

N-Hydroxyphthalimide/Benzoquinone-Catalyzed Chlorination of Hydrocarbon C–H Bond with N-Chlorosuccinimide

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| On leave to Institute of Process Engineering

|| Quantum chemical calculations

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1. General Method

Commercial reagents were purchased from TCI or ALADDIN and unless otherwise stated, all solvents were dried and distilled before use according to the standard methods. Analytical thin layer chromatography (TLC) was performed on precoated silica gel. Visualization on TLC was achieved with UV light (254 nm) as the visualization method. ^1H NMR spectra were recorded on a bruker ASCEND spectrometer (^1H , 600 MHz; $^{13}\text{C}\{^1\text{H}\}$, 151 MHz) or a JEOL JNM-ECA600 spectrometer. Chemical shifts were quoted in parts per million (ppm) referenced to the appropriate solvent peak. Data for ^1H NMR spectra are reported as follows: chemical shift (δ shift), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dd = double of doublet, ddd = double of dd, dt = double of triplet, td = triple of doublet), integration, coupling constant (Hz), and assignment. ^{13}C NMR spectra were recorded on bruker ASCEND spectrometer ($^{13}\text{C}\{^1\text{H}\}$, 151 MHz). Chemical shifts were reported in ppm referenced to the center line of a triplet at 77.0 ppm of chloroform-d or a heptet at 39.5 ppm of dimethyl sulfoxide-d₆. GC analysis was performed using Agilent GC-7890B equipped with a DB-FFAP capillary column, 30 m×0.32 mm; FID detector: 280 °C; injection: 220 °C; oven temperature: 85 °C for 15 min, raised to 190 °C at a rate of 30 °C/min, and held for 15 min. GC-MS analysis was performed using Shimadzu GCMS-QP2020 with HP-5MS column, 30 m×0.32 mm; FID detector: 250 °C; injection: 250 °C; oven temperature: 40 °C for 2 min, raised to 100 °C at a rate of 10 °C/min, then to 240 °C at a rate of 20 °C/min, and held for 8 min. All reagents were

commercially available and weighed out under Argon conditions. All chemicals were used without further purification.

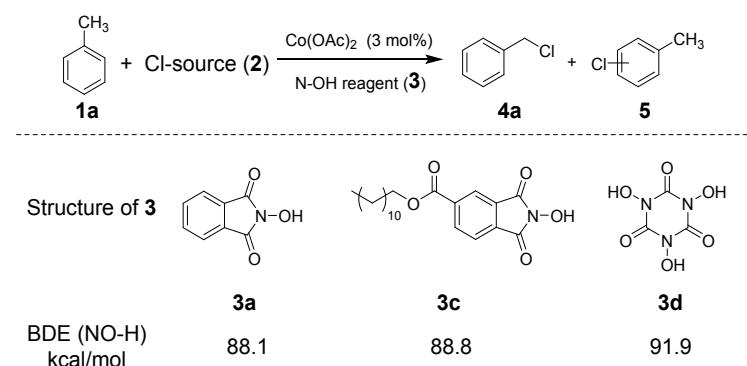
Chlorination of toluene

A typical chlorination procedure: NCS (0.5 mmol, 1 equiv.), toluene (2.5 mmol, 5 equiv.), CH₃CN (2 mL), DDQ (0.0025 mmol, 0.005 equiv.), NHPI (0.05mmol, 0.1equiv.) were added to an oven-dried tube. The reaction tube was equipped with a magnetic stir bar and sealed with a Teflon-lined cap for 3 hours or 16 hours at 80 °C.

After the reaction was finished, the reaction mixture was diluted and detected by GC containing biphenyl as an internal standard. Or concentrate the reaction mixture in vacuo to remove the solvent. Notably, the vacuum degree of concentration and the time should be controlled, avoiding the product loss. The crude reaction mixture was detected by ¹H NMR with CH₂Br₂ as an internal standard. Otherwise the residue was purified by column chromatography to give the product. The products were detected by ¹H NMR analysis.

2. Efforts on optimization of transition metal catalyzed chlorination reaction

Table S1. Condition screening for chlorination of toluene in transition metal catalyzed system.^a



entry	1a (mmol)	2a (mmol)	3 (mol%)	t (h)	varyations	conv. (%)	yield (%)
1	0.5	0.5	3a (10)	6	-	68	31
2	0.5	0.5	3a (10)	6	50 °C	30	n.d.
3	0.5	0.5	3a (10)	6	100 °C	100	6
4	0.5	0.5	3a (10)	6	160 °C	65	29
5	0.5	0.5	3a (10)	6	Co(acac) ₂ (3 mol%)	73	27
6	0.5	0.5	3a (10)	6	CoCl ₂ (3 mol%)	87	18
7	0.5	0.5	3c (10)	6	-	65	21
8	0.5	0.5	3d (10)	6	-	75	34
9	0.5	0.5	3a (10)	6	CH ₃ COOH (2 mL)	100	n.d.
10	0.5	0.5	3a (10)	6	1,2-DCE (2 mL)	49	7
11	0.5	0.5	3a (10)	6	CF ₃ COOH (2 mL)	100	n.d.
12	0.5	0.5	3a (10)	6	DMSO (2 mL)	43	n.d.
13	0.5	0.5	3a (10)	6	PhCN (2 mL)	83	17

14	0.5	0.5	3a (10)	6	CCl ₄ (2 mL)	76	21
15	0.5	0.5	3a (10)	6	O ₂ (10 atm)	89	5
16	0.5	0.5	3a (10)	6	pyridine (20 mol%)	65	12
17	0.5	0.5	3a (10)	6	TsOH (5 mol%)	71	31
18 ^b	1	0.5	3a (10)	16		-	41
19 ^b	1	0.5	3d (3)	16		-	48

^a Conditions: CH₃CN (2 mL), O₂ (1atm), 140 °C, conversions and yields are determined by GC using diphenyl as internal standard and are on the basis of toluene; ^b yields are determined by GC using diphenyl as internal standard and are on the basis of NCS.

3. N-Cl BDE calculation of chlorine reagents

Four different chlorine reagents (**2a-2d**) were compared and their N-Cl bond dissociation energy (BDE_{N-Cl}, **eq. 1**) values were calculated (Table S2).¹⁻⁴ **2a** has the highest BDE_{N-Cl}, followed by **2b**. The corresponding N-Cl BDEs of **2c** and **2d** are lower by more than 10 kcal/mol.



Table S2. Comparison of N-Cl BDE of the chlorine reagents based on B3LYP/6-31+G(d) and M06/6-31+G(d) level of theories.

	2a	2b	2c	2d
	B3LYP/6-31+G(d)		M06/6-31+G(d)	
	(kcal/mol)		(kcal/mol)	
2a	64.8		73.3	
2b	62.6		70.5	
2c	54.2		63.0	
2d	50.0		58.8	

The calculations were performed by the B3LYP⁵⁻⁷ and M06⁸ density functionals in combination with the 6-31+G(d) basis set as implemented in the Gaussian 09 program package.⁹

Normal mode analysis was carried out on each optimized structure in order to verify that they have the right number of imaginary frequencies.

Cartesian coordinates of 2b-2d at the B3LYP/6-31+G(d) level of theory

2b			
C	-0.99927	0.84365	-0.02339
C	-1.02052	-0.55119	-0.00164
C	-2.19889	-1.28507	0.02073
C	-3.39898	-0.56389	0.03151
C	-3.39764	0.83752	0.01561
C	-2.19782	1.55474	-0.01407
H	-2.19214	-2.37043	0.03189
H	-4.34236	-1.10178	0.05387
H	-4.34243	1.37327	0.02489
H	-2.18030	2.63998	-0.03107
C	0.36356	1.45002	-0.05432
O	0.64188	2.62688	0.02538
O	0.90634	-2.01631	-1.18940
N	1.28857	0.40812	-0.22576
Cl	2.96095	0.66319	0.01990
S	0.63746	-1.22141	0.00355
O	1.02668	-1.71611	1.32140
2b·			
C	-0.79063	0.70448	0.00009
C	-0.30744	-0.60384	0.00016
C	-1.14392	-1.71232	0.00008
C	-2.52153	-1.46726	0.00002
C	-3.02613	-0.15602	0.00001
C	-2.16654	0.94408	0.00003
H	-0.74955	-2.72360	0.00008
H	-3.21150	-2.30636	-0.00003
H	-4.10085	0.00110	-0.00005
H	-2.54227	1.96261	-0.00001
C	0.29613	1.72476	-0.00001
O	0.14571	2.93620	-0.00014
O	2.02623	-1.02628	-1.27287
N	1.60190	1.16024	-0.00002
S	1.48517	-0.54568	-0.00003
O	2.02661	-1.02589	1.27281
2c			
C	-0.82529	-0.69948	0.00015
C	-0.82530	0.69949	0.00008
C	-2.00813	1.42539	0.00013
C	-3.20977	0.70060	0.00032
2c·			
C	-3.20977	-0.70060	0.00043
C	-2.00812	-1.42539	0.00035
H	-1.99784	2.51109	0.00008
H	-4.15598	1.23453	0.00039
H	-4.15597	-1.23453	0.00057
H	-1.99783	-2.51108	0.00040
C	0.58538	1.18811	-0.00002
C	0.58539	-1.18811	-0.00009
O	1.01653	-2.31758	-0.00004
O	1.01655	2.31757	-0.00020
N	1.35940	-0.00001	-0.00015
Cl	3.06005	0.00000	-0.00039
2d			
C	-0.14419	-0.69974	0.00000
C	-0.14420	0.69975	-0.00008
C	-1.33076	1.42875	0.00005
C	-2.52697	0.70426	0.00024
C	-2.52696	-0.70427	0.00032
C	-1.33075	-1.42875	0.00020
H	-1.32157	2.51456	0.00000
H	-3.47577	1.23384	0.00035
H	-3.47576	-1.23386	0.00047
H	-1.32155	-2.51456	0.00025
C	1.27929	1.14730	-0.00028
C	1.27930	-1.14728	-0.00017
O	1.70851	-2.28724	0.00005
O	1.70851	2.28725	-0.00019
N	2.13284	-0.00001	-0.00022

Cl	-2.80065	1.24286	-0.00023
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2d·

C	1.20761	1.25617	-0.04420
C	-1.20762	1.25617	-0.04419
C	0.00001	-0.94187	-0.07845
N	-1.17366	-0.15311	-0.07450

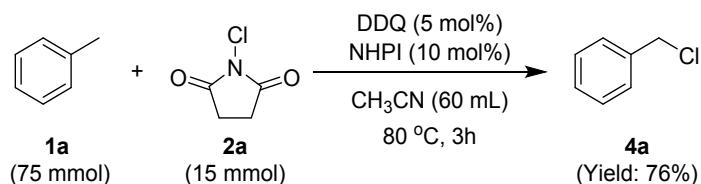
N	0.00000	1.85179	-0.41985
N	1.17367	-0.15310	-0.07453
O	0.00001	-2.14347	-0.06820
O	-2.19499	1.90640	0.21155
O	2.19496	1.90640	0.21162
Cl	2.66452	-0.98813	0.06303
Cl	-2.66451	-0.98814	0.06305

Cartesian coordinates of 2b-2d at the M06/6-31+G(d) level of theory

2b				2c				2d			
C	-0.99540	0.83975	-0.02245	C	-3.19275	-0.69795	0.00042	C	-1.99608	-1.42145	0.00034
C	-1.00831	-0.54972	-0.00440	H	-1.98074	2.50964	0.00009	H	-4.14142	1.23159	0.00040
C	-2.17725	-1.28872	0.01442	H	-4.14142	-1.23159	0.00056	H	-1.98073	-2.50964	0.00039
C	-3.37626	-0.57545	0.02716	C	0.58332	1.18501	-0.00005	C	0.58333	-1.18501	-0.00007
C	-3.38214	0.82074	0.01632	O	1.01261	-2.30893	-0.00003	O	1.01263	2.30891	-0.00021
C	-2.19116	1.54343	-0.01094	N	1.35394	-0.00001	-0.00015	Cl	3.03906	0.00000	-0.00037
H	-2.16044	-2.37668	0.02193								
H	-4.31937	-1.11780	0.04728								
H	-4.33208	1.35128	0.02784								
H	-2.17581	2.63131	-0.02486								
C	0.35795	1.44941	-0.05309								
O	0.63554	2.62050	0.02152								
O	0.91362	-1.99011	-1.17977								
N	1.27798	0.40641	-0.21200								
Cl	2.93641	0.64804	0.01640								
S	0.63734	-1.20581	0.00489								
O	1.02093	-1.69206	1.31483								
2b·				2c·				2d			
C	-0.78316	0.70166	0.00018	C	-0.14114	-0.69633	0.00001	C	1.45137	0.13180	-0.00011
C	-0.30004	-0.60013	0.00024	C	-0.14114	0.69634	-0.00007	C	-0.83998	1.19095	-0.00034
C	-1.13061	-1.70673	0.00003	C	-1.32161	1.42532	0.00005	C	-0.61150	-1.32257	0.00035
C	-2.50277	-1.46209	-0.00004	C	-2.51200	0.70195	0.00024	N	-1.33806	-0.12167	0.00008
C	-3.00706	-0.15584	0.00002	C	-2.51199	-0.70196	0.00031	N	0.56385	1.21962	-0.00060
C	-2.15322	0.94192	0.00010	C	-1.32161	-1.42532	0.00020	N	0.77427	-1.09782	0.00030
H	-0.73042	-2.71841	-0.00003	H	-1.30714	2.51350	0.00000	O	-1.11354	-2.40876	0.00059
H	-3.19407	-2.30257	-0.00018	H	-3.46334	1.23102	0.00034	O	-1.52984	2.16861	-0.00026
H	-4.08416	-0.00185	-0.00005	H	-3.46333	-1.23104	0.00047	O	2.64303	0.23983	-0.00023
H	-2.52725	1.96379	0.00008	H	-1.30713	-2.51350	0.00026	Cl	1.75333	-2.48648	-0.00021
C	0.29469	1.71940	0.00001	C	1.27274	1.14485	-0.00027	Cl	-3.03044	-0.27491	-0.00012
O	0.14210	2.92326	-0.00020	C	1.27275	-1.14483	-0.00016				
O	2.00610	-1.01890	-1.26593	O	1.69304	-2.28061	0.00003				
N	1.59600	1.15043	0.00006	O	1.69305	2.28061	-0.00020				
S	1.47617	-0.54422	-0.00002	N	2.12517	-0.00001	-0.00023				
O	2.00658	-1.01879	1.26575								
2c				2d							
C	-0.81943	-0.69646	0.00015								
C	-0.81943	0.69646	0.00007								
C	-1.99609	1.42146	0.00014								
C	-3.19275	0.69795	0.00032								

Cl	1.27729	2.76143	0.00042	N	0.00001	1.82133	-0.45685
2d·							
C	1.20584	1.25258	-0.04615	N	1.16906	-0.15428	-0.08595
C	-1.20586	1.25260	-0.04621	O	-0.00001	-2.13825	-0.06576
C	0.00000	-0.94293	-0.08594	O	-2.17805	1.90056	0.22996
N	-1.16907	-0.15427	-0.08600	O	2.17809	1.90054	0.22984
				Cl	2.64530	-0.97841	0.06822
				Cl	-2.64531	-0.97840	0.06820

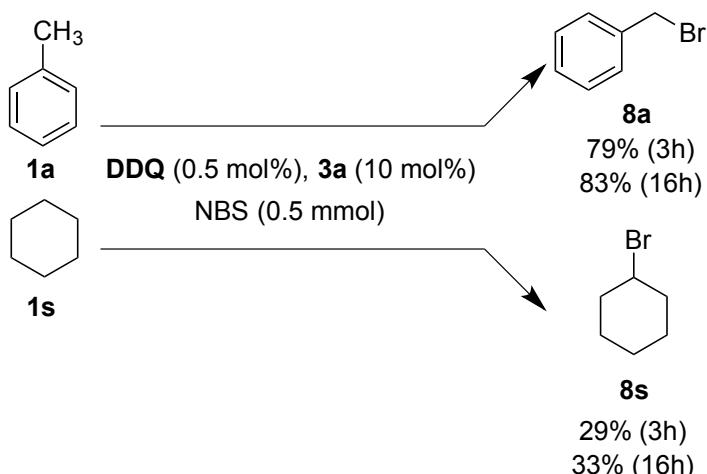
4. Toluene side-chain chlorination on gram-scale



Scheme S1. Toluene side-chain chlorination on gram-scale.^a

^a Conditions: toluene (**1a**, 75 mmol), NCS (**2a**, 15 mmol), NHPI (**3a**, 10 mol%), DDQ (5 mol%), CH₃CN (60 mL), 80 °C, 3h; GC yield with diphenyl as internal standard and is on the basis of NCS (**2a**).

5. Oxidative bromination of **1a** and **1s** with NBS.^a



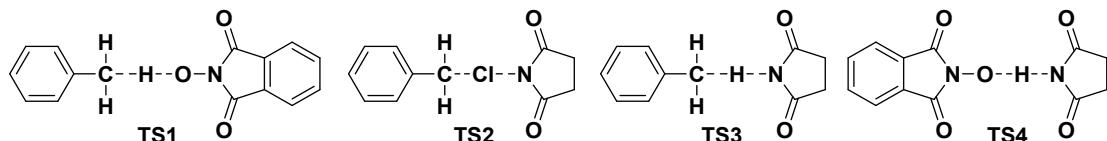
Scheme S2. Oxidative bromination of **1a** and **1s** with NBS.^a

^a Conditions: **1** (2.5 mmol, 5.0 equiv.), NBS (0.5 mmol), NHPI (**3a**, 10 mol%), DDQ (0.5 mol%), CH₃CN (2 mL), 80 °C, 3h; yield determined by GC using diphenyl as internal standard and is on the basis of NBS.

6. DFT calculation of mechanism



Table S3. Comparison of relative Gibbs free energies (ΔG_r , kcal/mol) and barrier heights (ΔG^\ddagger , kcal/mol) of the reaction steps. The calculations were performed to mimic the optimized experimental conditions (80 °C, CH₃CN).



	B3LYP/6-31+G(d)	M06/6-31+G(d)	M06-2X/6-31+G(d)
ΔG_r	eq. 2	13.7	8.8
	eq. 3	-1.7	1.0
	eq. 4	-28.4	-32.4
	eq. 5	-42.1	-41.2
ΔG^\ddagger	TS1	29.8	- **
	TS2	17.0	19.7
	TS3	13.5*	9.4
	TS4	12.5*	- **

* - Single-point calculations on gas phase optimized structures

** - Attempts to locate the corresponding transition states were not successful

Computational Methods

All calculations were carried out by using the Gaussian 09 program package. The calculations were performed by the B3LYP,⁵⁻⁷ M06⁸ and M06-2X⁸ density functionals in combination with the 6-31+G(d) basis set as implemented in the Gaussian 09

program package.⁹ Normal mode analysis was carried out on each optimized structure in order to verify that they have the right number of imaginary frequencies. To mimic the optimized experimental conditions the calculations were performed using the CPCM solvent model^{10, 11} with acetonitrile (CH_3CN) as solvent at 80 °C and 1 atm.

Cartesian coordinates of all computed structures at the B3LYP/6-31+G(d) (solvent: CH₃CN) level of theory

1a			
C	0.19873	-1.20276	0.00000
C	0.91709	0.00547	0.00000
C	0.19187	1.20629	0.00000
C	-1.20842	1.20517	0.00000
C	-1.90966	-0.00384	0.00000
C	-1.19865	-1.21031	0.00000
H	0.74108	-2.14651	0.00000
H	0.72767	2.15334	0.00000
H	-1.74810	2.14921	0.00000
H	-2.99670	-0.00823	0.00000
H	-1.73322	-2.15729	0.00000
C	2.42981	0.00291	0.00000
H	2.82733	-0.51504	-0.88195
H	2.83002	1.02193	-0.00002
H	2.82733	-0.51500	0.88197

2a			
C	1.89116	-0.77039	0.00000
C	1.89116	0.77039	0.00006
C	0.43187	1.18697	-0.00006
C	0.43187	-1.18697	-0.00009
H	2.37515	-1.19895	-0.88276
H	2.37508	-1.19901	0.88277
H	2.37520	1.19902	-0.88265
H	2.37504	1.19894	0.88288
O	-0.04245	-2.30336	0.00003
O	-0.04245	2.30336	0.00000
N	-0.31922	0.00000	-0.00004
Cl	-2.02724	0.00000	0.00002

3a			
C	0.50398	0.70124	0.00395
C	0.50398	-0.70123	0.00395
C	1.68766	-1.42552	0.00384
C	2.89016	-0.70044	-0.00081
C	2.89016	0.70044	-0.00088
C	1.68766	1.42553	0.00371
H	1.68489	-2.51110	0.00513

H	3.83578	-1.23439	-0.00497
H	3.83578	1.23439	-0.00506
H	1.68489	2.51110	0.00502
C	-0.90699	-1.17816	0.00791
C	-0.90699	1.17817	0.00834
N	-1.67282	-0.00001	0.09707
O	-1.35842	-2.30632	-0.04155
O	-1.35845	2.30630	-0.04161
O	-3.03107	0.00002	-0.11986
H	-3.44587	-0.00012	0.76458

4a			
C	-1.99588	-1.21093	-0.15970
C	-0.65937	-1.20996	0.24760
C	0.02156	0.00017	0.45244
C	-0.65962	1.21012	0.24730
C	-1.99612	1.21072	-0.15999
C	-2.66656	-0.00020	-0.36574
H	-2.51367	-2.15419	-0.31205
H	-0.14164	-2.15273	0.40879
H	-0.14208	2.15304	0.40826
H	-2.51411	2.15384	-0.31256
H	-3.70672	-0.00034	-0.68065
C	1.45297	0.00038	0.89240
H	1.71627	0.89354	1.45791
H	1.71633	-0.89232	1.45861
Cl	2.62375	-0.00016	-0.55025

I			
C	-0.45218	0.70415	0.00029
C	-0.45218	-0.70415	0.00032
C	-1.64034	-1.42645	0.00011
C	-2.83937	-0.70195	-0.00001
C	-2.83937	0.70195	0.00001
C	-1.64034	1.42645	0.00015
H	-1.63709	-2.51163	0.00012
H	-3.78560	-1.23426	-0.00013
H	-3.78560	1.23426	-0.00011
H	-1.63709	2.51163	0.00012

C	0.93597	-1.20174	0.00062
C	0.93597	1.20174	0.00035
N	1.75723	0.00000	-0.00020
O	1.39497	-2.32015	-0.00015
O	1.39497	2.32015	-0.00008
O	3.02202	0.00000	-0.00097

N	0.00000	0.95934	0.00000
H	0.00000	1.97498	-0.00001
C	-0.76946	-1.26012	-0.00344
H	-1.20682	-1.74751	0.87337
H	-1.19728	-1.73904	-0.88976

II			
C	0.00000	1.13594	1.21448
C	0.00000	-0.25291	1.22087
C	0.00000	-0.99712	0.00000
C	0.00000	-0.25291	-1.22087
C	0.00000	1.13594	-1.21448
C	0.00000	1.84351	0.00000
H	0.00000	1.67898	2.15640
H	0.00000	-0.79384	2.16442
H	0.00000	-0.79384	-2.16442
H	0.00000	1.67898	-2.15640
H	0.00000	2.93000	0.00000
C	0.00000	-2.40655	0.00000
H	0.00000	-2.96788	-0.92956
H	0.00000	-2.96788	0.92956

TS1			
C	2.96487	-0.70141	-0.08041
C	2.96481	0.70144	-0.08044
C	4.11822	1.42534	0.18707
C	5.28992	0.70063	0.45894
C	5.28998	-0.70039	0.45896
C	4.11833	-1.42521	0.18712
H	4.11561	2.51093	0.18630
H	6.21104	1.23470	0.67268
H	6.21114	-1.23437	0.67273
H	4.11581	-2.51080	0.18639
C	1.59172	1.17801	-0.39883
C	1.59182	-1.17811	-0.39879
N	0.83339	-0.00008	-0.57825
O	1.15564	2.31006	-0.49855
O	1.15583	-2.31019	-0.49847
O	-0.45635	-0.00014	-0.93490
C	-3.99910	1.21714	0.40176
C	-3.32175	-0.00005	0.66950
C	-3.99922	-1.21714	0.40169
C	-5.29767	-1.21465	-0.09746
C	-5.95285	0.00012	-0.35013
C	-5.29755	1.21481	-0.09739
H	-3.49181	2.15869	0.59736
H	-3.49203	-2.15877	0.59722
H	-5.80546	-2.15568	-0.29064
H	-6.96689	0.00018	-0.74032
H	-5.80524	2.15590	-0.29050
C	-1.95748	-0.00013	1.16300
H	-1.16776	-0.00015	0.07128
H	-1.61159	0.91567	1.64651
H	-1.61168	-0.91598	1.64648

III			
C	0.76277	1.24904	-0.02672
C	-0.76283	1.24908	0.02676
C	-1.11205	-0.23991	0.03834
C	1.11203	-0.24004	-0.03827
H	1.16360	1.71029	-0.93514
H	1.24343	1.72425	0.83288
H	-1.24344	1.72432	-0.83286
H	-1.16363	1.71045	0.93512
O	2.24441	-0.71907	0.02379
O	-2.24431	-0.71918	-0.02416
N	-0.00005	-1.06748	0.00033

7a			
C	0.76946	-1.26012	0.00343
C	1.16844	0.20911	0.00033
C	-1.16844	0.20911	-0.00031
H	1.20682	-1.74750	-0.87338
H	1.19728	-1.73905	0.88975
O	-2.29374	0.68093	0.00160
O	2.29374	0.68093	-0.00161

TS2			
C	4.09990	1.34984	-0.34525
C	4.49914	-0.13006	-0.47765
C	3.21136	-0.91277	-0.21298
C	2.60755	1.32898	-0.00695

H	4.63283	1.87363	0.45446	C	3.17871	1.30438	0.10831
H	4.24116	1.92171	-1.26775	C	3.20530	-1.45226	-0.35266
H	5.25402	-0.44228	0.25079	H	1.35827	-1.50168	0.75530
H	4.87071	-0.39241	-1.47314	C	4.21899	0.73429	-0.62202
O	1.89514	2.30179	0.19411	H	3.16529	2.37752	0.28479
O	3.08768	-2.12908	-0.21427	C	4.23750	-0.64867	-0.85562
N	2.20985	0.00685	0.02617	H	3.21277	-2.52382	-0.53494
C	-1.68723	-1.37047	1.14975	H	5.01710	1.36170	-1.01072
H	-1.51200	-1.17543	2.20295	H	5.04936	-1.09424	-1.42473
H	-1.49043	-2.38488	0.81729	H	0.07177	1.06556	0.79886
C	-2.68716	-0.60795	0.46801				
C	-3.20806	0.58247	1.04363				
C	-3.17897	-1.02804	-0.79743	TS4			
C	-4.19038	1.31009	0.38564	C	2.70259	-0.39377	0.01005
H	-2.83215	0.91369	2.00794	C	1.86289	0.68293	0.31877
C	-4.16113	-0.29467	-1.44905	C	2.34550	1.98229	0.38861
H	-2.78069	-1.93374	-1.24673	C	3.71266	2.17879	0.14055
C	-4.66876	0.87541	-0.86129	C	4.55255	1.10148	-0.17085
H	-4.59011	2.21448	0.83478	C	4.05405	-0.20859	-0.24247
H	-4.53809	-0.62568	-2.41228	C	0.48011	0.16313	0.52618
H	-5.43679	1.44760	-1.37404	C	1.89373	-1.65023	0.00822
Cl	0.29602	-0.55667	0.45913	N	0.57462	-1.20929	0.27746
			O	-0.54092	0.76868	0.80753	
			O	2.23392	-2.79507	-0.17732	
			O	-0.42833	-2.08757	0.59370	
			N	-2.99468	-0.77471	-0.16219	

TS3			
C	-3.61956	-1.30295	0.02981
C	-4.18098	0.05961	-0.35655
C	-2.92454	0.98543	-0.37480
C	-2.10795	-1.03636	0.10006
H	-3.96965	-1.67177	0.99918
H	-3.81389	-2.08607	-0.71031
H	-4.89525	0.47364	0.36106
H	-4.63885	0.09520	-1.34967
O	-1.25638	-1.85631	0.44218
O	-2.99647	2.18971	-0.58421
N	-1.77357	0.25813	-0.21084
C	1.00225	1.11441	1.39554
H	0.79788	0.55333	2.31640
H	1.19298	2.16106	1.65387
C	2.13686	0.50997	0.62620
C	2.16842	-0.88194	0.38057

TS4			
C	2.70259	-0.39377	0.01005
C	1.86289	0.68293	0.31877
C	2.34550	1.98229	0.38861
C	3.71266	2.17879	0.14055
C	4.55255	1.10148	-0.17085
C	4.05405	-0.20859	-0.24247
C	0.48011	0.16313	0.52618
C	1.89373	-1.65023	0.00822
N	0.57462	-1.20929	0.27746
O	-0.54092	0.76868	0.80753
O	2.23392	-2.79507	-0.17732
O	-0.42833	-2.08757	0.59370
N	-2.99468	-0.77471	-0.16219
C	-3.69681	0.13540	0.64964
C	-2.45799	-0.04599	-1.23929
C	-3.67426	1.55062	0.07196
O	-4.28512	-0.22426	1.65951
C	-2.98306	1.38231	-1.28953
O	-1.63672	-0.56332	-1.98756
H	1.68818	2.81174	0.63195
H	4.12742	3.18168	0.19205
H	5.60705	1.28401	-0.35849
H	4.69691	-1.04986	-0.48364
H	-1.05083	-2.08034	-0.16399
H	-4.69085	1.95151	0.02428
H	-3.09569	2.18338	0.75364
H	-3.68060	1.47487	-2.13139
H	-2.15541	2.07313	-1.46928

Cartesian coordinates of all computed structures at the M06/6-

31+G(d) (solvent: CH₃CN) level of theory

1a			
C	0.19152	1.20313	-0.00466
C	0.91331	0.00655	-0.00571
C	0.20001	-1.19819	-0.00466
C	-1.19205	-1.20695	0.00100
C	-1.90069	-0.00492	0.00441
C	-1.20318	1.20082	0.00109
H	0.73193	2.15087	-0.00914
H	0.74933	-2.14122	-0.00911
H	-1.72769	-2.15562	0.00090
H	-2.98960	-0.01000	0.00749
H	-1.74566	2.14556	0.00112
C	2.41379	0.00286	0.00454
H	2.82069	1.00766	-0.16198
H	2.80507	-0.35750	0.96617
H	2.81961	-0.65957	-0.77148

2a			
C	1.87566	0.75995	0.08126
C	1.87566	-0.75996	-0.08125
C	0.42978	-1.17769	-0.11711
C	0.42978	1.17769	0.11711
H	2.35450	1.09525	1.00828
H	2.36956	1.28496	-0.74402
H	2.36956	-1.28496	0.74402
H	2.35450	-1.09525	-1.00827
O	-0.04290	2.28360	0.22399
O	-0.04291	-2.28360	-0.22399
N	-0.32059	0.00000	0.00000
Cl	-2.01150	0.00000	-0.00001

3a			
C	0.49997	0.69818	0.00236
C	0.49996	-0.69817	0.00235
C	1.67750	-1.42161	0.00401
C	2.87499	-0.69780	0.00243
C	2.87499	0.69781	0.00237
C	1.67750	1.42162	0.00390
H	1.67054	-2.50966	0.00513

H	3.82305	-1.23144	0.00087
H	3.82306	1.23144	0.00080
H	1.67055	2.50967	0.00504
C	-0.90282	-1.17540	0.00461
C	-0.90282	1.17541	0.00501
N	-1.66595	-0.00001	0.09356
O	-1.35037	-2.29837	-0.04644
O	-1.35039	2.29835	-0.04651
O	-3.01171	0.00002	-0.10787
H	-3.42136	-0.00011	0.77748

4a			
C	-1.97373	-1.20731	-0.15908
C	-0.64445	-1.20620	0.25423
C	0.02888	0.00015	0.46469
C	-0.64465	1.20634	0.25398
C	-1.97394	1.20713	-0.15933
C	-2.64015	-0.00016	-0.36690
H	-2.49143	-2.15205	-0.31646
H	-0.12081	-2.14895	0.41770
H	-0.12118	2.14921	0.41725
H	-2.49180	2.15175	-0.31690
H	-3.68052	-0.00028	-0.68772
C	1.45490	0.00032	0.89250
H	1.72284	0.89410	1.46200
H	1.72289	-0.89309	1.46257
Cl	2.57758	-0.00013	-0.55994

I			
C	-0.44925	0.70100	0.00022
C	-0.44925	-0.70100	0.00024
C	-1.63115	-1.42249	0.00011
C	-2.82531	-0.69928	0.00002
C	-2.82531	0.69928	0.00003
C	-1.63115	1.42249	0.00012
H	-1.62360	-2.51011	0.00011
H	-3.77386	-1.23140	-0.00005
H	-3.77386	1.23140	-0.00004
H	-1.62360	2.51011	0.00010

C	0.93151	-1.19886	0.00043
C	0.93151	1.19886	0.00026
N	1.74848	0.00000	-0.00017
O	1.38898	-2.31171	-0.00012
O	1.38898	2.31171	-0.00007
O	3.00278	0.00000	-0.00074

N	0.00000	-0.95846	0.00000
H	0.00000	-1.97469	0.00000
C	0.75934	1.24837	-0.07918
H	1.28512	1.73458	0.75055
H	1.10237	1.73216	-1.00091

II			
C	0.00000	1.13127	1.21094
C	0.00000	-0.25251	1.21704
C	0.00000	-0.99221	0.00000
C	0.00000	-0.25251	-1.21704
C	0.00000	1.13127	-1.21094
C	0.00000	1.83613	0.00000
H	0.00000	1.67579	2.15432
H	0.00000	-0.79867	2.16096
H	0.00000	-0.79867	-2.16096
H	0.00000	1.67579	-2.15432
H	0.00000	2.92465	0.00000
C	0.00000	-2.39527	0.00000
H	0.00000	-2.95796	-0.93223
H	0.00000	-2.95796	0.93223

TS2			
C	4.22719	0.83330	-0.59769
C	4.26940	-0.69227	-0.63549
C	2.88340	-1.14019	-0.20864
C	2.81835	1.18230	-0.15269
H	4.93597	1.27445	0.11273
H	4.41105	1.30109	-1.57184
H	5.00296	-1.12832	0.05256
H	4.47595	-1.10105	-1.63138
O	2.37019	2.29826	0.02985
O	2.49978	-2.28742	-0.08093
N	2.13428	-0.00360	0.00123
C	-1.64579	-0.14647	1.76034
H	-1.47765	0.72672	2.38897
H	-1.49198	-1.10819	2.24826
C	-2.63569	-0.06188	0.73113
C	-3.10209	1.19567	0.28662
C	-3.13621	-1.23514	0.12351
C	-4.05231	1.27260	-0.71682
H	-2.70425	2.10061	0.74550
C	-4.08747	-1.15121	-0.87865
H	-2.76618	-2.20362	0.45931
C	-4.54657	0.10108	-1.30072
H	-4.41520	2.24173	-1.05258
H	-4.47805	-2.05715	-1.33727
H	-5.29309	0.16577	-2.09009
Cl	0.23220	-0.07178	0.68298

III			
C	0.74547	1.22421	-0.09193
C	-0.77211	1.21201	0.11753
C	-1.13966	-0.25817	0.18195
C	1.13965	-0.23992	-0.13419
H	1.06136	1.69285	-1.03086
H	1.29150	1.71801	0.71998
H	-1.33789	1.67433	-0.69916
H	-1.08558	1.69817	1.04852
O	2.18256	-0.72575	-0.50575
O	-2.23634	-0.76208	0.11002
N	0.05639	-0.96679	0.38783

7a			
C	-0.75934	1.24837	0.07918
C	-1.15735	-0.20744	0.11991
C	1.15735	-0.20744	-0.11991
H	-1.28512	1.73458	-0.75055
H	-1.10237	1.73216	1.00091
O	2.27345	-0.67163	-0.23497
O	-2.27345	-0.67163	0.23497

TS3			
C	1.72130	-1.75827	0.23654
C	2.90165	-1.13628	-0.49278
C	2.81881	0.34107	-0.13306
C	1.09189	-0.57703	0.95122
H	0.98018	-2.22064	-0.42799
H	2.00641	-2.51842	0.97425
H	2.84911	-1.23638	-1.58395
H	3.87649	-1.53010	-0.18073
O	0.02998	-0.61909	1.58014

O	3.54685	1.21095	-0.59709	H	-0.55777	0.00062	-2.00685
N	1.81874	0.56959	0.79065	C	-3.40556	-0.00646	0.64897
C	-0.43516	2.09912	-0.32277	H	-2.51059	1.79855	1.39990
H	-0.06271	2.33708	-1.32690	C	-3.32540	-1.02591	-0.31191
H	-0.75065	3.00212	0.21128	H	-2.24561	-1.82344	-2.00955
C	-1.42993	1.03324	-0.31393	H	-4.20572	-0.01863	1.38527
C	-1.36374	-0.01403	-1.27203	H	-4.06226	-1.82665	-0.31185
C	-2.46674	1.00369	0.65678	H	0.48722	1.71283	0.23003
C	-2.29875	-1.02563	-1.27235				

**Cartesian coordinates of all computed structures at the M06-2X/6-31+G(d)
(solvent: CH₃CN) level of theory**

1a				4a			
C	-0.19718	-1.20076	-0.00848	H	-1.67707	-2.50913	0.00502
C	-0.91284	0.00250	-0.01098	C	0.90863	1.17304	0.00639
C	-0.19400	1.20269	-0.00850	C	0.90863	-1.17303	0.00608
C	1.20154	1.20377	0.00199	N	1.66922	-0.00001	0.09623
C	1.90269	-0.00185	0.00792	O	1.35741	2.29256	-0.04707
C	1.19726	-1.20615	0.00194	O	1.35739	-2.29256	-0.04709
H	-0.74059	-2.14337	-0.01676	O	3.01507	0.00001	-0.11111
H	-0.73403	2.14708	-0.01689	H	3.42754	-0.00009	0.77186
H	1.74051	2.14726	0.00203				
H	2.98886	-0.00374	0.01330				
H	1.73367	-2.15112	0.00189				
C	-2.42140	0.00101	0.00870				
H	-2.82438	-0.82541	-0.58428				
H	-2.79675	-0.11579	1.03188				
H	-2.82371	0.93782	-0.38671				
2a							
C	0.01731	-0.76708	-1.88368				
C	-0.01731	0.76708	-1.88368				
C	-0.00337	1.18075	-0.42708				
C	0.00337	-1.18075	-0.42708				
H	-0.84436	-1.21877	-2.38040				
H	0.92193	-1.17310	-2.34319				
H	-0.92193	1.17310	-2.34319				
H	0.84436	1.21877	-2.38040				
O	0.00000	-2.28912	0.04799				
O	0.00000	2.28912	0.04799				
N	0.00000	0.00000	0.32347				
Cl	0.00000	0.00000	2.00848				
3a							
C	-0.50568	-0.69773	0.00476				
C	-0.50568	0.69773	0.00477				
C	-1.68323	1.42370	0.00425				
C	-2.88156	0.69869	0.00068				
C	-2.88156	-0.69869	0.00072				
C	-1.68322	-1.42370	0.00433				
H	-1.67708	2.50913	0.00496				
H	-3.82746	1.23101	-0.00267				
H	-3.82746	-1.23102	-0.00262				
I							
C	0.45532	-0.69979	0.00003				
C	0.45532	0.69979	0.00001				
C	1.63690	1.42416	0.00009				
C	2.83231	0.69986	0.00012				
C	2.83231	-0.69986	0.00010				
C	1.63690	-1.42416	0.00004				
H	1.63006	2.50929	0.00008				
H	3.77850	1.23127	0.00016				
H	3.77850	-1.23127	0.00013				
H	1.63006	-2.50929	0.00005				
C	-0.93795	1.19624	-0.00013				
C	-0.93795	-1.19624	0.00006				
N	-1.75198	0.00000	-0.00007				
O	-1.39557	2.30516	-0.00003				

O	-1.39557	-2.30516	-0.00008
O	-3.00789	0.00000	-0.00012

II

C	0.00000	1.21028	-1.13346
C	0.00000	1.21570	0.25303
C	0.00000	0.00000	0.98991
C	0.00000	-1.21570	0.25303
C	0.00000	-1.21028	-1.13346
C	0.00000	0.00000	-1.83814
H	0.00000	2.15160	-1.67562
H	0.00000	2.15773	0.79581
H	0.00000	-2.15773	0.79581
H	0.00000	-2.15160	-1.67562
H	0.00000	0.00000	-2.92363
C	0.00000	0.00000	2.40276
H	0.00000	-0.92936	2.96056
H	0.00000	0.92936	2.96056

III

C	0.76270	1.23717	-0.02819
C	-0.76502	1.23694	0.03834
C	-1.13552	-0.23476	0.01109
C	1.13518	-0.23405	0.00466
H	1.15430	1.67495	-0.95052
H	1.24130	1.74699	0.81152
H	-1.24587	1.74507	-0.80066
H	-1.15452	1.67566	0.96151
O	2.25387	-0.70696	-0.03263
O	-2.25170	-0.70883	-0.06325
N	0.00049	-1.07832	0.08425

7a

C	-0.76238	1.25588	0.07842
C	-1.15645	-0.20988	0.12160
C	1.15645	-0.20988	-0.12160
H	-1.28282	1.73194	-0.75628
H	-1.10255	1.73493	0.99986
O	2.27021	-0.67486	-0.23997
O	-2.27021	-0.67486	0.23997
N	0.00000	-0.96089	0.00000
H	0.00000	-1.97619	0.00000
C	0.76238	1.25588	-0.07843
H	1.28282	1.73194	0.75628

H	1.10255	1.73493	-0.99986
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TS2

C	4.46094	-0.04030	-0.21042
C	3.93183	-1.47868	-0.17000
C	2.41617	-1.33920	-0.06341
C	3.21756	0.84076	-0.13896
H	5.10555	0.20807	0.63648
H	5.00360	0.19770	-1.12862
H	4.28641	-2.04873	0.69240
H	4.16444	-2.05082	-1.07166
O	3.18703	2.05450	-0.11336
O	1.60800	-2.23962	0.03932
N	2.12159	0.00645	-0.11431
C	-1.53903	1.59167	0.84082
H	-1.41623	2.54727	0.33834
H	-1.33231	1.59981	1.90696
C	-2.57953	0.71153	0.36801
C	-3.16027	0.90531	-0.90345
C	-3.00812	-0.37233	1.16349
C	-4.15573	0.05014	-1.35452
H	-2.82461	1.73422	-1.52132
C	-4.00101	-1.22613	0.70613
H	-2.55281	-0.52834	2.13810
C	-4.57579	-1.01642	-0.55253
H	-4.60705	0.20736	-2.32903
H	-4.33081	-2.05599	1.32308
H	-5.35185	-1.68669	-0.90964
Cl	0.27072	0.67964	0.18862

TS3

C	-1.05916	0.69464	-1.66641
C	-1.74891	-0.66490	-1.51778
C	-2.01358	-0.77518	-0.02153
C	-1.24931	1.36241	-0.31344
H	0.02285	0.59685	-1.82115
H	-1.46084	1.32443	-2.46283
H	-1.14241	-1.51631	-1.83506
H	-2.70675	-0.70998	-2.04539
O	-0.85568	2.46020	0.02002
O	-2.25428	-1.78233	0.60768
N	-1.95692	0.50760	0.54871
C	0.11746	-0.00934	2.27746
H	-0.09558	-0.94381	2.80295

H	0.40343	0.78226	2.97478	H	1.80641	1.85119	1.25631
C	1.06987	-0.17066	1.14516	C	2.76082	-0.42474	-1.08460
C	1.14205	-1.38052	0.43511	H	2.02618	-2.44898	-1.20851
C	1.86017	0.90804	0.71819	H	3.30897	1.62515	-0.69999
C	1.97964	-1.50641	-0.67073	H	3.41512	-0.52327	-1.94584
H	0.53952	-2.22647	0.75870	H	-0.90266	0.32190	1.86480
C	2.70159	0.78180	-0.38499				

7. Time-course plot of the chlorination reaction

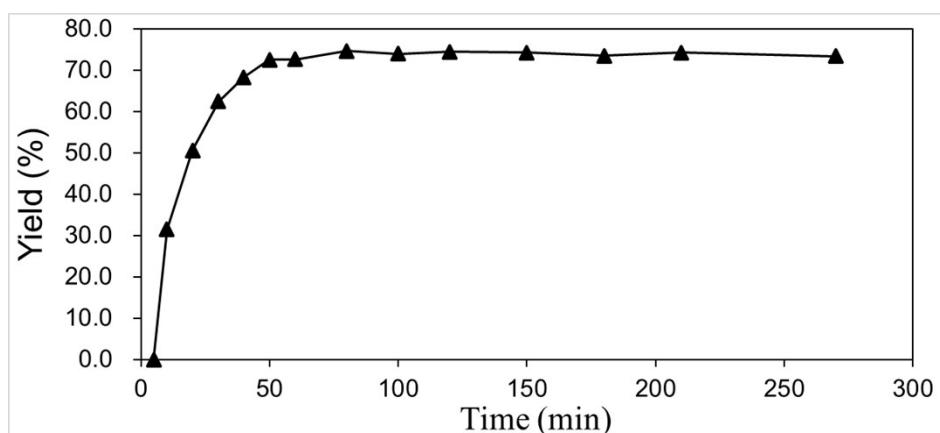
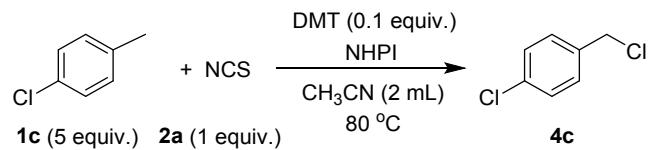


Figure S1. Time-course plot of the chlorination reaction.^a

^a Conditions: toluene (**1a**, 6.91g, 75 mmol), NCS (**2a**, 2g, 15 mmol), NHPI (**3a**, 10 mol%), DDQ (0.5 mol%), CH₃CN (60 ml), 80 °C; yield is determined by GC using diphenyl as internal standard; the calculation of yield is on the basis of NCS (**2a**).

8. Chain length experiment

Table S4. Radical chain length initiated by NHPI.^a

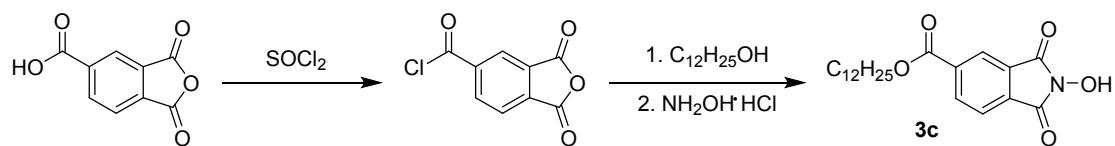


NHPI / equiv	20 min		40min	
	Y _{4c} / %	Chain Length 1 ^a	Y _{4c} / %	Chain Length 2 ^a
0.03	33.7 (±1.6)	11.2 (±0.5)	44.6 (±0.05)	14.9 (±0.02)
0.07	40.1 (±0.2)	5.7 (±0.03)	45.9 (±0.6)	6.6 (±0.1)
0.1	41.9 (±0.1)	4.2 (±0.01)	47.1 (±0.3)	4.7 (±0.03)
0.2	41.8 (±0.5)	2.1 (±0.02)	48.7 (±1.9)	2.4 (±0.1)
0.3	44.1 (±0.5)	1.5 (±0.02)	49.3 (±0.05)	1.6 (±0.002)
Average Chain Length ^b		4.9 (±0.1)	6.0 (±0.001)	

^a yield determined by GC with diphenyl as internal standard, on the basis of NCS; ^b chain length refers to yield divide NHPI in mol%; ^c average chain length refers to sum of chain length divide sum of experiments.

9. Preparation and characterization of NO-H agents

The procedures for the synthesis of 3c



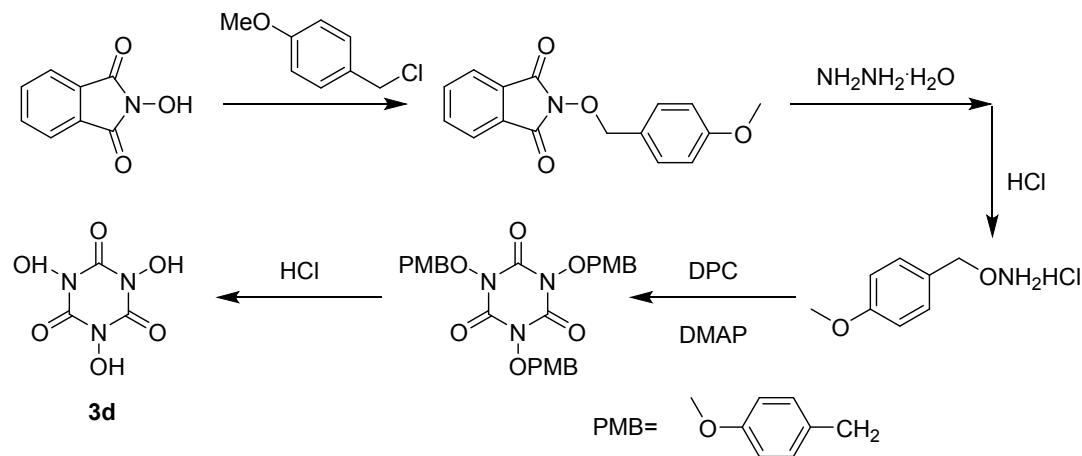
Scheme S3. Preparation of 3c.

3c was prepared according to the literature procedure in Scheme S3.^{12, 13} Trimellitic acid anhydride (7.68 g, 0.04 mol) was boiled in thionyl chloride (10 mL) for 12 h. The anhydride acid chloride was isolated by vacuum distillation of the solvent. Remove the thionyl chloride with 20ml petroleum ether three times giving rise to white solid. Pyridine (10 mmol, 0.75 mL) was added in a few portions to a stirred mixture of trimellitic anhydride chloride (10 mmol, 2.16 g) in toluene (10 mL) at 0°C. Next, a solution of dodecyl alcohol (10 mmol, 2.25 mL) in toluene (5 mL) was added dropwise over 30 min. The mixture was stirred for 4 h at room temperature, then for 10 h at 85°C and for 7 h at 100°C. Toluene was evaporated, and product in form of white paste was obtained. Dried pyridine (12.5 mL) and hydroxylamine hydrochloride (11 mmol, 0.76 g) were added and the reaction mixture was stirred at 90°C for 15 h. The product was poured to 30 mL of water and acidified by hydrochloric acid to pH 1. The precipitate was filtered, and washed with ice water. The product was recrystallised from methanol.

$^1\text{H NMR}$ (600 MHz, DMSO-D_6 , 298 K): $\delta = 11.01$ (s, 1H,N-OH), 8.35 (dd, $^3J_{\text{HH}} = 7.8$

Hz, $^4J_{\text{HH}} = 1.8$ Hz, 1H, Ar-H), 8.17 (s, 1H, Ar-H), 7.97 (d, $^3J_{\text{HH}} = 7.2$ Hz, 1H, Ar-H), 4.32 (t, $^3J_{\text{HH}} = 6.6$ Hz, 2H, CO-OCH₂), 1.73 (m, 2H, OCH₂-CH₂), 1.22 - 1.32 (m, 18H, CH₂), 0.84 (t, 3H, $^3J_{\text{HH}} = 6.6$ Hz, CH₃); $^{13}\text{C}\{\text{H}\}$ NMR (151 MHz, DMSO-D⁶, 298 K): $\delta = 164.2, 163.2$ (Ar-CO-N), 135.2, 135.1, 132.5, 129.4, 123.5, 122.3 (Ar), 65.6 (COO), 31.3 (OCH₂), 29.0, 28.94, 28.88, 28.7, 28.6, 28.0, 25.4 (CH₂), 22.1 (CH₂CH₃), 14.0 (CH₃).

The procedures for the synthesis of 3d



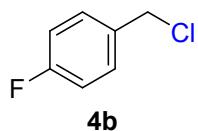
Scheme S4. Preparation of the THICA (**3d**) catalyst.

3d was prepared according to the literature procedure in Scheme S4.¹⁴

^1H NMR (600 MHz, DMSO-D⁶, 298 K): $\delta = 11.04$ (s, 3H, NO-H). $^{13}\text{C}\{\text{H}\}$ NMR (151 MHz, DMSO-D⁶, 298 K): $\delta = 146.7$ (C=O).

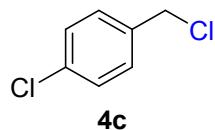
10. Characterization of Chlorinated products

1-(chloromethyl)-4-fluorobenzene (4b)



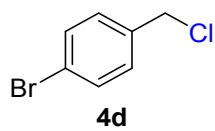
¹H NMR (600 MHz, CDCl₃, 298 K): $\delta = 7.37$ (m, 2H, Ar-H), 7.05 (t, $^3J_{HH} = 8.4$ Hz, 2H, Ar-H), 4.57 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): $\delta = 163.6$, 133.5, 130.5, 115.9 (Ar), 45.5 (CH₂). NMR data correspond to the reported values.¹⁵

Chloro-4-(chloromethyl)benzene (4c).



¹H NMR (600 MHz, CDCl₃, 298 K): $\delta = 7.35\text{--}7.31$ (m, 4H, Ar-H), 4.55 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): $\delta = 136.1$, 134.4, 130.0, 129.0 (Ar), 45.5 (CH₂). NMR data correspond to the reported values.¹⁶

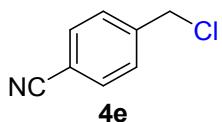
Bromo-4-(chloromethyl)benzene (4d).



¹H NMR (600 MHz, CDCl₃, 298 K): $\delta = 7.49$ (d, $^3J_{HH} = 8.4$ Hz, 2H, Ar-H), 7.27 (d, $^3J_{HH} = 8.4$ Hz, 2H, Ar-H), 4.54 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): $\delta =$

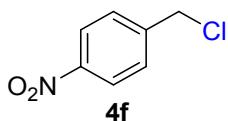
136.6, 132.0, 130.4, 122.6 (Ar), 45.5 (CH₂). NMR data correspond to the reported values.¹⁷

4-(chloromethyl)benzonitrile (4e).



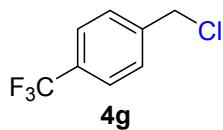
¹H NMR (600 MHz, CDCl₃, 298 K): δ = 7.64 (d, 2H, ³J_{HH} = 8.4 Hz, Ar-H), 7.50 (d, 2H, ³J_{HH} = 8.4 Hz, Ar-H), 4.59 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 142.5 (CN), 132.5, 129.2, 118.5, 112.2 (Ar), 45.0 (CH₂). NMR data correspond to the reported values.¹⁵

1-(chloromethyl)-4-nitrobenzene (4f).



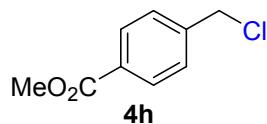
¹H NMR (600 MHz, CDCl₃, 298 K): δ = 8.21 (d, ³J_{HH} = 8.4 Hz, 2H, Ar-H), 7.57 (d, ³J_{HH} = 8.4 Hz, 2H, Ar-H), 4.64 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 147.8, 144.4, 129.4, 124.0 (Ar), 44.6 (CH₂). NMR data correspond to the reported values.¹⁸

1-(chloromethyl)-4-(trifluoromethyl)benzene (4g).



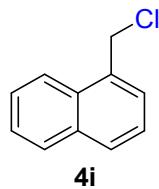
¹H NMR (600 MHz, CDCl₃, 298 K): δ = 7.64 (d, $^3J_{\text{HH}} = 8.4$ Hz, 2H, Ar-H), 7.52 (d, $^3J_{\text{HH}} = 8.4$ Hz, 2H, Ar-H), 4.62 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 141.4 (CF₃), 130.6, 129.0, 125.8, 123.2 (Ar), 45.2 (CH₂). NMR data correspond to the reported values.¹⁹

Methyl 4-(chloromethyl)benzoate (4h).



¹H NMR (600 MHz, CDCl₃, 298 K): δ = 8.02 (d, $^3J_{\text{HH}} = 8.4$ Hz, 2H, Ar-H), 7.45 (d, $^3J_{\text{HH}} = 8.4$ Hz, 2H, Ar-H), 4.60 (s, 2H, CH₂), 3.91 (s, 3H, CH₃). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 166.7 (C=O), 142.3, 130.2, 130.1, 128.6 (Ar), 52.3 (CH₃), 45.5 (CH₂). NMR data correspond to the reported values.²⁰

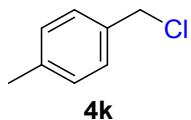
1-(chloromethyl)naphthalene (4i).



¹H NMR (600 MHz, CDCl₃, 298 K): δ = 8.18 (d, $^3J_{\text{HH}} = 8.4$ Hz, 1H, Ar-H), 7.91–7.86 (m, 2H, Ar-H), 7.64–7.61 (m, 1H, Ar-H), 7.57–7.54 (m, 2H, Ar-H), 7.46–7.43 (m, 1H, Ar-H), 5.07 (s, 2 H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 134.1, 133.1,

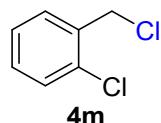
131.2, 129.9, 129.0, 127.8, 126.8, 126.3, 125.4, 123.8 (Ar), 44.7 (CH_2). NMR data correspond to the reported values.²¹

1-(chloromethyl)-4-methylbenzene (4k).



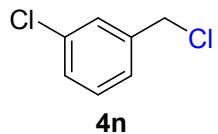
$^1\text{H NMR}$ (600 MHz, CDCl_3 , 298 K): δ = 7.3 (d, 2H, $^3J_{\text{HH}} = 8.4$ Hz), 7.19 (d, 2H, $^3J_{\text{HH}}$ = 7.8 Hz), 4.57 (s, 2H), 2.37 (s, 3 H). **$^{13}\text{C NMR}$** (151 MHz, CDCl_3 , 298 K): δ = 138.4, 134.7, 129.5, 128.7, 46.4 (Ar), 21.3 (CH_2). NMR data correspond to the reported values.¹⁵

1-chloro-2-(chloromethyl)benzene (4m).



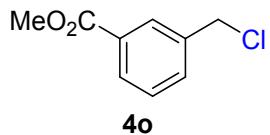
$^1\text{H NMR}$ (600 MHz, CDCl_3 , 298 K): δ = 7.48–7.47 (m, 1H, Ar-H), 7.42–7.40 (m, 1H, Ar-H), 7.29–7.26 (m, 2H, Ar-H), 4.72 (s, 2H, CH_2). **$^{13}\text{C NMR}$** (151 MHz, CDCl_3 , 298 K): δ = 135.1, 134.2, 131.0, 130.0, 129.9, 127.3 (Ar), 43.7 (CH_2). NMR data correspond to the reported values.²¹

1-chloro-3-(chloromethyl)benzene (4n).



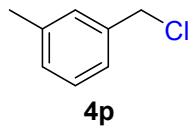
¹H NMR (600 MHz, CDCl₃, 298 K): δ = 7.40 (bs, 1H, Ar-H), 7.31–7.27 (m, 3H, Ar-H), 4.54 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 139.4, 134.6, 130.1, 128.8, 128.7, 126.8 (Ar), 45.3 (CH₂). NMR data correspond to the reported values.¹⁷

methyl 3-(chloromethyl)benzoate (4o).



¹H NMR (600 MHz, CDCl₃, 298 K): δ = 8.05 (s, 1H, Ar-H), 7.98 (d, $^3J_{HH}$ = 7.8 Hz, 1H, Ar-H), 7.57 (d, $^3J_{HH}$ = 7.8 Hz, 1H, Ar-H), 7.43 (t, $^3J_{HH}$ = 7.8 Hz, 1H, Ar-H), 4.61 (s, 2H, CH₂), 3.92 (s, 3H, CH₃). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 166.6 (C=O), 137.9, 133.2, 130.8, 129.7, 129.6, 128.9 (Ar), 52.3 (CH₃), 45.6 (CH₂). NMR data correspond to the reported values.²²

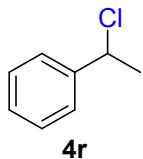
1-(chloromethyl)-3-methylbenzene (4p).



¹H NMR (600 MHz, CDCl₃, 298 K): δ = 7.28–7.15 (m, 4H, Ar-H), 4.57 (s, 2H, CH₂), 2.38 (s, 3H, CH₃). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 138.6, 137.5, 129.4, 129.3, 128.8, 125.8 (Ar), 46.5 (CH₂), 21.4 (CH₃). NMR data correspond to the reported

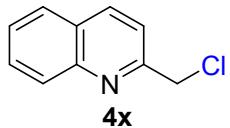
values.²¹

(1-chloroethyl)benzene (4r).



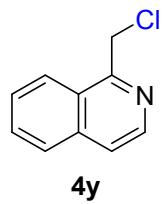
¹H NMR (600 MHz, CDCl₃, 298 K): δ = 7.44–7.30 (m, 5H, Ar), 5.12 (q, 1H, ³J_{HH} = 6.6 Hz, CH), 1.87 (d, 3H, ³J_{HH} = 6.6 Hz, CH₃). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 143.0, 128.8, 128.4, 126.6 (Ar), 58.9 (CH), 26.6 (CH₃). NMR data correspond to the reported values.²³

2-(chloromethyl)quinolone (4x).



¹H NMR (600 MHz, CDCl₃, 298 K): δ = 8.18 (d, 1H, ³J_{HH} = 8.4 Hz, Ar-H), 8.07 (d, 1H, ³J_{HH} = 8.4 Hz, Ar-H), 7.81 (d, 1H, ³J_{HH} = 7.8 Hz, Ar-H), 7.73 (t, 1H, ³J_{HH} = 7.8 Hz, Ar-H), 7.60–7.54 (m, 2H, Ar), 4.84 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 156.8, 147.5, 137.4, 130.1, 129.4, 127.7, 127.6, 127.1, 120.6 (Ar), 47.5 (CH₂). NMR data correspond to the reported values.¹⁵

1-(chloromethyl)isoquinoline (4y).



¹H NMR (600 MHz, CDCl₃, 298 K): δ = 8.48 (d, 1H, ³J_{HH} = 5.4 Hz, Ar-H), 8.25 (d, 1H, ³J_{HH} = 8.4 Hz, Ar-H), 7.86 (d, 1H, ³J_{HH} = 7.8 Hz, Ar-H), 7.74–7.65 (m, 3H, Ar), 5.15 (s, 2H, CH₂). **¹³C NMR** (151 MHz, CDCl₃, 298 K): δ = 155.8, 142.1, 136.8, 130.5, 128.0, 127.7, 126.6, 125.2, 121.9 (Ar), 45.1 (CH₂). NMR data correspond to the reported values.¹⁵

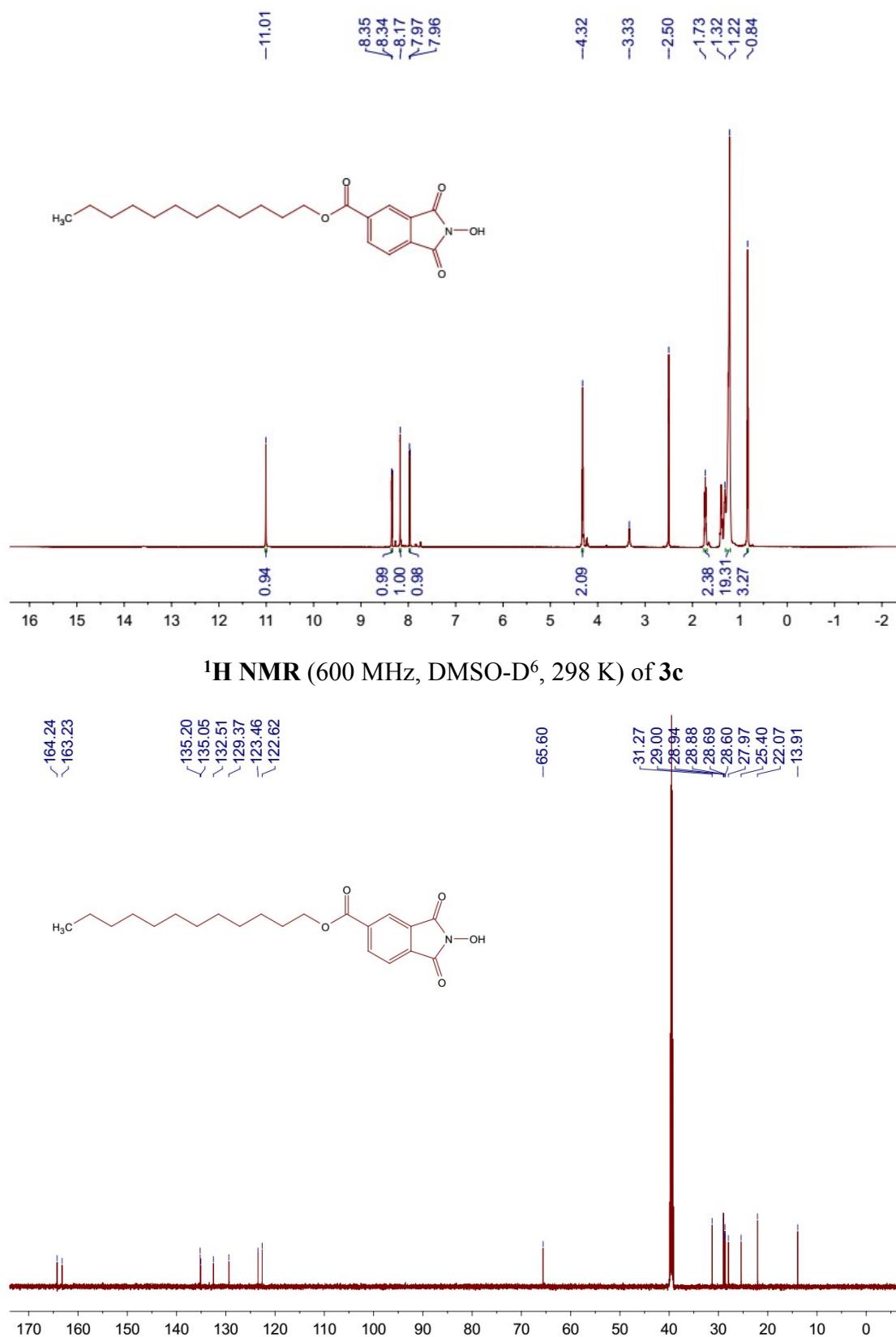
11. References

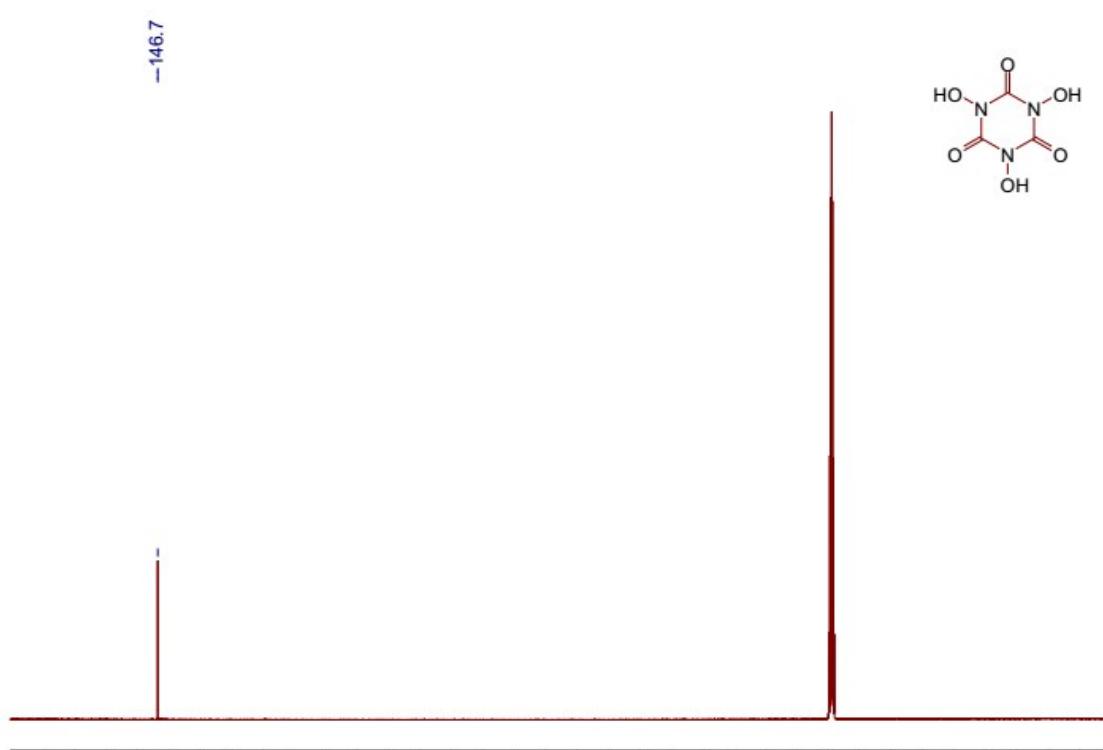
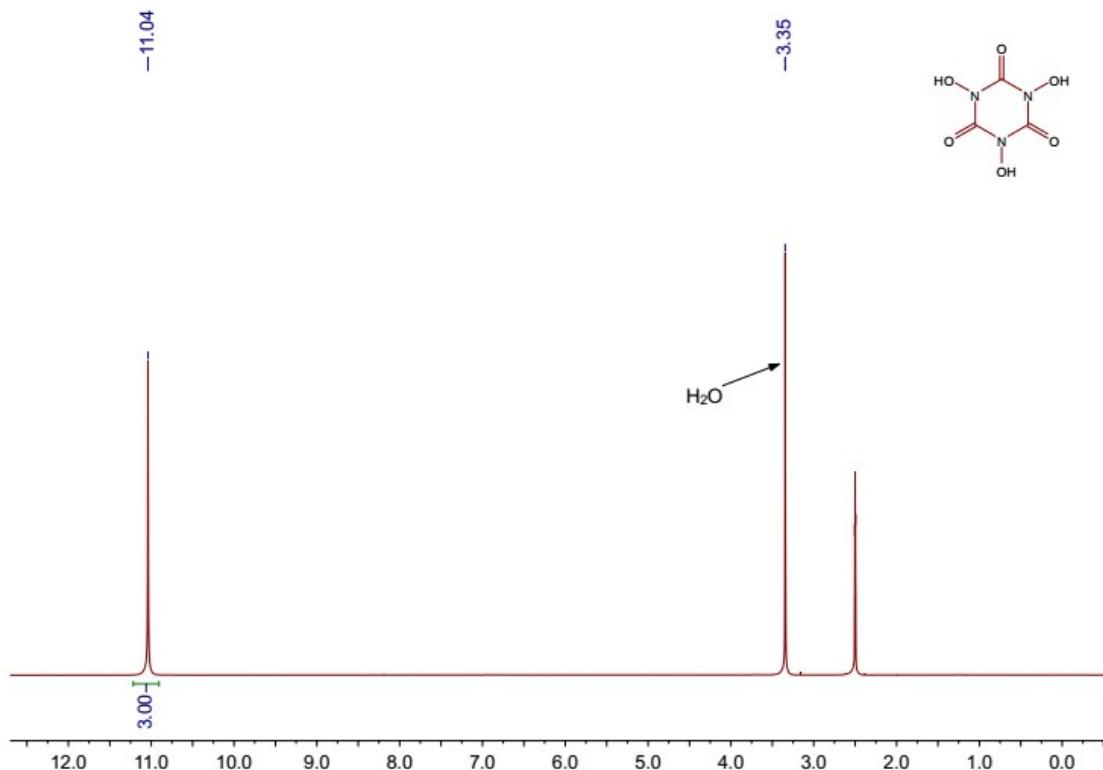
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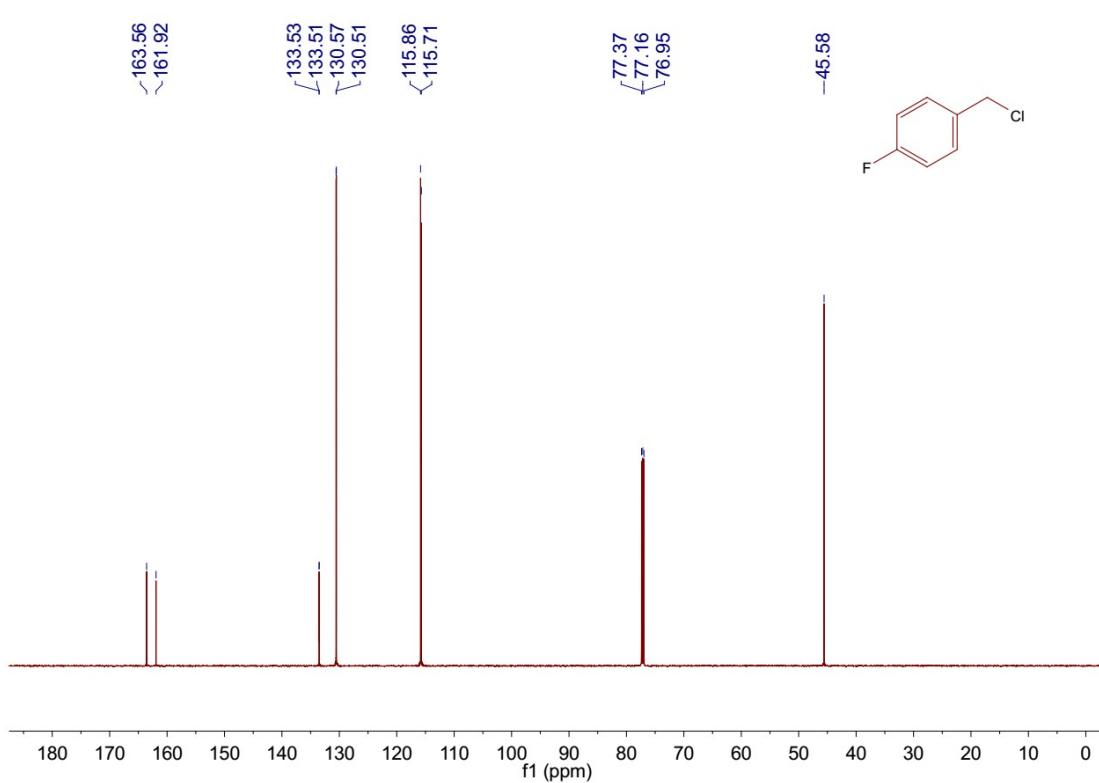
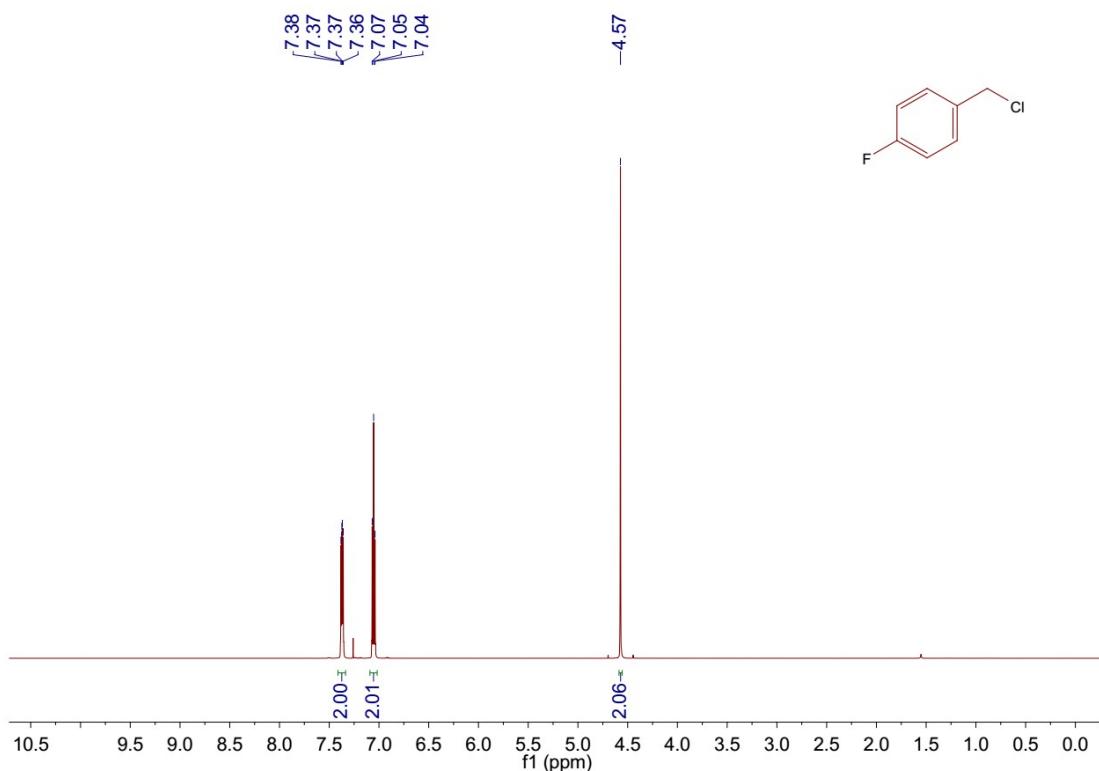
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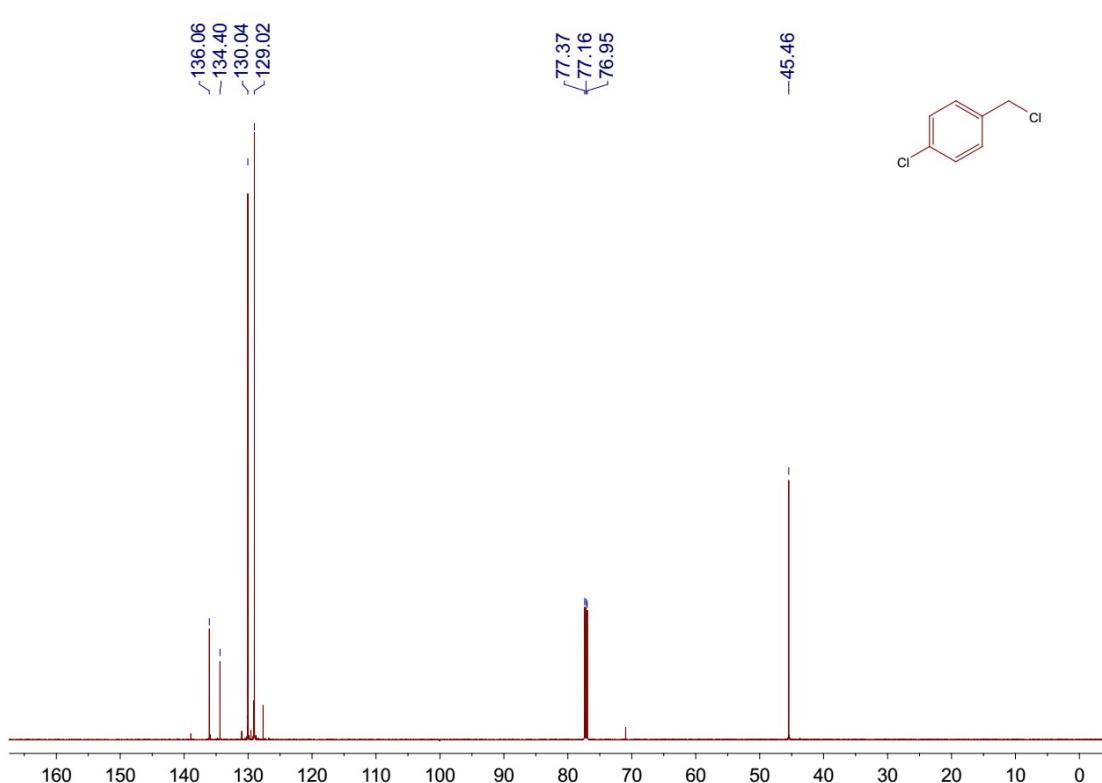
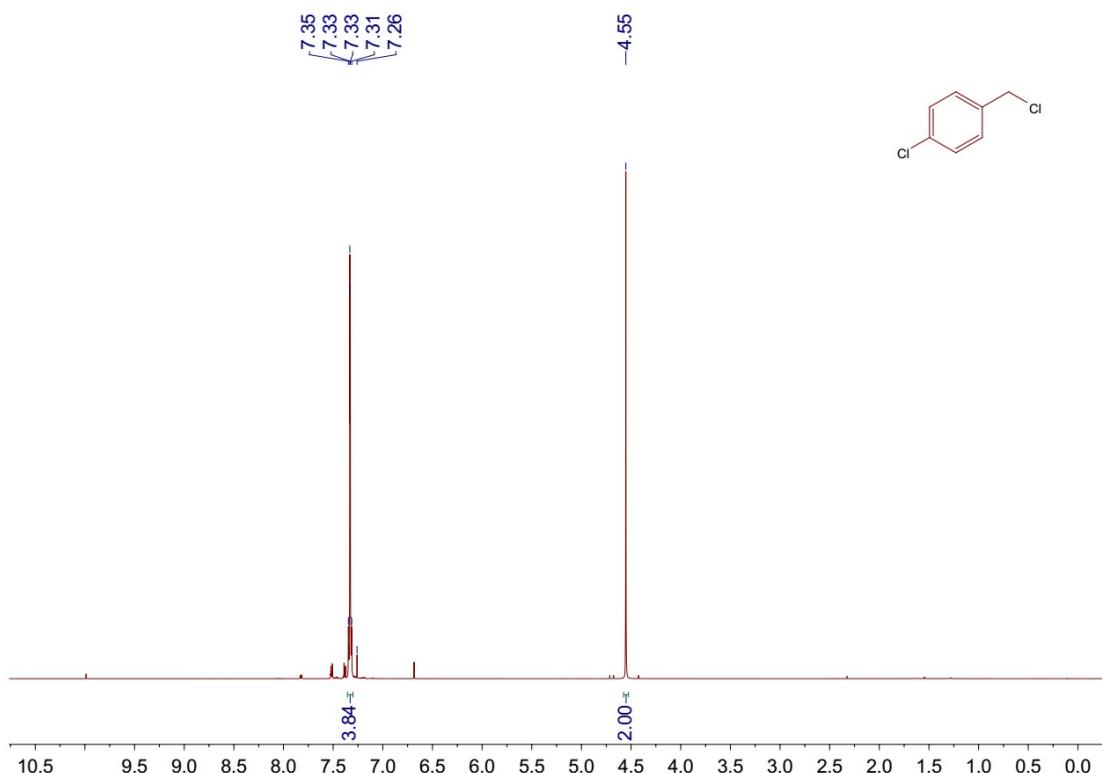
GC and NMR spectral



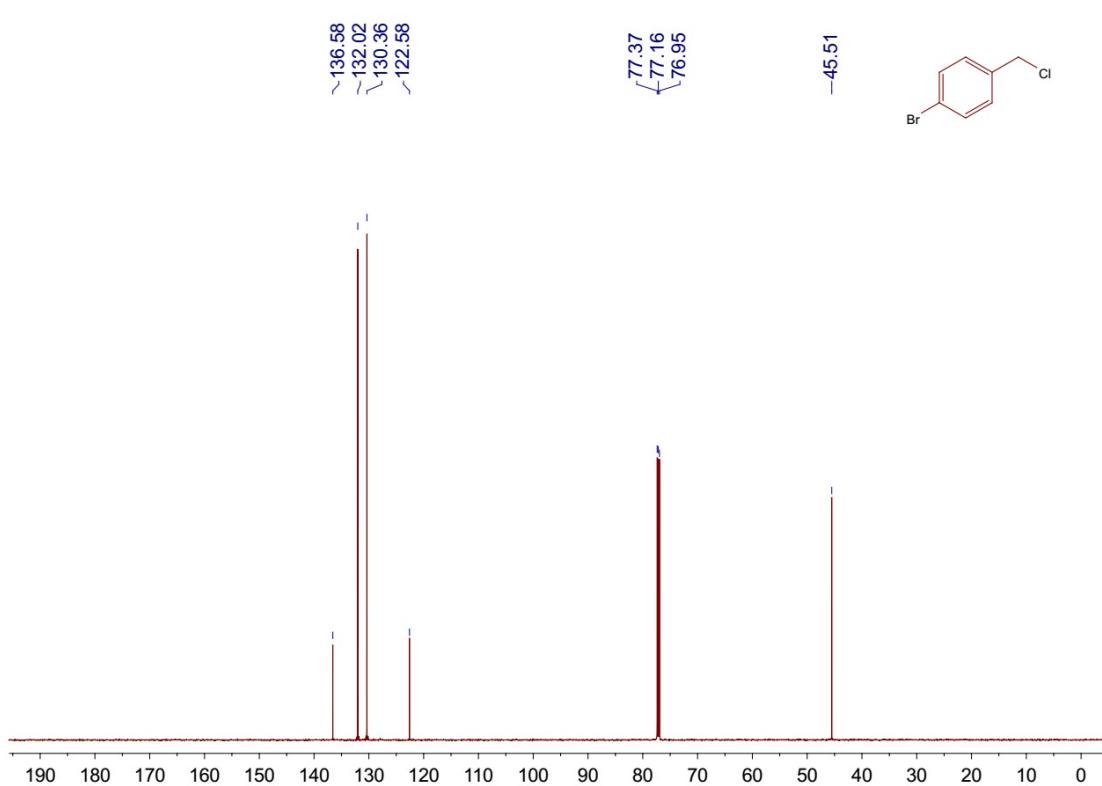
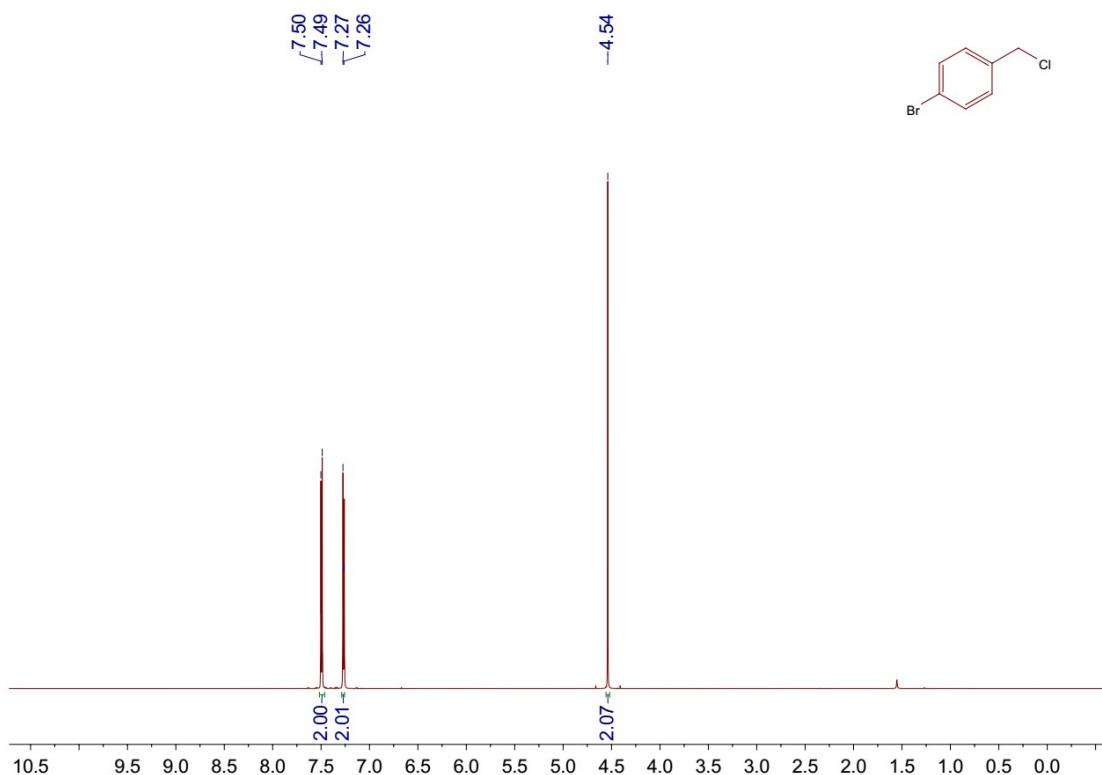


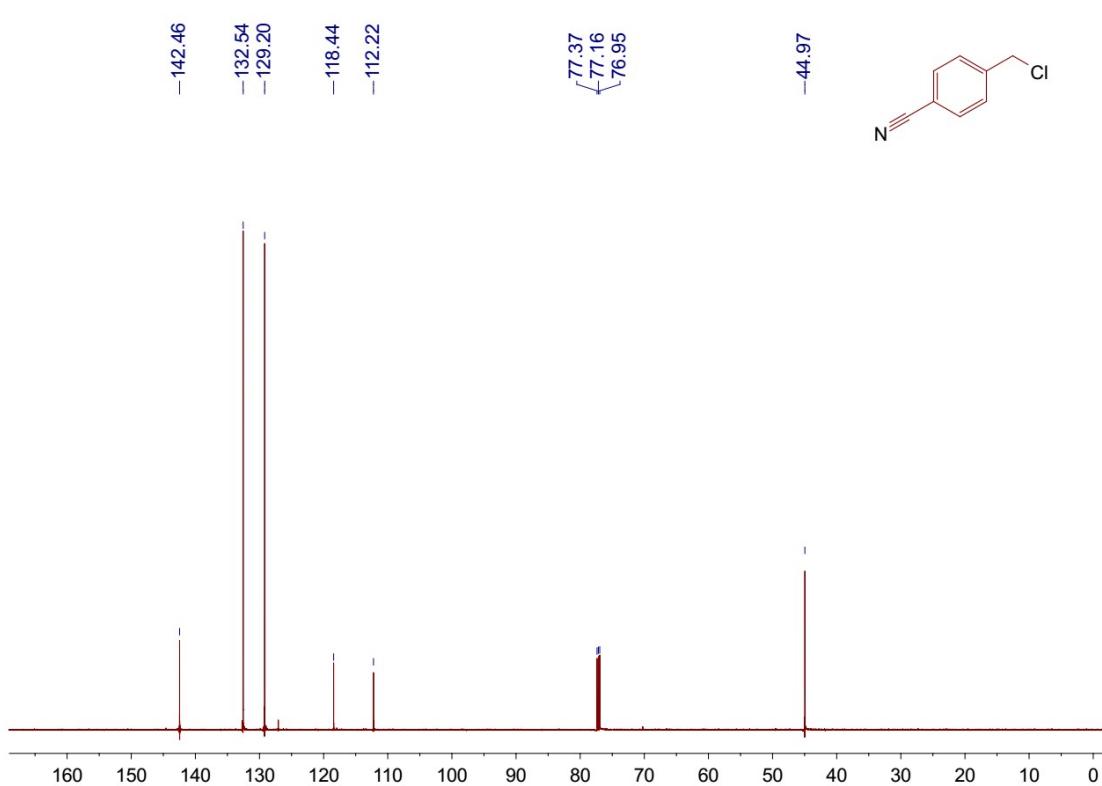
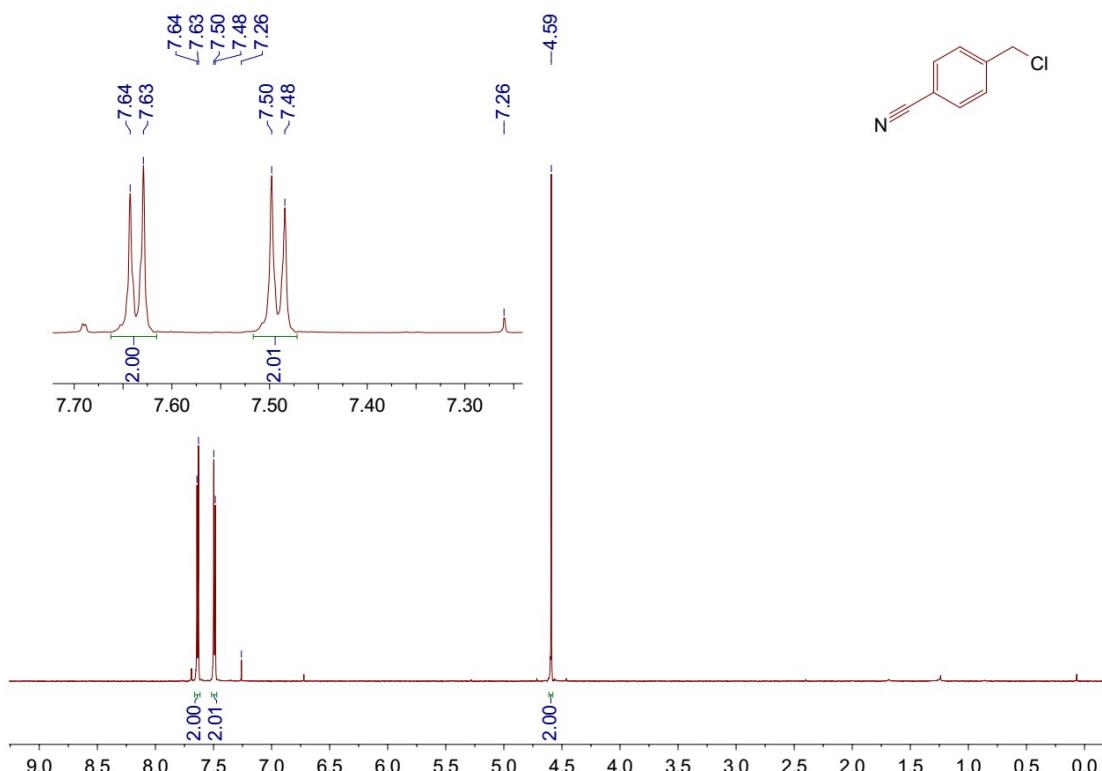
$^{13}\text{C}\{^1\text{H}\}$ NMR (151 MHz, DMSO- D^6 , 298 K) of **3d**

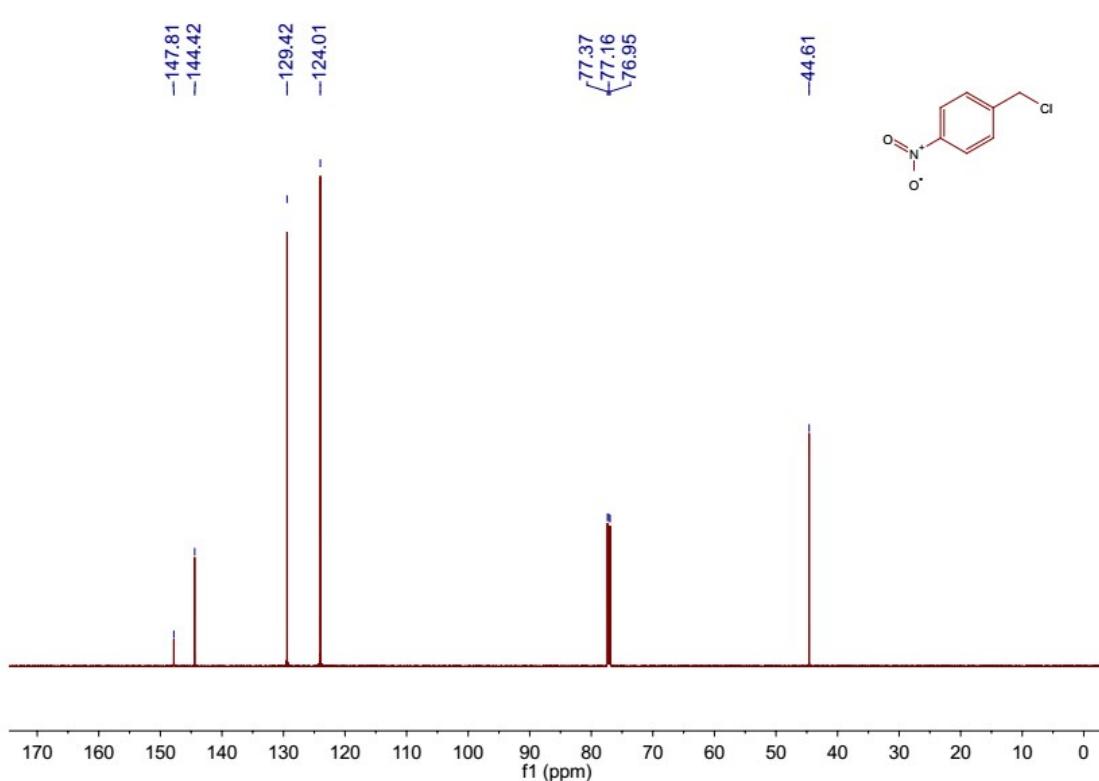
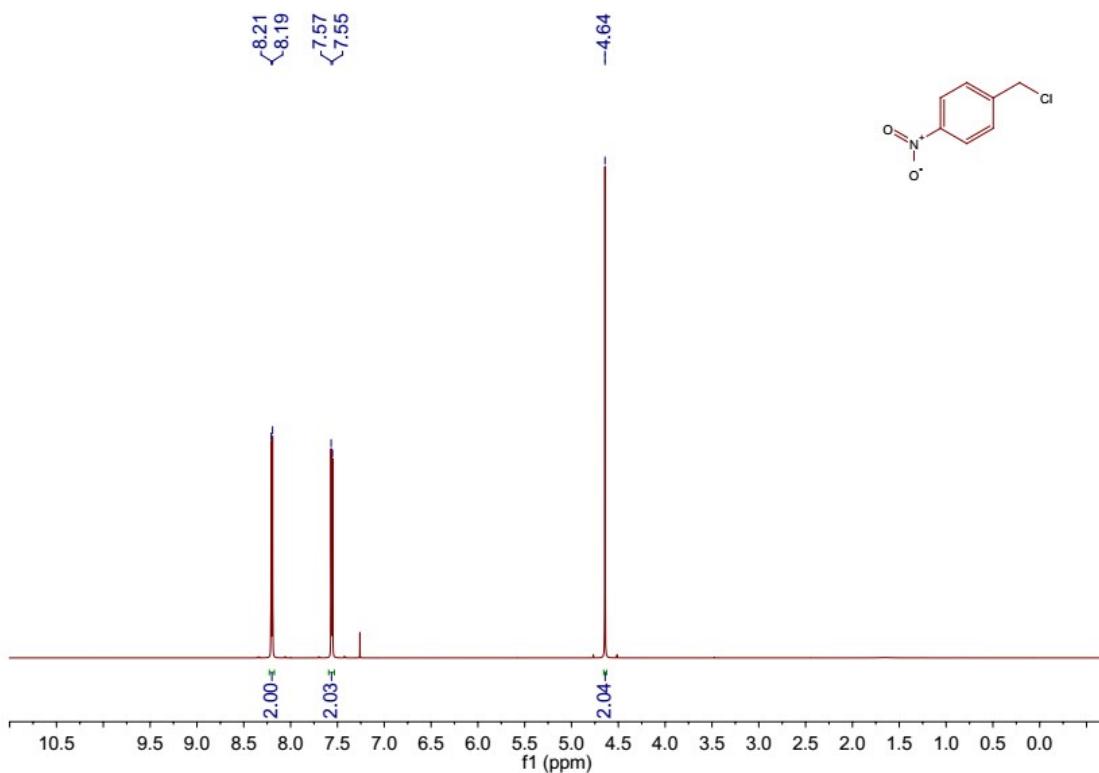


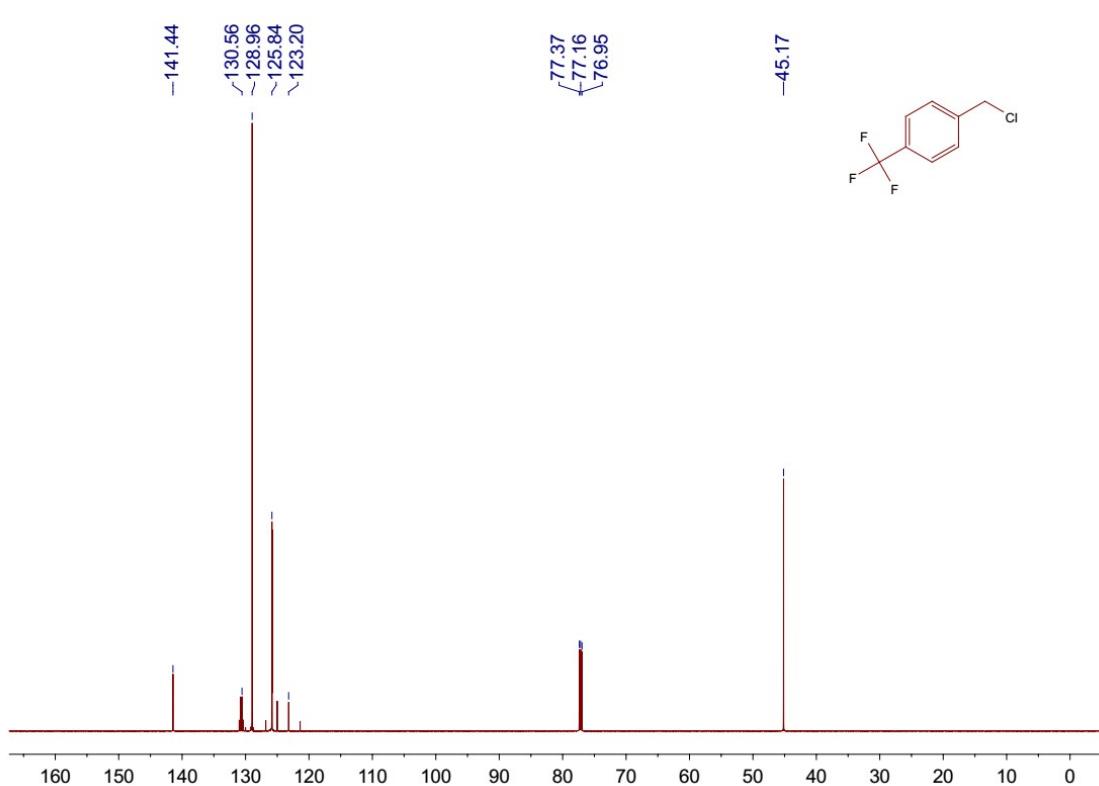
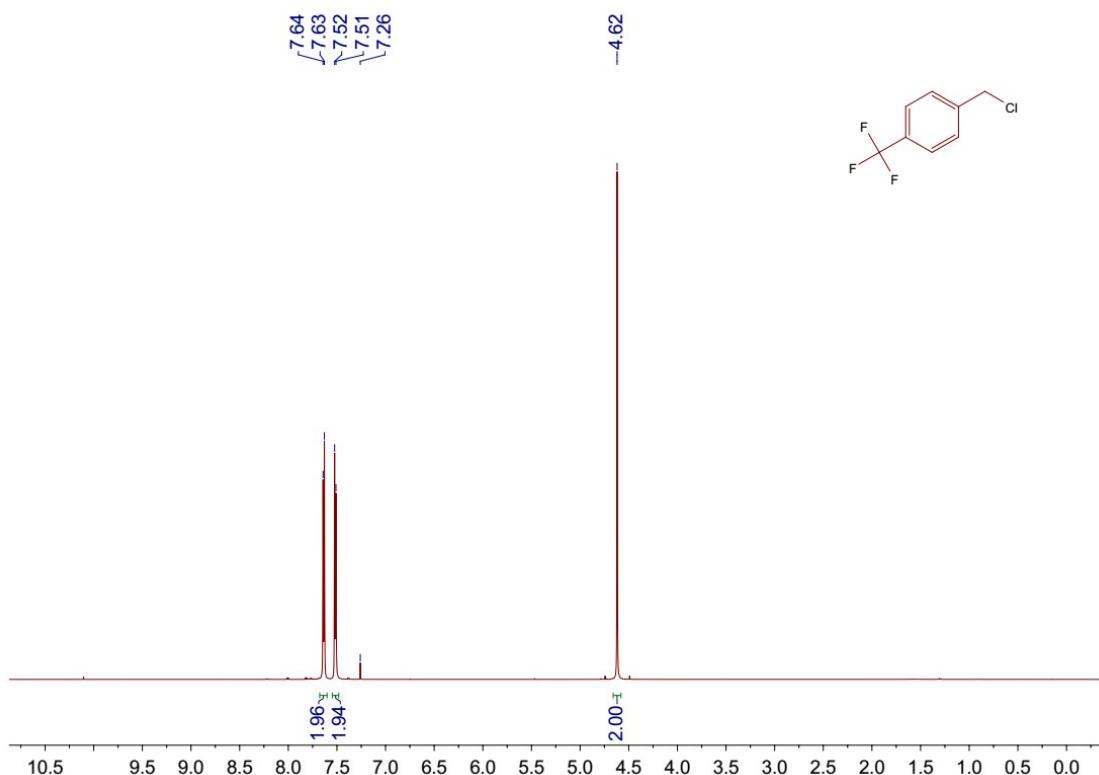


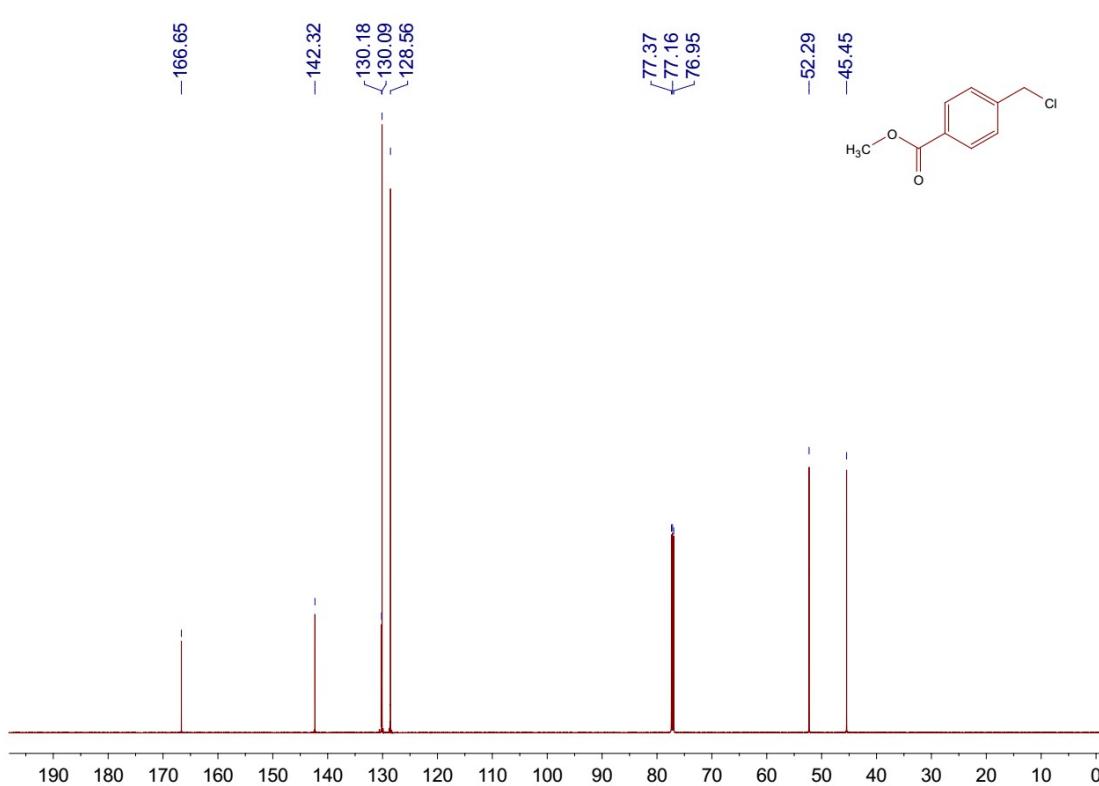
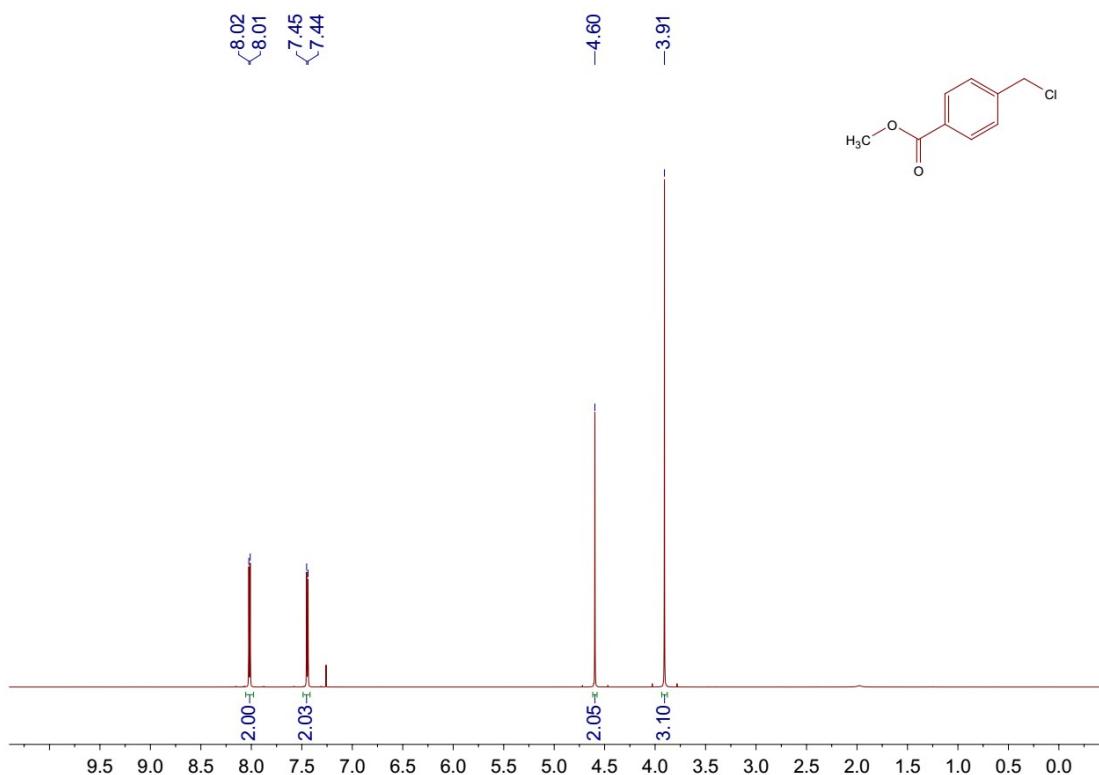
¹³C{¹H} NMR (151 MHz, CDCl₃, 298 K) of **4c**

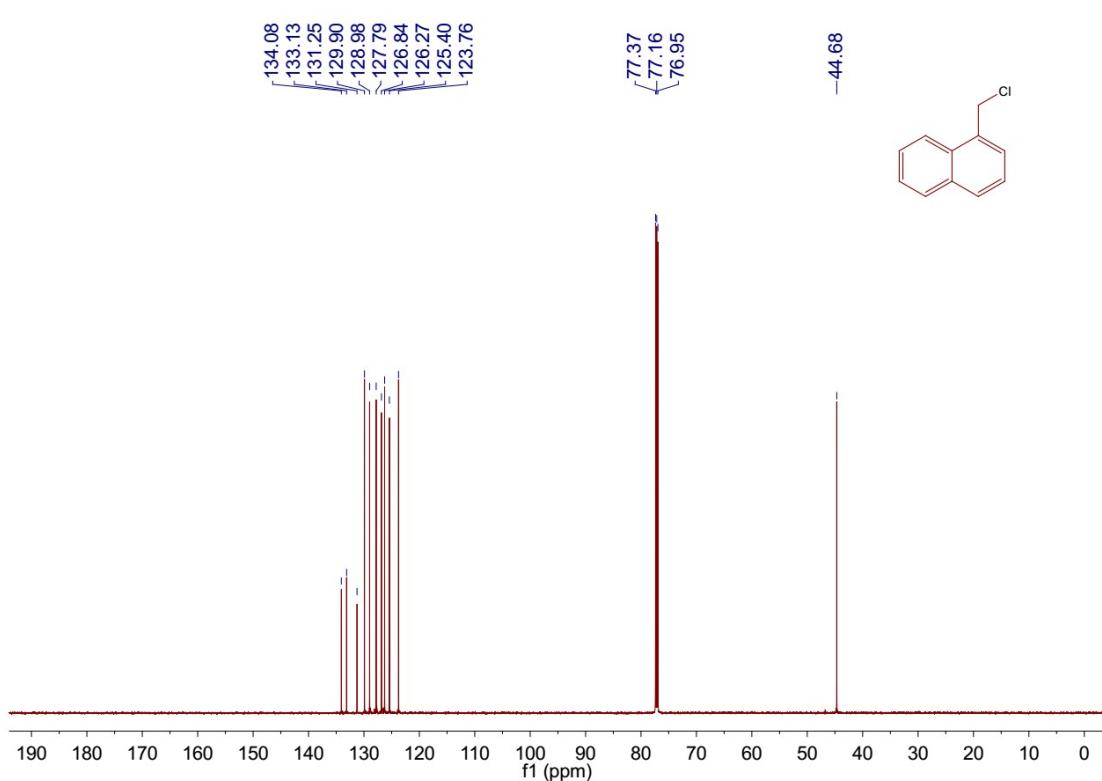
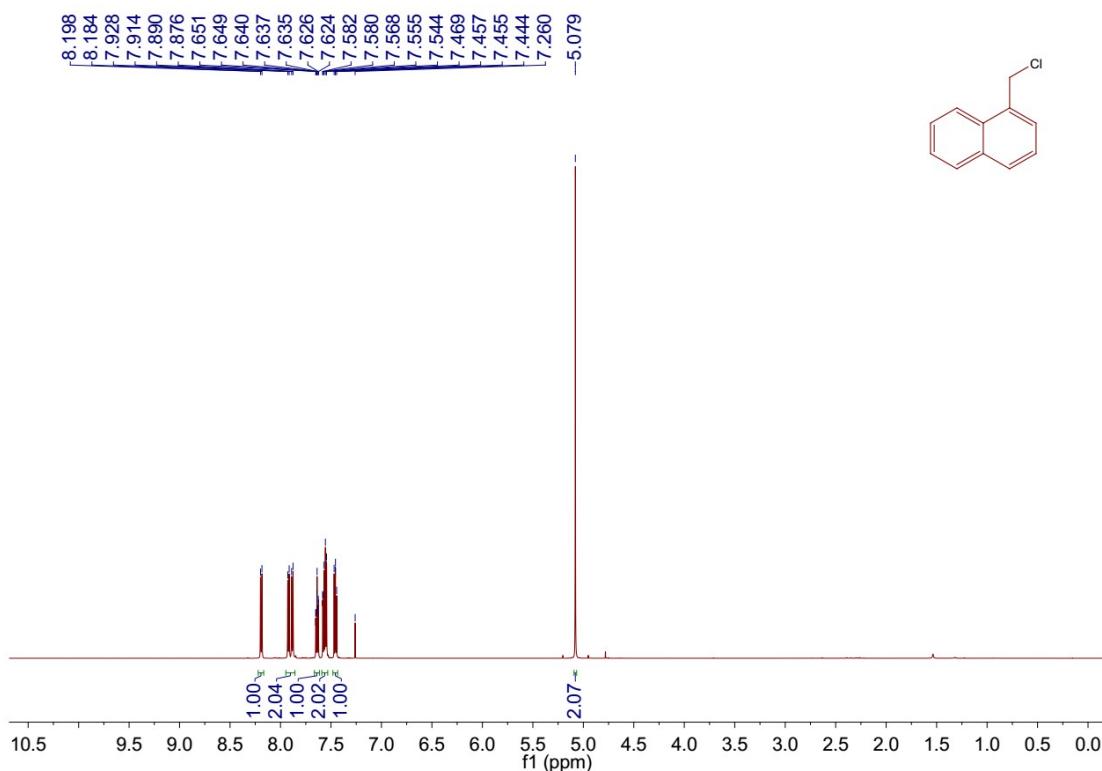


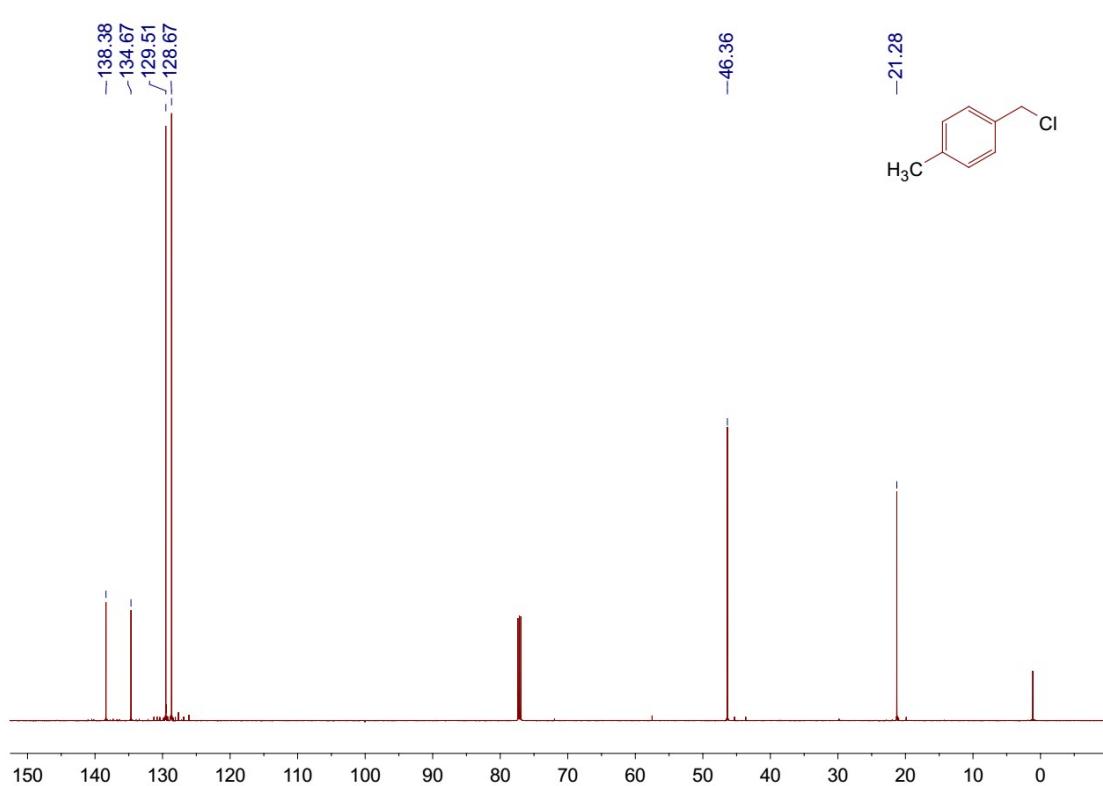
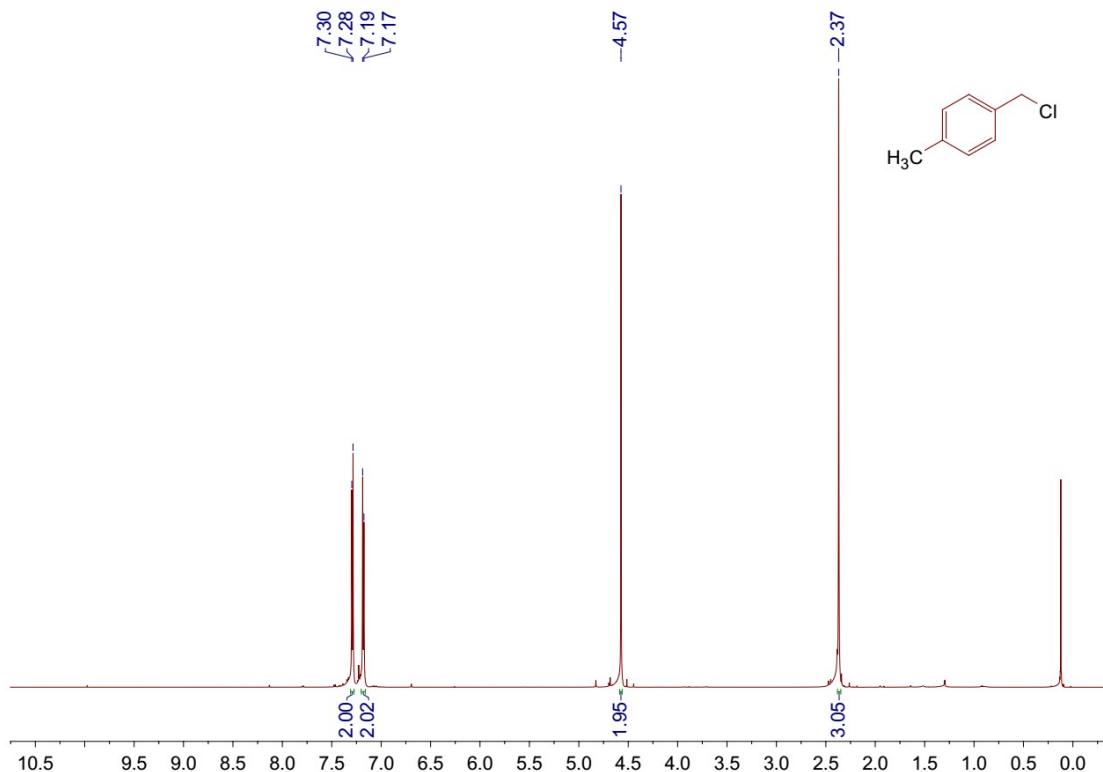


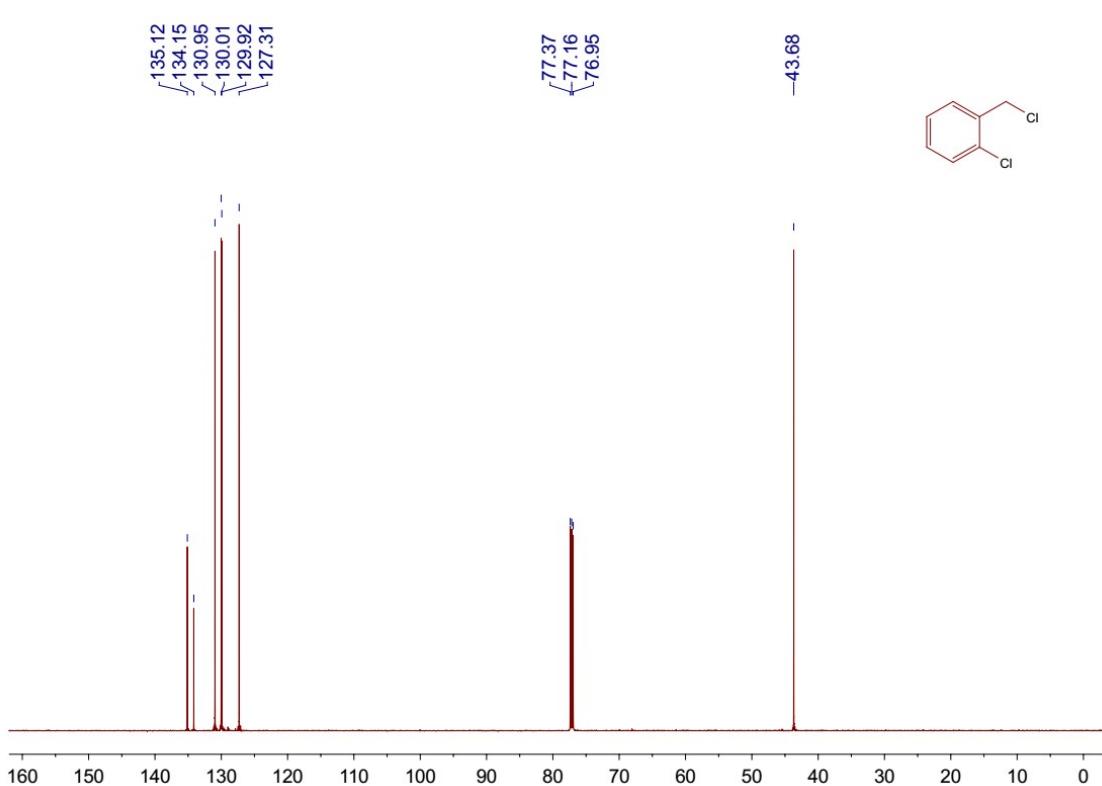
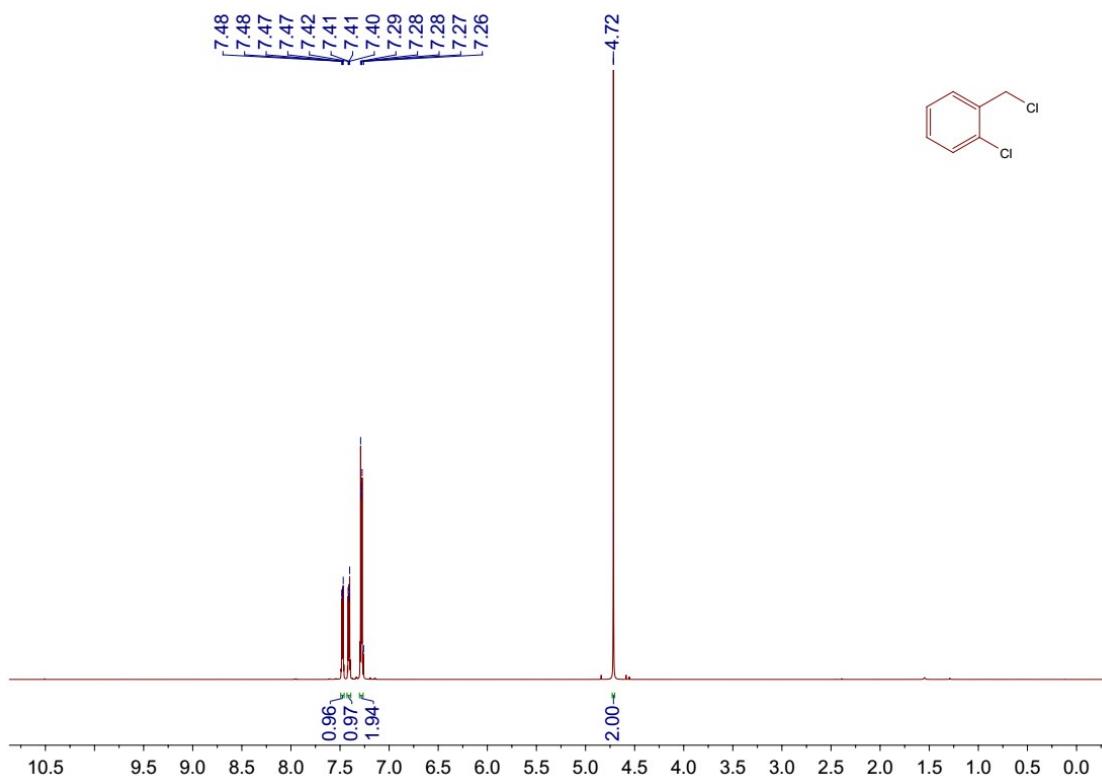




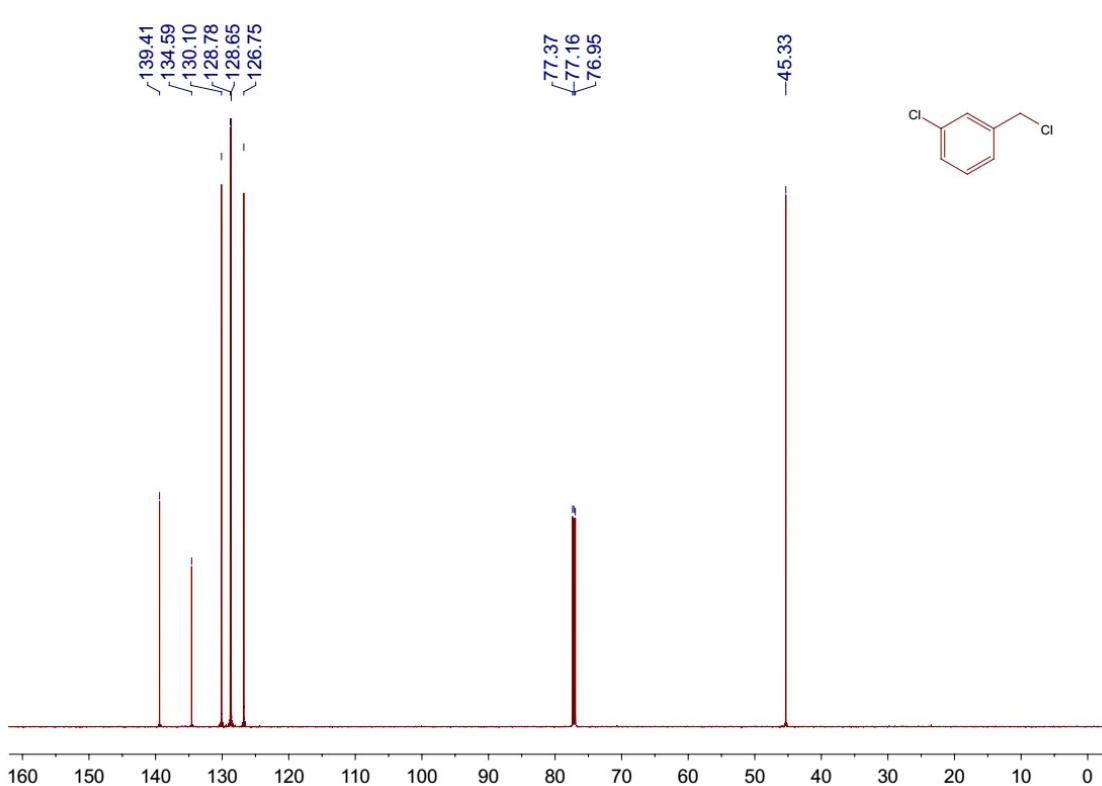
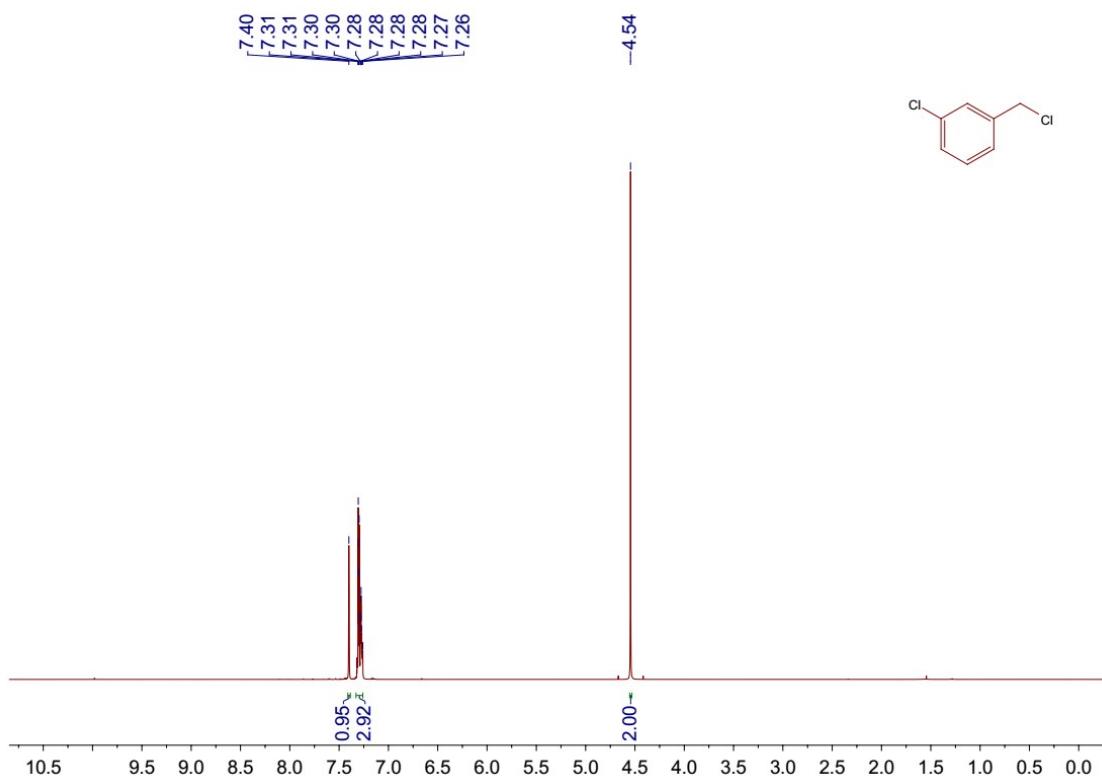


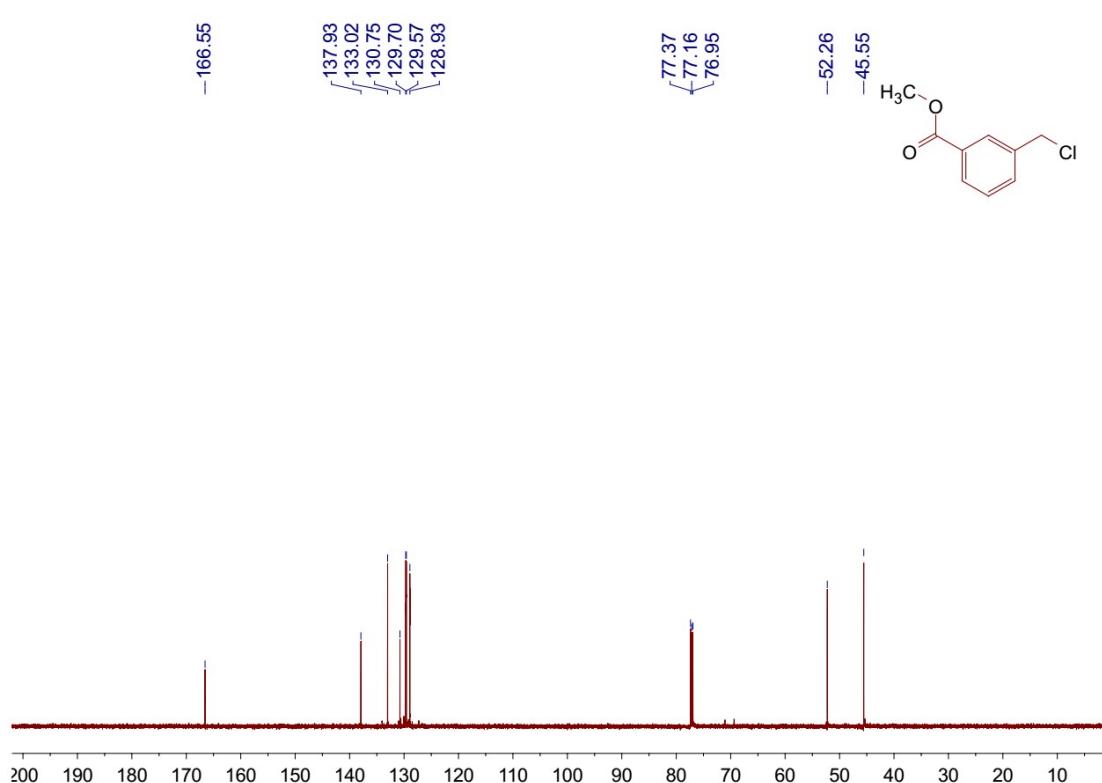
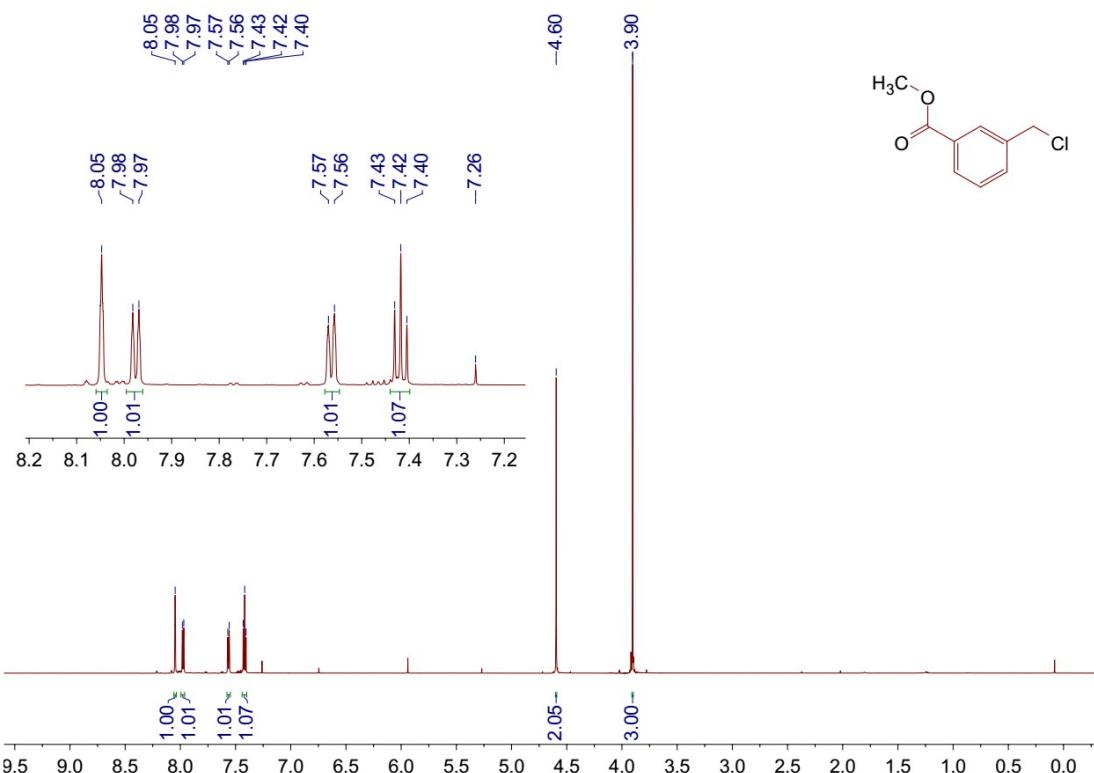


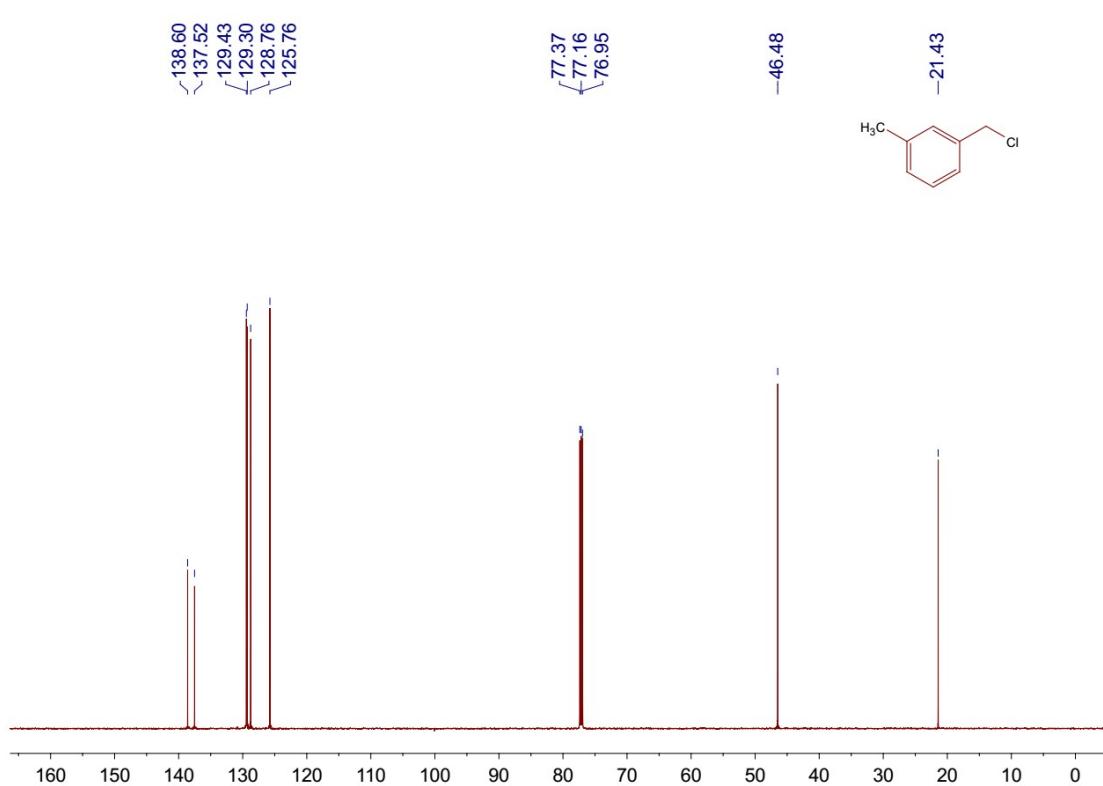
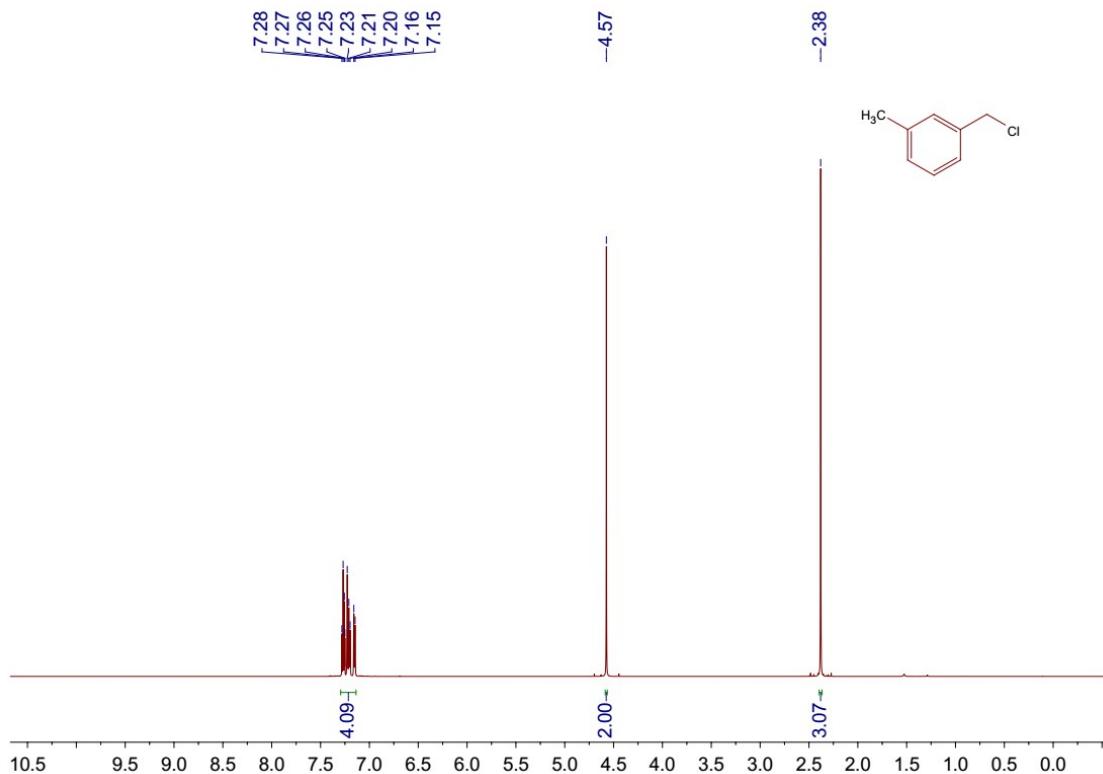




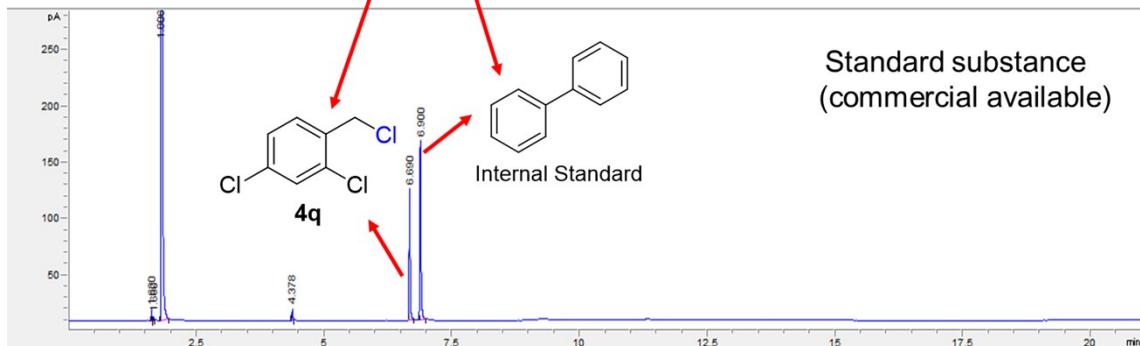
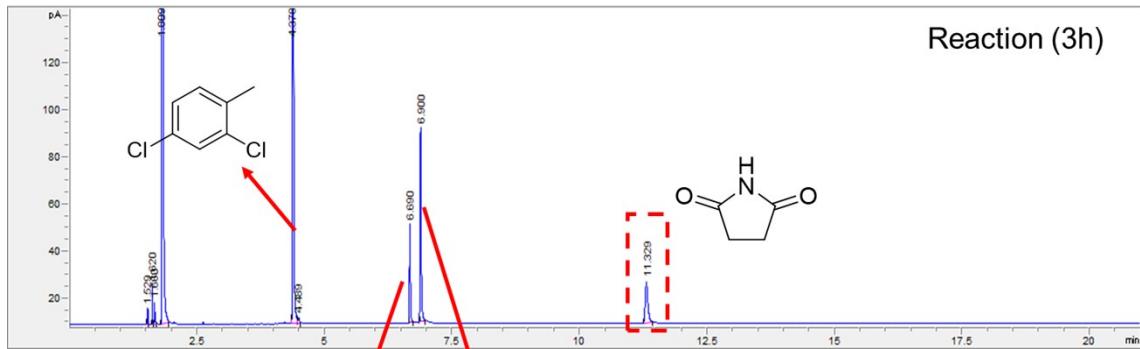
¹³C{¹H} NMR (151 MHz, CDCl₃, 298 K) of **4m**



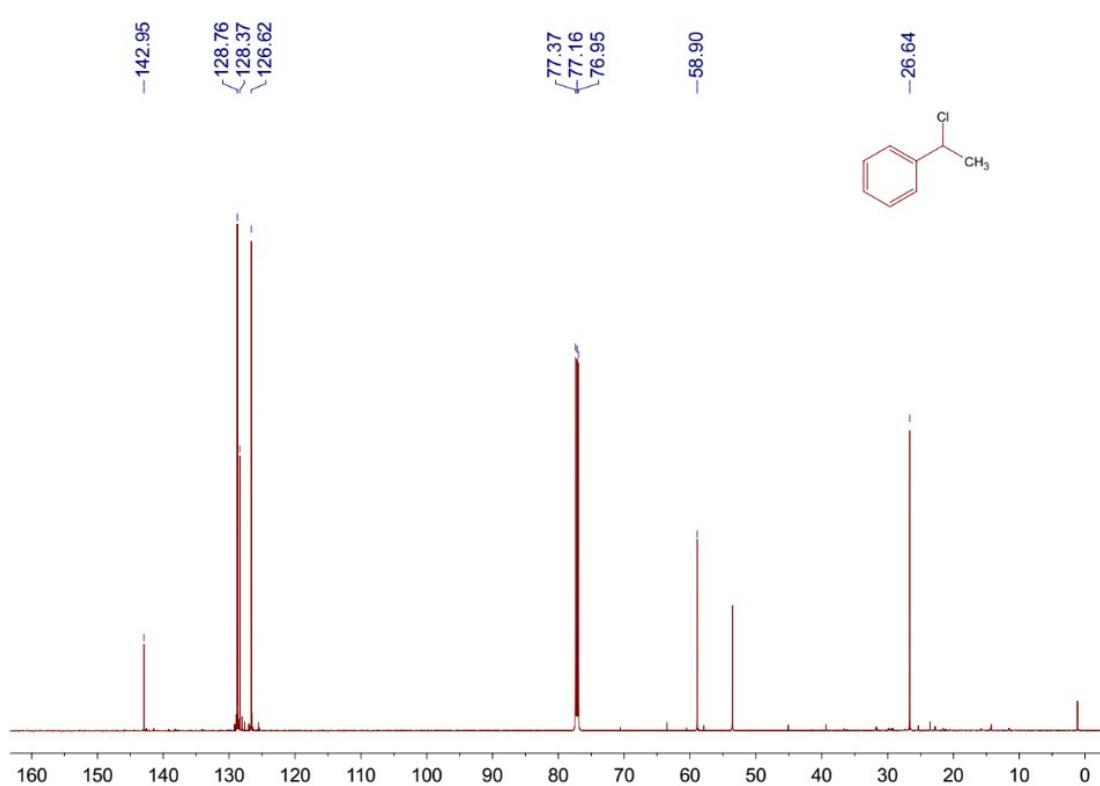
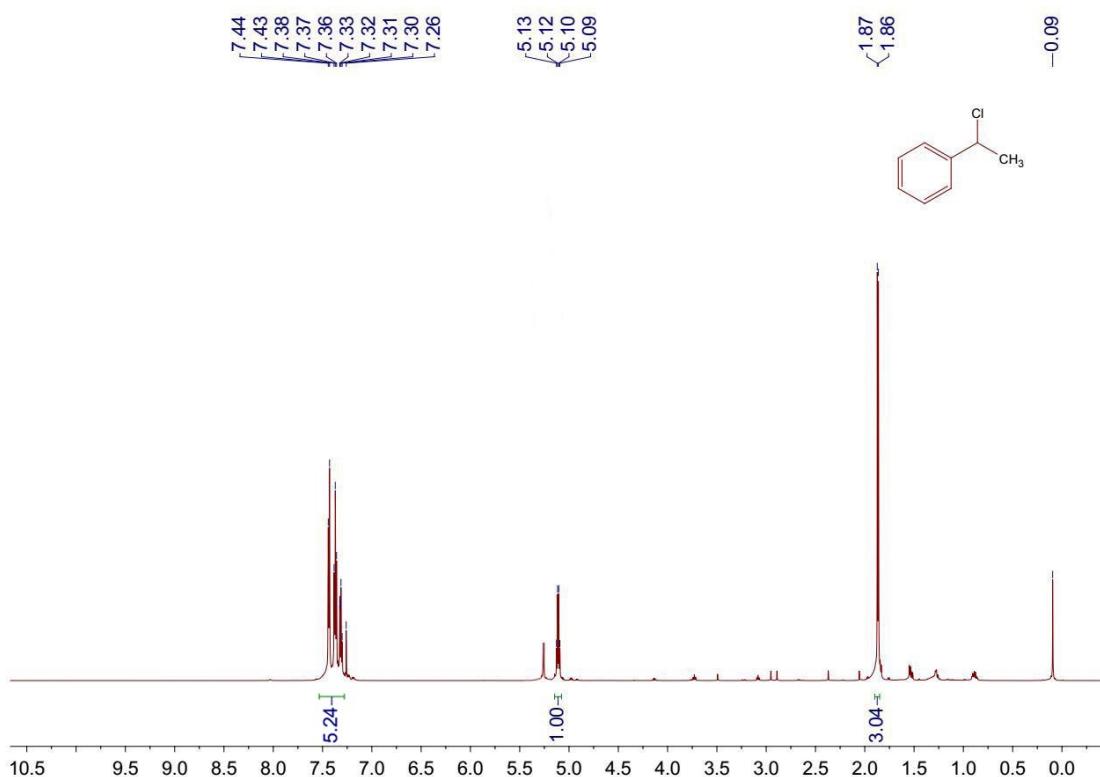


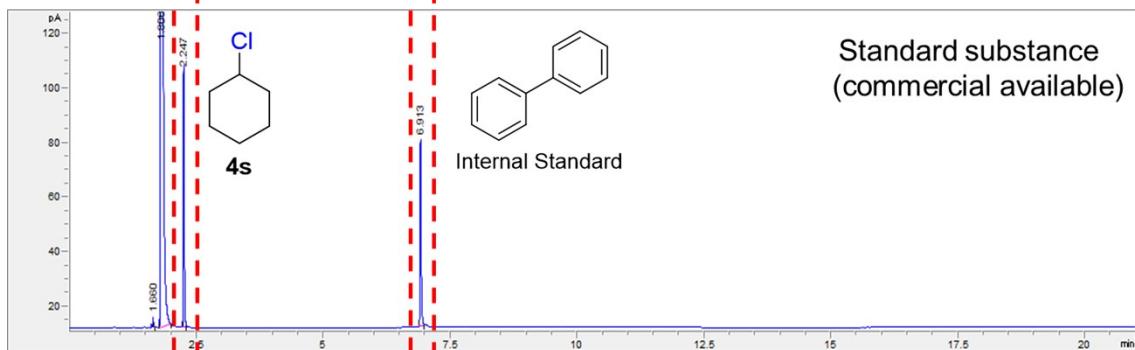
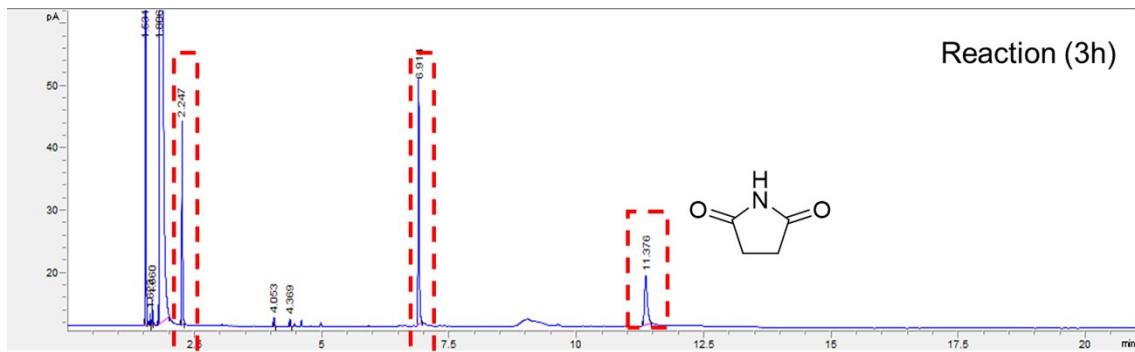


¹³C{¹H} NMR (151 MHz, CDCl₃, 298 K) of **4p**

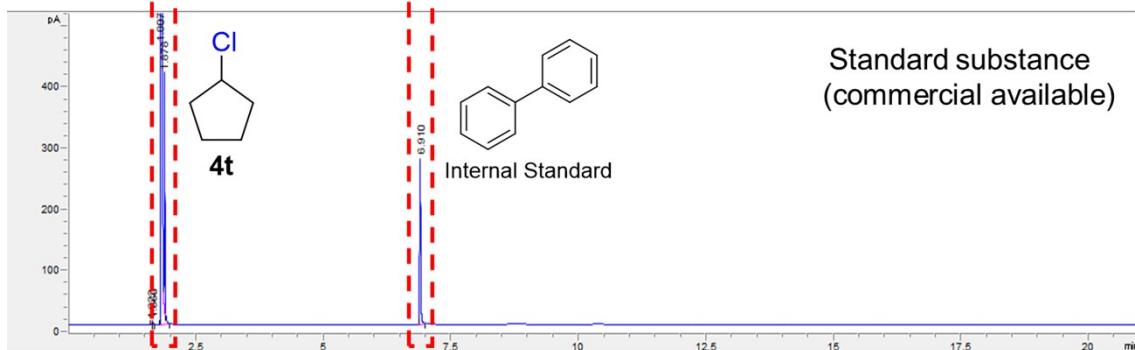
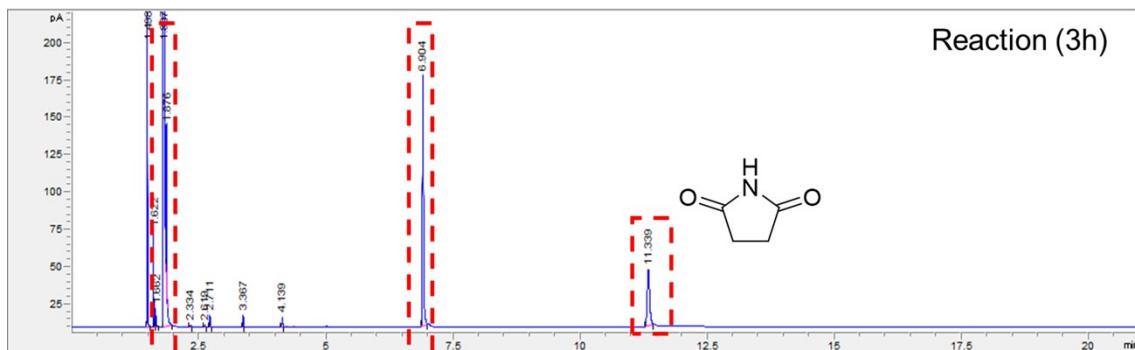


GC spectral of **4q** crude reaction mixture and standard substance

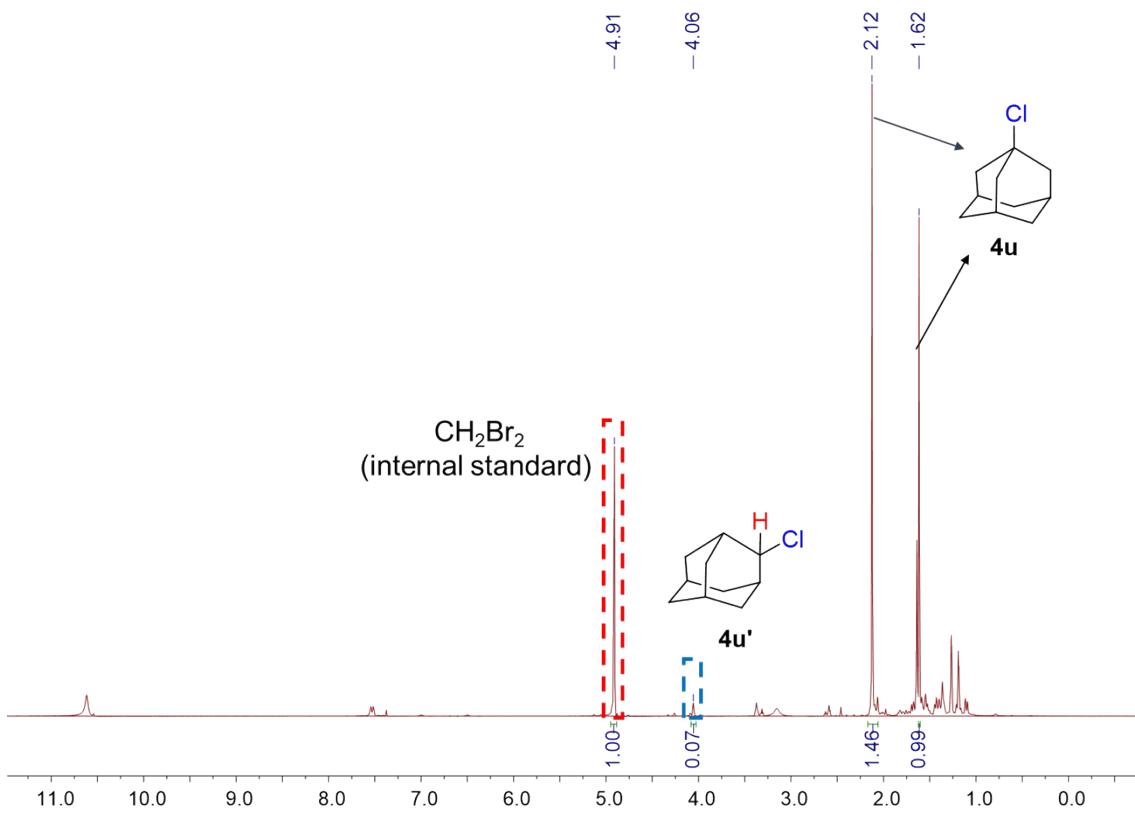




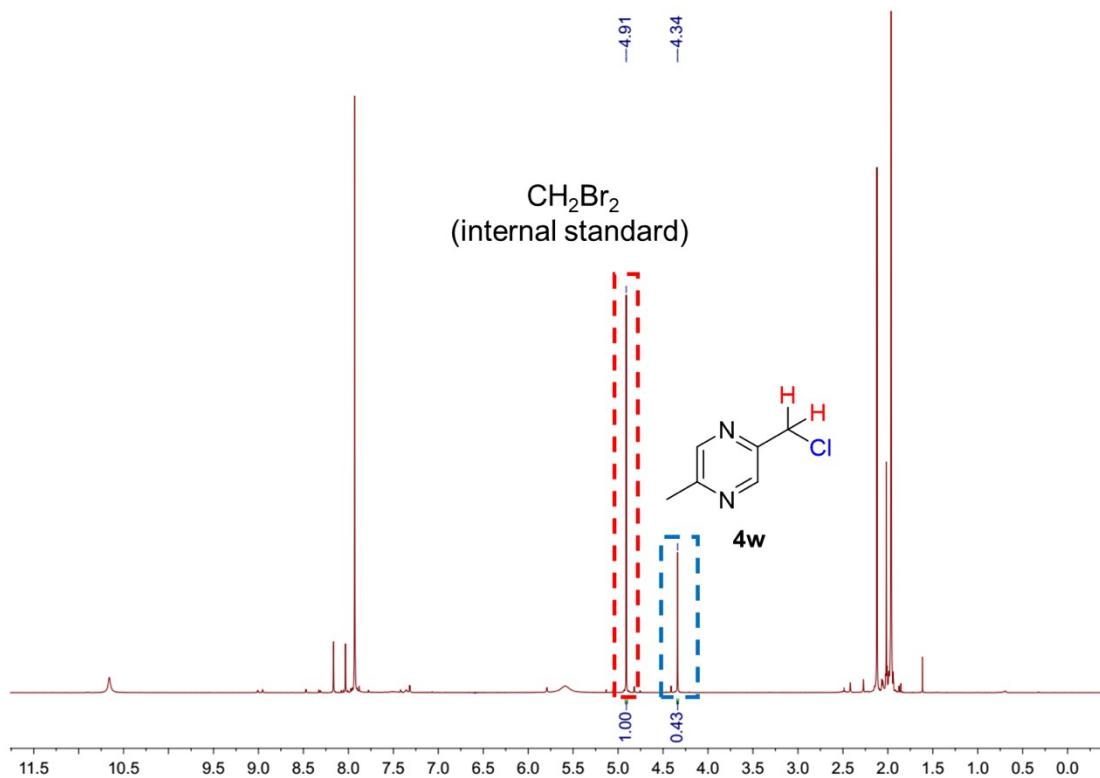
GC spectral of 4s crude reaction mixture and standard substance



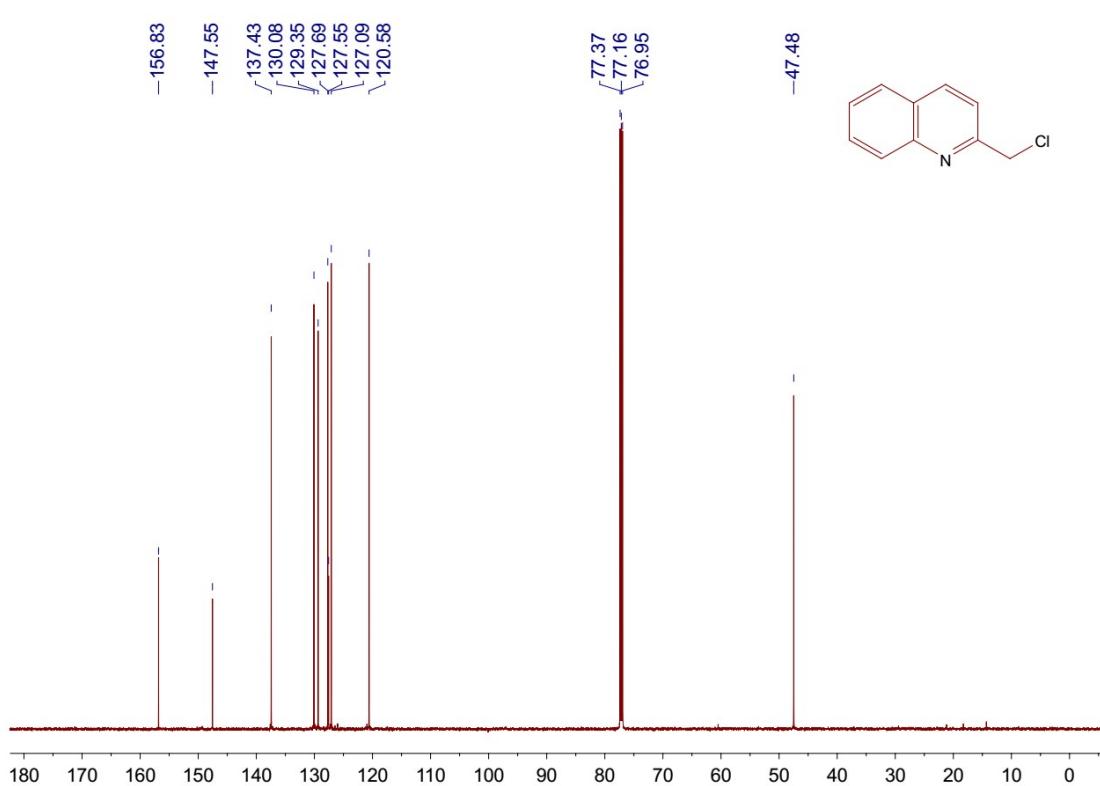
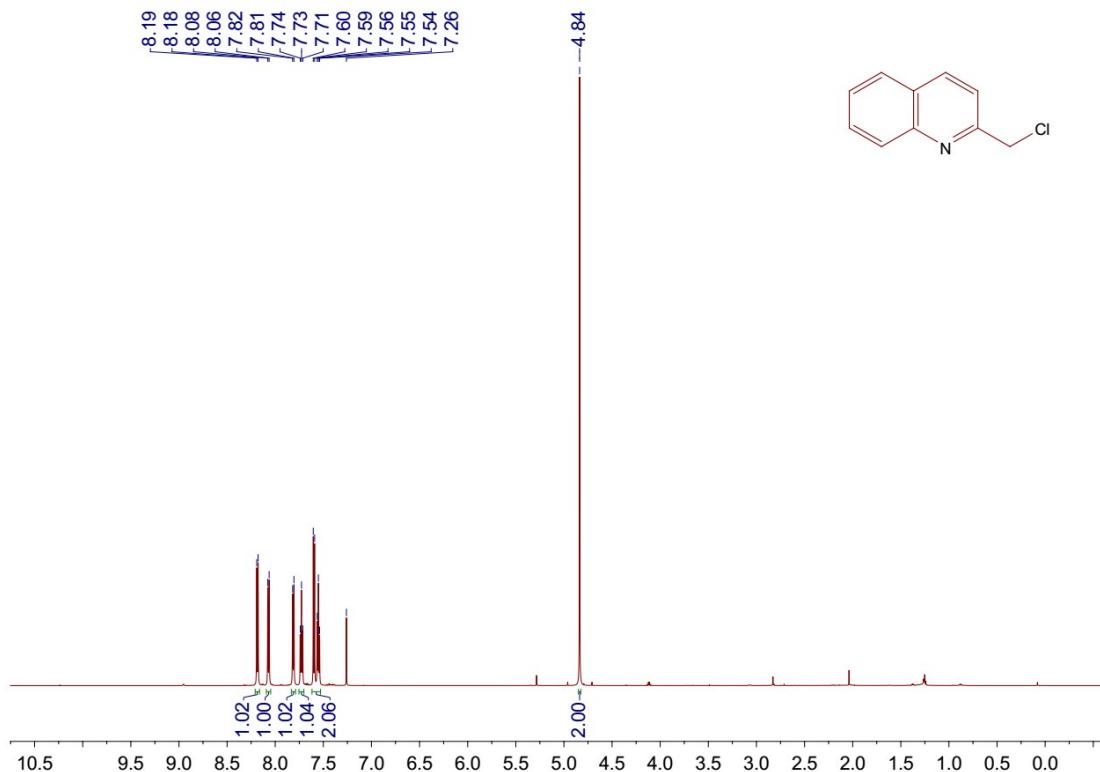
GC spectral of 4t crude reaction mixture and standard substance

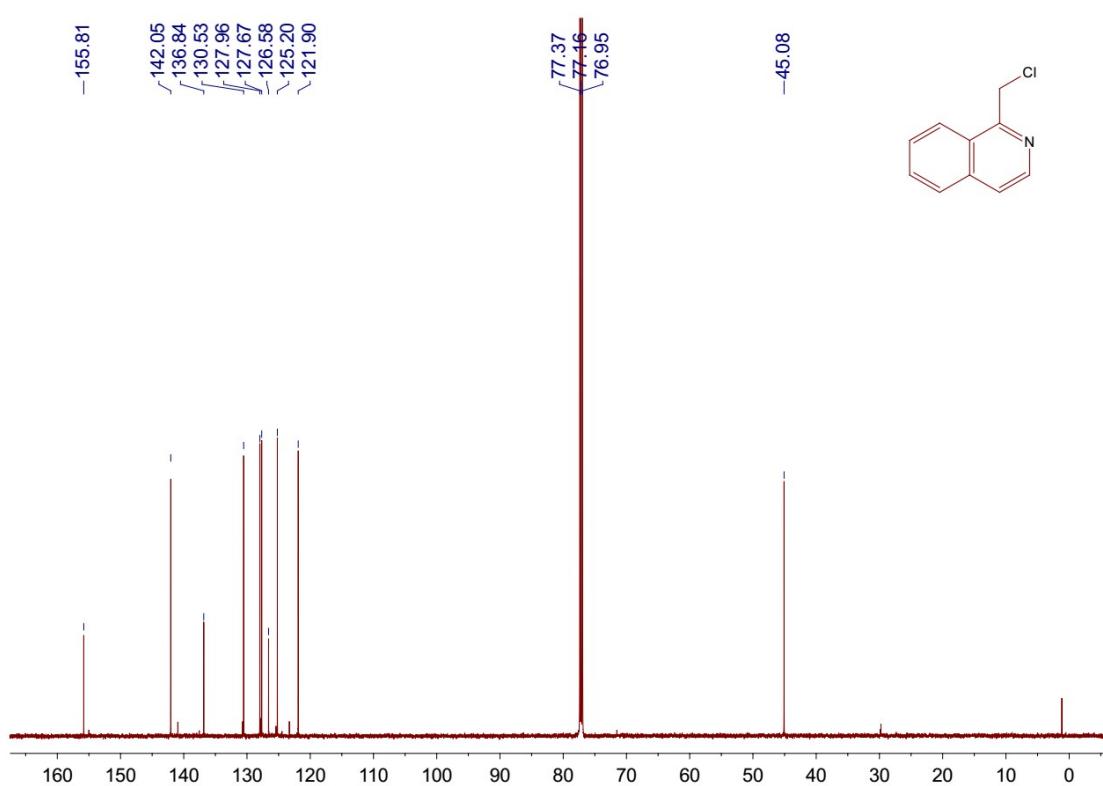
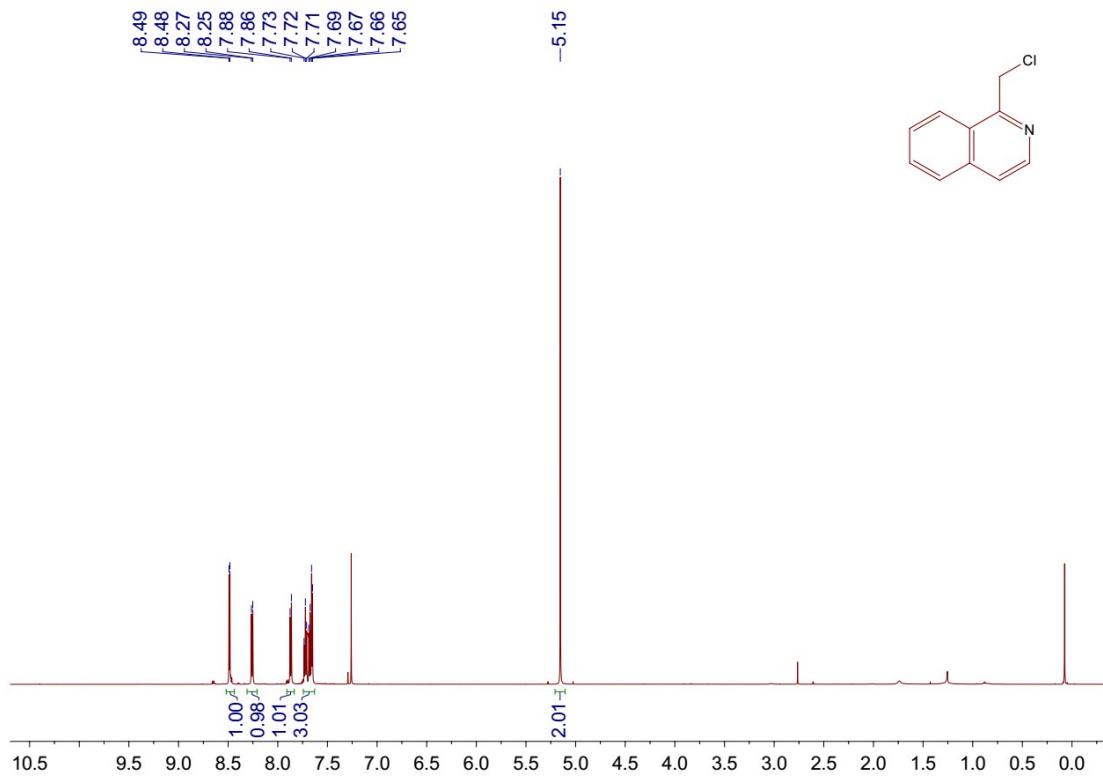


^1H NMR (600 MHz, CDCl_3 , 298 K) of **4u** and **4u'** crude reaction mixture

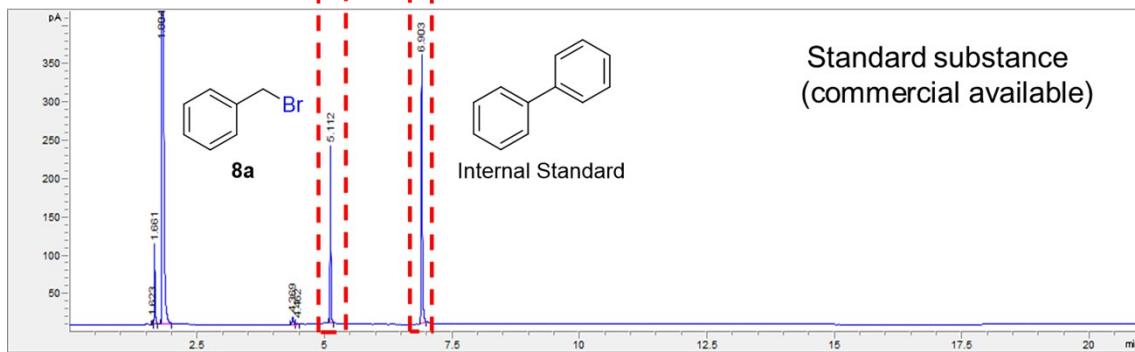
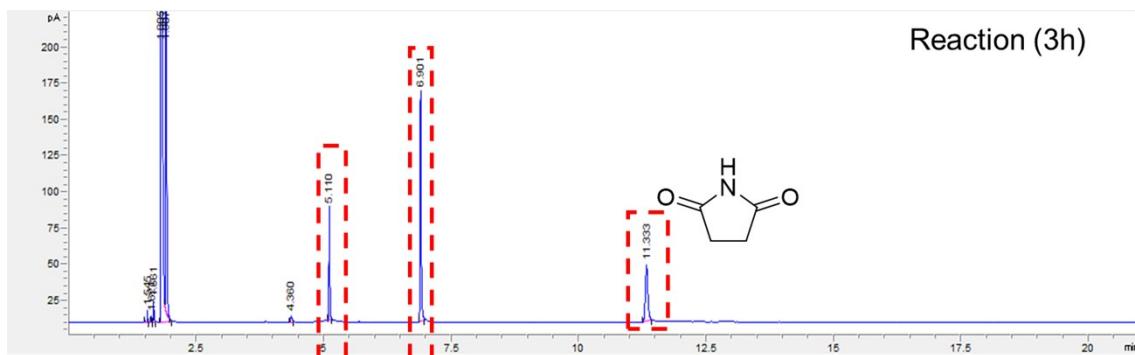


^1H NMR (600 MHz, CDCl_3 , 298 K) of **4w** crude reaction mixture

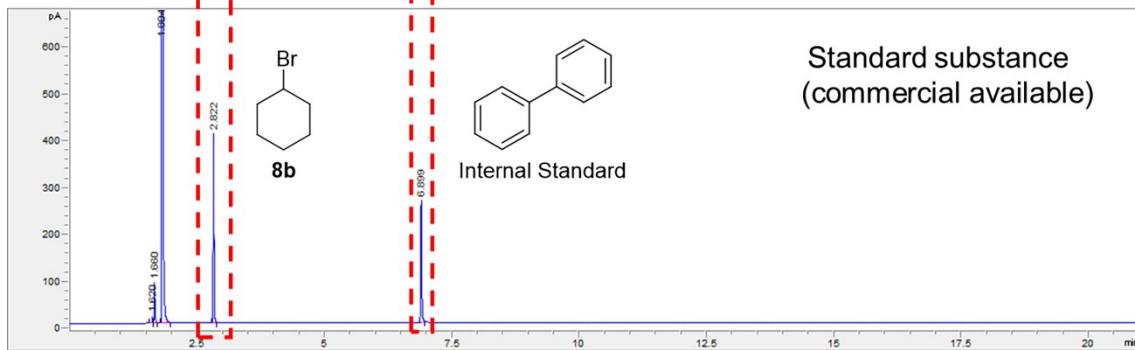
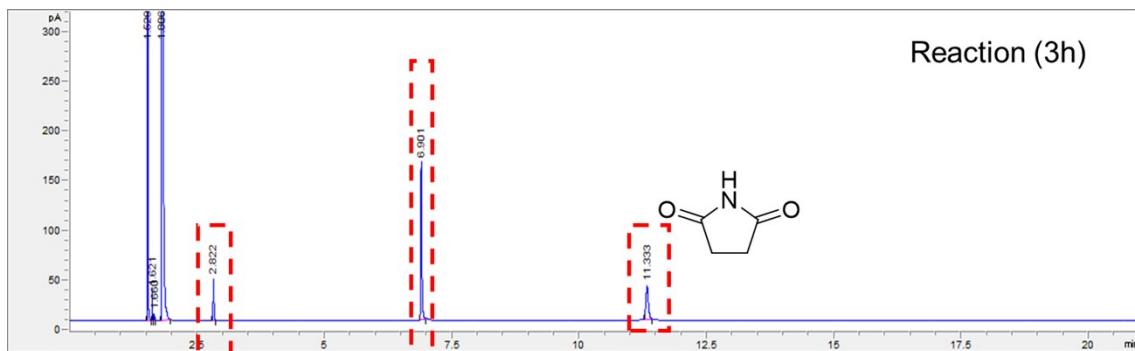




¹³C{¹H} NMR (151 MHz, CDCl₃, 298 K) of **4y**



GC spectral of **8a** crude reaction mixture and standard substance



GC spectral of **8b** crude reaction mixture and standard substance