Supplementary Information

Amine-Tethered Phenylboronic Acid-Enabling Ring-Opening Strategy for Carbon Chain Elongation from Double Aldol Cyclic Hemiacetals

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Materials and Methods

Reactions were carried out under argon atmosphere unless otherwise noted. Purified compounds were further dried under high vacuum. Diastereoselectivity of the products was determined by ¹H NMR analysis or LC/MS analysis of the crude mixture, comparing authentic samples. Yields refer to the diastereo mixture of compounds. Thin layer chromatography (TLC) was performed using EMD TLC plates pre-coated with 254 μ m thickness silica gel 60 F₂₅₄ plates and visualized by fluorescence quenching under UV light and p-anisaldehyde stains. Flash chromatography was performed using silica gel 60 (230-400 mesh ASTM) or silica gel 60N (40-100 µm) purchased from Merck or Kanto chemical, respectively. NMR spectra were recorded on either a JEOL ECX 500 spectrometer operating at 500 MHz and 125 MHz for ¹H and ¹³C acquisitions, respectively, or a JEOL ECS 400 spectrometer operating at 400 MHz, 125 MHz and 100 MHz for ¹H, ¹¹B and ¹³C acquisitions, respectively. Chemical shifts are reported in ppm with the solvent resonance as the internal standard (¹H: CDCl₃, δ 7.26; CD₃OD, δ 3.31; (CD₃)₂CO, δ 2.05), (¹³C: CDCl₃, δ 77.16; CD₃OD, δ 49.00; (CD₃)₂CO, δ 29.84). Data is reported as follows: s = singlet, br = broad, d = doublet, t = triplet, q = quartet, m = multiplet; coupling constants in Hz; integration. All deuterated solvents were purchased from Kanto chemical or Sigma-Aldrich. IR spectra were measured on a JASCO FT/IR 410 spectrophotometer. High-resolution mass spectra were obtained using a JEOL JMS-T100LC AccuTOF spectrometer. LC/MS data were obtained using a Shimazu Prominence-iLC-2030/LCMS-2020. Preparative HPLC was performed on a Shimazu SPD-20A/CTO-20AC using 2 cm × 25cm Daicel Chiralpak IA. Optical rotations were measured on a JASCO P-1010 polarimeter.

MesCu was either purchased from Strem or synthesized according to the literature.¹ DTBM-segphos was donated by Takasago International Corporation. Liquid aldehydes, Et₃N and Triisopropyl borate were purified by distillation. All the other chemicals were used as received. THF was deoxidized, stabilizer-free, and organic synthesis grade; acetone was super dehydrated, organic synthesis grade; toluene was JIS special grade. These solvents were purchased from Wako Pure Chemical Industries and used as received without further purification. Compound **2e** was prepared according to the literature.²

Experimental Data





Cyclic hemiacetal **2** (0.1 mmol) and boronic acid **7** (0.105 mmol) were added to a test tube, followed by the addition of toluene (2 mL) at 23 °C. After stirred for 20 hours at 100 °C, the solvent was evaporated.

Deuterated chloroform was added to this crude mixture, followed by the addition of 1,1,2,2tetrachloroethane as internal standard. The solution was transferred to an NMR tube. The yield was determined by ¹H-NMR taken at room temperature.



A Representative Procedure for Synthesis of Cyclic Hemiacetal

Under an argon atmosphere, mesitylcopper (1.8 mg, 0.01 mmol) and (S)-DTBM-segphos (11.8 mg, 0.01 mmol) were added to a round-bottom flask, followed by the addition of THF (0.4 mL) and 4-MeO-C₆H₄-OH (24.8 mg, 0.2 mmol) at 23 °C. After cooled to -60 °C, aldehyde (0.2 mmol) and a solution of boron enolate (0.8 mmol) in THF (0.8 mL), prepared according to the literature³, were added, and the mixture was stirred for 24 hours at -60 °C. Water was added, and the mixture was allowed to warm up to room temperature. The organic layer was separated, and the aqueous phase was extracted with EtOAc. The combined organic layers were washed with brine, dried over Na₂SO₄. After evaporation, the residue was purified by column chromatography on silica gel.



Effects of Stereochemistry of Cyclic Hemiacetal on the Ring-Opening Reaction

Using 7g as the optimized ring-opening boron reagent, we examined the reactivity of cyclic hemiacetals bearing stereochemistry different from 2a. 2a reacted with 7g to give thermodynamically stable boron ester 6ag in 98% yield. 2a and 7g could afford boron ester 2b'; however, 2b' is destabilized by 1,3-diaxial interaction of methyl groups. In the case of 2e, target product 6eg was obtained in only 27% yield. We assume that relatively stable boron ester 2e' is formed, and it disturbed ring-opening process of cyclic hemiacetal. In entry 3, we examined the reactivity of 7g with cyclic hemiacetal 2f which could form a similar boron ester 2f' such as 2e'. The ring opening reaction did not proceed at all. This result supports our hypothesis.

Synthesis of Boronic acids (2-(methoxymethyl)-6-methylphenyl)boronic acid (7e)



Under an argon atmosphere, tetramethylethylenediamine (96 μ L, 0.64 mmol) was added to a solution of 2-bromo-1-(methoxymethyl)-3-methylbenzene⁴ (536 mg, 2.49 mmol) in THF (20mL). The mixture was cooled at -78 °C, and buthyllithium (2.65M, 1.46 mL, 3.87 mmol) was slowly added to the solution. The reaction mixture was stirred for an hour. Triisopropyl borate (1.7 mL, 7.41 mmol) was added to the solution. After stirring for 20 hours at room temperature, the solvent was evaporated. The reaction mixture was dissolved in dichloromethane, and insoluble materials were filtrated. Water was added to the filtrate, and the mixture was stirred overnight at room temperature. The solvent was evaporated, and the resulting residue was washed with hexane. The obtained solid was dissolved in dichloromethane and the insoluble solid was removed by filtration. After evaporation of solvent, obtained solid was used for the ring-opening reaction without further purification.

(2-bromo-3-methylbenzyl)(methyl)sulfane (13)



2-Bromo-1-(bromomethyl)-3-methylbenzene⁵ (242.8 mg, 0.92 mmol) Et_2O (1.0 mL), H₂O (1.1 mL), sodium methanethiolate (0.193mg, 2.76 mmol), and sodium iodide (27.6 mg, 0.18 mmol) were added to a round bottom flask. After stirring the mixture for overnight at room temperature, the organic layer was separated, and the aqueous phase was extracted with Et_2O . The combined organic layers were washed with brine, dried over Na₂SO₄, filtered and concentrated. The residue was purified by column chromatography on silica gel eluting with hexane to afford desired product **13** (116.5 mg, 0.50 mmol, 55% yield) as yellow oil.

NMR spectroscopy: ¹H NMR (400 MHz, CDCl₃): δ = 7.17-7.14 (m, 3H), 3.85 (s, 2H), 2.44 (s, 3H), 2.08 (s, 3H). ¹³C NMR (100 MHz, CDCl₃): δ = 139.27, 137.94, 129.66, 128.24, 127.27, 126.77, 39.55, 24.12, 15.42. IR spectroscopy (CDCl₃, cm⁻¹): 2909, 2357, 1435, 1026, 786, 738. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₉H₁₁BrS [M+Ag]⁺: 338.8790, found: 338.8794.

(2-methyl-6-((methylthio)methyl)phenyl)boronic acid (7f)



Under an argon atmosphere, tetramethylethylenediamine (118 μ L, 0.79 mmol) was added to a solution of (2-bromo-3-methylbenzyl)(methyl)sulfane **13** (706 mg, 3.05 mmol) in THF (25 mL). The mixture was cooled at -78 °C, and buthyllithium (2.65M, 1.46 mL, 4.74 mmol) was slowly added to the solution. The reaction mixture was stirred for an hour. Triisopropyl borate (1.7 mL, 8.63 mmol) was added to the solution. After stirring for 20 hours at room temperature, the solvent was evaporated. The reaction mixture was dissolved in dichloromethane, and insoluble materials were filtrated. Water was added to the filtrate, and the mixture was stirred overnight at room temperature. After evaporation, the residue was collected and washed with hexane. The obtained solid was dissolved in THF and the insoluble solid was removed by filtration. After evaporation of solvent, obtained solid was used for the ring-opening reaction without further purification.

(2-((dimethylamino)methyl)-6-methylphenyl)boronic acid (7g)



(2-((Dimethylamino)methyl)-6-methylphenyl)boronic acid **7g** was prepared according to the literature⁶. ¹H NMR was taken using deuterated acetone and a drop of D₂O.

NMR spectroscopy: ¹H NMR (400 MHz, (CD₃)₂CO): $\delta = 6.99$ (dd, J = 7.8 Hz, 7.3 Hz, 1H), 6.87 (d, J = 7.8 Hz, 1H), 6.80 (d, J = 7.3 Hz, 1H), 3.78 (s, 2H), 2.45 (s, 6H), 2.34 (s, 3H). ¹³C NMR (100 MHz, (CD₃)₂CO): $\delta = 141.4$, 141.1, 128.4, 127.5, 120.8, 64.2, 44.8, 20.4. IR spectroscopy (CDCl₃, cm⁻¹): 3281, 2922, 2357, 1457, 1009, 755. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₀H₁₆ BNO₂ [M+Na]⁺: 216.1166, found: 216.1171.

Synthesis of Cyclic Hemiacetals (3*R*,4*S*,5*S*,6*S*)-3,5-dimethyl-6-phenethyltetrahydro-2H-pyran-2,4-diol (2a)



Under an argon atmosphere, mesitylcopper (82.2 mg, 0.45 mmol) and (S)-DTBM-segphos (530.8 mg, 0.45 mmol) were added to a round bottom flask, followed by the addition of THF (8 mL) and 4-MeO-C₆H₄OH (1117 mg, 9 mmol) at 23 °C. After cooled to -60 °C, a solution of boron enolate (36 mmol) in THF (30 mL) and aldehyde (4 mmol) were added, and the mixture was stirred for 24 hours at -60 °C. To this solution was added water, and the mixture was allowed to warm up to room temperature. The organic layer was separated, and the aqueous phase was extracted with EtOAc. The combined organic layers were washed with brine, dried over Na₂SO₄. After filtration and evaporation, the residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 4:1 to 1:1 to afford desired product (1.5g, 6.0 mmol, 67% yield) as a white solid (**2a** : anomer = 1 : 0.15).

NMR spectroscopy: ¹H NMR (400 MHz, CDCl₃): δ = 7.31-7.28 (m, 2H), 7.23-7.18 (m, 3H), 4.80 (dd, J = 6.5, 4.6 Hz, 1H), 4.17 (dt, J = 10.9, 4.3 Hz, 1H), 3.44 (dd, J = 12.6, 6.0 Hz, 1H), 3.03 (d, J = 6.5 Hz, 1H), 2.86-2.80 (m, 1H), 2.68-2.62 (m, 1H), 2.00 (d, J = 6.0 Hz, 1H), 1.89-1.79 (m, 2H), 1.77-1.64 (m, 2H), 1.12 (d, J = 7.2 Hz, 3H), 1.00 (d, J = 7.2 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃): δ = 142.2, 128.7, 128.5, 126.1, 96.0, 75.8, 69.9, 42.6, 39.9, 32.4, 31.4, 14.6, 13.1. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₅H₂₂O₃ [M+Na]⁺: 273.1461, found 273.1460.





Under an argon atmosphere, mesitylcopper (36 mg, 0.2 mmol) and (S)-DTBM-segphos (236 mg, 0.2 mmol) were added to a round bottom flask, followed by the addition of THF (4 mL) and isopropanol (304 μ L, 4 mmol) at 23 °C. After cooled to -60 °C, a solution of boron enolate (16 mmol) in THF (16 mL) and aldehyde (4.0 mmol) were added, and the mixture was stirred for 24

hours at -60 °C. To this solution was added water, and the mixture was allowed to warm up to room temperature. The organic layer was separated, and the aqueous phase was extracted with EtOAc. The combined organic layers were washed with brine, dried over Na₂SO₄. After filtration and evaporation, the residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 3:1 to 1:1 to afford (2S,3S)-2-ethyl-3-hydroxy-5-phenylpentanal (770.3 mg, 3.7 mmol, 93% yield) as a white solid.

Under an argon atmosphere, mesitylcopper (18 mg, 0.1 mmol) and (S)-DTBM-segphos (118 mg, 0.1 mmol) were added to a round bottom flask, followed by the addition of THF (2 mL), isopropanol (7.7 μ L, 0.1 mmol), and triethylamine (279 μ L, 2 mmol) at 23 °C. After cooled to – 60 °C, a solution of boron enolate (3.0 mmol) in THF (3.0 mL) and (2*S*,3*S*)-2-ethyl-3-hydroxy-5-phenylpentanal (1.0 mmol) were added, and the mixture was stirred for 24 hours at -60 °C. To this solution was added water, and the mixture was allowed to warm up to room temperature. The organic layer was separated, and the aqueous phase was extracted with EtOAc. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered and concentrated. The residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 3:1 to 1:1 to afford desired product (108.3 mg, 0.39 mmol, 39% yield) as a white solid (**2b** : anomer = 1 : 0.11).

NMR spectroscopy: ¹H NMR (500 MHz, CDCl₃): δ = 7.30-7.27 (m, 2H), 7.23-7.18 (m, 3H), 4.98 (dd, J = 6.5 Hz, 2.6 Hz, 1H), 4.29 (dt, J = 10.3 Hz, 3.6 Hz, 1H), 3.83-3.79 (m, 1H), 3.29 (d, J = 6.5 Hz, 1H), 2.88-2.82 (m, 1H), 2.67-2.66 (m, 1H), 2.20 (d, J = 5.7 Hz, 1H), 1.92-1.84 (m, 1H), 1.70-1.63 (m, 1H), 1.61-1.38 (m, 6H), 0.99 (t, J = 6.9 Hz, 3H), 0.94 (t, J = 7.4 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃): δ = 142.2, 128.7, 128.5, 126.0, 94.7, 71.6, 67.6, 48.4, 46.8, 32.6, 32.1, 22.0, 19.6, 13.0, 12.6. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₇H₂₆O₃ [M+Na]⁺: 301.1774, found: 301.1773.

(3R,4S,5S,6S)-5-ethyl-3-methyl-6-phenethyltetrahydro-2H-pyran-2,4-diol (2c)



Under an argon atmosphere, mesitylcopper (27 mg, 0.2 mmol) and (S)-DTBM-segphos (177 mg, 0.2 mmol) were added to a round bottom flask, followed by the addition of THF (2 mL) and isopropanol (152 μ L, 2.0 mmol) at 23 °C. After cooled to -60 °C, a solution of boron enolate (6.0 mmol) in THF (6.0 mL) and (2*S*,3*S*)-2-ethyl-3-hydroxy-5-phenylpentanal (2.0 mmol) were added, and the mixture was stirred for 24 hours at -60 °C. To this solution was added water, and the mixture was allowed to warm up to room temperature. The organic layer was separated, and the

aqueous phase was extracted with EtOAc. The combined organic layers were washed with brine, dried over Na_2SO_4 , filtered and concentrated. The residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 3:1 to 1:1 to afford desired product (228.9 mg, 0.86 mmol, 43% yield) as a white solid (**2c** : anomer = 1 : 0.09).

NMR spectroscopy: ¹H NMR (500 MHz, CDCl₃): δ = 7.30-7.26 (m, 2H), 7.23-7.18 (m, 3H), 4.72 (dd, J = 6.6 Hz, 6.3 Hz, 1H), 4.17 (dt, J = 11.5, 4.0 Hz, 1H), 3.45-3.41 (m, 1H), 3.17 (d, J = 6.6 Hz, 1H), 2.85-2.83 (m, 1H), 2.66-2.64 (m, 1H), 1.96 (d, J = 6.6 Hz, 1H), 1.86-1.82 (m, 1H), 1.72-1.58 (m, 4H), 1.29-1.23 (m, 1H), 1.12 (d, J = 6.7 Hz, 3H), 0.89 (t, J = 7.2 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃): δ = 142.0, 128.6, 128.5, 126.1, 95.2, 73.6, 70.9, 47.1, 44.4, 32.4, 29.8, 20.4, 13.9, 12.1. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₆H₂₄O₃ [M+Na]⁺: 287.1618, found 287.1620.

(3R,4S,5S,6R)-3,5-diethyl-6-phenyltetrahydro-2H-pyran-2,4-diol (2d)



Under an argon atmosphere, mesitylcopper (46 mg, 0.25 mmol) and (S)-DTBM-segphos (295 mg, 0.25 mmol) were added to a round bottom flask, followed by the addition of THF (5 mL), isopropanol (19 μ L, 0.25 mmol) and triethylamine (1.4 mL, 10 mmol) at 23 °C. After cooled to – 60 °C, a solution of boron enolate (20 mmol) in THF (20 mL) and aldehyde (5 mmol) were added, and the mixture was stirred for 24 hours at –60 °C. To this solution was added water, and the mixture was allowed to warm up to room temperature. The organic layer was separated, and the aqueous phase was extracted with EtOAc. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered and concentrated. The residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 4:1 to 1:1 to afford desired product (402.8 mg, 1.6 mmol, 32% yield) as a white solid (**2d** : aldehyde form = 1 : 0.08).

NMR spectroscopy: ¹H NMR (500 MHz, CDCl₃): $\delta = 7.33-7.32$ (m, 4H), 7.26-7.22 (m, 1H), 5.62 (d, J = 3.4 Hz, 1H), 5.28 (d, J = 5.2 Hz, 1H), 4.08-4.01 (m, 1H), 3.70 (brs, 1H), 3.01 (brs, 1H), 1.81-1.77 (m, 2H), 1.67-1.60 (m, 1H), 1.54-1.46 (m, 1H), 1.28-1.13 (m, 2H), 1.03 (t, J = 5.7 Hz, 3H), 0.58 (t, J = 7.4 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃): $\delta = 140.8$, 128.2, 126.9, 126.0, 96.3, 72.0, 67.1, 48.2, 46.3, 23.0, 19.4, 13.4, 13.3. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₅H₂₂O₃ [M+Na]+: 273.1461, found 273.1462.





Under an argon atmosphere, mesitylcopper (36 mg, 0.2 mmol) and (S)-DTBM-segphos (236 mg, 0.2 mmol) were added to a round bottom flask, followed by the addition of THF (4 mL) and isopropanol (304 µL, 4 mmol) at 23 °C. After cooled to -60 °C, a solution of boron enolate (4.8 mmol) in THF (4.8 mL) and aldehyde (4 mmol) were added, and the mixture was stirred for 24 hours at -60 °C. To this solution was added water, and the mixture was allowed to warm up to room temperature. The organic layer was separated, and the aqueous phase was extracted with EtOAc. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered and concentrated. The residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 2:1 to afford desired product (607.5 mg, 3.16 mmol, 79% yield) as a colorless oil. Under an argon atmosphere, mesitylcopper (30 mg, 0.16 mmol) and (R)-DTBM-segphos (194 mg, 0.16 mmol) were added to a round bottom flask, followed by the addition of THF (3 mL) and isopropanol (125 µL, 1.6 mmol) at 23 °C. After cooled to -60 °C, a solution of boron enolate (3.2 mmol) in THF (3.2 mL) and aldehyde (1.6 mmol) were added, and the mixture was stirred for 24 hours at -60 °C. To this solution was added water, and the mixture was allowed to warm up to room temperature. The organic layer was separated, and the aqueous phase was extracted with EtOAc. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered and concentrated. The residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 4:1 to 1:1 to afford desired product (102.6 mg, 0.41 mmol, 26% yield) as a white solid (2f: anomer = 1 : 0.5).

NMR spectroscopy: ¹H NMR (500 MHz, CDCl₃): δ = 7.30-7.27 (m, 2H), 7.21-7.18 (m, 3H), 4.30 (dd, J = 8.0, 6.3 Hz, 1H), 3.43-3.39 (m, 2H), 3.21 (d, J = 6.3 Hz, 1H), 2.80-2.71 (m, 1H), 2.67-2.57 (m, 1H), 2.10-1.85 (m, 2H), 1.73-1.53 (m, 3H), 1.08 (d, J = 6.3 Hz, 3H), 0.95 (d, J = 6.9 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃): δ = 141.9, 128.6, 128.5, 126.0, 99.7, 74.2, 71.6, 40.2, 38.2, 34.1, 32.3, 12.7, 5.7. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₅H₂₂O₃ [M+Na]⁺: 273.1461, found 273.1461.

Ring-Opening Reaction of 2a



Cyclic hemiacetal **2a** (12.5 mg, 0.05 mmol) and **7g** (10.1 mg, 0.053 mmol) were added to a test tube, followed by the addition of toluene (1 mL) at 23 °C. After stirred for 20 hours at 100 °C, the solvent was evaporated. Deuterated chloroform was added to this crude mixture, followed by the addition of 1,1,2,2-tetrachloroethane (10.6 μ L, 0.1 mmol) as an internal standard. The solution was transferred to an NMR tube. ¹H-NMR and ¹¹B-NMR were taken at room temperature. The yield was determined by ¹H NMR using the internal standard.

NMR spectroscopy: ¹H NMR (500 MHz, CDCl₃): δ = 9.82 (d, J = 2.3 Hz, 1H), 7.30-7.27 (m, 2H), 7.25-7.15 (m, 3H), 7.10 (dd, J = 7.3 Hz, 7.3 Hz, 1H), 7.01 (d, J = 7.3 Hz, 1H), 6.84 (d, J = 7.3 Hz, 1H), 4.20 (dd, J = 8.9 Hz, 2.5Hz, 1H), 4.11-4.07 (m, 1H), 3.74 (d, J = 13.7 Hz, 1H), 3.70 (d, J = 13.7 Hz, 1H), 2.90-2.83 (m, 1H), 2.77-2.69 (m, 1H), 2.67-2.59 (m, 1H), 2.53 (s, 6H), 2.47 (s, 3H), 2.01-1.92 (m, 1H), 1.80-1.68 (m, 2H), 1.22 (d, J = 7.3 Hz, 3H), 1.07 (d, J = 7.3 Hz, 3H). ¹¹B NMR (125 MHz, CDCl₃): δ = 13.7.

ethyl (4S,5R,6R,7S,E)-5,7-dihydroxy-4,6-dimethyl-9-phenylnon-2-enoate (9)



To a solution of **6ag** (0.05 mmol) in THF (1.0 mL) was added ethyl(triphenylphosphoranylidene) acetate (34.8 mg, 0.10 mmol) at room temperature. After stirring for 5 hours, the solvent was removed by evaporation. To this residue, MeOH (1.0 mL) and 2-amino-2-methyl-1,3-propanediol (15.8 mg, 015 mmol) were added. After stirring for 24 hours at 50 °C, the reaction mixture was concentrated under reduced pressure. The resulting mixture was diluted with EtOAc and washed with brine, dried over Na₂SO₄, filtered and concentrated. The residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 2:1 to afford desired product **9** (12.1

mg, 0.037 mmol, 76% yield) as a colorless oil.

NMR spectroscopy: ¹H NMR (500 MHz, CDCl₃): $\delta = 7.31 \cdot 7.28$ (m, 2H), 7.21-7.18 (m, 3H), 6.75 (dd, J = 15.6, 9.5 Hz, 1H), 5.85 (d, J = 15.6 Hz, 1H), 4.19 (q, J = 7.2 Hz, 2H), 3.83-3.82 (m, 1H), 3.60 (d, J = 8.6 Hz, 1H), 3.07 (brs, 1H), 2.75-2.71 (m, 1H), 2.67-2.63 (m, 1H), 2.51-2.46 (m, 1H), 2.29 (brs, 1H), 1.88-1.82 (m, 1H), 1.74-1.70 (m, 1H), 1.56-1.53 (m, 1H), 1.30 (t, J = 7.2 Hz, 3H) , 1.13 (d, J = 6.7 Hz, 3H) , 0.90 (d, J = 6.7 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃): $\delta = 166.7$, 150.3, 141.7, 128.7, 128.6, 126.2, 121.7, 80.2, 77.0, 60.6, 41.1, 39.1, 37.2, 32.6, 17.0, 14.4, 4.4. IR spectroscopy (CDCl₃, cm⁻¹): 3413, 2932, 1717, 1651, 700. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₉H₂₈O₄ [M+Na]⁺: 343.1880, found 343.1881. Optical rotation: $[\alpha]_D^{25.6} = -28.5$ (c = 0.23, MeOH).





To a solution of **6ag** (0.05 mmol) in toluene (1.0 mL) was added diisopropyl allylboronate (20.8 μ L, 0.10 mmol) at room temperature. After stirring for 2 hours, the solvent was removed by evaporation. To this residue, MeOH (1.0 mL) and 2-amino-2-methyl-1,3-propanediol (15.8 mg, 015 mmol) were added. After stirring for 24 hours at 50 °C, the reaction mixture was concentrated under reduced pressure. The resulting mixture was diluted with EtOAc and washed with brine, dried over Na₂SO₄, filtered and concentrated. After the determination of diastereoselectivity by ¹H NMR, the residue was purified by column chromatography on silica gel eluting with hexane/EtOAc = 1:1 to afford desired product **10** (10.7 mg, 0.036 mmol, 73% yield, 2.6/1 dr) as a colorless oil.

NMR spectroscopy: Major isomer: ¹H NMR (500 MHz, CDCl₃): $\delta = 7.31-7.28$ (m, 2H), 7.22-7.18 (m, 3H), 5.83-5.74 (m, 1H), 5.16 (d, J = 5.7 Hz, 1H), 5.13 (s, 1H), 3.87 (dd, J = 4.6 Hz, 4.6Hz, 1H), 3.84-3.79 (m, 2H), 2.75 (m, 1H), 2.64 (m, 1H), 2.23 (m, 1H), 2.17 (brs, 1H), 2.04 (brs, 1H), 1.87 (m, 1H), 1.71 (m, 2H), 0.98 (d, J = 7.4 Hz, 3H), 0.96 (d, J = 6.9 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃): $\delta = 142.0, 135.0, 128.6, 128.6, 126.1, 118.5, 79.7, 75.1, 74.0, 40.1, 40.0, 39.7, 37.3, 32.7, 6.9, 6.7.$ IR spectroscopy (CDCl₃, cm⁻¹): 3383, 2924, 1716, 1456, 699. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₇H₂₈O₃ [M+Na]⁺: 315.1931, found 315.1931.

(4S,5S,6R,7S)-4,6-dimethyl-9-phenylnonane-1,3,5,7-tetraol (14)



To a solution of dicyclohexylboron trifluoromethanesulfonate (122.4 mg, 0.38 mmol) in hexane (0.85 mL), triethylamine (62.8 μ L, 0.45 mmol) and S-t-butyl phenylthiothioacetate (85.6 μ L, 0.6 mmol) were added at 0 °C under an argon atmosphere. After stirring for 2 hours, this boron-enolate solution (0.5 mL) was transferred to a solution of **6ag** (0.1 mmol) in dichloromethane (0.33 mL) at -78 °C. The mixture was allowed to stir for 2 hours. To this solution, a solution of LiBH₄ (0.5 mmol) in THF (0.25 mL) was added. The mixture was allowed to warm up to room temperature, and stirred for overnight. To this mixture was added HCl (1 M). The organic layer was separated, and the aqueous phase was extracted with Et₂O. To the combined organic layers, water and an excess of H₂O₂ and NaOH were added. After stirring for overnight at 23 °C, Na₂SO₃ was added to the solution. The layers were separated, and the aqueous phase was extracted with Et₂O. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered and concentrated. After the determination of diastereoselectivity by LC/MS analysis, the residue was purified by column chromatography on silica gel eluting with EtOAc to afford the title compound **14** (18.0 mg, 0.061 mmol, 61%, 3.4/1 dr) as colorless oil. Stereoisomers were separated by preparative HPLC using Daicel Chiralpak IA.

NMR spectroscopy: Major isomer: ¹H NMR (500 MHz, CD₃OD): $\delta = 7.27$ -7.20 (m, 4H), 7.14 (t, J = 7.16 Hz, 1H), 3.79-3.64 (m, 5H), 2.77-2.72 (m, 1H), 2.65-2.59 (m, 1H), 1.85-1.59 (m, 6H), 1.00 (d, J = 6.9 Hz, 3H), 0.89 (d, J = 6.9 Hz, 3H). ¹³C NMR (100 MHz, CD₃OD): $\delta = 143.5$, 129.4, 129.4, 126.8, 75.2, 73.0, 72.3, 60.6, 41.4, 41.3, 38.1, 37.3, 33.5, 10.5, 8.7. IR spectroscopy (CDCl₃, cm⁻¹): 3375, 2924, 1681, 1455, 1203. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₇H₂₈O₄ [M+Na]+: 319.1880, found: 319.1880. Optical rotation: $[\alpha]_D^{29.4} = -61.5$ (c = 0.26, CHCl₃).

Minor isomer: ¹H NMR (500 MHz, CD₃OD): $\delta = 7.27$ -7.20 (m, 4H), 7.14 (t, J = 7.16 Hz, 1H), 3.88-3.84 (m, 1H), 3.72-3.63 (m, 4H), 2.80-2.74 (m, 1H), 2.65-2.59 (m, 1H), 1.86-1.62 (m, 6H), 0.96 (d, J = 6.9 Hz, 3H), 0.90 (d, J = 6.9 Hz, 3H). ¹³C NMR (100 MHz, CD₃OD): $\delta = 143.6$, 129.5, 129.4, 126.8, 78.2, 73.9, 71.8, 60.4, 41.0, 40.7, 38.7, 38.3, 33.4, 8.1, 8.1. IR spectroscopy (CDCl₃, cm⁻¹): 3381, 2924, 1682, 1455. Mass spectroscopy: HRMS-ESI (m/z): calcd for C₁₇H₂₈O₄ [M+Na]⁺: 319.1880, found: 319.1880. Optical rotation: $[\alpha]_D^{29.4} = -138.5$ (c = 0.13, CHCl₃).



(2S,3R,4S,5S,6R,7S)-2,4,6-trimethyl-9-phenylnonane-1,3,5,7-tetraol (15)



Under an argon atmosphere, mesitylcopper (0.9 mg, 0.005 mmol) and (S)-DTBM-segphos (5.9 mg, 0.005 mmol) were dissolved in THF (0.2 mL) at 23 °C. After cooled to -60 °C, a solution of boron enolate (0.15 mmol) in THF (0.15 mL) and **6ag** (0.1 mmol) were added, and the mixture was stirred for 24 hours at -60 °C. To this solution, a solution of LiBH₄ (0.5 mmol) in THF (0.25 mL) was added. The mixture was allowed to warm up to room temperature, and stirred for overnight. To this mixture was added HCl (1 M). The layers were separated, and the aqueous phase was extracted with Et₂O. To the combined organic layers, water and an excess of H₂O₂ and NaOH were added. After stirring for overnight at 23 °C, Na₂SO₃ was added to the solution. The organic layer was separated, and the aqueous phase was extracted with Et₂O. The combined organic layer was extracted with Et₂O.

dried over Na₂SO₄, filtered and concentrated. Yield and diastereoselectivity was determined by LC/MS analysis according to the literature³.



Control Experiments of Transformation of Ring-Opened Aldehyde 6ag



In order to demonstrate that ring-opening of hemiacetal **2a** by amine-tethered phenylboronic acid **7g** is critical for the transformations of **2a**, control experiments were conducted using **2a** without the ring-opening procedure. In the Wittig reaction, the desired product **9** was not obtained, and the starting material was recovered quantitatively. The allylation reaction afforded the desired product **10** in 1% yield. The starting material was recovered mainly.

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S-32



S-33

















