

Synthesis and Chiroptical Properties of Stereogenic Cyclic Dimers Based on 2,2'- Biselenophene and [2.2]Paracyclophane

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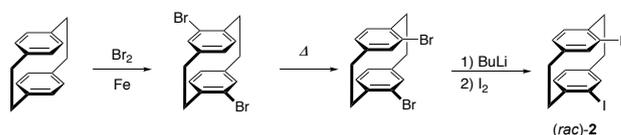
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S1. General

^1H and ^{13}C NMR spectra were recorded on Bruker AVANCE-II 600 (600 MHz for ^1H NMR, 150 MHz for ^{13}C and 114 MHz for ^{77}Se NMR) or Bruker AVANCE-II 400 (400 MHz for ^1H NMR, 100 MHz for ^{13}C and 76 MHz for ^{77}Se NMR) instruments. Spectra are reported in δ , referenced to internal tetramethylsilane (Me_4Si). Mass spectra and High-resolution mass spectra were recorded on Thermo Scientific, Exactive Plus Orbitrap Mass Spectrometer for electrospray ionization (ESI). Melting points were determined with Yanaco melting point apparatus. IR spectra were recorded on JASCO FT/IR-610 spectrometer. UV-Vis spectra were recorded on JASCO V-560 spectrometer. Circular dichroism (CD) spectra were recorded on JASCO J-720 spectrometer. Optical rotation measurements were recorded on JASCO P-1030 spectrometer. Recycling Preparative HPLC was carried out using HPLCLC-9204 instrument by Japan Analytical Industry (JAI) equipped with a CHIRALPAK®IA ($\phi = 30$ mm, $l = 200$ mm) and JASCO OR-2090 Plus.

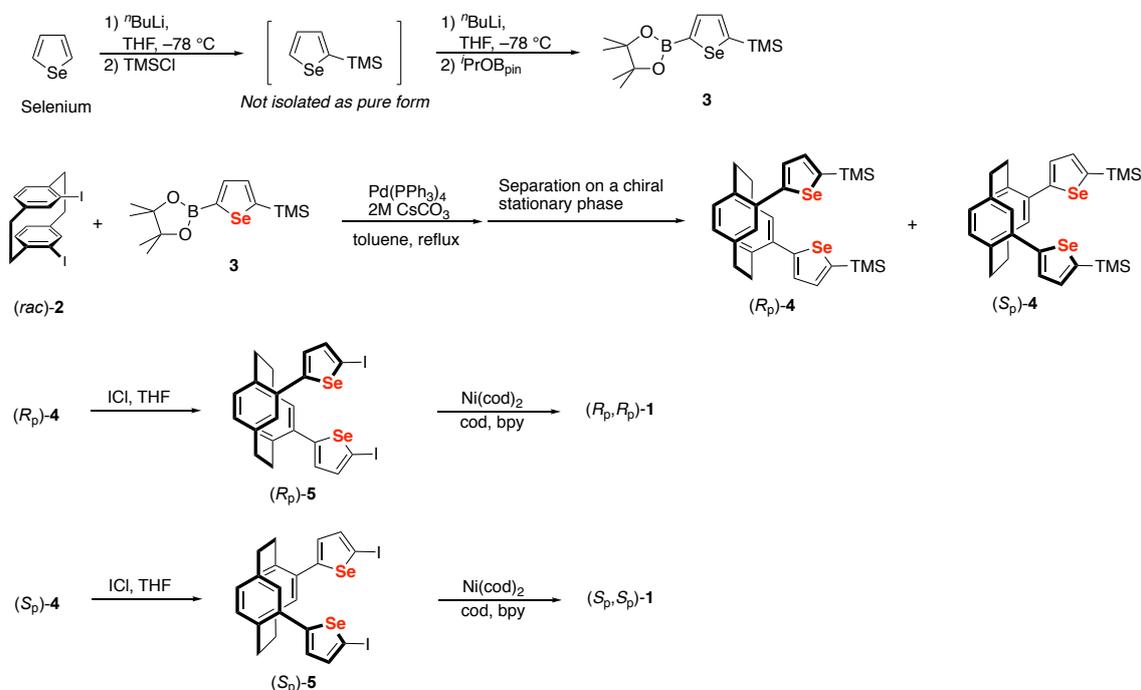
Starting racemic 4,12-diiodo[2.2]paracyclophane (**2**) was prepared according to literature procedure as shown in Scheme S1.¹ [2.2]paracyclophane was purchased from Strem Chemicals, Inc. and stored under Ar at 4 °C. Selenophene (CAS RN. 288-05-1) and 2-Isopropoxy-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (CAS RN. 61676-62-8) were purchased from Tokyo Chemical Industry Co. Ltd, (TCI). Other chemicals were purchased from FUJIFILM Wako Pure Chemical Corporation. Column chromatography was carried out using Kanto Chemical silica gel 60N, 60-210mm meshes. All solvents were dried by conventional procedures and distilled before use.



Scheme S1. Synthesis of racemic **2**

S2. Preparation of 1

Compounds **1** in (*R*_p) and (*S*_p) configuration were prepared as shown in Scheme S2. Optical resolution was carried out in compound **4**, and thereafter conversions were performed separately.



Scheme S2. Synthetic route to compounds **1** in (*R*_p) and (*S*_p) configuration

Synthesis of (*rac*)-**3**

To a solution of selenophene (655 mg, 5.00 mmol) in THF (5.0 mL) ⁿBuLi (3.1 mL, 5.0 mmol) was added dropwise at -78 °C under Ar atmosphere. After 30 min stirring at the same temperature, a solution of trimethylsilyl chloride (TMSCl) (0.64 mL, 5.1 mmol) was added dropwise. The mixture was stirred for 30 min at -78 °C, and then allowed to warm up rt. Further continuous stirring was carried out at rt for 1h. Then, aqueous NH₄Cl solution was added to the mixture. The products were extracted by Et₂O. The combined organic solution was washed with saturated brine, and dried over MgSO₄. The solvent was carefully removed under reduced pressure due to highly volatile properties of the product. Residual oil was purified on silica gel column chromatography with the elution of hexane. The resultant colorless oil (1.02 g) contains almost pure form of 2-trimethylsilylselenophene (> 99%). However, due to the extreme high volatility under ambient conditions, we carried out next reaction without further treatment. Data for 2-trimethylsilylselenophene: MS (ESI-orbitrap) *m/z* = 204 (M⁺); ¹H NMR (400 MHz, CDCl₃) δ 7.99 (dd, *J* = 5.4 and 1.2 Hz, ¹H, together with ⁷⁷Se-¹H satellite band: *J* = 36 Hz), 7.19 (dd, *J* = 3.6 and 1.2 Hz, 1H), 7.08 (dd, *J* = 5.4 and 3.6 Hz, 2H), -0.3 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 148.3, 136.3, 135.3, 130.9, 0.40; ⁷⁷Se NMR (76 MHz, CDCl₃) δ 672.

To a solution of above mixture (1.02 g) in THF (10 mL) *n*BuLi (3.1 mL, 5.0 mmol) was added dropwise carefully at $-78\text{ }^{\circ}\text{C}$ under Ar atmosphere. After 1h stirring at the same temperature, a solution of 2-Isopropoxy-4,4,5,5-tetramethyl-1,3,2-dioxaborolane ($\text{B}_{\text{pin}}\text{-O}^i\text{Pr}$) (1.2 mL, 5.8 mmol) was added dropwise. The mixture was stirred for 2h at $-78\text{ }^{\circ}\text{C}$, and then allowed to warm up rt. Further continuous stirring was carried out at rt for 14h. Then, aqueous NH_4Cl solution was added to the mixture. The products were extracted by EtOAc. The combined organic solution was washed with saturated brine, and dried over MgSO_4 . After the removal of the solvent under reduced pressure, the residue was purified recrystallization from MeOH at $-20\text{ }^{\circ}\text{C}$ to give colorless solid of **3** (742 mg, 45 % from selenophene). The molecular structure was also confirmed by X-ray crystallographic analysis. Data for **3**: Mp. $86.1\text{-}86.8\text{ }^{\circ}\text{C}$; HRMS (ESI-orbitrap): m/z calcd for $\text{C}_{13}\text{H}_{23}\text{SeBSiO}_2$ $[\text{M}+\text{H}^+]$: 331.07984, Found: 331.07980; ^1H NMR (400 MHz, CDCl_3) δ 7.99 (d, $J = 3.6$ Hz, 1H), 7.63 (d, $J = 3.6$ Hz, 1H), 1.34 (s, 12H), 0.32 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 156.7, 140.5, 137.5, 84.1, 24.8, 0.34; ^{77}Se NMR (76 MHz, CDCl_3) δ 742.8; IR (KBr) 2977, 2956, 1528, 1447, 1373, 1340, 1250, 1142, 1067 cm^{-1} .

Synthesis of (*rac*)-**4**

A mixture of 4,12-diiodo[2.2]paracyclophane **2** (140 mg, 0.30 mmol), compound **3** (400 mg, 1.22 mmol), $\text{Pd}(\text{PPh}_3)_4$ (74 mg, 0.064 mmol), and aqueous solution of Cs_2CO_3 (2M, 4.0 mL, 8.0 mmol) in toluene (10 mL) was stirred for 14h at $120\text{ }^{\circ}\text{C}$ under Ar atmosphere. Then, the reaction mixture was filtered off through Celite pad with AcOEt. Resultant solution was washed with saturated brine and dried over MgSO_4 . After the removal of the solvent under reduced pressure, the residue was purified on silica gel column chromatography with the elution of CH_2Cl_2 -hexan solution ($v/v = 1:1$). Further purification was accomplished by GPC separation to give pale yellow oil of the crude product. From the analysis of NMR spectrum, the crude product contains **4** together with **4'**, whose selenophene rings are connected at 4,5-positions of [2.2]paracyclophane. In our previous report, Suzuki coupling reaction of 4,12-diiodo[2.2]paracyclophane occasionally led to rearranged compound at 4,5 positions.^{s2} The former Fr.1-**4'** was characterized by ^1H , ^{13}C , and ^{77}Se NMR together with MS spectra. The reaction yields were determined by NMR ration to be 81 and 13 %, respectively. The mixture was subject to a recycling HPLC method a recycling HPLC method without further treatment. Thus, separation on a hiral stationary phase (chiralpack IA3) with the elution of CHCl_3 -hexane ($v/v = 1:8$) gave Fr.1-**4'**, Fr.2-**4**, and Fr.3-**4** (Figure S1).

The obtained fraction of Fr.2-**4** exhibited large optical specific rotation of $[\alpha]_{\text{D}}^{25} = -26.5^{\circ}$ ($c = 5.0 \times 10^{-3}$) in CH_2Cl_2 , while the latter Fr.3-**4** exhibited $[\alpha]_{\text{D}}^{25} = +26.0$ ($c = 3.4 \times 10^{-3}$) in the identical conditions. Fortunately, slow evaporation of Fr.2-(-)-**4** from CH_2Cl_2 -MeOH solution gave a single crystal suitable for X-ray analysis, which suggests (S_p) configuration with sufficient flack parameter (Table S1). In summary, the fraction of 1st (-)-**4** 2nd (+)-**4** could be determined as (S_p)- and (R_p)-

isomer, respectively.

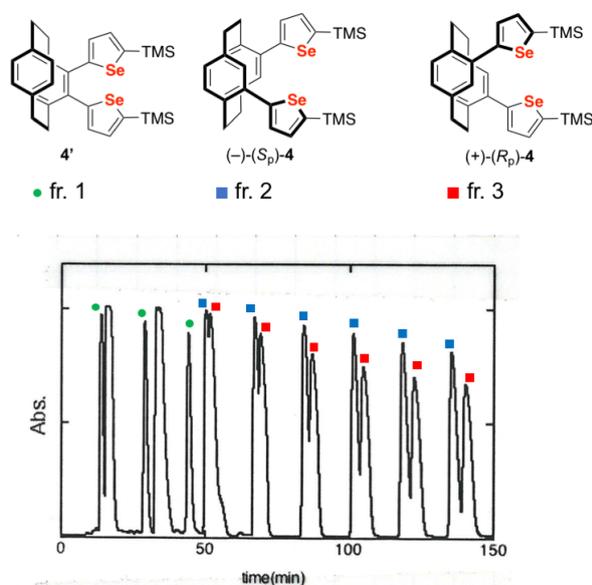


Figure S1. HPLC chart (CHCl_3 -hexane 8:1) of crude product.

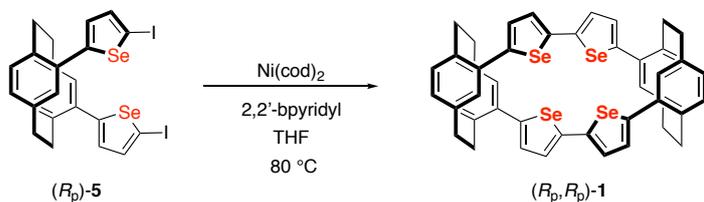
Data for (*R_p*)-**4**: Mp. 144-145 °C; HRMS (ESI-orbitrap): m/z calcd for $\text{C}_{30}\text{H}_{37}\text{Se}_2\text{Si}_2$ [$\text{M}+\text{H}^+$]: 613.07587, Found: 613.07587; ^1H NMR (400 MHz, CDCl_3) δ 7.50 (d, $J = 3.6$ Hz, 2H), 7.25 (d, $J = 3.6$ Hz, 2H), 6.84 (d, $J = 1.6$ Hz, 2H), 6.65 (d, $J = 7.8$ Hz, 2H), 6.58 (dd, $J = 7.8$ and 1.6 Hz, 2H), 3.72-3.80 (m, 2H), 3.03-3.11 (m, 2H), 2.92-2.96 (m, 4H), 0.35 (s, 18H); ^{13}C NMR (150 MHz, CDCl_3) δ 155.0, 148.7, 139.8, 136.7, 136.5, 136.4, 135.5, 132.2, 130.9, 129.2, 34.8, 33.9, 0.44; ^{77}Se NMR (76 MHz, CDCl_3) δ 678.7; $[\alpha]_{25}^{\text{D}}$ = +26.0 for (*R_p*)-**5** (CH_2Cl_2 , $c = 0.019$) and -26.5 for (*S_p*)-**5** (CH_2Cl_2 , $c = 0.020$); IR (KBr) 3030, 2956, 2926, 2884, 2854, 1587, 1443, 1401, 1247, 1205, 1071 cm^{-1} .

Data for **4'**: colorless oil; HRMS (ESI-orbitrap): m/z $\text{C}_{30}\text{H}_{37}\text{Se}_2\text{Si}_2$ [$\text{M}+\text{H}^+$]: 613.07587, Found: 613.07581; ^1H NMR (600 MHz, CDCl_3) δ 7.29-7.31 (m, 2H), 6.95-6.98 (m, 2H), 6.84 (d, $J = 1.8$ Hz, 2H), 6.76 (d, $J = 1.8$ Hz, 2H), 6.52 (s, 2H), 3.19-3.22 (m, 2H), 3.09-3.11 (m, 2H), 3.00-3.05 (m, 2H), 2.93-2.96 (m, 2H); ^{13}C NMR (150 MHz, CDCl_3) δ 153.2, 149.5, 139.8, 128.8, 125.6, 135.0, 134.1, 132.7, 131.7, 131.7, 139.4, 35.3, 33.9, 0.39; ^{77}Se NMR (114 MHz, CDCl_3) δ 717.7; IR (KBr) 3034, 2954, 2928, 2894, 2854, 1454, 1249, 1203 cm^{-1} .

Synthesis of (*R_p*)-**5**

To a solution of (*R_p*)-**4** (77 mg, 0.13 mmol) in THF (5 mL), ICl (0.19 M in THF, 3.0 mL, 0.57 mmol) was added dropwise at -78°C under Ar. After the reaction mixture was stirred for 4h at -78°C , the reaction was quenched by the addition of saturated aqueous NaHCO_3 and 10% $\text{Na}_2\text{S}_2\text{O}_4$ solution. The products were extracted with AcOEt and the combined organic phase was washed with saturated brine and dried over Na_2SO_4 , subsequently. After removal of the solvent under reduced pressure, the residue was purified by column chromatography on silica gel with the elution of hexane- CH_2Cl_2 ($v/v = 1:1$) to give yellow solid (*R_p*)-**5** (90 mg, 99%). The other enantiomer (*S_p*)-**5** was prepared in a similar manner from (*S_p*)-**4** (y. 99%). Data for (*R_p*)-**5**: Mp $122\text{--}125^{\circ}\text{C}$; HRMS (ESI-orbitrap): m/z calcd for $\text{C}_{24}\text{H}_{19}\text{Se}_2\text{I}_2$ $[\text{M}+\text{H}^+]$: 720.79004, Found: 720.79011; ^1H NMR (400 MHz, CDCl_3) δ 7.54 (d, $J = 4.0$ Hz, 2H), 6.80 (d, $J = 4.0$ Hz, 2H), 6.70 (d, $J = 1.6$ Hz, 2H), 6.65 (d, $J = 8.0$ Hz, 2H), 6.59 (dd, $J = 8.0$ and 1.6 Hz, 2H), 3.64-3.71 (m, 2H), 3.07-3.14 (m, 2H), 2.83-2.94 (m, 4H); ^{13}C NMR (150 MHz, CDCl_3) δ 160.0, 140.7, 140.1, 136.9, 135.9, 135.6, 132.8, 130.35, 129.2, 75.5, 34.9, 34.0; ^{77}Se NMR (76 MHz, CDCl_3) δ 722.6; $[\alpha]_{\text{D}_{25}}^{25} = +65.7$ for (*R_p*)-**5** (CH_2Cl_2 , $c = 0.019$) and -66.5 for (*S_p*)-**5** (CH_2Cl_2 , $c = 0.020$); IR (KBr) 2922, 2851, 1585, 1477, 1399 cm^{-1} .

Synthesis of (*R_p,R_p*)-**1**

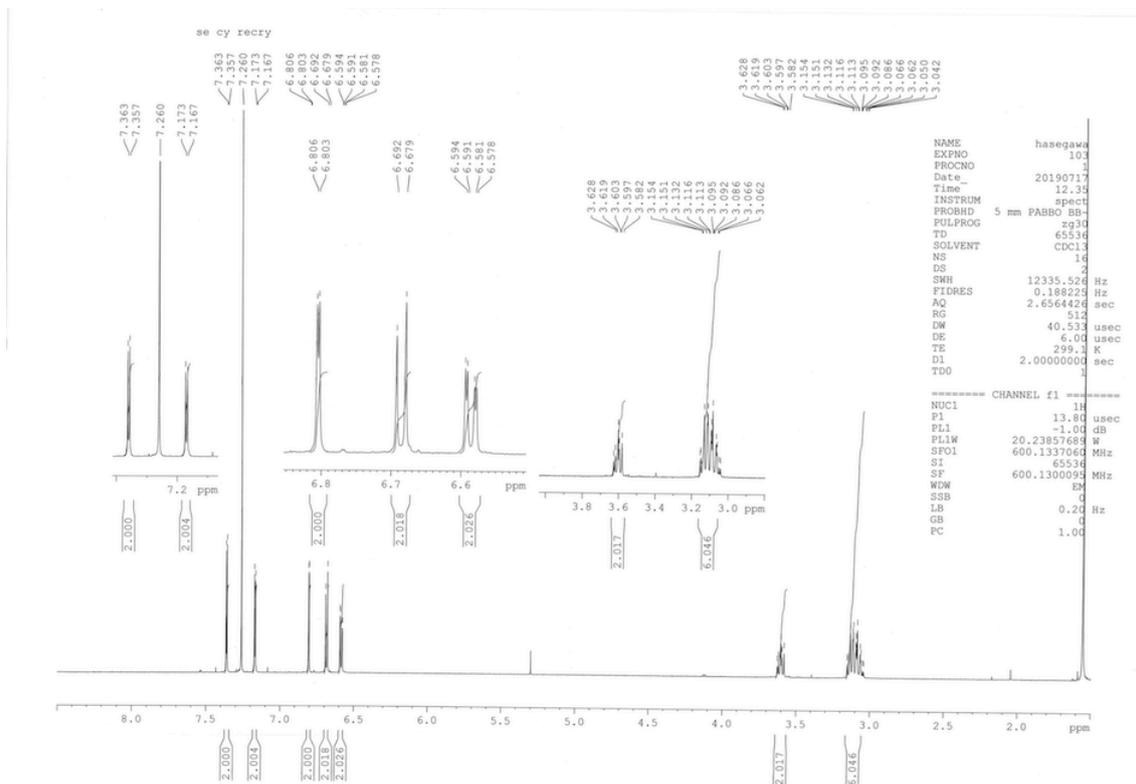


To solution of 2,2'-bipyridyl (78 mg, 0.50 mmol), Ni(cod)_2 (138mg, 0.50 mmol), and (*R_p*)-**5** (91 mg, 0.13 mmol) in THF (2.0 mL) were added under Ar atmosphere. The mixture was stirred for 5 days at 75°C . After the reaction was quenched by the addition of 2M HCl, then filtrated through celite pad. The resultant filtrate was extracted with CHCl_3 . The combined organic phase was washed with saturated brine and dried over Na_2SO_4 . The resulting solution was evaporated, and then the residue was purified by column chromatography on silica gel with the elution of CS_2 hexane- CH_2Cl_2 ($v/v = 1:2:1$). Further purification on GPC with the elution of CHCl_3 gave cyclic dimer (*R_p,R_p*)-**1** (12 mg, 20%) together with insoluble byproducts. The other enantiomer of (*S_p,S_p*)-**1** was prepared from (*S_p*)-**5** in a similar manner (y. 18 %). Data for (*R_p,R_p*)/(*S_p,S_p*)-**1**: Mp 314°C (decomp.); HRMS (ESI-orbitrap): m/z calcd for $\text{C}_{48}\text{H}_{34}\text{Se}_4$ $[\text{M}+\text{H}^+]$: 930.95585, Found: 930.95593; ^1H NMR (600 MHz, CDCl_3) δ 7.36 (d, $J = 3.6$ Hz, 4H), 7.17 (d, $J = 3.6$ Hz, 4H), 6.80 (d, $J = 1.8$ Hz, 4H), 6.68 (d, $J = 8.4$ Hz, 4H), 6.58 (dd, $J = 8.4$ and 1.8 Hz, 4H), 3.58-3.63 (m, 4H), 3.04-3.15 (m, 12H); ^{13}C NMR (150 MHz, CDCl_3) δ 151.1, 145.2, 139.5, 136.4, 136.1, 135.0, 132.9, 132.0, 128.8, 126.0, 34.5, 33.4; ^{77}Se NMR (114 Hz, CDCl_3) δ 631.4; $[\alpha]_{\text{D}_{25}}^{25} = +563$ for (*R_p,R_p*)-**1** (CH_2Cl_2 , $c = 0.8 \times 10^{-3}$) and -561 for (*S_p,S_p*)-**1** (CH_2Cl_2 , c

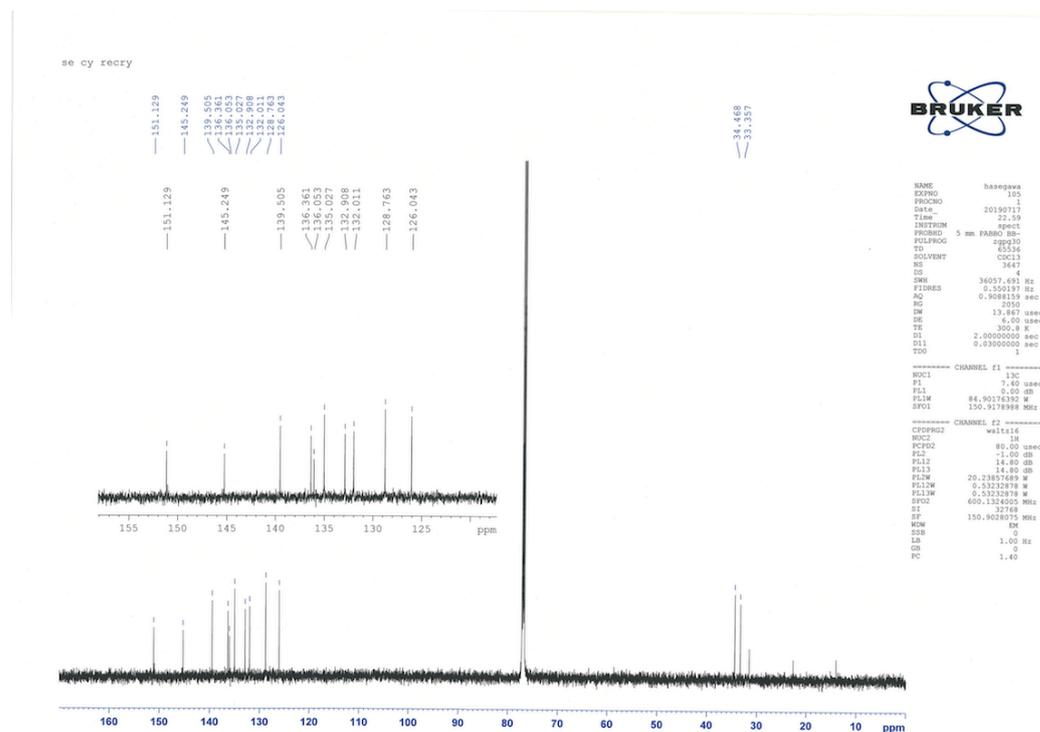
= 0.8×10^{-3} ; IR (KBr) 2925, 2855, 1582, 1462, 1399 cm^{-1}

S3. Figure S2. ¹H, ¹³C and ⁷⁷Se NMR Charts of (*R_p*,*R_p*)-1

(a) ¹H NMR (600 MHz, CDCl₃)

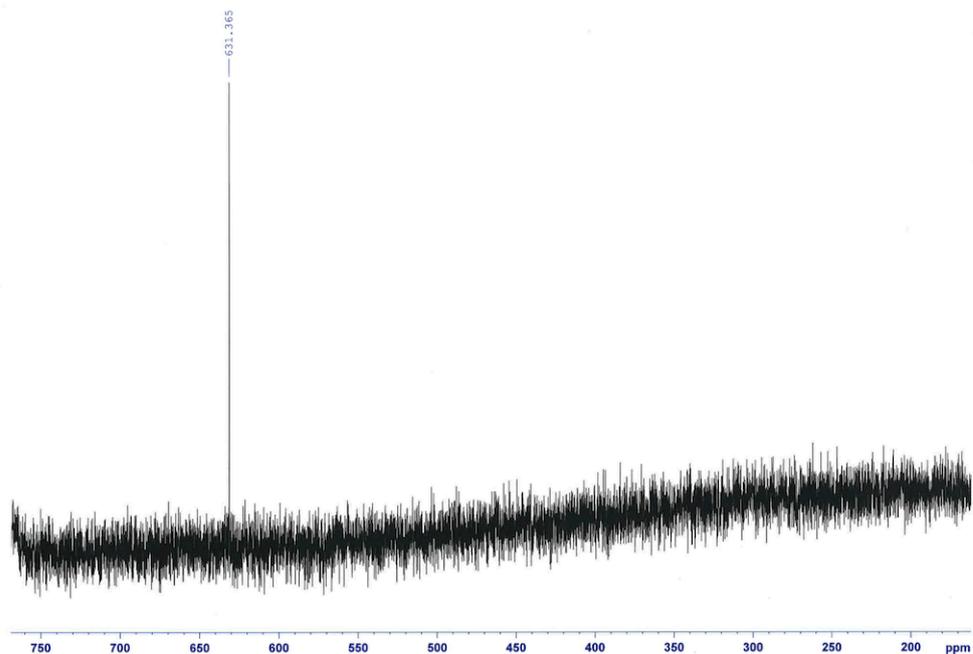


(b) ¹³C NMR (150 MHz, CDCl₃)



(c) ^{77}Se NMR (114 Hz, CDCl_3)

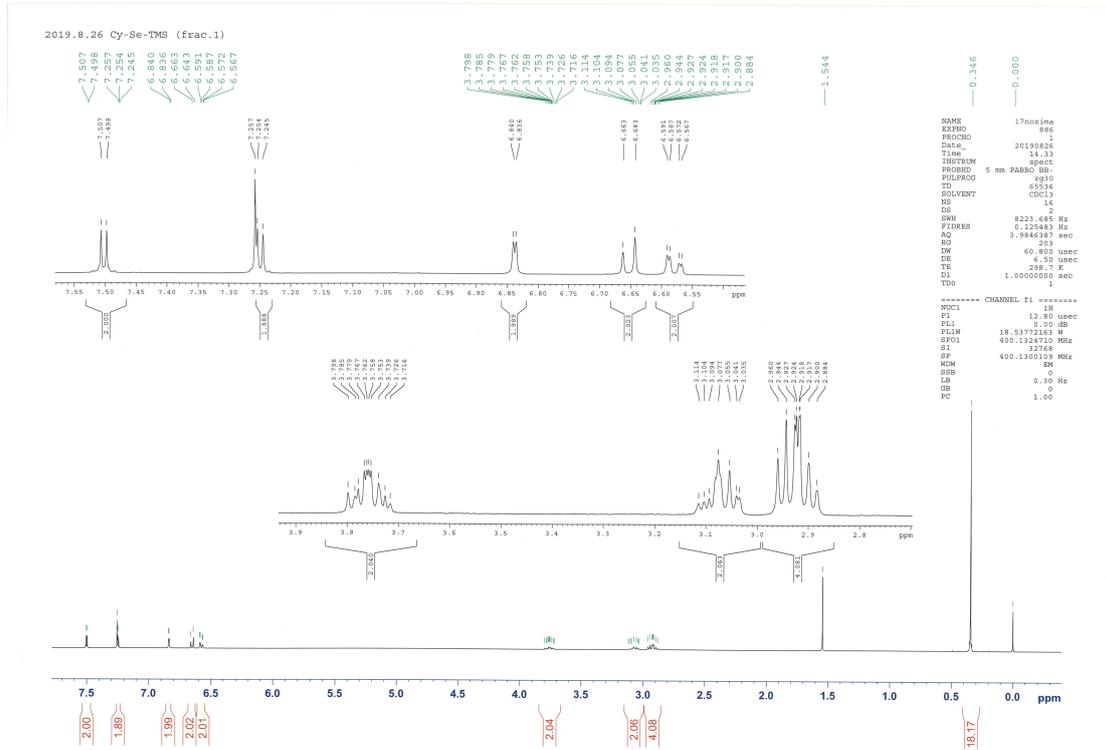
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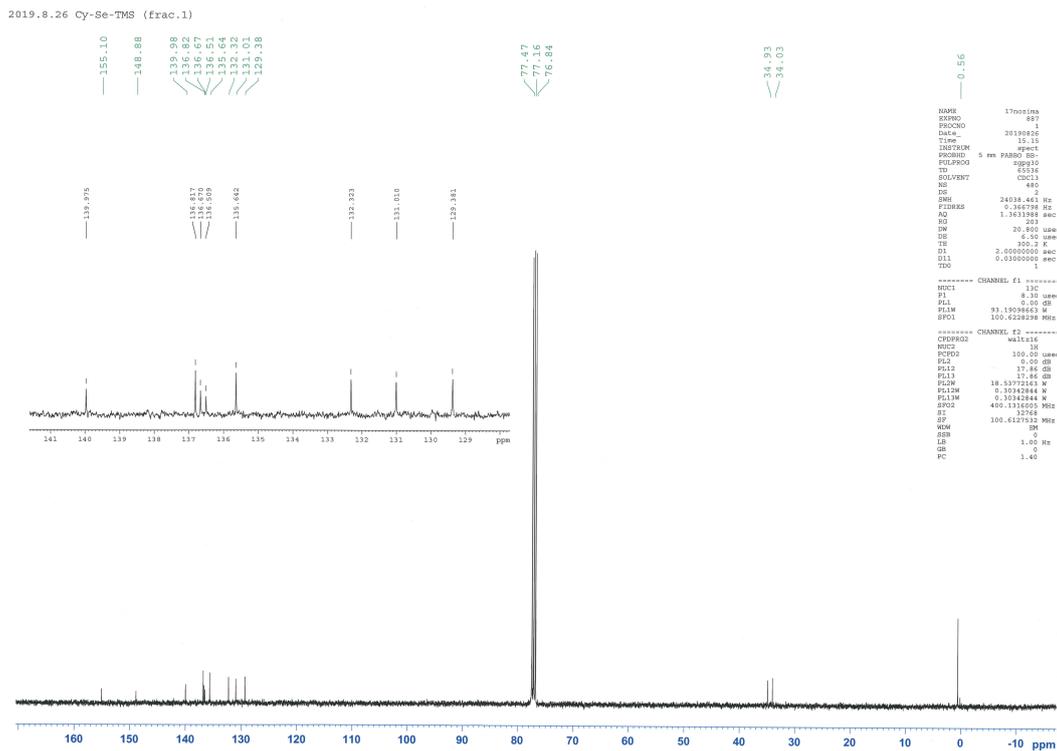
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DE         6.00 usec
TE         298.2 K
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TDO        1
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PL1        0.00 dB
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SI         32768
SF         114.4537347 MHz
WDW        EM
SSB        0
LB         5.00 Hz
GB         0
PC         1.40
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S4. Figure S3, 4 and 5. ¹H, ¹³C and ⁷⁷Se NMR Charts of 4, 4' and 5

(a) ¹H NMR of (R_p,R_p)-4



(b) ¹³C NMR of (R_p,R_p)-4



(c) ^{77}Se NMR of (R_p,R_p) -4

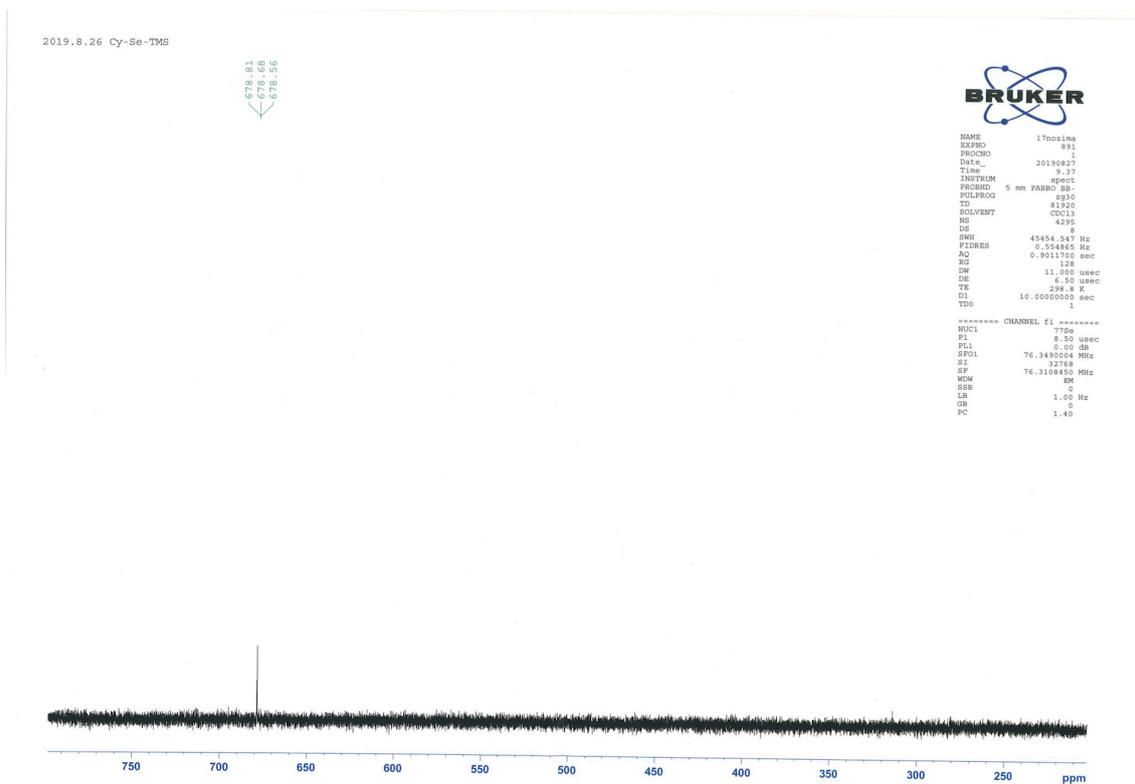
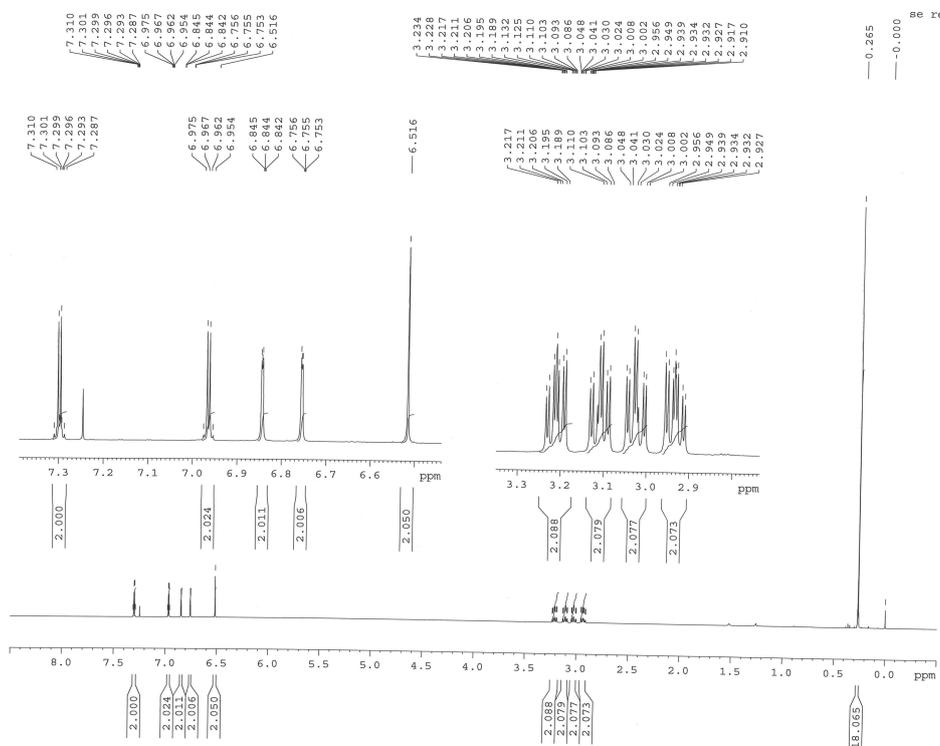
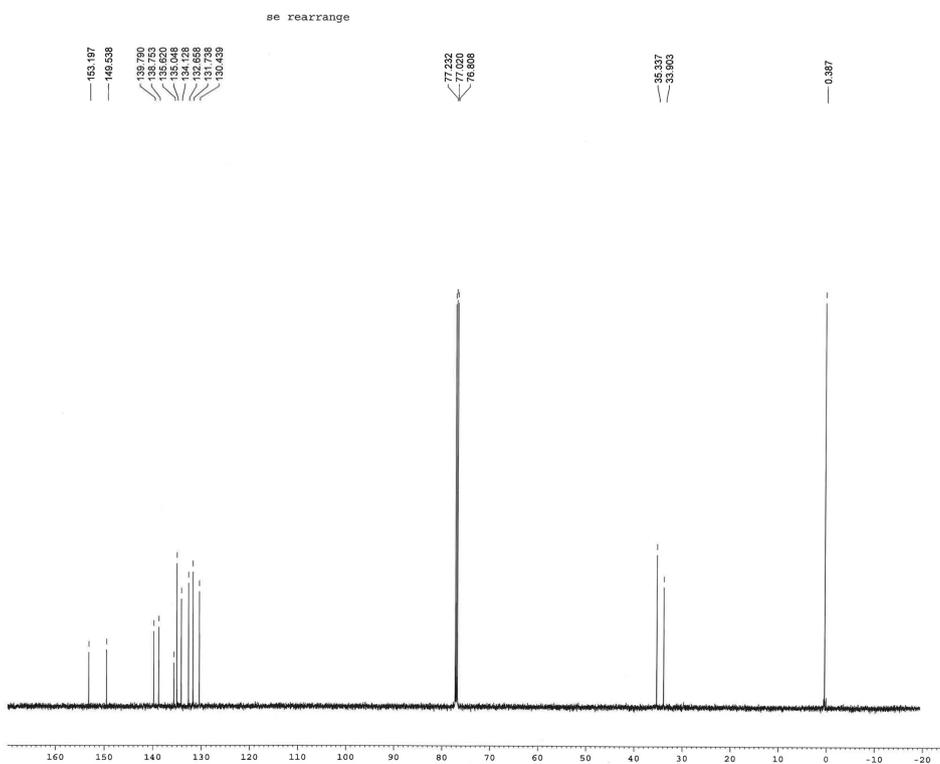


Figure S4.

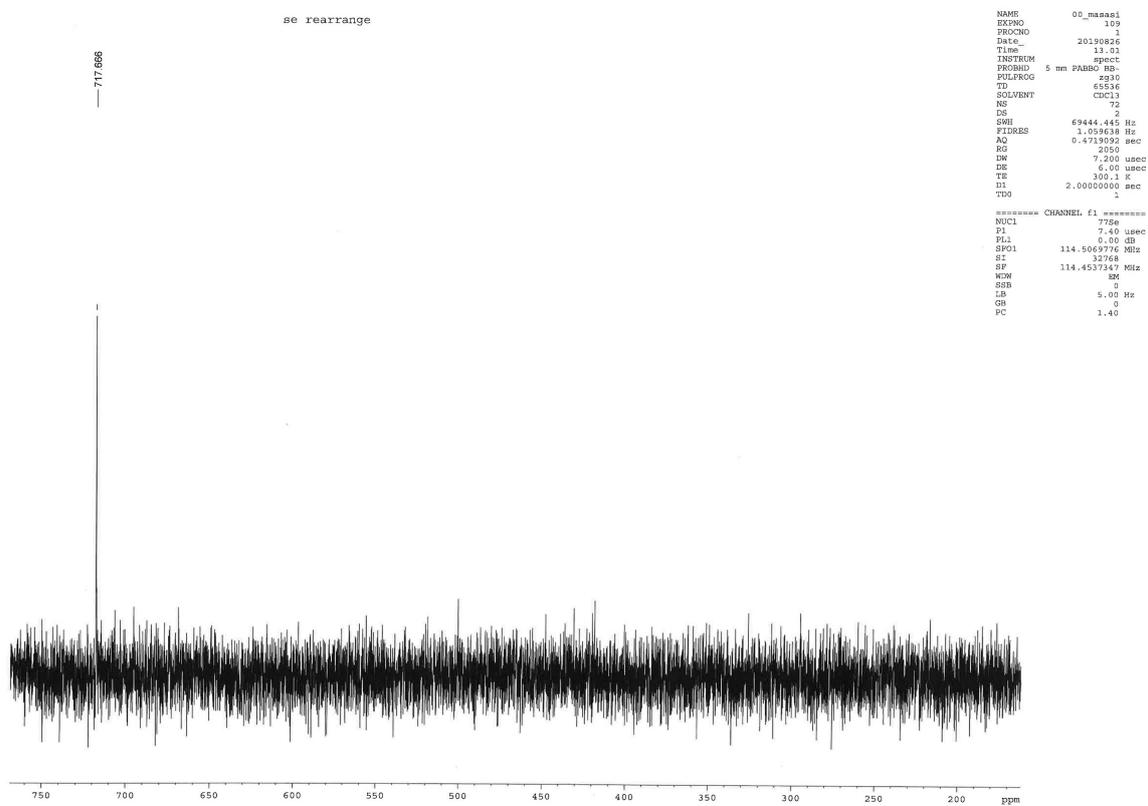
(a) ^1H NMR of **4'**



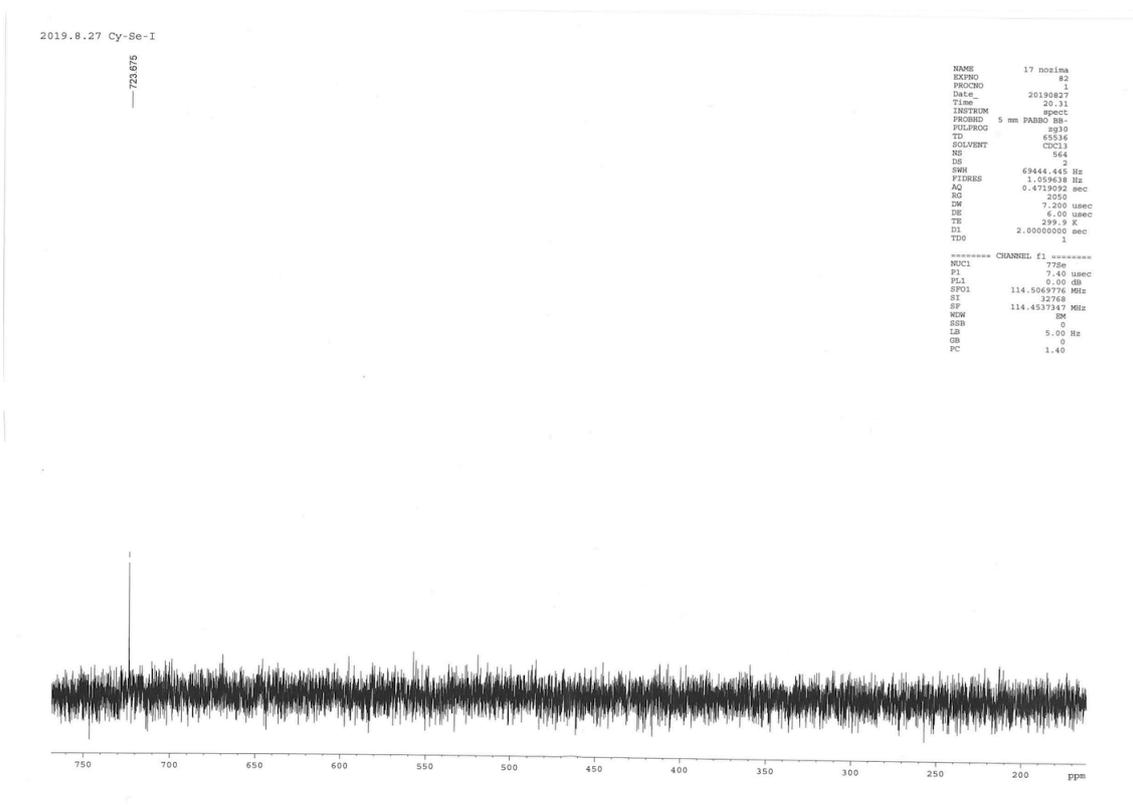
(b) ^{13}C NMR of **4'**



(c) ⁷⁷Se NMR of 4'



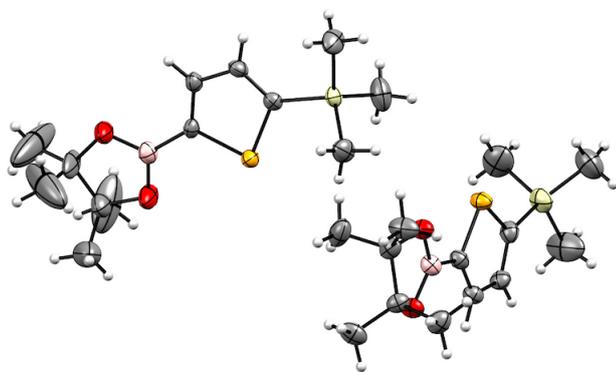
(c) ^{77}Se NMR of **5**



S5. X-ray Analysis of **3**, (*S_p*)-**4** and (*R_p*,*R_p*)-**1**

Single crystals suitable for X-ray analysis were obtained from recrystallization from CH₂Cl₂-hexane (for **3**), CH₂Cl₂-MeOH (for (*S_p*)-**4**) or CHCl₃ (for (*R_p*,*R_p*)-**1**). The intensity data were collected on Rigaku XtaLAB Synergy-R/DW (for **3**) or a Bruker Apex-II CCD diffractometer (**1** and **4**) using monochromated Mo-K α radiation ($\lambda = 0.71073 \text{ \AA}$). The detailed crystallographic data are summarized in Table S1. All structures were solved by the direct method (SHELXT) and refined by full-matrix least-squares method by using Olex2. Non-hydrogen atoms were refined anisotropically. Hydrogen atoms were treated as the riding model.

(a)



(b)

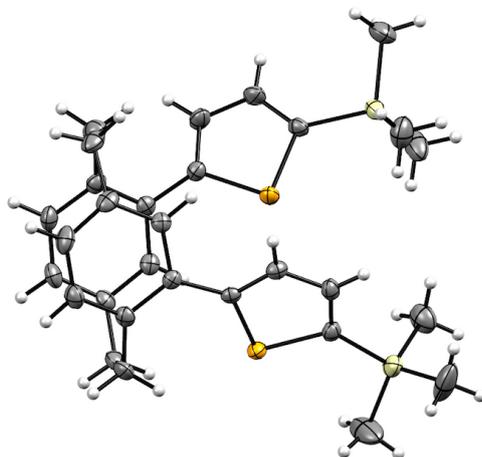


Figure S6. ORTEP Drawing of (a) **3** and (b) (*S_p*)-**4**

Table S1. X-ray Analyses of **3**, (*S_p*)-**4**, and (*R_p*,*R_p*)-**1**

Identification code	3	(<i>S_p</i>)- 4	(<i>R_p</i> , <i>R_p</i>)- 1
Crystal description	Colorless	Colorless	Yellow
Empirical formula	C ₂₆ H ₄₆ B ₂ O ₄ Se ₂ Si ₂	C ₃₀ H ₃₆ Se ₂ Si ₂	C ₄₉ H ₃₇ Cl ₃ Se ₄
Moiety formula	2(C ₁₃ H ₂₃ BO ₂ SeSi)	C ₃₀ H ₃₆ Se ₂ Si ₂	C ₄₈ H ₃₄ Se ₄ •CHCl ₃
Formula weight	658.35	610.69	2124.1(3)
Temperature	173 K	173 K	173 K
Wavelength	Mo- <i>K</i> α 0.71073 Å	Mo- <i>K</i> α 0.71073 Å	Mo- <i>K</i> α 0.71073 Å
Crystal system	Orthorhombic	Orthorhombic	Monoclinic
Space group	<i>Pna</i> 2 ₁ (#33)	<i>P2</i> ₁ 2 ₁ 2 ₁ (#19)	<i>P2</i> ₁ (#4)
Unit cell dimensions	<i>a</i> = 44.640(2) Å <i>b</i> = 6.5823(3) Å <i>c</i> = 11.7225(7) Å <i>α</i> = 90° <i>β</i> = 90° <i>γ</i> = 90°	<i>a</i> = 9.4081(4) Å <i>b</i> = 14.8474(6) Å <i>c</i> = 20.8540(8) Å <i>α</i> = 90° <i>β</i> = 90° <i>γ</i> = 90°	<i>a</i> = 9.6143(9) Å <i>b</i> = 16.007(2) Å <i>c</i> = 13.842(1) Å <i>α</i> = 90° <i>β</i> = 94.363(1)° <i>γ</i> = 90°
Volume	3444.5(3) Å ³	2913.0(2) Å ³	2124.1(3) Å ³
<i>Z</i>	4	4	2
Density (calc.)	1.270 gcm ⁻¹	1.392 gcm ⁻¹	1.639 gcm ⁻¹
Absorption coefficient	2.243	2.637	3.678
<i>F</i> (000)	1360	1248	1036
Crystal size	0.2 x 0.2 x 0.1 cm ³	0.4 x 0.25 x 0.2 cm ³	0.2 x 0.2 x 0.1 mm ³
Index ranges	-56 ≤ <i>h</i> ≤ 63 -9 ≤ <i>k</i> ≤ 7 -14 ≤ <i>l</i> ≤ 11	-9 ≤ <i>h</i> ≤ 12 -19 ≤ <i>k</i> ≤ 19 -28 ≤ <i>l</i> ≤ 18	-12 ≤ <i>h</i> ≤ 12 -14 ≤ <i>k</i> ≤ 21 -17 ≤ <i>l</i> ≤ 17
Reflections collected	20867	17167	12591
Independent reflections	7337 [<i>R</i> _{int} = 0.0717, <i>R</i> _{sigma} = 0.0821]	6914 [<i>R</i> _{int} = 0.0205, <i>R</i> _{sigma} = 0.0464]	7631 [<i>R</i> _{int} = 0.0237, <i>R</i> _{sigma} = 0.0573]
Data/restraints/parameters	7337/1/339	6914/0/313	7631/43/505
Goodness-of-fit on <i>F</i> ²	1.032	0.941	1.035
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0524 <i>wR</i> ₂ = 0.1193	<i>R</i> ₁ = 0.0215, <i>wR</i> ₂ = 0.0460	<i>R</i> ₁ = 0.0335, <i>wR</i> ₂ = 0.0810
Final <i>R</i> indices [all data]	<i>R</i> ₁ = 0.0727 <i>wR</i> ₂ = 0.1271	<i>R</i> ₁ = 0.0257, <i>wR</i> ₂ = 0.0470	<i>R</i> ₁ = 0.0393, <i>wR</i> ₂ = 0.0835
Flack parameters	–	0.007(3)	0.011(7)
CCDC deposition No.	1939516	1939517	1939518

S6. Geometry Optimization of (R_p,R_p)-1

Geometry optimization of (R_p,R_p)-1 were performed by DFT calculation (Gaussian 16, Revision A.03) at B3LYP/6-31G(d,p) level of theory. The minimum energy of each conformation was confirmed by frequency calculations. We obtained three stable conformer of i, ii and iii having a local minimum energy (Table S2). The anti (R_p,R_p)-1-i conformation with C_1 symmetry was found as the most stable geometry. The atomic coordinate of (R_p)-4-i was listed below (Table S3).

Table S2. Geometry and total energy of each conformer in (R_p,R_p)-1

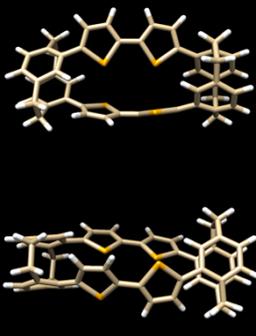
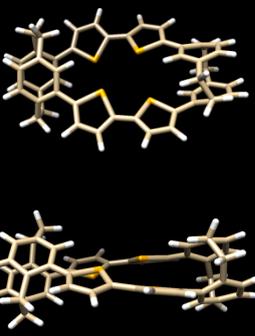
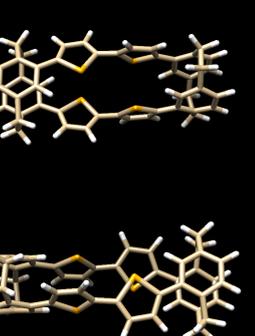
ID code	(R_p,R_p)-1-i	(R_p,R_p)-1-ii	(R_p,R_p)-1-iii
Geometry			
Point Group	C_1	C_2	C_2
# of img	0	0	0
Energy in HF	-11448.2931431	-11448.2880409	-11448.279405
Relative energy (kJ/mol)	0.00	+13.4	+36.0

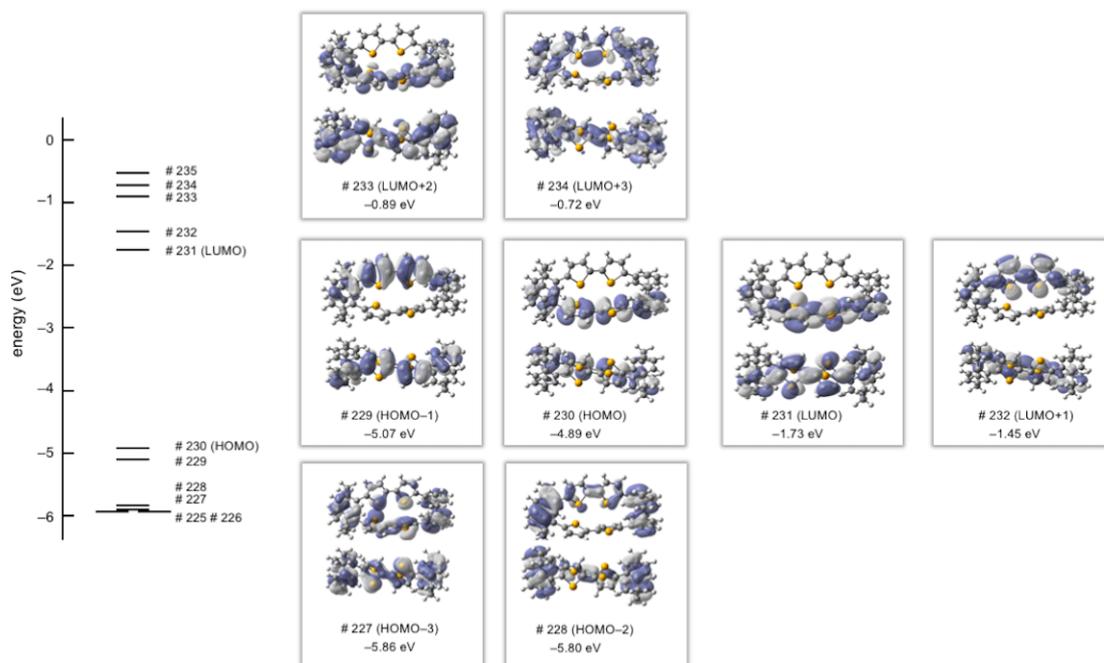
Table S3. Atomic coordinate of optimized structure of (R_p)-4-i

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-4.627274	-1.515430	-0.583341
2	6	0	-4.639795	-0.481821	-1.545237
3	1	0	-3.693236	-0.015678	-1.804806
4	6	0	-5.819266	0.122525	-1.979374
5	6	0	-7.017305	-0.538981	-1.683175
6	1	0	-7.958512	-0.168732	-2.081855
7	6	0	-7.028004	-1.551816	-0.726831
8	1	0	-7.983815	-1.950429	-0.394943

9	6	0	-5.869063	-1.940080	-0.041445
10	6	0	-4.606529	1.663540	0.785263
11	6	0	-4.729319	0.626384	1.728224
12	1	0	-3.817339	0.237244	2.171938
13	6	0	-5.952582	0.000926	1.989853
14	6	0	-7.105383	0.645849	1.524381
15	1	0	-8.086633	0.259034	1.788399
16	6	0	-5.763418	2.076663	0.074124
17	6	0	-6.027270	-2.531364	1.343553
18	1	0	-5.257773	-3.277450	1.559226
19	1	0	-6.988512	-3.051608	1.392462
20	6	0	-6.012862	-1.420916	2.511857
21	1	0	-6.914234	-1.567842	3.115812
22	1	0	-5.162238	-1.614083	3.173604
23	6	0	-5.802044	1.546839	-2.491715
24	1	0	-6.706825	1.740493	-3.076690
25	1	0	-4.951241	1.700072	-3.163353
26	6	0	-5.707873	2.657447	-1.327606
27	1	0	-4.785766	3.219189	-1.476998
28	1	0	-6.536992	3.358172	-1.470852
29	6	0	4.627251	-1.515455	0.583325
30	6	0	4.639777	-0.481858	1.545231
31	1	0	3.693221	-0.015720	1.804819
32	6	0	5.819253	0.122478	1.979376
33	6	0	7.017286	-0.539028	1.683170
34	1	0	7.958494	-0.168783	2.081852
35	6	0	7.027982	-1.551854	0.726810
36	1	0	7.983790	-1.950467	0.394911
37	6	0	5.869041	-1.940099	0.041418
38	6	0	4.606559	1.663575	-0.785248
39	6	0	4.729343	0.626428	-1.728234
40	1	0	3.817356	0.237307	-2.171943
41	6	0	5.952592	0.000956	-1.989858
42	6	0	7.105407	0.645856	-1.524363
43	1	0	8.086651	0.259032	-1.788392
44	6	0	7.003234	1.665590	-0.581056
45	1	0	7.910808	2.044828	-0.116895
46	6	0	5.763440	2.076657	-0.074095
47	6	0	5.802033	1.546786	2.491738
48	1	0	6.706804	1.740423	3.076732
49	1	0	4.951217	1.700010	3.163359

50	6	0	5.707893	2.657417	1.327647
51	1	0	4.785794	3.219172	1.477041
52	1	0	6.537023	3.358123	1.470915
53	6	0	6.027253	-2.531351	-1.343594
54	1	0	5.257762	-3.277436	-1.559293
55	1	0	6.988500	-3.051582	-1.392509
56	6	0	6.012845	-1.420876	-2.511884
57	1	0	6.914200	-1.567824	-3.115858
58	1	0	5.162201	-1.614014	-3.173612
59	6	0	-3.311563	-1.964898	-0.105152
60	6	0	-2.832260	-2.261851	1.152245
61	1	0	-3.462615	-2.266420	2.029988
62	6	0	-1.432265	-2.473395	1.261221
63	1	0	-0.951965	-2.670000	2.215020
64	6	0	-0.714159	-2.379005	0.092246
65	6	0	0.714133	-2.379011	-0.092245
66	6	0	1.432233	-2.473416	-1.261221
67	1	0	0.951944	-2.670022	-2.215024
68	6	0	2.832232	-2.261884	-1.152250
69	1	0	3.462581	-2.266463	-2.029996
70	6	0	3.311539	-1.964930	0.105143
71	34	0	-1.877957	-2.004733	-1.338091
72	34	0	1.877937	-2.004754	1.338090
73	6	0	3.273466	2.216013	-0.502824
74	6	0	2.907701	3.538154	-0.417864
75	1	0	3.635149	4.334132	-0.542027
76	6	0	1.529307	3.795648	-0.203769
77	1	0	1.138596	4.804825	-0.130532
78	6	0	0.714744	2.686743	-0.108723
79	6	0	-0.714724	2.686740	0.108736
80	6	0	-1.529291	3.795639	0.203803
81	1	0	-1.138580	4.804819	0.130581
82	6	0	-2.907681	3.538136	0.417898
83	1	0	-3.635138	4.334103	0.542083
84	6	0	-3.273443	2.215993	0.502836
85	34	0	1.749308	1.116176	-0.274198
86	34	0	-1.749280	1.116167	0.274181
87	6	0	-7.003204	1.665605	0.581089
88	1	0	-7.910781	2.044854	0.116942

S7. Figure S7. MO Diagram of (*R_p*,*R_p*)-1



S8. TD-DFT Calculation of (R_p,R_p)-1

The optimized C_1 (B3LYP/6-31G(d,p)) structure of (R_p,R_p)-1 was applied for excitation energy and simulation of the ECD calculation using time-dependent density functional theory (TD-DFT) with B3LYP/6-311+G(2d,p) basis set. Selected excitation energies and their transition nature are listed as follow.

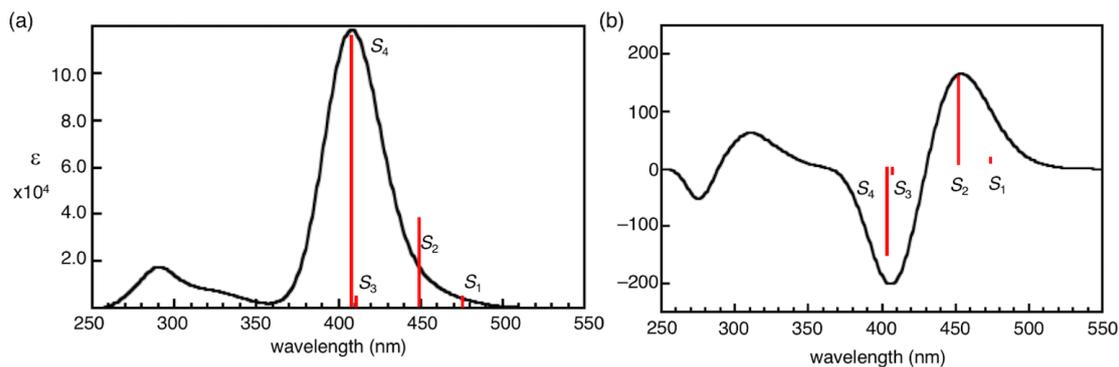


Figure S8. Simulated (a) UV-vis and (b) CD spectra of (R_p,R_p)-1 from TD-DFT calculation (isovalue : 0.15).

Table S4. Selected electronic transition for (R_p,R_p)-1

State	Excitation energies / nm (Oscillator strengths, R_{vel})	Nature	
S_1	476.96 (0.002, 0.0976)	230 -> 231	0.70585
S_2	451.01 (0.0765, 476.829)	229 -> 231	0.59368
		230 -> 232	-0.37490
S_3	409.28 (0.003, -0.5115)	229 -> 232	0.70532
S_4	407.99 (1.3103, -513.8583)	229 -> 231	0.26679
		230 -> 232	0.59178

S9. References

- [s1] G. Meyer-Eppler, E. Vogelsang, C. Benkhäuser, A. Schneider, G. Schnakenburg, A. Lützen, *Eur. J. Org. Chem.*, 2013, **21**, 4523.
- [s2] M. Hasegawa, K. Kobayakawa, H. Matsuzawa, T. Nishinaga, T. Hirose, K. Sako, Y. Mazaki, *Chem. Eur. J.*, 2017, **23**, 326.