LED-induced Controlled Radical Polymerization with an in situ bromine-iodine transformation and block polymerization combined with Ring-opening Polymerization using one organocatalyst

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Figure S1. The picture of white LED light (380 -780 nm, 13 W/m, 1.5 mW/cm²): actual light intensity (at the position of the reaction solution) experimentally measured by a luminometer.



Figure S2. ¹H NMR spectrum of Ph-BrOH in CDCl₃.



Figure S3. Color changes in the solution. The condition sees in Figure 2.



Figure S4. UV-vis spectra of CP-Br (0.02 mmol) with NaI (0.022 mmol) and HMPA (0.02 mmol) (blue line), and CP-Br (0.02 mmol) with NaI (0.022 mmol) and HMPA (0.02 mmol) in LED light 1h (black line). The solvent was MMA in all cases.



Figure S5. UV-vis spectra of CP-Br (0.02 mmol), (0.022 mmol) with TBA (0.005 mM) (black line), and CP-Br (0.02 mmol) with NaI (0.022 mmol) and TBA (0.005 mM) in LED light 1 h (purple line). The solvent was MMA in all cases.



Figure S6. Plot of $In([M]_0/[M])$ vs t (left) and M_n and M_w/M_n vs conversion (right) for the solution(entry3,entry4 in table1). The symbols are indicated in the figure.



Figure S7. Plot of $In([M]_0/[M])$ vs t (left) and M_n and M_w/M_n vs conversion (right) for the solution(entry6,entry7 in table1). The symbols are indicated in the figure.