Electronic Supplementary Information (ESI)

Coordination-insertion polymerization of polar allylbenzene monomers

Xiaoqiang Hu,^{a,b,#} Xin Ma^{a,b,#} and Zhongbao Jian^{a*}

^aState Key Laboratory of Polymer Physics and Chemistry, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, Renmin Street 5625, Changchun 130022, China

^bUniversity of Science and Technology of China, Hefei 230026, China

[#] These authors contributed to this work equally.

Contents

1. General information	S 2
2. General procedures for the polymerizations	S 3
3. NMR figures of copolymers	S 4
4. GPC traces and DSC data of copolymers	S21
5. References	S34

General Procedures and Materials: All syntheses involving air- and moisture sensitive compounds were carried out using standard Schlenk-type glassware (or in a glove box) under an atmosphere of nitrogen. All solvents were purified from the MBraun SPS system. NMR spectra for monomers and polymers were recorded on a Bruker AV400 (¹H: 400 MHz, ¹³C: 100 MHz) or a Bruker AV500 (¹H: 500 MHz, ¹³C: 125 MHz). NMR assignments were confirmed by ¹H–¹H COSY, ¹H–¹³C HSQC and ¹H–¹³C HMBC experiments when necessary. The molecular weights (M_n) and molecular weight distributions (M_w/M_n) of copolymers were measured by means of gel permeation chromatography (GPC) on a PL-GPC 220-type high-temperature chromatograph equipped with three PL-gel 10 µm Mixed-B LS type columns at 150 °C. Melting points (Tm) of copolymers were measured through DSC analyses, which were carried out on a Q 100 DSC from TA Instruments under a nitrogen atmosphere at heating and cooling rates of 10 °C/min (temperature range: 20–180 °C).

Materials: $[(2,6^{-i}PrC_6H_3NC(Me)C(Me)N-2,6^{-i}PrC_6H_3)Pd(CH_3)(Cl)]^1$, $Pd(P(C_6H_4OCH_3)_2-C_6H_4SO_3)(CH_3)(DMSO)^2$, $Pd(P(C_6H_{11})_2C_6H_4SO_3)(CH_3)(SO(CH_3)_2)^3$, 2-methoxy allylbenzene⁴, 2-acetoxy allylbenzene⁵, 2-allylbenzaldehyde⁶, 2-allylbromobenzene⁶ and 2-(*N*,*N*-dimethylamino) allylbenzene⁷ were prepared according to the literature procedures. All other reagents were commercially available and used as received.

A general procedure for the copolymerization of polar monomer with ethylene.

In a typical experiment, a 150 mL glass pressure reactor connected with a high pressure gas line was firstly dried at 90 $^{\circ}$ C under vacuum for at least 1 h. The glass reactor was then adjusted to the desired polymerization temperature. 23 mL of toluene and the desired polar allylbenzene monomer were added to the reactor under N₂ atmosphere, then the desired amount of Pd(II) catalyst in 2 mL of CH₂Cl₂ was injected into the polymerization system via syringe subsequently. With a rapid stirring, the reactor was pressurized and maintained at the desired pressure of ethylene. After 1 h, the pressure reactor was vented and the copolymer was precipitated in ethanol, filtered and dried at 50 $^{\circ}$ C for at least 24 h under vacuum.



Figure S1. ¹H NMR spectrum (500 MHz, 298 K, CDCl₃) of 2-methoxy allylbenzene.



Figure S2. ¹H NMR spectrum (500 MHz, 298 K, CDCl₃) of 2-acetoxy allylbenzene.



Figure S3. ¹H NMR spectrum (500 MHz, 298 K, CDCl₃) of 2-allylbenzaldehyde.



Figure S4. ¹H NMR spectrum (500 MHz, 298 K, CDCl₃) of 2-allylbromobenzene.



Figure S5. ¹H NMR spectrum (500 MHz, 298 K, CDCl₃) of **2-**(*N*,*N*-dimethylamino) allylbenzene.



Figure S6. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 4.



Figure S7. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 5.



Figure S8. ¹³C NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 5.



Figure S9. 1 H- 1 H COSY NMR spectrum (400 MHz, C₂D₂Cl₄, 110 °C) of the copolymer from table 1, entry 5.



Figure S10. ¹H-¹³C HSQC NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 5.



Figure S11. ¹H-¹³C HMBC NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 5.



Figure S12. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 6.



Figure S13. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 7.



Figure S14. ¹³C NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 7.



Figure S15. ¹H-¹H COSY NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 7.



Figure S16. ¹H-¹³C HSQC NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 7.



Figure S17. ¹H-¹³C HMBC NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 7.



Figure S18. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 8.



Figure S19. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 9.



Figure S20. ¹³C NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 9.



Figure S21. ¹H-¹H COSY NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 9.



Figure S22. 1 H- 13 C HSQC NMR spectrum (400 MHz, C₂D₂Cl₄, 110 °C) of the copolymer from table 1, entry 9.



Figure S23. ¹H-¹³C HMBC NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 9.



Figure S24. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 10.



Figure S25. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 11.



Figure S26. ¹³C NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 11.



Figure S27. 1 H- 1 H COSY NMR spectrum (400 MHz, C₂D₂Cl₄, 110 °C) of the copolymer from table 1, entry 11.



Figure S28. ¹H-¹³C HSQC NMR spectrum (400 MHz, C₂D₂Cl₄, 110 °C) of the copolymer from table 1, entry 11.



Figure S29. ¹H-¹³C HMBC NMR spectrum (400 MHz, C₂D₂Cl₄, 110 °C) of the copolymer from table 1, entry 11.



Figure S30. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 12.



Figure S31. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 13.



Figure S32. ¹³C NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 13.



Figure S33. ¹H-¹H COSY NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 13.



Figure S34. ¹H-¹³C HSQC NMR spectrum (400 MHz, C₂D₂Cl₄, 110 °C) of the copolymer from table 1, entry 13.



Figure S35. ¹H-¹³C HMBC NMR spectrum (400 MHz, C₂D₂Cl₄, 110 °C) of the copolymer from table 1, entry 13.



Figure S36. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 14.



Figure S37. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 15.



Figure S38. ¹H NMR spectrum (400 MHz, $C_2D_2Cl_4$, 110 °C) of the copolymer from table 1, entry 17.



Figure S39. GPC trace of the copolymer from table 1, entry 4.



Figure S40. GPC trace of the copolymer from table 1, entry 5.



Figure S41. GPC trace of the copolymer from table 1, entry 6.



Figure S42. GPC trace of the copolymer from table 1, entry 7.



Figure S43. GPC trace of the copolymer from table 1, entry 8



Figure S44. GPC trace of the copolymer from table 1, entry 9



Figure S45. GPC trace of the copolymer from table 1, entry 10



Figure S46. GPC trace of the copolymer from table 1, entry 11



Figure S47. GPC trace of the copolymer from table 1, entry 12



Figure S48. GPC trace of the copolymer from table 1, entry 13



Figure S49. GPC trace of the copolymer from table 1, entry 14







Figure S50. GPC trace of the copolymer from table 1, entry 15



Figure S51. GPC trace of the copolymer from table 1, entry 17



Figure S52. DSC data of the polymer from table 1, entry 4.



Figure S53. DSC data of the polymer from table 1, entry 8



Figure S54. DSC data of the polymer from table 1, entry 10



Figure S55. DSC data of the polymer from table 1, entry 11



Figure S56. DSC data of the polymer from table 1, entry 12



Figure S57. DSC data of the polymer from table 1, entry 14



Figure S58. DSC data of the polymer from table 1, entry 15



Figure S59. DSC data of the polymer from table 1, entry 17

References

(1) L. K. Johnson, S. Mecking and M. Brookhart, J. Am. Chem. Soc., 1996, 118, 267-268.

(2) H. Leicht, I. Göttker-Schnetmann and S. Mecking, *Angew. Chem. Int. Ed.*, 2013, **52**, 3963-3966.

(3) D. Guironnet, P. Roesle, T. Rünzi, Göttker-Schnetmann and S. Mecking, *J. Am. Chem. Soc.*, 2009, **131**, 422–423.

(4) M. J. Gresser, S. M. Wales and P. A. Keller, Tetrahedron, 2010, 66, 6965-6976.

(5) T. N. Gieshoff, M. Villa, A. Welther, M. Plois, U. Chakraborty, R. Wolf and A. J. von Wangelin, *Green Chem.*, 2015, **17**, 1408-1413.

(6) I. D. G. Watson, S. Ritter and F. D. Toste, J. Am. Chem. Soc., 2009, 131, 2056–2057.

(7) H. F Jiang, W. F. Yang, H. J. Chen, J. X. Li and W. Q. Wu, *Chem. Commun.*, 2014, **50**, 7202-7204.