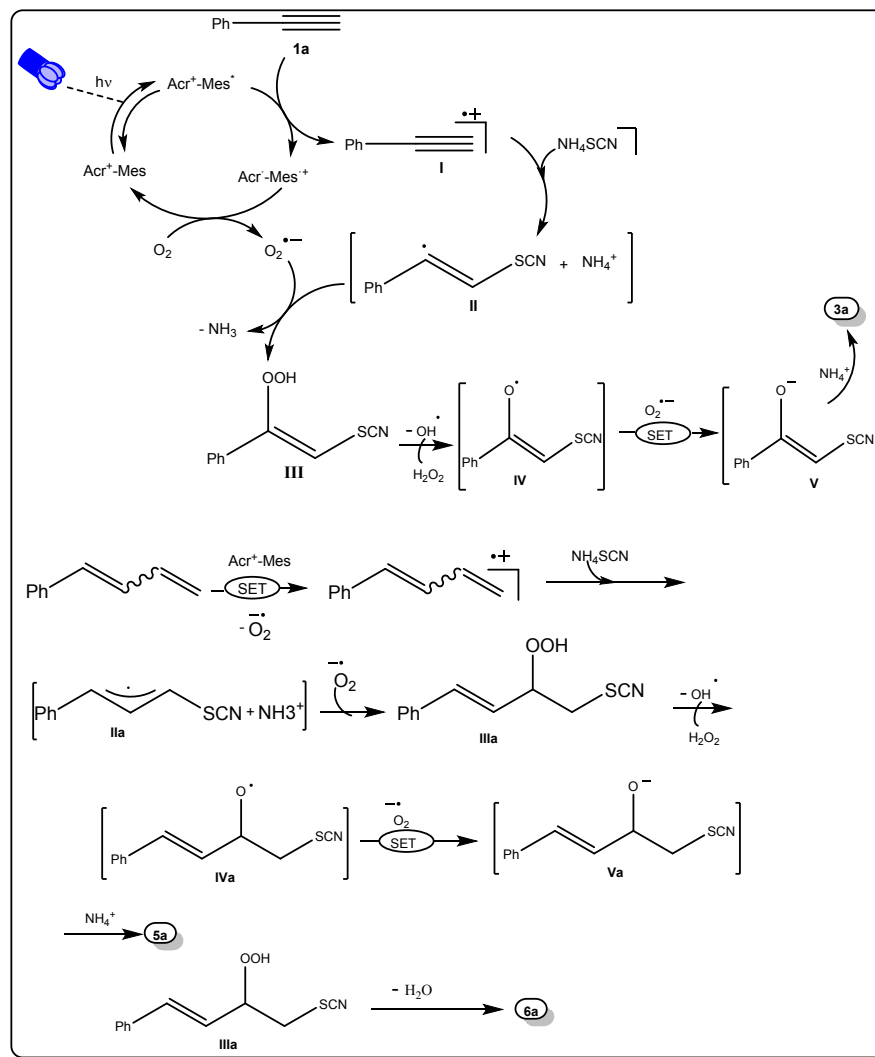


Figure 7: Stern-Volmer plot for $\text{Acr}^+ - \text{Mes ClO}_4^-$ emission quenching by Phenyl acetylene. The efficient quenching of Phenyl acetylene is upto 1.6mM . The Stern-Volmer constant (K_{SV}) of Phenyl acetylene is six times less than NH_4SCN . Hence NH_4SCN is better quencher than Phenyl acetylene.

IV. Alternative mechanism



- The excited state of photo-catalyst also quenched by ground state of **1a**. On the basis of this, alternative mechanism cannot be ruled out

V . 5 mmol Scale-up reaction for the synthesis of 4-Phenylthiazol-2-ol(7a)-Image

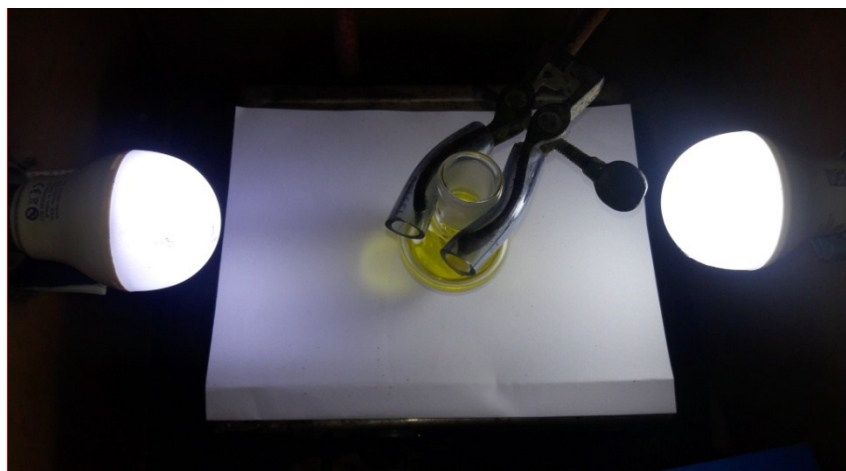


Figure 8: 5 mmol Scale-up reaction for the synthesis of 4-Phenylthiazol-2-ol, (**7a**) with two 7W day light white LED lamps (PHILIPS 6500K, 42mA 220-240V, Made in India).

