Electronic Supplementary Information (ESI):

Typical procedure for asymmetric autocatalysis initiated by γ -glycine (Table 1, entry 1):

An enantiomorphic crystal of γ -glycine (right-handed, space group: P3₁, CD(+)215_{KBr}, 37.5 mg, 0.5 mmol) was ground into a fine powder together with pyrimidine-5carbaldehyde 1 (9.4 mg, 0.05 mmol) using a pestle and mortar. After drying under reduced pressure, 1 M toluene solution of *i*-Pr₂Zn (0.4 mL, 0.4 mmol) was slowly added dropwise over a period of 2 h at 0 °C. After the mixture was stirred for 12 h and following addition of toluene (1 mL), 1M toluene solution of *i*-Pr₂Zn (0.3 mL, 0.3 mmol) and aldehyde 1 (18.8 mg, 0.1 mmol) dissolved in toluene (0.5 mL) were then added dropwise over a period of 2 h at 0 °C. The reaction was guenched with a mixture of 30% agueous ammonia and a saturated aqueous ammonium chloride (2/1, v/v) solution (5 mL). The mixture was extracted three times using ethyl acetate. The combined organic layers were dried over anhydrous sodium sulfate and evaporated in vacuo. Purification of the residue using silica gel column chromatography (hexane/ethyl acetate = 2/1, v/v) gave (R)-5-pyrimidyl alkanol 2 with 68% ee in 73% yield. The ee value was determined by HPLC using a chiral stationary phase (Daicel Chiralpak IB column (250 x 4.6 Φ mm ID), eluent = 5% 2propanol in hexane, flow rate 1.0 mL/min, 254 nm UV detector, retention time 10.2 min for (S)-2, 13.6 min for (R)-2).



Fig. S1. The chemical structures of 20 natural amino acids. Only glycine is achiral and the rest of 19 are L-configuration.

Single-crystallization of γ **-glycine**: Glycine (6.0 g) was dissolved in 28-30% aqueous ammonia (20 mL) by stirring for 5 h at room temperature. After the filtration, the filtrate was concentrated under an ambient atmosphere to form the single-crystal of γ -glycine (Figure S2).



Fig. S2. Photos of single-crystals of γ -glycine.

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Entr	y Flack parameter	Parsons Z	Hooft Y	R1	wR	Space group	CD^{a}
1	-0.2 (2)	-0.24 (14)	-0.29 (13)	3.41	9.27	<i>P</i> 3 ₁	(+)
2	-0.08 (15)	-0.37 (8)	-0.01 (12)	2.81	8.22	<i>P</i> 3 ₁	(+)
3	-0.12 (19)	-0.16 (8)	-0.18 (10)	2.78	7.58	<i>P</i> 3 ₁	(+)
4	0.11 (12)	0.07 (11)	0.07 (10)	3.34	8.05	<i>P</i> 3 ₁	(+)
5	-0.1 (3)	0.03 (11)	-0.02 (16)	4.03	10.37	<i>P</i> 3 ₁	(+)
6	0.27 (16)	0.22 (9)	0.23 (12)	3.64	8.74	<i>P</i> 3 ₁	(+)
7	0.04 (19)	-0.56 (14)	0.10 (18)	4.84	12.08	<i>P</i> 3 ₁	(+)
8	-0.5 (3)	0.35 (7)	-0.3 (2)	3.17	8.11	P3 ₂	(-)
9	0.15 (19)	0.02 (10)	0.09 (11)	2.88	7.52	P3 ₂	(-)
10	-0.08 (15)	-0.02 (10)	0.00 (10)	3.02	7.78	P32	(-)

Table S1. Single-crystal X-ray structure analysis of γ -glycine.

^aObserved sigh of Cotton effect in solid-state CD with KBr disk at 215 nm.

Preparation of KBr pellet: After determination of absolute structure of γ -glycine by X-ray single-crystal analysis, the single crystal itself (0.1 mg) was ground in fine powder together with KBr (27 mg). The pellet for solid-state CD was prepared from this powderd mixture.

Formation of enantioenriched powder-like crystals of γ -glycine: Glycine (6.0 g) was dissolved in 28-30% aqueous ammonia (20 mL) by stirring for 5 h at room temperature. After the filtration and addition of seed crystal whose absolute structure was determined by X-ray analysis, the mixture was vigorously stirred with glass beads under ambient atmosphere for 6 d to form the powder-like crystal of γ -glycine.



Fig. S3. ¹H NMR spectrum (600 MHz, CDCl₃, 7.26 ppm for residual CHCl₃) of 5pyrimidyl alkanol **2**: δ 0.883 ppm (d, J = 6.6 Hz, 3H), 0.961 (d, J = 6.6, 3H), 1.375 (s, 9H), 1.36 (d, J = 3.0, 1H), 1.99 (m, 1H), 4.51 (dd, J = 3.0, 5.0, 1H), 8.63 (s, 2H). See also the following reference: T. Shibata, S. Yonekubo, K. Soai *Angew. Chem. Int. Ed.* **1999**, *38*, 659-661.