

## Supporting Information

### Iodine-mediated aminosulfonylation of alkenyl sulfonamides with sulfonyl hydrazides: Synthesis of sulfonylmethyl pyrazolines and pyrrolidines

Zhong-Qi Xu,<sup>a</sup> Wen-Bo Wang,<sup>a</sup> Lin-Chuang Zheng,<sup>a</sup> Lin Li,<sup>a</sup> Lili Duan<sup>a</sup> and Yue-Ming Li<sup>a,b\*</sup>

<sup>a</sup> State Key Laboratory of Medicinal Chemical Biology, College of Pharmacy and Tianjin Key Laboratory of Molecular Drug Research, Nankai University, Tianjin, People's Republic of China.

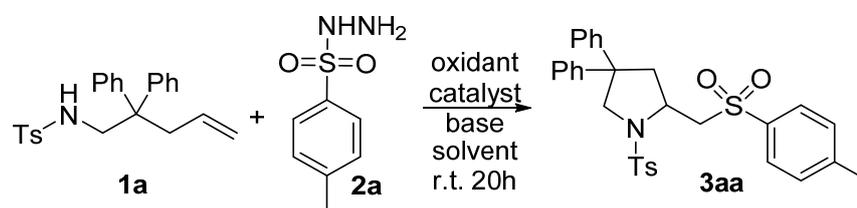
<sup>b</sup> CAS Key Laboratory of Synthetic Chemistry of Natural Substances, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, People's Republic of China.

### Contents

1. Optimization of Reaction Conditions.....	S1
2. Control Reactions Using $\beta,\gamma$ -Unsaturated Hydrazone <b>1r</b> as the Substrate.....	S3
3. X-ray Diffraction Data for <b>3ga</b> and <b>3qa</b> .....	S5
4. Copies of <sup>1</sup> H and <sup>13</sup> C NMR Spectra. ....	S20

## 1. Optimization of Reaction Conditions

Table S1. Optimization of reaction conditions to synthesize sulfone pyrrolidines.<sup>a</sup>

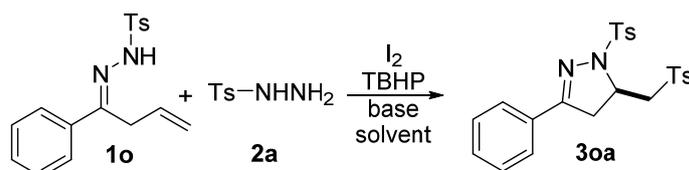


entry	base	solvent	oxidant	catalyst	yield (%) <sup>b</sup>
1	K <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	20
2	NaOAc	1,4-dioxane	TBHP	I <sub>2</sub>	71
3	KF	1,4-dioxane	TBHP	I <sub>2</sub>	N.R. <sup>c</sup>
4	NaHCO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	79
5	DBU	1,4-dioxane	TBHP	I <sub>2</sub>	N.R.
6	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	95
7	<i>t</i> -BuOK	1,4-dioxane	TBHP	I <sub>2</sub>	N.R.
8	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	NaI	82
9	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	NIS	87
10	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	KI	75
11	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	FeI <sub>2</sub>	35
12	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	MnI <sub>2</sub>	54
13	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	CuI	trace
14	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	TBAI	22
15	None	1,4-dioxane	TBHP	I <sub>2</sub>	N.R.
16	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	I <sub>2</sub>	trace
17	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	DTBP	I <sub>2</sub>	trace
18	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	PIDA	I <sub>2</sub>	N.R.
19	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	H <sub>2</sub> O <sub>2</sub>	I <sub>2</sub>	N.R.
20	Na <sub>2</sub> CO <sub>3</sub>	THF	TBHP	I <sub>2</sub>	81
21	Na <sub>2</sub> CO <sub>3</sub>	toluene	TBHP	I <sub>2</sub>	45
22	Na <sub>2</sub> CO <sub>3</sub>	MeCN	TBHP	I <sub>2</sub>	61
23	Na <sub>2</sub> CO <sub>3</sub>	acetone	TBHP	I <sub>2</sub>	47
24	Na <sub>2</sub> CO <sub>3</sub>	DCE	TBHP	I <sub>2</sub>	63
25	Na <sub>2</sub> CO <sub>3</sub>	DMF	TBHP	I <sub>2</sub>	15
26 <sup>d</sup>	NaOAc	1,4-dioxane	TBHP	-	0
27 <sup>e</sup>	NaOAc	1,4-dioxane	TBHP	I <sub>2</sub>	27

entry	base	solvent	oxidant	catalyst	yield (%) <sup>b</sup>
28 <sup>f</sup>	NaOAc	1,4-dioxane	TBHP	I <sub>2</sub>	41
29 <sup>g</sup>	NaOAc	1,4-dioxane	TBHP	I <sub>2</sub>	0

Reaction conditions: <sup>a</sup>**1a** (0.25 mmol), **2a** (0.5 mmol, 2.0 equiv), oxidant (3.0 equiv), solvent (2.0 mL), base (1.5 equiv), catalyst (0.3 equiv), under argon and stirred at room temperature for 20 h. TBHP= *tert*-butyl hydroperoxide (70% aqueous solution), DTBP= Di-*tert*-butyl peroxide, PIDA= (Diacetoxyiodo)benzene. <sup>b</sup>Isolated yield based on **1a**. <sup>c</sup>N.R=No reaction. <sup>d</sup> reaction in the absence of iodine. <sup>e</sup> 10 mol% of iodine was used. <sup>f</sup> 20 mol% of iodine was used. <sup>g</sup> the reaction was carried out in open air system.

Table S2. Optimization of reaction conditions to synthesize sulfone pyrazolines.<sup>a</sup>



entry	base	solvent	oxidant	catalyst	yield (%) <sup>b</sup>
1	Na <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	23
2	NaOAc	1,4-dioxane	TBHP	I <sub>2</sub>	55
3	KF	1,4-dioxane	TBHP	I <sub>2</sub>	N.R. <sup>c</sup>
4	NaHCO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	75
5	DBU	1,4-dioxane	TBHP	I <sub>2</sub>	N.R.
6	K <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	trace
7	CsF	1,4-dioxane	TBHP	I <sub>2</sub>	N.R.
8	NaHCO <sub>3</sub>	MeCN	TBHP	I <sub>2</sub>	40
9	NaHCO <sub>3</sub>	DCE	TBHP	I <sub>2</sub>	37
10	NaHCO <sub>3</sub>	Toluene	TBHP	I <sub>2</sub>	15
11	NaHCO <sub>3</sub>	MeCN/H <sub>2</sub> O (1:1)	TBHP	I <sub>2</sub>	33

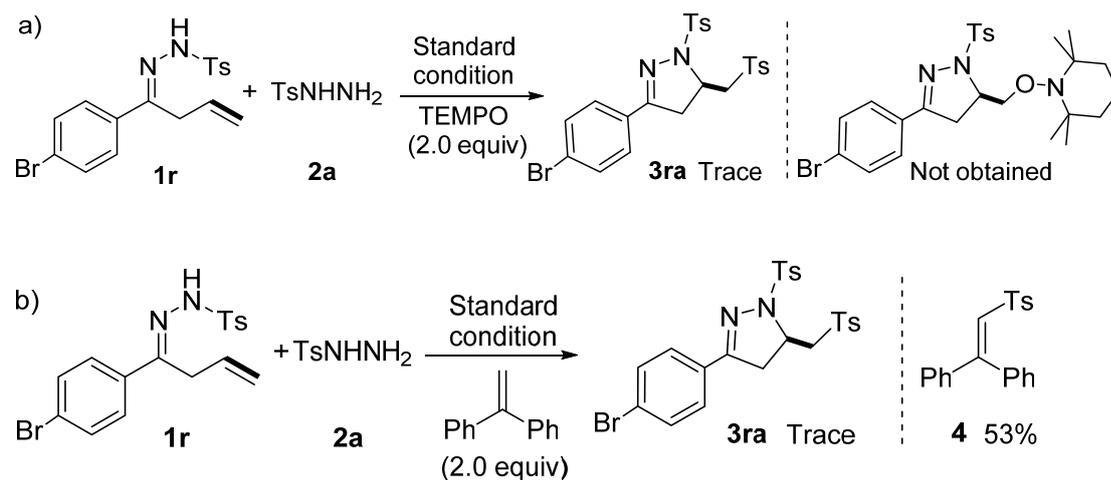
entry	base	solvent	oxidant	catalyst	yield (%) <sup>b</sup>
12	NaHCO <sub>3</sub>	THF	TBHP	I <sub>2</sub>	51
13 <sup>d</sup>	NaHCO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	0
14 <sup>e</sup>	NaHCO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	22
15 <sup>f</sup>	NaHCO <sub>3</sub>	1,4-dioxane	TBHP	I <sub>2</sub>	36

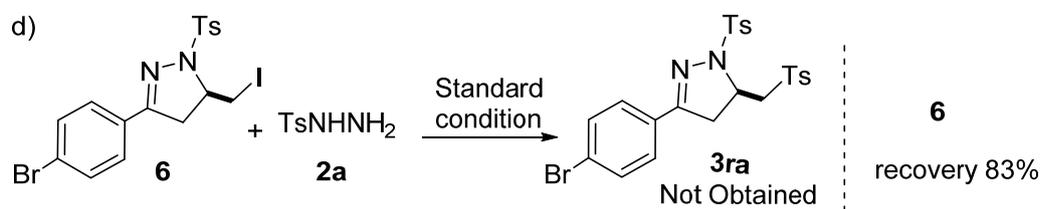
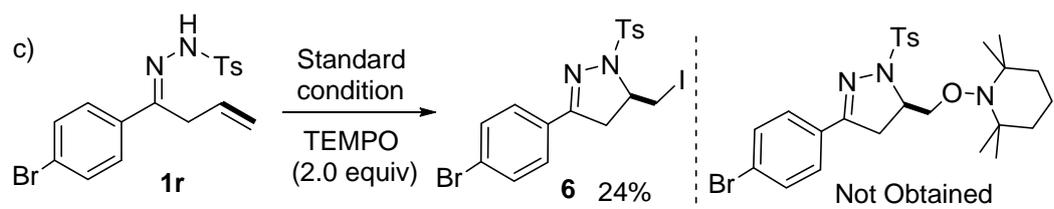
Reaction conditions: <sup>a</sup>**1o** (0.25 mmol), **2a** (0.5 mmol, 2.0 equiv), TBHP (70% aqueous solution, 0.75 mmol, 3.0 equiv), solvent (2.0 mL), base (1.5 equiv), catalyst (0.3 equiv), under argon and stirred at room temperature for 20 h. <sup>b</sup>Isolated yield based on **1o**. <sup>c</sup>N.R.=No reaction. <sup>d</sup> reaction carried out in the absence of iodine. <sup>e</sup> 10 mol% of iodine was used. <sup>f</sup> 20 mol% of iodine was used.

## 2. Control Reactions Using $\beta,\gamma$ -Unsaturated Hydrazone **1r** as the Substrate

The control reactions were proceeded using  $\beta,\gamma$ -unsaturated hydrazone **1r** as substrate and the reaction results did confirm with the results showed in scheme 2, implying that the sulfone pyrazolines and pyrrolidines might be synthesized in the same reaction path way.

Scheme S1. Control Reactions.





### 3. X-ray Diffraction Data for 3ga and 3qa

The crystals of **3ga** was obtained by crystallization from a solution in acetone/petroleum ether after purification by column chromatography.

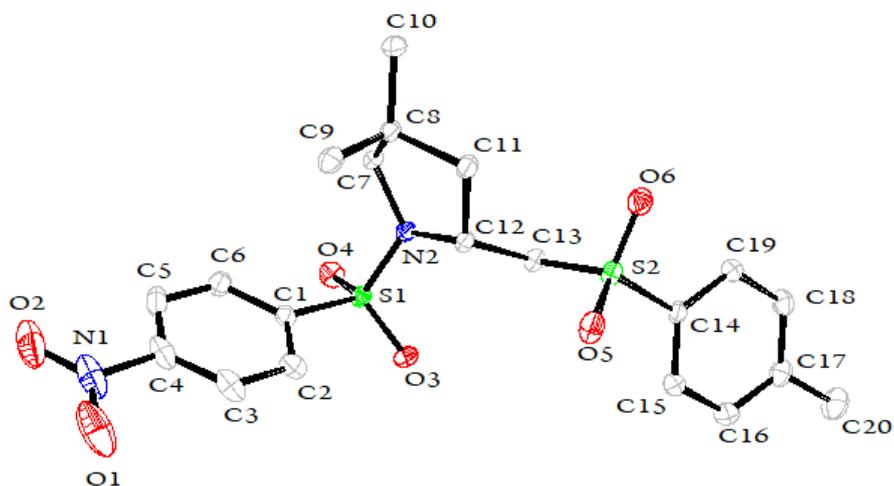
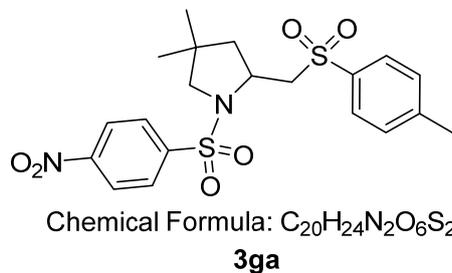


Figure S1. ORTEP drawing of compound **3ga**.

Table S3. Crystal data and structure refinement for **3ga**.

Identification code	3ga
Empirical formula	C <sub>20</sub> H <sub>24</sub> N <sub>2</sub> O <sub>6</sub> S <sub>2</sub>
Formula weight	452.53
Temperature/K	113
Crystal system	monoclinic
Space group	P 21/c
a/Å	11.026(2)
b/Å	15.703(3)
c/Å	12.684(3)
α/°	90

$\beta/^\circ$	98.83(3)
$\gamma/^\circ$	90
Volume/ $\text{\AA}^3$	2170.0(8)
Z	4
$\rho_{\text{calc}}/\text{g cm}^3$	1.385
$\mu/\text{mm}^{-1}$	0.284
F (000)	952.0
Crystal size/ $\text{mm}^3$	0.200×0.180×0.120
Radiation	MoK $\alpha$ (0.71073)
2 $\Theta$ range for data collection/ $^\circ$	2.275 to 27.893
Index ranges	-14 $\leq$ h $\leq$ 13, -20 $\leq$ k $\leq$ 20, -16 $\leq$ l $\leq$ 16
Reflections collected	24978
Independent reflections	5168 [ $R_{\text{int}} = 0.0570$ ]
Data/restraints/parameters	5168/0/275
Goodness-of-fit on $F^2$	1.084
Final R indexes [ $I > 2\sigma(I)$ ]	R1 = 0.0644, wR2 = 0.1619
Final R indexes [all data]	R1 = 0.0775, wR2 = 0.1720
Largest diff. peak/hole / $e \text{\AA}^{-3}$	0.445/ -0.401
Flack parameter	n/a

Table S4. Fractional Atomic Coordinates ( $\times 10^4$ ) and Equivalent Isotropic Displacement Parameters ( $\text{\AA}^2 \times 10^3$ ) for **3ga**. Ueq is defined as 1/3 of the trace of the orthogonalized.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	Ueq
S(1)	4972(1)	2292(1)	1366(1)	26(1)
S(2)	2555(1)	4736(1)	-476(1)	30(1)
O(1)	10878(2)	3281(2)	2502(3)	99(1)
O(2)	10694(3)	2554(2)	3935(3)	89(1)
O(3)	4890(2)	2380(1)	230(1)	35(1)
O(4)	4651(2)	1505(1)	1826(2)	37(1)

Atom	<i>x</i>	<i>y</i>	<i>z</i>	Ueq
O(5)	3693(2)	5010(1)	-780(2)	43(1)
O(6)	1916(2)	5319(1)	132(2)	41(1)
N(1)	10277(3)	2871(2)	3067(3)	67(1)
N(2)	4123(2)	3031(1)	1751(2)	24(1)
C(1)	6532(2)	2475(2)	1907(2)	28(1)
C(2)	7240(2)	2983(2)	1340(2)	37(1)
C(3)	8464(3)	3115(2)	1734(3)	46(1)
C(4)	8950(3)	2739(2)	2676(3)	47(1)
C(5)	8269(3)	2238(2)	3268(3)	48(1)
C(6)	7030(3)	2100(2)	2872(2)	36(1)
C(7)	4024(2)	3134(2)	2901(2)	27(1)
C(8)	4257(2)	4091(2)	3114(2)	28(1)
C(9)	5646(3)	4278(2)	3343(2)	35(1)
C(10)	3630(3)	4417(2)	4031(2)	39(1)
C(11)	3680(3)	4482(2)	2041(2)	32(1)
C(12)	3926(2)	3861(2)	1166(2)	26(1)
C(13)	2844(2)	3761(2)	249(2)	28(1)
C(14)	1554(2)	4464(2)	-1651(2)	28(1)
C(15)	2032(2)	4115(2)	-2503(2)	36(1)
C(16)	1251(3)	3947(2)	-3451(2)	41(1)
C(17)	2(3)	4135(2)	-3557(2)	38(1)
C(18)	-452(2)	4480(2)	-2690(2)	37(1)
C(19)	306(2)	4646(2)	-1729(2)	33(1)
C(20)	-849(3)	3970(2)	-4592(3)	56(1)

Table S5. Anisotropic Displacement Parameters ( $\text{\AA}^2 \times 10^3$ ) for **3ga**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+ \dots ]$ .

Atom	U11	U22	U33	U23	U13	U12
------	-----	-----	-----	-----	-----	-----



<b>Atom</b>	<b>U11</b>	<b>U22</b>	<b>U33</b>	<b>U23</b>	<b>U13</b>	<b>U12</b>
S(1)	29(1)	25(1)	24(1)	-3(1)	2(1)	-1(1)
S(2)	30(1)	32(1)	26(1)	5(1)	-1(1)	-2(1)
O(1)	34(1)	81(2)	176(4)	9(2)	-1(2)	-9(1)
O(2)	55(2)	88(2)	110(3)	-33(2)	-32(2)	22(2)
O(3)	41(1)	43(1)	22(1)	-7(1)	3(1)	4(1)
O(4)	38(1)	24(1)	49(1)	-1(1)	7(1)	-4(1)
O(5)	30(1)	52(1)	44(1)	18(1)	-2(1)	-12(1)
O(6)	52(1)	37(1)	32(1)	-4(1)	-1(1)	9(1)
N(1)	36(2)	48(2)	110(3)	-14(2)	-10(2)	8(1)
N(2)	27(1)	25(1)	21(1)	0(1)	4(1)	1(1)
C(1)	31(1)	25(1)	28(1)	-2(1)	3(1)	6(1)
C(2)	33(1)	32(1)	47(2)	2(1)	12(1)	2(1)
C(3)	31(1)	32(2)	75(2)	-2(2)	10(1)	1(1)
C(4)	29(1)	38(2)	72(2)	-13(2)	-2(1)	6(1)
C(5)	45(2)	48(2)	43(2)	-8(1)	-12(1)	18(1)
C(6)	39(1)	37(1)	32(1)	1(1)	2(1)	4(1)
C(7)	31(1)	31(1)	19(1)	1(1)	3(1)	2(1)
C(8)	36(1)	27(1)	22(1)	-2(1)	1(1)	6(1)
C(9)	42(1)	32(1)	30(1)	-4(1)	0(1)	-3(1)
C(10)	49(2)	41(2)	27(1)	-4(1)	6(1)	15(1)
C(11)	41(1)	28(1)	25(1)	-1(1)	-1(1)	6(1)
C(12)	27(1)	28(1)	22(1)	3(1)	2(1)	-1(1)
C(13)	30(1)	29(1)	25(1)	2(1)	0(1)	-2(1)
C(14)	25(1)	32(1)	24(1)	6(1)	1(1)	1(1)
C(15)	29(1)	45(2)	34(1)	1(1)	4(1)	7(1)
C(16)	44(2)	48(2)	31(1)	-5(1)	6(1)	6(1)
C(17)	41(2)	36(1)	33(1)	1(1)	-4(1)	-3(1)
C(18)	27(1)	42(2)	39(2)	1(1)	-4(1)	3(1)

Atom	U11	U22	U33	U23	U13	U12
C(19)	28(1)	38(1)	33(1)	2(1)	5(1)	4(1)
C(20)	56(2)	63(2)	41(2)	-5(2)	-14(2)	-4(2)

Table S6. Bond lengths for **3ga**.

Atom	Length/Å	Atom	Length/Å
S(1)-O(4)	1.4332(19)	C(3)-C(4)	1.366(5)
S(1)-O(3)	1.4355(19)	C(4)-C(5)	1.386(5)
S(1)-N(2)	1.613(2)	C(5)-C(6)	1.398(4)
S(1)-C(1)	1.775(3)	C(7)-C(8)	1.541(3)
S(2)-O(5)	1.4343(19)	C(8)-C(10)	1.531(3)
S(2)-O(6)	1.448(2)	C(8)-C(11)	1.539(3)
S(2)-C(14)	1.765(3)	C(8)-C(9)	1.542(4)
S(2)-C(13)	1.788(3)	C(11)-C(12)	1.533(3)
O(1)-N(1)	1.230(5)	C(12)-C(13)	1.540(3)
O(2)-N(1)	1.231(5)	C(14)-C(15)	1.386(4)
N(1)-C(4)	1.486(4)	C(14)-C(19)	1.394(3)
N(2)-C(7)	1.488(3)	C(15)-C(16)	1.393(4)
N(2)-C(12)	1.500(3)	C(16)-C(17)	1.394(4)
C(1)-C(2)	1.391(4)	C(17)-C(18)	1.386(4)
C(1)-C(6)	1.392(4)	C(17)-C(20)	1.515(4)
C(2)-C(3)	1.381(4)	C(18)-C(19)	1.393(4)

Table S7. Bond angles for **3ga**.

Atom	Angle/°	Atom	Angle/°
O(4)-S(1)-O(3)	120.70(12)	C(5)-C(4)-N(1)	119.1(3)
O(4)-S(1)-N(2)	107.77(11)	C(4)-C(5)-C(6)	118.3(3)
O(3)-S(1)-N(2)	106.54(11)	C(1)-C(6)-C(5)	118.9(3)

Atom	Angle/°	Atom	Angle/°
O(4)-S(1)-C(1)	105.48(12)	N(2)-C(7)-C(8)	104.06(19)
O(3)-S(1)-C(1)	106.26(12)	C(10)-C(8)-C(11)	111.4(2)
N(2)-S(1)-C(1)	109.85(11)	C(10)-C(8)-C(7)	112.1(2)
O(5)-S(2)-O(6)	118.04(13)	C(11)-C(8)-C(7)	101.49(19)
O(5)-S(2)-C(14)	107.82(12)	C(10)-C(8)-C(9)	110.1(2)
O(6)-S(2)-C(14)	108.05(12)	C(11)-C(8)-C(9)	110.8(2)
O(5)-S(2)-C(13)	107.54(12)	C(7)-C(8)-C(9)	110.7(2)
O(6)-S(2)-C(13)	109.24(12)	C(12)-C(11)-C(8)	106.9(2)
C(14)-S(2)-C(13)	105.45(12)	N(2)-C(12)-C(11)	102.87(18)
O(1)-N(1)-O(2)	124.9(3)	N(2)-C(12)-C(13)	108.78(19)
O(1)-N(1)-C(4)	118.0(4)	C(11)-C(12)-C(13)	114.1(2)
O(2)-N(1)-C(4)	117.1(4)	C(12)-C(13)-S(2)	110.94(17)
C(7)-N(2)-C(12)	111.27(18)	C(15)-C(14)-C(19)	121.0(2)
C(7)-N(2)-S(1)	120.63(16)	C(15)-C(14)-S(2)	119.33(19)
C(12)-N(2)-S(1)	121.47(16)	C(19)-C(14)-S(2)	119.5(2)
C(2)-C(1)-C(6)	121.3(2)	C(14)-C(15)-C(16)	119.3(2)
C(2)-C(1)-S(1)	118.7(2)	C(15)-C(16)-C(17)	120.9(3)
C(6)-C(1)-S(1)	120.0(2)	C(18)-C(17)-C(16)	118.6(3)
C(3)-C(2)-C(1)	119.6(3)	C(18)-C(17)-C(20)	120.2(3)
C(4)-C(3)-C(2)	118.8(3)	C(16)-C(17)-C(20)	121.2(3)
C(3)-C(4)-C(5)	123.2(3)	C(17)-C(18)-C(19)	121.7(2)
C(3)-C(4)-N(1)	117.8(3)	C(18)-C(19)-C(14)	118.5(3)

Table S8. Hydrogen Atom Coordinates ( $\text{\AA}\times 10^4$ ) and Isotropic Displacement Parameters ( $\text{\AA}^2\times 10^3$ )

for **3ga**.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U <sub>eq</sub>
H(2)	6883	3238	687	44
H(3)	8960	3461	1357	55

<b>Atom</b>	<b><i>x</i></b>	<b><i>y</i></b>	<b><i>z</i></b>	<b>Ueq</b>
H(5)	8634	1994	3926	57
H(6)	6536	1757	3254	43
H(7A)	4646	2784	3353	33
H(7B)	3197	2969	3043	33
H(9A)	5776	4892	3435	53
H(9B)	6035	4078	2743	53
H(9C)	6008	3982	3996	53
H(10A)	3702	5039	4075	58
H(10B)	4025	4165	4704	58
H(10C)	2760	4258	3904	58
H(11A)	2786	4561	2022	38
H(11B)	4054	5043	1939	38
H(12)	4686	4031	878	31
H(13A)	2100	3591	544	34
H(13B)	3034	3306	-241	34
H(15)	2883	3991	-2440	43
H(16)	1573	3701	-4034	49
H(18)	-1303	4607	-2754	45
H(19)	-20	4878	-1140	39
H(20A)	-1086	3368	-4631	83
H(20B)	-426	4110	-5196	83
H(20C)	-1584	4324	-4621	83

The crystals of **3qa** was obtained by crystallization from a solution in acetone/petroleum ether after purification by column chromatography.

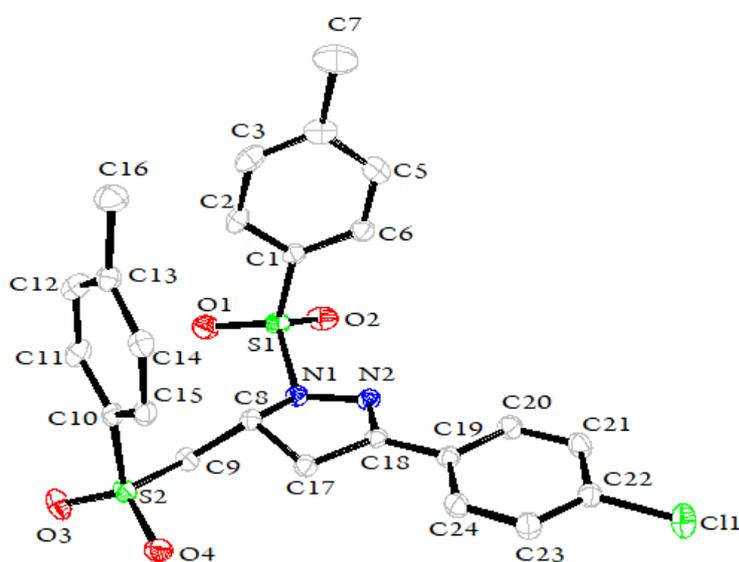
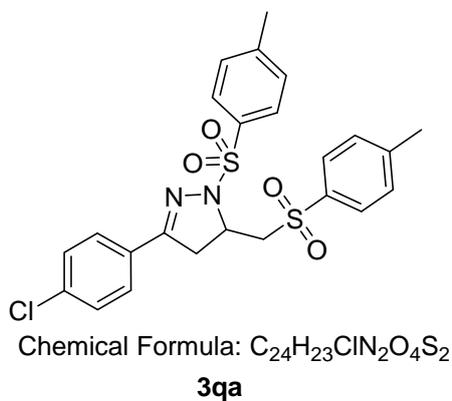


Figure S2. ORTEP drawing of compound **3qa**.

Table S9. Crystal data and structure refinement for **3qa**.

Identification code	3qa
Empirical formula	$C_{24}H_{23}ClN_2O_4S_2$
Formula weight	503.01
Temperature/K	113
Crystal system	Triclinic
Space group	P-1
a/Å	10.622(2)
b/Å	10.722(2)

$c/\text{\AA}$	11.698(2)
$\alpha/^\circ$	78.63(3)
$\beta/^\circ$	98.83(3)
$\gamma/^\circ$	67.46(3)
Volume/ $\text{\AA}^3$	1196.0(5)
Z	2
$\rho_{\text{calc}}/\text{g cm}^3$	1.397
$\mu/\text{mm}^{-1}$	0.368
F (000)	524
Crystal size/ $\text{mm}^3$	0.200×0.180×0.120
Radiation	MoK $\alpha$ (0.71073)
2 $\Theta$ range for data collection/ $^\circ$	2.076 to 27.853
Index ranges	-13 $\leq$ h $\leq$ 13, -14 $\leq$ k $\leq$ 14, -15 $\leq$ l $\leq$ 15
Reflections collected	14274
Independent reflections	5654 [ $R_{\text{(int)}} = 0.0405$ ]
Data/restraints/parameters	5654 / 0 / 300
Goodness-of-fit on $F^2$	1.040
Final R indexes [ $I \geq 2\sigma(I)$ ]	R1 = 0.0424, wR2 = 0.1104
Final R indexes [all data]	R1 = 0.0570, wR2 = 0.1190
Largest diff. peak/hole / $e \text{\AA}^{-3}$	0.269/ -0.361
Flack parameter	n/a

Table S10. Fractional Atomic Coordinates ( $\times 10^4$ ) and Equivalent Isotropic Displacement Parameters ( $\text{\AA}^2 \times 10^3$ ) for **3qa**. Ueq is defined as 1/3 of the trace of the orthogonalized.

Atom	$x$	$y$	$z$	Ueq
S(1)	9159(1)	-491(1)	2383(1)	33(1)
S(2)	12000(1)	2480(1)	1232(1)	32(1)
Cl(1)	3474(1)	5778(1)	-2764(1)	46(1)

<b>Atom</b>	<b>x</b>	<b>y</b>	<b>z</b>	<b>Ueq</b>
O(1)	10424(1)	-880(1)	2861(1)	44(1)
O(2)	8786(2)	-1495(1)	2035(1)	45(1)
O(3)	13407(1)	1779(1)	1451(1)	46(1)
O(4)	11698(1)	3536(1)	239(1)	38(1)
N(1)	9294(2)	600(1)	1187(1)	29(1)
N(2)	8085(2)	1147(1)	634(1)	30(1)
C(1)	7811(2)	392(2)	3385(1)	30(1)
C(2)	8108(2)	802(2)	4333(2)	36(1)
C(3)	7043(2)	1436(2)	5153(2)	43(1)
C(4)	5690(2)	1671(2)	5049(2)	45(1)
C(5)	5415(2)	1301(2)	4078(2)	43(1)
C(6)	6456(2)	666(2)	3250(2)	36(1)
C(7)	4536(3)	2332(3)	5971(2)	72(1)
C(8)	9776(2)	1724(2)	1276(2)	28(1)
C(9)	11344(2)	1223(2)	1109(2)	32(1)
C(10)	11006(2)	3177(2)	2502(2)	29(1)
C(11)	11161(2)	2373(2)	3590(2)	38(1)
C(12)	10265(2)	2839(2)	4564(2)	42(1)
C(13)	9212(2)	4109(2)	4484(2)	37(1)
C(14)	9115(2)	4920(2)	3402(2)	35(1)
C(15)	9982(2)	4458(2)	2401(2)	32(1)
C(16)	8185(2)	4546(2)	5547(2)	52(1)
C(17)	9077(2)	2826(2)	308(2)	30(1)
C(18)	7958(2)	2367(2)	139(1)	28(1)
C(19)	6852(2)	3189(2)	-572(1)	28(1)
C(20)	5817(2)	2725(2)	-665(2)	36(1)
C(21)	4779(2)	3525(2)	-1334(2)	38(1)
C(22)	4774(2)	4786(2)	-1910(2)	33(1)

Atom	<i>x</i>	<i>y</i>	<i>z</i>	Ueq
C(23)	5767(2)	5273(2)	-1813(2)	37(1)
C(24)	6803(2)	4462(2)	-1138(2)	36(1)

Table S11. Anisotropic Displacement Parameters ( $\text{\AA}^2 \times 10^3$ ) for **3qa**. The anisotropic displacement

factor exponent takes the form:  $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$

Atom	U11	U22	U33	U23	U13	U12
S(1)	38(1)	26(1)	31(1)	-1(1)	0(1)	-11(1)
S(2)	30(1)	35(1)	35(1)	-7(1)	-1(1)	-14(1)
Cl(1)	36(1)	43(1)	58(1)	-6(1)	-18(1)	-8(1)
O(1)	36(1)	42(1)	42(1)	8(1)	-5(1)	-7(1)
O(2)	63(1)	30(1)	43(1)	-7(1)	6(1)	-24(1)
O(3)	26(1)	51(1)	62(1)	-13(1)	-4(1)	-13(1)
O(4)	45(1)	40(1)	33(1)	-3(1)	1(1)	-23(1)
N(1)	35(1)	28(1)	27(1)	-3(1)	-3(1)	-15(1)
N(2)	35(1)	31(1)	25(1)	-5(1)	-3(1)	-15(1)
C(1)	36(1)	26(1)	26(1)	1(1)	-4(1)	-13(1)
C(2)	44(1)	31(1)	33(1)	1(1)	-14(1)	-11(1)
C(3)	64(1)	31(1)	32(1)	-6(1)	-10(1)	-10(1)
C(4)	53(1)	31(1)	42(1)	-5(1)	6(1)	-10(1)
C(5)	36(1)	39(1)	50(1)	-4(1)	0(1)	-14(1)
C(6)	41(1)	39(1)	32(1)	-2(1)	-5(1)	-21(1)
C(7)	76(2)	54(2)	67(2)	-26(1)	24(1)	-10(1)
C(8)	32(1)	28(1)	26(1)	-7(1)	-2(1)	-12(1)
C(9)	32(1)	30(1)	35(1)	-9(1)	-1(1)	-11(1)
C(10)	31(1)	30(1)	31(1)	-6(1)	-6(1)	-14(1)
C(11)	49(1)	30(1)	36(1)	-1(1)	-14(1)	-12(1)
C(12)	62(1)	38(1)	30(1)	-2(1)	-9(1)	-23(1)
C(13)	44(1)	47(1)	32(1)	-12(1)	-2(1)	-25(1)



<b>Atom</b>	<b>U11</b>	<b>U22</b>	<b>U33</b>	<b>U23</b>	<b>U13</b>	<b>U12</b>
C(14)	32(1)	34(1)	40(1)	-9(1)	-5(1)	-11(1)
C(15)	34(1)	32(1)	32(1)	0(1)	-9(1)	-15(1)
C(16)	60(1)	64(2)	41(1)	-20(1)	6(1)	-32(1)
C(17)	34(1)	31(1)	29(1)	-2(1)	-4(1)	-15(1)
C(18)	34(1)	31(1)	21(1)	-7(1)	0(1)	-16(1)
C(19)	32(1)	31(1)	23(1)	-8(1)	0(1)	-12(1)
C(20)	41(1)	36(1)	36(1)	-3(1)	-5(1)	-21(1)
C(21)	35(1)	45(1)	43(1)	-8(1)	-4(1)	-23(1)
C(22)	29(1)	36(1)	32(1)	-10(1)	-3(1)	-8(1)
C(23)	39(1)	32(1)	42(1)	1(1)	-9(1)	-15(1)
C(24)	37(1)	36(1)	39(1)	-1(1)	-8(1)	-21(1)

Table S12. Bond lengths for **3qa**.

<b>Atom</b>	<b>Length/Å</b>	<b>Atom</b>	<b>Length/Å</b>
S(1)-O(2)	1.4301(13)	C(5)-C(6)	1.376(3)
S(1)-O(1)	1.4360(14)	C(8)-C(9)	1.527(2)
S(1)-N(1)	1.6582(15)	C(8)-C(17)	1.534(2)
S(1)-C(1)	1.7518(19)	C(10)-C(15)	1.387(2)
S(2)-O(3)	1.4401(14)	C(10)-C(11)	1.389(2)
S(2)-O(4)	1.4429(14)	C(11)-C(12)	1.375(3)
S(2)-C(10)	1.7628(19)	C(12)-C(13)	1.391(3)
S(2)-C(9)	1.7763(18)	C(13)-C(14)	1.384(3)
Cl(1)-C(22)	1.7393(19)	C(13)-C(16)	1.507(3)
N(1)-N(2)	1.4157(19)	C(14)-C(15)	1.388(2)
N(1)-C(8)	1.505(2)	C(17)-C(18)	1.507(2)
N(2)-C(18)	1.288(2)	C(18)-C(19)	1.465(2)
C(1)-C(6)	1.388(2)	C(19)-C(24)	1.381(2)
C(1)-C(2)	1.399(2)	C(19)-C(20)	1.398(2)

Atom	Length/Å	Atom	Length/Å
C(2)-C(3)	1.382(3)	C(20)-C(21)	1.382(3)
C(3)-C(4)	1.386(3)	C(21)-C(22)	1.385(3)
C(4)-C(5)	1.389(3)	C(22)-C(23)	1.375(2)
C(4)-C(7)	1.514(3)	C(23)-C(24)	1.387(3)

Table S13. Bond angles for **3qa**.

Atom	Angle/°	Atom	Angle/°
O(2)-S(1)-O(1)	120.30(9)	C(5)-C(6)-C(1)	119.41(17)
O(2)-S(1)-N(1)	106.79(8)	N(1)-C(8)-C(9)	110.57(13)
O(1)-S(1)-N(1)	104.66(8)	N(1)-C(8)-C(17)	101.09(12)
O(2)-S(1)-C(1)	108.24(9)	C(9)-C(8)-C(17)	115.40(15)
O(1)-S(1)-C(1)	108.27(9)	C(8)-C(9)-S(2)	113.47(12)
N(1)-S(1)-C(1)	108.00(8)	C(15)-C(10)-C(11)	120.39(17)
O(3)-S(2)-O(4)	119.07(9)	C(15)-C(10)-S(2)	120.14(14)
O(3)-S(2)-C(10)	109.14(9)	C(11)-C(10)-S(2)	119.13(14)
O(4)-S(2)-C(10)	107.97(8)	C(12)-C(11)-C(10)	119.54(17)
O(3)-S(2)-C(9)	107.26(9)	C(11)-C(12)-C(13)	121.29(18)
O(4)-S(2)-C(9)	108.89(8)	C(14)-C(13)-C(12)	118.23(17)
C(10)-S(2)-C(9)	103.39(8)	C(14)-C(13)-C(16)	121.75(19)
N(2)-N(1)-C(8)	109.94(12)	C(12)-C(13)-C(16)	119.98(19)
N(2)-N(1)-S(1)	112.06(11)	C(13)-C(14)-C(15)	121.48(17)
C(8)-N(1)-S(1)	118.47(10)	C(10)-C(15)-C(14)	118.94(17)
C(18)-N(2)-N(1)	108.16(14)	C(18)-C(17)-C(8)	101.94(14)
C(6)-C(1)-C(2)	120.12(17)	N(2)-C(18)-C(19)	121.91(15)
C(6)-C(1)-S(1)	120.27(13)	N(2)-C(18)-C(17)	114.16(15)
C(2)-C(1)-S(1)	119.61(14)	C(19)-C(18)-C(17)	123.86(15)
C(3)-C(2)-C(1)	119.26(18)	C(24)-C(19)-C(20)	119.05(17)
C(3)-C(2)-H(2)	120.4	C(24)-C(19)-C(18)	119.73(16)

Atom	Angle/°	Atom	Angle/°
C(1)-C(2)-H(2)	120.4	C(20)-C(19)-C(18)	121.20(16)
C(2)-C(3)-C(4)	121.06(18)	C(21)-C(20)-C(19)	120.07(18)
C(2)-C(3)-H(3)	119.5	C(20)-C(21)-C(22)	119.46(17)
C(4)-C(3)-H(3)	119.5	C(23)-C(22)-C(21)	121.46(17)
C(3)-C(4)-C(5)	118.70(18)	C(23)-C(22)-Cl(1)	119.36(15)
C(3)-C(4)-C(7)	120.6(2)	C(21)-C(22)-Cl(1)	119.18(14)
C(5)-C(4)-C(7)	120.7(2)	C(22)-C(23)-C(24)	118.56(18)
C(6)-C(5)-C(4)	121.37(19)	C(19)-C(24)-C(23)	121.38(17)

Table S14. Hydrogen Atom Coordinates ( $\text{\AA} \times 10^4$ ) and Isotropic Displacement Parameters

( $\text{\AA}^2 \times 10^3$ ) for **3qa**.

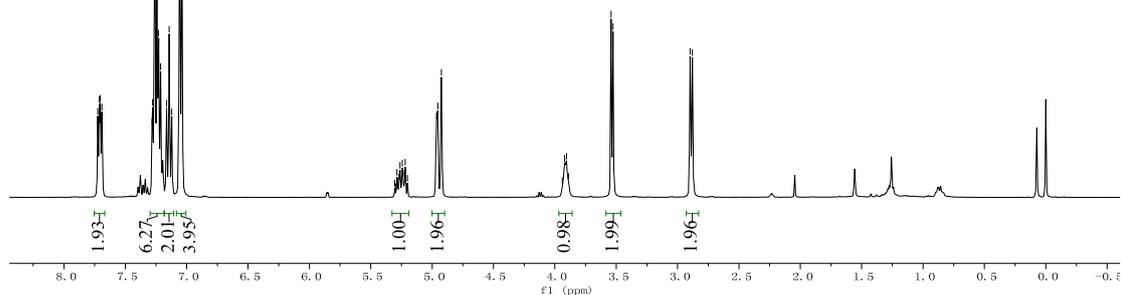
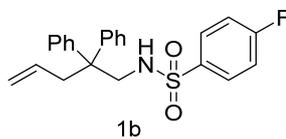
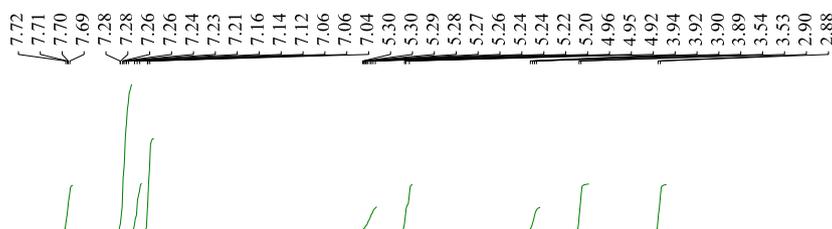
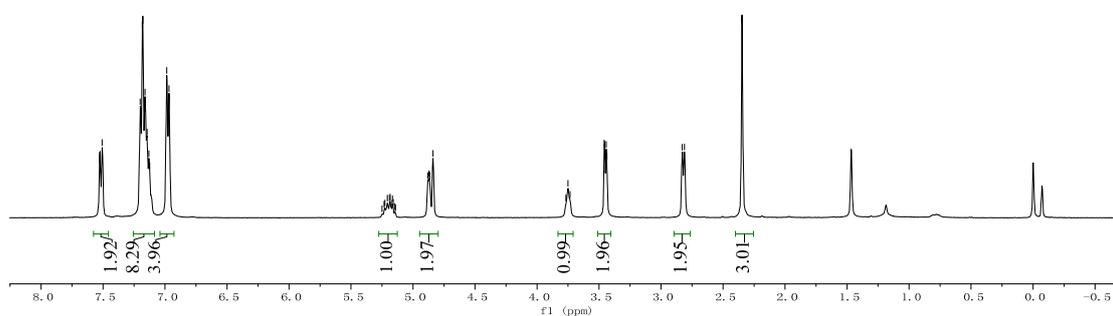
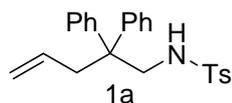
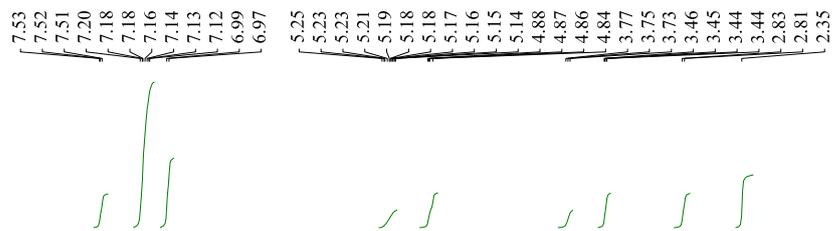
Atom	<i>x</i>	<i>y</i>	<i>z</i>	U <sub>eq</sub>
H(2)	9031	647	4412	44
H(3)	7241	1715	5799	52
H(5)	4488	1490	3983	52
H(6)	6250	418	2592	43
H(7A)	4428	3290	5924	108
H(7B)	4753	1855	6754	108
H(7C)	3678	2279	5832	108
H(8)	9405	2051	2060	34
H(9A)	11695	922	323	38
H(9B)	11694	423	1705	38
H(11)	11881	1508	3662	46
H(12)	10368	2283	5305	50
H(14)	8440	5813	3343	42
H(15)	9876	5010	1658	38
H(16A)	7310	4487	5461	78
H(16B)	8042	5488	5617	78

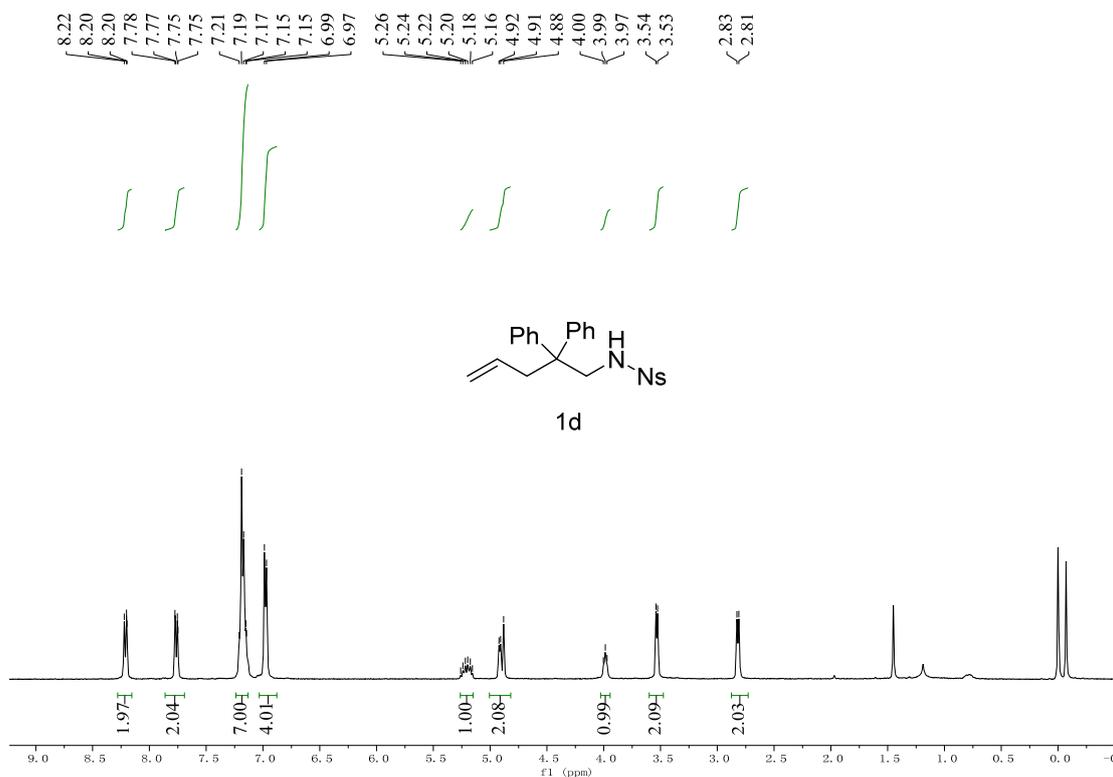
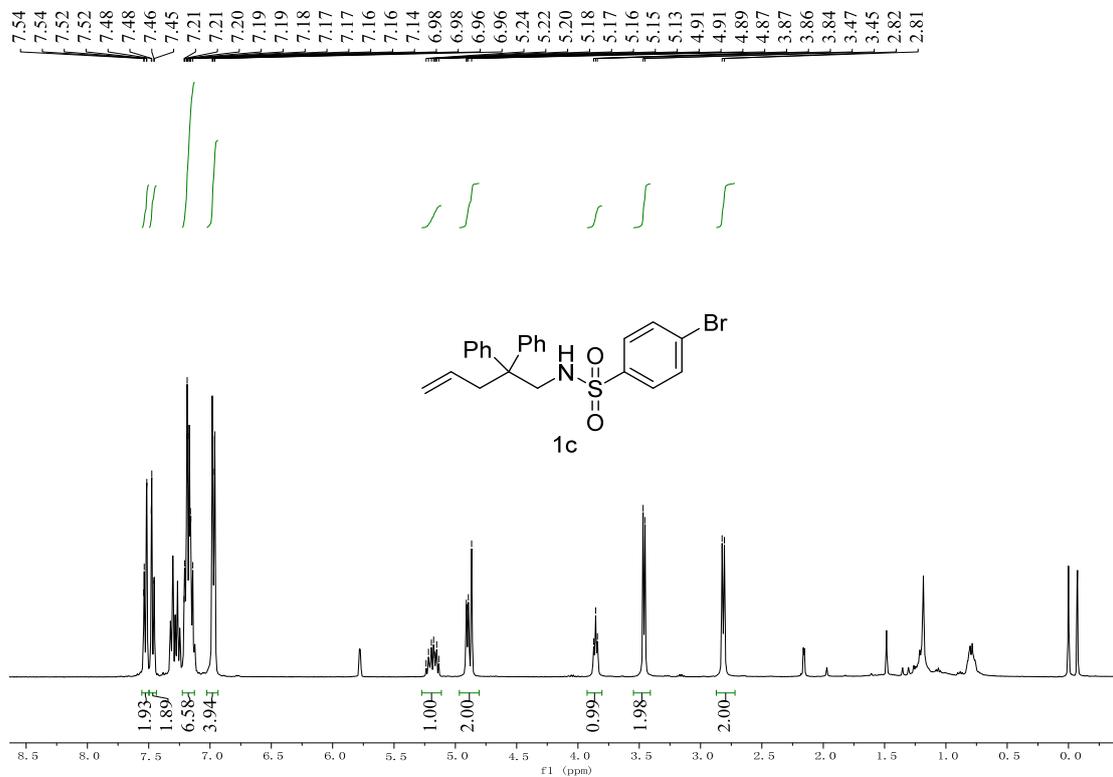
---

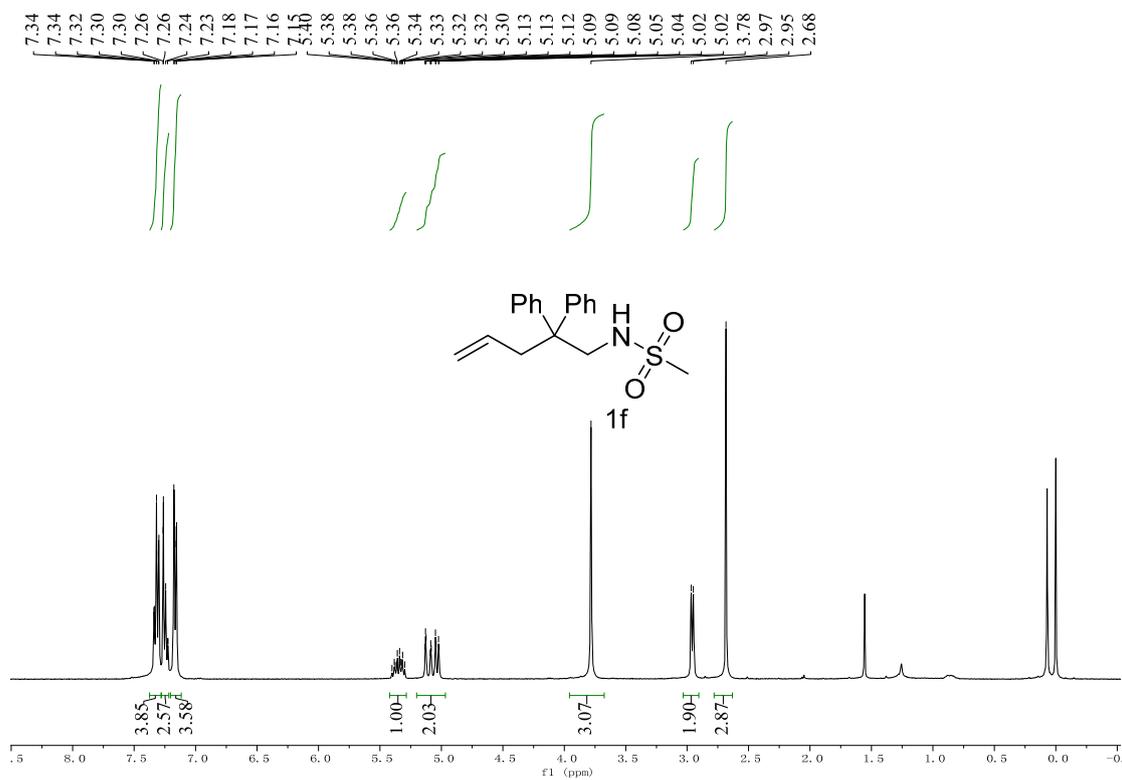
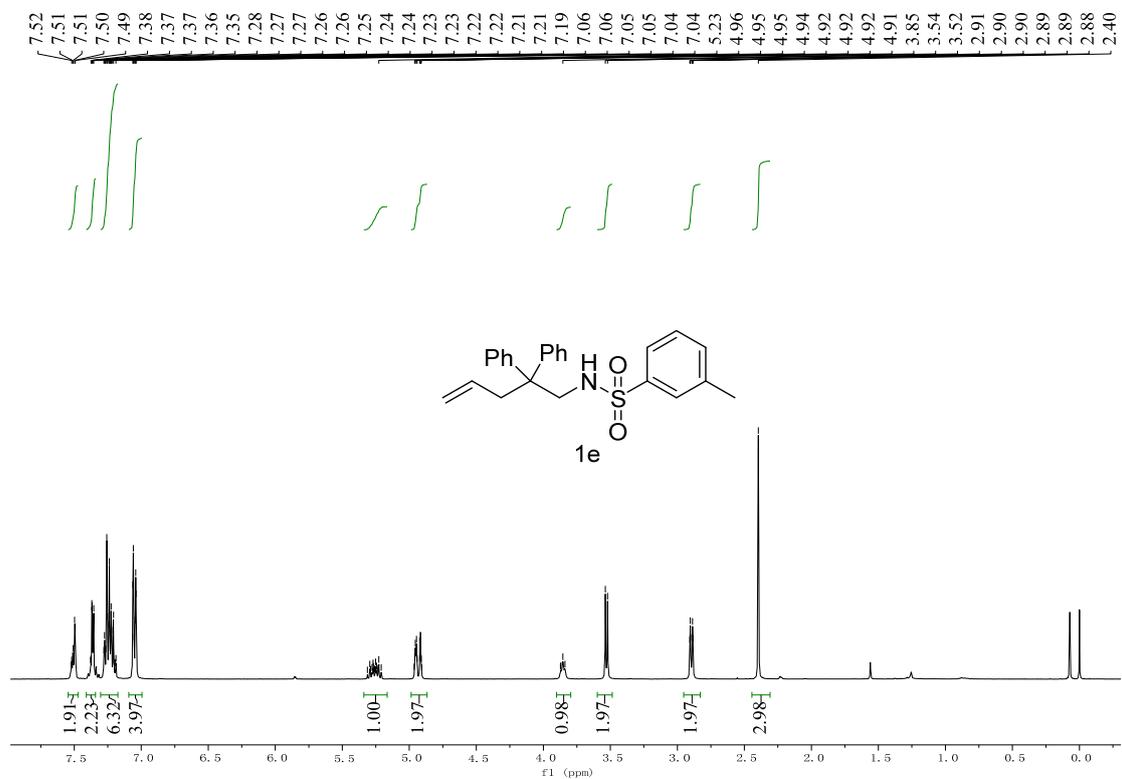
<b>Atom</b>	<b><i>x</i></b>	<b><i>y</i></b>	<b><i>z</i></b>	<b>Ueq</b>
H(16C)	8534	3947	6254	78
H(17A)	8691	3739	566	36
H(17B)	9725	2843	-423	36
H(20)	5828	1859	-268	43
H(21)	4073	3214	-1398	46
H(23)	5743	6147	-2200	45
H(24)	7494	4788	-1064	43

---

#### 4. Copies of $^1\text{H}$ and $^{13}\text{C}$ NMR Spectra.

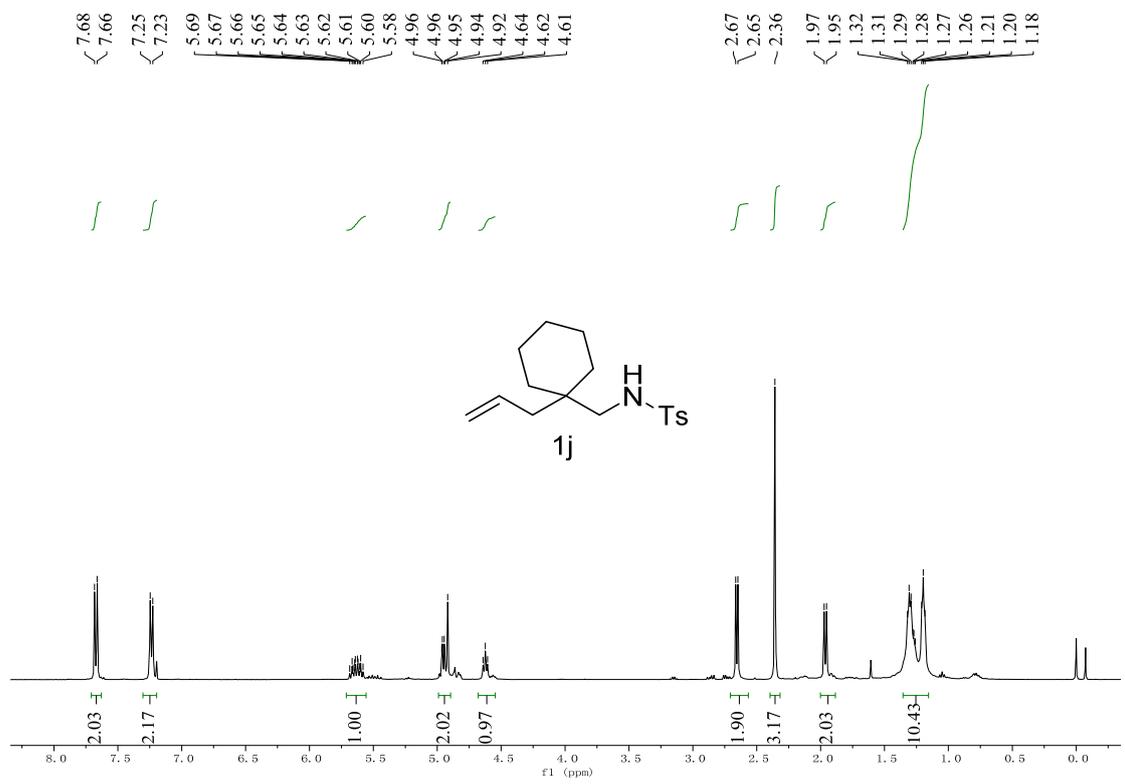
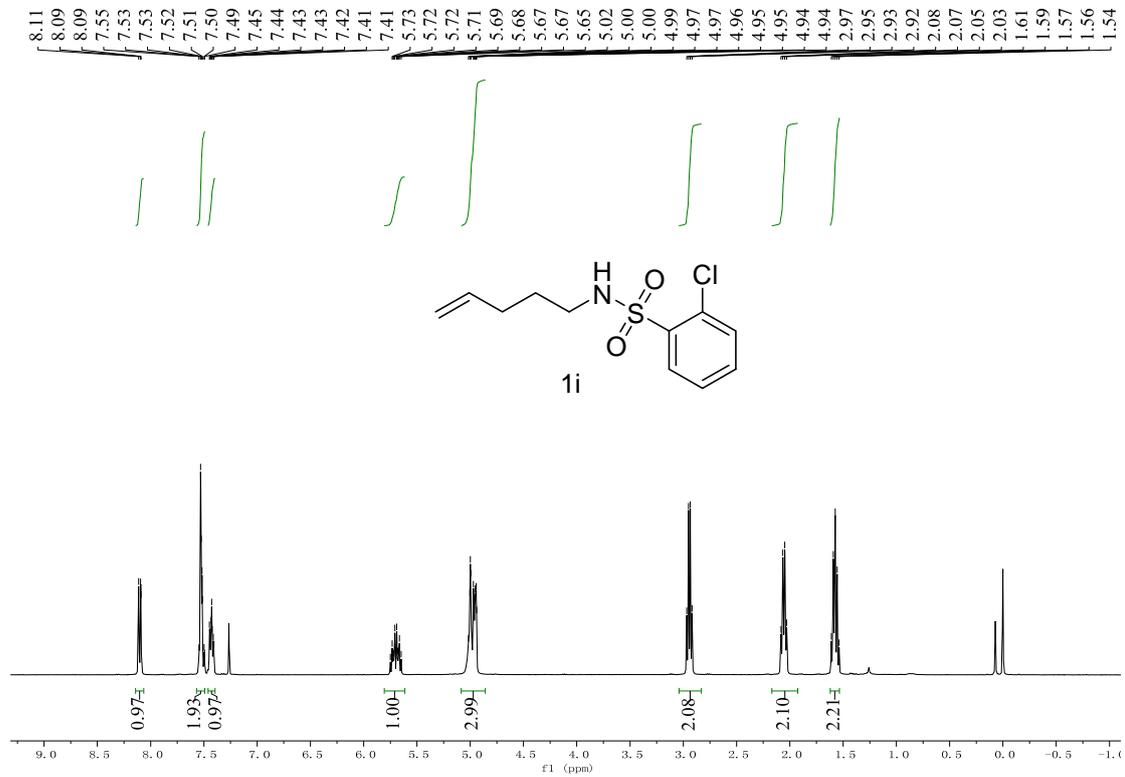


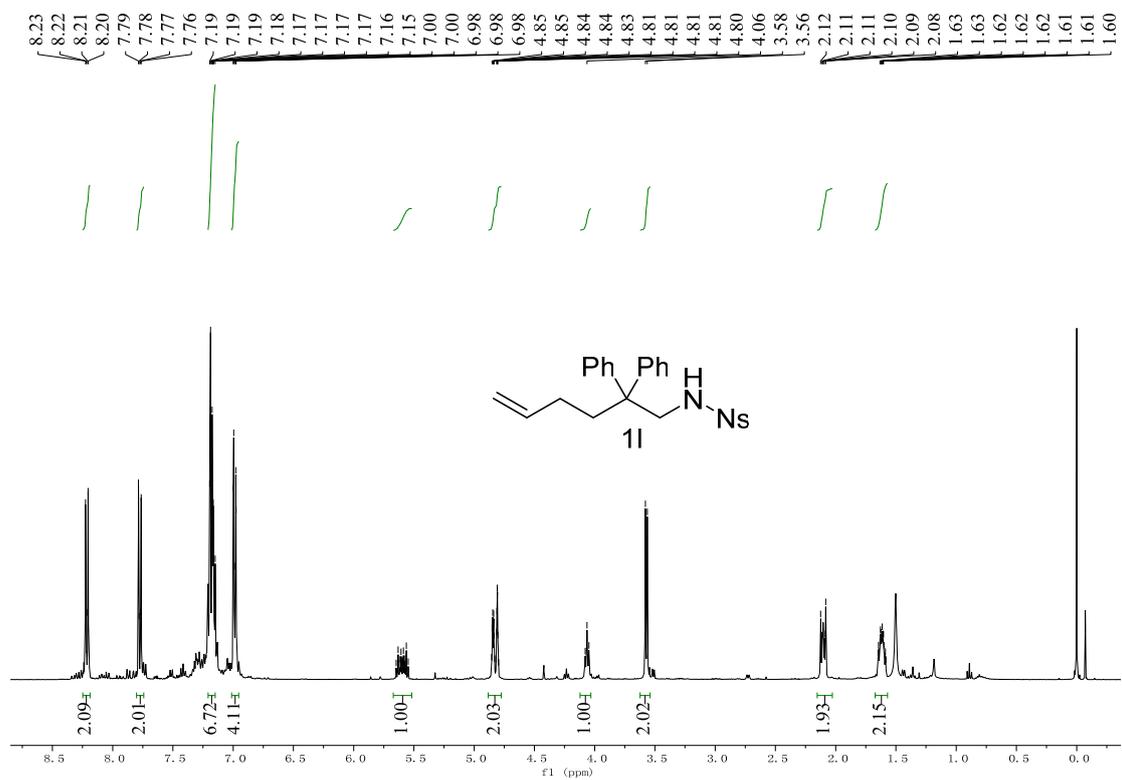
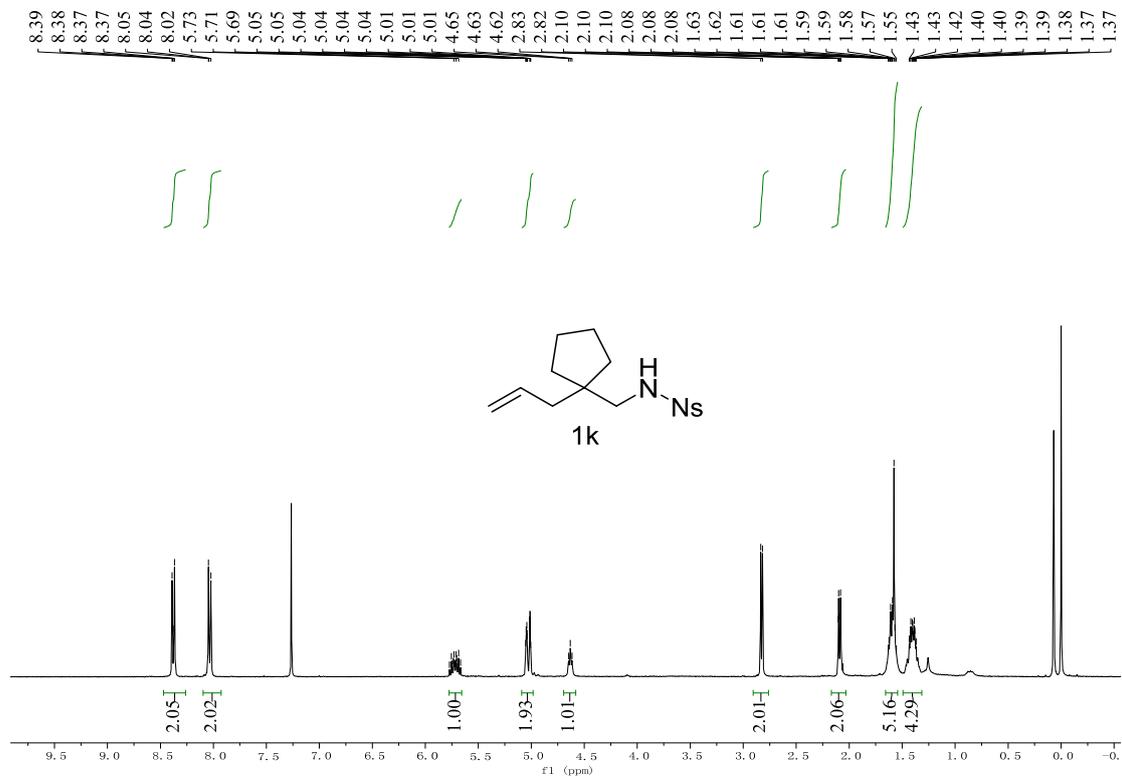


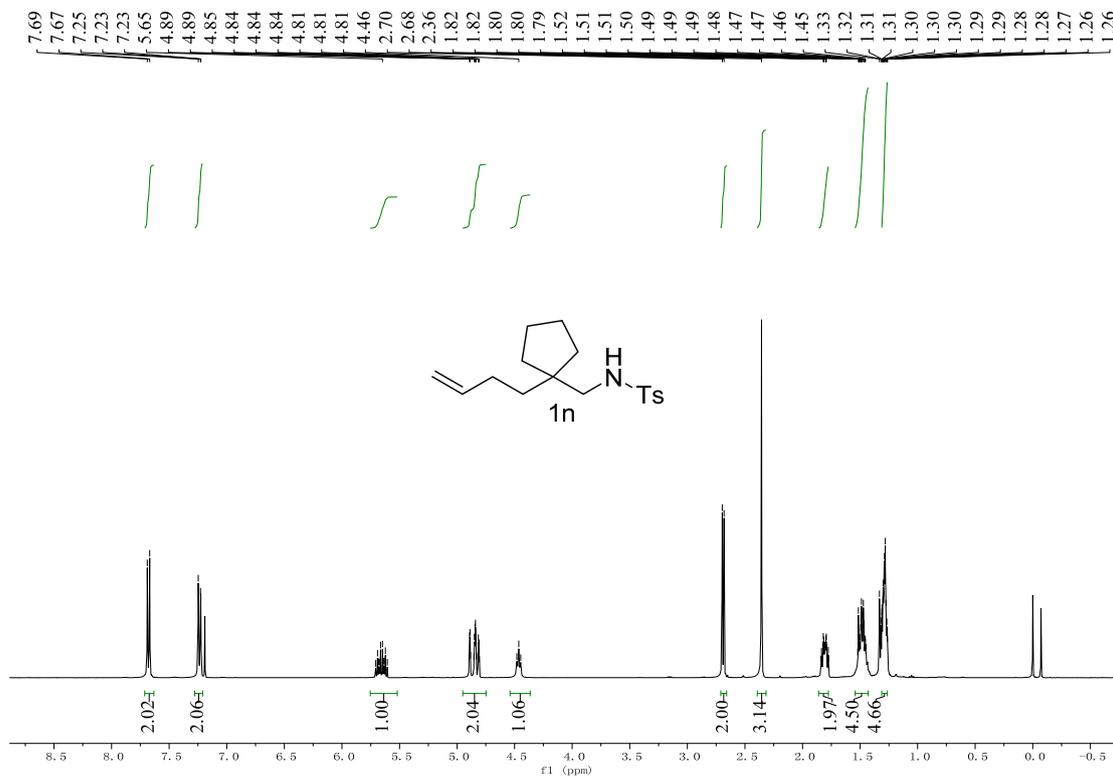
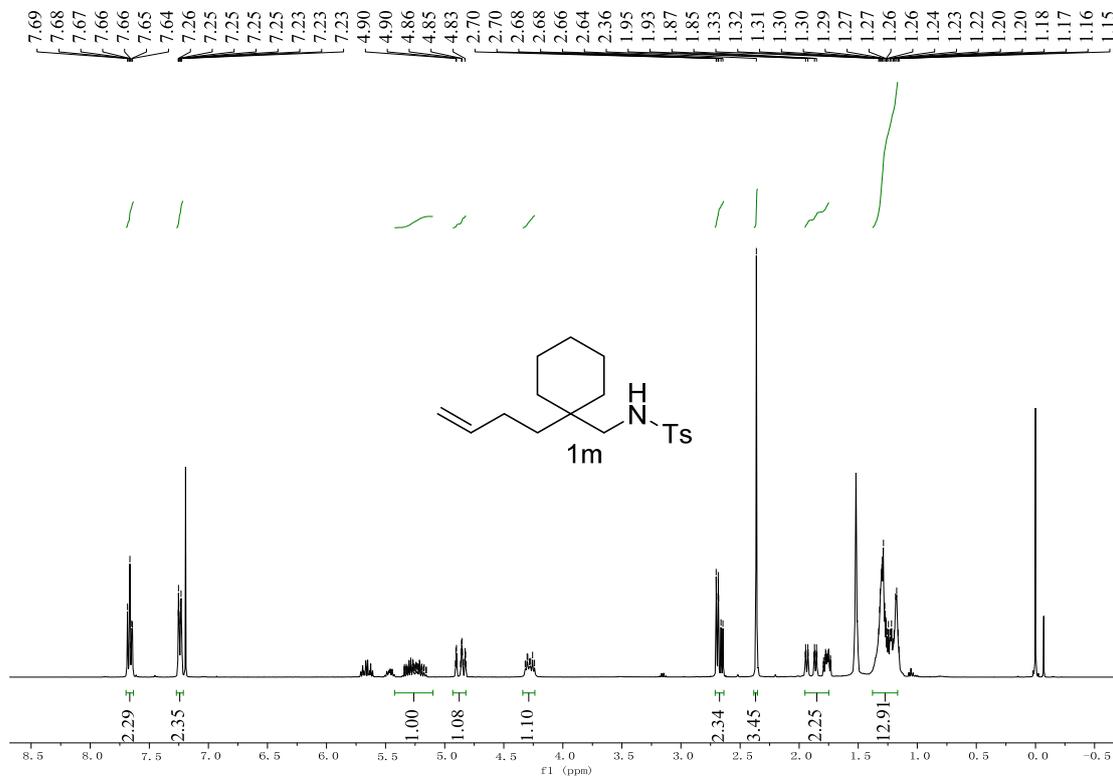


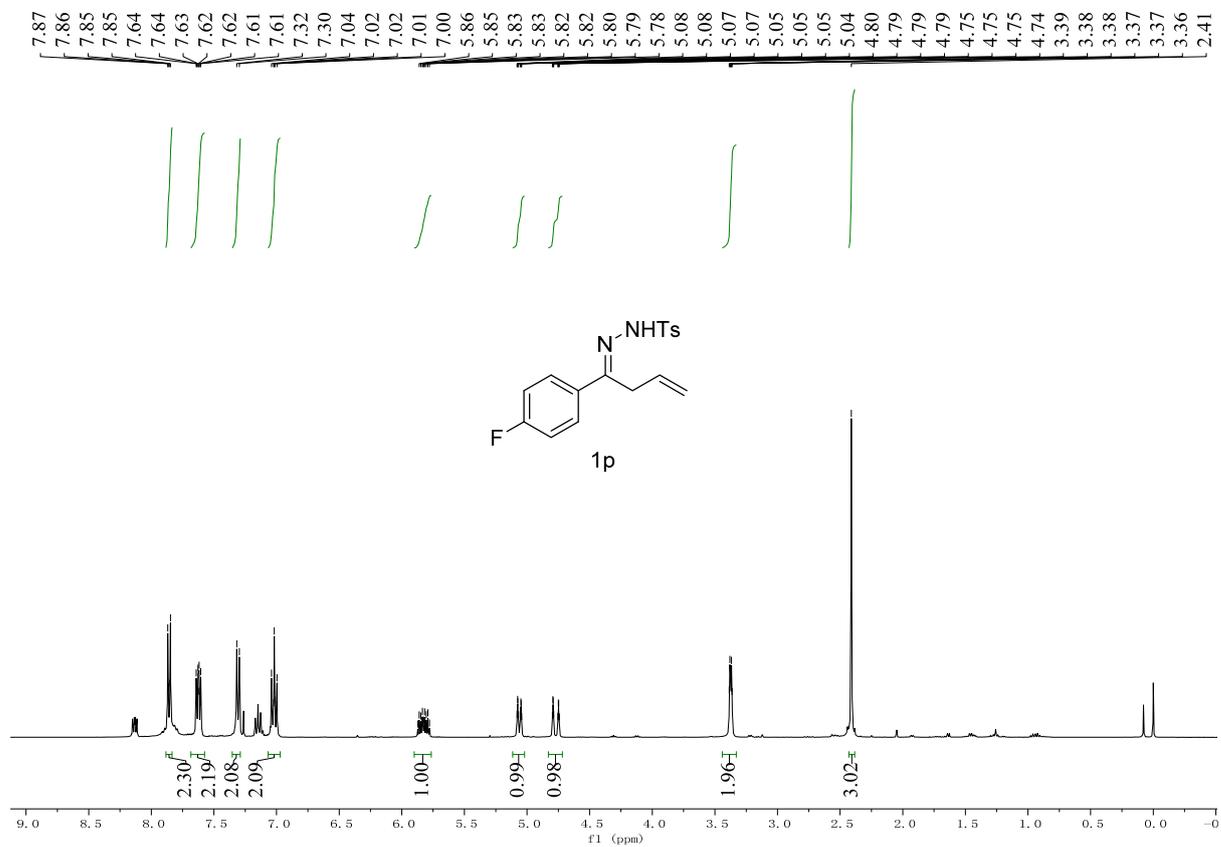
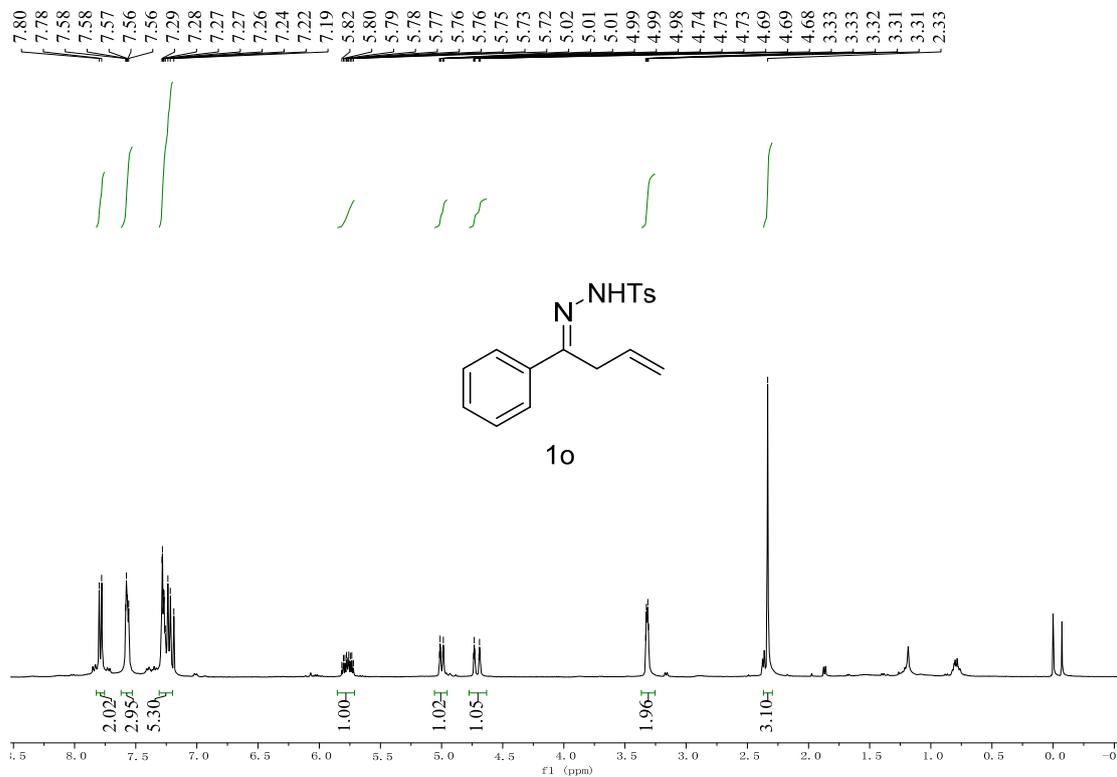


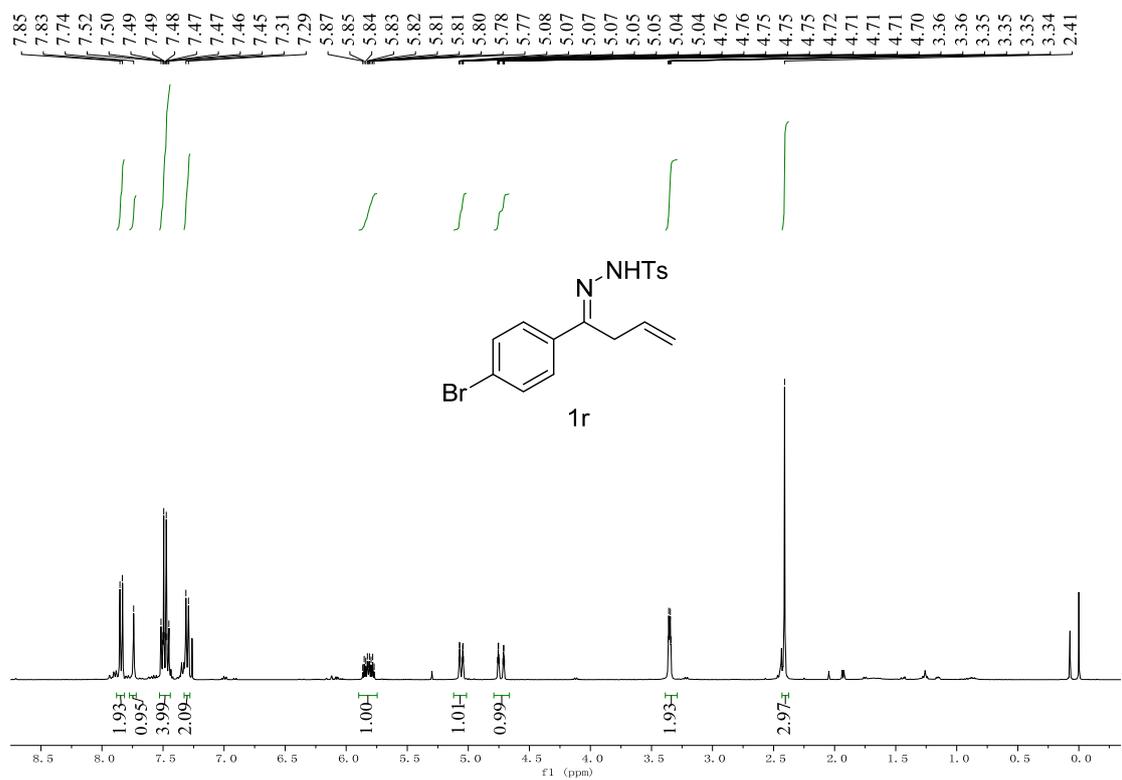
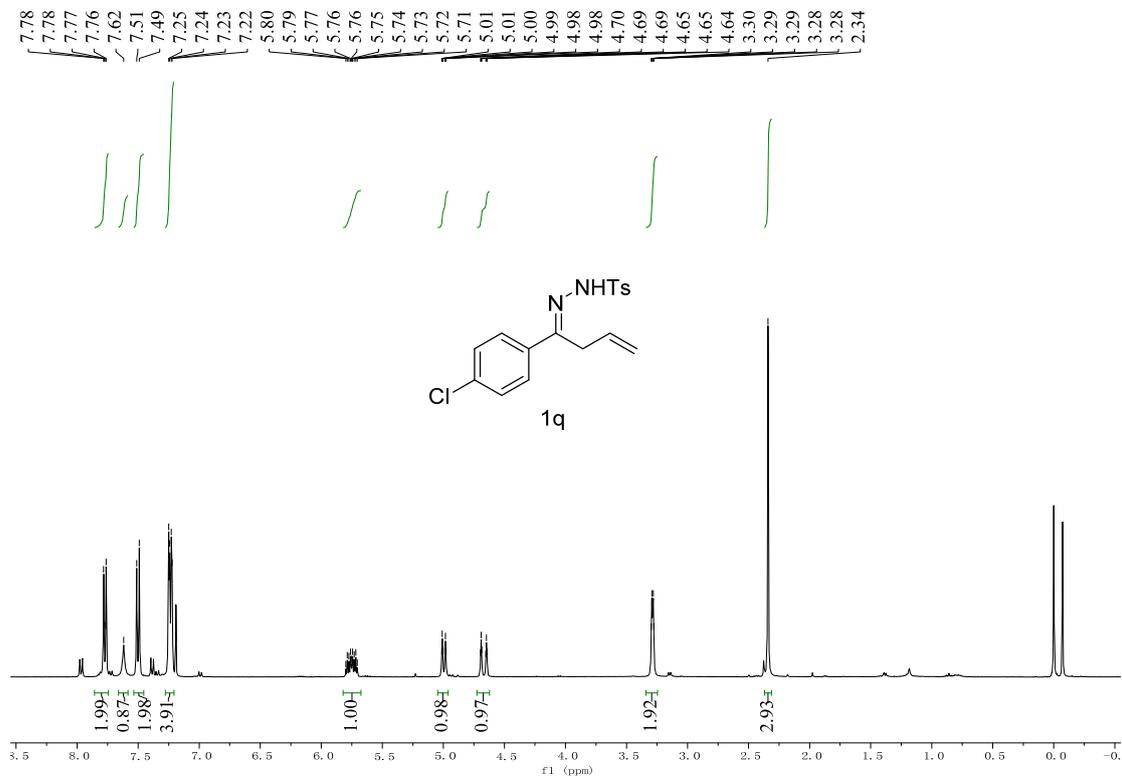


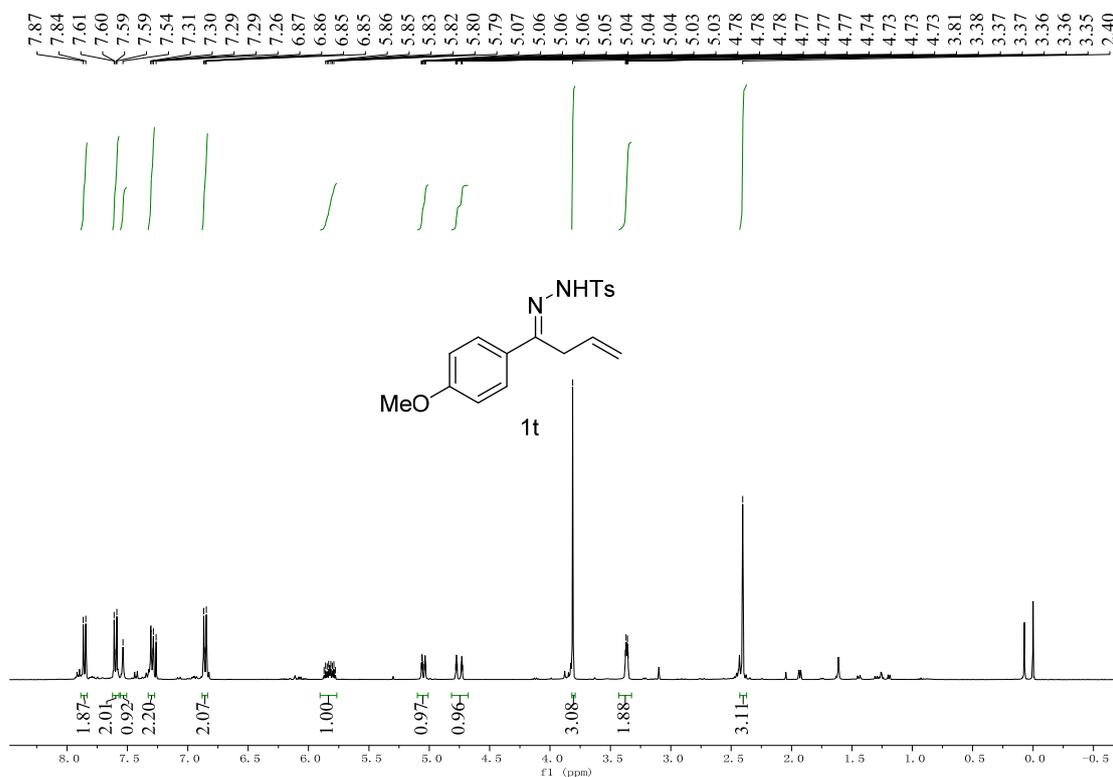
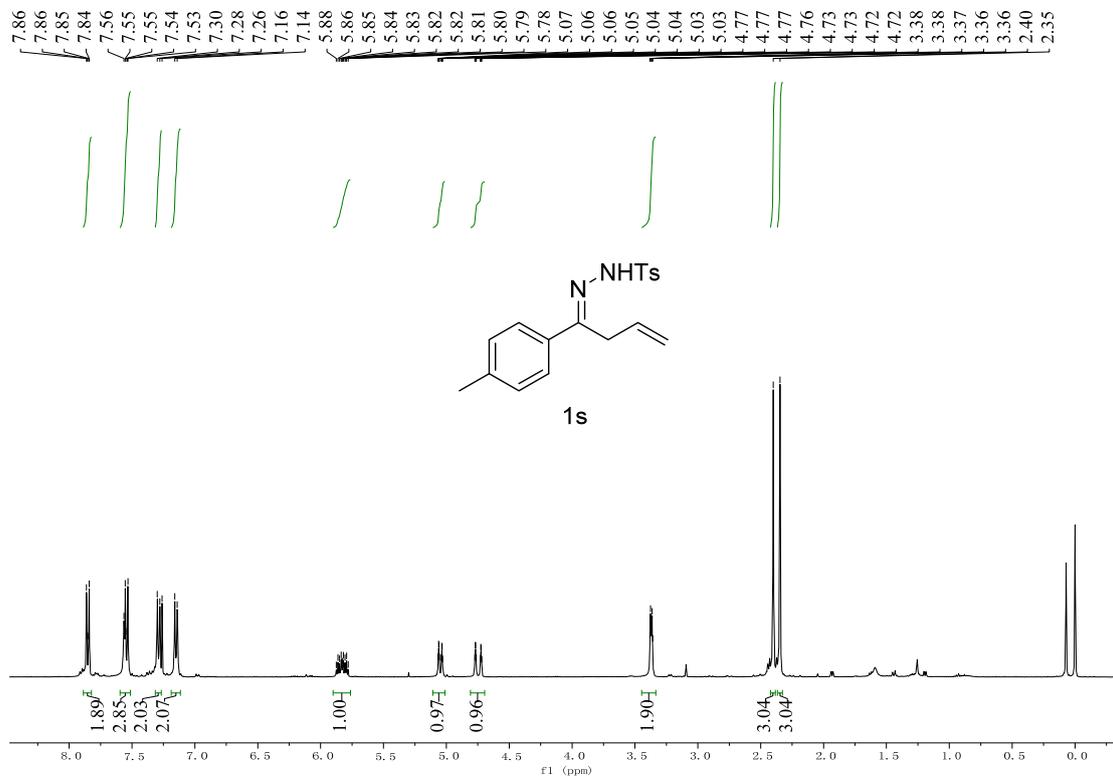


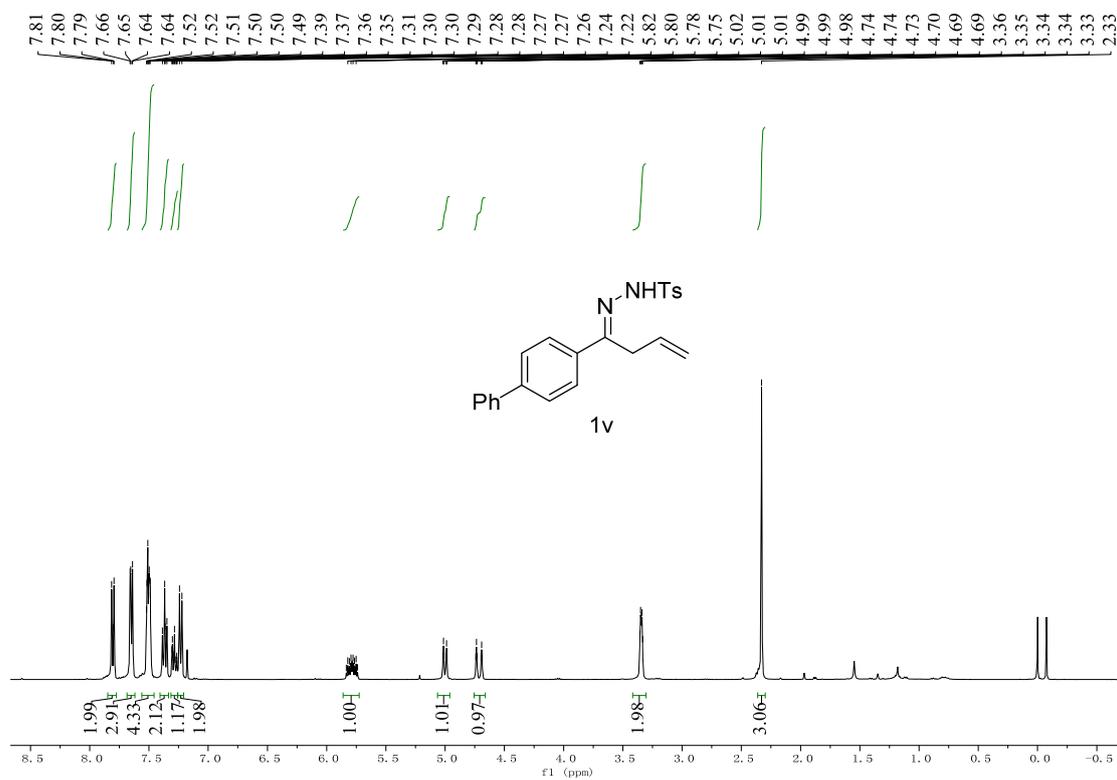
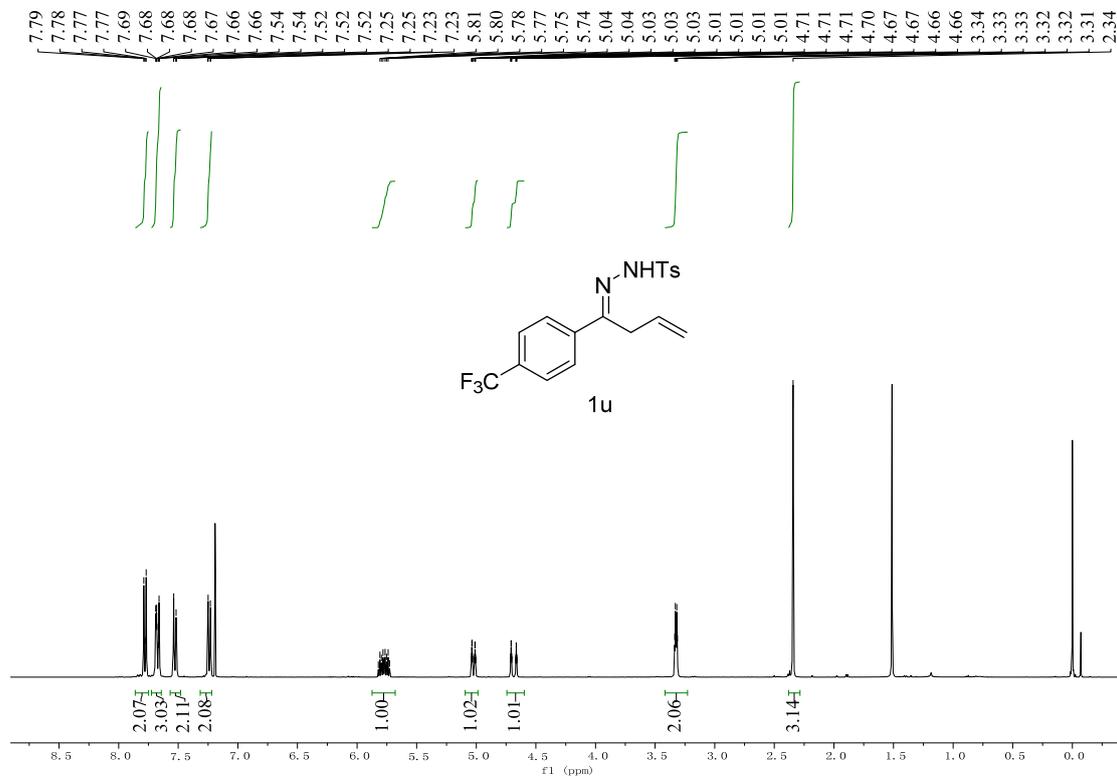


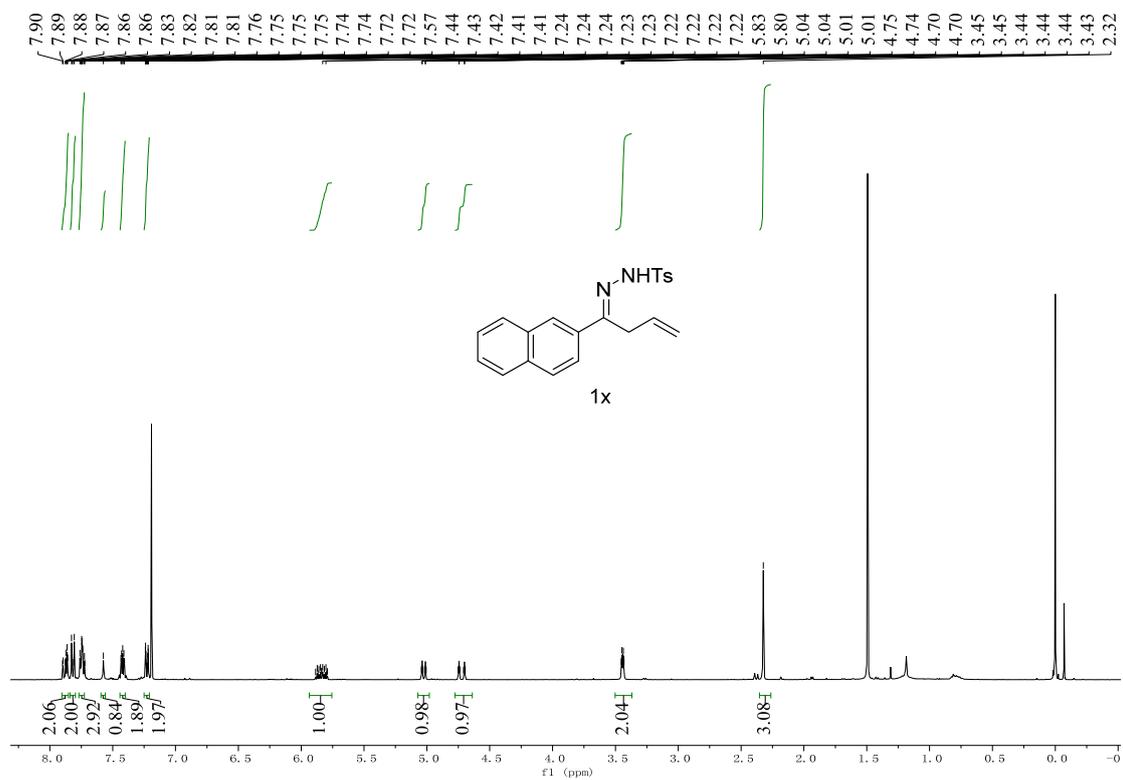
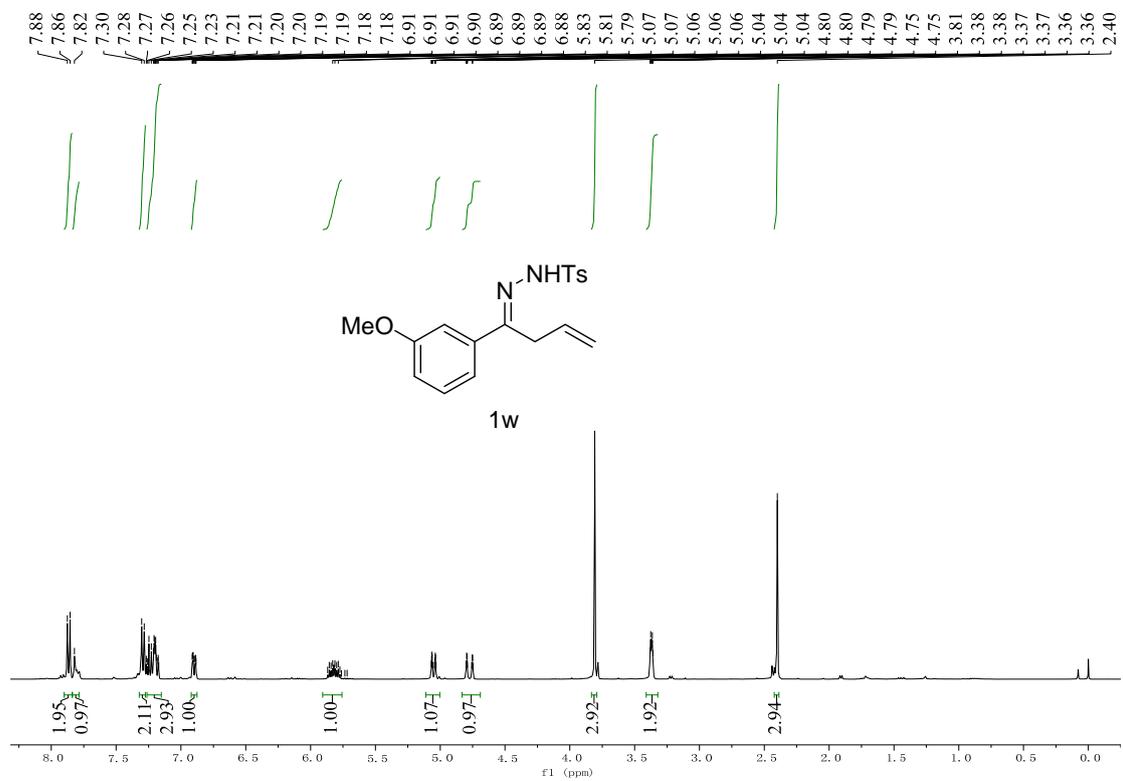




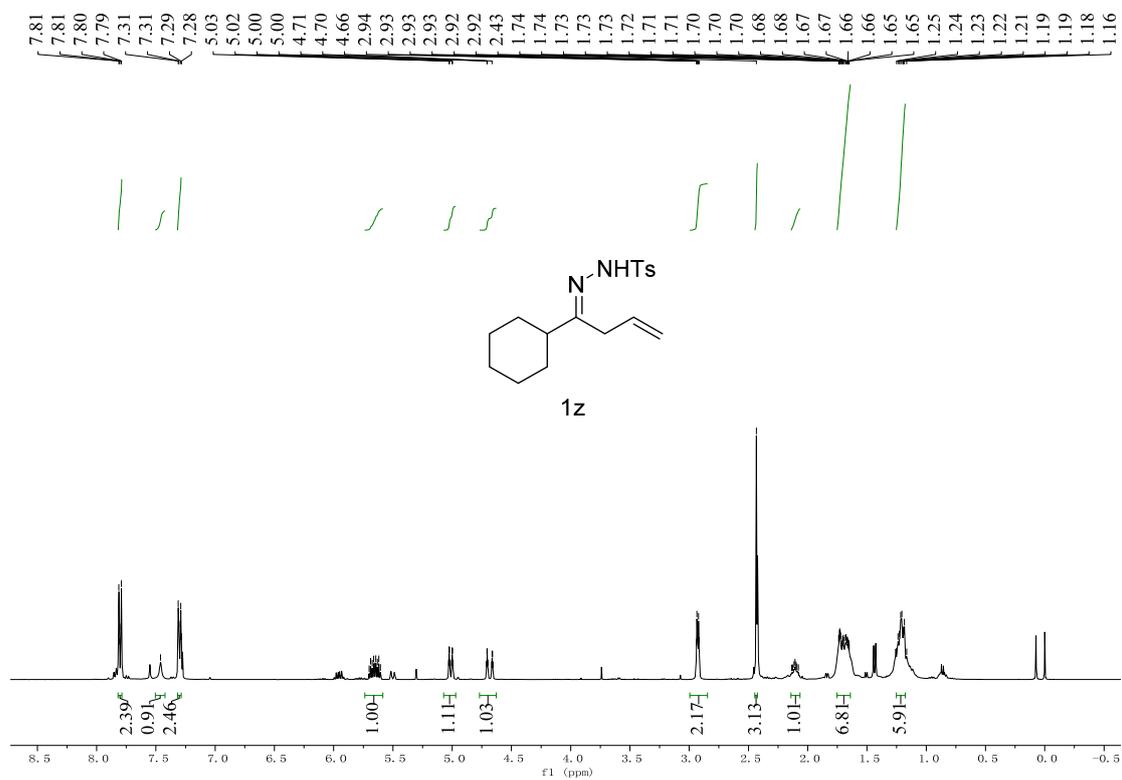
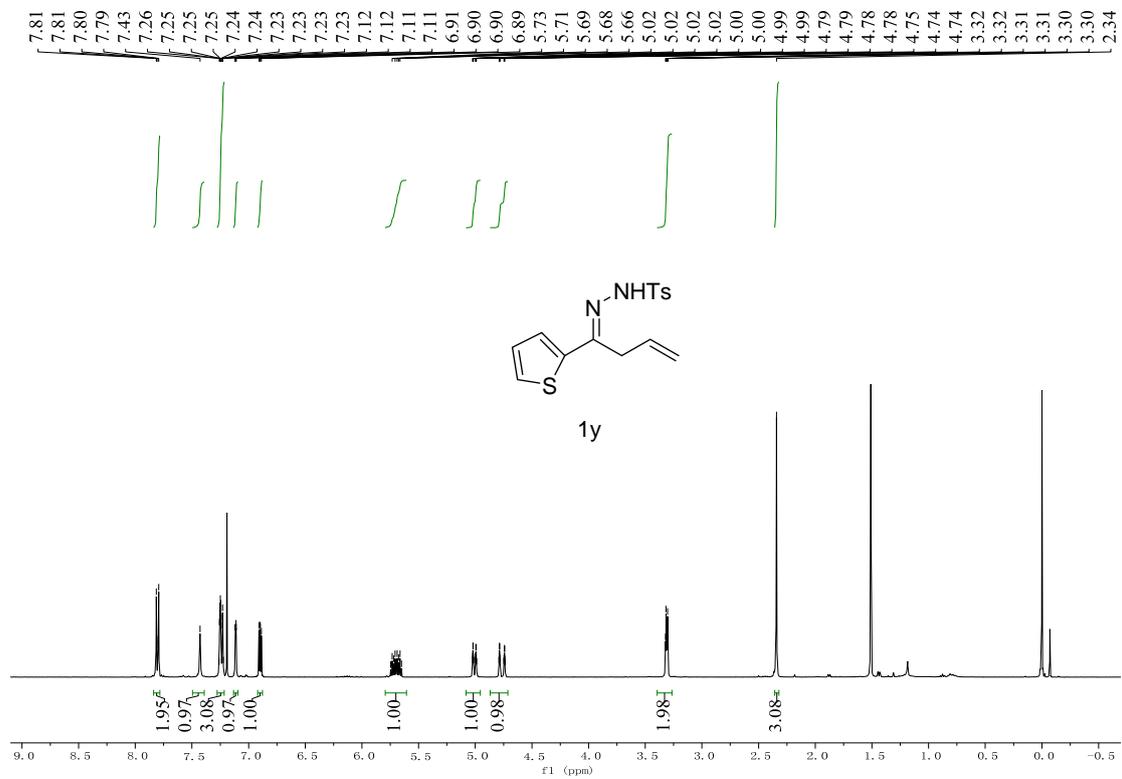


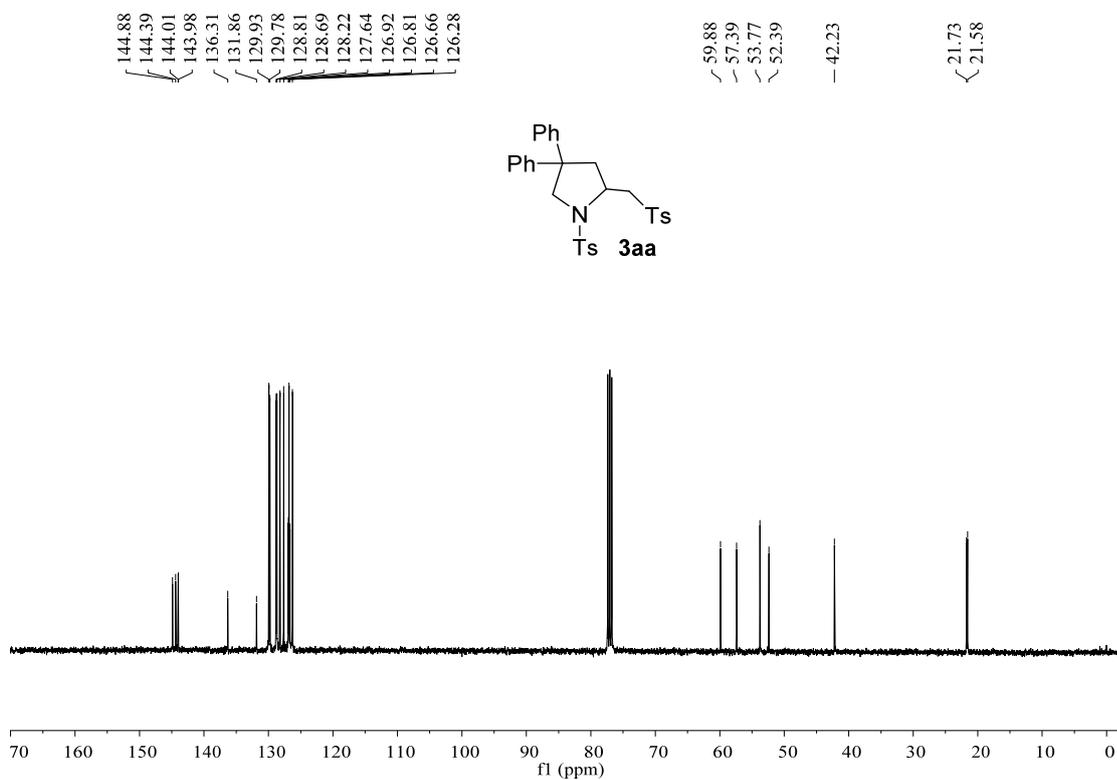
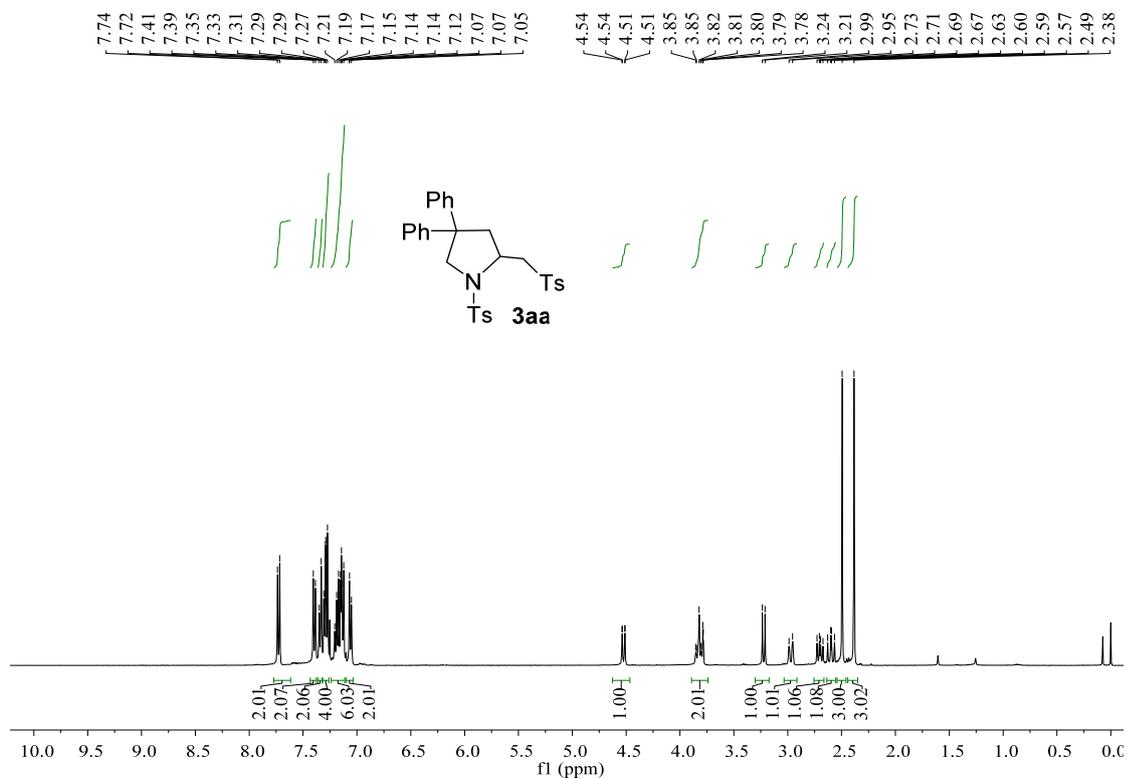


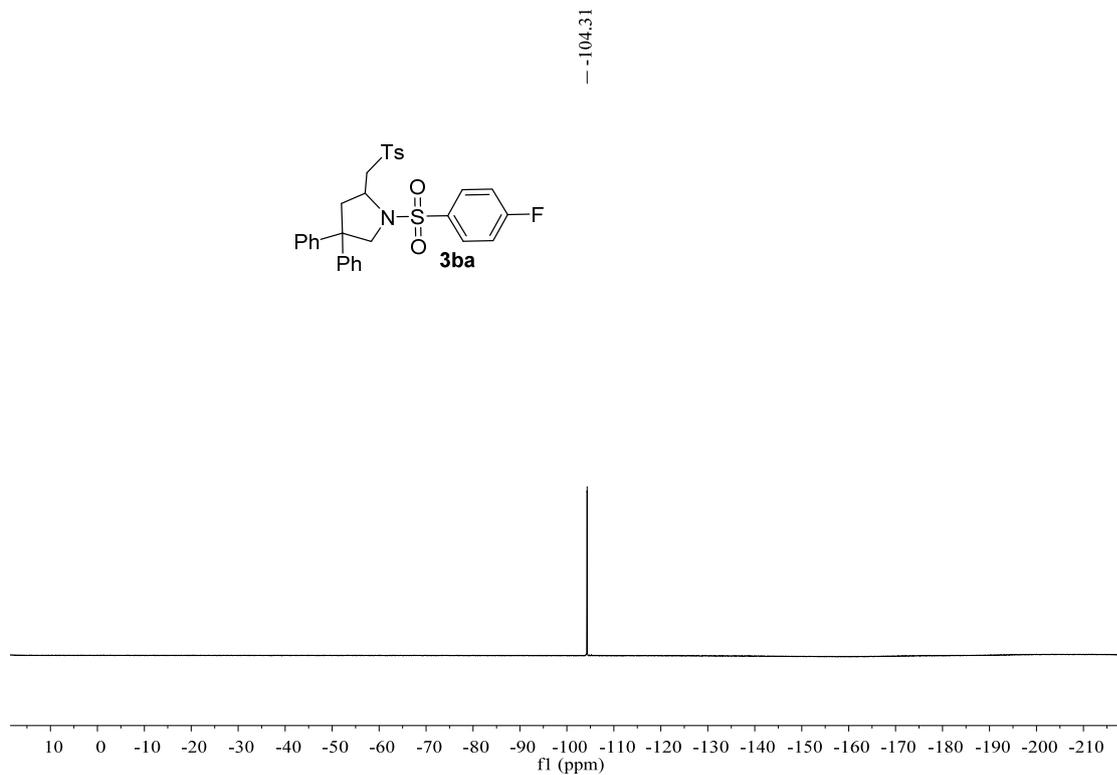
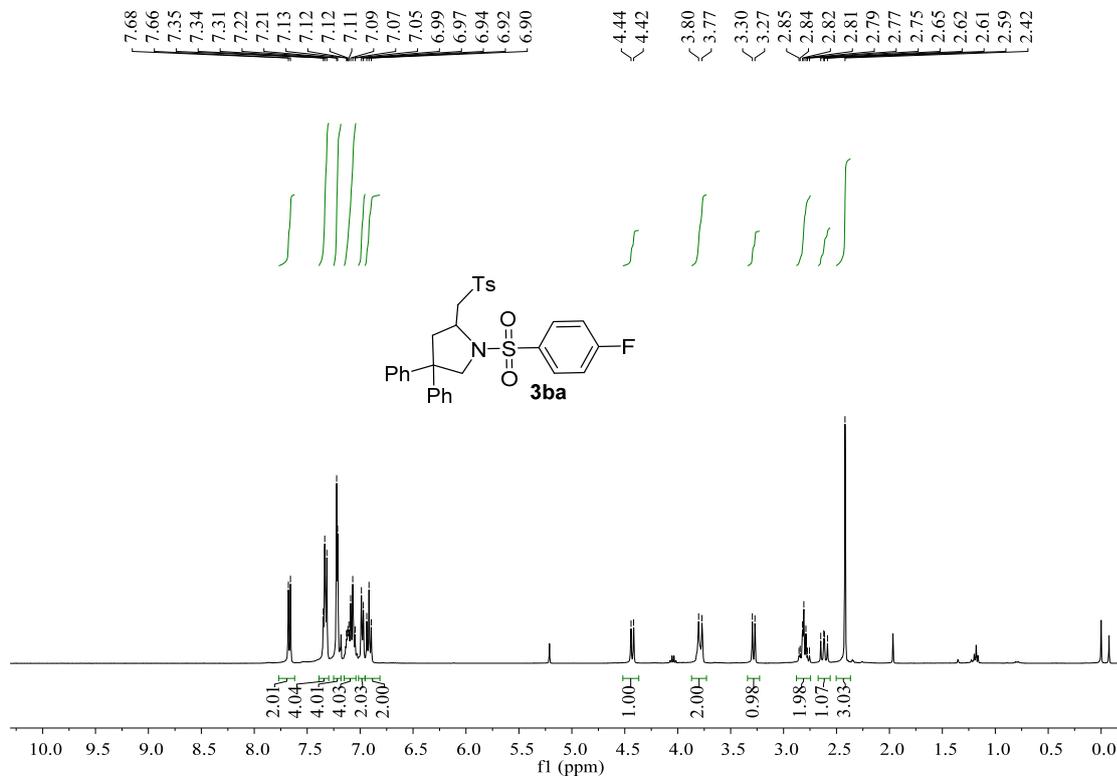


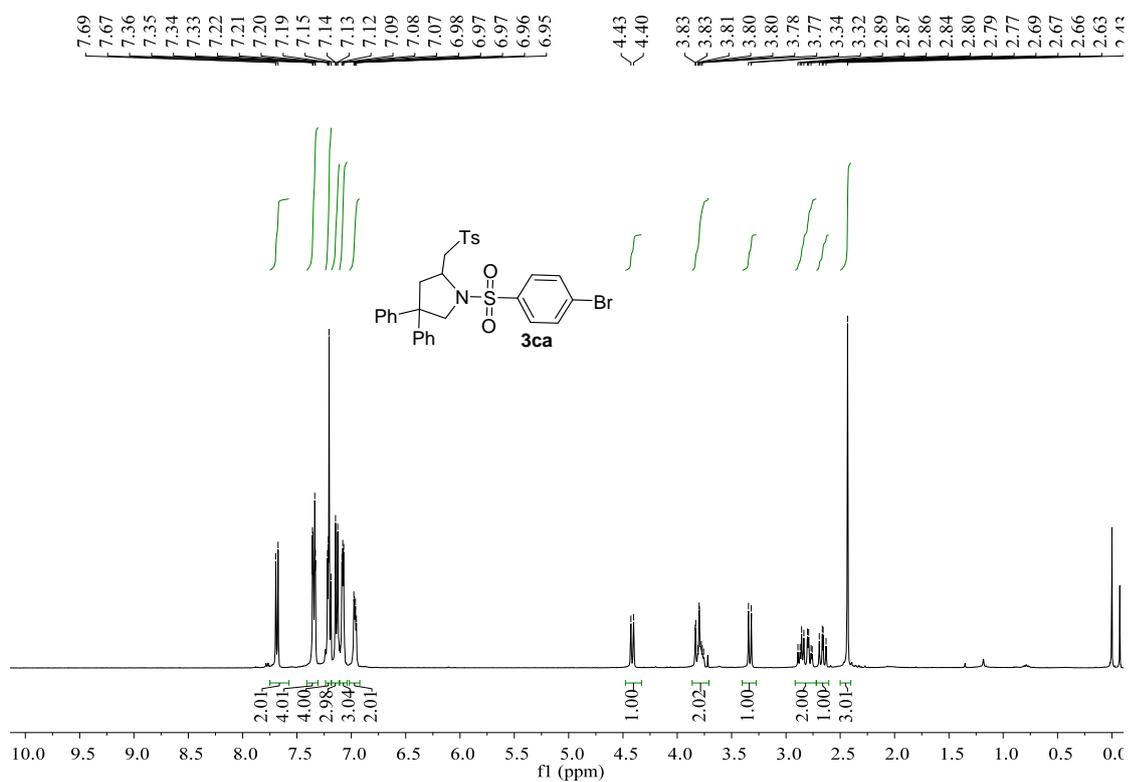
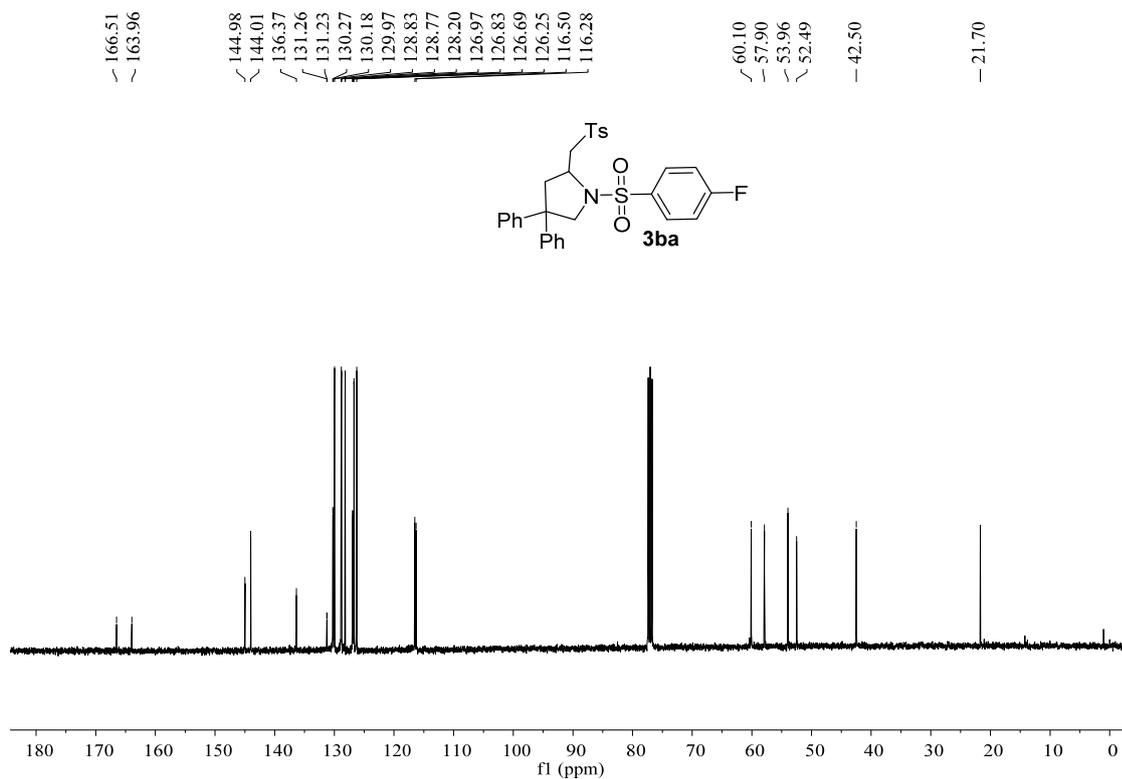


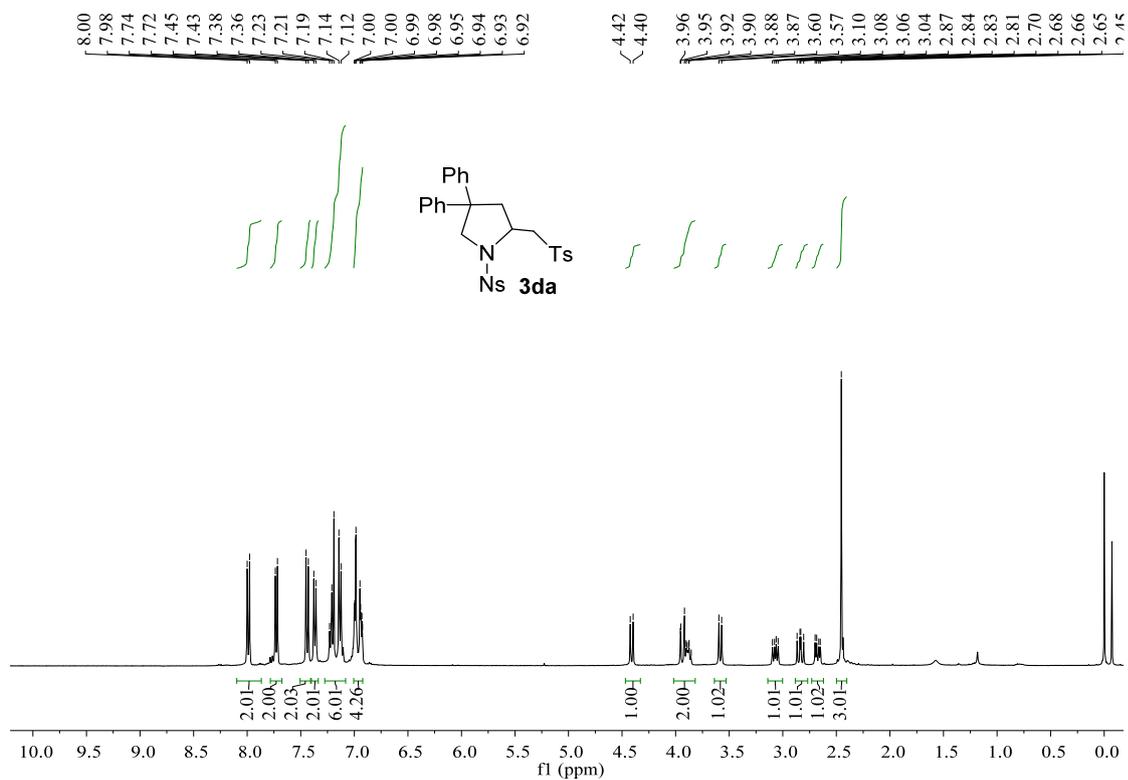
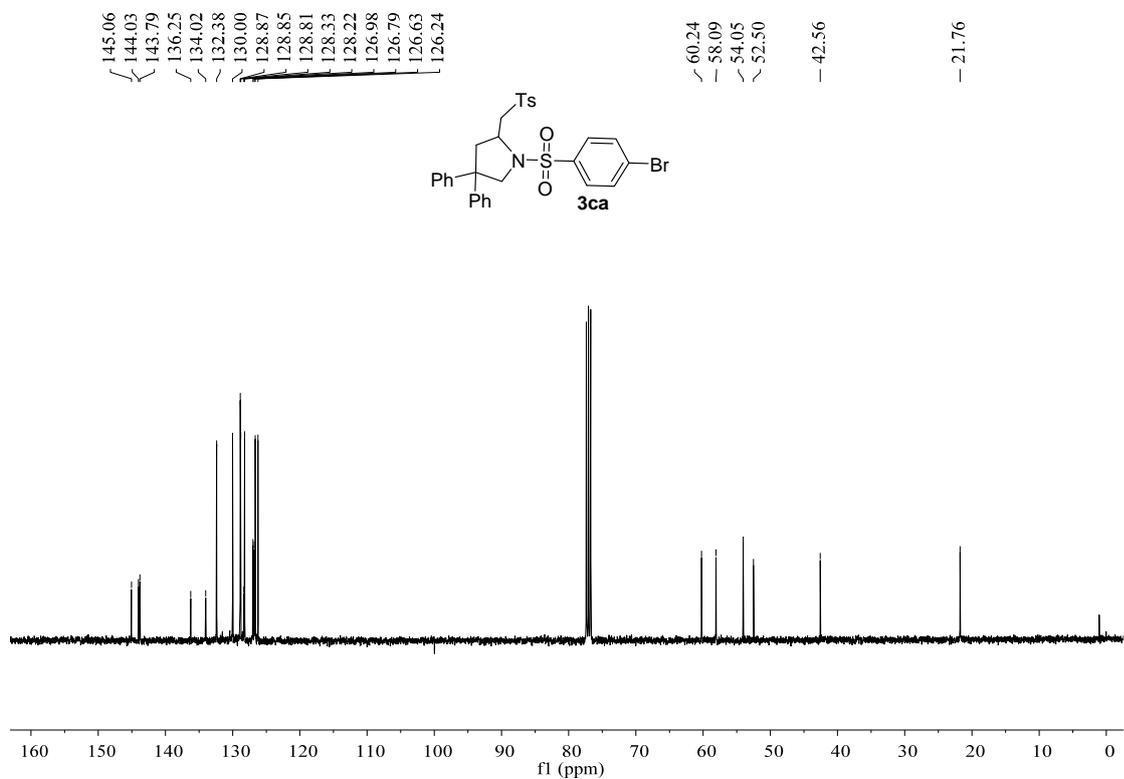


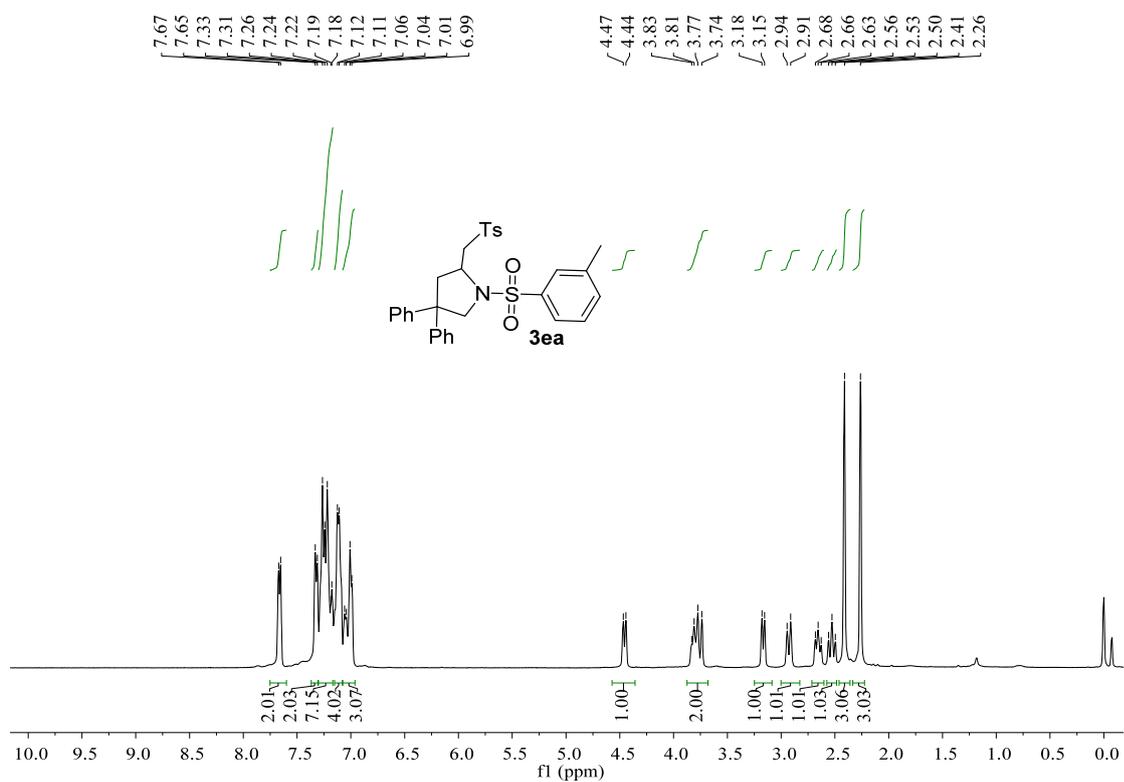
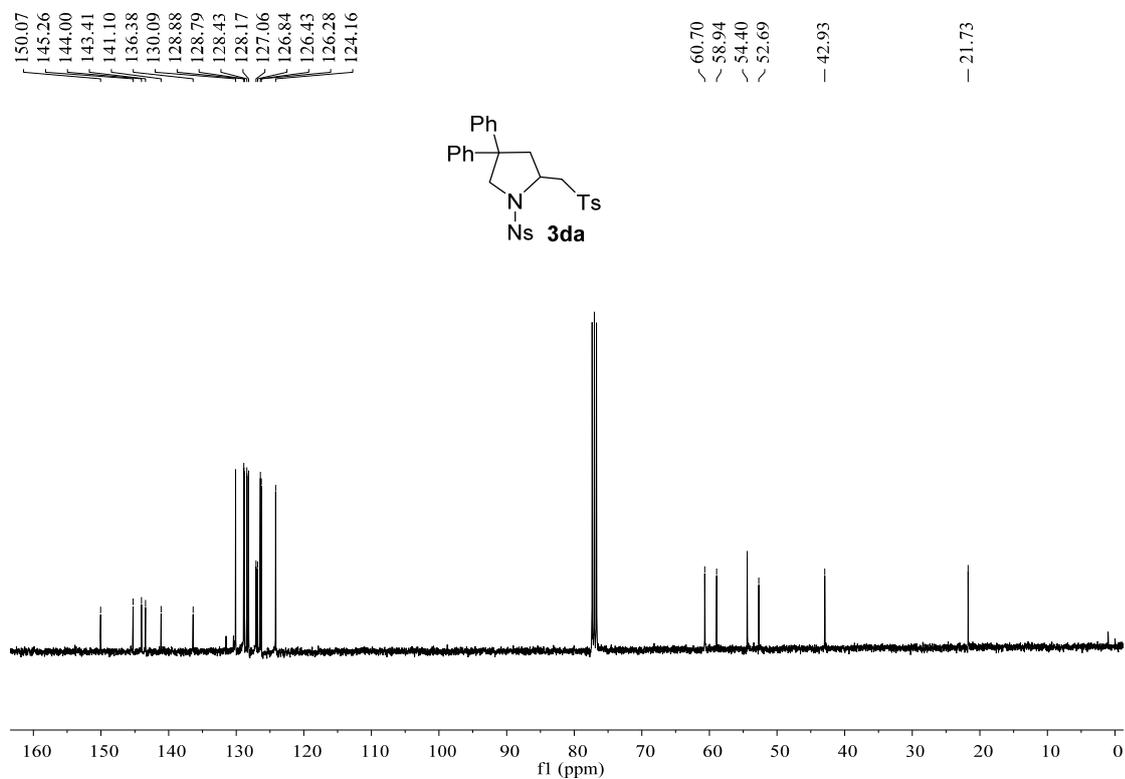


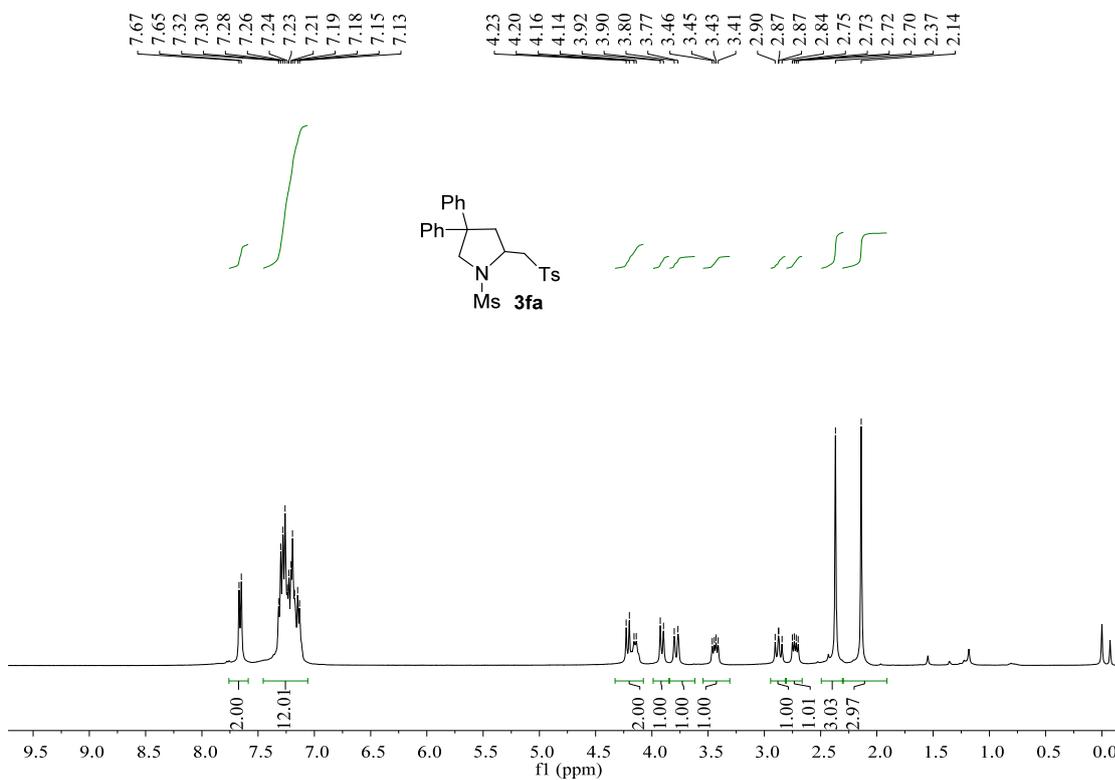
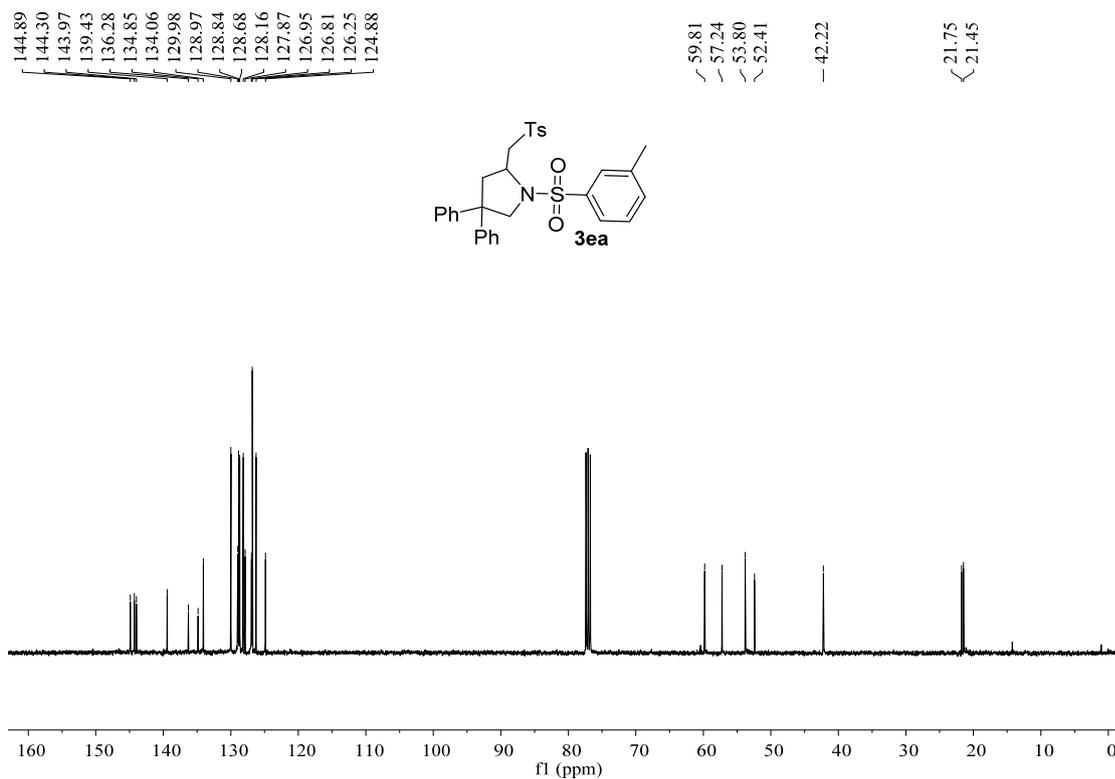


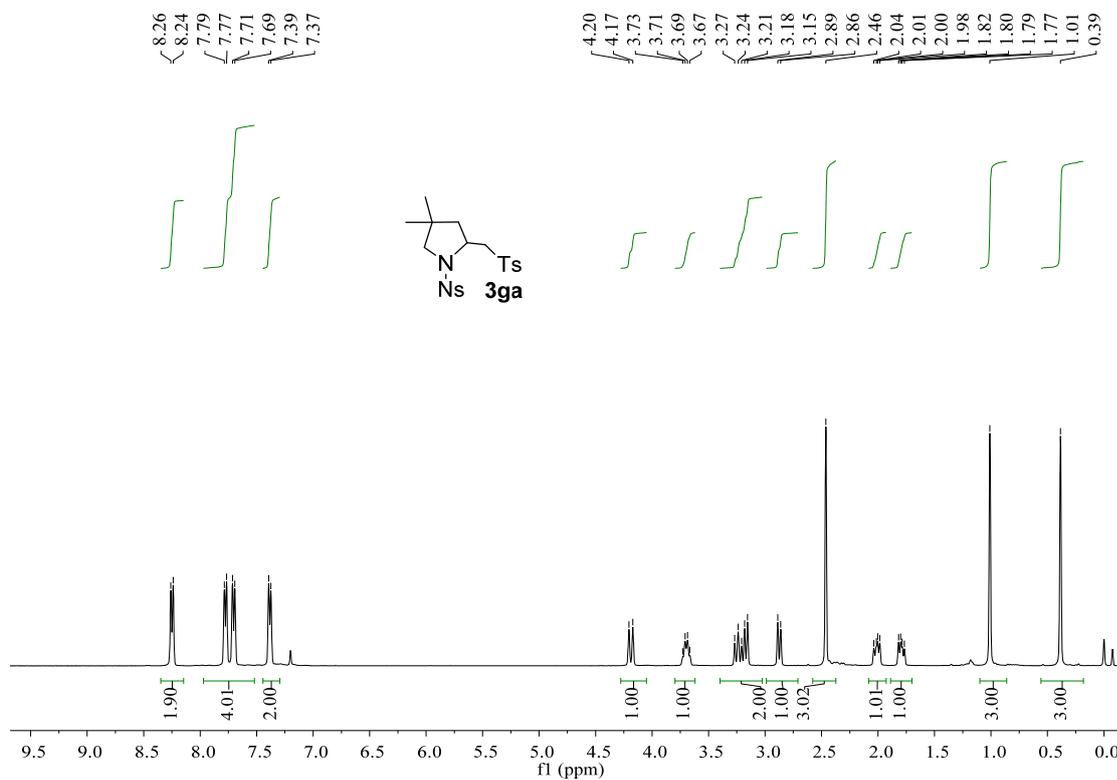
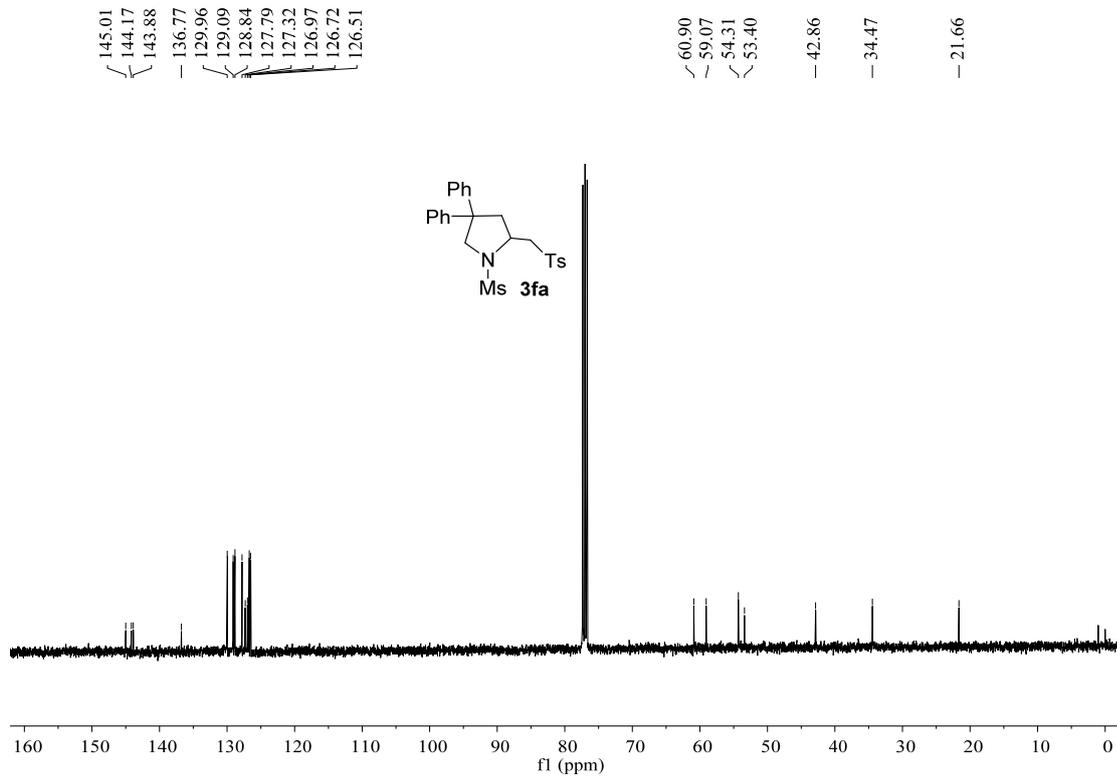








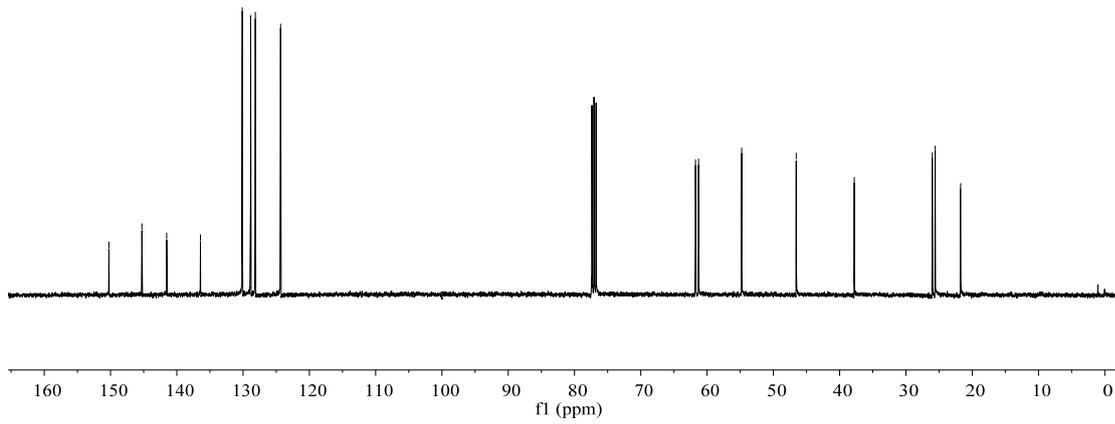
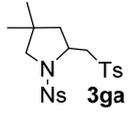




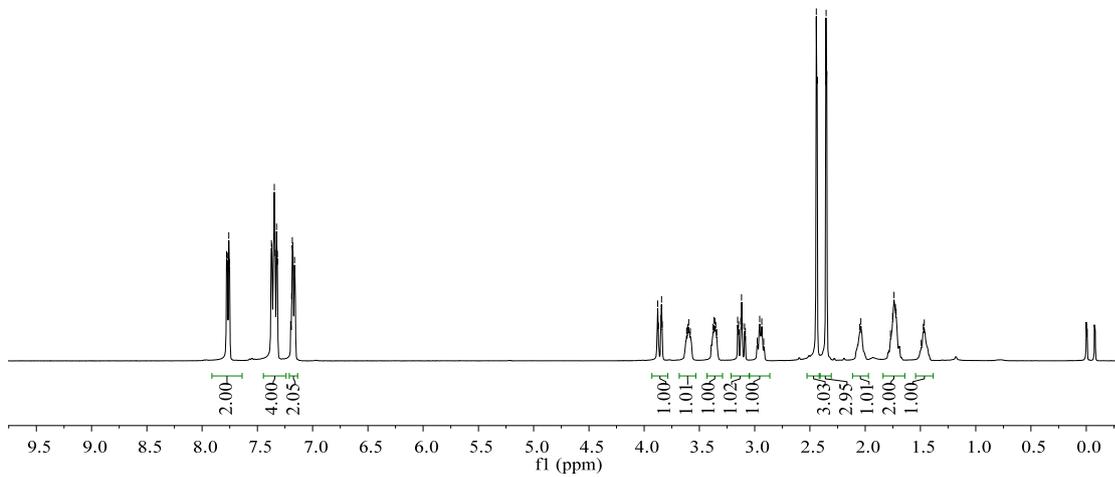
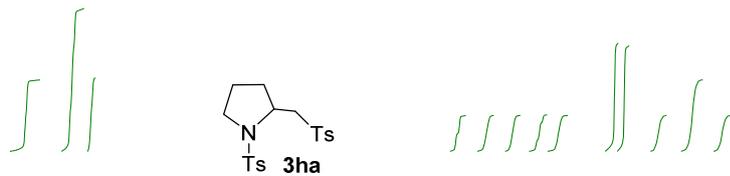


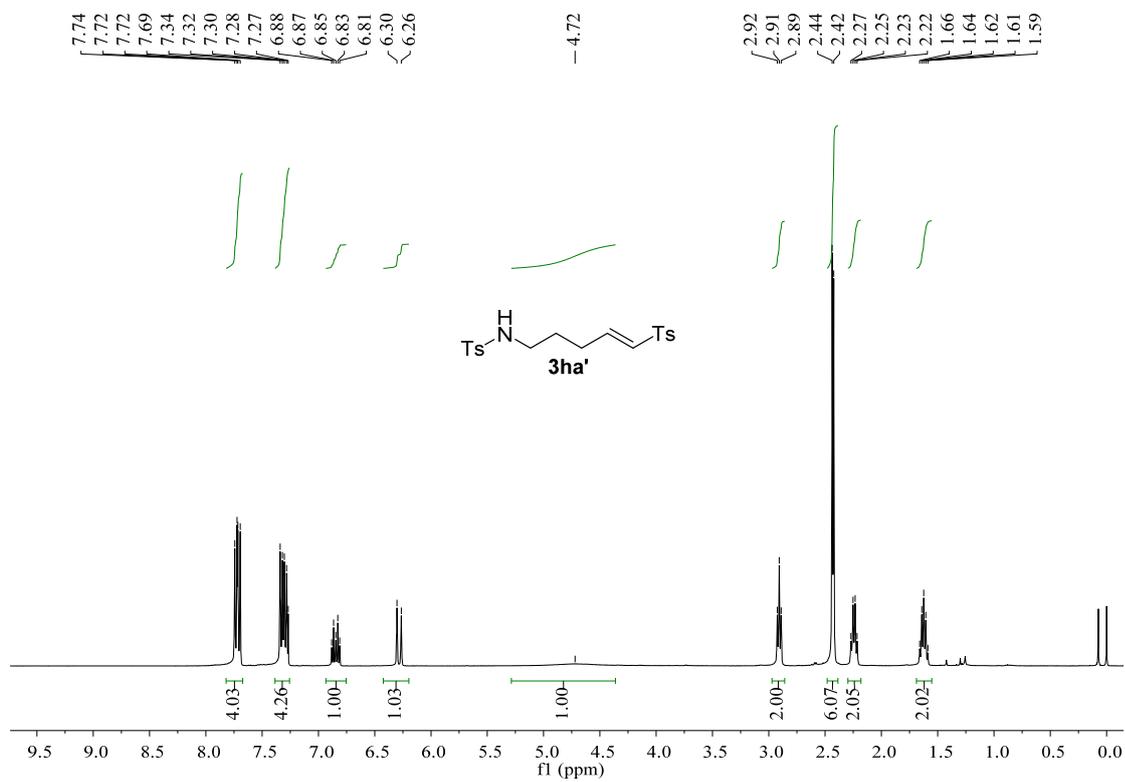
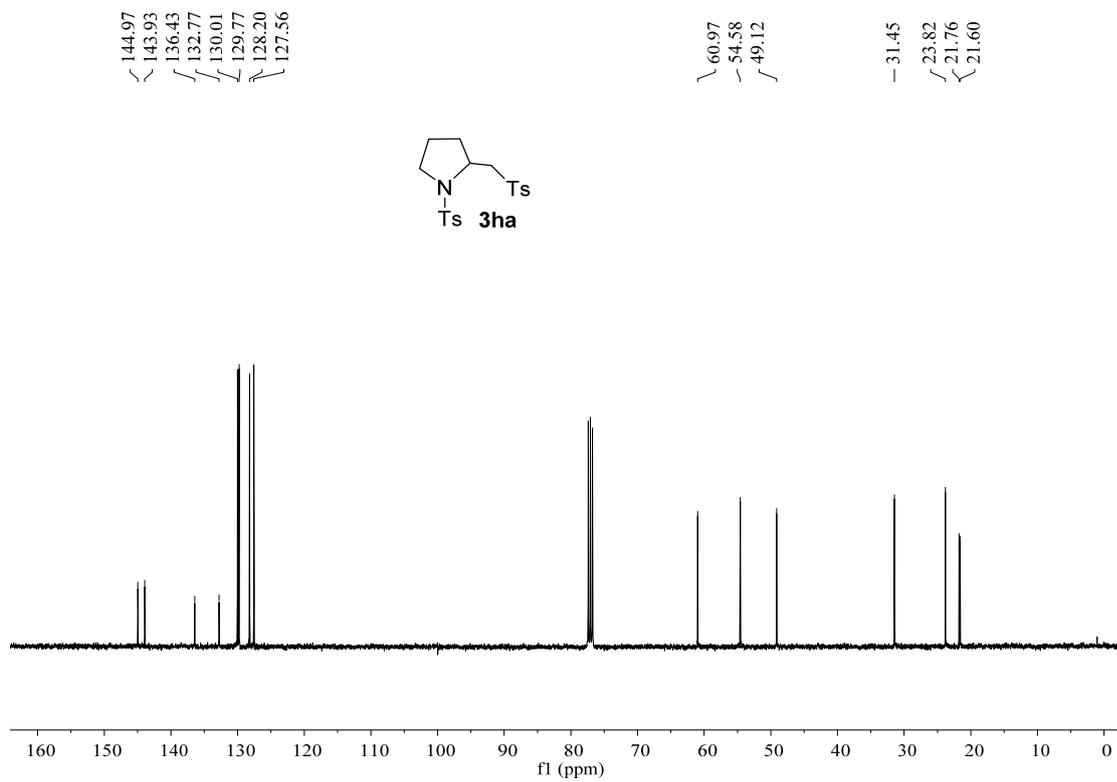
150.25  
145.26  
141.54  
136.43  
130.14  
128.87  
128.16  
124.36

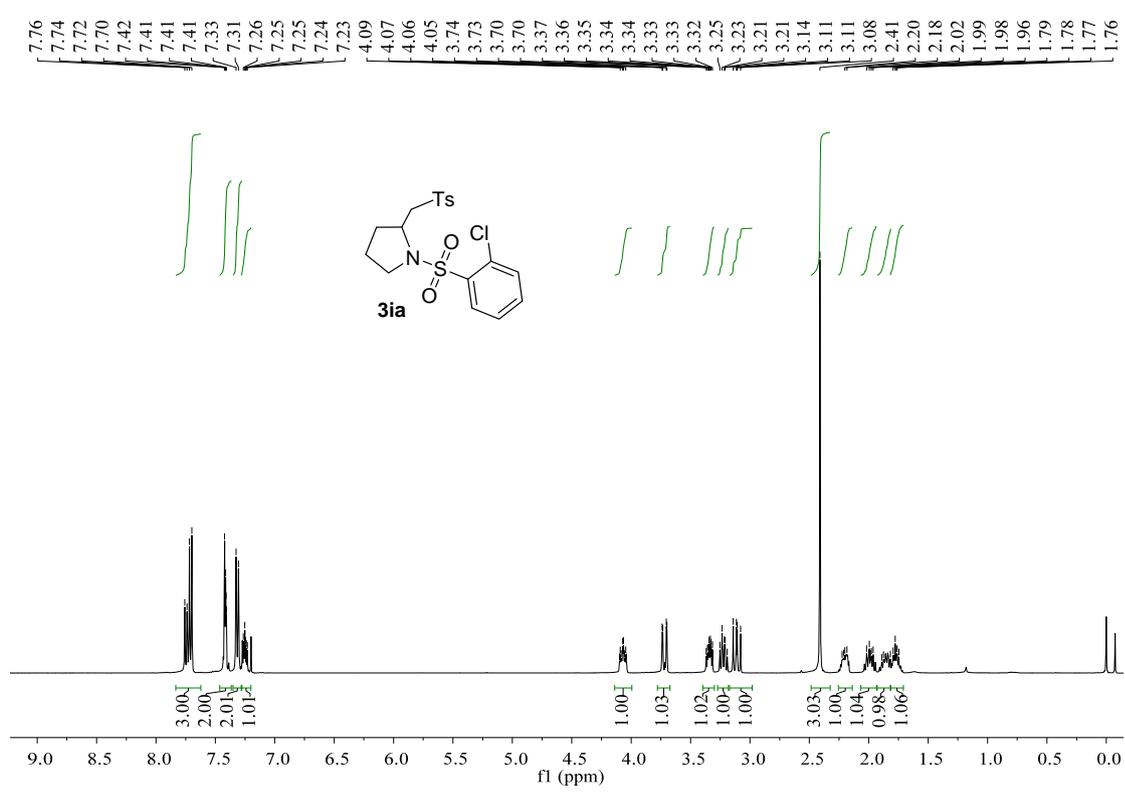
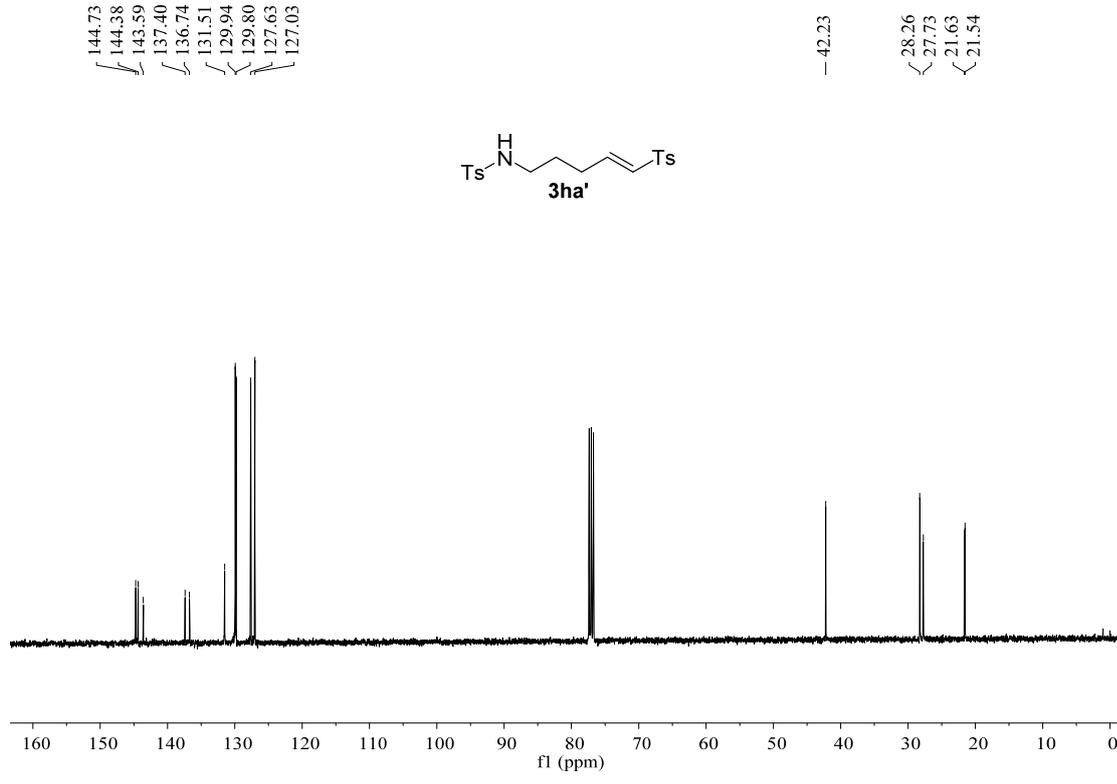
61.75  
61.30  
54.79  
46.55  
37.80  
26.03  
25.60  
21.76

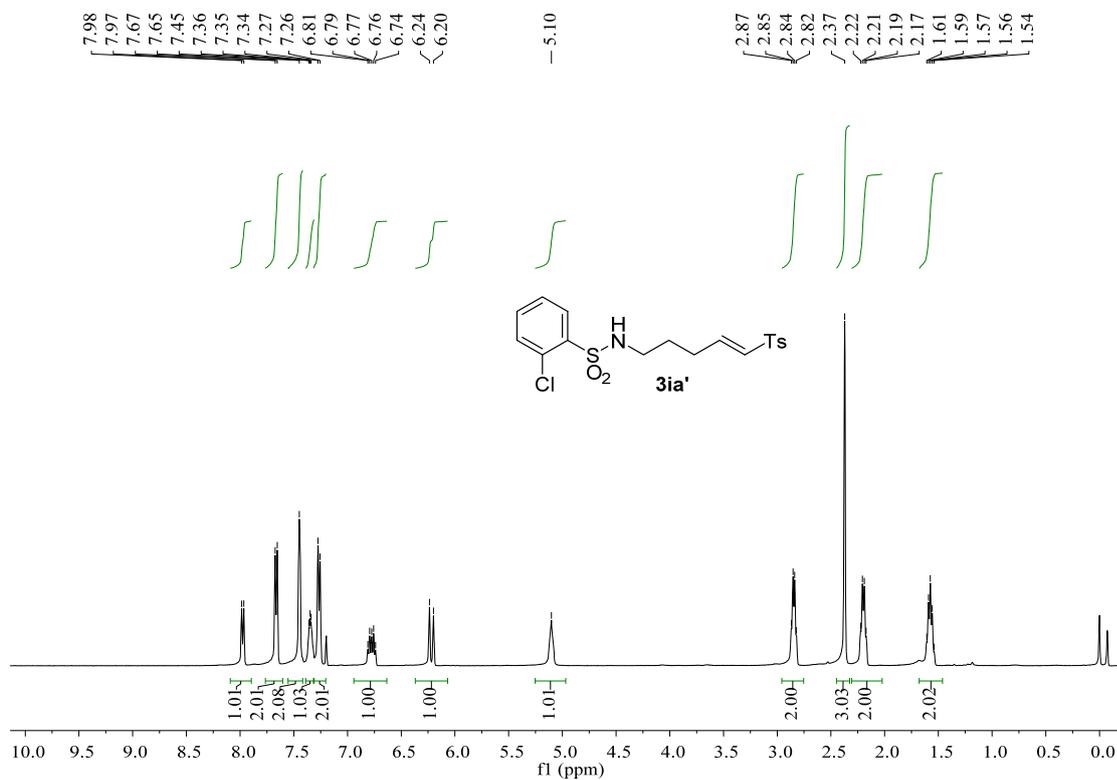
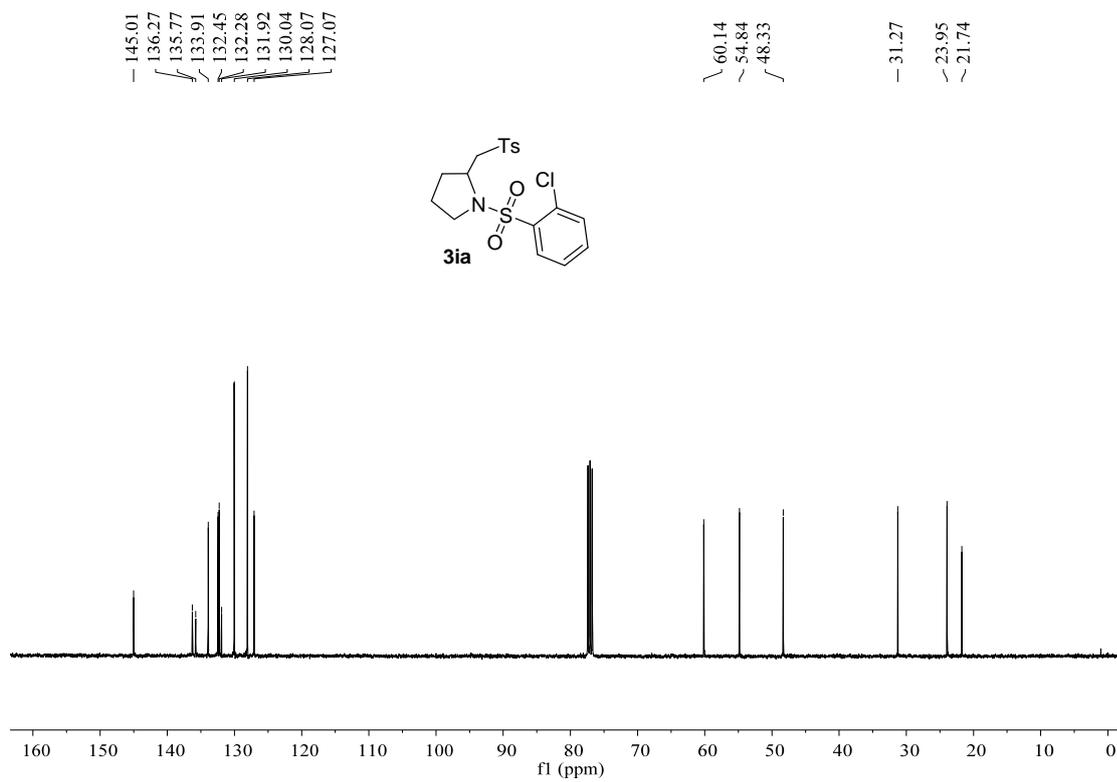


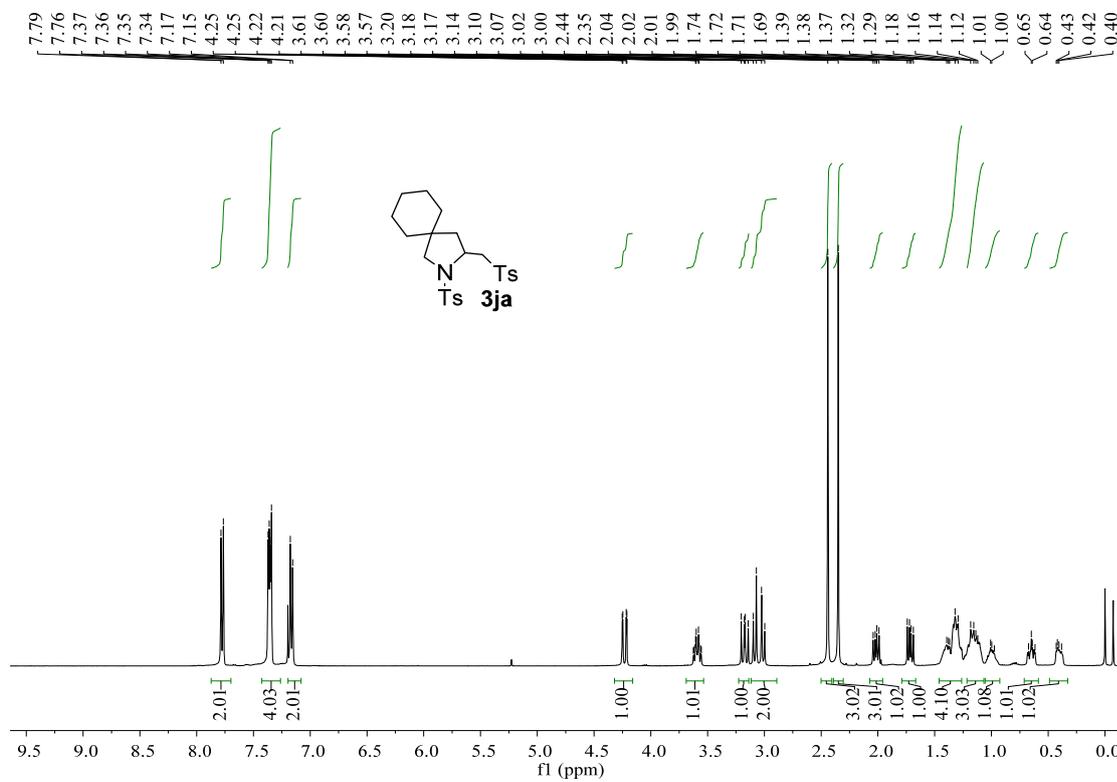
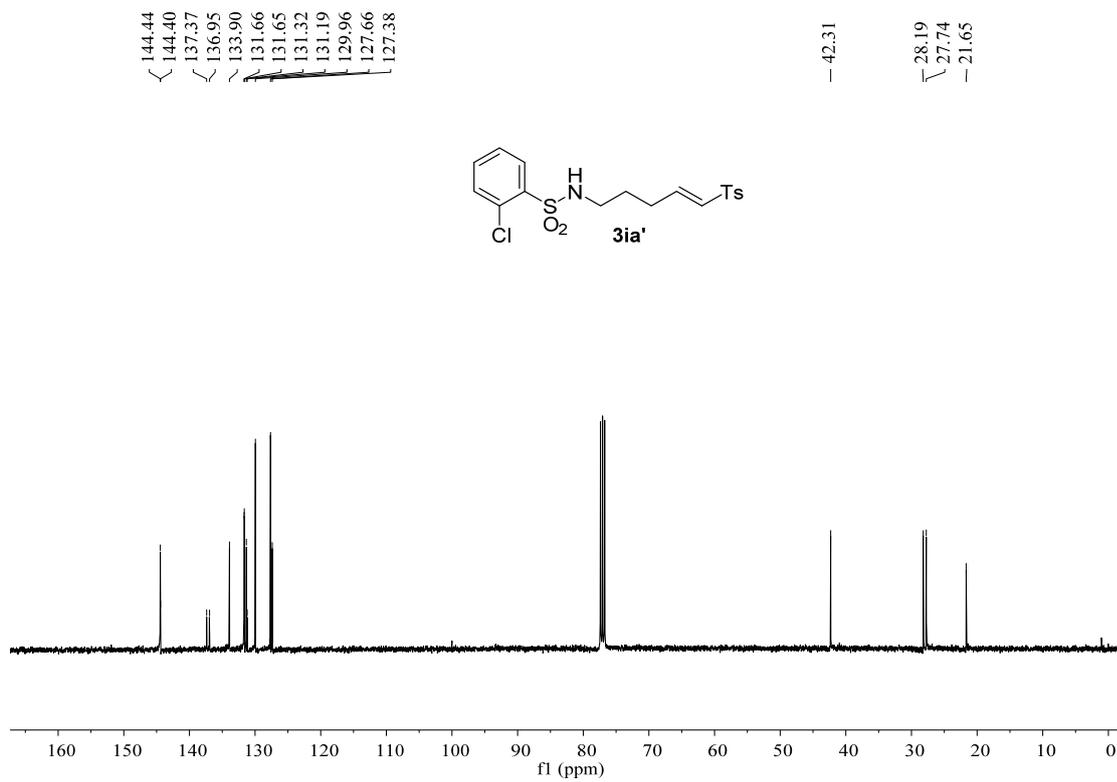
7.78  
7.78  
7.77  
7.76  
7.75  
7.37  
7.35  
7.35  
7.34  
7.33  
7.32  
7.19  
7.18  
7.18  
7.16  
3.88  
3.87  
3.85  
3.84  
3.84  
3.62  
3.61  
3.61  
3.60  
3.58  
3.38  
3.37  
3.36  
3.35  
3.34  
3.15  
3.15  
3.13  
3.12  
3.11  
3.09  
3.08  
2.96  
2.94  
2.44  
2.35  
2.05  
2.04  
1.75  
1.74  
1.73  
1.72  
1.47  
1.47

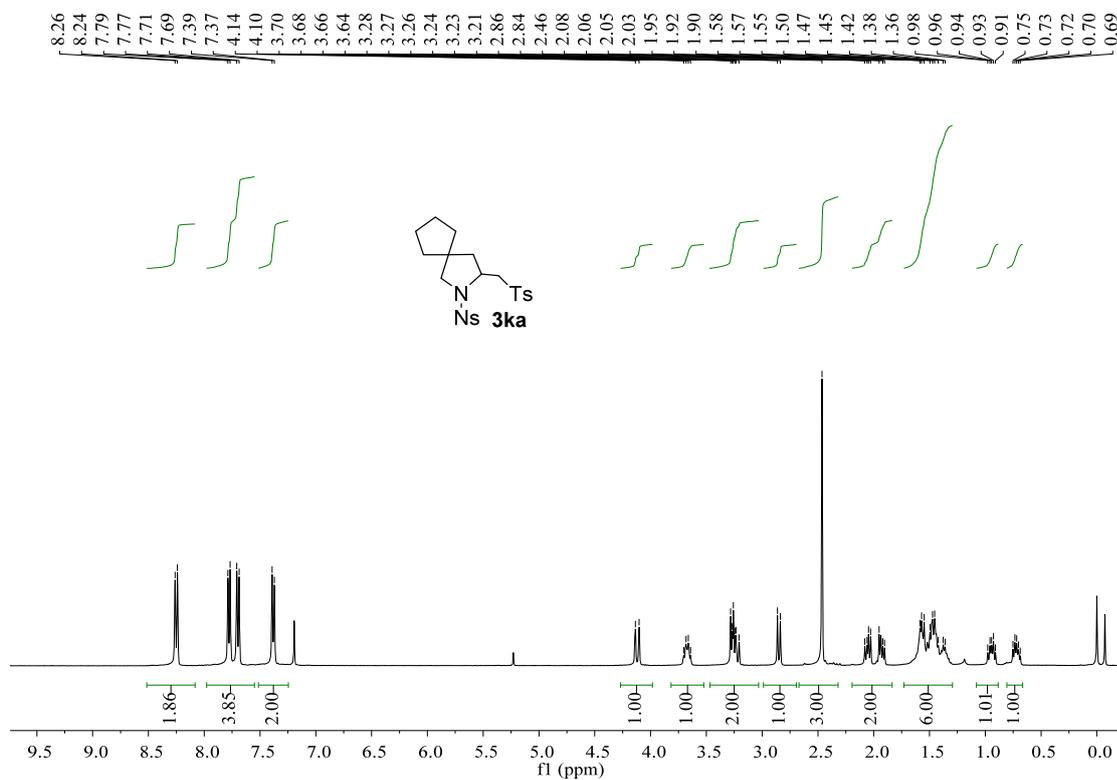
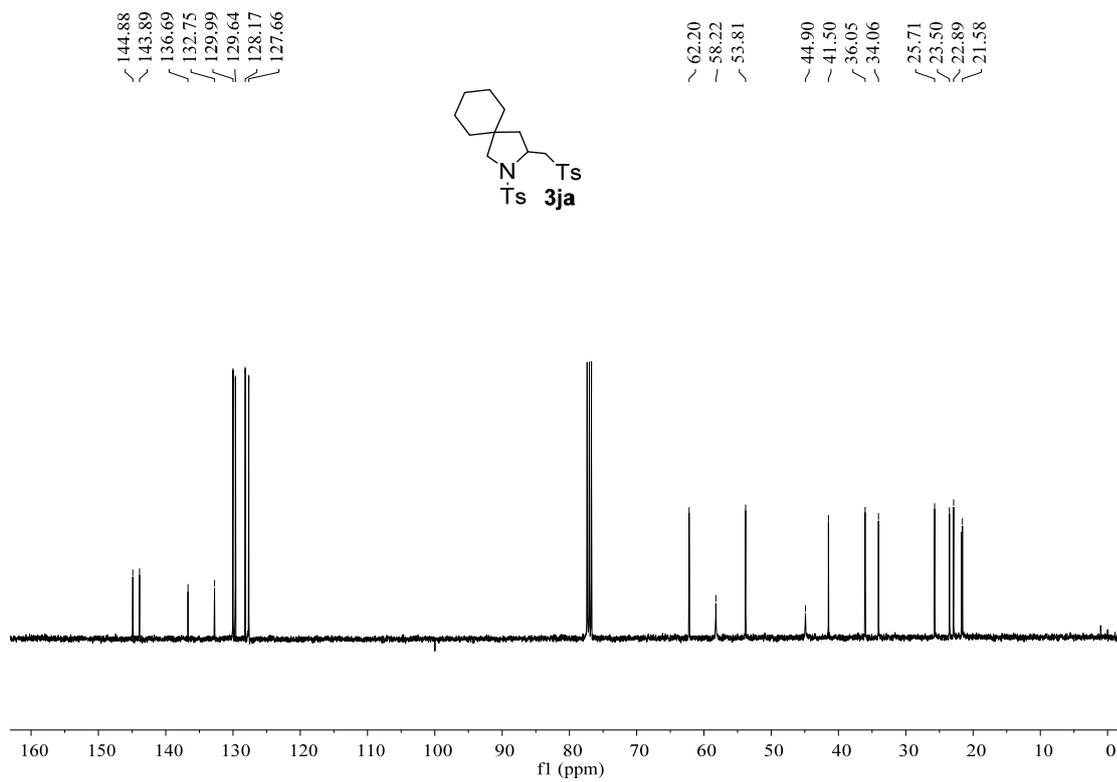


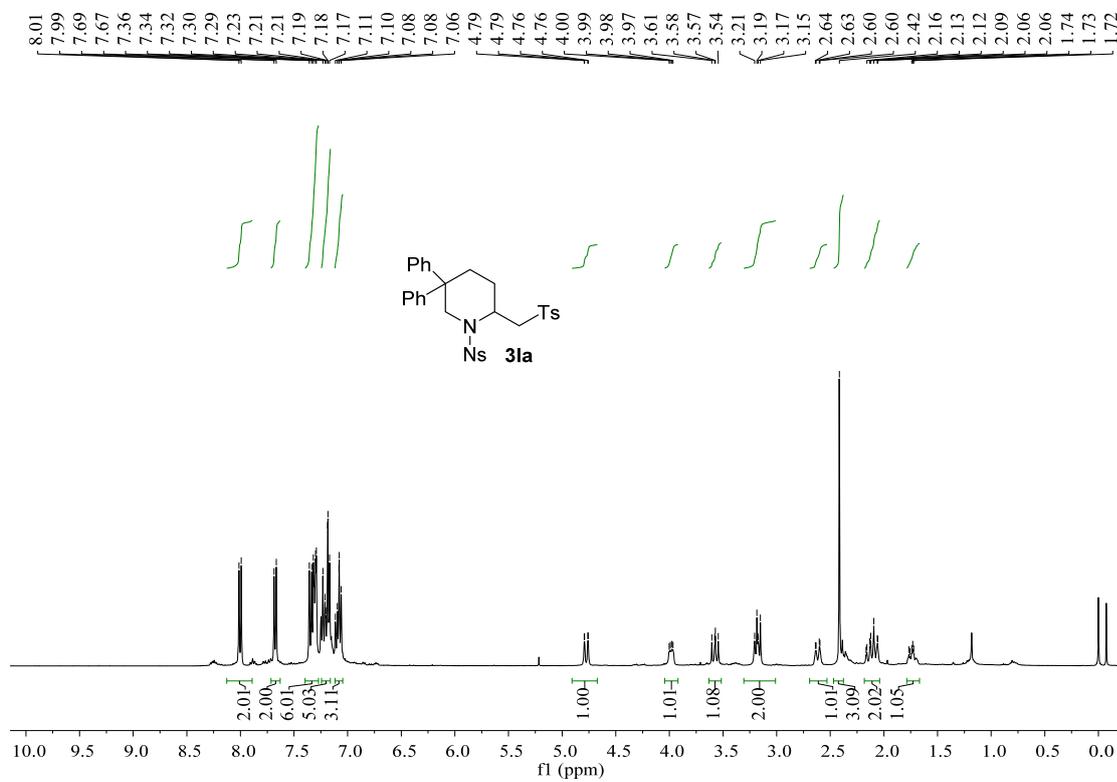
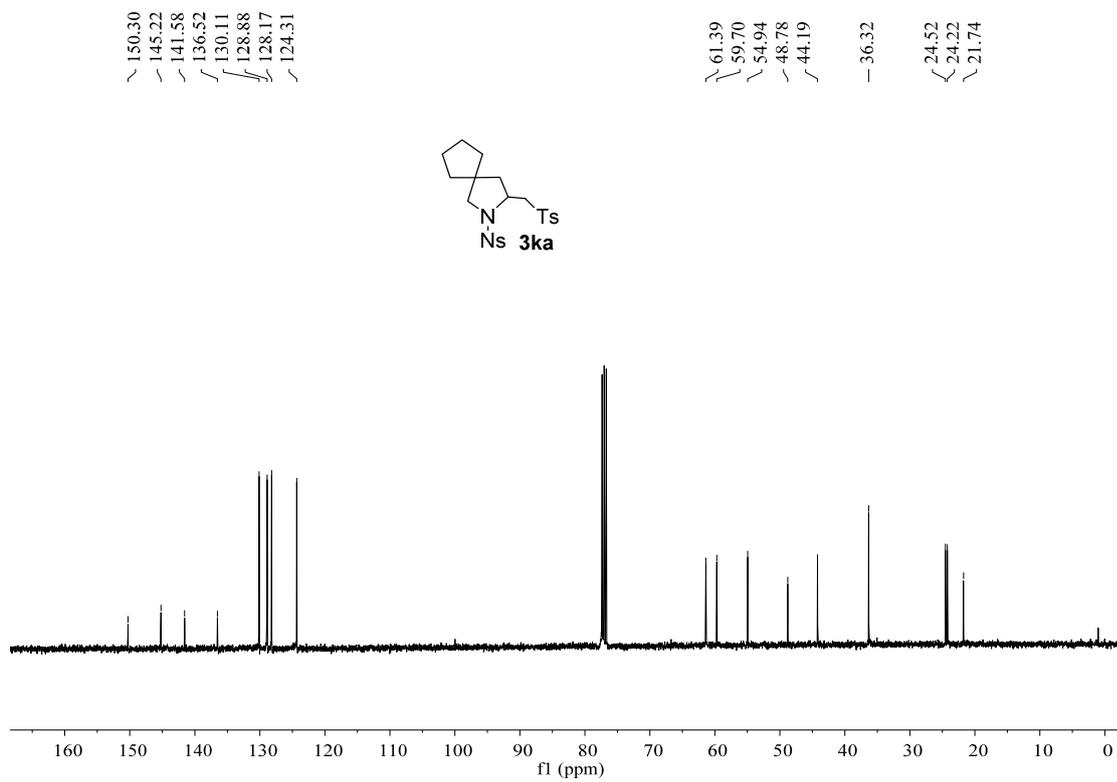


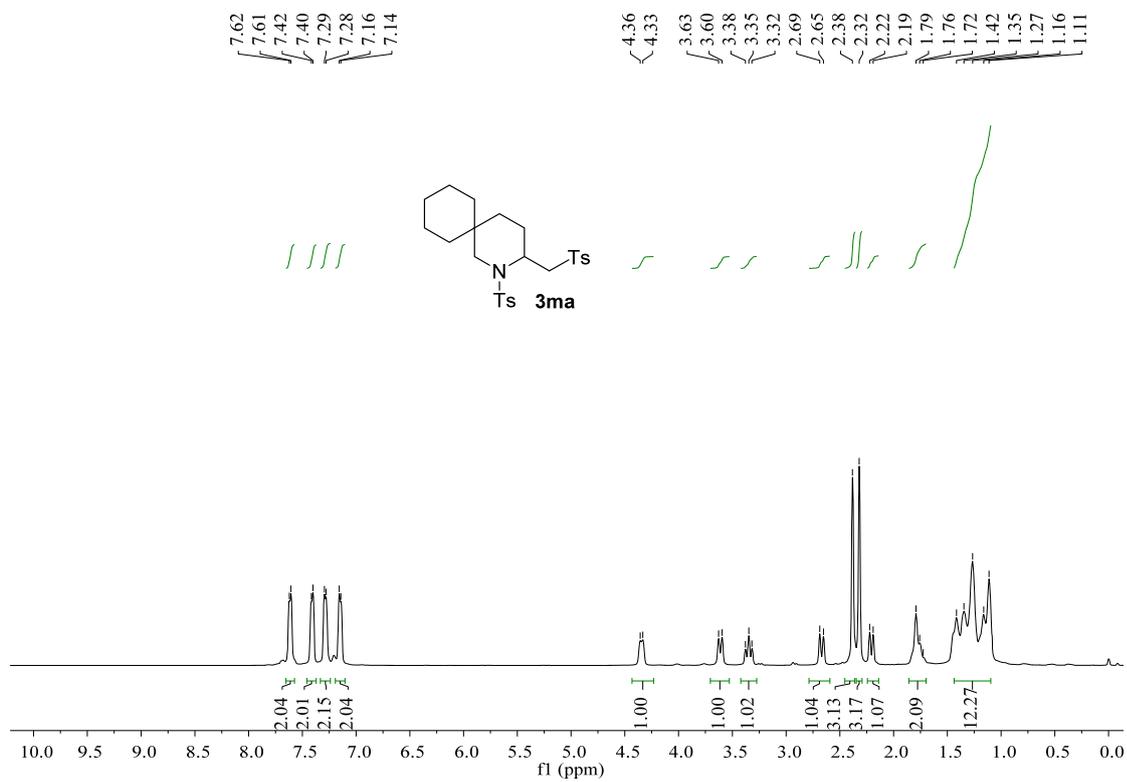
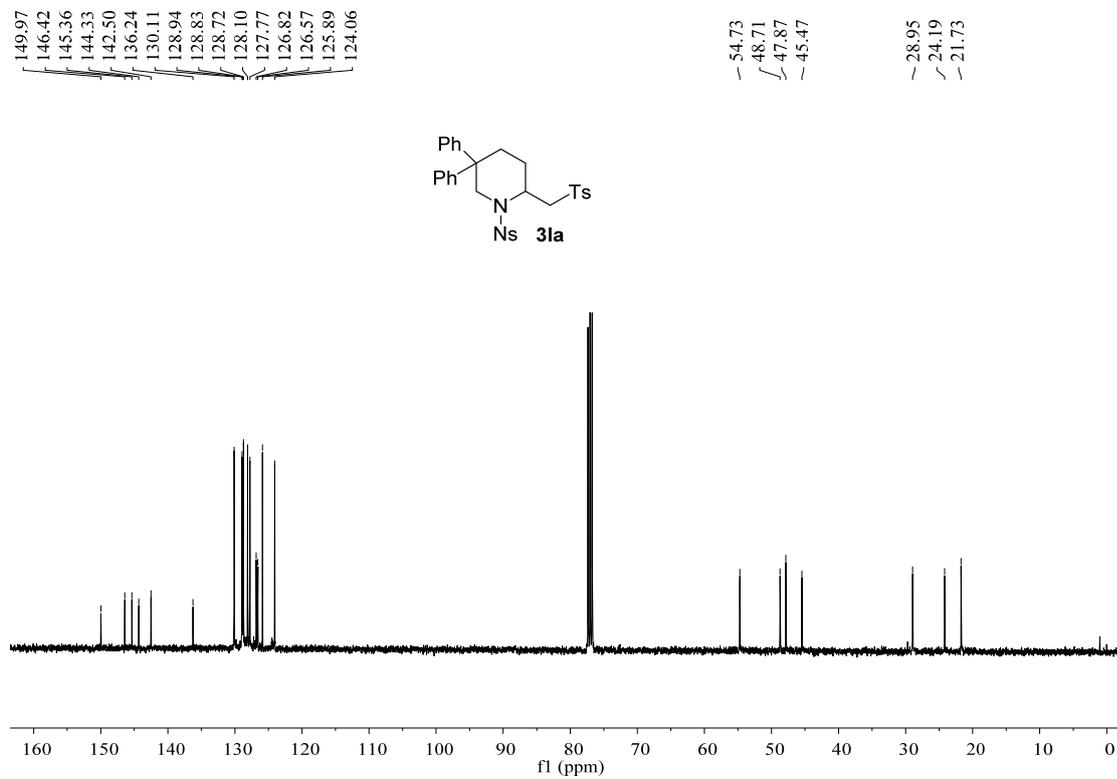




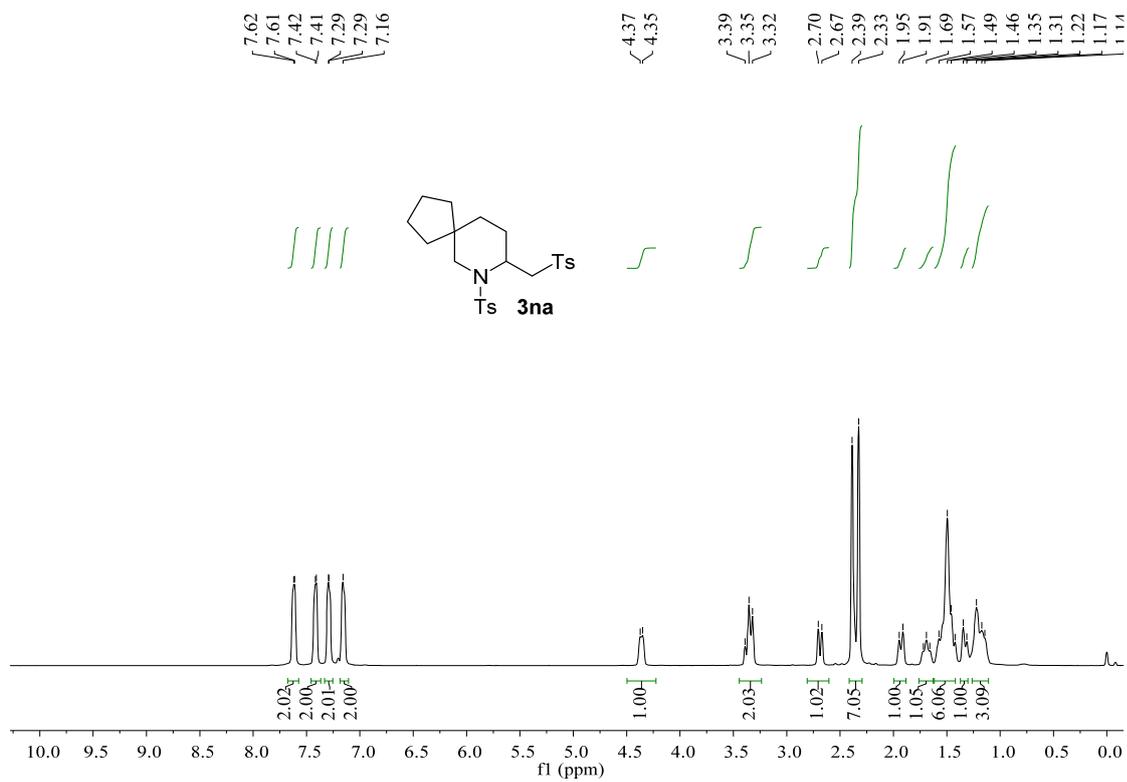
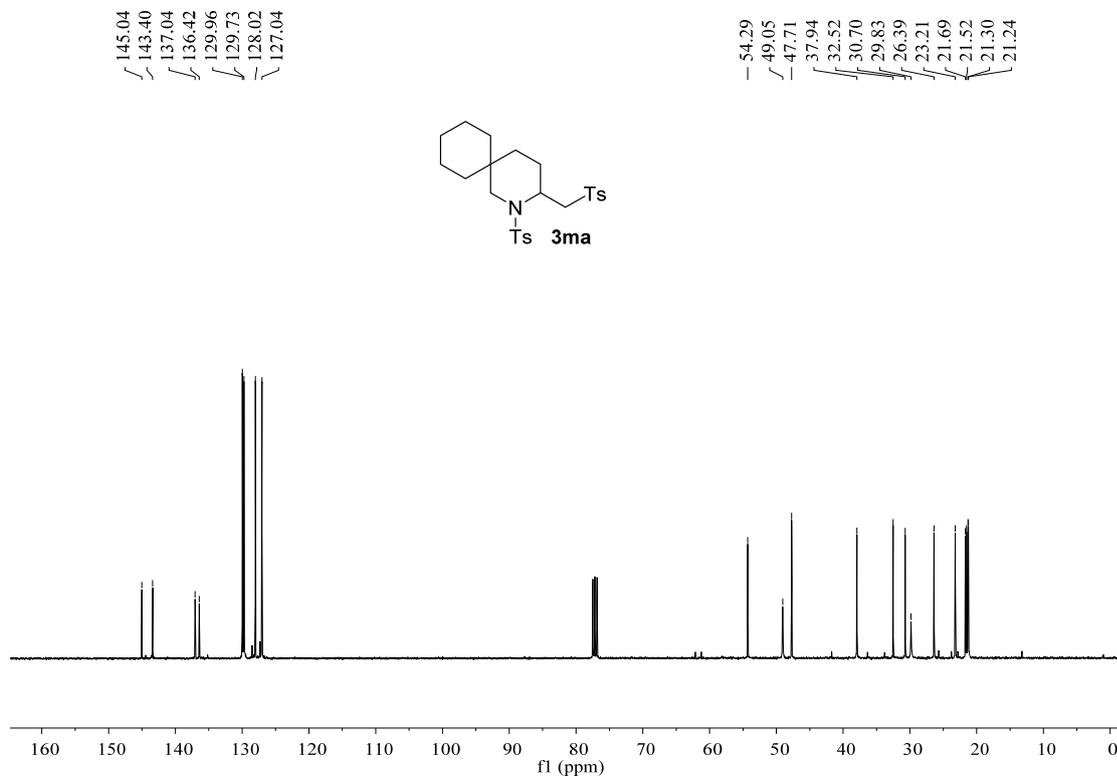


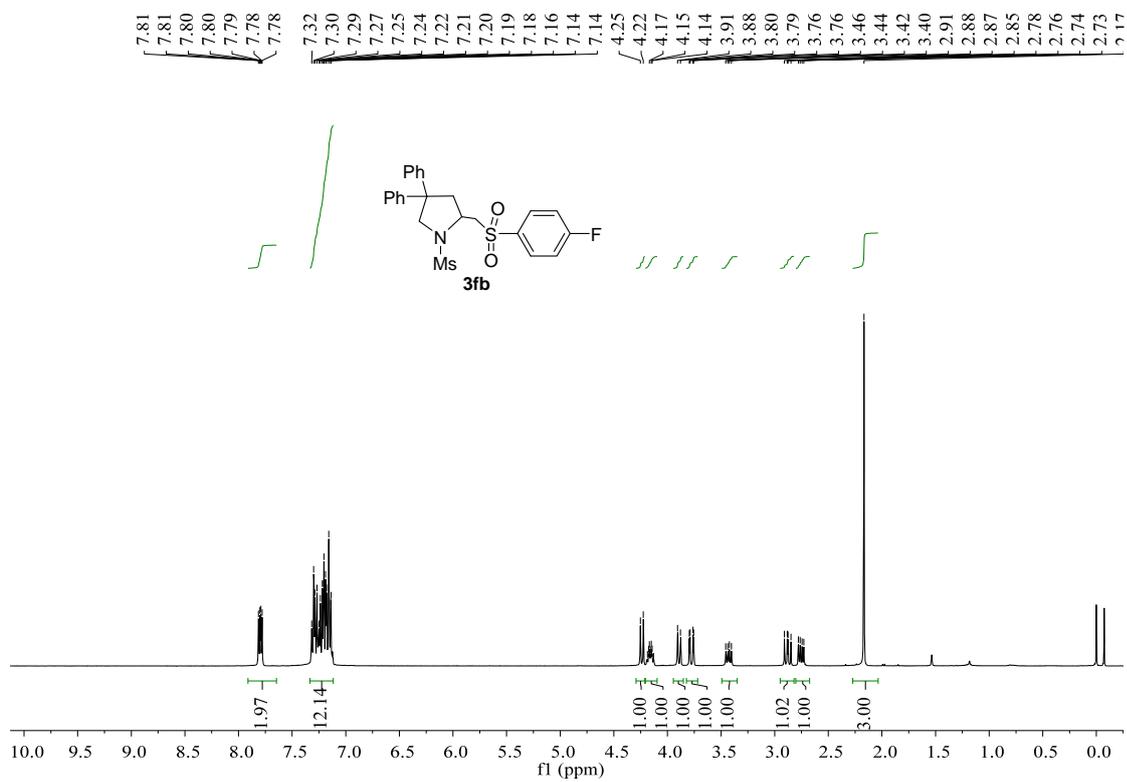
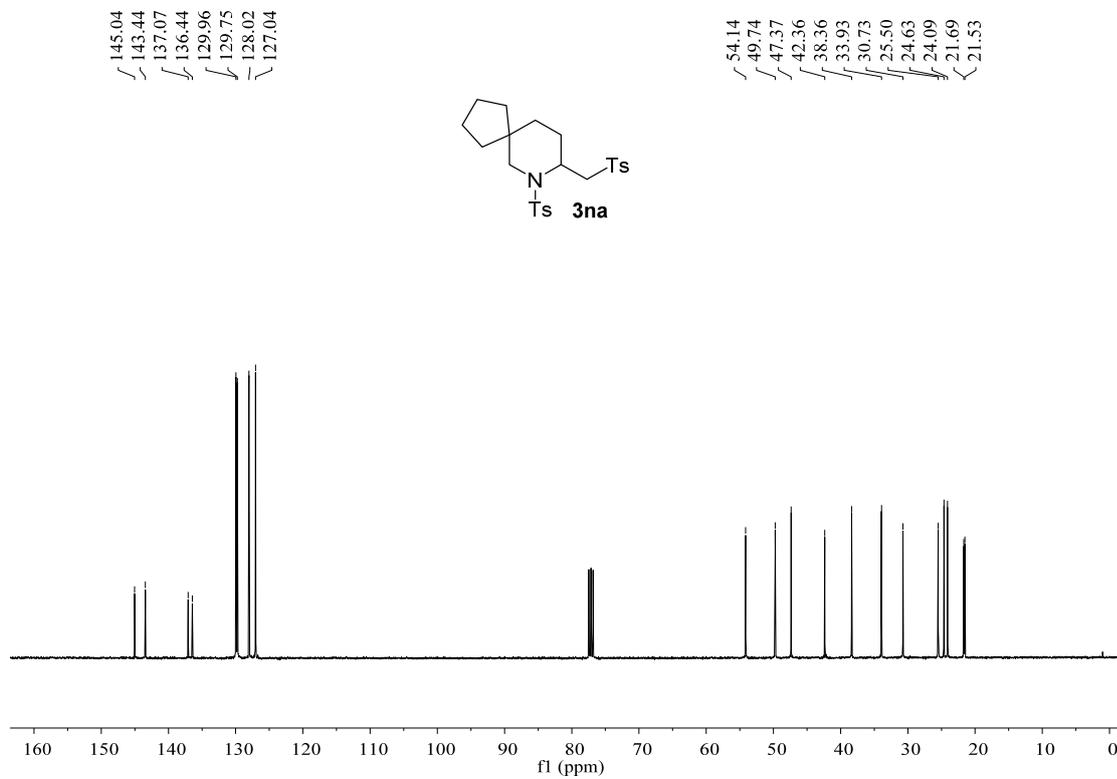


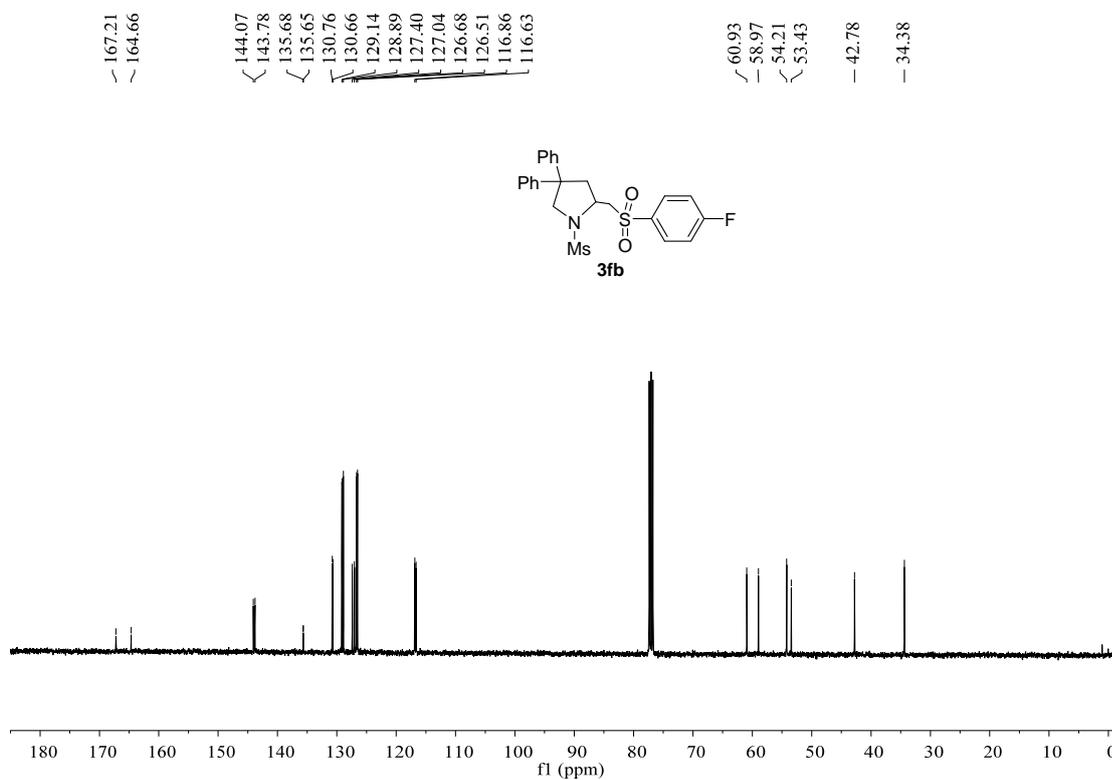
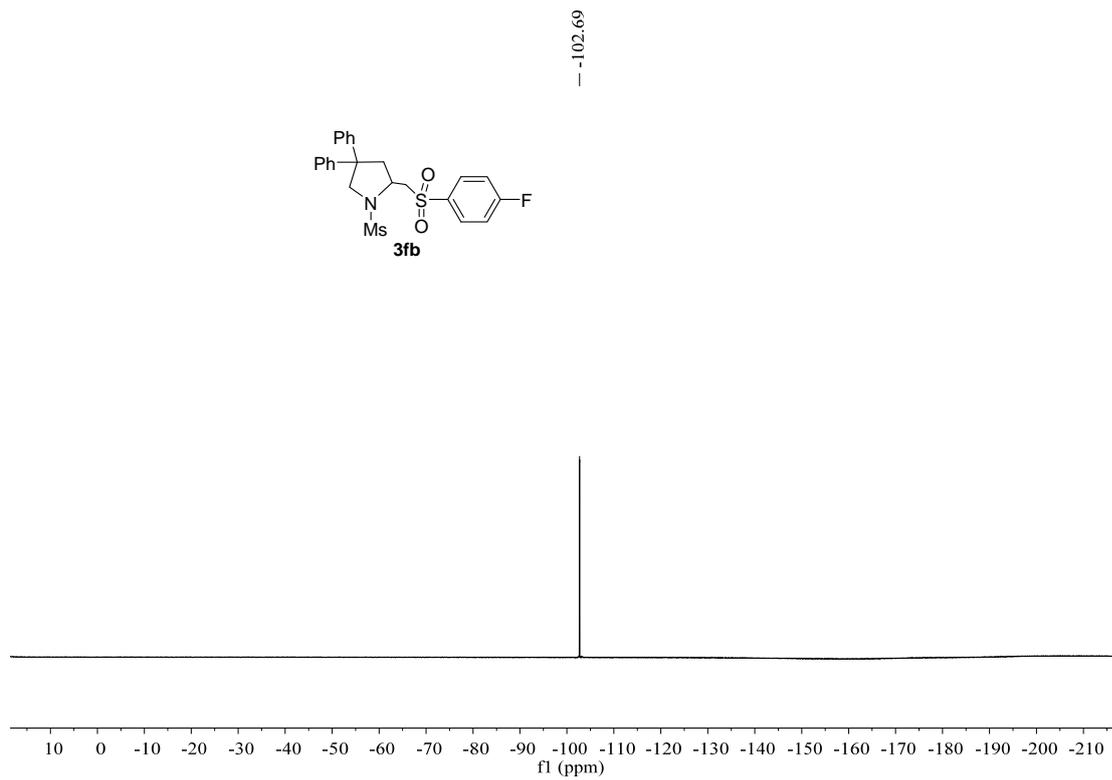


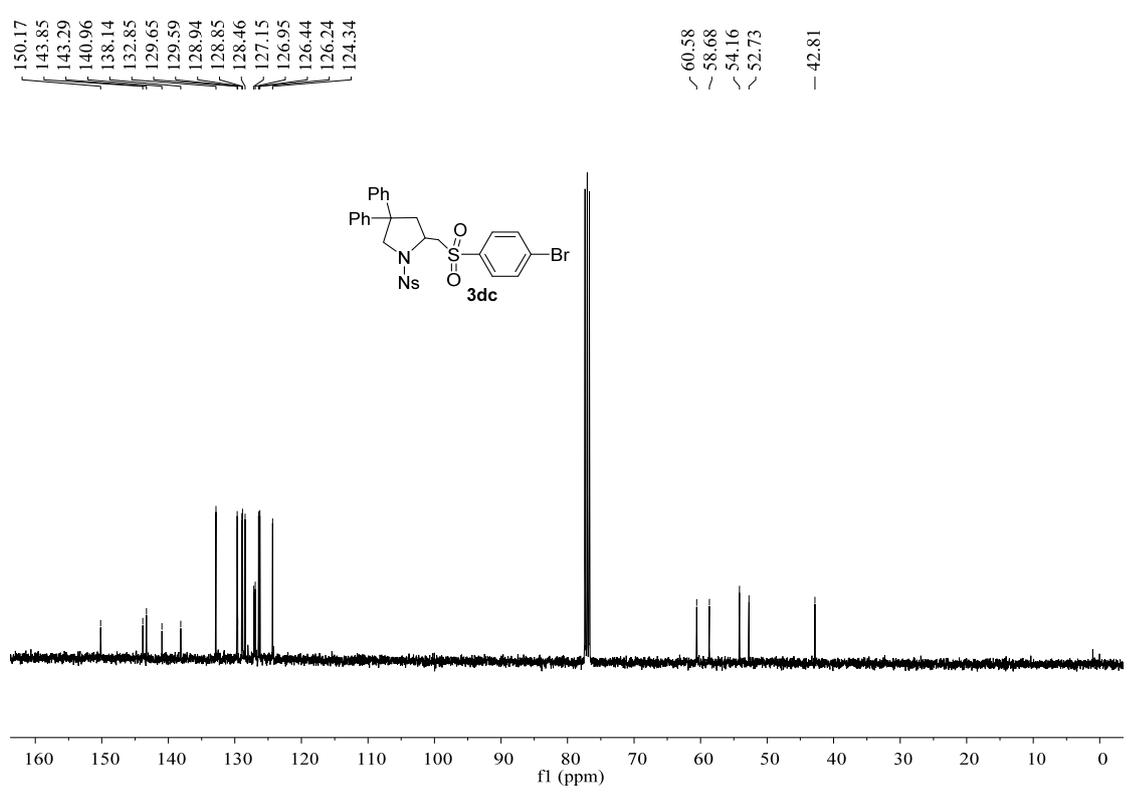
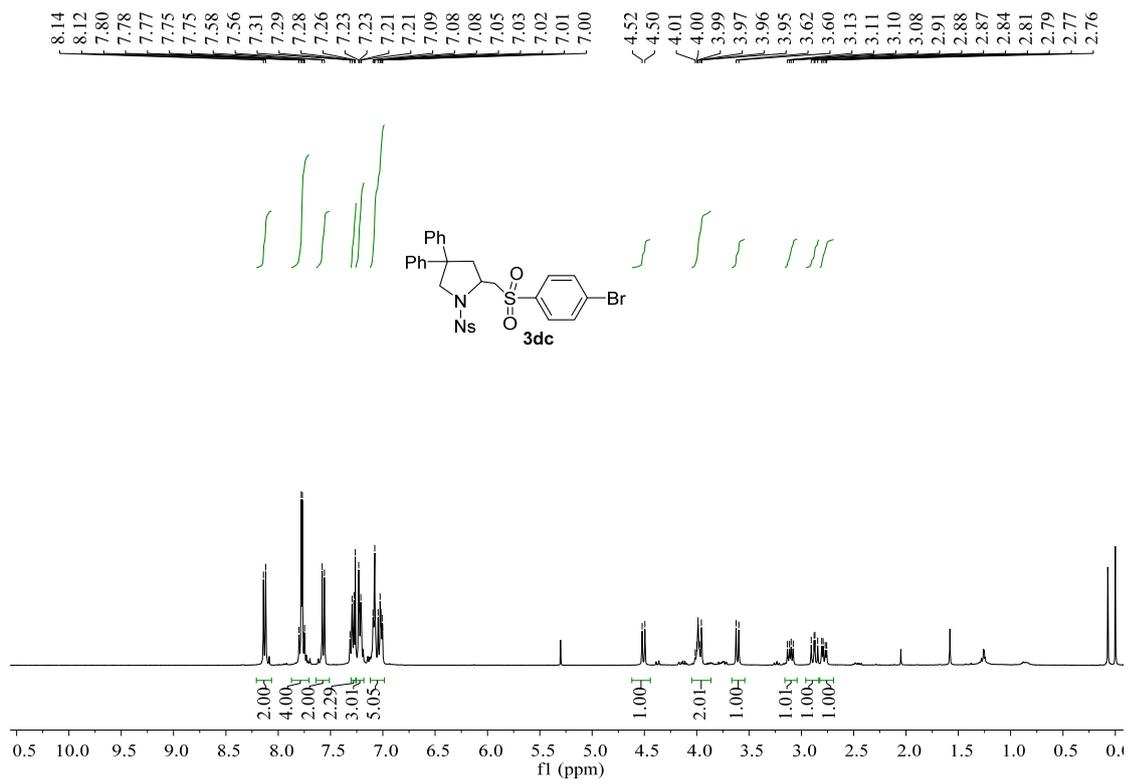


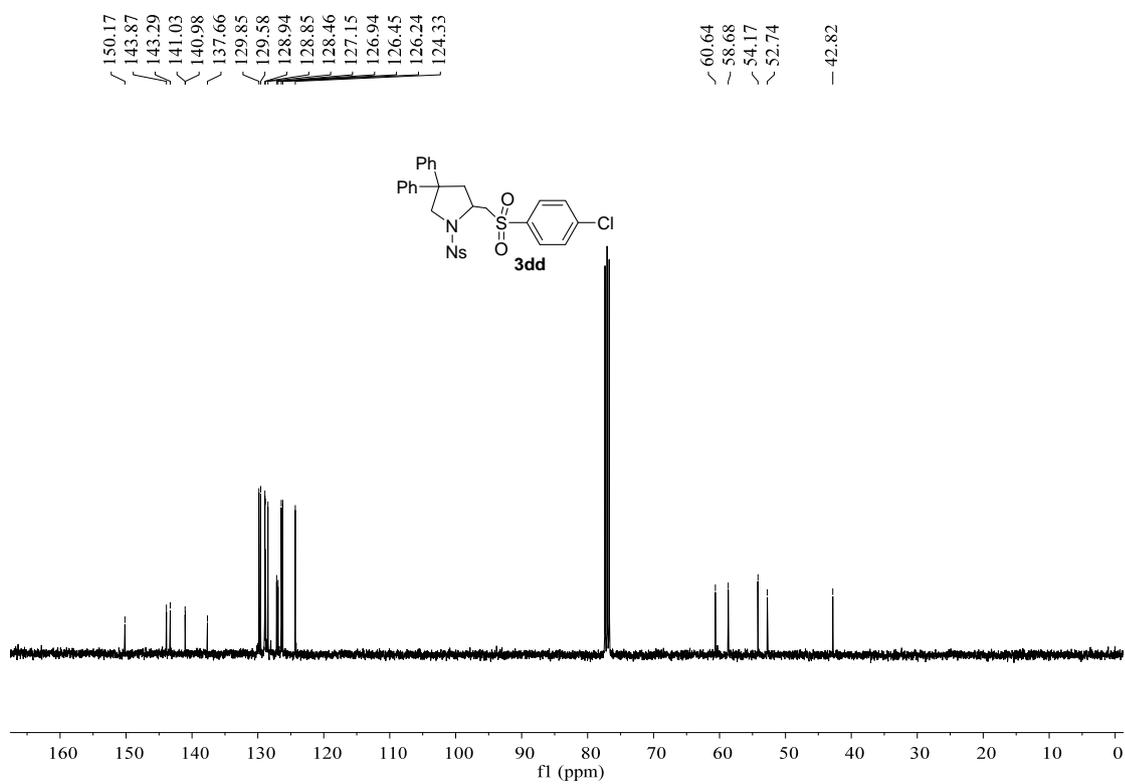
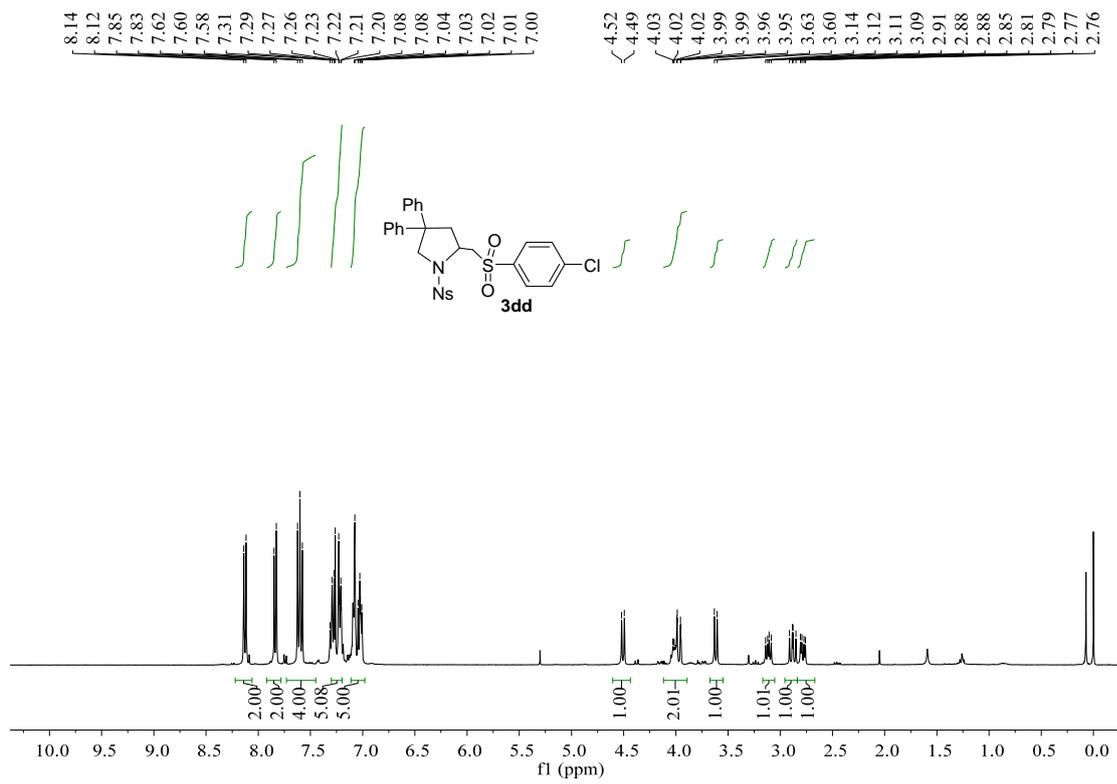


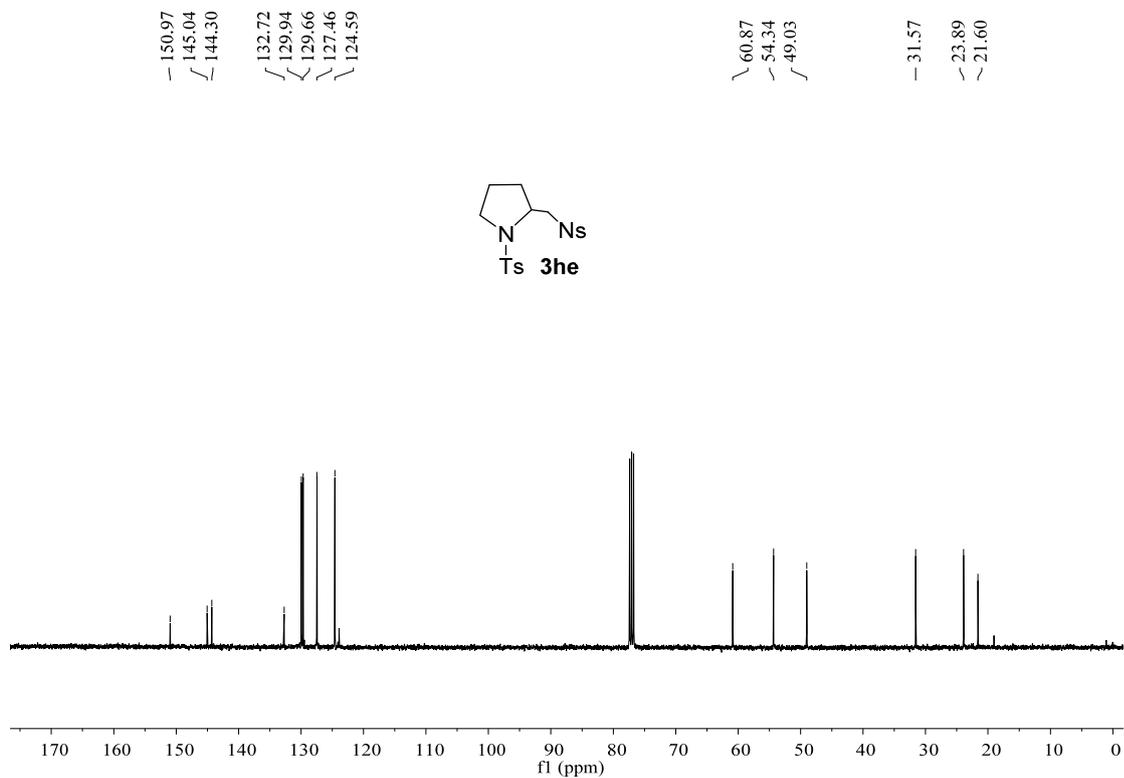
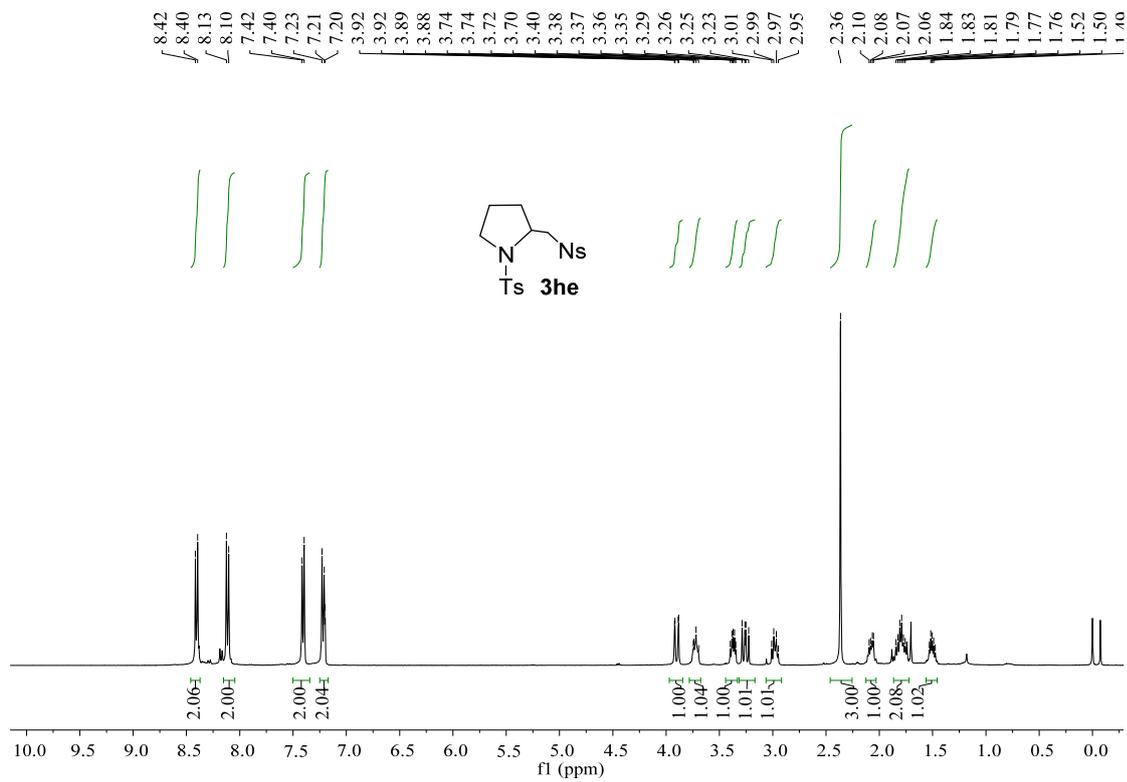


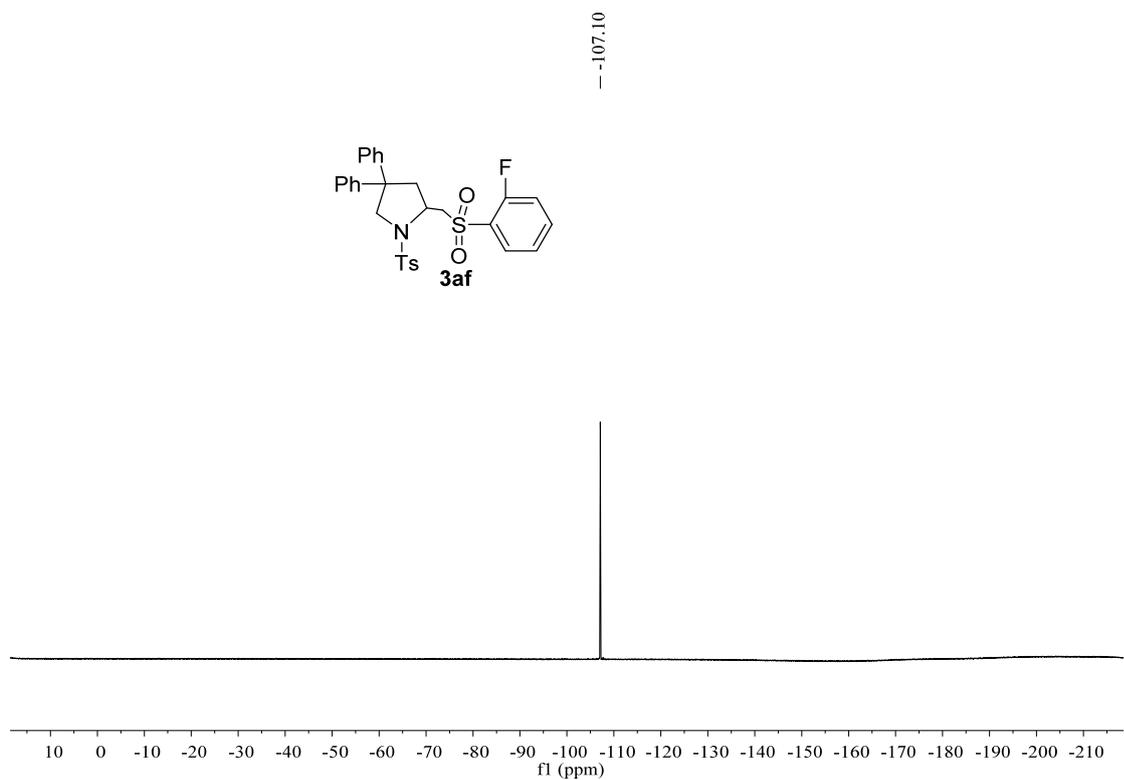
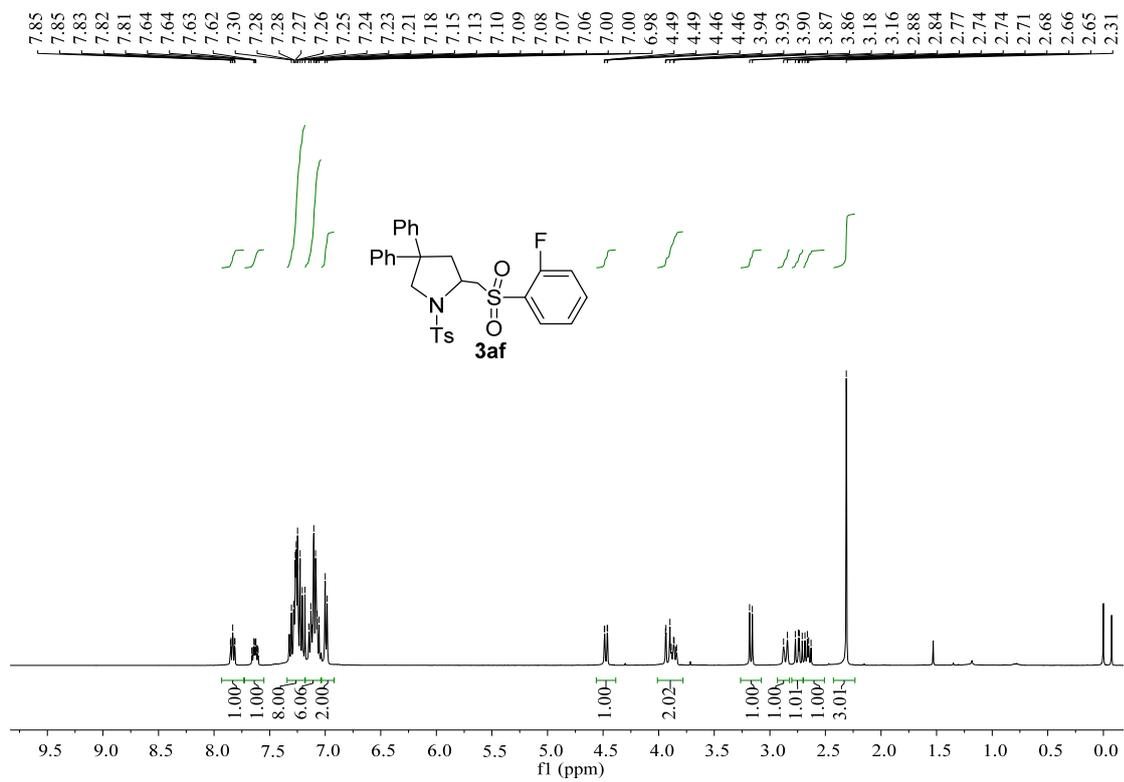


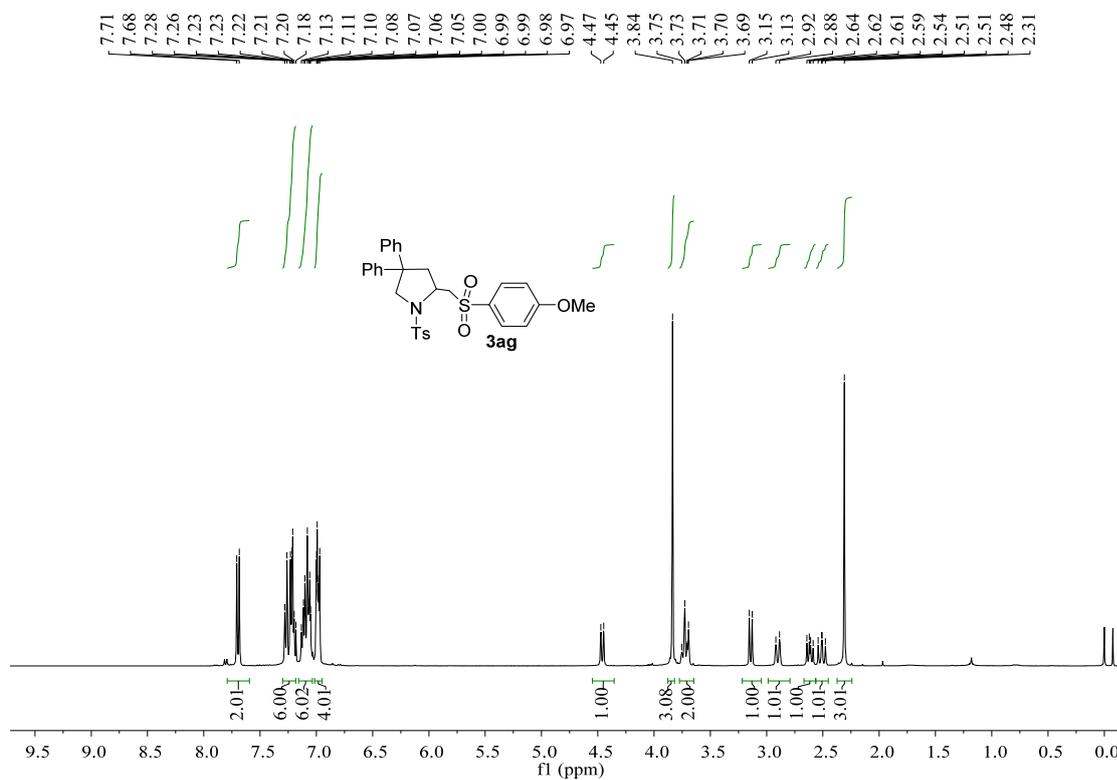
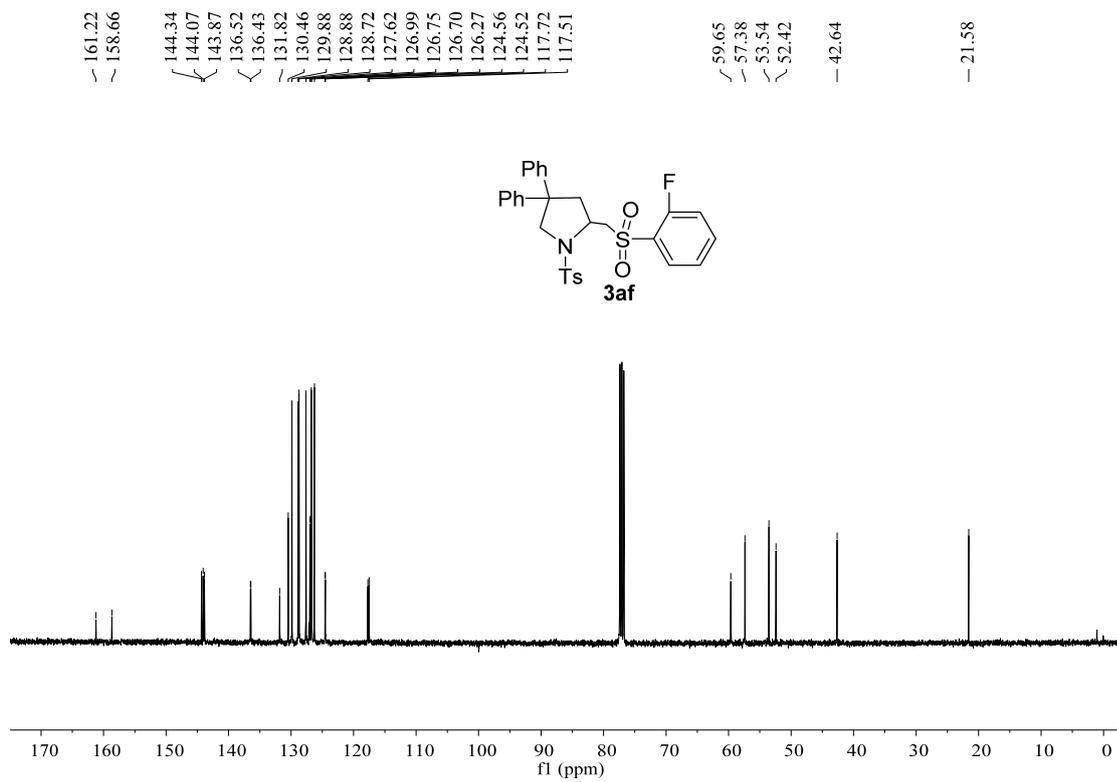




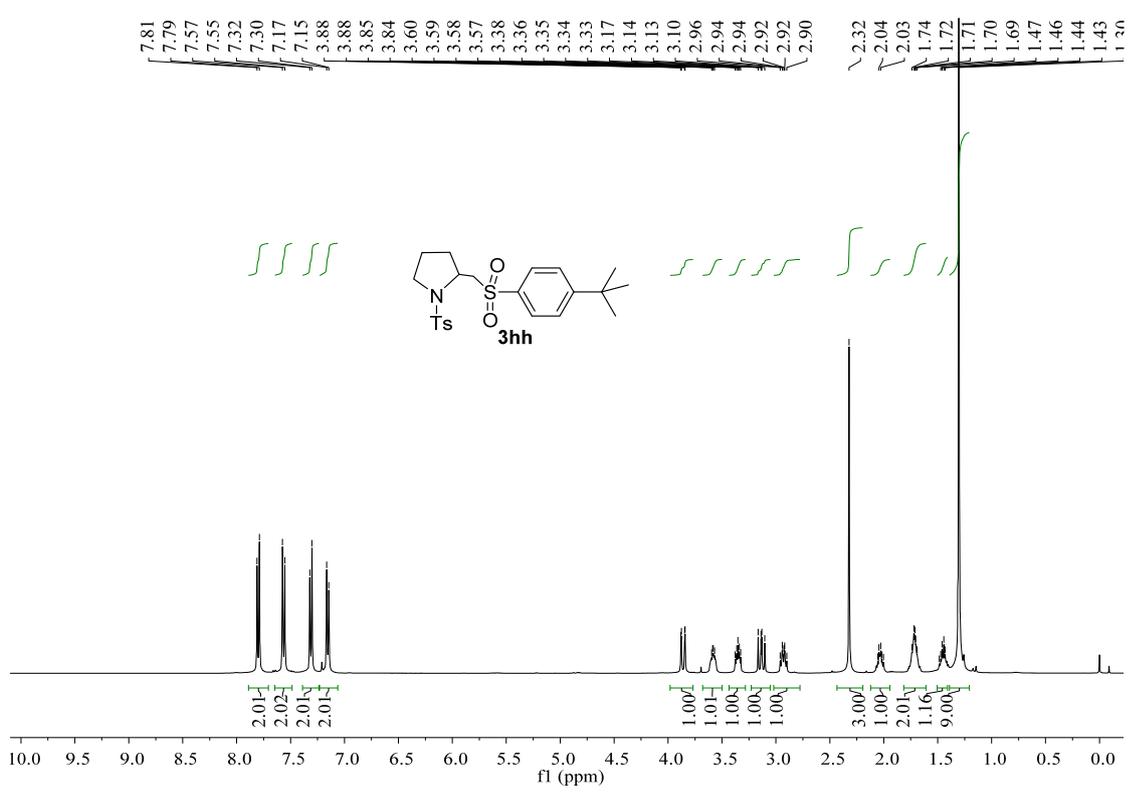
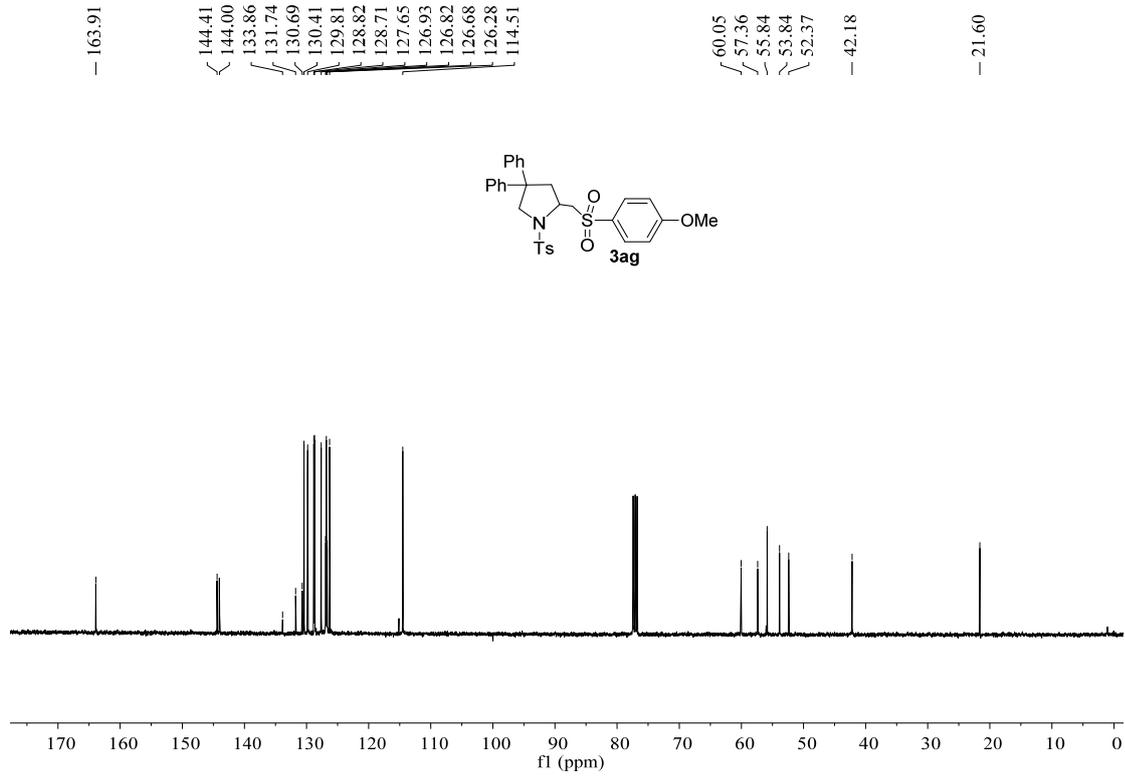


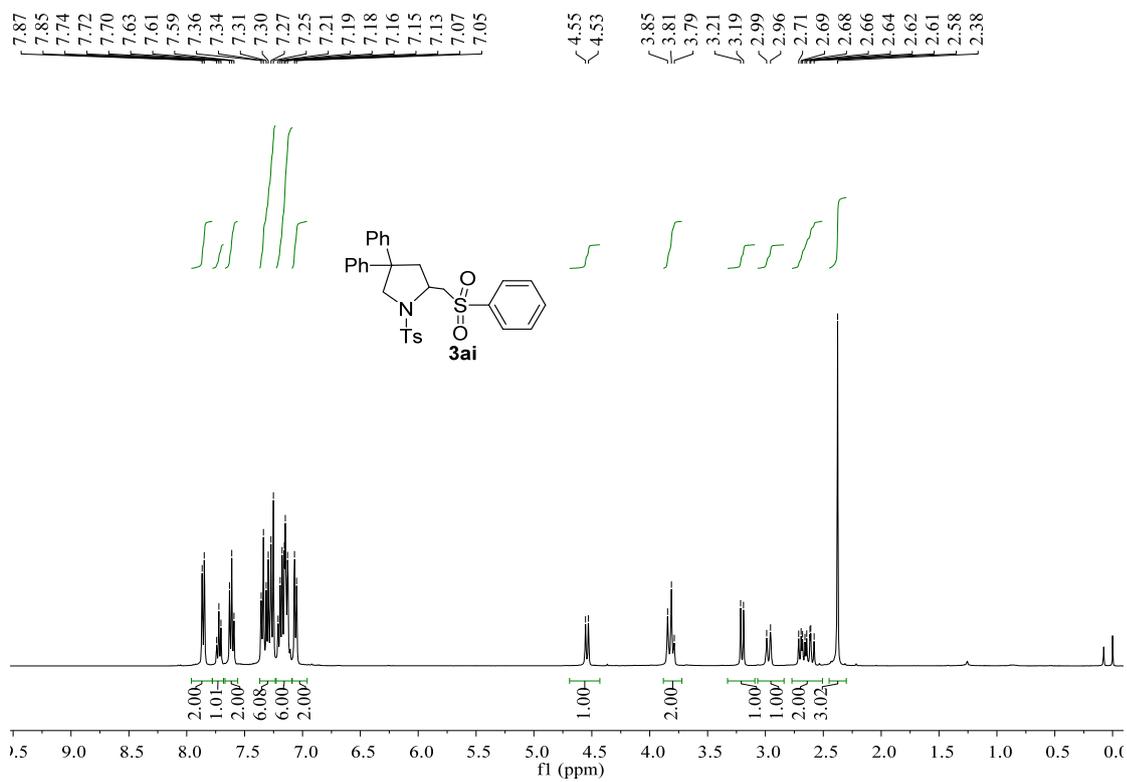
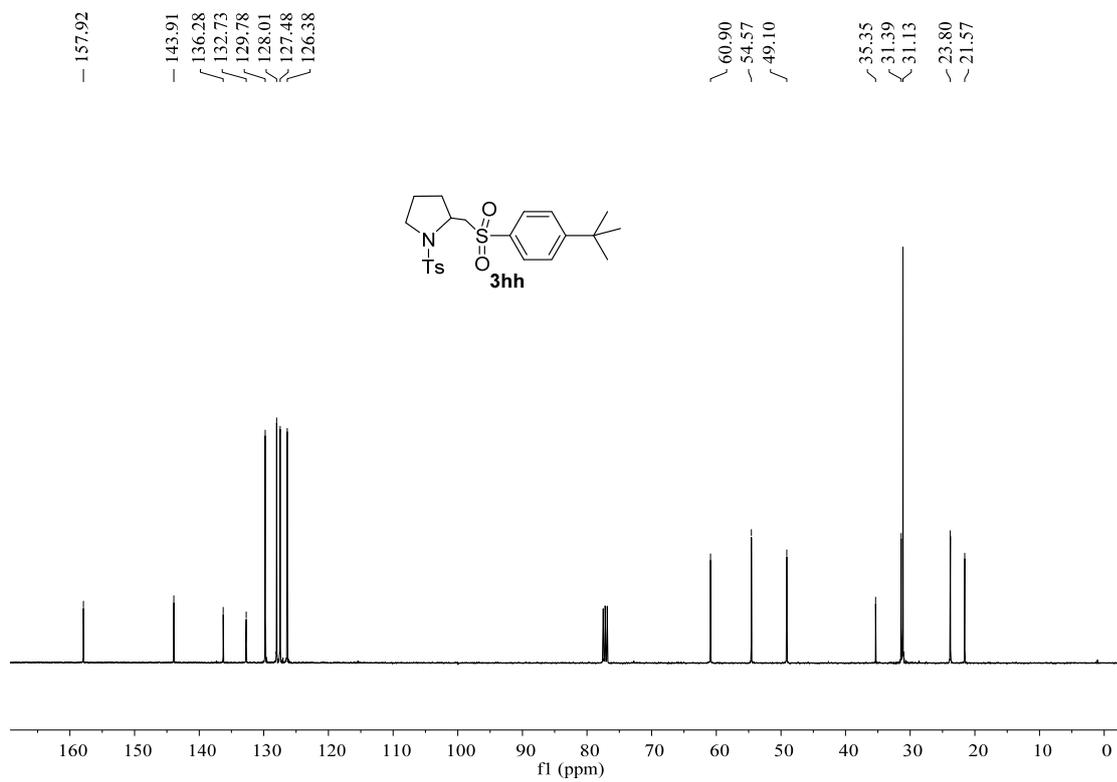


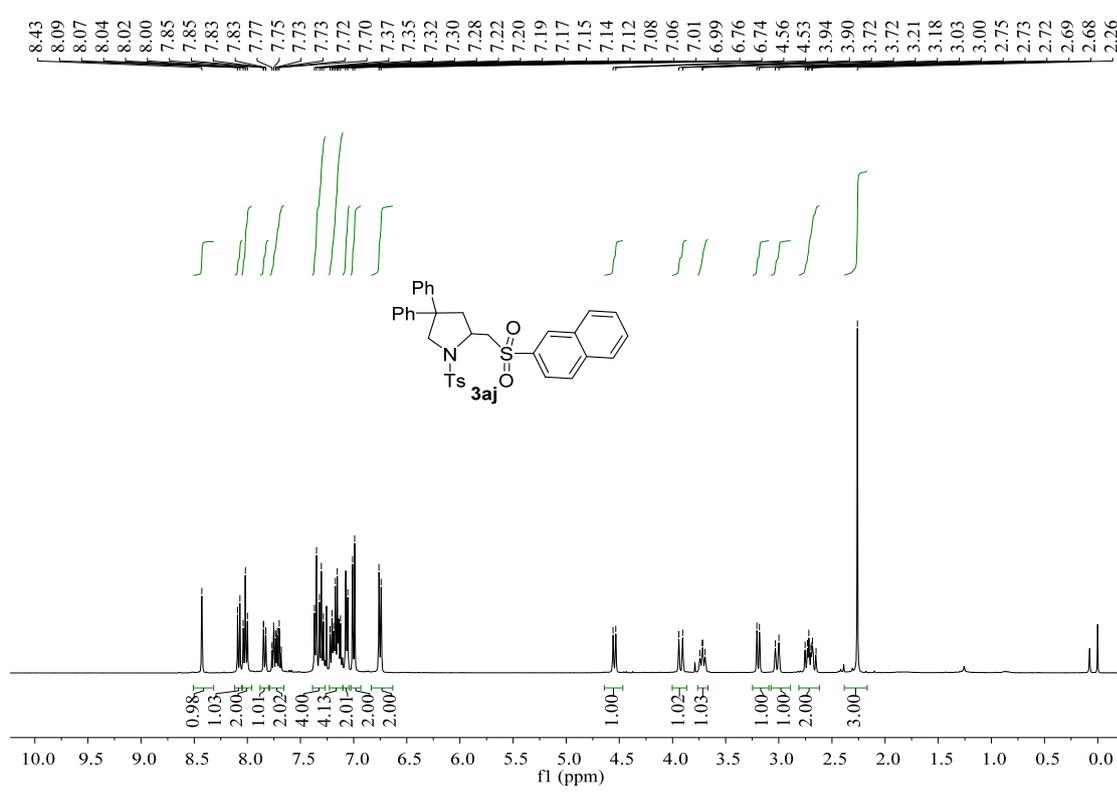
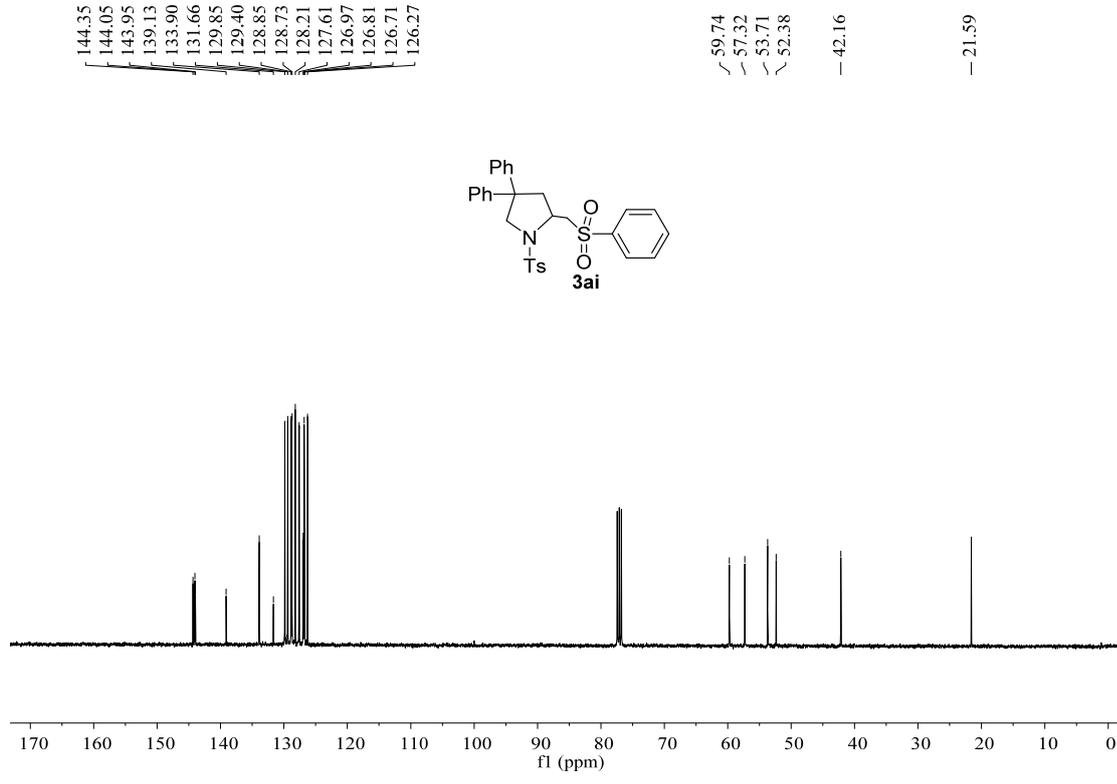


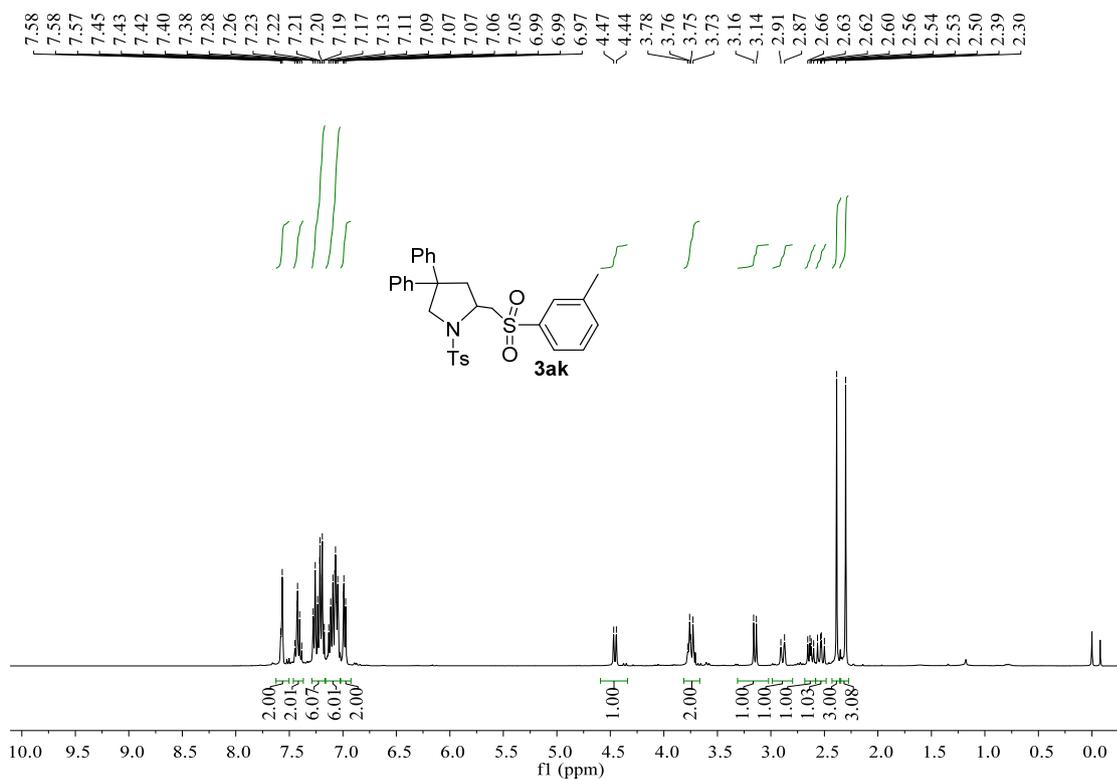
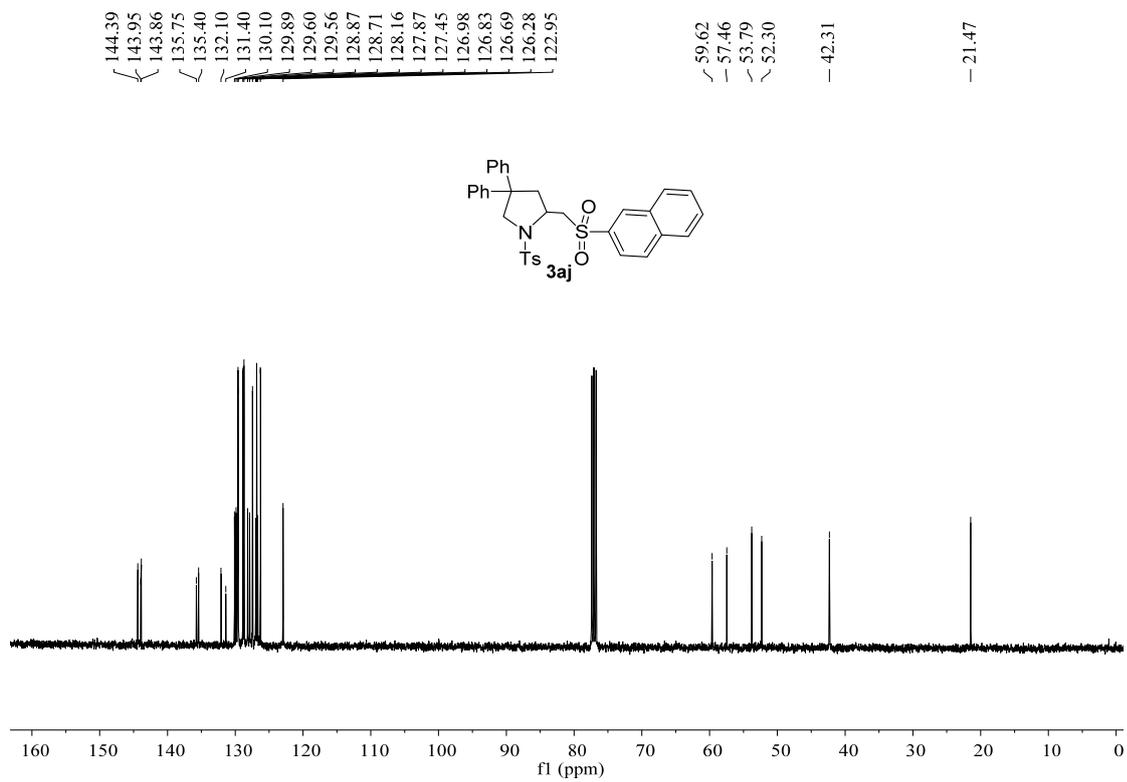


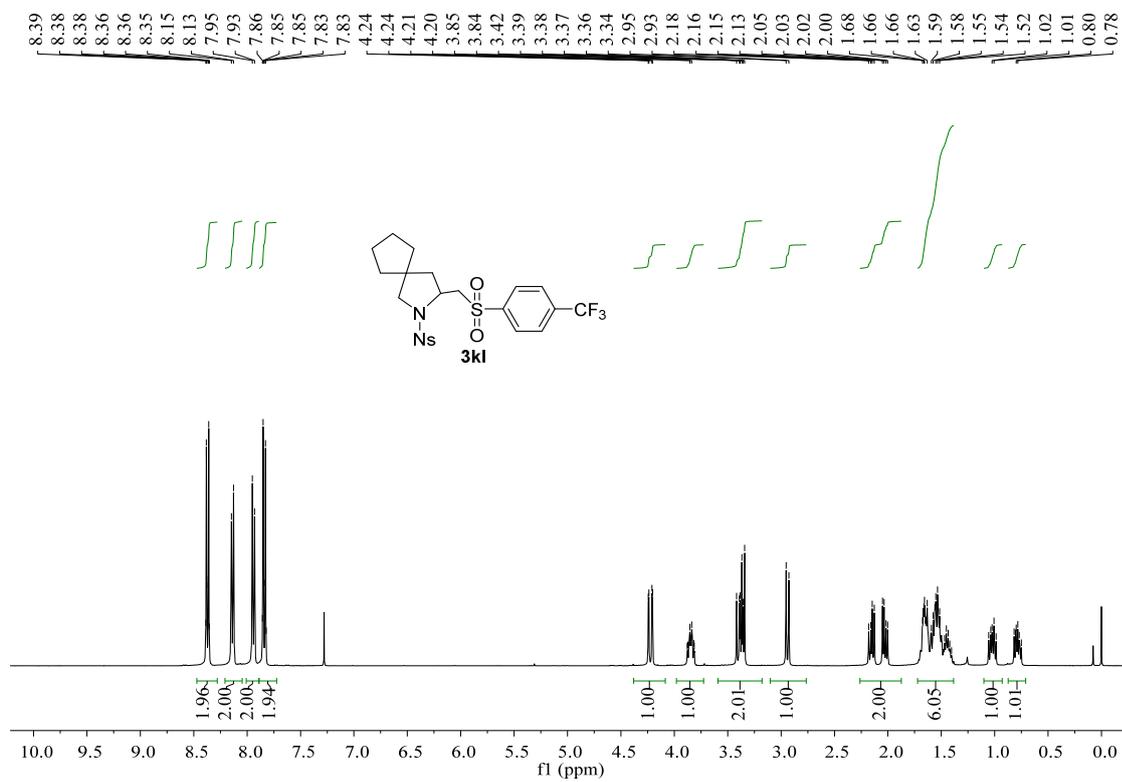
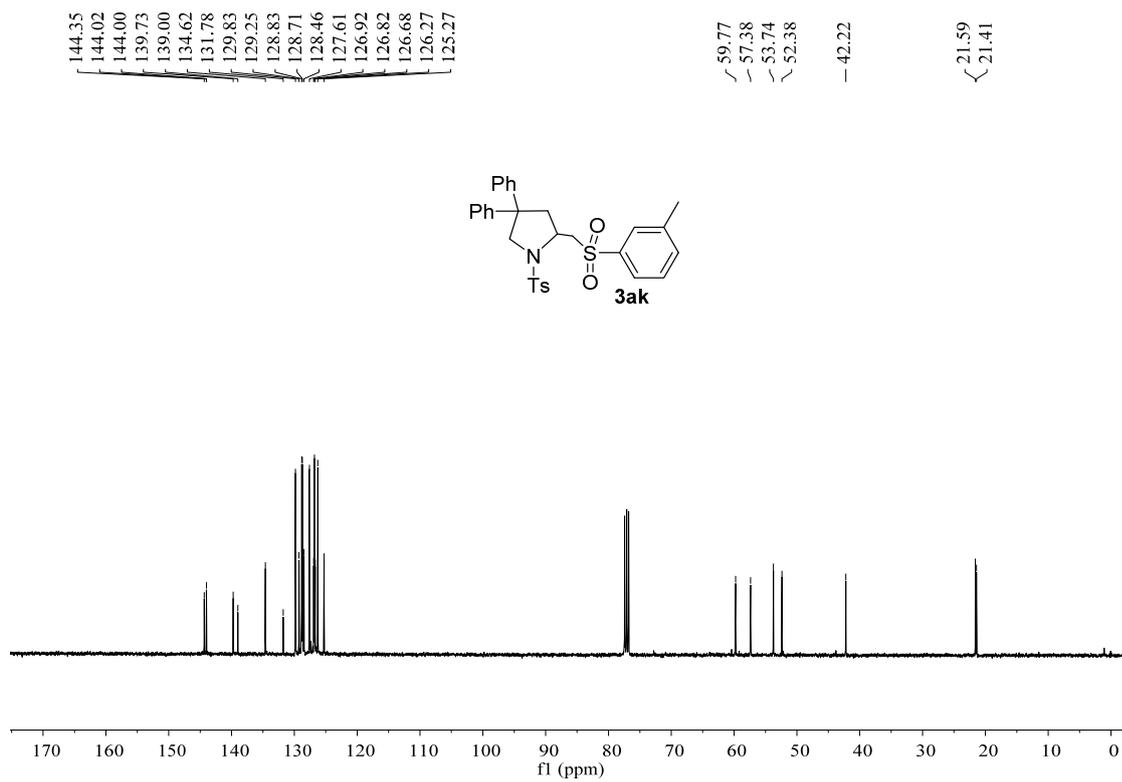


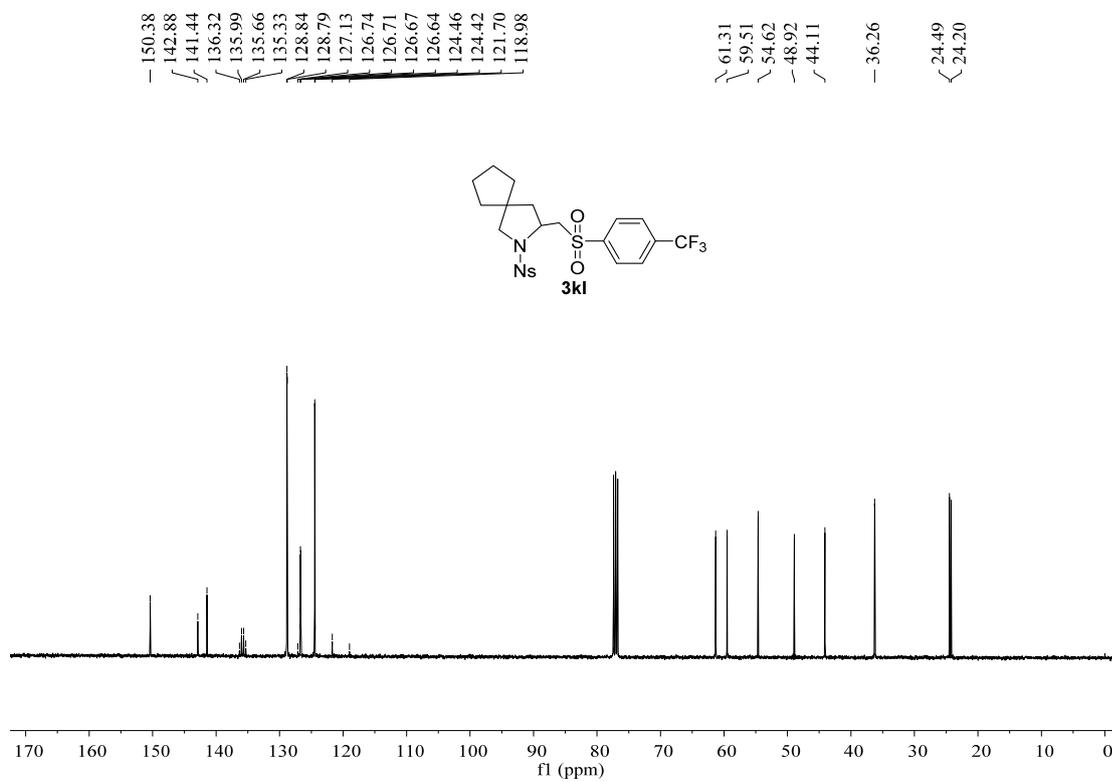
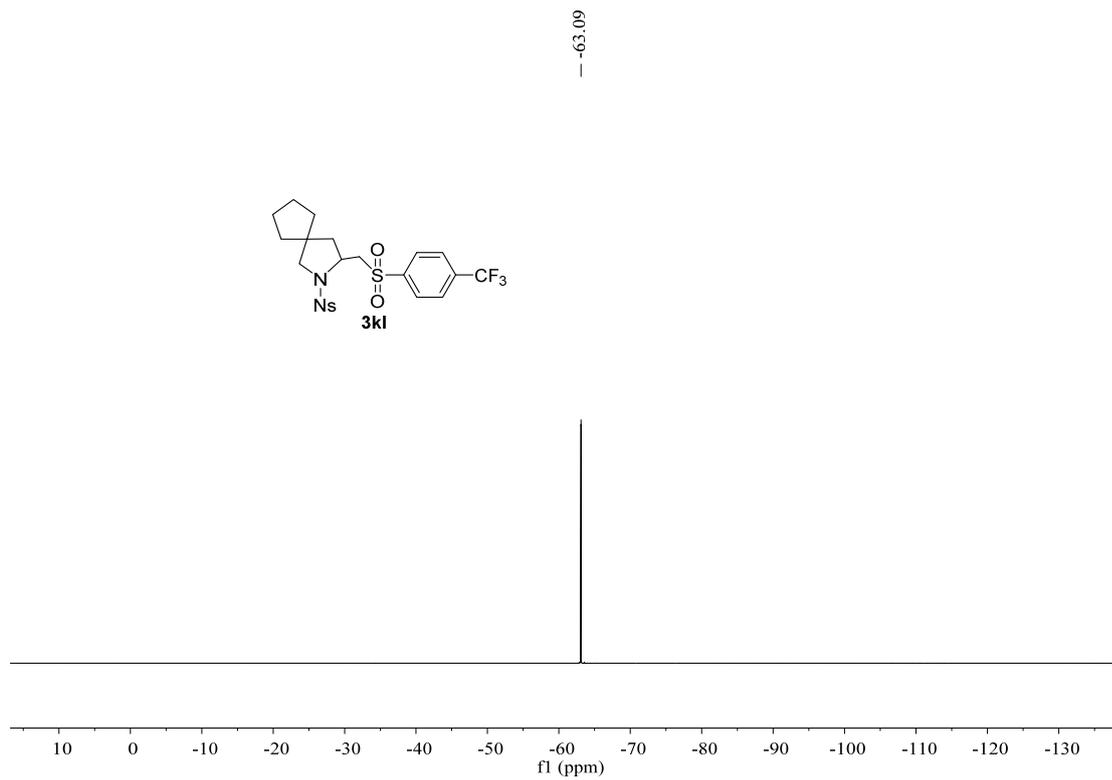


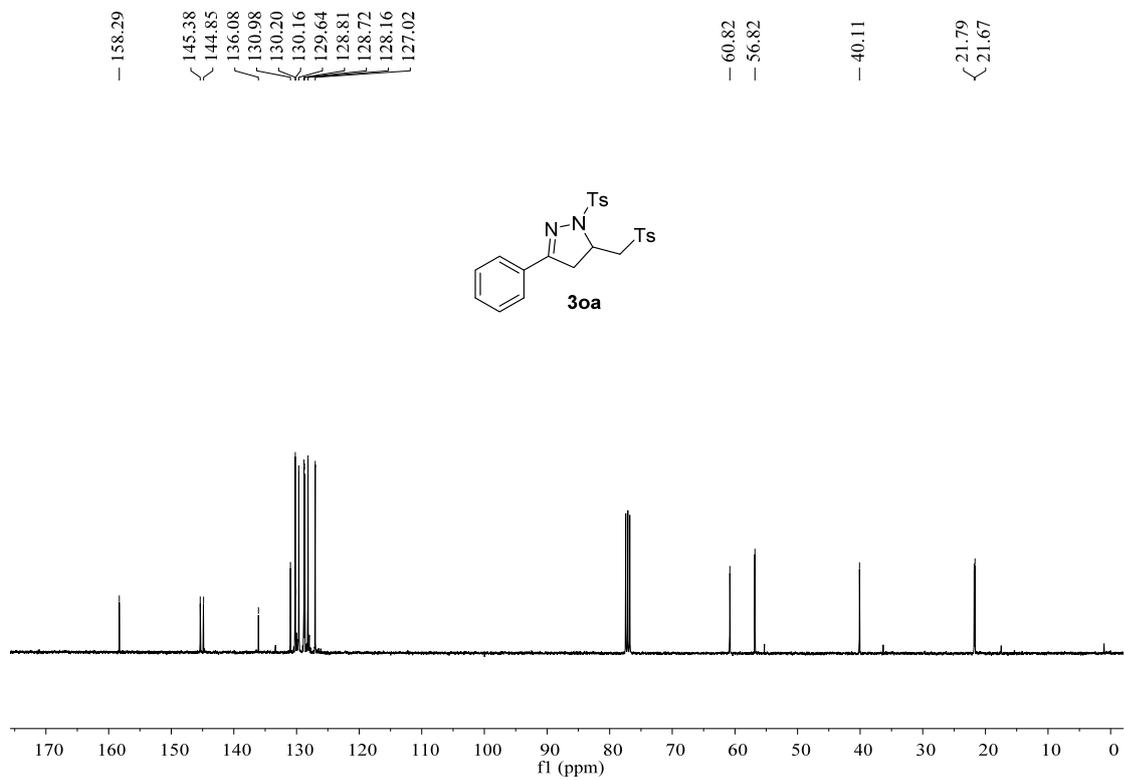
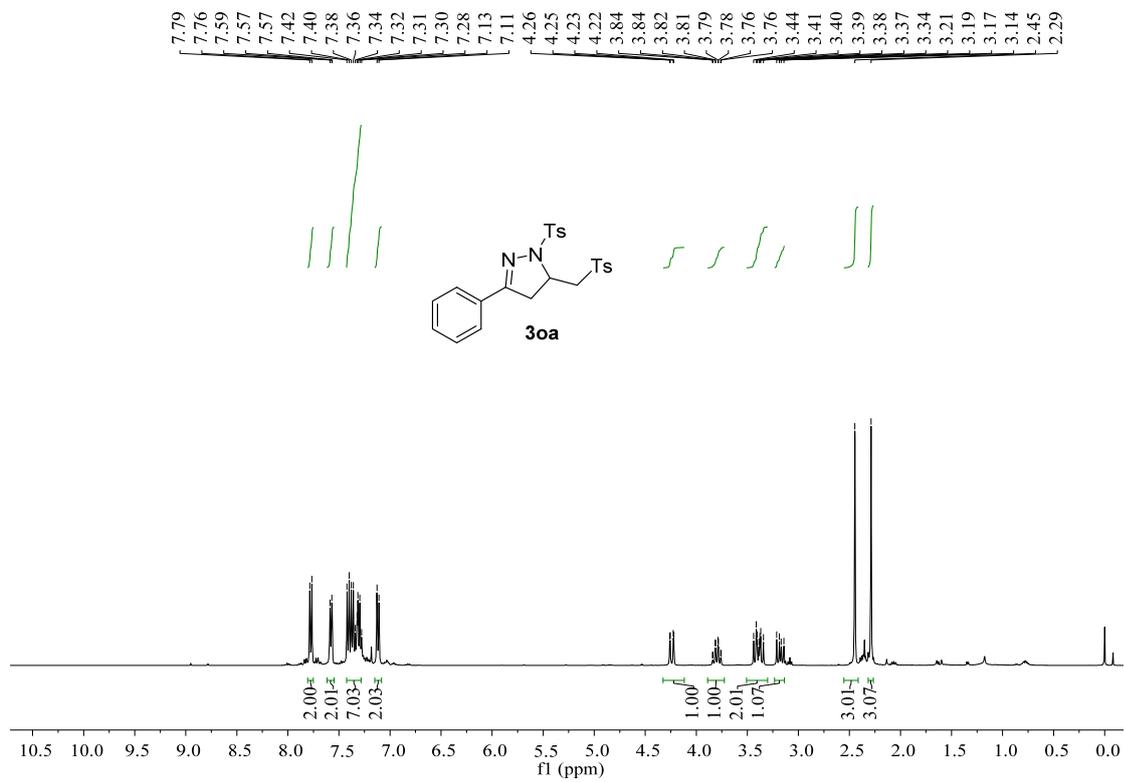


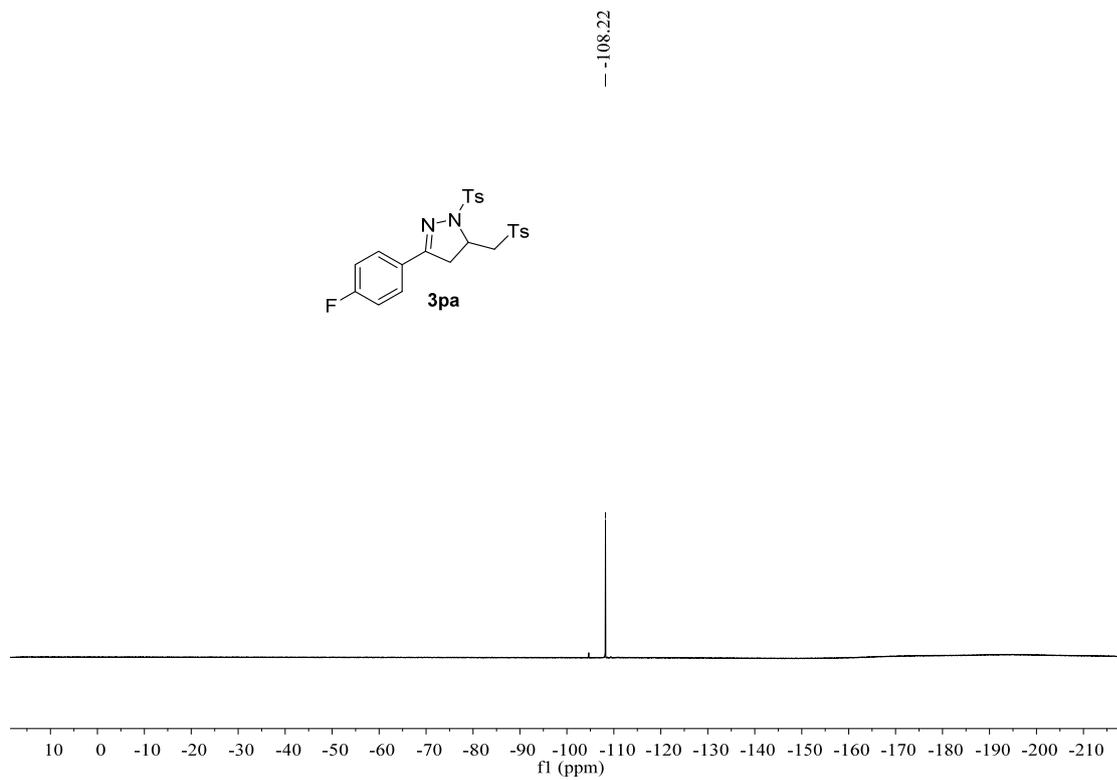
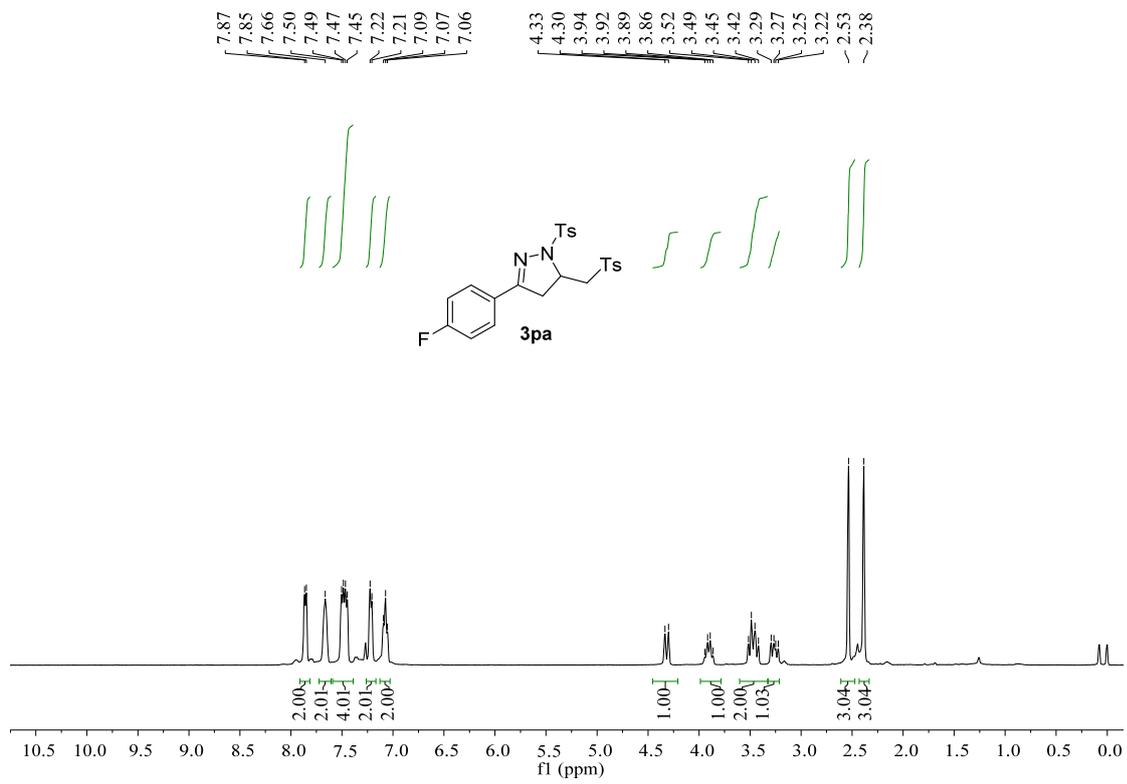




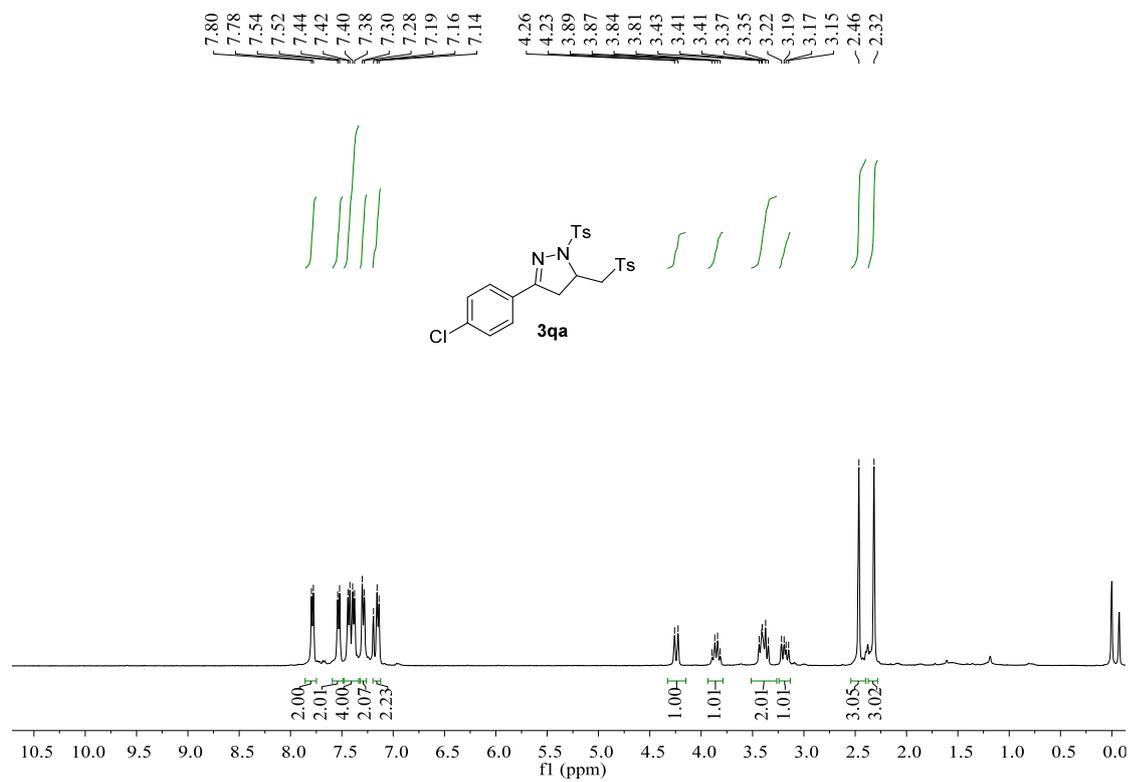
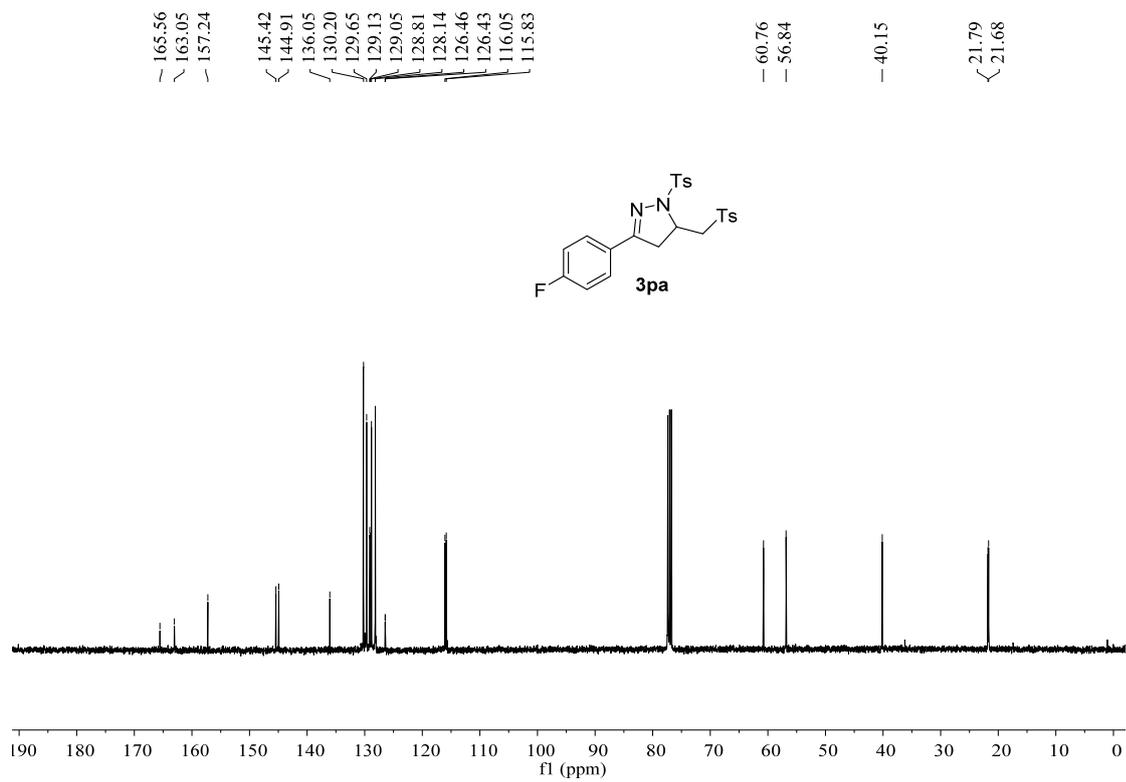


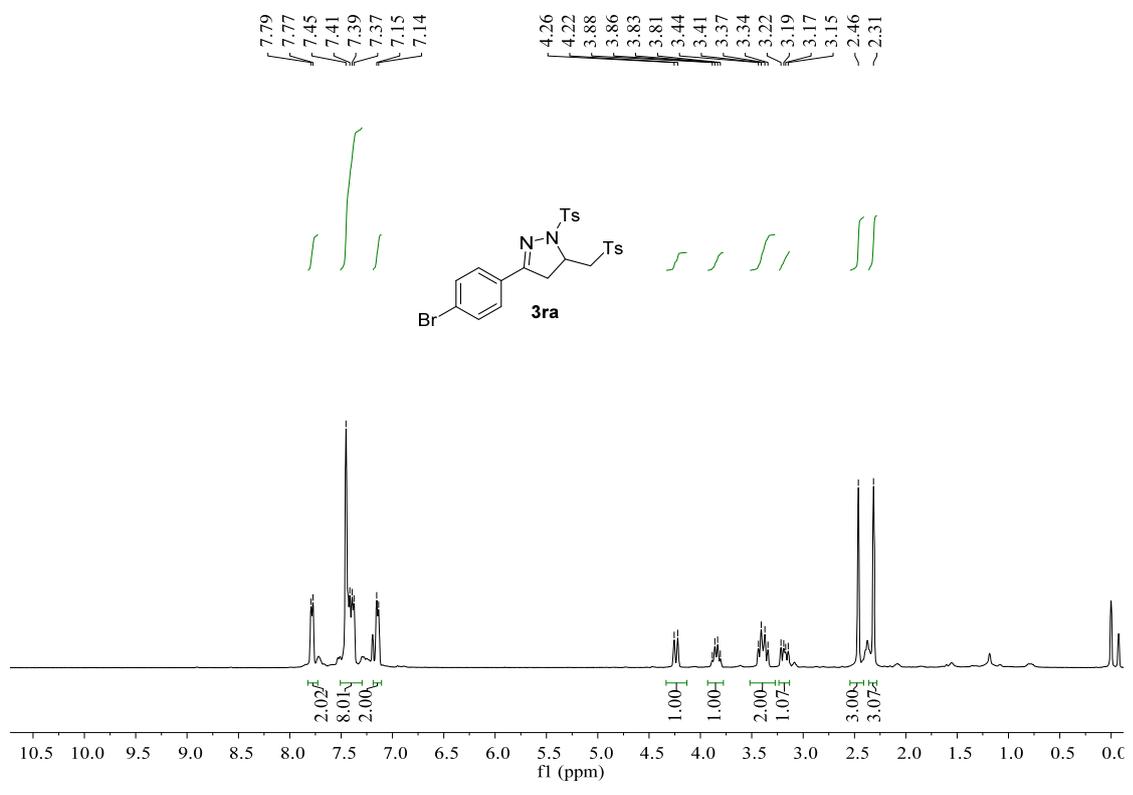
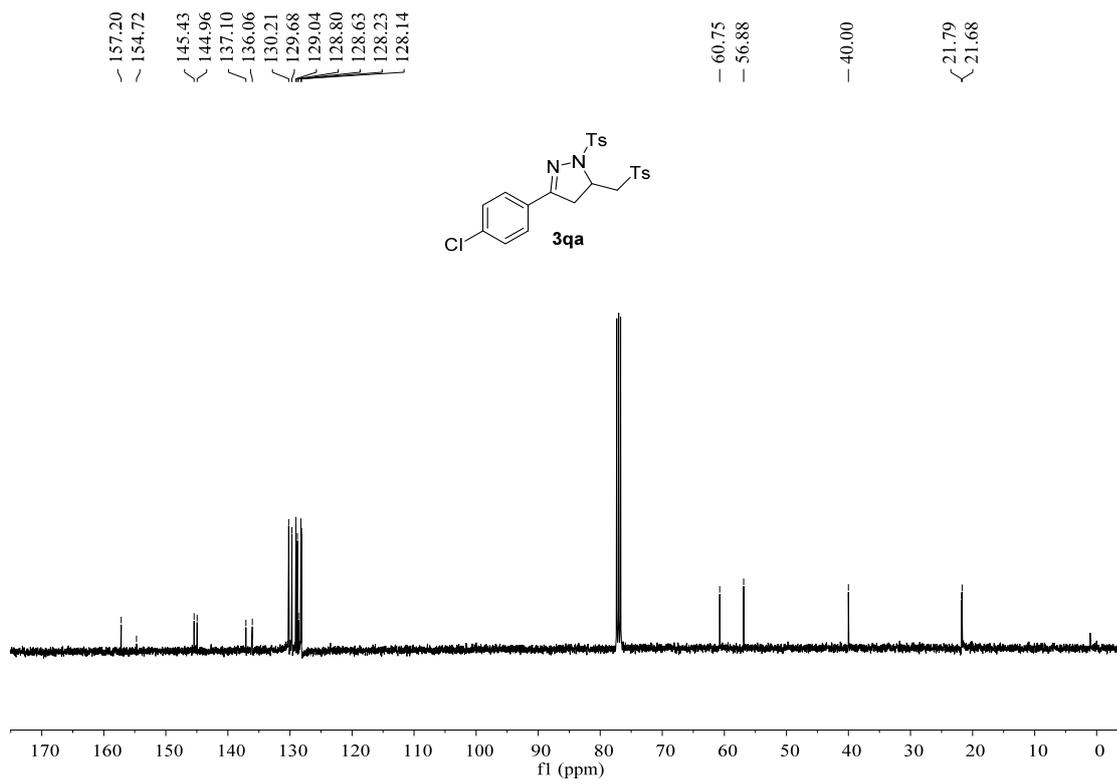


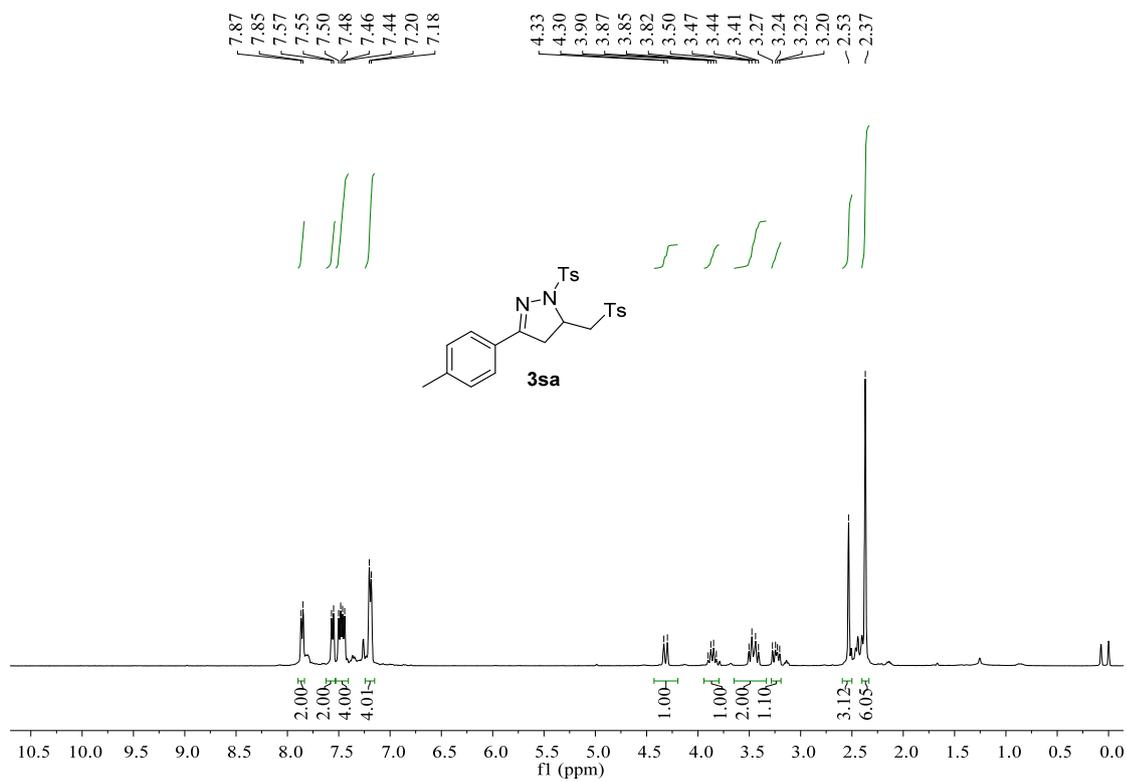
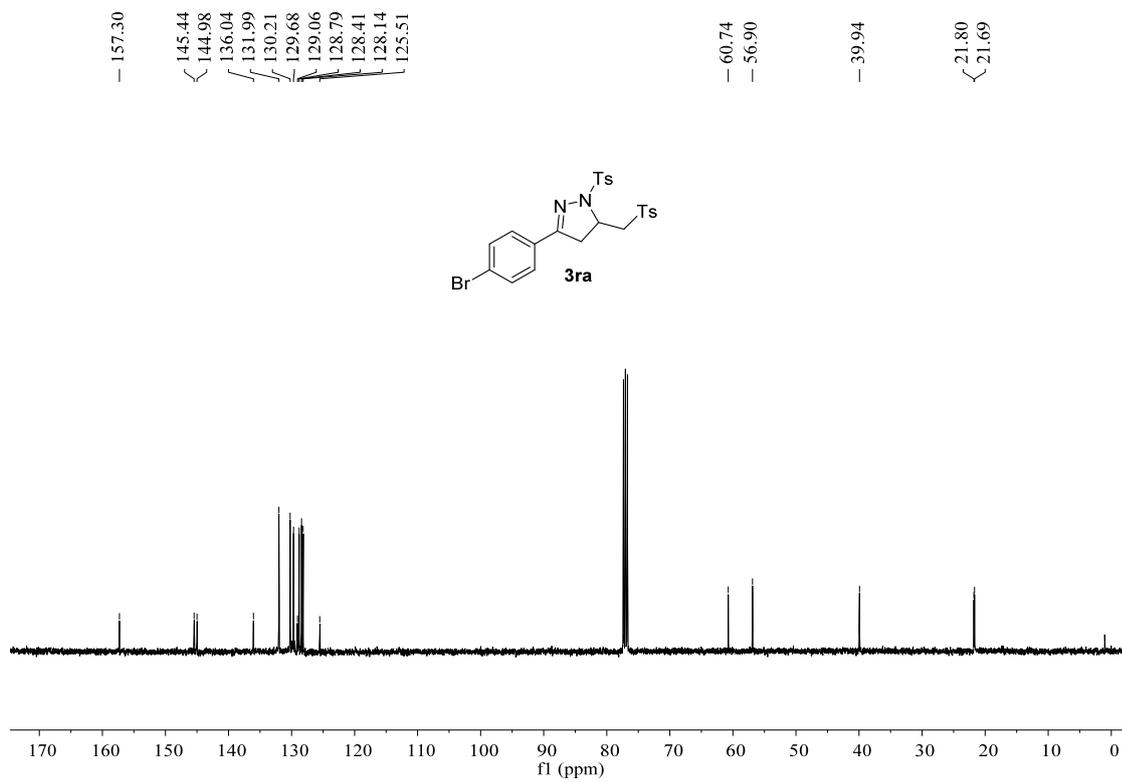


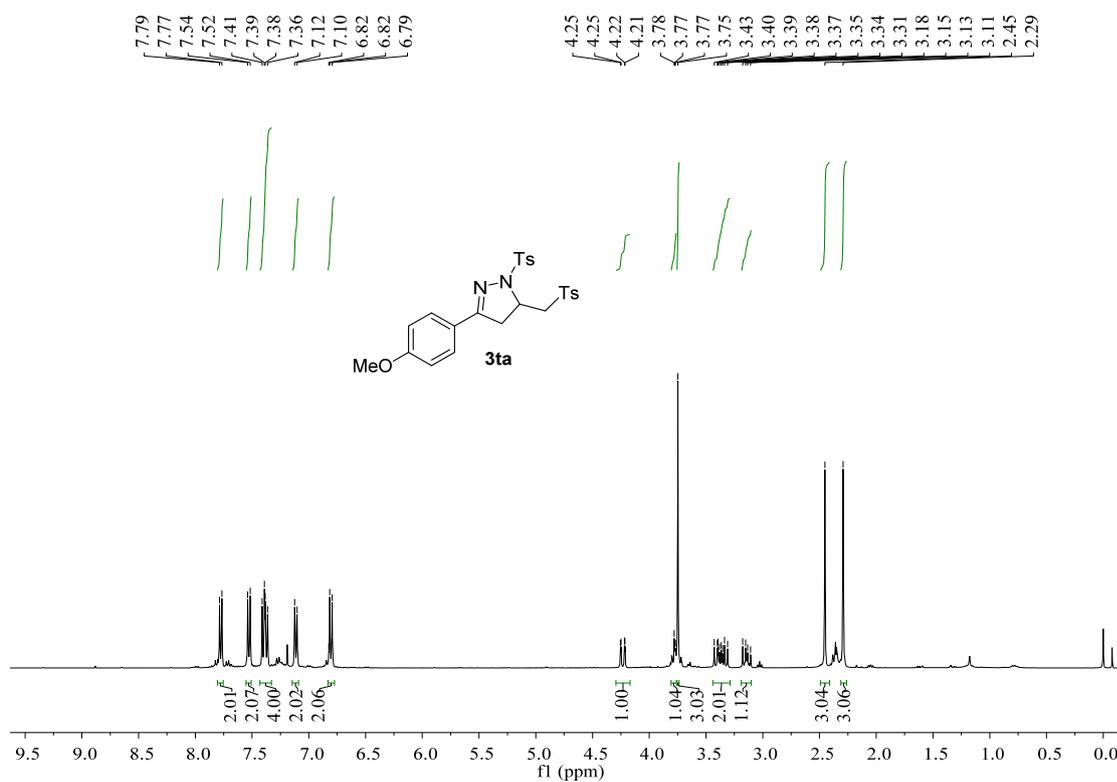
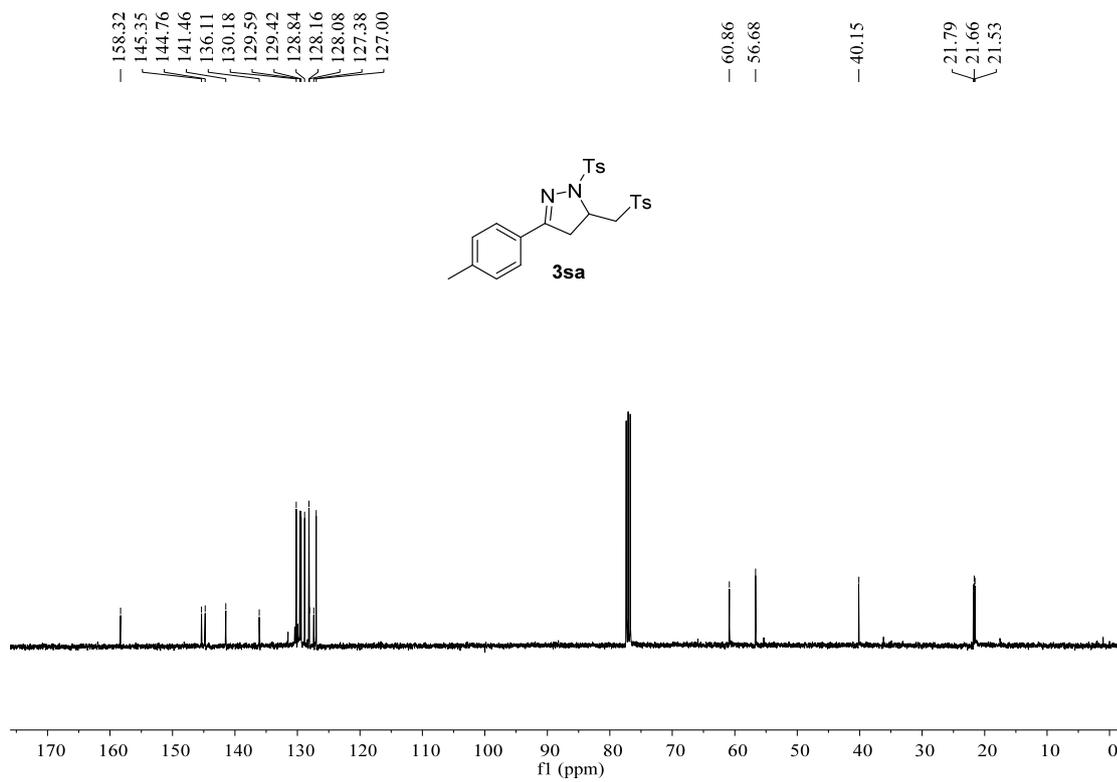




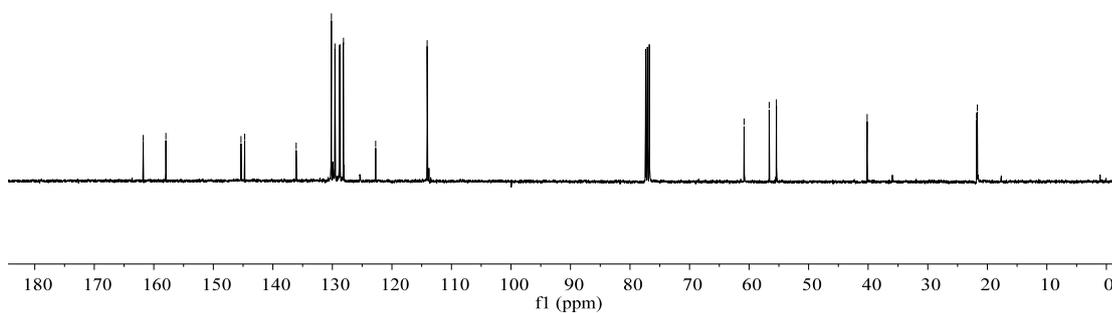
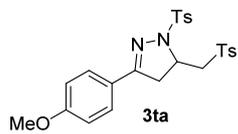




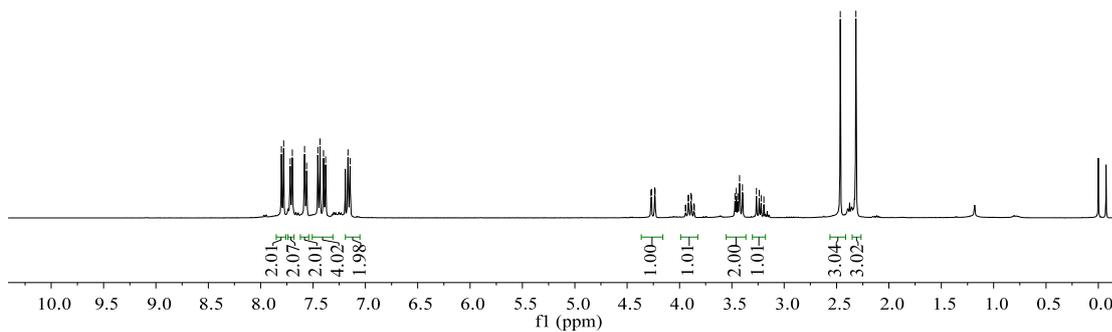
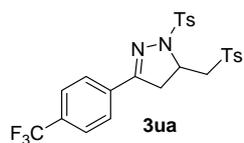


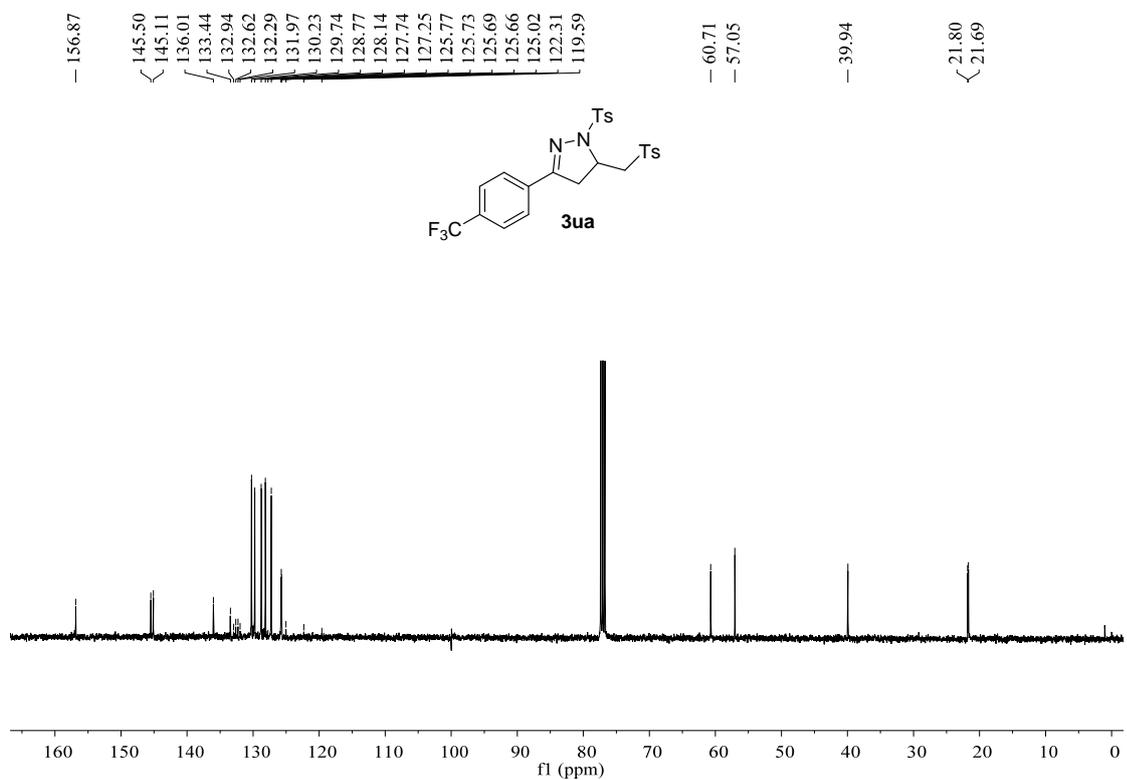
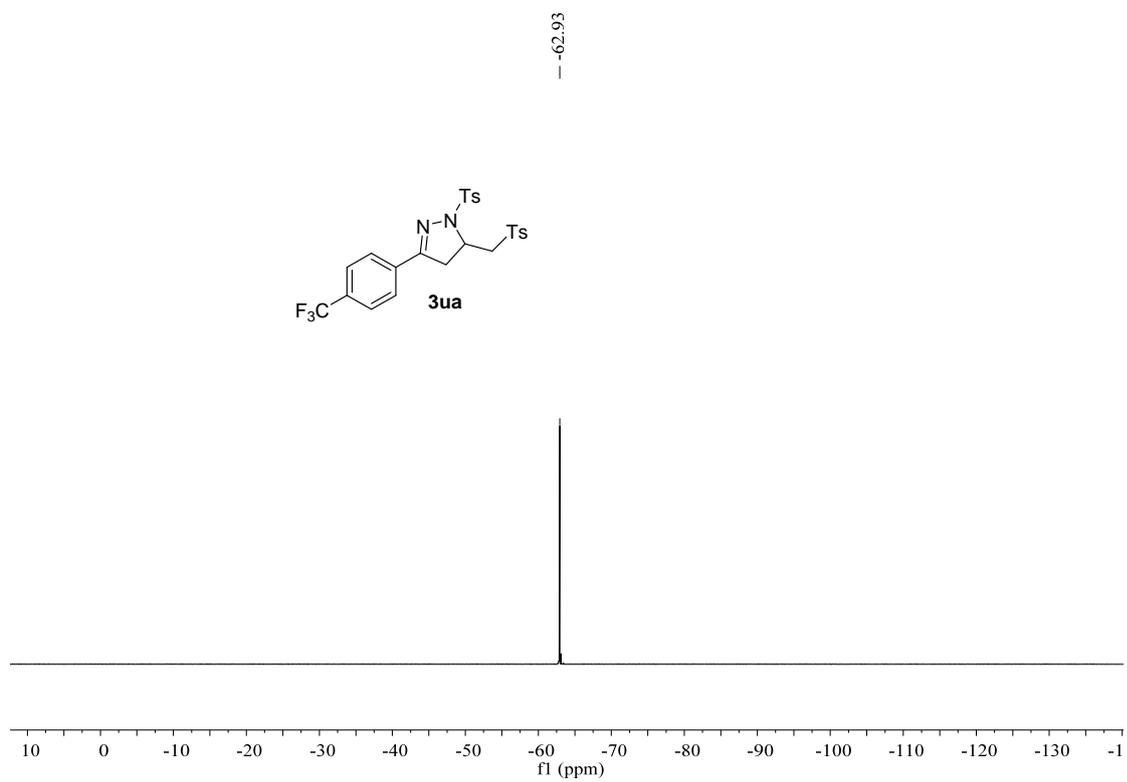


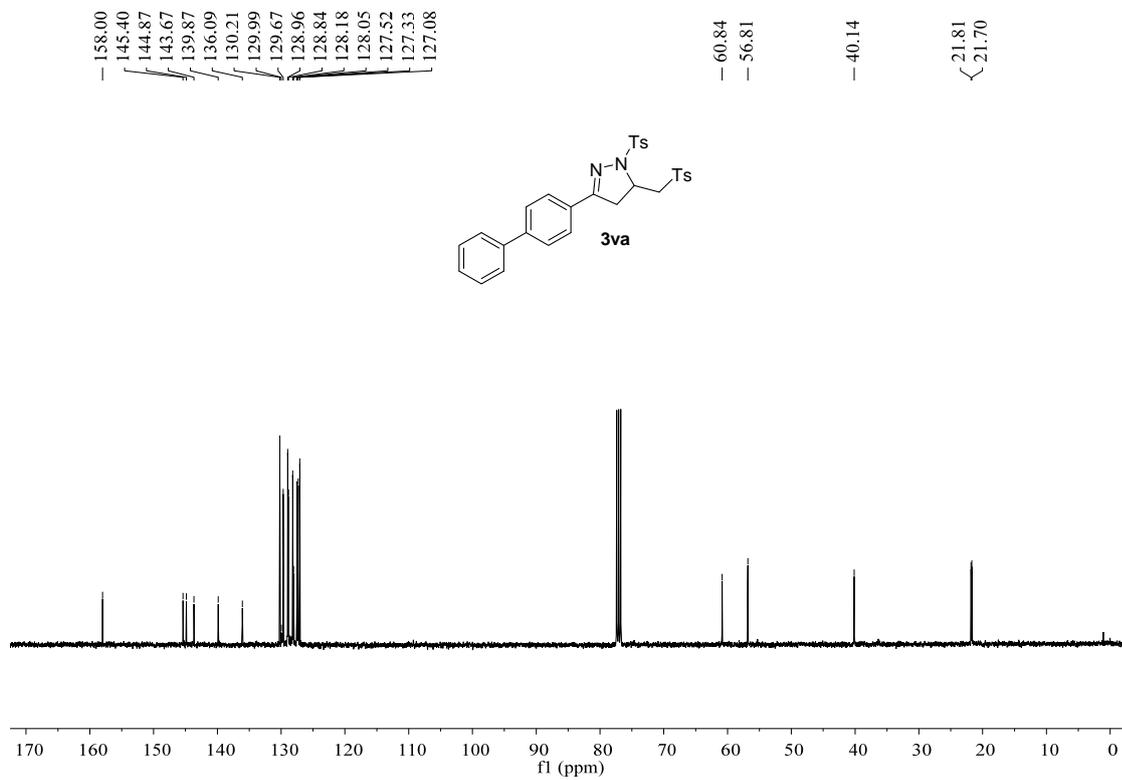
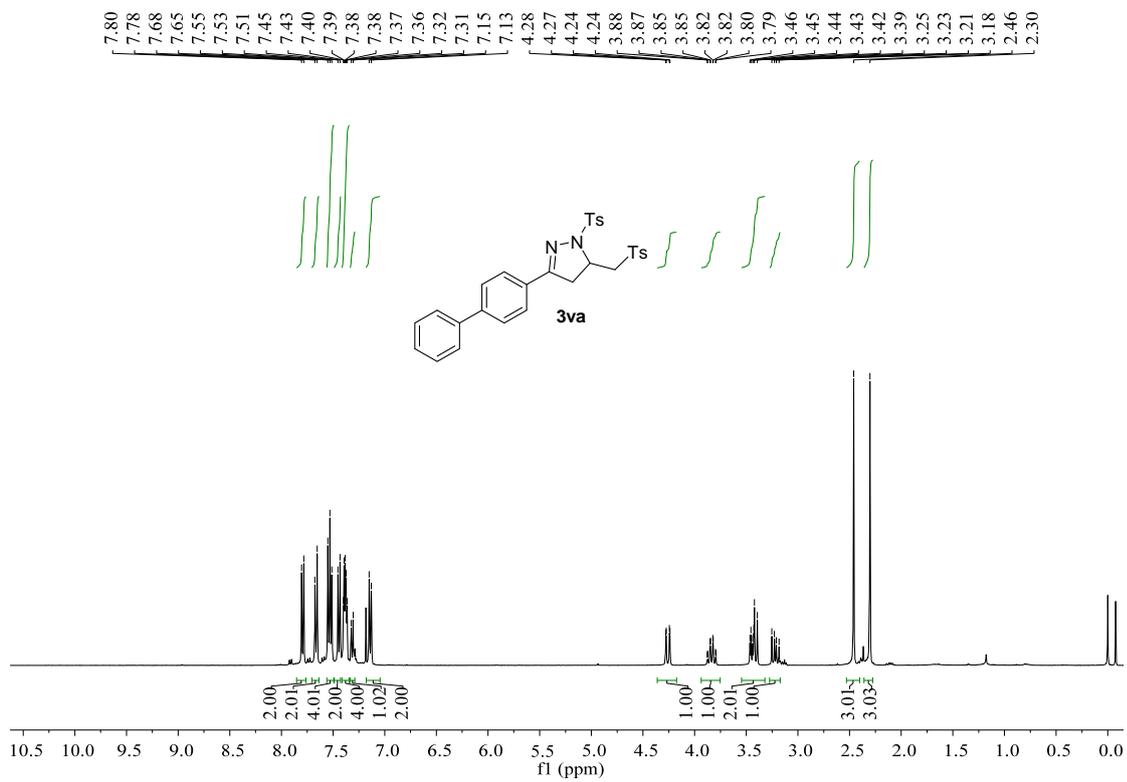
— 161.78  
 — 157.98  
 < 145.35  
 < 144.73  
 < 136.09  
 < 130.18  
 < 129.57  
 < 128.85  
 < 128.70  
 < 128.16  
 < 122.73  
 < 114.09  
 < 60.84  
 < 56.64  
 < 55.43  
 — 40.17  
 < 21.79  
 < 21.66



7.80  
 7.78  
 7.72  
 7.70  
 7.58  
 7.56  
 7.45  
 7.43  
 7.40  
 7.38  
 7.17  
 7.15  
 4.27  
 4.27  
 4.24  
 4.23  
 3.94  
 3.94  
 3.92  
 3.91  
 3.89  
 3.89  
 3.86  
 3.86  
 3.47  
 3.46  
 3.46  
 3.44  
 3.43  
 3.40  
 3.27  
 3.24  
 3.22  
 3.19  
 2.46  
 2.32

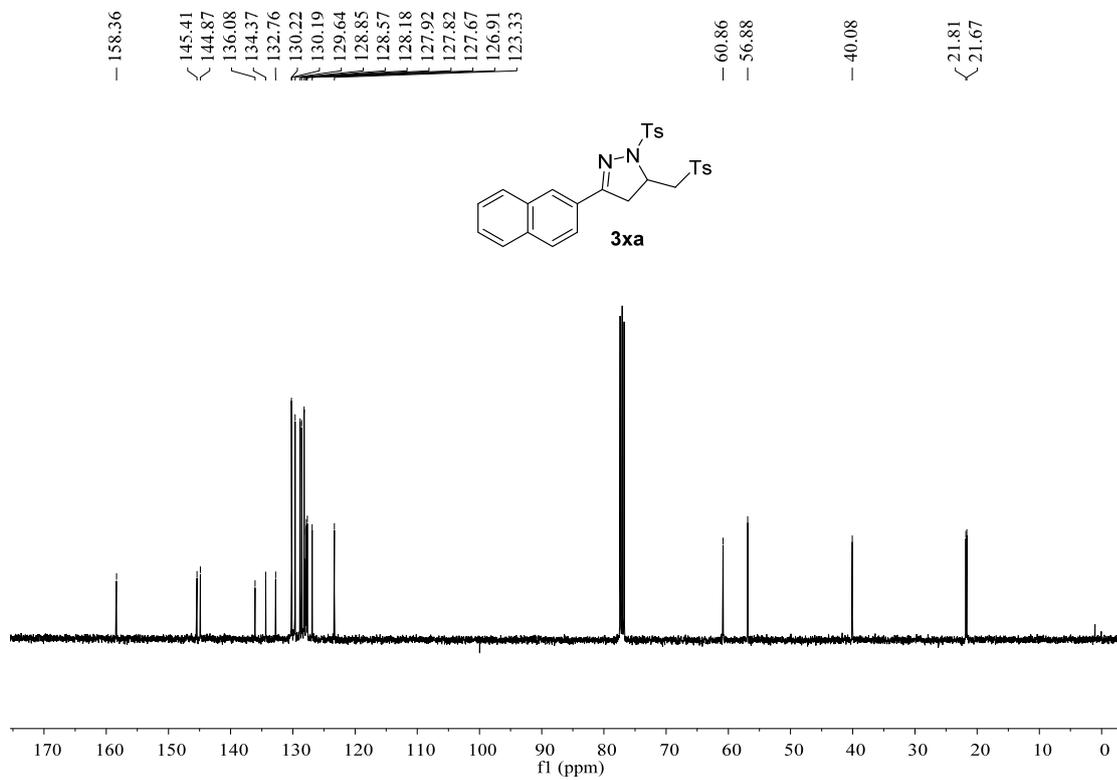
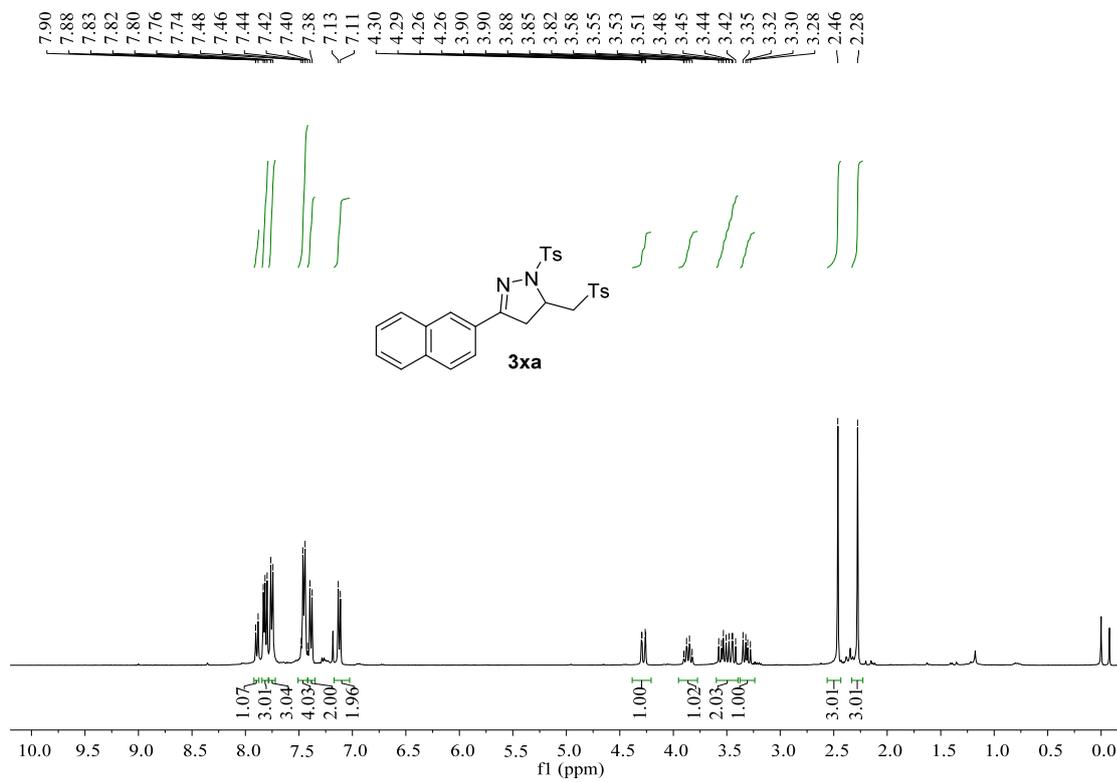


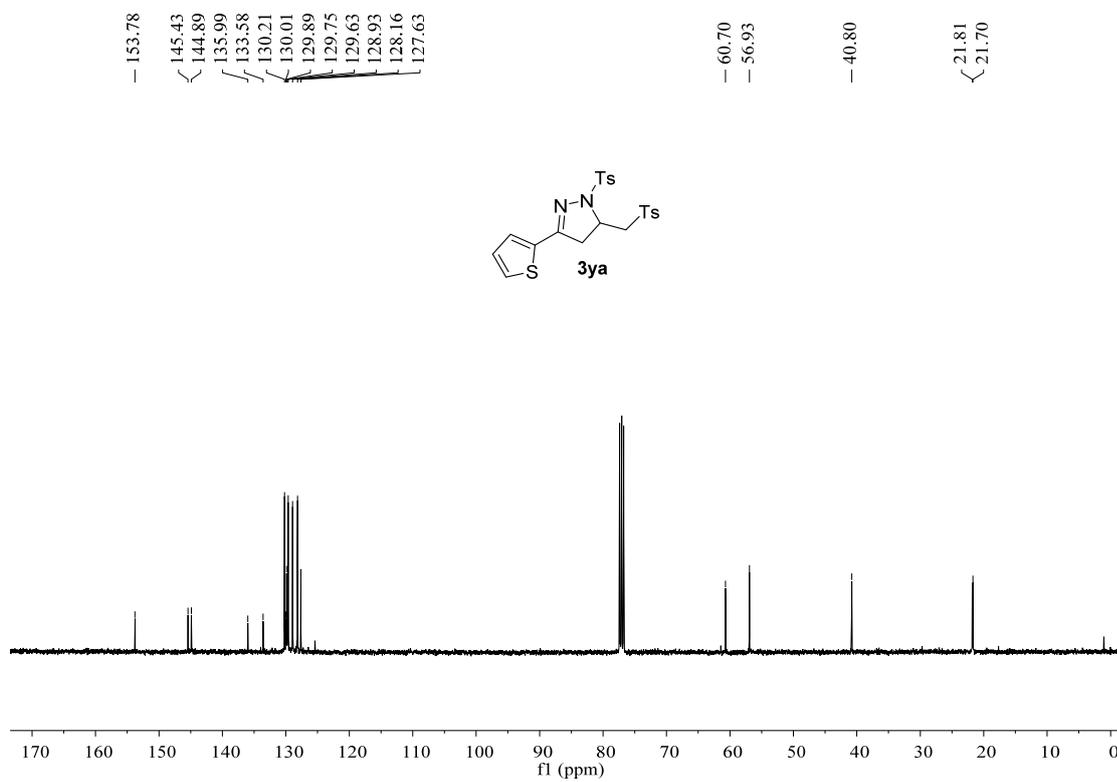
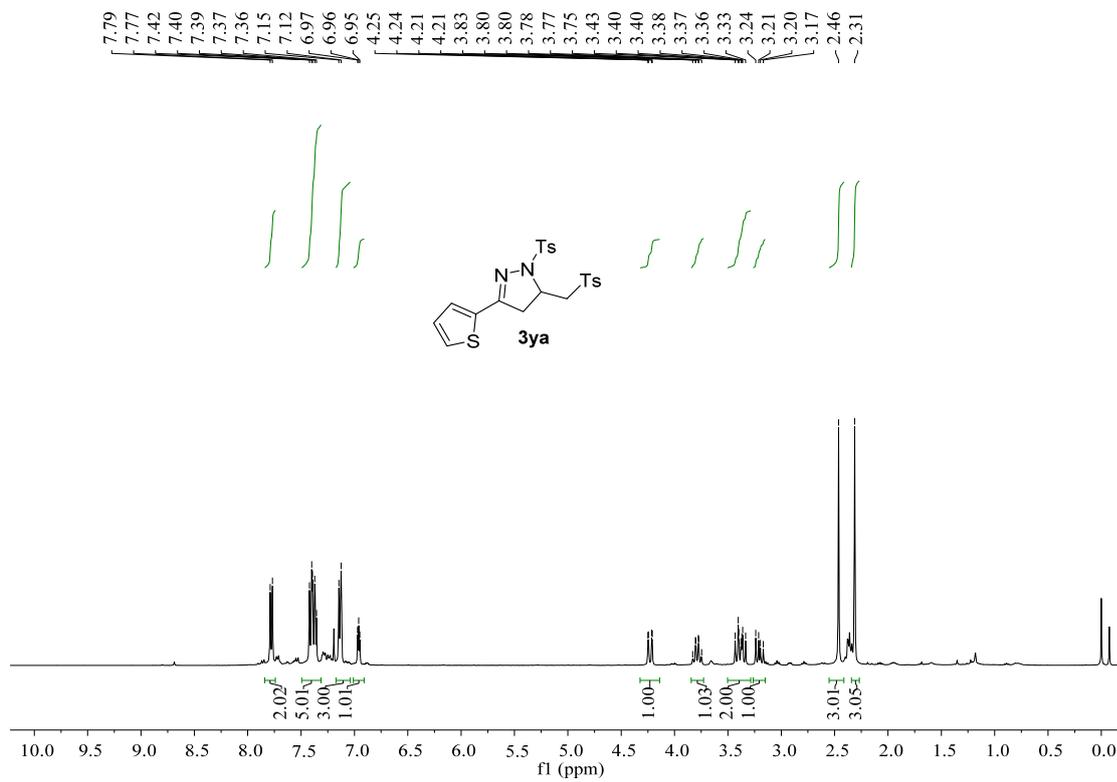


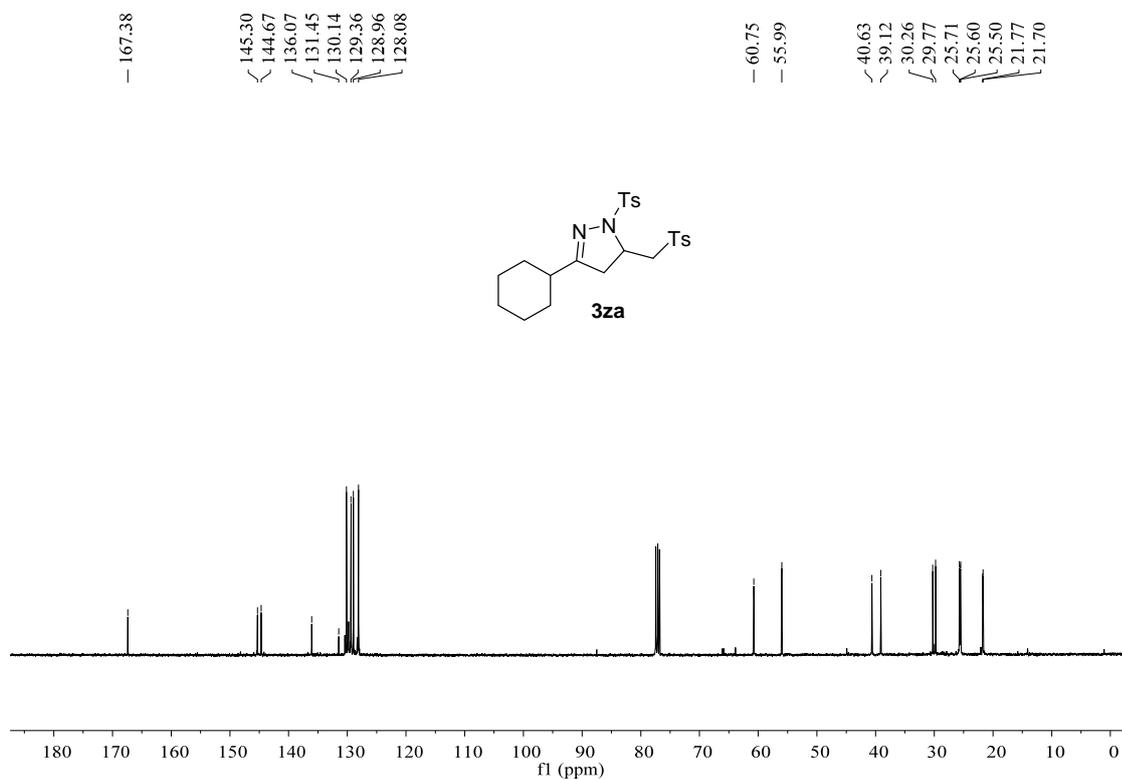
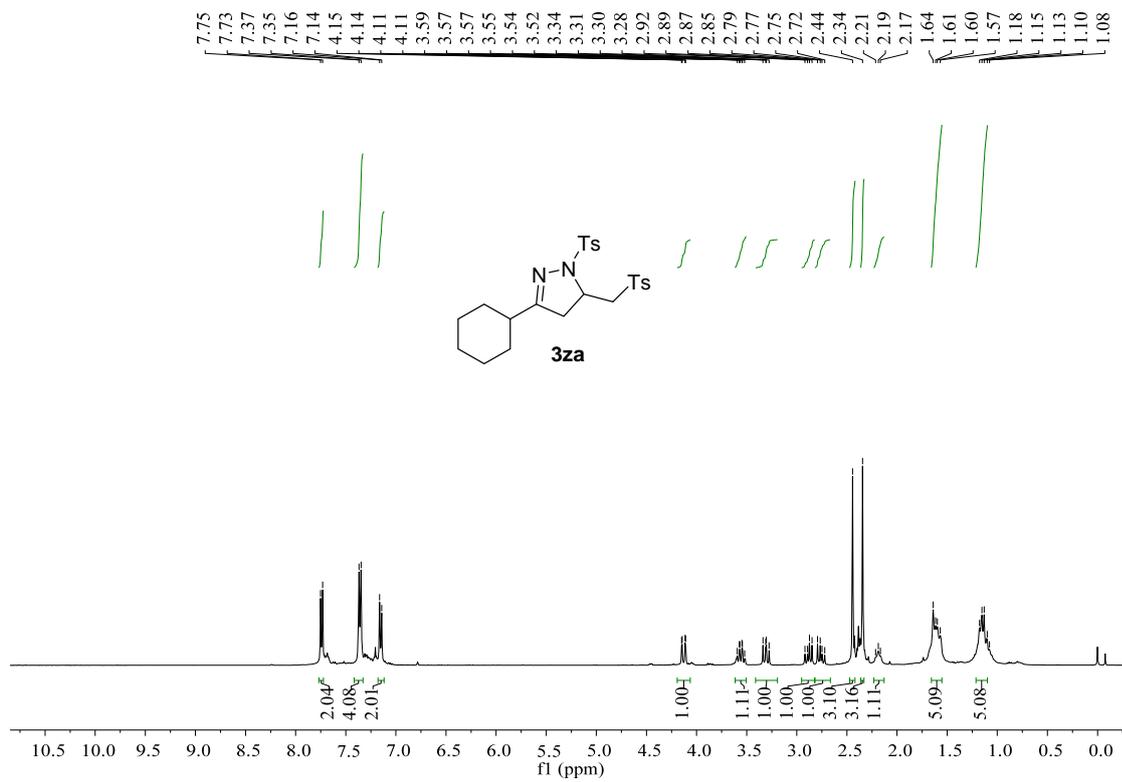


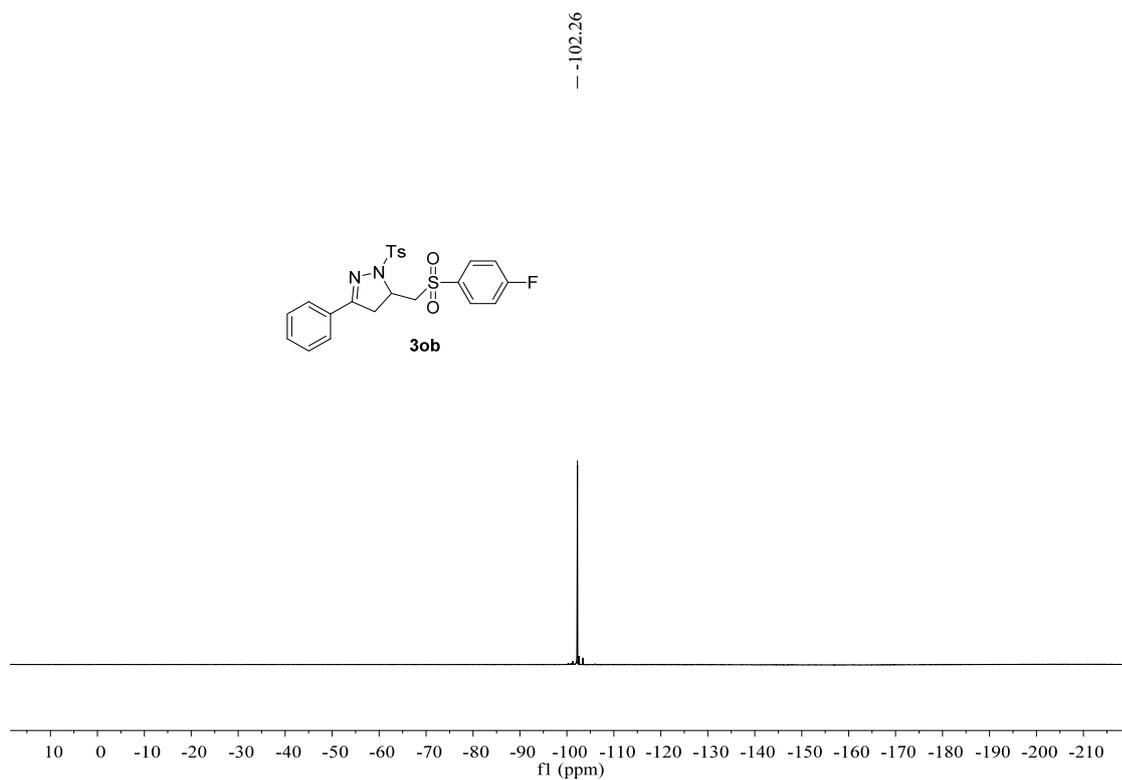
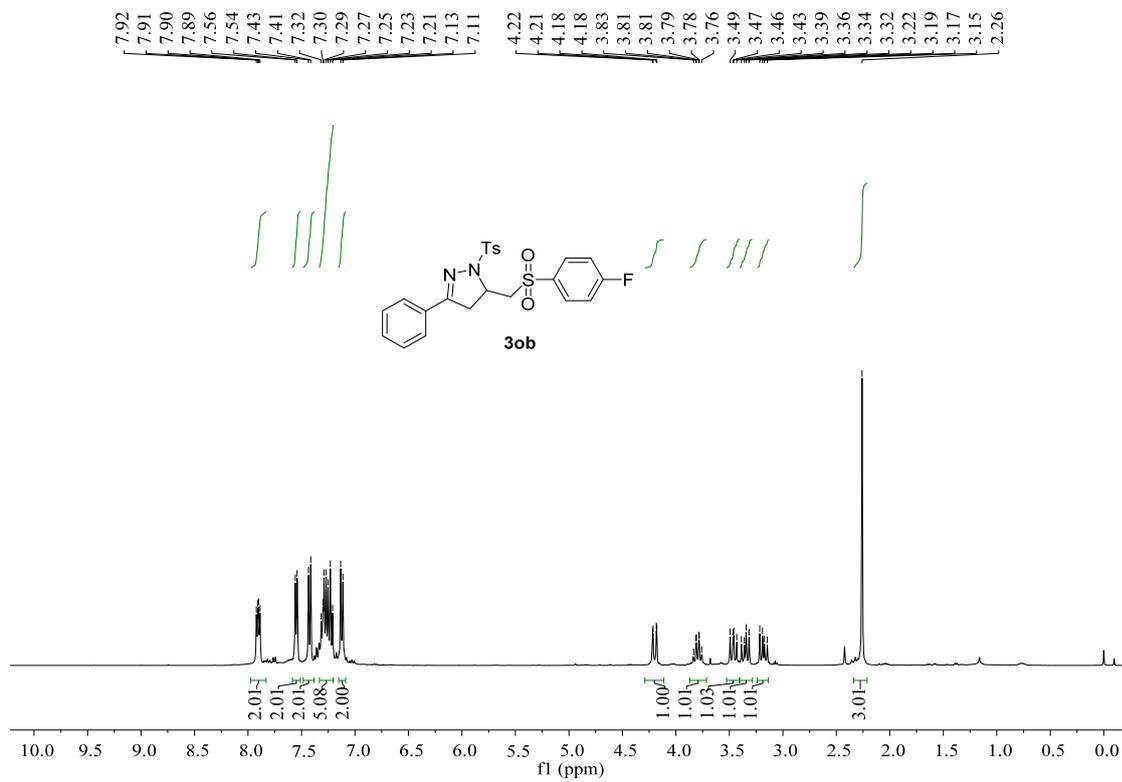


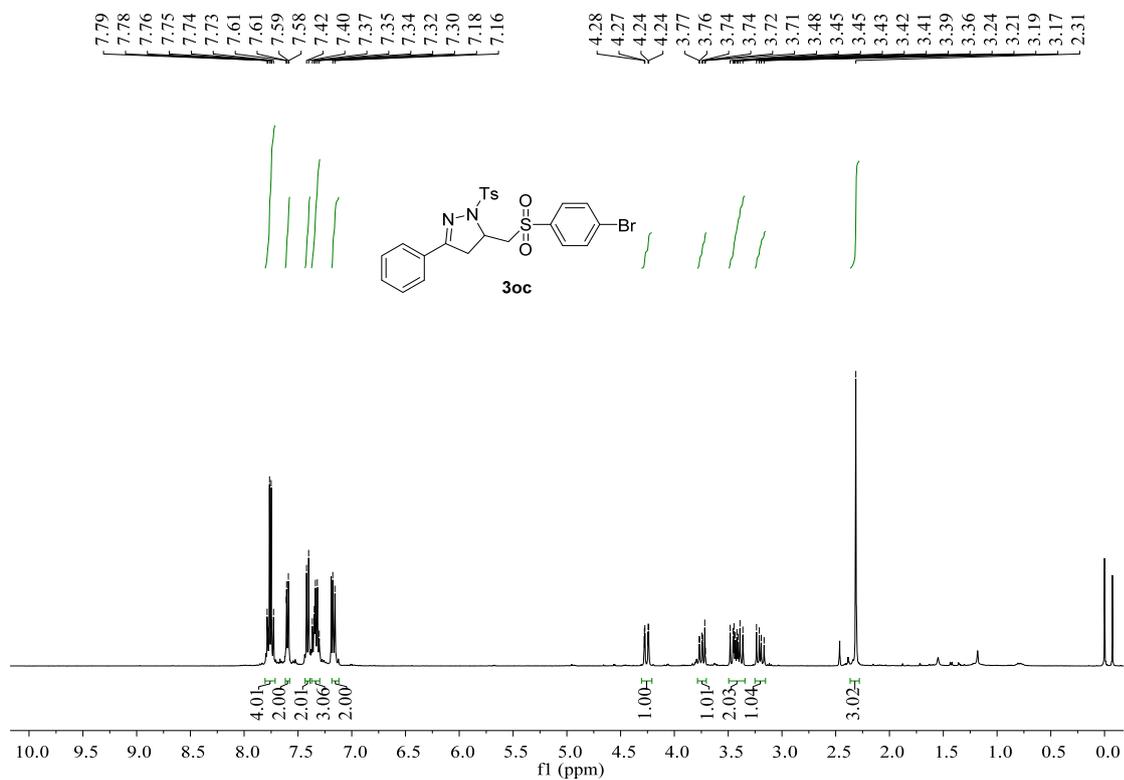
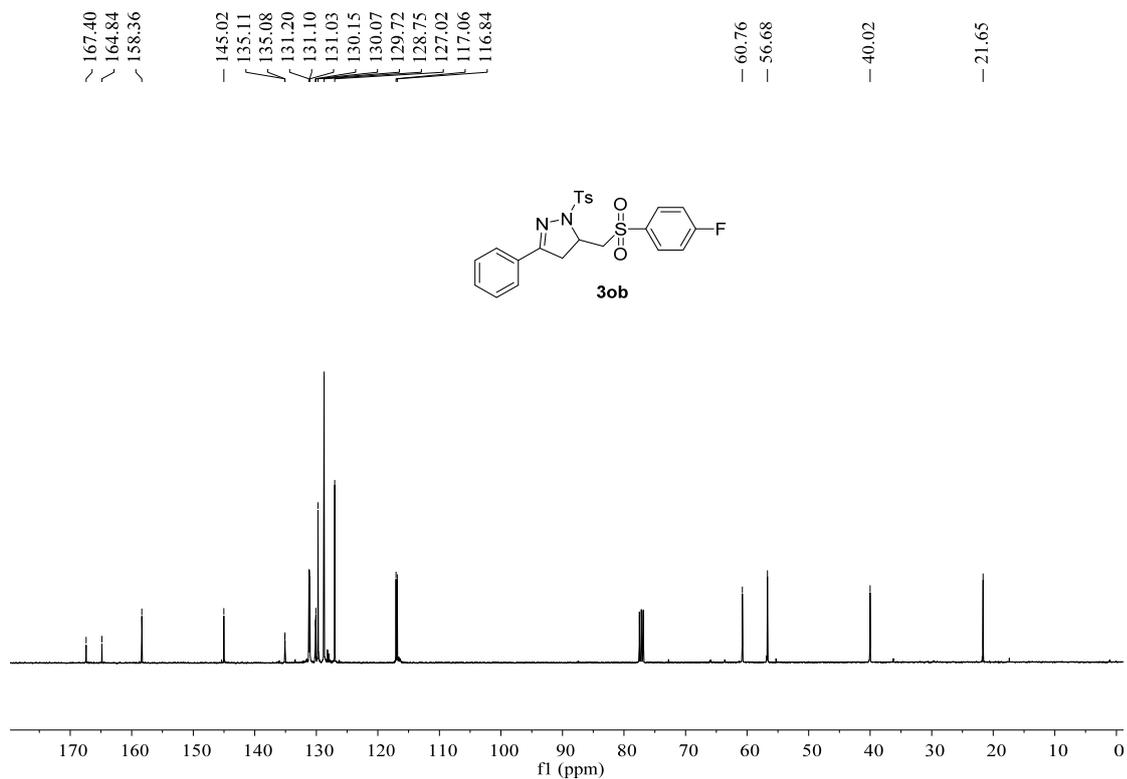


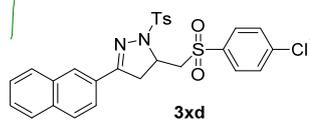
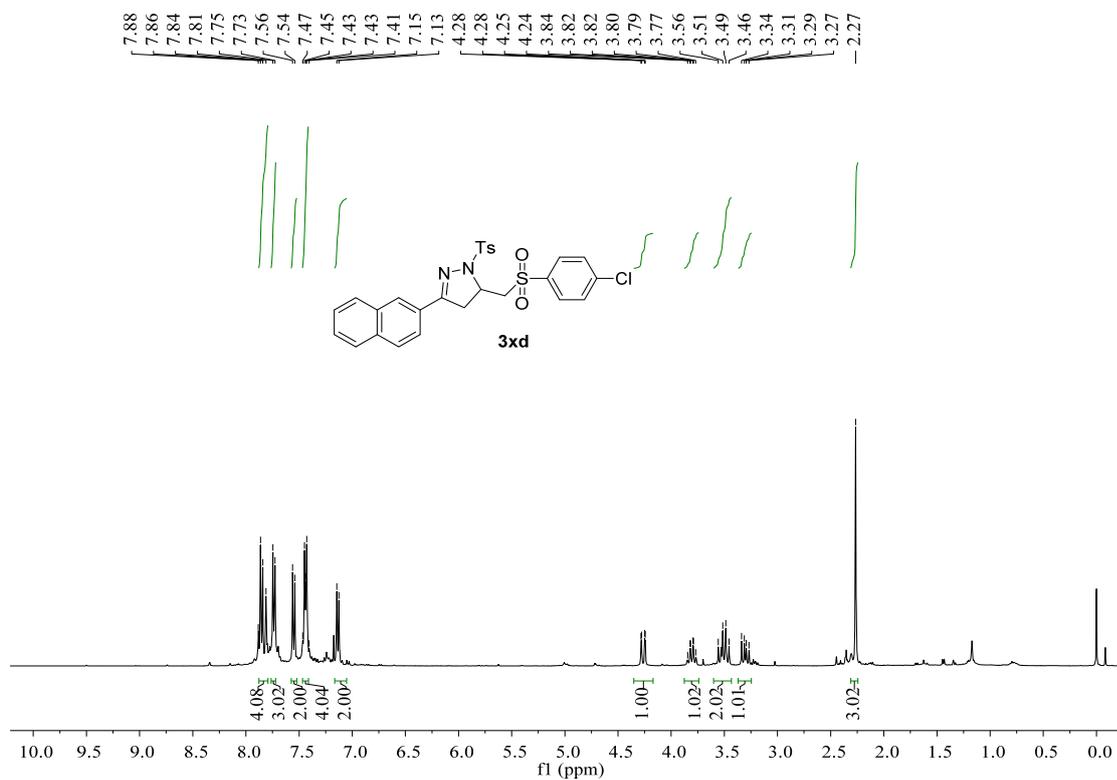
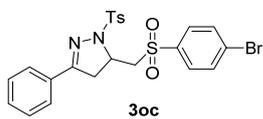
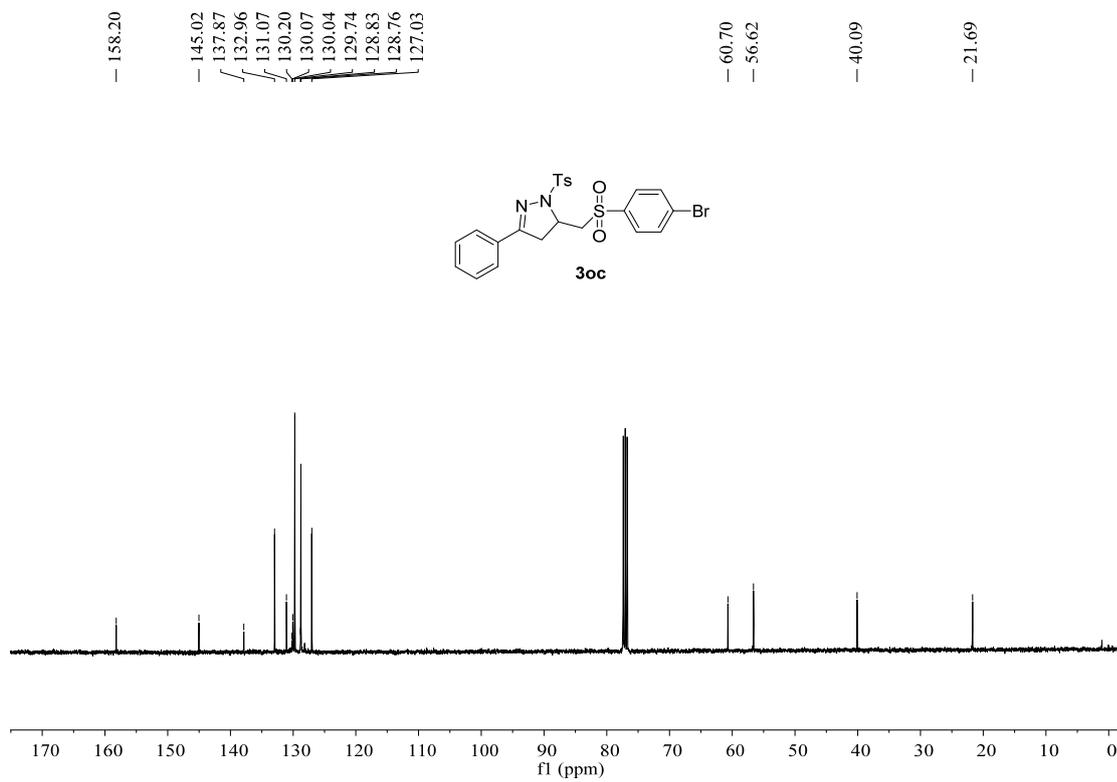


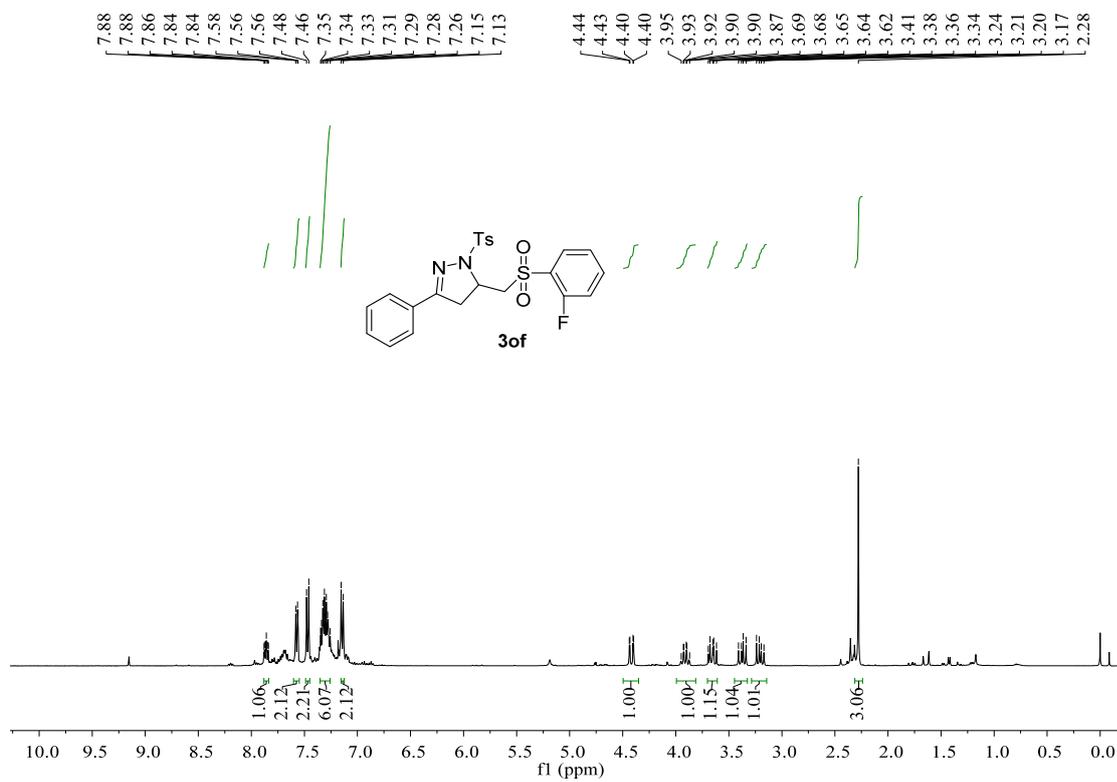
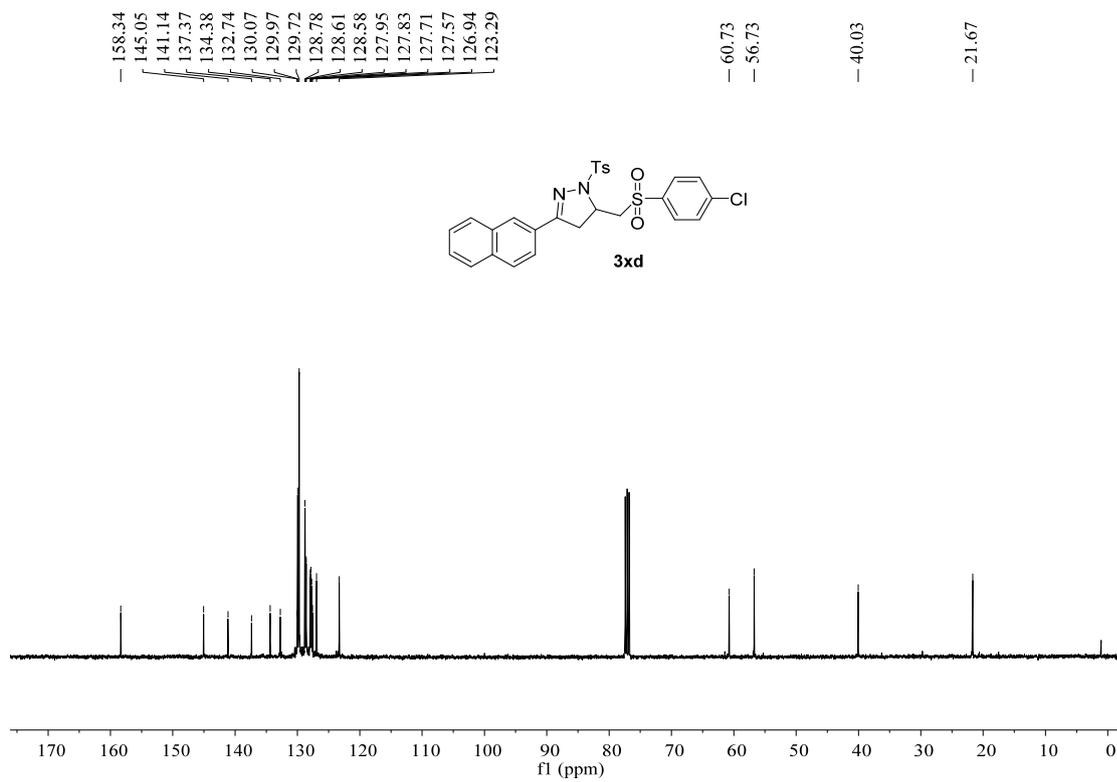


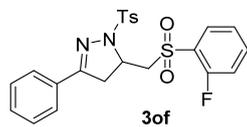




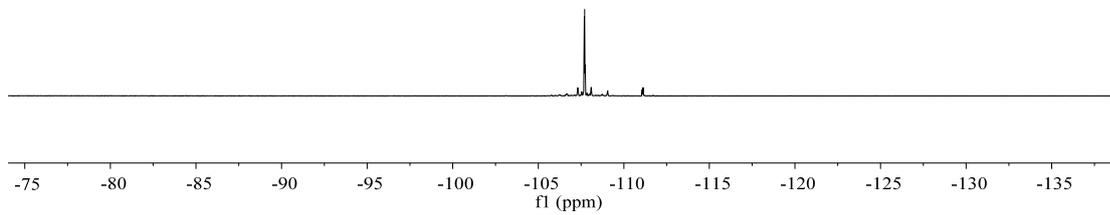








-107.71



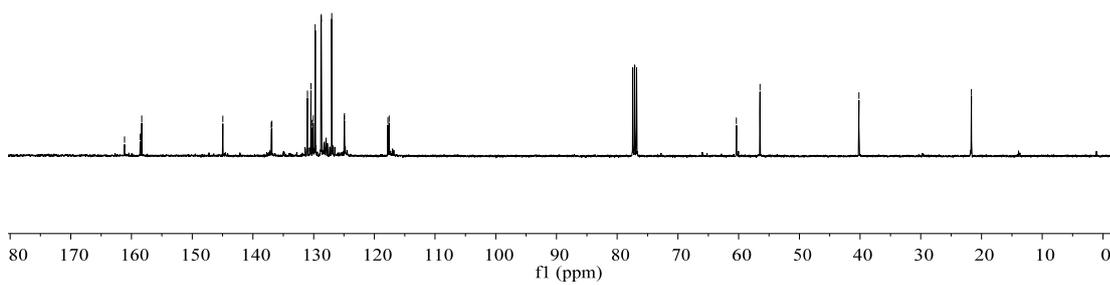
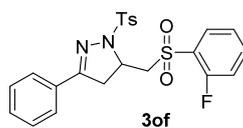
161.14  
158.58  
158.32

144.96  
136.97  
136.89  
131.03  
130.45  
130.22  
130.07  
129.74  
128.78  
128.74  
127.03  
124.96  
124.92  
117.77  
117.56

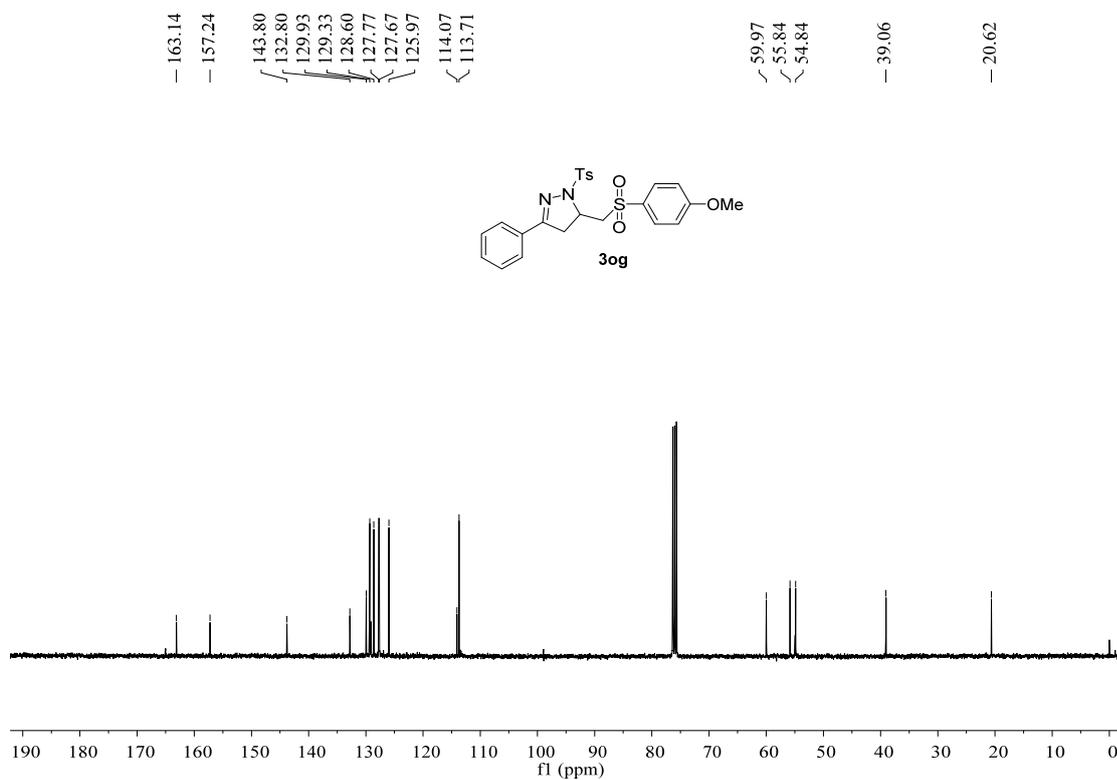
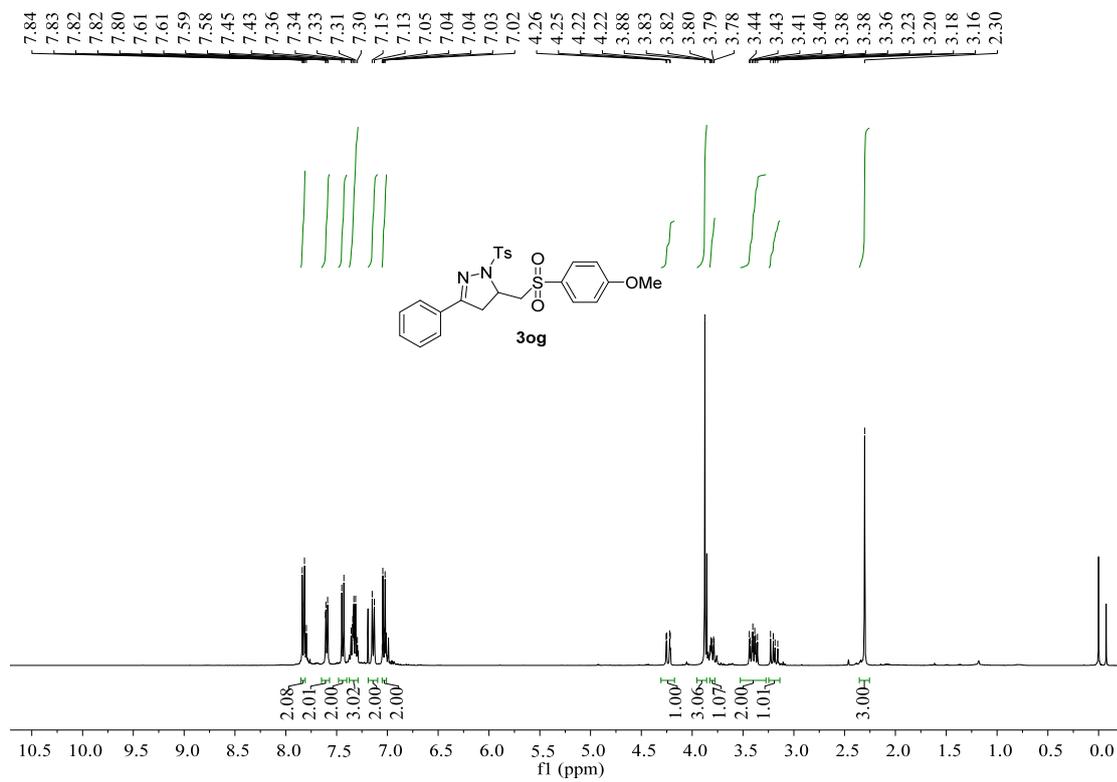
60.36  
56.49

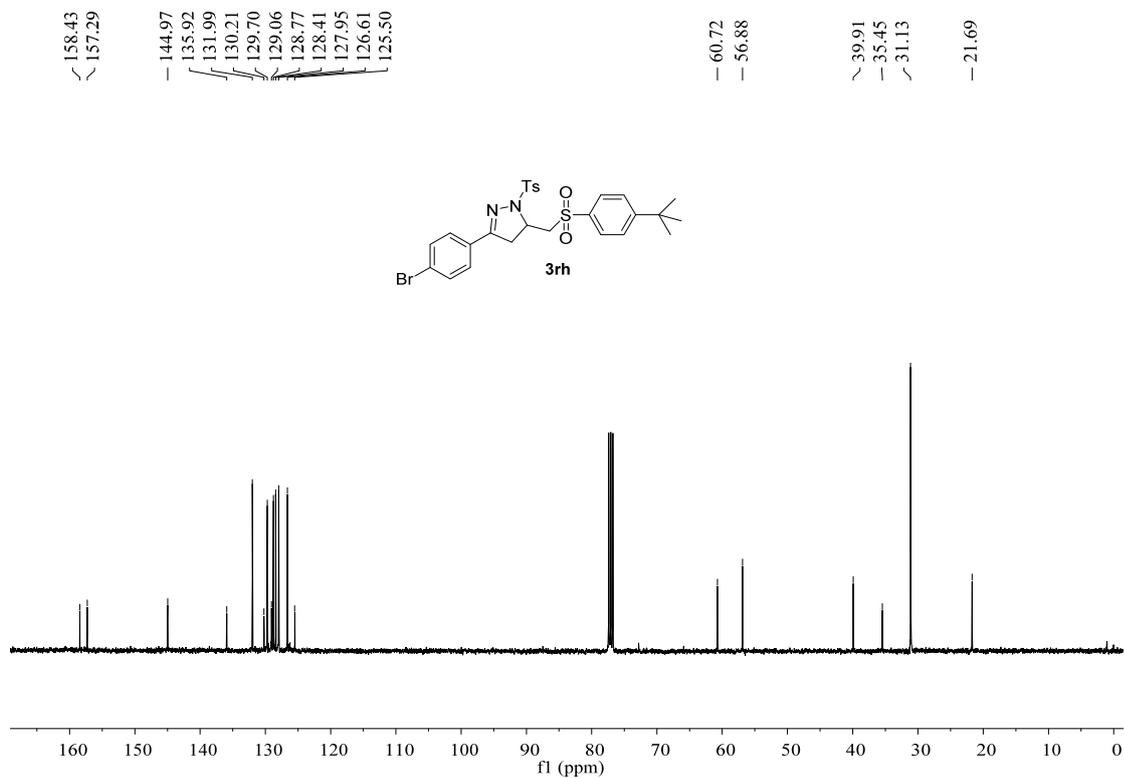
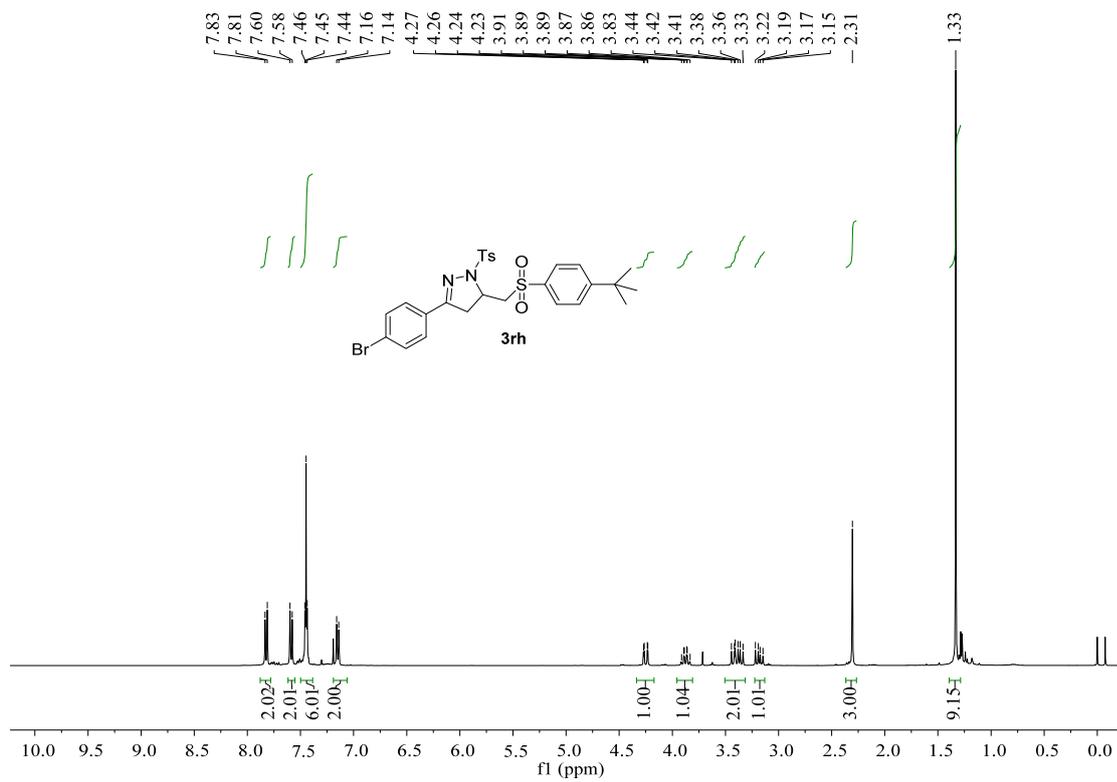
40.20

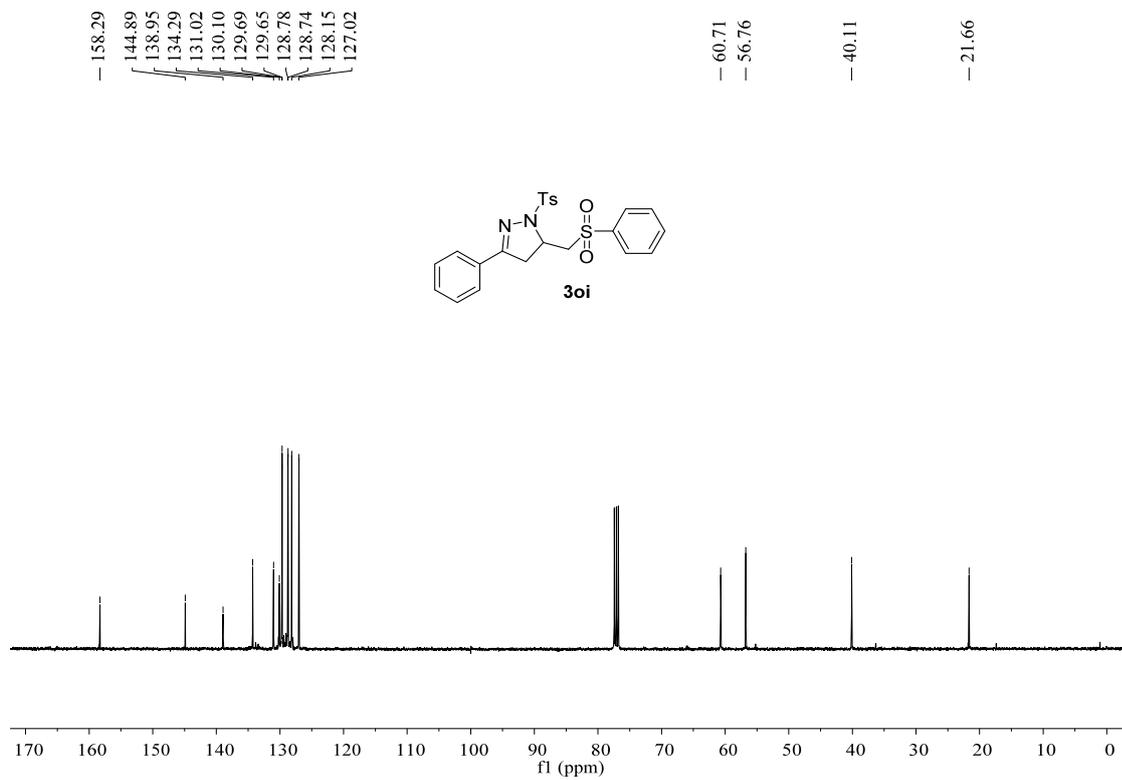
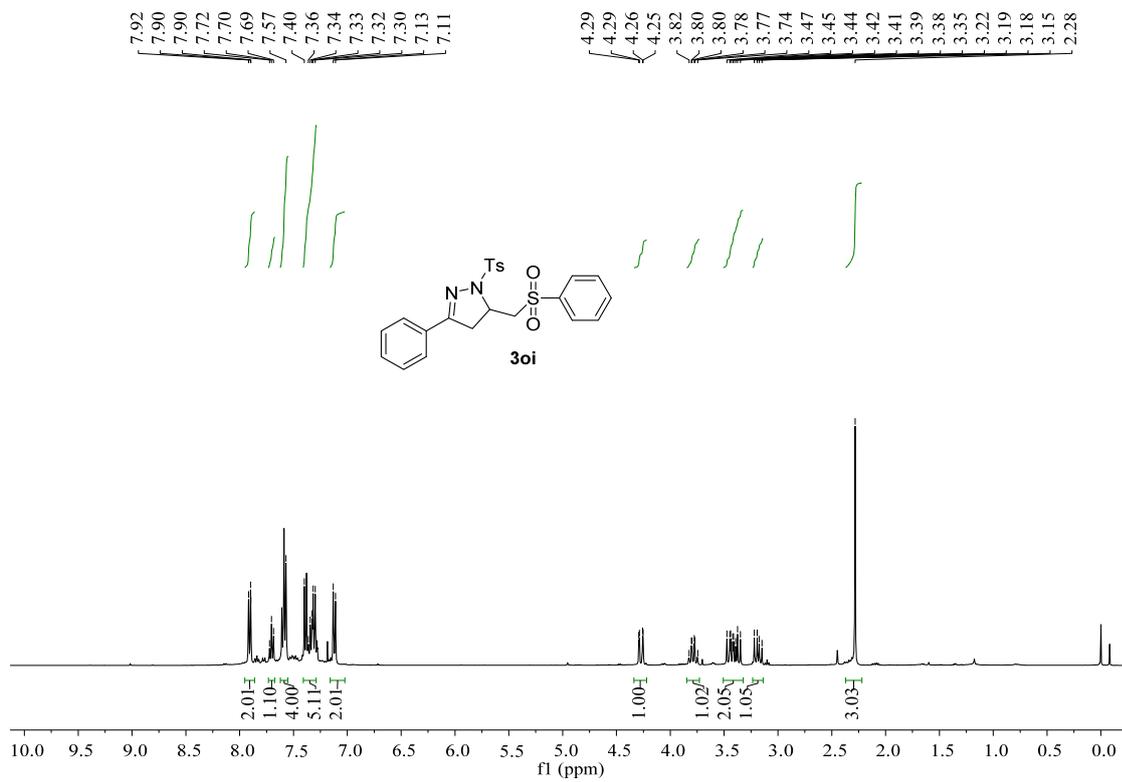
21.66

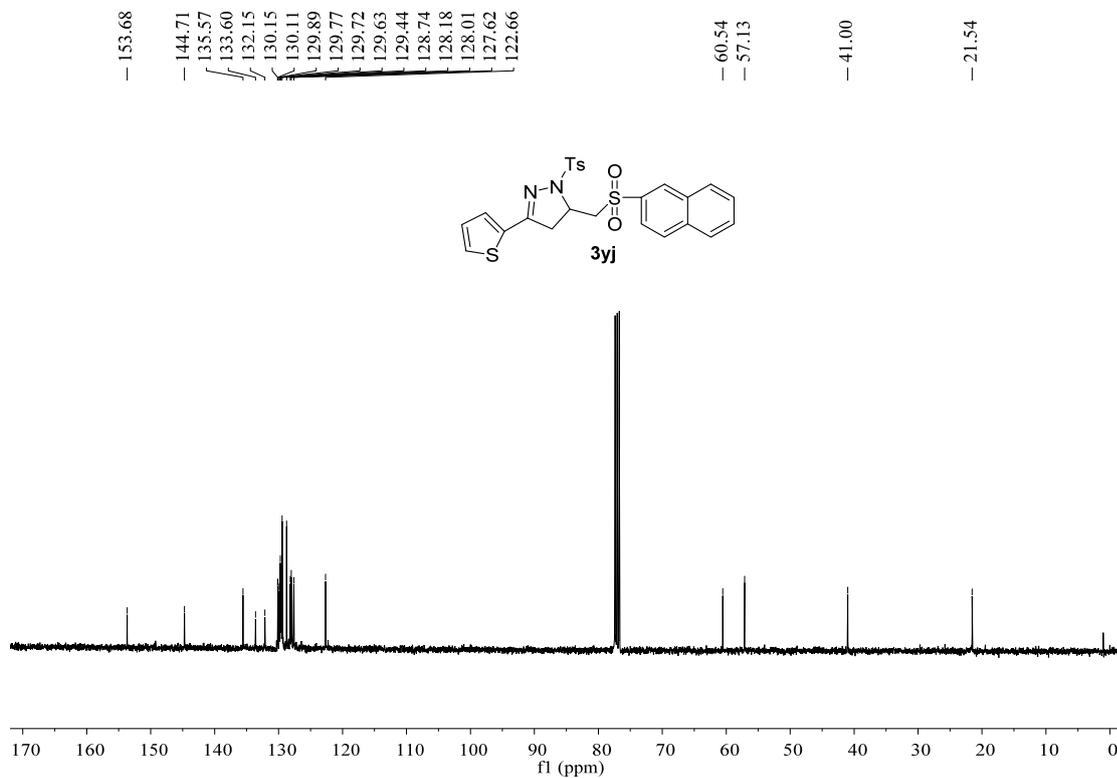
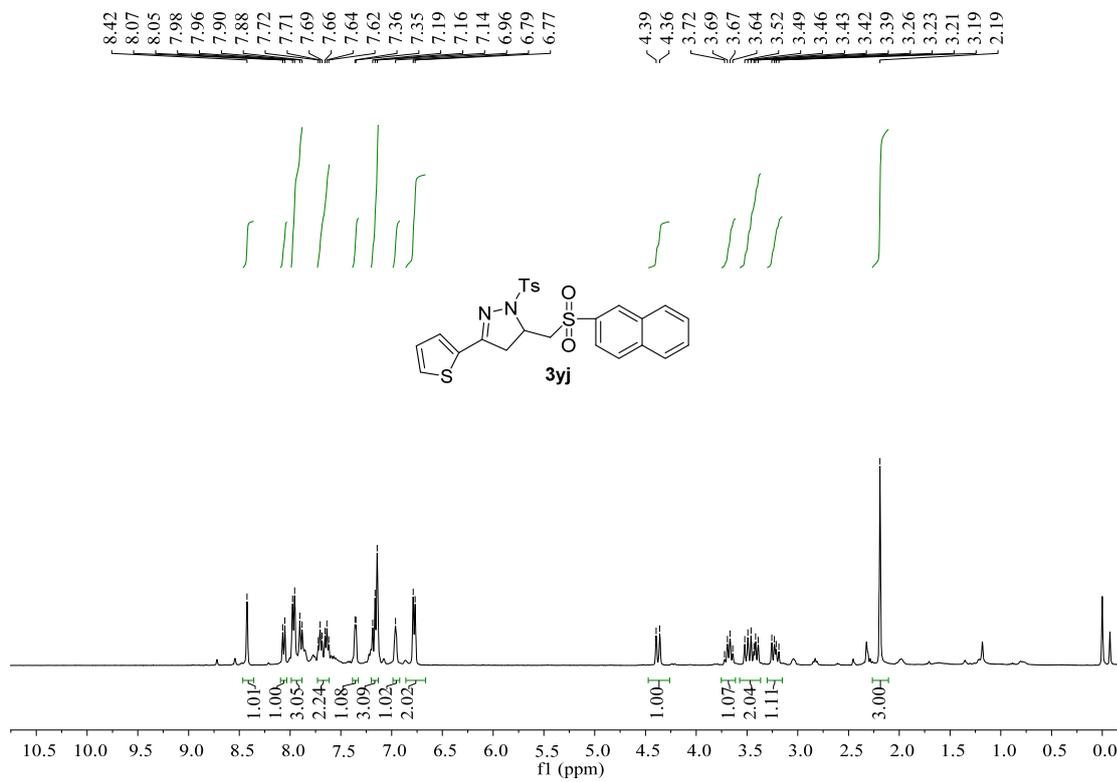


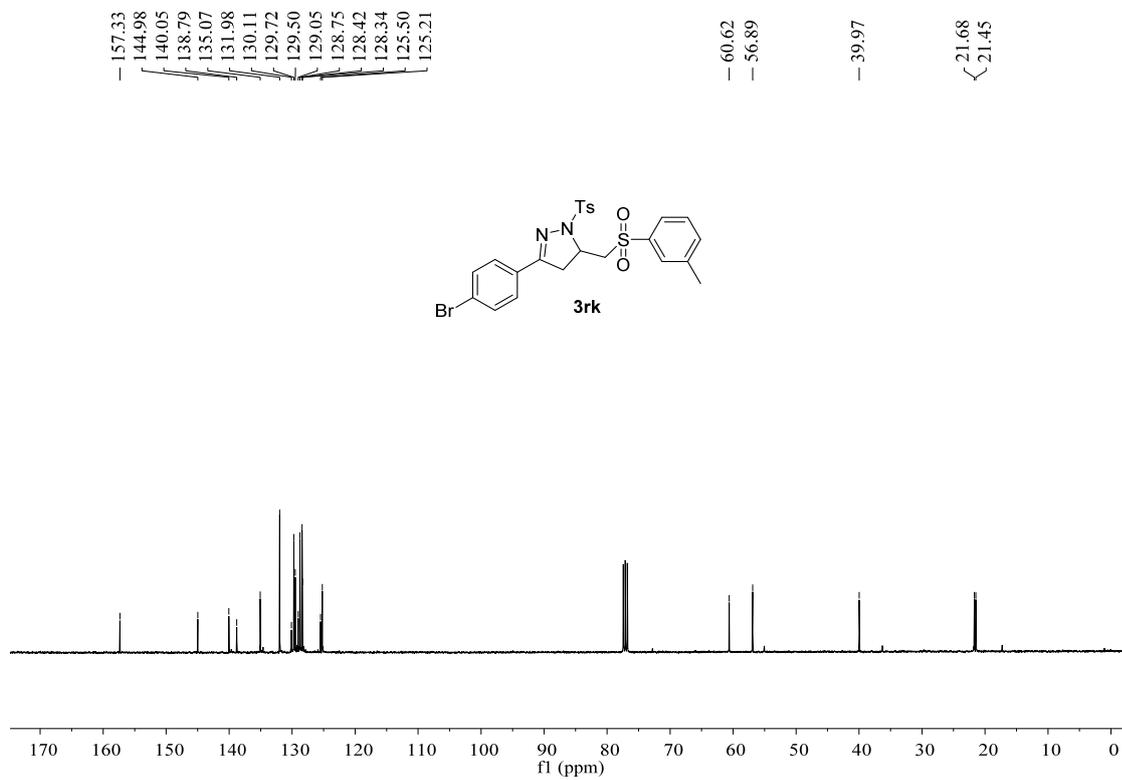
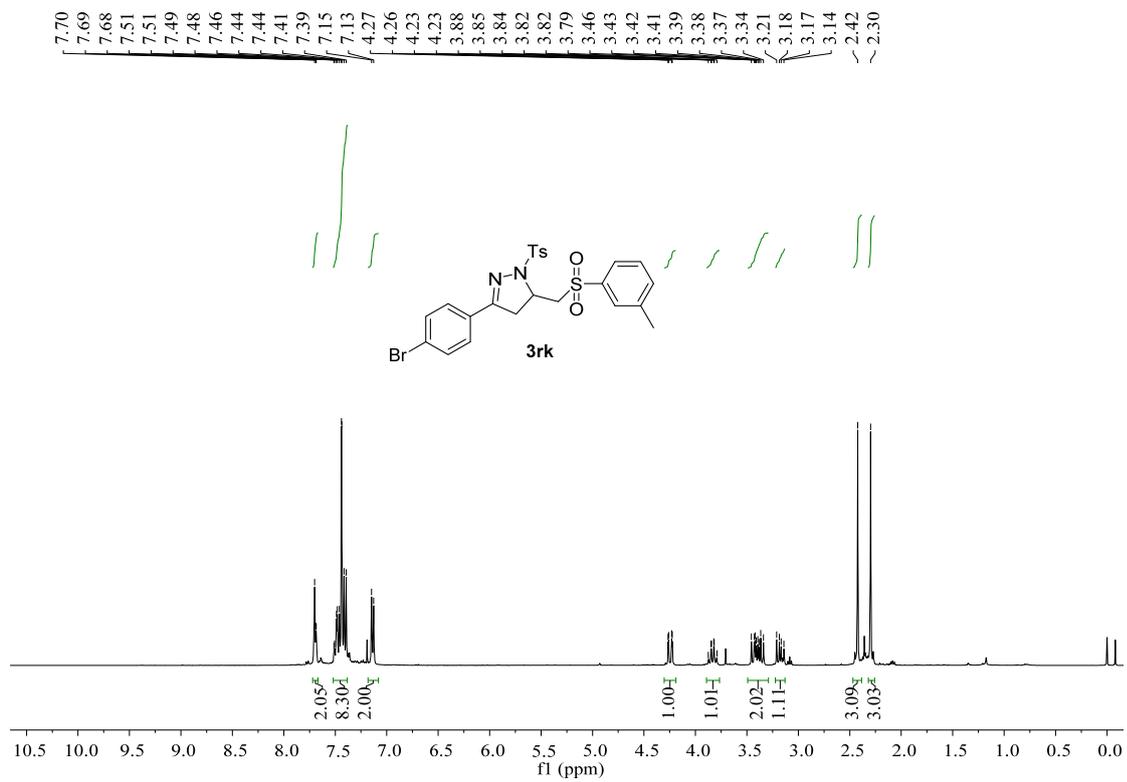


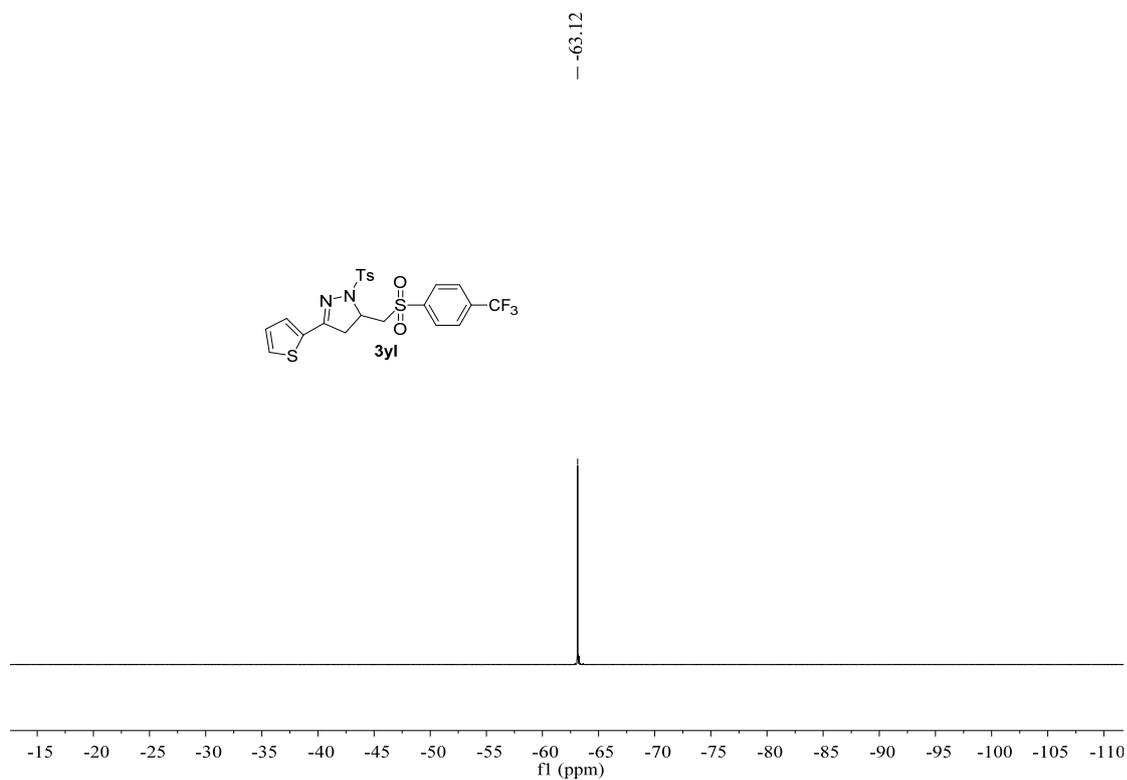
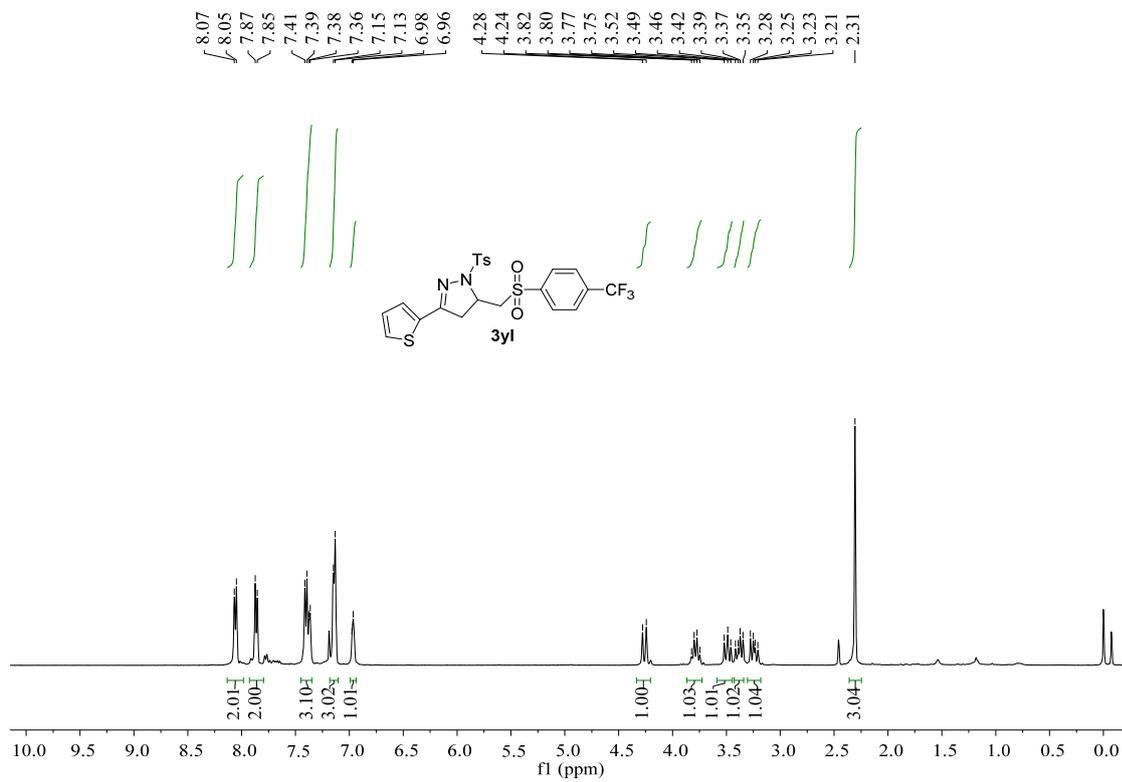


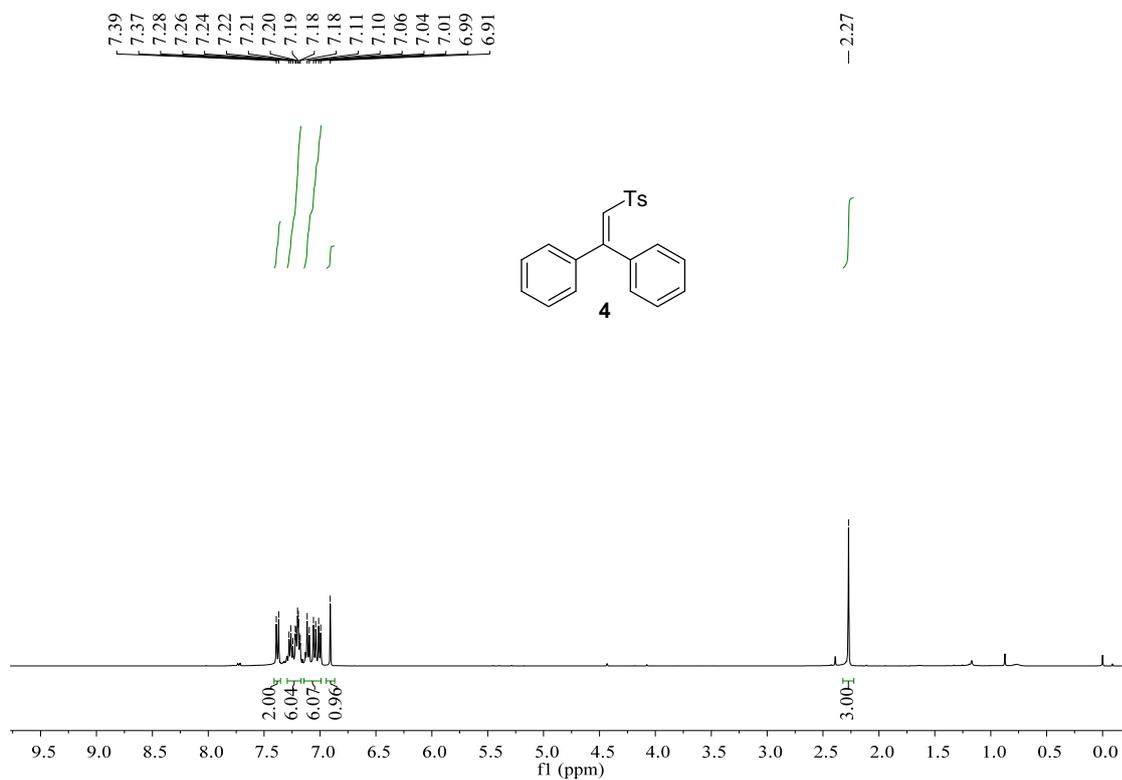
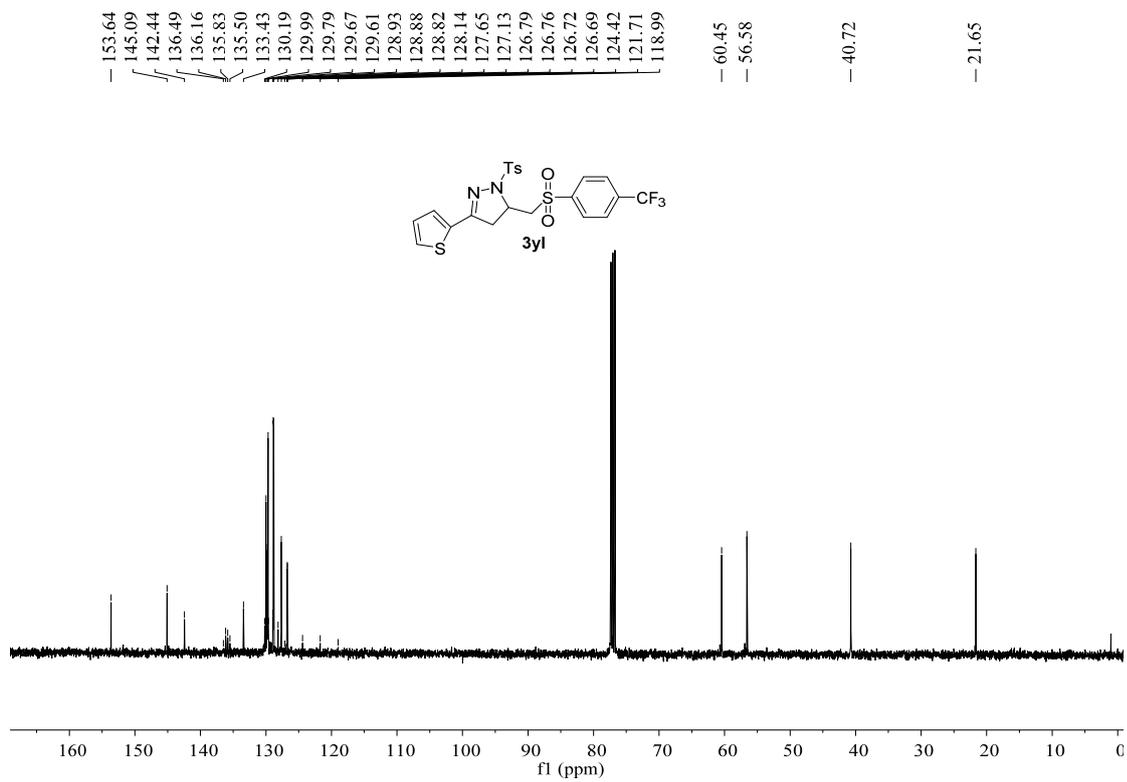


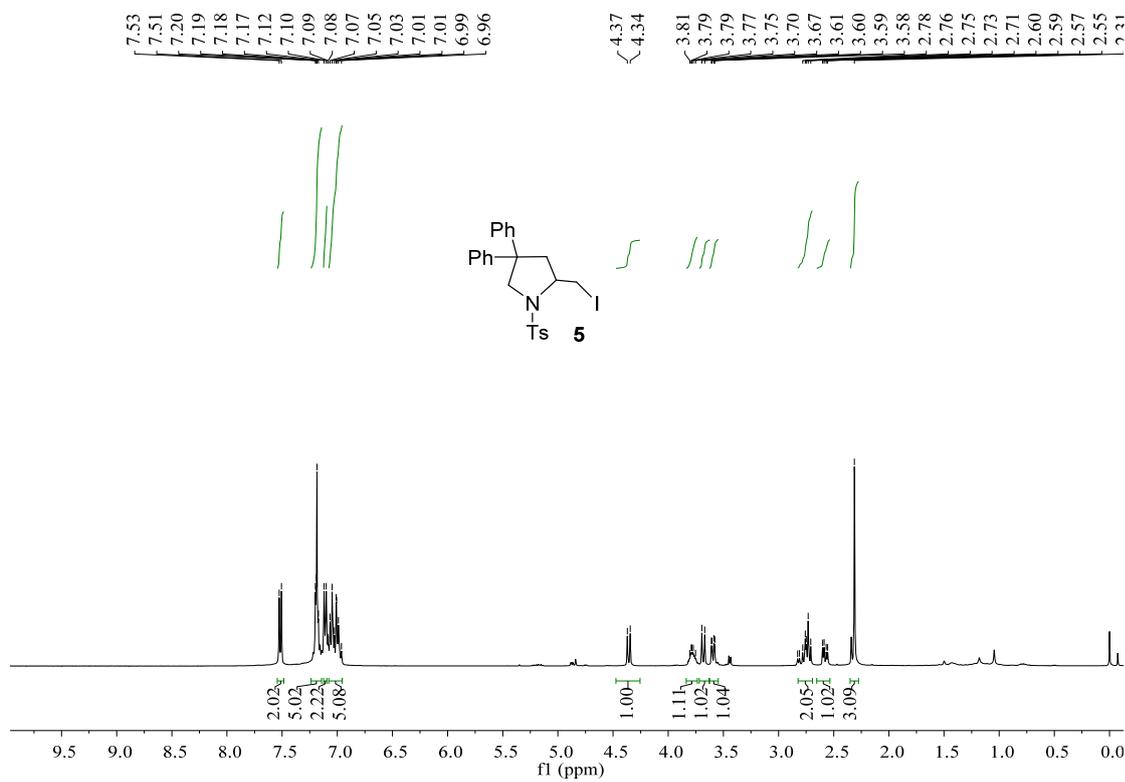
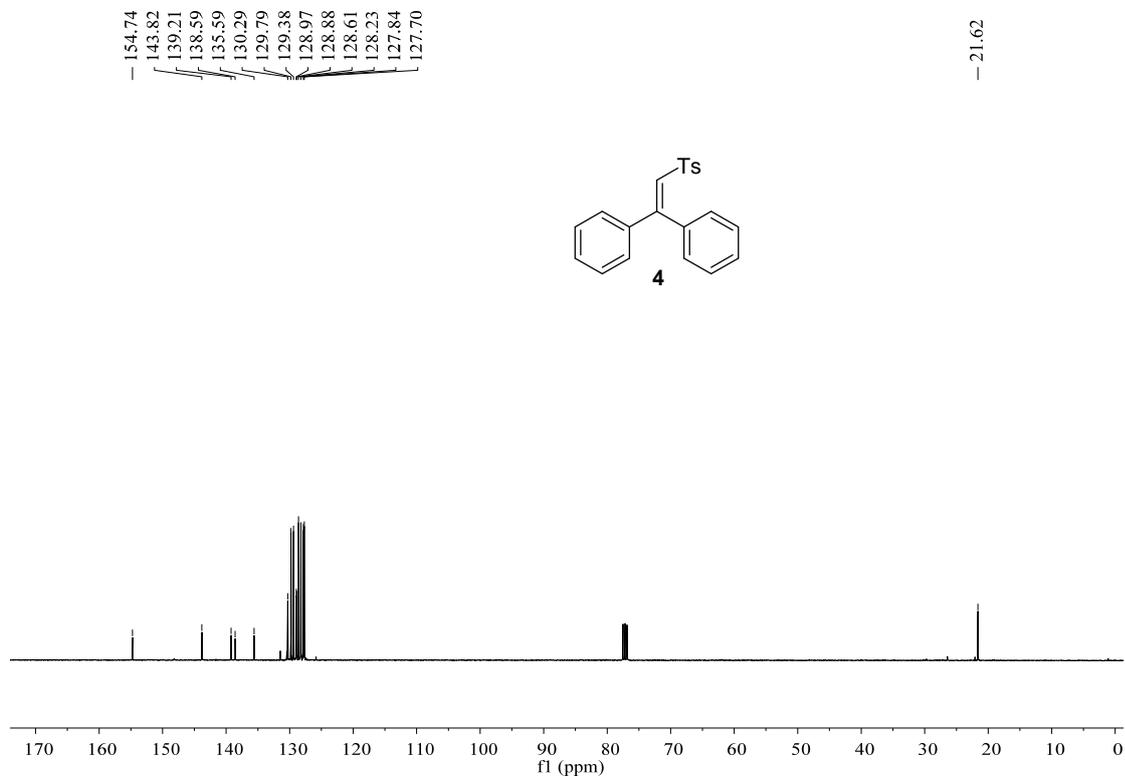




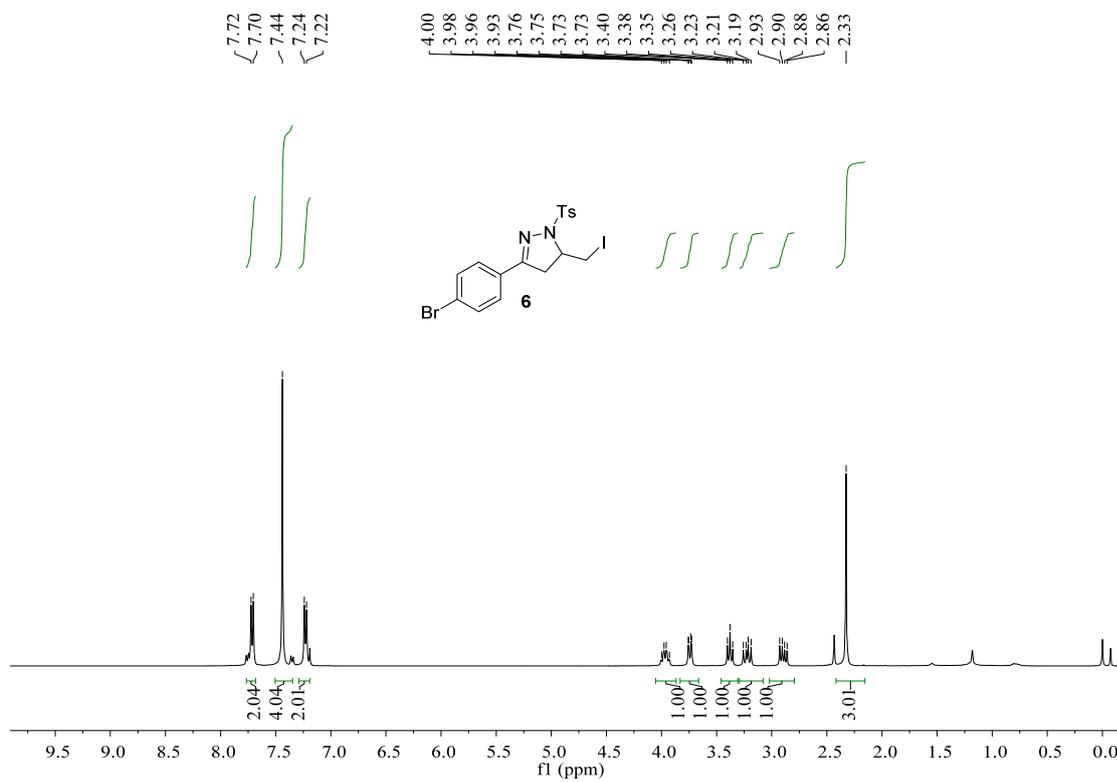
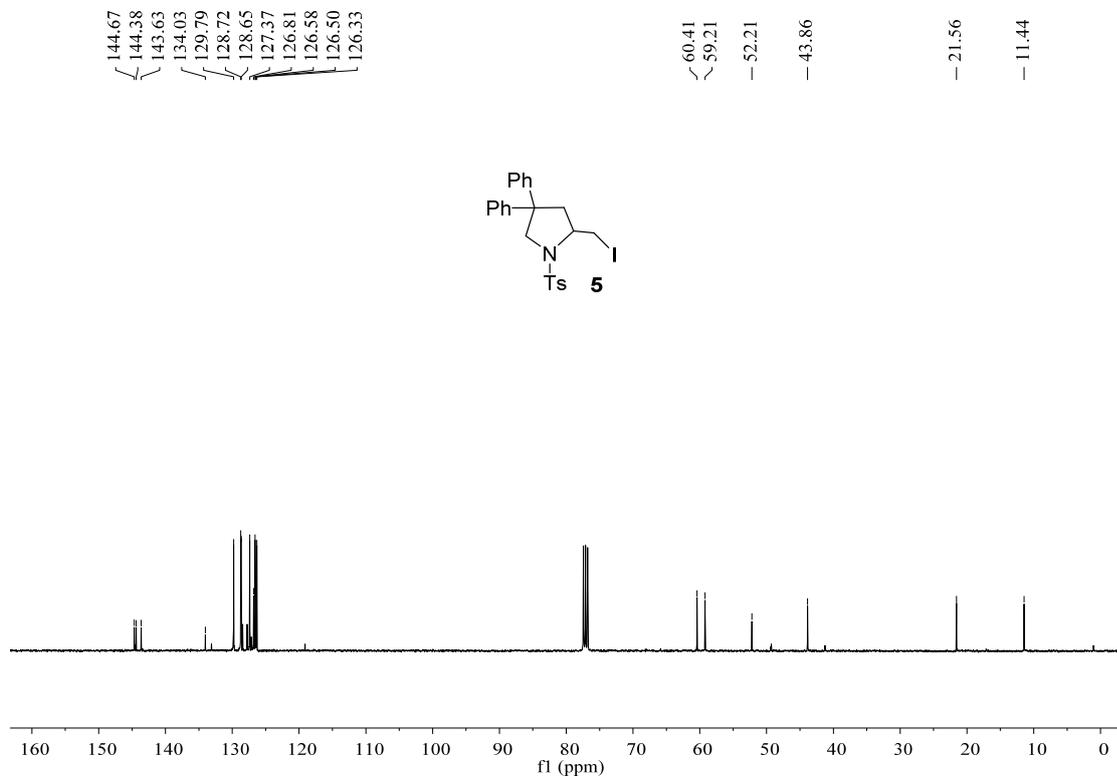












— 155.65  
— 144.75  
131.94  
130.42  
129.74  
129.32  
128.59  
128.35  
125.29

— 62.33

— 41.46

— 21.68

— 9.70

