Anisotropic Polymer Nanoparticles with Solvent and Temperature Dependent Shape and Size from Triblock Copolymers

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1. NMR Spectra H₁ - Phth-poly(NIPAM)₇₉-TTC



Figure S1: ¹H-NMR of H₁ (400 MHz, CDCl₃, 303 K) *δ* = 7.75 (m, 2H, ArH), 7.67 (m, 2H, ArH), 6.22 (br, 79 H, N-H), 4.62 (s, 1 H, -CH(S)C), 4.00 (s, 79 H, -CH(NH)(CH₃)₂), 3.73-3.63 (m, 2H, -CH₂-Phth), 3.34 (br, 2H, -CH₂S(CS₂)), 2.5-1 (br, acrylic backbone), 1.07 (s, (CH₃)₂-CHN-), 0.93 (m, -CH₃) ppm.

D₁ - Phth-poly[(NIPAM)₇₉-b-(PEGA)₆₆]-TTC



Figure S2: ¹H-NMR of D₁ (400 MHz, CDCl₃, 303 K) δ = 7.82 (m, ArH), 7.73 (m, ArH), 4.16 (s, -CH₂OC(O)), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-), 0.93 (m,-CH₃) ppm.

D₁ - H₂N-poly[(NIPAM)₇₉-b-(PEGA)₆₆]-H



Figure S3: ¹H-NMR of D₁ after end cleavages (400 MHz, CDCl₃, 303 K) δ = 4.16 (s, -CH₂OC(O)), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-) ppm.

D₂ - Phth-poly[(NIPAM)₇₉-b-(PEGA)₉]-TTC



Figure S4: ¹H-NMR of D₂ (400 MHz, CDCl₃, 303 K) δ = 7.82 (m, ArH), 7.73 (m, ArH), 4.16 (s, -CH₂OC(O)), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-), 0.93 (m,-CH₃) ppm.

D₂ - H₂N-poly[(NIPAM)₇₉-b-(PEGA)₉]-H



Figure S5: ¹H-NMR of D₂ after end cleavages (400 MHz, CDCl₃, 303 K) δ = 4.16 (s, -CH₂OC(O)), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-) ppm.

T_{1.1} - poly[(BLG)₆₀-*b*-(NIPAM)₇₉-*b*-(PEGA)₆₆]



Figure S6: ¹H-NMR of T _{1.1} (400 MHz, CDCl₃, 303 K) *δ* = 7.25 (m, Bz), 5.03 (br, -CH₂Ph), 4.17 (s, -CH₂OC(O) PEGA unit), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN) ppm.

T_{1.2} - poly[(BLG)₃₀-b-(NIPAM)₇₉-b-(PEGA)₆₆]



Figure S7: ¹H-NMR of T _{1.2} (400 MHz, CDCl₃, 303 K) *δ* = 7.25 (m, Bz), 5.03 (br, -CH₂Ph), 4.17 (s, -CH₂OC(O) PEGA unit), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-) ppm.

T_{1.3} - poly[(BLG)₇-b-(NIPAM)₇₉-b-(PEGA)₆₆]



Figure S8: ¹H-NMR of T _{1.3} (400 MHz, CDCl₃, 303 K) δ = 7.25 (m, Bz), 5.03 (br, -CH₂Ph), 4.17 (s, -CH₂OC(O) PEGA unit), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-) ppm.

T_{2.1} - poly[(BLG)₆₀-b-(NIPAM)₇₉-b-(PEGA)₉]



Figure S9: ¹H-NMR of T _{2.1} (400 MHz, CDCl₃, 303 K) δ = 7.25 (m, Bz), 5.03 (br, -CH₂Ph), 4.17 (s, -CH₂OC(O) PEGA unit), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-) ppm.

T_{2.2} - poly[(BLG)₃₀-*b*-(NIPAM)₇₉-*b*-(PEGA)₉]



Figure S10: ¹H-NMR of T _{2.2} (400 MHz, CDCl₃, 303 K) δ = 7.25 (m, Bz), 5.03 (br, -CH₂Ph), 4.17 (s, -CH₂OC(O) PEGA unit), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-) ppm.

T_{2.3} - poly[(BLG)₇-b-(NIPAM)₇₉-b-(PEGA)₉]



Figure S 11: ¹H-NMR of T _{2.3} (400 MHz, CDCl₃, 303 K) δ = 7.25 (m, Bz), 5.03 (br, -CH₂Ph), 4.17 (s, -CH₂OC(O) PEGA unit), 4.00 (s, -CH(NH)(CH₃)₂), 3.65 (m, -OCH₂CH₂O-), 3.38 (s, -OCH₃), 2.2-1.2 (br, alkyl chain), 1.13 (s, (CH₃)₂CHN-) ppm.

2. GPC



Figure S12: SEC trace of Phth-poly[(NIPAM)₇₉-*b*-(PEGA)₆₆]-TTC Phth-D₁-TTC (black) and poly[(NIPAM)₇₉-*b*-(PEGA)₆₆] (red). (DMF/LiBr, MALLS detection, 1 mL min⁻¹, molecular mass determined against poly(styrene) standards).



Figure S13: SEC traces (DMF/LiBr, dRI detection, molecular masses determined against poly(styrene) standards) of T_{1.1}. $\overline{M}_w = 17.3 \times 10^3 \text{ g mol}^{-1}$; $\overline{M}_n = 15.1 \times 10^3 \text{ g mol}^{-1} D = 1.15$.



Figure S14: SEC traces (DMF/LiBr, dRI detection, molecular masses determined against poly(styrene) standards) of T_{1.2}. $\bar{M}_w = 13.7 \times 10^3 \text{ g mol}^{-1}$; $\bar{M}_n = 11.3 \times 10^3 \text{ g mol}^{-1} D = 1.21$.



Figure S15: SEC traces (DMF/LiBr, dRI detection, molecular masses determined against poly(styrene) standards) of T_{1.3}. $\bar{M}_w = 13.9 \times 10^3 \text{ g mol}^{-1}$; $\bar{M}_n = 11.1 \times 10^3 \text{ g mol}^{-1} D = 1.26$.



Figure S16: SEC traces (DMF/LiBr, dRI detection, molecular masses determined against poly(styrene) standards) of T_{2.1}. $\bar{M}_w = 13.3 \times 10^3 \text{ g mol}^{-1}$; $\bar{M}_n = 12.3 \times 10^3 \text{ g mol}^{-1} D = 1.08$.



Figure S17: SEC traces (DMF/LiBr, dRI detection, molecular masses determined against poly(styrene) standards) of T_{2.2}. $\bar{M}_w = 8.60 \times 10^3 \text{ g mol}^{-1}$; $\bar{M}_n = 7.84 \times 10^3 \text{ g mol}^{-1}$ D = 1.10.



Figure S18: SEC traces (DMF/LiBr, dRI detection, molecular masses determined against poly(styrene) standards) of T_{2.3}. $\bar{M}_w = 6.32 \times 10^3 \text{ g mol}^{-1}$; $\bar{M}_n = 5.76 \times 10^3 \text{ g mol}^{-1} D = 1.10$.

3. FTIR spectra



Figure S19: FTIR spectra of solid polymers, dashed lines representing the amide I and II peaks at 1546 and 1650 cm⁻¹.

4. CD spectra

4.1 CD spectra of particles from THF procedure



Figure S20: CD spectra of polymers solutions from solvent exchange procedure with THF.

4.2 CD spectra of particles from HFIP procedure



Figure S21: CD spectra of polymers dissolved in HFIP.



Figure S22: CD spectra of polymers solutions from solvent exchange procedure with HFIP.

5. AF4 experiments



Figure S23: Raw Data Fractogram of AF4 - MALS & UV sample T2.1. The fractogram shows a system peak at ~7 min.



Figure S24: Data fitting of region I and II of T2.1 using a sphere fit and a random coil fit (see Figure 4, main manuscript).



Figure S25: Differential Particle Size Distribution (red trace) and cumulative Particle Size Distribution (blue trace) of sample T2.1.



Figure S26: Overlay of the hydrodynamic radius (R_h from DLS; red), the radius of gyration (R_g from MALS; blue) and the concentration profile (green) of sample T2.1.



Figure S27: Raw Data Fractogram of AF4 – MALS & UV sample T2.2. The fractogram shows a system peak at ~7 min.



Figure S28: Differential Particle Size Distribution (red trace) and cumulative Particle Size Distribution (blue trace) of sample T2.2.



Figure S29: Overlay of the ratio of the R_g to R_h (purple) and the concentration profile (green) against elution time of sample T2.2.



Figure S30: Overlay of the hydrodynamic radius (R_h from DLS; red), the radius of gyration (R_g from MALS; blue) and the concentration profile (green) of sample T2.2.

6. DLS experiments

6.1 DLS from THF solvent exchange procedure



Figure S31: DLS traces of $T_{1,1}$ particles solutions obtained from THF.



Figure S32: DLS traces of T_{1.2} particles solutions obtained from THF.



Figure S33: DLS traces of T_{1.3} particles solutions obtained from THF.



Figure S34: DLS traces of T 2.1 particles solutions obtained from THF.



Figure S35: DLS traces of T 2.2 particles solutions obtained from THF.



Figure S36: DLS traces of T 2.3 particles solutions obtained from THF.



6.2 DLS from HFIP solvent exchange procedure

Figure S37: DLS traces of T 2.1 particles solutions obtained from HFIP.



Figure S38: DLS traces of T $_{2.2}$ particles solutions obtained from HFIP.