Supplementary information

Influence of different sequences of L-proline dipeptide derivatives in the pendants on the helix of poly(phenylacetylene)s and their enantioseparation

properties

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Figure S1. ¹H NMR spectrum of the intermediate product (i) in CDCl₃ at 20 °C.



Figure S2. ¹H NMR spectrum of the intermediate product (ii) in CDCl₃ at 20 °C.



Figure S3. ¹H NMR spectrum of PA-1 in CDCl₃ at 20 °C.



Figure S4. ¹H NMR spectrum of PA-2 in CDCl₃ at 20 °C.



Figure S5. ¹H NMR spectrum of **PPA-1** in DMSO- d_6 at 80 °C.



Figure S6. ¹H NMR spectrum of PPA-2 in DMSO- d_6 at 80 °C.



Figure S7. IR spectra of (A) monomer PA-1 and (B) its polymer PPA-1.



Figure S8. IR spectra of (A) monomer PA-2 and (B) its polymer PPA-2.



Figure S9. Temperature dependence of the CD and UV-Vis spectra of (A) PPA-

1 in DMF (B) **PPA-2** in DMSO (c = 1 mg/mL).

Dynamic Light Scattering Measurements (DLS). Dynamic Light Scattering Measurements (DLS). DLS measurements were performed on a Nicomp 380ZLS (PSS. NICOPMP, USA) equipped with a 15 mW Laser at 25 °C. The obtained diameters of **PPA-1** in CHCl₃ (29 nm) was almost comparable to that in the other solvents (26-32 nm). And for **PPA-2**, the obtained diameters of **PPA-2** in the solvents were in the range of 30-33 nm. These results indicate that the polymers could be well soluble in the solvents and the formation of aggregates of the polymer main chains in the presence of a polar solvent such as DMF can be excluded.