Supporting Information

Covalently Linked meso-Tetraaryl Triphyrin(2.1.1)-Ferrocene(s) Conjugates: Synthesis and Properties

Kamakshya Nath Panda,[†] Kishor G. Thorat,[†] and Mangalampalli Ravikanth*

Indian Institute of Technology, Powai, Mumbai, 400076, India, Fax: 91-22-5723480;

Tel: 91-22-5767176; E-mail: ravikanth@chem.iitb.ac.in

*Corresponding author, †Authors contributed equally

Sr. no	Contents	Page no.
1	Figures S1 to S26: Characterization data (1D, 2D NMR and HR mass spectra) of all new compounds	S2-S27
2	Figures S27 and S28: Absorption, emission and electrochemical data of the compounds 2, 6, 9 and 10	S28-29
3	Figure S29, Tables S1 and S4. X-ray and DFT data	S30-S35



Figure S1. ¹H NMR spectrum of the compound 5 (*E*/*Z* mixture) recorded in CDCl₃.



Figure S2. ¹³C NMR spectrum of the compound 5 (*E*/*Z* mixture) recorded in CDCl₃.



Figure S3. HR mass spectrum of the compound 5 (*E*/*Z* mixture).



Figure S4. ¹H NMR spectrum of the compound 6 recorded in CDCl₃.



Figure S5. ¹³C NMR spectrum of the compound 6 recorded in CDCl₃.



Figure S6. HR mass spectrum of the compound 6.



Figure S7. ¹H NMR spectrum of the triad **2** recorded in CDCl₃; Peaks marked with asterisk (*) were due to triethyl amine added during experiment to avoid protonation of the macrocycle.



Figure S8. ¹³C NMR spectrum of the triad **2** recorded in CDCl₃; Peaks marked with asterisk (*) were due to triethyl amine added during experiment to avoid protonation of the macrocycle.



Figure S9. HR mass spectrum of the triad 2.



Figure S10. Comparison of partial ¹H NMR spectra of the compound 6 and triad 2 recorded in CDCl₃.



Figure S11. ¹H NMR spectrum of the compound 8 recorded in CDCl₃.



Figure S12. ¹³C NMR spectrum of the compound 8 recorded in CDCl₃.



Figure S13. HR mass spectrum of the compound 8



Figure S14. ¹H NMR spectrum of the compound 9 recorded in CDCl₃.



Figure S15. Partial ¹H (a) and ¹H-¹H (b) NMR spectra of the compound **9** recorded in CDCl₃.



Figure S16. ¹³C NMR spectrum of the compound 9 recorded in CDCl₃.



Figure S17. HR mass spectrum of the compound 9.



Figure S18. ¹H NMR spectrum of the compound 10 recorded in CDCl₃.



Figure S19. Partial ¹H (a) and ¹H-¹H (b) NMR spectra of the compound 10 recorded in CDCl₃.



Figure S20. ¹³C NMR spectrum of the compound 10 recorded in CDCl₃.



Figure S21. HR mass spectrum of the compound 10.



Figure S22. ¹H NMR spectrum of the dyad 3 recorded in CDCl₃.



Figure S23. Partial ¹H NMR (a) and ¹H-¹H COSY (b) spectra of the dyad 3 recorded in CDCl₃.



Figure S24. ¹³C NMR spectrum of the dyad 3 recorded in CDCl₃.



Figure S25. HR mass spectrum of the dyad 3.



Figure S26. Comparision of ¹H NMR spectra of the triphyrin 9, monobromo triphyrin 10 and dyad 3 recorded in CDCl₃.



Figure S27. (a) comparison of absorption spectra of the triphyrin(2.1.1) **6** (red solid line) and triphyrin(2.1.1)-ferrocene triad **2** (black solid line) recorded in CHCl₃ (10 μ M), and (b) comparison of cyclic voltammograms (black solid lines) and differential pulse voltammograms (red solid lines) of the triphyrin(2.1.1) **6** and triphyrin(2.1.1)-ferrocene triad **2** recorded in CH₂Cl₂ with TBAP as a supporting electrolyte (0.1 M) at scan rate of 50 mVs⁻¹.



Figure S28. Comparison of absorption (a) and emission (b) spectra of the triphyrin(2.1.1) **9** and monobromo triphyrin(2.1.1) **10** recorded in CHCl₃ (1 μ M), and (c) comparison of cyclic voltammograms (black solid lines) and differential pulse voltammograms (red solid lines) of the triphyrin(2.1.1) **9** and monobromo triphyrin(2.1.1) **10** recorded in CH₂Cl₂ with TBAP as a supporting electrolyte (0.1 M) at scan rate of 50 mVs⁻¹.



Figure S29. Crystal structure of the dyad 3; Hydrogens in the side view were omitted for clarity.

MR-KGT-617_Mo
$C_{56}H_{43}FeN_3$
813.826
293(2)
monoclinic
P2 ₁ /c
10.3859(10)
17.6141(10)
22.8451(19)
90
94.387(8)
90
4167.0(6)
4
1.292
0.405
1692.0
$0.116 \times 0.106 \times 0.048$
MoK α ($\lambda = 0.71073$)
3.576 to 62.386
$-14 \le h \le 15, -24 \le k \le 24, -32 \le l \le 32$
56659
12392 [$R_{int} = 0.2662, R_{sigma} = 0.3106$]
12392/0/545
1.106
$R_1 = 0.1807, wR_2 = 0.3922$
$R_1 = 0.4362, wR_2 = 0.5288$
1.26/-0.87

Table S1. Crystal data and structure refinement for the dyad **3.**

Atom	Atom	Length/Å	Atom	Atom	Length/Å	Atom	Atom	Length/Å
Fe1	C19	1.968(18)	C20	C19	1.48(2)	С9	C10	1.437(12)
Fe1	C20	2.015(15)	C20	C21	1.37(2)	C10	C11	1.395(11)
Fe1	C21	1.990(15)	C21	C22	1.31(2)	C10	C36	1.464(11)
Fe1	C22	2.025(15)	C22	C23	1.458(18)	C11	C12	1.453(12)
Fe1	C23	2.026(13)	C25	C24	1.49(3)	C13	C12	1.345(11)
Fe1	C24	1.950(19)	C25	C26	1.38(2)	C14	C13	1.452(13)
Fe1	C25	2.05(2)	C26	C27	1.347(19)	C15	C14	1.425(11)
Fe1	C26	2.071(17)	C28	C24	1.43(4)	C15	C16	1.369(12)
Fe1	C27	2.030(12)	C28	C27	1.41(2)	C15	C43	1.527(11)
Fe1	C28	2.026(16)	C29	C34	1.411(12)	C17	C18	1.291(18)
N1	C1	1.334(10)	C30	C29	1.387(12)	C19	C18	1.448(18)
N1	C4	1.364(11)	C30	C31	1.381(12)	C19	C23	1.417(18)
N2	C6	1.346(11)	C32	C31	1.386(12)	C46	C47	1.364(14)
N2	C9	1.360(10)	C32	C35	1.518(13)	C46	C49	1.559(12)
N3	C11	1.383(10)	C33	C32	1.395(13)	C48	C47	1.368(13)
N3	C14	1.354(10)	C33	C34	1.352(12)	C50	C16	1.501(11)
C1	C2	1.460(12)	C36	C37	1.411(11)	C50	C51	1.375(12)
C1	C16	1.476(12)	C37	C38	1.336(12)	C50	C55	1.377(11)
C2	C3	1.412(12)	C39	C38	1.401(13)	C51	C52	1.380(13)
C4	C3	1.441(12)	C39	C42	1.543(12)	C53	C52	1.372(13)
C4	C5	1.466(11)	C40	C39	1.377(12)	C53	C54	1.393(13)
C5	C6	1.357(12)	C41	C36	1.393(12)	C53	C56	1.574(13)
C5	C29	1.464(12)	C41	C40	1.385(12)	C55	C54	1.373(12)
C6	C7	1.496(13)	C43	C44	1.379(12)	C8	C17	1.350(17)
C8	C7	1.358(15)	C43	C48	1.394(12)	С9	C8	1.485(13)
C46	C45	1.417(13)	C44	C45	1.403(12)			

Table S2. Bond lengths for the dyad **3.**

Atom	Atom	Atom	Angle/°												
C19	Fe1	C20	43.7(6)	C16	C15	C14	131.8(8)	C25	Fe1	C22	115.2(9)	C28	C27	Fe1	69.5(8)
C19	Fel	C21	68.7(7)	C16	C15	C43	117.2(7)	C25	Fe1	C26	39.0(7)	C24	C28	Fe1	66.1(10)
C19	Fel	C22	67.5(5)	C1	C16	C50	109.6(8)	C25	Fe1	C27	67.3(7)	C27	C28	Fe1	69.8(8)
C19	Fe1	C23	41.5(5)	C15	C16	C1	126.9(8)	C27	Fe1	C26	38.3(6)	C27	C28	C24	97.0(18)
C19	Fel	C25	164.8(8)	C15	C16	C50	123.4(8)	C28	Fe1	C22	159.8(8)	C30	C29	C5	123.6(8)
C19	Fel	C26	156.2(9)	C18	C17	C8	174.5(16)	C28	Fe1	C23	120.9(7)	C30	C29	C34	117.5(9)
C19	Fe1	C27	124.7(6)	C17	C18	C19	179.4(15)	C28	Fe1	C25	74.9(10)	C34	C29	C5	118.8(9)
C19	Fel	C28	107.6(8)	C18	C19	Fe1	121.5(10)	C28	Fe1	C26	69.4(8)	C31	C30	C29	120.4(9)
C20	Fe1	C21	40.1(6)	C18	C19	C20	122.7(13)	C28	Fe1	C27	40.7(6)	C30	C31	C32	121.5(9)
C20	Fel	C22	67.2(8)	C18	C19	C23	125.9(15)	C1	N1	C4	109.2(7)	C31	C32	C35	122.1(10)
C20	Fe1	C23	72.6(8)	C20	C19	Fe1	69.8(10)	C6	N2	C9	110.1(7)	C33	C32	C31	118.0(10)
C20	Fel	C25	122.1(8)	C23	C19	Fe1	71.4(9)	C14	N3	C11	108.7(8)	C33	C32	C35	119.9(9)
C20	Fe1	C26	157.5(8)	C23	C19	C20	111.2(13)	N1	C1	C2	109.8(8)	C34	C33	C32	120.8(9)
C20	Fe1	C27	162.4(6)	C19	C20	Fe1	66.5(9)	N1	C1	C16	124.8(8)	C33	C34	C29	121.7(9)
C20	Fe1	C28	123.9(8)	C21	C20	Fe1	69.0(10)	C2	C1	C16	125.4(8)	C37	C36	C10	121.3(8)
C21	Fel	C22	38.2(6)	C21	C20	C19	102.9(16)	C3	C2	C1	105.5(8)	C41	C36	C10	121.3(8)
C21	Fe1	C23	69.9(6)	C20	C21	Fe1	70.9(9)	C2	C3	C4	106.0(8)	C41	C36	C37	117.3(8)
C21	Fe1	C25	103.7(9)	C22	C21	Fe1	72.3(9)	N1	C4	C3	109.5(8)	C38	C37	C36	122.0(9)
C21	Fe1	C26	121.9(8)	C22	C21	C20	112.5(16)	N1	C4	C5	122.9(8)	C37	C38	C39	120.1(9)
C21	Fe1	C27	157.0(7)	C21	C22	Fe1	69.5(10)	C3	C4	C5	127.5(10)	C38	C39	C42	120.2(9)
C21	Fe1	C28	160.5(6)	C21	C22	C23	112.2(16)	C4	C5	C29	120.9(8)	C40	C39	C38	119.6(9)
C22	Fe1	C26	106.6(7)	C23	C22	Fe1	68.9(7)	C6	C5	C4	117.7(8)	C40	C39	C42	120.1(9)
C22	Fe1	C27	124.6(8)	C19	C23	Fe1	67.0(9)	C6	C5	C29	121.2(8)	C39	C40	C41	119.9(9)
C23	Fe1	C22	42.2(6)	C22	C23	Fe1	68.9(8)	N2	C6	C5	121.4(8)	C40	C41	C36	120.9(9)
C23	Fe1	C25	150.0(8)	C22	C23	C19	101.1(15)	N2	C6	C7	110.9(8)	C44	C43	C15	120.1(8)
C23	Fe1	C26	118.3(9)	C25	C24	Fe1	71.6(10)	C5	C6	C7	126.7(9)	C44	C43	C48	118.8(8)
C23	Fe1	C27	106.8(6)	C28	C24	Fe1	71.8(14)	C8	C7	C6	102.0(10)	C48	C43	C15	121.1(9)
C24	Fe1	C19	128.9(6)	C28	C24	C25	116(2)	C7	C8	C9	111.6(9)	C43	C44	C45	121.7(8)
C24	Fe1	C20	110.7(9)	C24	C25	Fe1	64.6(12)	C17	C8	C7	120.9(11)	C44	C45	C46	119.1(9)
C24	Fe1	C21	124.5(12)	C26	C25	Fe1	71.4(11)	C17	C8	C9	127.3(13)	C45	C46	C49	120.5(10)
C24	Fe1	C22	155.7(12)	C26	C25	C24	99(2)	N2	C9	C8	105.3(9)	C47	C46	C45	116.8(9)
C24	Fe1	C23	162.1(11)	C25	C26	Fe1	69.6(13)	N2	C9	C10	118.9(8)	C47	C46	C49	122.7(9)
C24	Fe1	C25	43.8(8)	C27	C26	Fe1	69.2(8)	C10	C9	C8	134.0(9)	C48	C47	C46	124.9(9)
C24	Fe1	C26	65.6(7)	C27	C26	C25	112.2(19)	C9	C10	C36	119.6(8)	C47	C48	C43	118.6(10)
C24	Fe1	C27	64.5(10)	C26	C27	Fe1	72.5(10)	C11	C10	C9	118.5(8)	C51	C50	C16	120.5(8)
C24	Fe1	C28	42.1(11)	C26	C27	C28	115.7(17)	C11	C10	C36	121.9(9)	C51	C50	C55	117.6(8)
C12	C13	C14	106.6(9)	C52	C53	C56	122.8(10)	N3	C11	C10	123.6(9)	C55	C50	C16	121.8(7)
N3	C14	C13	109.1(8)	C54	C53	C56	117.7(10)	N3	C11	C12	107.0(7)	C50	C51	C52	121.5(10)
N3	C14	C15	123.9(9)	C53	C54	C55	118.9(10)	C10	C11	C12	129.4(8)	C53	C52	C51	119.9(9)
C13	C12	C11	108.6(8)	C52	C53	C54	119.5(9)	C15	C14	C13	127.0(8)	C50	C55	C54	122.3(9)

Table S3. Bond angles for the dyad **3**.

Atom	X	Y	Z	Atom	X	Y	Z	Atom	X	Y	Z
N	-1.24982	1.26905	-0.25923	Н	2.71011	-1.18774	1.76049	Н	-10.36752	11.58168	-0.44286
Ν	-3.47464	0.05739	-0.83023	С	4.12121	-2.85293	-1.43911	Н	-9.45478	9.17119	-1.25169
Ν	-1.34337	-1.29987	-0.38959	Η	4.51087	-3.31844	-2.34020	Fe	-7.57855	11.25904	-0.84984
Н	-1.61198	-0.27199	-0.33254	С	-4.88848	-5.62698	0.83757	С	-6.81803	13.08349	-1.49685
С	-1.09114	3.52132	-0.68110	Η	-4.60736	-6.39307	1.55300	С	-5.75273	12.16199	-1.25935
Н	-1.38631	4.54055	-0.88772	С	4.97149	2.60900	-0.35689	С	-7.66918	12.52492	-2.49759
С	0.04594	1.57777	-0.41779	С	4.53657	1.94633	-1.50641	Н	-6.97188	14.02312	-0.98415
С	-5.45303	0.81787	0.05415	Н	5.15541	1.94145	-2.39935	С	-5.94547	11.03504	-2.11403
Н	-6.19511	1.50489	0.43830	С	-3.74588	4.70713	0.57286	Н	-4.95784	12.28168	-0.53603
С	-1.99394	2.40341	-0.42864	Н	-2.86008	4.56114	1.18199	С	-7.13017	11.25819	-2.87869
С	-4.12516	1.16892	-0.46187	С	2.92748	1.96865	0.76676	Н	-8.58054	12.96771	-2.87547
С	2.48701	1.28690	-0.37751	Н	2.29493	1.98605	1.64979	Н	-5.33494	10.14279	-2.13632
С	1.08942	-0.78202	-0.38937	С	2.92272	-2.13937	-1.49176	Н	-7.56077	10.57404	-3.59685
С	-0.04246	-1.67126	-0.44157	Н	2.38258	-2.05505	-2.43068	С	-8.55442	-8.93691	0.70989
С	-4.22603	-1.00450	-0.48351	С	-6.09317	-5.76838	0.11754	С	-9.88582	-9.07500	0.17501
С	0.17394	3.01184	-0.68392	С	-4.46745	5.88823	0.68410	С	-10.44405	-10.28293	0.67834
Н	1.09495	3.54505	-0.86943	Н	-4.13725	6.66640	1.36475	С	-9.47139	-10.90949	1.51617
С	-0.03127	-3.10839	-0.56623	С	-5.34391	3.87607	-1.03484	С	-8.30700	-10.09166	1.53568
Н	0.85356	-3.72239	-0.63209	Η	-5.67539	3.10030	-1.71757	Н	-10.35848	-8.37668	-0.50086
С	-2.18615	-2.38045	-0.48345	С	4.30397	-2.36543	0.91663	Н	-11.42297	-10.67492	0.43940
С	-5.51579	-0.54765	0.04293	Η	4.83552	-2.45306	1.86007	Н	-9.58623	-11.85857	2.02121
Н	-6.31640	-1.17095	0.41775	С	-6.06835	5.05542	-0.93168	Н	-7.38460	-10.29351	2.06142
С	2.40265	-1.52428	-0.34342	Η	-6.96837	5.19642	-1.52144	Fe	-8.71650	-10.68730	-0.41128
С	-4.16296	3.67460	-0.29142	С	-5.64381	6.08752	-0.06848	С	-8.29283	-12.57824	-1.16504
С	-3.41381	2.40967	-0.41058	Н	5.93152	3.11723	-0.34946	С	-9.23162	-11.91229	-2.00968
С	-3.60717	-2.28463	-0.43563	Η	5.74908	-3.52245	-0.19159	С	-7.10287	-11.78947	-1.12015
С	1.13185	0.62649	-0.38264	С	-6.38884	7.29430	0.04069	Н	-8.46517	-13.49999	-0.62645
С	3.30549	1.29022	-1.51592	С	-7.03991	8.31831	0.13298	С	-8.62211	-10.71165	-2.48660
Н	2.97078	0.77945	-2.41327	С	-7.81437	9.49992	0.24380	Н	-10.23945	-12.24150	-2.22253
С	-5.63924	-3.63888	-0.96599	С	-7.56400	10.61354	1.12369	С	-7.30651	-10.63711	-1.93760
Н	-5.92295	-2.87270	-1.68028	С	-8.60156	11.56946	0.93689	Н	-6.21528	-12.01003	-0.54319
С	4.81630	-2.96742	-0.23403	С	-9.49567	11.07029	-0.05892	Н	-9.08648	-9.97310	-3.12539
С	-1.33844	-3.53622	-0.60019	С	-9.01543	9.80325	-0.49314	Н	-6.60675	-9.82398	-2.07450
Н	-1.68307	-4.55324	-0.71711	Н	-6.72117	10.69777	1.79491	С	-6.93502	-6.89838	0.31152
С	-4.07966	-4.51287	0.65628	Н	-8.67950	12.52409	1.43853	С	-7.67413	-7.84860	0.48964
Н	-3.17690	-4.40649	1.24837	С	-4.43355	-3.49347	-0.25023	Н	-7.37210	-4.84793	-1.35782
С	4.16278	2.61807	0.78069	С	-6.45163	-4.75065	-0.79127	С	3.10629	-1.65184	0.86229

Table S4. Cartesian coordinates of the S_0 optimized geometry of the triad 2

Atom	X	Y	Z	Atom	X	Y	Z	Atom	Х	Y	Z
С	-1.04993	-1.52366	0.01855	С	-0.34794	-3.61888	6.93704	Н	6.49661	-1.69622	9.05788
С	1.07459	-1.14810	-0.32676	Н	-0.25931	-4.42694	6.21607	Н	5.35451	-0.75841	10.02235
Ν	0.07997	-1.05806	0.57059	С	3.79083	-0.94736	3.49301	Н	5.32031	-2.52812	10.07536
С	2.43112	-1.00636	0.10861	Н	4.49867	-0.78161	4.29165	С	-7.58932	-2.56212	-1.03024
С	-2.13445	-1.83337	0.88968	С	3.98980	-0.74760	2.15859	Н	-8.26432	-1.91154	-0.46001
С	2.74027	-1.09215	1.48975	Н	4.87998	-0.36374	1.68046	Н	-7.69740	-2.29056	-2.08536
С	-1.89495	-1.94347	2.28816	С	4.71407	-1.53887	-0.81929	Н	-7.94303	-3.58956	-0.90503
С	2.41180	-1.41489	3.64393	Н	4.86124	-2.25267	-0.01536	С	6.62237	-0.27150	-3.87751
Ν	1.84043	-1.50359	2.43329	С	5.70575	-1.36605	-1.78151	Н	6.20730	-0.06402	-4.86861
С	-0.71968	-2.05940	4.21161	Н	6.62715	-1.93832	-1.70523	Н	7.25881	0.58107	-3.60811
С	1.74872	-1.72833	4.88737	С	4.32037	0.21953	-2.93606	Н	7.27112	-1.14878	-3.95692
С	0.39209	-2.02384	5.12646	Н	4.15764	0.91148	-3.75896	С	-1.68043	-2.40168	-2.36350
С	2.67528	-1.71153	6.07723	С	3.32299	0.04943	-1.98049	С	-2.41005	-2.88130	-3.21284
С	-3.50264	-2.06207	0.37550	Н	2.39924	0.61429	-2.05756	Fe	-3.58615	-5.44626	-4.62090
С	3.50109	-0.82930	-0.89481	С	0.54768	-1.59973	-1.60202	С	-3.27894	-3.42517	-4.18984
С	5.53232	-0.48088	-2.85415	Н	1.10554	-1.78556	-2.50918	С	-2.95565	-3.70868	-5.56556
С	-6.16151	-2.40682	-0.56273	С	-0.79942	-1.85071	-1.40977	С	-4.66192	-3.78127	-3.99725
Ν	-0.64751	-1.94394	2.86518	С	-2.83394	-2.08390	3.37119	С	-4.12506	-4.21098	-6.20373
С	4.47975	-1.67971	8.26880	Н	-3.90851	-2.09720	3.26412	Н	-1.98291	-3.57023	-6.01565
С	3.98506	-0.48892	7.72262	С	-2.12054	-2.15896	4.54358	С	-5.17604	-4.25655	-5.23670
Н	4.29351	0.46346	8.14766	Н	-2.51816	-2.25269	5.54250	Н	-5.18994	-3.70681	-3.05773
С	3.10089	-0.50240	6.64357	С	-4.08080	-1.14392	-0.51981	Н	-4.19506	-4.53455	-7.23313
Н	2.73352	0.43554	6.23621	Н	-3.50187	-0.28483	-0.84363	Н	-6.17922	-4.62221	-5.40767
С	3.17061	-2.90430	6.62027	С	-5.38521	-1.31165	-0.97210	С	-4.27740	-7.39115	-4.36570
Η	2.85725	-3.85347	6.19577	Н	-5.81094	-0.57941	-1.65371	С	-3.25272	-7.36180	-5.35957
С	4.05428	-2.88650	7.69813	С	-5.57682	-3.33545	0.30880	С	-3.72150	-6.90111	-3.14443
Н	4.42042	-3.82737	8.10254	Н	-6.14900	-4.20327	0.62780	Η	-5.30292	-7.69763	-4.52017
С	-0.58908	-1.56811	8.79171	С	-4.27396	-3.16706	0.77477	С	-2.06360	-6.85400	-4.75211
Н	-0.68234	-0.75550	9.50854	Н	-3.83848	-3.90810	1.43754	Η	-3.36675	-7.64262	-6.39769
С	-0.89697	-2.87652	9.18948	Н	0.29211	-1.76047	2.39455	С	-2.35339	-6.57098	-3.38334
С	-0.76718	-3.89579	8.23892	С	-1.37811	-3.16870	10.59146	Н	-4.25141	-6.77020	-2.21113
Н	-0.99726	-4.92112	8.51897	Н	-2.44573	-2.94018	10.70199	Н	-1.11883	-6.68216	-5.24921
С	-0.16521	-1.28833	7.49419	Н	-1.24167	-4.22282	10.85001	Н	-1.67333	-6.12884	-2.66810
Н	0.06746	-0.26609	7.21226	Н	-0.84208	-2.56671	11.33231				
С	-0.03056	-2.31151	6.54556	С	5.46037	-1.66434	9.41801				

Table S4. Cartesian coordinates of the S_0 optimized geometry of the dyad 3