

**Theoretical Insights into the Effect of Ligand on Platinum(II) Complexes with
Bidentate Bis(o-carborane) Ligands structure**

Ancong Zhao, Wanlin Cai, Xi Yan, Huize Zhang, Jian Wang, Wei Shen

Table S1. The optimized ground-state geometries of complex 1 obtained by different functionals with the available experimental structure parameters.

	B3LYP	PBE0	BMK	M062X	M052X	CAM- B3LYP	BP86- D3	Exp. ^a
Band length 1	2.065	2.043	2.068	2.029	2.037	2.059	2.044	2.036
Band length 2	2.089	2.075	2.076	2.041	2.047	2.081	2.068	2.073
Band length 5	1.690	1.675	1.728	1.663	1.656	1.669	1.709	1.704
Band length 6	1.529	1.522	1.530	1.526	1.519	1.526	1.527	1.517
Band length 7	1.726	1.721	1.584	1.720	1.725	1.716	1.736	1.725
Band length 3	2.309	2.255	2.259	2.349	2.309	2.277	2.257	2.175
Band length 4	2.146	2.107	2.169	2.173	2.137	2.127	2.110	2.053
Band angle 1-Pt-2	86.22	86.26	85.26	86.69	86.80	86.15	86.56	87.02
Band angle 1-C-5	116.29	116.68	113.52	116.83	116.35	116.34	116.59	115.78
Band angle 5-C-6	110.70	110.68	109.88	110.20	110.66	110.97	109.91	110.73
Band angle 6-C-7	114.29	113.99	115.47	113.69	113.61	114.10	113.95	114.11
Band angle 7-C-2	112.33	112.22	115.70	112.43	111.86	112.26	112.66	112.04
Band angle 1-Pt-4	173.82	174.78	171.71	173.45	174.09	174.55	174.93	176.55
Band angle 2-Pt-3	171.54	171.72	175.82	171.85	170.28	171.66	170.55	168.82
Dihedral angle 5-6-7	4.14	4.44	-2.20	4.41	9.18	4.49	6.18	6.10
Dihedral angle 8-9-10	10.49	8.28	-4.30	10.02	13.68	9.50	6.83	4.43
	11.11%	7.55%	6.20%	10.60%	16.37%	9.33%	4.07%	

^a The experimental value obtains from Reference 2.

Table S2. Main optimized geometry structural parameters of the investigated complexes in the ground state S_0 and the lowest-lying triplet state T_1 along the RMSD at the DFT/BP86-D3 level.

	1		1a		1a-N		1a-O		1a-PO	
	S_0	T_1	S_0	T_1	S_0	T_1	S_0	T_1	S_0	T_1
Band length 1	2.044	2.059	2.038	2.083	2.037	2.059	2.037	2.075	2.042	2.086
Band length 2	2.068	2.082	2.055	2.130	2.057	2.077	2.057	2.165	2.053	2.116
Band length 3	2.257	2.242	2.272	2.212	2.282	2.238	2.277	2.286	2.287	2.216
Band length 4	2.110	2.062	2.110	2.133	2.110	2.102	2.107	2.142	2.105	2.123
Band angle 1-Pt-2	86.56	86.49	87.74	90.03	87.38	87.39	87.41	91.23	87.91	89.30
Band angle 3-Pt-4	82.97	83.07	82.96	83.07	81.69	81.20	81.35	78.90	86.045	87.30
Band angle 1-Pt-4	174.93	175.95	178.75	177.18	178.59	177.80	178.05	174.00	173.35	178.43
Band angle 2-Pt-3	170.55	172.40	168.06	132.09	166.86	161.12	167.79	110.69	170.30	139.77
Dihedral angle 5-6-7	6.18	8.40	4.14	-11.49	4.60	3.54	4.42	-12.43	-0.20	-9.69
Dihedral angle 8-9-10	6.83	-5.82	2.06	-15.65	5.36	18.01	5.32	23.34	12.84	-13.70
RMSD	0.329		0.705		0.915		2.153		1.011	
	1		1b		1b-N		1b-O		1b-PO	
	S_0	T_1	S_0	T_1	S_0	T_1	S_0	T_1	S_0	T_1
Band length 1	2.044	2.059	2.043	2.061	2.024	2.041	2.021	2.038	2.034	2.049
Band length 2	2.068	2.082	2.070	2.084	2.044	2.061	2.041	2.061	2.061	2.078
Band length 3	2.257	2.242	2.245	2.267	2.411	2.399	2.463	2.418	2.304	2.269
Band length 4	2.110	2.062	2.107	2.047	2.144	2.079	2.146	2.087	2.122	2.103
Band angle 1-Pt-2	86.56	86.49	86.53	86.44	88.14	88.14	88.34	88.37	87.15	87.28
Band angle 3-Pt-4	82.97	83.07	75.12	75.63	80.17	81.44	80.54	82.35	77.20	79.11
Band angle 1-Pt-4	174.93	175.95	174.85	174.09	173.82	170.64	173.47	170.05	174.62	174.99
Band angle 2-Pt-3	170.55	172.40	171.19	172.52	176.02	177.85	176.21	178.12	173.70	173.23

Electronic supplementary information

Dihedral angle 5-6-7	6.18	8.40	5.34	6.76	3.27	2.38	3.22	1.19	3.55	1.86
Dihedral angle 8-9-10	6.83	-5.82	1.35	0.43	0.41	-0.08	0.48	-0.39	0.55	0.01
RMSD	0.329	0.103	0.075	0.077	0.056					

Table S3. Simulated emissive spectra of the complex 1 obtained by different functions with the available experimental spectra.

nm	BP86-D3	PBE0	CAM-B3LYP	wB97X	M05X	M062X	Exp.a
λem	790	618	619	580	531	498	497

^aThe experimental value obtains from Reference 2.

Table S4. The analysis of the molecular orbital compositions in the lowest-lying triplet state for all complexes at the TD-DFT/M062X level.

	MO compositions					
	Orbitals	Energy(eV)	Et-bc	Pt	ntp-bpy	Composition contribution>10%
1	L	-2.42	1	3	96	ntp-bpy
	H-1	-8.04	18	57	26	Et-bc+ Pt+ ntp-bpy
	H-5	-8.57	44	8	48	Et-bc+ ntp-bpy
1a	H	-6.24	40	43	17	Et-bc+ Pt+ ntp-bpy
	L	-1.39	10	10	80	Et-bc+ Pt+ ntp-bpy
	L+3	-0.3	16	14	71	Et-bc+ Pt+ ntp-bpy
	L+4	0.12	15	34	51	Et-bc+ Pt+ ntp-bpy
1a-N	L	-1.39	10	10	80	Et-bc+ Pt+ ntp-bpy
	H	-6.24	40	43	17	Et-bc+ Pt+ ntp-bpy
	H-1	-7.98	21	74	5	Et-bc+ Pt
1a-O	H	-6.49	26	50	25	Et-bc+ Pt+ ntp-bpy
	L	-1.58	28	18	54	Et-bc+ Pt+ ntp-bpy
	L+2	-0.58	17	25	58	Et-bc+ Pt+ ntp-bpy
1a-PO	H	-6.76	42	41	16	Et-bc+ Pt+ ntp-bpy
	L	-1.8	6	6	88	ntp-bpy
	L+4	-0.1	22	55	23	Et-bc+ Pt+ ntp-bpy
1b	L	-2.25	1	2	96	ntp-bpy
	H-1	-7.83	11	41	48	Et-bc+ Pt+ ntp-bpy
1b-N	L	-2.12	1	3	96	ntp-bpy
	H	-7.54	30	66	3	Et-bc+ Pt
	H-1	-7.81	17	54	29	Et-bc+ Pt+ ntp-bpy
1b-O	L	-2.36	1	3	96	ntp-bpy
	H	-7.66	32	65	3	Et-bc+ Pt
	H-1	-7.99	21	60	19	Et-bc+ Pt+ ntp-bpy

	H-5	-8.75	20	7	73	Et-bc+ dtp-bpy
	L	-2.79	1	2	97	dtp-bpy
1b-PO	H-1	-8.27	23	59	18	Et-bc+ dtp-bpy
	H-5	-8.86	27	11	62	Et-bc+ dtp-bpy
	H-6	-8.89	48	26	26	Et-bc+ dtp-bpy

Table S5. Transition dipole moments μ_j (Debye) for S_0 - S_n transitions, singlet-triplet splitting energies $\Delta E(S_n-T_1)$ (eV) and the SOC matrix elements $\langle T_1 | H_{\text{SOC}} | S_n \rangle$ (cm^{-1}) of complexes.

	S_n	μ_j	$\Delta E(S_n-T_1)$	$\langle T_1 H_{\text{SOC}} S_n \rangle$		S_n	μ_j	$\Delta E(S_n-T_1)$	$\langle T_1 H_{\text{SOC}} S_n \rangle$
1	S_1	0.11	0.064	29.918					
	S_2	0.18	0.423	1628.919					
	S_3	1.52	0.797	1443.740					
	S_4	0.89	1.171	312.805					
	S_5	0.55	1.207	146.871					
	S_6	0.16	1.282	115.923					
	S_7	0.07	1.298	144.080					
	S_8	0.24	1.337	78.091					
	S_9	0.11	1.419	50.353					
	S_{10}	0.19	1.459	47.585					
1a	S_1	0.73	0.214	47.074	1b	S_1	0.09	0.053	7.645
	S_2	0.27	0.548	79.900		S_2	0.12	0.343	70.858
	S_3	0.49	0.887	91.130		S_3	0.38	0.444	1617.387
	S_4	0.57	1.116	93.561		S_4	0.18	0.691	158.115
	S_5	0.47	1.627	1392.679		S_5	1.42	0.753	1305.894
	S_6	0.59	1.732	422.079		S_6	1.59	1.049	516.842
	S_7	0.17	1.830	1139.549		S_7	1.74	1.130	454.413
	S_8	0.80	2.021	252.019		S_8	0.15	1.265	55.623
	S_9	0.37	2.216	200.323		S_9	0.40	1.281	273.223
	S_{10}	0.61	2.309	222.704		S_{10}	0.15	1.322	40.896
1a-N	S_1	0.21	0.073	183.615	1b-N	S_1	0.24	0.083	216.247
	S_2	0.29	0.345	1448.872		S_2	0.47	0.441	1621.410
	S_3	0.35	0.610	1172.835		S_3	2.14	0.807	1417.158
	S_4	0.78	0.670	746.907		S_4	0.37	1.067	111.384
	S_5	0.24	0.911	500.061		S_5	0.29	1.138	168.396
	S_6	0.80	0.937	384.013		S_6	0.07	1.330	161.727
	S_7	0.64	1.130	50.199		S_7	0.07	1.403	101.442
	S_8	0.82	1.179	148.284		S_8	0.18	1.469	28.322
	S_9	0.86	1.374	41.059		S_9	1.79	1.537	377.533
	S_{10}	0.15	1.396	84.860		S_{10}	0.76	1.600	140.305
1a-O	S_1	0.61	0.169	31.127	1b-O	S_1	0.29	0.082	252.036
	S_2	0.76	0.900	179.100		S_2	0.49	0.437	1606.019
	S_3	0.54	1.156	234.710		S_3	2.07	0.783	1416.350
	S_4	0.32	1.397	1372.351		S_4	0.31	1.120	218.355
	S_5	1.20	1.618	718.850		S_5	0.41	1.176	95.051
	S_6	1.17	1.717	1136.311		S_6	0.16	1.290	206.929
	S_7	0.68	2.058	217.452		S_7	0.24	1.368	171.842

1a-PO	S ₈	1.05	2.098	165.283	1b-PO	S ₈	0.08	1.429	48.091
	S ₉	0.46	2.120	171.222		S ₉	1.59	1.583	51.860
	S ₁₀	0.45	2.219	241.21		S ₁₀	0.14	1.623	264.417
	S ₁	0.70	0.212	45.312		S ₁	0.12	0.042	61.303
	S ₂	0.58	0.571	61.447		S ₂	0.11	0.397	1632.214
	S ₃	0.47	0.889	88.369		S ₃	1.23	0.726	1399.484
	S ₄	0.35	1.035	110.091		S ₄	0.57	0.768	429.292
	S ₅	0.50	1.367	1360.640		S ₅	0.30	1.111	145.111
	S ₆	0.25	1.612	1088.240		S ₆	0.56	1.120	82.949
	S ₇	0.48	1.769	127.664		S ₇	0.03	1.232	154.310
S ₈	0.47	1.959	204.027	S ₈	0.04	1.306	92.081		
S ₉	0.48	2.018	365.148	S ₉	0.06	1.328	124.059		
S ₁₀	0.58	2.071	298.735	S ₁₀	0.21	1.406	52.755		

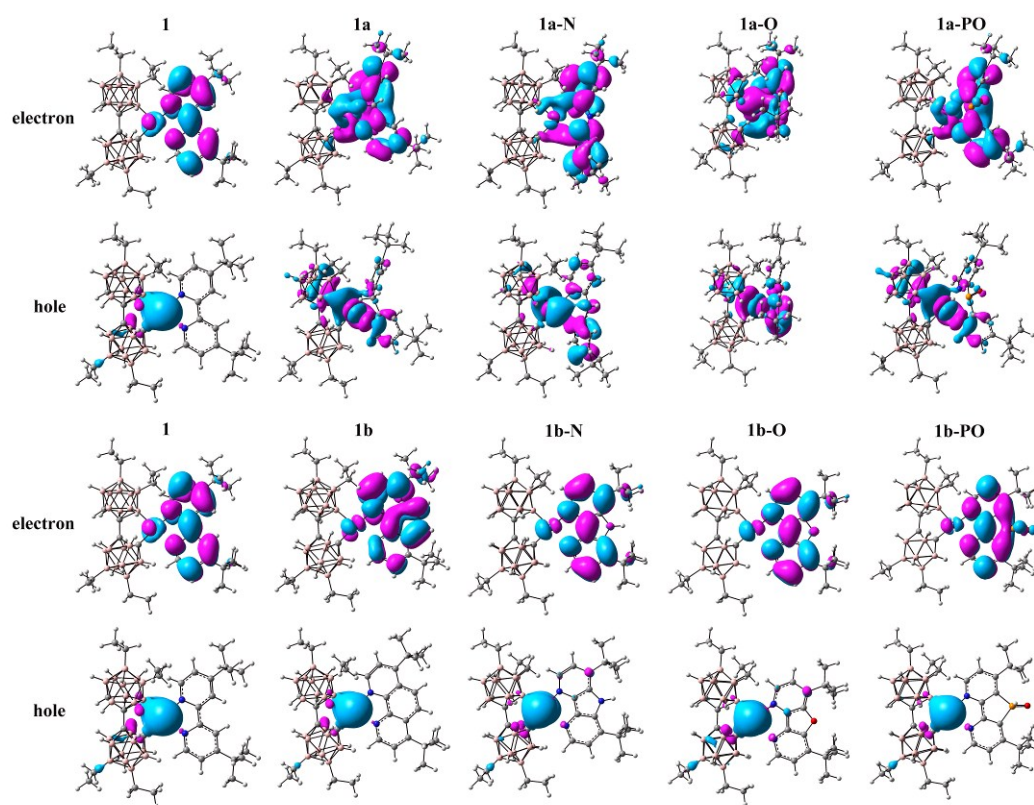
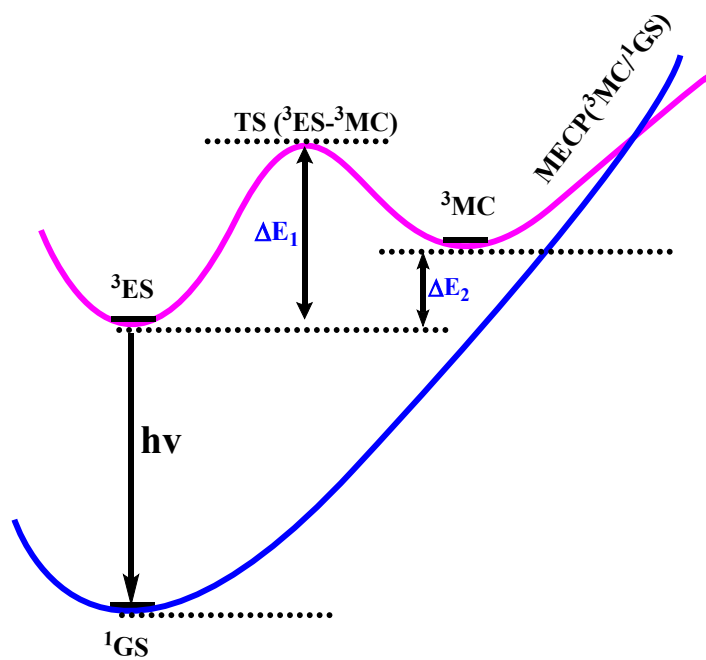


Figure 2 NTO pairs calculated at the excited T₁ state for the two class complexes



Scheme 2 The potential energy curves of thermal decay channels