## **Supporting Information**

## Synthesis and Supramolecular Assembly of MRI Active Biogenic Amine Host Polymers

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### 1. General information

All metathesis reactions were conducted under nitrogen atmosphere using *Schlenk*-technique or under argon using a glovebox. CD<sub>3</sub>OD, CDCl<sub>3</sub>, THF-d<sub>8</sub>, CD<sub>2</sub>Cl<sub>2</sub> (Eurisotop), dopamine • HCl (**12**), **G2**, other reagents and solvents (Aldrich), were used as received.

Dimethyl 4-oxo-1,4-dihydropyridine-2,6-dicarboxylate  $(3)^1$  bicyclo[2.2.1]hept-5-en-2ylmethanol  $(4)^2$ , bicyclo[2.2.1]hept-5-ene-2,2-diyldimethanol  $(9)^3$ , bicyclo[2.2.1]hept-5-en-2ylmethyl 4-methylbenzenesulfonate<sup>4</sup>, tetraethylene glycol ditosylate<sup>5</sup>, G3<sup>6</sup> and Tos-4<sup>7</sup> were synthetized according to literature resources.

NMR spectra were obtained on a Varian Unity INOVA spectrometer operating at an equivalent <sup>1</sup>H frequency of 500 MHz. Notation for the <sup>1</sup>H NMR spectral splitting patterns includes singlet (s), doublet (d), triplet (t), broad (br) and multiplet/overlapping peaks (m). Signals are given as  $\delta$  values in ppm, coupling constants (*J*) are expressed in Hertz.

Solid-state NMR magic angle spinning (MAS) spectra of samples were recorded on Varian NMR System spectrometer operating at an equivalent <sup>1</sup>H frequency of 600 MHz with a Chemagnetics 3.2 mm narrow-bore triple resonance with T3 probe in double resonance mode. The spinning rate of the rotor was 10 kHz in all cases. Adamantane was used as external chemical shift reference (38.55 and 29.50 ppm). The 90° pulse lengths were 3.0  $\mu$ s for carbon and 2.4  $\mu$ s for the proton channel. Cross polarization techniques were used to record solid-state <sup>13</sup>C{<sup>1</sup>H} NMR spectra with SPINAL-64 decoupling at 18 °C with a recycle delay of 10 s which is 5 times larger than T<sub>1H</sub>.

GPC measurements were carried out using Waters 2695 separation unit and Waters 2414 RI detectors (Waters, Milford, USA) at 35 °C column and detector temperature. The sample compartment was used at 25 °C. Column bank contained four columns (4.6\*300 mm): Styragel HR 0.5, Styragel HR 1, Styragel HR 2 and Styragel HR 4 (Waters, Milford, USA). The third order calibration curve was used, and the calibration standards were Polystyrenes in the 500-310000 Da molecular weight range. The eluent was HPLC grade THF (VWR International, Leuven, Belgium). The flow rate was 0.5 ml/min. For the calculations of the molecular weights, Millennium<sup>32</sup> Chromatography Manager was used. The concentration of the samples was 5 mg/mL, and the volumes of injections were in the range of 10-50 μL.

The Matrix-Assisted Laser Desorption/Ionization Time-of-Flight Mass Spectrometric (MALDI-TOF MS) measurements were performed with an AutoFlex Speed mass spectrometer equipped with a time-of-flight (TOF) mass analyzer. In all cases, 19.5 kV (IonSource 1) and 18.3 kV (IonSource 2) acceleration voltages were used with pulsed ion extraction (PIE<sup>TM</sup>). The

positive ions were detected in the linear mode. A Bruker smartbeam<sup>TM</sup>-II solid phase laser (355 nm,  $\geq 100 \mu$ J/pulse) operating at 500 Hz was used to produce laser desorption, and 5000 shots were summed. The MALDI-TOF MS spectra were externally calibrated using polyethylene glycol (PEG) standard (Mn = 1450 g/mol, Sigma-Aldrich, Taufkirchen, Germany).

Samples for MALDI-TOF MS were prepared with 2,5-dihydroxybenzoic acid (DHB) (Sigma-Aldrich, Taufkirchen, Germany) matrix dissolved in tetrahydrofuran (VWR International, Leuven, Belgium) at a concentration of 20 mg/mL. The concentrations of the analyte solutions were 5 mg/mL, and sodium trifluoroacetate (NaTFA) (Sigma-Aldrich, Taufkirchen, Germany) was dissolved in THF at a concentration of 5 mg/mL (used as the cationization agent to promote ionization). The solutions were mixed in a 10:2:1 (v/v) ratio (matrix/analyte/cationization agent). A volume of 0.5  $\mu$ L of the solution was deposited onto a metal sample plate and allowed to air-dry.

The glass transition temperature ( $T_g$ ) was determined by a Setaram DSC92 differential scanning calorimeter from -140 °C to 130 °C. The samples, with an average mass of 30-40 mg were pressed in 120 µL aluminum pans and sealed with pierced lids. On each sample, at least three measurements were performed, with two different scanning rates (two measurements with 20 °C/min, and one measurement with 10 °C/min), during the measurement, the calorimeter was purged with high purity nitrogen (flow rate 20 mL/min). Calcined  $\alpha$ -alumina powder was used as reference material. The calorimeter was calibrated with high purity metals (5 different metals, with at least 3 different scanning rates). The data obtained from the first measurement was not evaluated (the thermal history of the sample is removed). Thermal stability was investigated by simultaneous thermogravimetry-differential scanning calorimetry (TG-DSC) on a Setaram LabsysEvo system. The measurements were performed under pyrolytic conditions in a flowing (80 mL/min) high purity argon (99.999%) atmosphere, in 25-500 °C temperature range, with a scanning rate of 10 °C/min. An average of 10-15 mg sample was placed in 100 µL aluminum pan; the samples were used "as received". The measurements were blank corrected. All measurement results were evaluated with Calisto Processing software.

GC-MS analyses were carried out using a Shimadzu GC-MS-QP2010 instrument equipped with a Rxi-5Sil MS column coupled with a quadrupole mass filter with pre-rods.

HRMS and MS-MS analyses were performed on a Thermo Velos Pro Orbitrap Elite (Thermo Fisher Scientific) system. The ionization method was ESI operated in positive ion mode. The protonated molecular ion peaks were fragmented by CID at a normalized collision energy of 35%. For the CID experiment helium was used as the collision gas. The samples were dissolved in methanol. Data acquisition and analysis were accomplished with Xcalibur software

version 2.0 (Thermo Fisher Scientific). In cases of fluorinated samples **10** and **11**, HRMS analyses were performed on a Thermo Q Exactive GC Orbitrap (Thermo Fisher Scientific) system. The ionization method was EI operated in positive ion mode. Electron energy was set at 70 eV. Data acquisition and analysis were accomplished with Xcalibur software version 2.0 (Thermo Fisher Scientific).

TLC was performed on Merck Kieselgel 60  $F_{254}$  plates or Merck Aluminium oxide 60  $F_{254}$  plates and spots were visualized by UV light or by exposing it with iodine or the aqueous solution of (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub>, Ce(SO<sub>4</sub>)<sub>2</sub> and sulfuric acid.

Flash column chromatography was performed by a CombiFlash Rf 150 apparatus using gradient elution in normal (silica or alumina column; hexane–ethyl acetate, dichloromethane–methanol or dichloromethane–ethanol as eluent) phase mode. Sample loadings were performed in the case of silica flash chromatography by drying the sample onto a silica cartridge.

In some cases, gradient elution preparative HPLC was applied (Armen) on a Gemini 250x50.00 mm; 10  $\mu$ m, C18, 110A column using 10 ml of TFA in 5 l water and acetonitrile as the two solvents.

- 2. Synthesis of norbornene functionalized pyridino-18-crown-6 ether (7).
  - 2.1. Synthesis of dimethyl 4-(bicyclo[2.2.1]hept-5-en-2-ylmethoxy)pyridine-2,6dicarboxylate (5) by *Mitsunobu* reaction



Chelidamic acid dimethyl ester (dimethyl 4-oxo-1,4-dihydropyridine-2,6-dicarboxylate, **3**, 10.3 g, 48.8 mmol), 5-norbornene-2-methanol (bicyclo[2.2.1]hept-5-en-2-ylmethanol, **4**, 9.09 g, 73.2 mmol, 1.5 equivalent of **3**) and triphenylphosphine (25.6 g, 97.6 mmol, 2.0 equivalent of **3**) were dissolved in pure and dry THF (310 mL) and cooled to 0 °C. Then DIAD (diisopropyl-diazene-1,2-dicarboxylate, 14.4 mL, 73.2 mmol, 1.5 equivalent of **3**) was added dropwise in nitrogen atmosphere. The reaction mixture was allowed to warm to room temperature and stirred for additional 12 hours. Then the solvent was removed under reduced pressure, and the

crude product (65.9 g) was dissolved in 40 mL of ethyl acetate at 50 °C. 80 ml hexane was added and the solution was cooled down to -10 °C. In 20 hours the formed solid precipitate (triphenylphosphine oxide) was filtered off, and the filtrate was concentrated by vacuum. The crude product (45 g) was purified by flash chromatography (eluent: hexane and ethyl acetate from 0% to 40% using gradient elution). **5** (9.44 g, 61%) was obtained as a white solid containing a mixture of *endo* (74%) and *exo* (26%) isomers. Pure *endo* isomer can be achieved by multiple chromatography (eluent: hexane and ethyl acetate from 10% to 30%). Mixture of *endo* and *exo* isomers in a ratio of 74/26 of **5**.

### endo isomer of 5

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.75 (s, 2H, Ar), 6.19 (dd, 1H, *J*= 5.7 Hz, 3.0 Hz, CH=), 5.92 (dd, 1H, *J*=5.7 Hz, 3.0 Hz, CH=), 3.99 (s, 6H, OCH<sub>3</sub>), 3.86 (dd, 1H, *J*<sub>1</sub>=9.2 Hz, *J*<sub>2</sub>=6.6 Hz, OCH<sub>a</sub>H<sub>b</sub>), 3.72 (t, 1H, *J*=9.3 Hz, OCH<sub>a</sub>H<sub>b</sub>), 3.01 (bs, 1H, CH-CH=), 2.86 (bs, 1H, CH-CH=), 2.58 (m, 1H, *J*<sub>1</sub>=9.3 Hz, *J*<sub>2</sub>=4.0 Hz, CH-CH<sub>2</sub>O), 1.93 (ddd, 1H, *J*<sub>1</sub>=11.8 Hz, *J*<sub>2</sub>=9.3 Hz, *J*<sub>3</sub>=4.0 Hz, CH<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O), 1.49 (ddd, 1H, *J*<sub>1</sub>=8.3 Hz, *J*<sub>2</sub>=4.0 Hz, *J*<sub>3</sub>=2.0 Hz =CH-CH-CH<sub>a</sub>H<sub>b</sub>-CH-CH=), 1.31 (bd, 1H, *J*=8.3 Hz, =CH-CH-CH<sub>a</sub>H<sub>b</sub>-CH-CH=), 0.64 (ddd, 1H *J*<sub>1</sub>=11.8 Hz, *J*<sub>2</sub>=4.4 Hz, *J*<sub>3</sub>=2.6 Hz, CH<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O). <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl3)  $\delta$ : 167.20, 165.35, 149.85, 138.17, 132.08, 114.63, 72.63, 53.28, 49.59, 43.94, 42.35, 38.11, 29.06. HRMS (ESI) m/z: [M + H] + Calcd for C<sub>17</sub>H<sub>20</sub>O<sub>5</sub>N 318.13360; Found 318.13247. HR-ESI-MS-MS (CID=35%; rel. int. %): 252(2); 212(100); 107(2). *exo* isomer of **5** 

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ: 7.80 (s, 2H, Ar), 6.16-6.13 (m, 1H, CH=), 6.12-6.09 (m, 1H, CH=), 4.19 (dd, *J*= 9.0 Hz, 6.3 Hz, 1H, OC*H*<sub>a</sub>H<sub>b</sub>), 4.11 (q, *J*= 9.0 Hz, 1H, OCH<sub>a</sub>H<sub>b</sub>), 4.00 (s, 6H, OCH<sub>3</sub>), 2.89(bs, 1 H, CH-CH=), 2.85(bs, 1H, CH-CH=), 2.58 (m, 1H, CH-CH<sub>2</sub>O) 1.98-1.91 (m, 1H, C*H*<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O), 1.71 (bs, 1H, =CH-CH-C*H*<sub>a</sub>H<sub>b</sub>-CH-CH=), 1.42-1.22 (m, 2H, =CH-CH-C*H*<sub>a</sub>H<sub>b</sub>-CH-CH=, *CH*<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O).





## 2.2. Synthesis of (4-(bicyclo[2.2.1]hept-5-en-2-ylmethoxy)pyridine-2,6diyl)dimethanol (6)



To an ethanolic suspension of **5** (0.50 g, 1.58 mmol) (5.0 ml) sodium borohydride (0.276 g, 7.30 mmol) was gradually added at 0 °C. The mixture was stirred at 0 °C for 1 hour then at room temperature for additional 2 hours and finally at reflux temperature for 14 hours. The solvent was evaporated by vacuum giving a waxy solid. Acetone (7 ml) was added, and the mixture was refluxed for 1 hour. When the solvent was evaporated, aqueous potassium carbonate solution (2 g in 5 ml water) was added, and the resulting mixture was refluxed for 2 hours. The mixture was concentrated, and 2 ml of brine was added. The mixture was extracted with chloroform (5×10 ml). The combined organic phase was dried over anhydrous sodium sulfate, and the solvent was evaporated under reduced pressure to give a pale yellow solid (0.466 g). This crude product was purified by column-chromatography (eluent: DCM/MeOH = 4/1) to give a white solid (6, 0.336 g, 81%).

### endo isomer of 6

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 6.67 (s, 2H, 2×Py-H), 6.18 (dd, 1H,  $J_1$ =5.7 Hz,  $J_2$ =3.0 Hz, CH=), 5.93 (dd, 1H,  $J_1$ =5.7 Hz,  $J_2$ =3.0 Hz, CH=), 4.67 (s, 4H, 2×C $H_2$ OH), 3.76 (dd, 1H,  $J_1$ =9.0 Hz,  $J_2$ =6.4 Hz, Ar-OC $H_a$ H<sub>b</sub>), 3.60 (t, 1H, J= 9.0 Hz, Ar-OCH<sub>a</sub>H<sub>b</sub>), 3.00 (s, 1H, CH-CH=), 2.86 (s, 1H, CH-CH=), 2.56 (m, 1H, CH-CH<sub>2</sub>O), 1.92 (ddd, 1H,  $J_1$ =11.7 Hz,  $J_2$ =9.3 Hz,  $J_3$ =3.8 Hz, C $H_a$ H<sub>b</sub>-CH-CH<sub>2</sub>O), 1.49 (dd, 1H,  $J_1$ =11.7 Hz,  $J_2$ =9.3 Hz, =CH-CH-CH<sub>a</sub>H<sub>b</sub>-CH-CH=), 1.31 (dd, 1H,  $J_1$ =11.7 Hz,  $J_2$ =9.3 Hz, =CH-CH-CH<sub>a</sub>H<sub>b</sub>-CH-CH=), 1.31 (dd, 1H,  $J_1$ =11.7 Hz,  $J_2$ =9.3 Hz, =CH-CH-CH<sub>a</sub>H<sub>b</sub>-CH-CH=), 1.62 (ddd, 1H,  $J_1$ =11.7 Hz,  $J_2$ =4.3 Hz,  $J_3$ =2.6 Hz, CH<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O). <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl3)  $\delta$ : 166.77, 160.46, 138.01, 132.23, 105.80, 71.81, 64.62, 49.60, 43.99, 42.39, 38.22, 29.12. HRMS (ESI) m/z: [M + H] + Calcd for C<sub>15</sub>H<sub>20</sub>O<sub>3</sub>N 262.14377; Found 262.14307. HR-ESI-MS-MS (CID=35%; rel. int. %): 196(33); 156(100).





# 2.3. Synthesis of 1<sup>4</sup>-(bicyclo[2.2.1]hept-5-en-2-ylmethoxy)-3,6,9,12,15-pentaoxa-1(2,6)-pyridinecyclohexadecaphane (7)



A THF (30 mL) solution of **6** (1.73 g, 6.63 mmol) was added dropwise to a 5 mL THF suspension of sodium hydride (0.48 g, 21.0 mmol, 80% dispersion in mineral oil) in nitrogen atmosphere at 0 °C. Then the mixture was stirred at 0 °C for 10 minutes, later on at room temperature for 30 minutes and finally at reflux temperature for 4 hours. The mixture was cooled to -60 °C, and a THF solution (30 mL) of tetraethylene glycol ditosylate (3.50 g, 6.96 mmol) was added. The mixture was stirred at -60 °C for 20 minutes and at room temperature for 7 days. The solvent was evaporated under reduced pressure, and the residue was dissolved in a mixture of dichloromethane (100 mL) and ice cold water (50 mL). The mixture was stirred for 5 minutes, the phases were separated, and the aqueous phase was extracted with dichloromethane (3×100 mL). The combined organic phase was dried over anhydrous magnesium sulfate, and the solvent was evaporated under reduced pressure. This residue (4.00 g) was purified by flash chromatography on alumina (eluent: dichloromethane and ethanol from 0 to 5% using gradient elution) to give 1.41 g (60%) of 7 as a colorless oil. Purity: 85% (<sup>1</sup>H NMR).

Further purification was made by preparative HPLC using water (5% TFA) and acetonitrile eluent. The TFA salt was liberated to 7 by dissolving in dichloromethane (50 mL), followed by the addition of tetramethyl ammonium hydroxide solution (10 m/m%) in water (10 mL). The phases were separated; the aqueous phase was washed with dichloromethane ( $3 \times 50$  mL); the organic phases were combined. The solvent was evaporated under reduced pressure to give 0.99 g (42%) of 7 as colorless oil.

 $R_{f} = 0.75$  (alumina TLC, eluent: 5% MeOH in dichloromethane, using iodine to visualization). TFA salt of 7

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ: 9.60 (bs, 1 H, HN), 7.15 (s, 2H, Ar), 6.18 (dd, 1H, J= 5.6 Hz, 3.0 Hz, CH=), 5.90 (dd, 1H, J= 5.6 Hz, 2.8 Hz, CH=), 4.73 (s, 4H, 2×ArCH<sub>2</sub>), 3.99 (dd, 1H, J=

9.4 Hz, 6.9 Hz, Ar-OC $H_a$ H<sub>b</sub>), 3.88-3.81 (m, 5H, Ar-OCH<sub>a</sub> $H_b$ , 1×OC $H_2$ C $H_2$ O), 3.72-3.58 (m, 12H, 3×OC $H_2$ C $H_2$ O), 2.98 (bs, 1H, CH-CH=), 2.84 (bs, 1H, CH-CH=), 2.60-2.55 (m, 1H, CH-CH<sub>2</sub>O), 1.90 (ddd, 1H, J= 12.1 Hz, 9.2 Hz, 3.7 Hz, OCH<sub>2</sub>CHC $H_a$ H<sub>b</sub>), 1.47 (dd, 1H, J= 8.3 Hz, 1.5 Hz, CHC $H_a$ H<sub>b</sub>CH), 1.29 (d, 1H, J= 8.3 Hz, CHCH<sub>a</sub> $H_b$ CH), 0.64 (ddd, 1H J= 11.9 Hz, 4.0 Hz, 2.7 Hz, OCH<sub>2</sub>CHCH<sub>a</sub> $H_b$ ). <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>): 171.21, 154.03, 138.41, 131.77, 110.75, 74.66, 70.58, 70.56, 69.95, 69.86, 69.17, 49.51, 43.83, 42.32, 37.82, 28.74.



S14





Non-TFA salt of 7

<sup>1</sup>H NMR (500 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $\delta$ : 6.73 (s, 2H, Ar), 6.19 (dd, 1H, *J*= 5.5 Hz, 2.9 Hz, CH=), 5.95 (dd, 1H, *J*= 5.5 Hz, 2.7 Hz, CH=), 4.58 (s, 4H, 2×ArCH<sub>2</sub>), 3.78 (dd, 1H, *J*= 9.2 Hz, 6.6 Hz, Ar-OCH<sub>a</sub>H<sub>b</sub>), 3.70-3.51 (m, 17H, Ar-OCH<sub>a</sub>H<sub>b</sub>, 4×OCH<sub>2</sub>CH<sub>2</sub>O), 3.01 (bs, 1H, CH-CH=), 2.85 (bs, 1H, CH-CH=), 2.59-2.51 (m, 1H, CH-CH<sub>2</sub>O), 1.92 (ddd, 1H, *J*= 12.2 Hz, 9.0 Hz, 3.5 Hz, OCH<sub>2</sub>CHCH<sub>a</sub>H<sub>b</sub>), 1.47 (dd, 1H, *J*= 8.3 Hz, 1.7 Hz, CHCH<sub>a</sub>H<sub>b</sub>CH), 1.33 (m, 1H, HCH<sub>a</sub>H<sub>b</sub>CH), 0.64 (ddd, 1H, *J*= 11.9 Hz, 4.0 Hz, 2.8 Hz, OCH<sub>2</sub>CHCH<sub>a</sub>H<sub>b</sub>). <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CD<sub>2</sub>Cl<sub>2</sub>): 166.89, 160.11, 138.30, 132.71, 107.97, 74.30, 72.11, 71.30, 71.00, 70.80, 70.07, 49.95, 44.53, 42.94, 38.74, 29.50. HRMS (ESI) m/z: [M + H] + Calcd for C<sub>23</sub>H<sub>34</sub>O<sub>6</sub>N 420.23806; Found 420.23677. HR-ESI-MS-MS (CID=35%; rel. int. %): 367(7); 354(100); 314(60).





**Figure S8.**  ${}^{13}C{}^{1}H$  NMR spectrum of *endo* isomer of **7** in CD<sub>2</sub>Cl<sub>2</sub>.

## 3. Synthesis of fluorinated norbornene derivatives 10 and 11 3.1. 5-(((1,1,1,3,3,3-hexafluoro-2-(trifluoromethyl)propan-2-yl)oxy)methyl) bicyclo[2.2.1]hept-2-ene (10)

$$\bigcirc OTs + Na O \leftarrow CF_3 \\ CF_3 \\ CF_3 \\ CF_3 \\ DMF \\ DMF \\ 10 \\ CF_3 \\ 10 \\ CF_3 \\ CF_3 \\ DMF \\ CF_3 \\ DMF \\ CF_3 \\$$

А Schlenk-tube charged with bicyclo[2.2.1]hept-5-en-2-ylmethyl 4was (5.7)methylbenzenesulfonate<sup>4</sup> 20.5 mmol), sodium 1,1,1,3,3,3-hexafluoro-2g, (trifluoromethyl)propan-2-olate<sup>7</sup> (10.6 g, 41.0 mmol) in N,N-dimethyl formamide (32 mL) and fitted to a condenser. The solution was heated overnight at 120 °C then cooled down to room temperature. The reaction mixture was diluted with water (70 mL) and extracted with diethyl ether (3  $\times$  20 mL). The combined organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent was evaporated under reduced pressure (400 mbar). The crude product (4.6 g, pale yellow oil) was filtered trough a silica plug (washed with pentane) than concentrated under vacuum (400 mbar) to give 10 as a mixture of endo (81%) and exo (19%) isomers (4.20 g, 60%) as a colorless oil. Mixture of endo and exo isomers in a ratio of 74/26 of 5.

### endo isomer of 10

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 6.18 (dd, 1H,  $J_1$ =5.7 Hz,  $J_2$ =3.1 Hz, CH=), 5.93 (dd, 1H,  $J_1$ =5.7 Hz,  $J_2$ =3.1 Hz, CH), 3.78 (t, 1H, J= 7.5 Hz, OCH<sub>a</sub>H<sub>b</sub>), 3.56 (t, 1H, J= 7.5 Hz, OCH<sub>a</sub>H<sub>b</sub>), 2.97 (bs, 1H, CH-CH=), 2.83 (bs, 1H, CH-CH=), 2.43 (m, 1H, CH-CH<sub>2</sub>O), 1.84 (ddd, 1H,  $J_1$ =11.8 Hz,  $J_2$ =9.3 Hz,  $J_3$ =3.7 Hz, CH<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O), 1.48 (m, 1H, =CH-CH-CH<sub>a</sub>H<sub>b</sub>), 1.28 (m, 1H, =CH-CH-CH<sub>a</sub>H<sub>b</sub>), 0.51 (ddd, 1H,  $J_1$ =11.8 Hz,  $J_2$ =4.6 Hz,  $J_3$ =2.6 Hz, CH<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O). <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 138.08, 132.15, 73.01, 49.47, 43.68, 42.26, 39.01, 28.59. <sup>19</sup>F NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : -70.83 (s, 9F. HRMS (HR-EI-MS) m/z: [M] + Calcd for C<sub>12</sub>H<sub>11</sub>OF<sub>9</sub> 342,06662; Found 342.06618.

### exo isomer of 10

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 6.12 (dd, 0.19H,  $J_1$ =5.6 Hz,  $J_2$ =3.0 Hz, CH= *exo*), 6.09 (dd, 0.19H,  $J_1$ =5.6 Hz,  $J_2$ =3.0 Hz, CH= *exo*), 4.06 (t, 0.19H, J=7.5 Hz, OCH<sub>a</sub>H<sub>b</sub> *exo*), 3.89 (t, 0.19H, J=7.5 Hz, OCH<sub>a</sub>H<sub>b</sub> *exo*), 2.85 (bs, 0.19H, CH-CH= *exo*), 2.78 (bs, 0.19H, CH-CH= *exo*), 1.76 (m, 0.19H, CH-CH<sub>2</sub>O *exo*), 1.38 (m, 0.19H, =CH-CH-CH<sub>a</sub>H<sub>b</sub> *exo*), 1,28 (m, 2H, =CH-CH-CH<sub>a</sub>H<sub>b</sub>, CH<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O), 1.14 (m, 0.2H, CH<sub>a</sub>H<sub>b</sub>-CH-CH<sub>2</sub>O *exo*). <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz,

CDCl<sub>3</sub>)  $\delta$ : 137.17, 136.37, 121.82, 119.48, 73.73, 45.06, 43.33, 41.69, 39.23, 29.22. <sup>19</sup>F NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : -70.72 (s, 9F). HRMS (HR-EI-MS) m/z: [M] + Calcd for C<sub>12</sub>H<sub>11</sub>OF<sub>9</sub> 342,06662; Found 342.06618.







**Figure S9.** <sup>1</sup>H NMR spectrum of **10** in CDCl<sub>3</sub>. (\* peaks of the ~19% *exo* isomer)



**Figure S10.** <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of **10** in CDCl<sub>3</sub>. (\* peaks of the ~19% *exo* isomer)



## 5,5-bis(((1,1,1,3,3,3-hexafluoro-2-(trifluoromethyl)propan-2yl)oxy)methyl)bicyclo[2.2.1]hept-2-ene (11)



A *Schlenk*-tube was charged with bicyclo[2.2.1]hept-5-ene-2,2-diyl)dimethanol<sup>3</sup> (3.0 g, 19.5 mmol), THF (80 mL), triphenylphosphine (15.3 g, 58.4 mmol) and cooled down to 0 °C. Then diisopropyl-azodicarboxylate (11.5 mL, 58.4 mmol) in THF (20 mL) was added dropwise under nitrogen. The reaction mixture was allowed to warm to room temperature and stirred for 20 minutes. Then nonafluoro-*tert*-butanol (8.14 mL, 58.4 mmol) was added and the solution was stirred for additional 30 hours. The solvent was evaporated under reduced pressure to give a pale yellow oil (10 g). The crude product was purified by distillation (5 mbar, 65 °C) to give **11** (3.11 g, 27%) as a colourless oil.

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 6.25 (bs, 1H, CH=), 6.06 (bs, 1H, CH=), 4.23 (d, 1H, *J*= 8.3 Hz, OCH<sub>a</sub>H<sub>b</sub>), 4.05 (d, 1H, *J*= 8.3 Hz, OCH<sub>a</sub>H<sub>b</sub>), 3.91 (d, 1H, *J*= 7.9 Hz, OCH<sub>a</sub>H<sub>b</sub>), 3.78 (d, 1H, *J*= 7.9 Hz, OCH<sub>a</sub>H<sub>b</sub>), 2.90 (1H, bs, CH-CH=), 2.74 (1H, bs, CH-CH=), 1.60-1.46 (m, 3H, CH-CH<sub>2</sub>-CH, CH<sub>a</sub>H<sub>b</sub>-C), 0.86 (d, 1H, CH<sub>a</sub>H<sub>b</sub>-C). <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 138.56, 133.78, 121.65, 119.32, 71.83, 70.72, 48.43, 47.43, 45.71, 42.45, 32.42. <sup>19</sup>F NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : -69.12, -69.23. HRMS (HR-EI-MS) m/z: [M] + Calcd for C<sub>17</sub>H<sub>12</sub>O<sub>2</sub>F<sub>18</sub> 590.05499; Found 590.05721







Figure S12. <sup>1</sup>H NMR spectrum of 11 (CDCl<sub>3</sub>).



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2 8	6 6 7 4
5 P	***

<del>R</del>	8
7	-33
r	

2



**Figure S14.** <sup>19</sup>F NMR spectrum of **11** (CDCl<sub>3</sub>).

### 4. Homopolymer synthesis

4.1. Polymerization of crown ether 7



In a glovebox a vial was charged with crown ether **7** (50 mg, 119 mmol, 10 mg/mL) and DCM (4.7 mL). Then DCM solution (0.3 mL) of **G2** (2.0 mg, 2.3 mmol, 2 mol% of 7) or **G3** (2.1 mg, 2.3 mmol, 2 mol% of 7) was added. The reaction mixture was stirred for 2 hours at 30 °C. The solution was transparent. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the mixture was stirred for additional 30 minutes. The solvent was evaporated under reduced pressure to give 50 mg (99%) of **poly-7** (**G2**) or 46.5 mg (93%) of **poly-7** (**G3**) as a vaxy solid. NMR measurements showed characteristic broad signals.



Figure S15. MALDI-TOF MS spectrum of poly-7 (G2) polymer, 7 (m/z: 419), (Linear mode, DHB/NaTFA)







GPC Sample Results									
	Sample Name	Inj	RT (min)	Mn	Mw	MP	Poly- dispersity		
	BP 97	1	20 850	1303	1525	1662	1 170		

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Figure S16. GPC chromatogram of poly-7 (G3) polymer.



Figure S17. <sup>1</sup>H NMR spectrum of poly-7 prepared by G2 (CDCl<sub>3</sub>).



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Figure S19. Superimposed <sup>1</sup>H NMR spectra of 7 monomer (blue) and poly-7 (dark red) (CD<sub>2</sub>Cl<sub>2</sub>).

#### 5. Synthesis of copolymers 7 and 8-11

# 5.1. The general method for copolymerization of crown ether 7 and other norbornenes

In a glovebox a vial was charged with 50 mg of the corresponding norbornene monomers [crown ether 7 (20-40 mg) and norbornene 8, 9, 10 or 11 (10-30 mg, 1 or 5 equivalent of 7)] and DCM (4.7 mL, 10 mg/mL). Then DCM (0.3 mL) solution of catalyst (G2 or G3 ( $2 \mod \% /$  norbornene units)) was added. The mixture was stirred for 2 hours at 30 °C. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the solution was stirred for 30 minutes. After the removal of the solvent a solid glue-like product was obtained.

### 5.2. Copolymerization of crown ether 7 and norbornene-methanol (4)



In a glovebox crown ether 7 (50.0 mg, 0.119 mmol), norbornene-methanol 4 (14.8 mg, 0.119 mmol, 1 equivalent of 7) were measured into a vial, dissolved in DCM (6.0 mL). G2 catalyst (4.1 mg, 0.0048 mmol (2 mol% / norbornene monomers)) in DCM (0.5 mL) was added to the solution, and the mixture was stirred for 2 hours at 30 °C. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the solution was stirred for 30 minutes. After the removal of the solvent 60 mg (92%) brownish solid product was obtained.



Figure S20. <sup>1</sup>H NMR spectrum of cp-7-4 (1/1 equivalent, G2) (1 : 1 mixture of MeOD and CD<sub>2</sub>Cl<sub>2</sub>).



Figure S21. DEPT-Q NMR spectrum of cp-7-4 (1/1 equivalent, G2) ( $CD_2Cl_2$ ).



Figure S22. Stacked <sup>1</sup>H NMR spectra of 7 (red) and cp-7-4 (1/1 equivalent, G2) (blue) (MeOD :  $CD_2Cl_2 = 1 : 1$ ).
#### 5.3. Copolymerization of crown ether 7 and norbornene (8)



In a glovebox a vial was charged with DCM (4.7 mL), crown ether **7** (40.8 mg, 0.0973 mmol) and norbornene (**8**, 9.2 mg, 0.0973 mmol, 1 equivalent of **7**). DCM (0.3 mL) solution of **G2** (3.3 mg, 0.0039 mmol (2 mol% / norbornene monomers)) or **G3** (3.5 mg) in DCM (0.3 mL) was added. The mixture was stirred for 2 hours at 30 °C. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the solution was stirred for 30 minutes. After the removal of the solvent 49 mg (98%) **cp-7-8** (1/1 equivalent, **G2**) or 47.5 mg (95%) **cp-7-8** (1/1 equivalent, **G3**) brownish product was obtained.



**Figure S23.** MALDI-TOF MS spectrum of **cp-7-8** (1/1 equivalent, **G3**) copolymer, **7** (m/z: 419), **8** (m/z: 94) (**7**/**8** = 1/1) (Linear mode, DHB/NaTFA)



	GPC Sample Results									
	Sample Name	Inj	RT (min)	Mn	Mw	MP	Poly- dispersity			
1	BP_98	1	18.462	2676	4491	7873	1.678			

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Figure S24. GPC chromatogram of poly-7-8 (7/8 = 1/1, G3) polymer.



Figure S25. <sup>1</sup>H NMR spectrum of cp-7-8 (1/1 equivalent, G2) (CD<sub>2</sub>Cl<sub>2</sub>).



#### CP-7-8 (1/5 equivalent)

In a glovebox a vial was charged with DCM (2.2 mL), crown ether **7** (13.2 mg, 0.0315 mmol) and norbornene (**8**, 14.8 mg, 0.1573 mmol, 5 equivalent of **7**). DCM (0.3 mL) solution of **G2** (3.2 mg, 0.0038 mmol (2 mol% / norbornene monomers)) or **G3** (3.3 mg) was added. The mixture was stirred for 2 hours at 30 °C. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the solution was stirred for 30 minutes. After the removal of the solvent 27.2 mg (97%) **cp-7-8** (1/5 equivalent, **G2**) or 25.5 mg (91%) **cp-7-8** (1/5 equivalent, **G3**) brownish product was obtained.



**Figure S27.** MALDI-TOF MS spectrum of **cp-7-8** (1/5 equivalent, **G3**) copolymer, **7** (m/z: 419), **8** (m/z: 94) (**7**/**8** = 1/5) (Linear mode, DHB/NaTFA)



## Empower2



	GPC Sample Results									
	Sample Name	Inj	RT (min)	Mn	Mw	MP	Poly- dispersity			
1	BP_99	1	18.084	3834	7185	10545	1.874			

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Figure S28. GPC chromatogram of cp-7-8 (1/5 equivalent, G3) copolymer



	GPC Sample Results									
	Sample Name	Inj	RT (min)	Mn	Mw	MP	Poly- dispersity			
1	BP_96	1	18.500	3311	5966	7655	1.802			

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## Figure S29. GPC chromatogram of cp-7-8 (1/5 equivalent, G2) copolymer



Figure S30. <sup>1</sup>H NMR spectrum of cp-7-8 (1/5 equivalent, G2) (CD<sub>2</sub>Cl<sub>2</sub>).



Figure S31.  ${}^{13}C{}^{1}H$  NMR spectrum of cp-7-8 (1/5 equivalent, G2) (CD<sub>2</sub>Cl<sub>2</sub>).



Figure S32. <sup>1</sup>H NMR spectrum of cp-7-8 (1/1 equivalent, G3) (CDCl<sub>3</sub>).



Figure S33. <sup>1</sup>H NMR spectrum of cp-7-8 (1/5 equivalent, G3) (CDCl<sub>3</sub>).



Figure S34. Superimposed <sup>1</sup>H NMR spectra of poly-7 (blue), cp-7-8 (1/1 equivalent) (green), cp-7-8 (1/5 equivalent (red), prepared by G3 (CDCl<sub>3</sub>).

#### 5.4. Copolymerization of crown ether 7 and norbornene-dimethanol 9



In a glovebox a vial was charged with DCM (2.3 mL), crown ether **7** (9.0 mg, 0.0021 mmol), **9** dimethanol (16.5 mg, 0.107 mmol, 5 equivalent of **7**). DCM (0.2 mL) solution of **G2** catalyst (2.2 mg, 0.0026 mmol (2 mol% / norbornene monomers)) was added. The mixture was stirred for 2 hours at 30 °C. The solid precipitate was formed. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the solution was stirred for 30 minutes. After the removal of the solvent 23 mg (90%) brownish product was obtained which was non-soluble in THF, DCM, chloroform, toluene, methanol, ethanol or 2-propanol.

5.5. Copolymerization of crown ether 7 and norborneneylmethyl-perfluoro-*tert*-butyl ether (10)



#### **CP-7-10** (1/1 equivalent, G2)

In a glovebox a vial was charged with DCM (9.7 mL), crown ether **7** (55.0 mg, 0.131 mmol) and norbornene **10** (44.9 mg, 0.131 mmol, 1 equivalent of **7**). DCM (0.3 mL) solution of **G2** catalyst (4.6 mg, 0.0052 mmol (2 mol% / norbornene monomers)) was added. The mixture was stirred for 2 hours at 30 °C. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the solution was stirred for 30 minutes. After the removal of the solvent 95.9 mg (96%) brownish product was obtained.



**Figure S35.** MALDI-TOF MS spectrum of **cp-7-10** (1/5 equivalent, **G2**) copolymer, **7** (m/z: 419), **10** (m/z: 342) (**7**/**10** = 1/5) (Linear mode, DHB/NaTFA)





Figure S37.  ${}^{13}C{}^{1}H$  NMR spectrum of cp-7-10 (1/1 equivalent, G2) (CD<sub>2</sub>Cl<sub>2</sub>).

#### CP-7-10 (1/5 equivalent, G2)

In a glovebox a vial was charged with DCM (4.7 mL), crown ether **7** (10.0 mg, 0.0238 mmol) and norbornene ether **10** (40.8 mg, 0.1192 mmol, 5 equivalent of **7**). DCM (0.3 mL) solution of **G2** catalyst (2.43 mg, 0.0029 mmol (2 mol% / norbornene monomers)) was added. The mixture was stirred for 2 hours at 30 °C. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the solution was stirred for 30 minutes. After the removal of the solvent 48.3 mg (95%) brownish product was obtained.



**Figure S38.** MALDI-TOF MS spectrum of **cp-7-10** (1/5 equivalent, **G2**) copolymer **7** (m/z: 419), **10** (m/z: 342) (**7**/**10** = 1/5) (Linear mode, DHB/NaTFA)



	GPC Sample Results										
	Sample Name	Inj	RT (min)	Mn	Mw	MP	Poly- dispersity				
1	BP88I	1	18.071	6905	9845	10660	1.426				

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Figure S39. GPC chromatogram of cp-7-10 (1/5 equivalent, G2) copolymer



Figure S40. <sup>1</sup>H NMR spectrum of cp-7-10 (1/5 equivalent, G2) ( $CD_2Cl_2$ ).



Figure S41.  ${}^{13}C{}^{1}H$  NMR spectrum of cp-7-10 (1/5 equivalent, G2) in CD<sub>2</sub>Cl<sub>2</sub>.



## 5.6. Copolymerization of crown ether 7 and bis(perfluoro-*tert*butyloxymethyl)norbornene (11)



In a glovebox a vial was charged with DCM (4.7 mL), crown ether **7** (6.3 mg, 0.0150 mmol) and norbornene **11** (44.3 mg, 0.0751 mmol, 5 equivalent of **7**). DCM (0.3 mL) solution of **G2** catalyst (1.53 mg, 0.0018 mmol (2 mol% / norbornene monomers)) was added. The mixture was stirred for 2 hours at 30 °C. The vial was removed from the glovebox, ethyl vinyl ether (0.1 mL) was added, and the solution was stirred for 30 minutes. After the removal of the solvent 49 mg (97%) brownish product was obtained.



	GPC Sample Results									
	Sample Name	Inj	RT (min)	Mn	Mw	MP	Poly- dispersity			
1	BP_94A	1	18.324	2592	5589	8744	2.156			

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## Figure S43. GPC chromatogram of cp-7-11 (1/5 equivalent, G2) copolymer



Figure S44. <sup>1</sup>H NMR spectrum of cp-7-11 (1/5 equivalent, G2) (CD<sub>2</sub>Cl<sub>2</sub>).



Figure S45. Solid state <sup>13</sup>C NMR spectrum of cp-7-11 (1/5 equivalent, G2) (\* spinning side band)



Figure S46. <sup>19</sup>F NMR spectrum of cp-7-11 (1/5 equivalent, G2) in CD<sub>2</sub>Cl<sub>2</sub>.

## 6. Complexation of biogenic amines with norbornene functionalized pyridino-18crown-6 ether (7) monomer and its polymers

Crown ether **7** (2.5 mg, 0.00596 mmol) was measured to a vial, dissolved in the mixture of  $CD_2Cl_2$  (0.3 mL) and  $CD_3OD$  (0.3 mL) 1 equivalent of dopamine hydrochloride (**12**, 1.30 mg, 0.00596 mmol) or L-alanine-L-Lysine hydrochloride (**13**, 1.51 mg, 0.00596 mmol) was added to a sample. The mixture was placed into an NMR tube, <sup>1</sup>H NMR measurements were carried out.



[7] = [12] = 0.01 mmol/mL).



[7] = [13] = 0.01 mmol/mL).

# 6.1. <sup>1</sup>H NMR investigation of poly-7 and cp-7-10 with dopamine hydrochloride (12) and 13

Polymer [**poly-7** or **cp-7-10**, 0.00596 mmol)] was measured to a vial, dissolved in the mixture of  $CD_2Cl_2$  (0.3 mL) and  $CD_3OD$  (0.3 mL) 1 equivalent [per crown ether unit (7)] of **12** or **13** (0.00596 mmol) was added to a sample. The mixture was placed into an NMR tube, <sup>1</sup>H NMR measurements were carried out.



Figure S49. Superimposed <sup>1</sup>H NMR spectra of poly-7 (marine), 12 (blue) and the complex of poly-7 and 12 (green) (1/1 mixture of CD<sub>3</sub>OD and CD<sub>2</sub>Cl<sub>2</sub>, [poly-7] = [12] = 0.01 mmol/mL).



Figure S50. Superimposed <sup>1</sup>H NMR spectra of poly-7 (marine), L-alanine-L-lysine hydrochloride (13) (blue) and their complex (green) (1/1 mixture of CD<sub>3</sub>OD and CD<sub>2</sub>Cl<sub>2</sub>, [poly-7] = [13] = 0.01 mmol/mL).



7-10 = [12] = 0.01 mmol/mL).



Figure S52. Superimposed <sup>1</sup>H NMR spectra of **cp-7-10** (marine), L-alanine-L-lysine hydrochloride (13, blue) and their complex (green) (1/1 mixture of CD<sub>3</sub>OD and CD<sub>2</sub>Cl<sub>2</sub>, [**cp-7-10**] = [13] = 0.01 mmol/mL).

# 7. Optimized geometries

## Structure: 7 : norbornene functionalized pyridino crown-ether

CHA	ARGE:0	SPI	ΝM	ULI	IPL	ICITY	: 1	L
0	-0.353	636	2.	803	3941	1.4	111	326
0	0.497	597	4.	193	3889	-1.0	)47(	018
0	0 212	976	_0	nar	1/25	1 9	2721	508
0	0.212	370	-0.	1050	142J	1.0	210	500
0	2.427	/06	-2.	101	811	1.2	212:	509
0	4.198	749	-1.	940	)102	-1.1	465	526
С	-0.231	169	4.	756	5928	0.0	)321	149
н	0 459	148	5	216	577	0 -	7600	924
ч	-0.856	850	5	551	701	-0 3	2981	265
	1 1 2 0	630	5.	331	. / U I	0.1	1202	200
C	-1.132	6/8	3.	161	901	Ο.	304	4/3
Η	-1.793	491	3.	271	.934	-0.0	069	978
Η	-1.777	090	4.	313	891	1.4	461	110
С	-1.180	227	1.	867	895	2.0	)738	377
ц	_1 810	031	2	300	2060	2 9	21.20	253
11	1.010	070		200	010	2.0		2JJ
Н	-1.854	070	⊥.	365	1212	1.3	5506	042
С	-0.350	555	Ο.	835	5215	2.7	1898	370
Η	-1.013	520	Ο.	301	.072	3.4	1946	689
Н	0.445	265	1.	328	3275	3.3	3782	289
C	0 610	912	_1	271	800	2 1	5510	671
	1 410	242 CE7	1		2090	2		
н	1.410	657	- <u> </u>	052	. 384	3.2	10.	3/0
Н	-0.252	339	-1.	674	1363	3.1	.125	577
С	1.079	508	-2.	326	5678	1.5	5816	665
Н	0.429	982	-2.	330	)397	0.6	5839	920
н	0 988	730	-3	312	174	2 (	)736	606
	2 047	010	3. 2	212		0.0	,, <u>)</u> ,	2000
C	2.947		-3.	224	1000	0	)T.)(	202
Н	3.066	/99	-4.	082	2054	1.2	2054	496
Η	2.253	044	-3.	534	664	-0.2	2878	802
С	4.287	708	-2.	874	207	-0.0	)819	904
Н	4.729	673	-3.	787	095	-0.5	5060	0.01
U	1 963	0,01	_2.	10/	1272	0.0	7030	212
п	4.903	221	-2.	4 2 4	$L \leq I \leq I$	0.	-050	100
C	1.68/	097	3.	51/	558	-0.6	96.	128
Η	2.508	252	3.	915	694	-1.3	3146	609
Η	1.952	366	3.	699	9712	0.3	3583	393
С	4.149	678	-0.	581	.392	-0.7	1664	415
н	4 443	287	-0	454	1422	0 3	289	244
11	4 000	207	0.	000	005	1 1	.0.72	
н	4.880.	232	-0.	020	0000	-1.3	5//:	964
С	2.798	415	0.	062	2921	-0.9	966	109
С	1.621	637	2.	024	644	-0.9	9225	541
С	1.676	507	-0.	667	371	-1.3	3254	469
C	0 457	541	0	007	747	-1 4	1452	222
ц	1 722	000	_1	711	1125	_1 /	170/	205
	1.722		1 ·	200	100	1.1		4 < 1
C	0.428	050	⊥.	392	628	-1.2	1694	40T
Η	-0.478	762	1.	981	.335	-1.3	381'	780
Ν	2.785	110	1.	389	9574	-0.7	1635	519
0	-0.628	077	-0.	749	993	-1.7	/123	322
C	-1 909	416	-0	113	815	-1 6	569	510
ц	-2 017	103	0.	560	0101	_2 5	2220	207
п	-2.017	405	0.	102	.704	-2	1550	251
Н	-1.986	83I	0.	483	\$704	-0.	460	128
С	-2.962	739	-1.	199	9816	-1.6	5850	07
С	-2.812	991	-2.	249	039	-0.5	5310	029
C	-4 392	705	-0	638	8638	-1 4	1548	859
ц	-2 906	951	_1	732	631	_2 0	5/50	220
11	2.900	240	±.	01/	.054	2.0		220
C	-4.244	349	-2.	ø⊥4	1/20	-0.5	1038	5/3
Η	-1.991	355	-2.	957	821	-0.6	814	452
С	-4.891	047	-1.	447	361	-0.2	2164	461
Н	-4.392	928	0.	444	1565	-1.2	260-	112
ц	-5 032	734	-0	826	2223	-2 3	222	191
11	4 2020	, J I 2 0 F	· ·	U Z C		4.5	, <u> </u>	7 0 C
Н	-4.399	205	-3.	541	UTT	υ.:	>058	596
Η	-4.556	900	-3.	245	428	-1.4	1671	165
Η	-5.976	312	-1.	420	605	-0.0	)681	199
С	-2.800	300	-1.	461	319	0.7	167	105
н	-1 903	168	_1	210	1685	1 3	3361	511
~	_1 010	221		000	2112	 	, , , , , , , , , , , , , , , , , , ,	> ± ± > 7 1
C	-4.040.	224	-0.	703	) I I J	0.5	,JJ2	∠ / ⊥
Н	-4.366	845	-0.	282	:385	1.7	124(	JU2

Structure: 7-12 : pyridino crown-ether+ dopaminium ion complex (PERPENDICULAR) CHARGE:1 SPIN MULTIPLICITY: 1

				1			
0	2.880019	-1.743333	2.637091	Н	-8.358099	2.648387	0.986952
$\cap$	0 215567	-0 852351	2 279989	ц	-9 80/31/	_0 103884	-1 272109
0	0.215507	0.052551	2.219909	11	9.004514	0.103004	1.2/2109
0	4.016190	-3.121148	0.417547	Н	-9.277421	1.621741	-1.250241
	1.010190	0.121110	0.11,011		J.2.7.7.121	1.021/11	1.200211
0	2.970992	-2.720405	-2.186090	H	-10.091608	0.842345	1.206955
~	0 252105	1 (44124	0 00000	~	7 (74())	1 000404	0 000000
0	0.352195	-1.644134	-2.366053	C	-/.6/4625	-1.036484	0.022680
C	0 808868	-0 745850	3 510320	ц	-7 002203	-1 803750	-0 036359
C	0.090000	0.745050	5.519529	11	1.002205	1.095759	0.050555
H	1.414331	0.230216	3.585436	C	-8.569200	-0.773904	0.986853
		0.200220			0.000200		
H	0.182335	-0.813193	4.357144	H	-8.776789	-1.372479	1.875363
~	1 000000	1 072010	2 644602	~	2 246022	0 044670	0 010007
C	1.888299	-1.8/3818	3.644603	C	3.246923	2.2446/9	0.21096/
н	1 373769	-2 845738	3 540347	C	2 652255	3 219593	-0 591716
11	1.5/5/09	2.045750	5.540547	C	2.052255	J.ZI9J9J	0.391/10
H	2.356626	-1.829958	4,643004	C	4,590035	2.400803	0.581028
~	2.000020	2.020000	0.000000	~		4.005405	1 01 5005
С	3.906309	-2.719134	2.753524	C	3.379749	4.335425	-1.015825
	4 515011	0 516611	2 (50750		1 (0(04)	2 112004	0 000706
н	4.515811	-2.516611	3.650/59	н	1.606943	3.113664	-0.889/26
н	3 459902	-3 724697	2 849026	C	5 320111	3 507232	0 163163
11	5.455502	5.724057	2.049020	$\sim$	5.520111	5.507252	0.105105
C	4.774408	-2.666572	1.523604	H	5.089123	1.658469	1.208580
		0.01-0-0-0	1 (20522	~	4 20000	4 400000	0 640480
Н	5.655129	-3.315058	1.6/85//	С	4./09992	4.483338	-0.6434/3
TT	E 1222E0	1 622024	1 2/0210	TT	2 016211	E 102/12	1 620124
п	3.132339	-1.033024	1.349319	п	2.910211	3.103413	-1.039134
C	4 795896	-3 247707	-0 757611	$\cap$	5 503808	5 534673	-1 002619
0	1.755050	5.217707	0./0/011	$\cup$	5.505000	0.001070	1.002019
H	5.243612	-2.273979	-1.030935	H	5.002554	6.144223	-1.567962
	F 61 4000	0.000000	0 500105	~	6 601060	0 00011	0 5 4 1 0 0 0
H	5.614302	-3.972665	-0.599137	0	6.621869	3.639211	0.541232
C	2 000001	2 720222	1 071007	TT	6 060127	1 161070	0 155710
C	3.908884	-3./38223	-1.8/198/	н	0.90913/	4.4012/8	0.155/18
н	3 381228	-4 656614	-1 557614				
	3.301220	1.000014	1.00/014				
H	4.523257	-3.971693	-2.758751				
~	0 040000	2 1 1 4 0 4 1	2 102 000				
C	2.042296	-3.114041	-3.183689				
ц	2 577300	-3 /10017	_1 000770				
п	2.377309	-3.419017	-4.099770				
H	1,440214	-3.969097	-2.827111				
	1 1 1 0 0 1 1						
С	1.151226	-1.942454	-3.499711				
	0 510400	0 1 0 7 1 0 2	4 262207				
Н	0.510403	-2.19/103	-4.36238/				
н	1 766180	-1 064842	-3 772628				
11	1./00100	1.004042	5.772020				
C	-0.826747	0.095013	2.198041				
		0.00010					
H	-1.566715	-0.074553	2.999437				
TT	0 400177	1 117266	2 224054				
н	-0.4221//	1.11/200	2.334854				
C	-0 491370	-0 538325	-2 618126				
0	0.491970	0.550525	2.010120				
H	0.120029	0.360931	-2.829117				
		0.000002					
H	-1.123322	-0.727331	-3.502830				
~	1 265716	0 0 0 1 4 7 0	1 401010				
C	-1.303/10	=0.2614/2	-1.421213				
C	-1 514343	0 031040	0 855832				
0	1.011010	0.001010	0.000002				
C	-2.700972	0.057973	-1.599204				
~	0 470060	0 00 0 4 5 0	0 485801				
C	-3.4/3869	0.386459	-0.4/5/81				
ц	-3 160129	0 056102	_2 507700				
п	-3.100120	0.030102	-2.307709				
C	-2 867793	0 366890	0 781871				
0	2.001195	0.000000	0./010/1				
H	-3.415194	0.598888	1.693821	1			
3.7	1 000570	1 200051	0 001 334				
N	⊥.४४≾५४0	-1.296051	-0.001334				
ц	0 866822	-1 045237	-0 010016				
п	0.000022	1.04020/	0.010010				
H	2.073708	-2.052723	-0.674432				
	0 1 400 4 5	1 00000	0 0 4 7 7 1 1				
н	∠.⊥4//46	-1.60/334	0.94//11				
NT	-0 772/37	-0 276562	-0 212077				
IN	0.112431	0.2/0302	0.212011				
0	-4.761901	0.689172	-0.700523				
~		1 007001	0 400050	1			
С	-5.586072	1.027894	0.423852	1			
ц	-5 150720	1 001035	0 030660	1			
п	J. IJ0/JZ	T. 204022	0.90000	1			
H	-5.608409	0.179985	1.128434				
-	0.000109	0.11000		1			
C	2.678237	-0.113665	-0.384413				
тт	3 731305	_0 /16100	_0 //12/0	1			
н	J./J43U5	-0.410190	-0.441340				
н	2.351262	0.194175	-1.388902				
L	2.501202	0.101110	1.000002	1			
C	2.481962	1.013450	0.625000				
	1 407570	1 0/000/		1			
н	1.40/5/0	⊥.∠43684	0.695536	1			
ц	2 820182	0 66/197	1 612200	1			
п	2.020402	0.00410/	T.017300	1			
C	-6.968128	1.335256	-0.105908	1			
Ŭ		1.000200		1			
C	-7.641238	0.167133	-0.902477	1			
	7 000000	1 (1)775	1 000000	1			
C	-1.990988	1.013/35	1.028862	1			
ц	-6 202350	2 205600	-0 774600				
п	0.092559	2.20009	0.114090				
C	-9.110970	0.620882	-0.823888				
Ľ.	5.110570	0.00002	1 000				
H	-7.212666	0.006445	-1.897677				
$\sim$	_0 1/0007	0 606060	0 716050				
	-9.14088/	0.000869	0.110038				
н	-7.565349	1,445266	2,029519				
· · ·				1			
CHARGE:1 SPIN	MOLTIPLIC	Т.Т.Х: Т					
---------------	---------------	-----------	-------------	---------------	-----------		
0 3.917992	-2.452020	-1.068781	н -7.129537	-1.347775	1.010030		
0 1.099674	-1.967779	-0.830081	н -8.740478	1.742755	-0.610565		
0 5 707690	-0 260967	-0 935071	н -8 120641	1 085423	0 951416		
0 1 025010	2 254102	0.026010	11 0.120011	0.016741	0.660524		
0 4.023040	2.234102	0.020919	п -0.957570	-0.910/41	-0.000034		
0 2.084414	2.//0964	-0.244/21	C -6.654965	0.954952	-2.063390		
C 1.673803	-3.229388	-1.142093	н -6.030351	1.351624	-2.865258		
Н 1.856464	-3.724778	-0.176668	C -7.526330	-0.061478	-2.143165		
н 0,960700	-3.838164	-1.722776	H -7.759082	-0.663698	-3.022961		
C = 2.978029	-3 117046	-1 894562	C 1 006604	-0 475466	2 379247		
2.970029	0 5 6 7 2 0 1	2.045465	C 1.000004	1 0 0 0 0 0	2.379247		
H 2.850310	-2.56/391	-2.845465	C 1.018034	-1.869985	2.298251		
Н 3.332322	-4.135545	-2.138140	C -0.229989	0.184961	2.393247		
C 5.218490	-2.475165	-1.628981	C -0.177169	-2.589662	2.197929		
н 5.610426	-3.507702	-1.649615	н 1,962864	-2.417156	2.297857		
н 5 194998	-2 091485	-2 665326	C -1 422428	-0 523734	2 312457		
G ( 124050	1 (107(0	2.0000020	U 0 200C0E	1 272422	2.312437		
0.124958	-1.610760	-0./949/3	H -0.280685	1.2/3433	2.469/26		
Н /.164400	-1./20/93	-1.149492	C -1.395480	-1.922048	2.192913		
н 6.083513	-1.916556	0.266496	н -0.170189	-3.679171	2.116427		
C 6.561478	0.644789	-0.252518	0 -2.616410	-2.524923	2.077359		
н 6.542892	0.444132	0.834228	н -2,502932	-3.487864	2.029547		
н 7 598573	0 521703	-0 610119	0 -2 610626	0 142033	2 340099		
	2 040007	0.010119	U 2 200010	0 = 1 4 0 0 0	2.310033		
0.106149	2.04988/	-0.542039	н -3.328210	-0.514209	2.364/16		
н 6.064355	2.208753	-1.635216					
H 6.835105	2.762605	-0.116420					
C 4.328143	3.548229	-0.258554					
н 5.003330	4.321411	0.150073					
н 4 256013	3 692739	-1 352023					
	2 701070	0.264007					
C 2.96/12/	3.701870	0.364907					
H 2.609465	4.732029	0.196660					
Н 3.012824	3.520259	1.454718					
C 0.350971	-1.382586	-1.890204					
н 1.016393	-1.062590	-2.709781					
H =0 374407	-2 11/150	-2 281039					
п -0.374407	2.114139	-2.201039					
C 0.738507	3.016862	0.1000/1					
н 0.631754	3.085152	1.199762					
н 0.407952	3.982990	-0.322131					
C -0.151337	1.904627	-0.402564					
C = 0.358368	-0 184352	-1 313217					
C 1 526070	2 012014	0 211145					
C -1.326070	2.012014	-0.211145					
C -2.3368/2	0.933635	-0.562558					
н -1.968874	2.896969	0.250134					
C -1.737187	-0.200317	-1.128665					
н -2.312675	-1.077908	-1.417888					
N 3.050120	0.024013	0.142845					
н 2 666065	0 964696	-0 061533					
11 2.00000J	0.110770	0 426000					
п 3.89/032	-0.110//8	-0.430909					
H 2.344116	-0.6/661/	-0.145444					
N 0.432335	0.838737	-0.958490					
0 -3.654509	1.048943	-0.314708					
C -4.453346	-0.137209	-0.405555					
H -3 976958	-0 938721	0 184294					
11 0.570550	0 462447	1 457107					
H -4.510277	-0.403447	-1.4J/10/					
0 3.415458	-0.10/906	1.3/261/					
н 3.703120	-1.153778	1.742960					
Н 4.293241	0.528021	1.736666					
C 2.280853	0.337147	2.482891					
H 2,664437	0.303769	3,515269					
н 2 051002	1 2026/05	2 266243					
	107200	0 1/221/					
0 -5.021310	0.19/380	0.143314					
C -6.563065	1.357722	-0.602146					
C -6.819241	-0.985589	0.020219					
н -5.700930	0.490397	1.197555					
C -8.005461	1.073333	-0.143029					
H -6 150370	2.350930	-0.394494					
C _8 02/527	-0 347642	_0 737330					
	1 000000	0.131330					
н -0.396041	-1.036365	-0.334/3/					

Structure: 7-12 : pyridino crown-ether+dopaminium ion complex (FOLDED)

CHARGE:I SPIN MULTIPLICITY: I	
0 -3.260590 -2.971564 -2.085666	C 7.915530 -2.491163 0.456424
0 -0 770723 -1 495256 -2 177647	H 9 070167 _3 507155 0 090400
0 -0.110125 -1.485258 -2.111841	Н 8.070107 -3.307133 0.089409
0 -4.705883 -3.036762 0.365402	N 1.023568 2.403497 -1.240025
0 -4 056257 -1 176180 2 425803	C -0 388592 2 564238 -1 681562
0 -1.323451 -0.439898 2.452535	C = 0.5951/3  4.0/4428  = 1.813131
C -1.359525 -2.179583 -3.270811	O 0.074999 4.656686 -2.670385
U _2 0//555 _1 /66557 _3 753161	и 1 652405 2 050613 <u>-</u> 1 945450
H -2.044555 -1.400557 -5.755101	H 1.052405 2.950015 -1.045450
н -0.585950 -2.472873 -4.000027	н -0.999801 2.117425 -0.895995
C -2 143808 -3 402499 -2 848644	C -0 598084 1 872780 -3 016698
C 2.143000 5.402499 2.040044	C 0.590004 1.072700 5.010090
н -1.524021 -4.099949 -2.256426	н 0.077935 2.295467 -3.772828
н -2 488522 -3 934290 -3 753334	н -1 633703 2 021889 -3 353069
C -4.129502 -4.031658 -1./15430	H -0.414991 0.793594 -2.910998
н -4.553162 -4.503898 -2.618491	н 1.071538 2.760182 -0.263226
H -3.5/4/61 -4./9/828 -1.144402	H 1.322096 1.419314 -1.2/4402
C -5.243369 -3.464868 -0.873293	N -1.492095 4.725397 -1.054896
$\mu = 6 0.00200 = 1 211001 = 0 710107$	C -2 150601 / 360/38 0 202727
11 0.009299 4.244091 0.710197	C 2.150051 4.505450 0.202727
H -5.718915 -2.617840 -1.401627	C -1.257403 3.628699 1.230911
C -5 677878 -2 422253 1 192924	0 -0 028964 3 484460 0 983288
п -0.020221 -1.4/4/22 0./4044/	U-3.352298 3.765677 U.U23475
н -6.551865 -3.086294 1.317215	C -3.636111 2.419441 -0.690104
C = 5 0.69896 = 2 1.64104 - 2 547212	C _3 158070 1 215350 0 120010
C = J. UU JU	C = 3.1300/3 1.213338 0.120910
H -4.635691 -3.099315 2.943246	C -3.186176 -0.043167 -0.727984
н -5,852047 -1 816611 з 243961	H -2.313944 5 340999 0 693788
C = 3.314964 = 1.013356 3.626778	H -4.1482/3 4.499560 -0.540121
н -3.985812 -0.691339 4.442065	н -3.999679 3.688132 1.024935
H _2 853806 _1 976123 3 911263	и _3 08/280 2 /73007 _1 6/5176
11 2.000000 1.070120 0.011200	11 5.004209 2.475907 1.045170
C -2.243744 0.027003 3.427696	Н -4.686733 2.243679 -0.972684
н -1.730749 0.180695 4.393987	н -3.799448 1.080140 1.006769
H -2.6/4964 0.992932 3.111131	H -2.134230 1.354973 0.507738
C 0.478266 -2.020654 -1.789013	н -2.619101 0.100116 -1.659600
н 0 361197 -3 055886 -1 417885	н -4 216589 -0 322726 -0 992942
11 0.301137 3.0330000 1.117003	
H 1.169408 -2.04/013 -2.648004	H -1.528690 5.715416 -1.292680
C -0.151047 0.348364 2.431266	0 -1.843207 3.279487 2.277829
$\mu = 0.306092 + 1.11100 + 2.277303$	
п -0.590902 1.414494 2.277505	
н 0.378114 0.269873 3.398288	
C 0 775088 -0 113343 1 334169	
C 1.074381 -1.186549 -0.681232	
C 2.133485 0.159411 1.441423	
C 2 992737 -0 260478 0 420739	
0.200470 0.420735	
н 2.538676 0.679714 2.309946	
H 2.538676 0.679714 2.309946 C 2.448894 -0.951609 -0.671067	
H 2.538676 0.679714 2.309946 C 2.448894 -0.951609 -0.671067 H 3.063267 -1.312714 -1.493932	
H 2.538676 0.679714 2.309946 C 2.448894 -0.951609 -0.671067 H 3.063267 -1.312714 -1.493932	
H 2.538676 0.679714 2.309946 C 2.448894 -0.951609 -0.671067 H 3.063267 -1.312714 -1.493932 N -2.574844 -1.183570 -0.017360	
H 2.538676 0.679714 2.309946 C 2.448894 -0.951609 -0.671067 H 3.063267 -1.312714 -1.493932 N -2.574844 -1.183570 -0.017360 H -1.549866 -1.026797 0.140398	
H 2.538676 0.679714 2.309946 C 2.448894 -0.951609 -0.671067 H 3.063267 -1.312714 -1.493932 N -2.574844 -1.183570 -0.017360 H -1.549866 -1.026797 0.140398 H -3 034303 -1 324346 0.897451	
H 2.538676 0.679714 2.309946 C 2.448894 -0.951609 -0.671067 H 3.063267 -1.312714 -1.493932 N -2.574844 -1.183570 -0.017360 H -1.549866 -1.026797 0.140398 H -3.034303 -1.324346 0.897451	
H 2.538676 0.679714 2.309946 C 2.448894 -0.951609 -0.671067 H 3.063267 -1.312714 -1.493932 N -2.574844 -1.183570 -0.017360 H -1.549866 -1.026797 0.140398 H -3.034303 -1.324346 0.897451 H -2.702640 -2.031377 -0.590594	
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**Structure:** 7-13 : pyridino crown-ether+diamid.H<sup>+</sup> ion complex (Lysine coord.) CHARGE:1 SPIN MULTIPLICITY: 1

CHARGE:1 SPIN	MULTIPLICITY: 1	
0 -2.286429	0.793901 2.355799	н -7.158262 -2.791693 0.145484
0 0 249140	-0.206794 1.930402	u = 9 704076 = 3 209974 0 314062
0 0.249140	-0.290704 1.030402	H -0./940/0 -3.2000/4 0.314002
0 -2.841497	3.349081 1.419073	H -8.347686 -2.288072 -0.978267
0 -1.282102	4.467540 -0.680750	C -2.054182 1.267674 -0.923698
0 1 125126	3 160140 -1 247284	C = 2 393050 $= 0$ 194347 $= 0$ 662826
0 1.120120	5.100140 1.247204	
C -0.728592 -	-0.989929 2.590702	H -2.928961 1.902433 -0./32698
н -1.350032 -	-1.612977 1.924436	C -1.564505 1.437141 -2.354069
H -0.240854	-1.644420 3.335520	н -1.296904 2.487071 -2.535268
G 1 E00E2C	0.000040 0.000020	
C =1.398338	0.002240 3.312200	H =0.000307 0.790079 =2.323027
н -0.992646	0.651707 3.969447	н -2.357906 1.147318 -3.056533
н -2.325019 -	-0.550507 3.932624	N -3.700680 -0.490144 -0.686366
C = 3 295144	1 597132 2 945460	C = 4 226865 = 1 831422 = 0 487396
0 3.293144	1.597152 2.945400	
H -4.090774	0.954266 3.36001/	C -4.80991/ -1.942351 0.94/1/6
н -2.863718	2.203914 3.761527	0 -5.966456 -2.442320 1.092459
C = 3.872197	2.497644 1.886050	C -5.214887 -2.217548 -1.581448
4 _1 698228	3 088024 2 320391	C = 5 / 25078 = 3 / 737032 = 1 / 651079
11 4.090220	3.000024 2.320391	
H -4.277442	1.889559 1.054470	C -6.778745 -4.147925 -2.242422
C -3.318859	4.369118 0.561742	C -7.866343 -4.310462 -1.192819
н -3 810033	3 932592 -0 328496	н -3 355459 -2 503043 -0 521048
II A 0E0201	4 007602 1 007700	$\mathbf{H} = \begin{bmatrix} 0.000109 & 2.000010 & 0.021040 \\ \mathbf{H} = \begin{bmatrix} 0.00006 & 1 & 405000 \end{bmatrix}$
п -4.059391	4.99/093 1.00//98	п -0.102120 -1.002900 -1.400236
C -2.152616	5.226096 0.146812	н -4.834214 -1.860220 -2.549968
н -1.609488	5.572324 1.044318	н -5.329312 -4.173723 -0.644233
H -2 519251	6 109610 -0 403763	H -4 611300 -4 168620 -2 253366
	5 101576 0 000070	$\mathbf{H} = \mathbf{C} \mathbf{C} \mathbf{C} \mathbf{C} \mathbf{A} \mathbf{C} = \mathbf{C} \mathbf{C} \mathbf{C} \mathbf{C} \mathbf{C} \mathbf{C} \mathbf{C} \mathbf{C}$
C -U.U95146	3.1013/0 -0.9928/0	п -0.090/49 -3.12048/ -2./48/94
н -0.344587	6.136406 -1.487354	н -7.099794 -3.420841 -3.007318
н 0.462897	5.402596 -0.065296	н -7.598982 -5.102952 -0.481444
C 0 750443	4 350237 -1 919622	н -8 831466 -4 560951 -1 649579
0.750445	4.550257 1.515022	
H 1.64/33/	4.929113 -2.203460	H -4.363321 U.2/9606 -0./403//
Н 0.192263	4.110810 -2.843570	0 -4.088927 -1.536953 1.878052
C 1.916991	2.330524 -2.073447	C 1.100331 -1.213684 1.175925
1 1 212754	1 052004 2 021045	T = 1 = 67240 = 1 = 007104 = 1 = 014242
H 1.313734	1.952904 -2.921045	H 1.007349 -1.007104 1.914343
H 2.761669	2.902096 -2.494447	H 0.497748 -1.913220 0.568225
C 2.451294	1.168084 -1.274827	0 -1.490108 -1.021413 -0.523084
C 2.064374 -	-0.493453 0.266773	
G 2 7 C1 8 O C	0.750500 1.450000	
C 3.761896	0.758599 -1.459896	
C 4.234598 -	-0.346905 -0.740472	
Н 4.429677	1.283043 -2.143543	
C 3 367735 -	-0 980213 0 152554	
U 2 (7(002	1 020247 0 750002	
H 3.676902 -	-1.830247 0.758092	
N -0.993931	1.675959 0.020021	
н -0.081986	1.180205 -0.164561	
H -0 843811	2 692342 -0 085592	
II 1 000014	1 470520 0 00000000	
н -1.292314	1.4/9529 0.992064	
N 1.602372	0.549406 -0.432359	
0 5.510595 -	-0.704628 -0.955034	
C 6 035842	-1 817362 -0 217956	
U E 407410	2.716662 0.420200	
н 5.43/410	-2.110002 -0.438206	
Н 5.969155 -	-1.604040 0.861866	
C 7.472284 ·	-2.006703 -0.649440	
C 8 396766 -	-0.763774 -0 419638	
0 0 101410	2 124055 0 150000	
0 8.191416	-3.124955 0.152890	
н 7.478082 -	-2.242802 -1.724051	
C 9.771223 -	-1.456532 -0.474679	
H 8 214658	0 058108 -1 120568	
	2 200570 0 702250	
L 9.458499 -	-2.3993/9 0./02358	
Н 7.568632 -	-3.525371 0.966964	
Н 8.474614 -	-3.962145 -0.499782	
н 10 602222	-0 769613 -0 263637	
TT 0 040454		
н 9.948454 -	-1.990130 -1.420/44	
Н 10.254348 -	-3.074164 1.036830	
C 8.319183	-0.423004 1.058102	
U 7 7/7360	0 102961 1 102010	
п /./4/302	U.4UZJU4 I.403910	
C 8.952372 -	-1.398524 1.726162	
Н 9.002939 -	-1.532030 2.808039	
N -8.055340 -	-3.074601 -0.385881	
	0.00000T	

Structure: 7-13 : pyridino crown-ether+diamid.H<sup>+</sup> ion complex (N-terminal coord.) CHARGE:1 SPIN MULTIPLICITY: 1

## Structure: all-trans H-(7-10)<sub>4</sub>-H

-					
0	12.7198	-8.7607	2.8417	Н	4.6024
0	13.6061	-7.7292 -	-0.1125	C	2.4191
	10.0001	0.0000	6.4.688		0.54.45
0	11.5256	-9.321/	6.16//	H	2.5145
0	8 9384	-7 6047	5 2653	н	3 2479
	7 0500	6 5 7 4 0	0 2002		1 0 6 1 5
0	1.2505	-0.3/48	2.1221	C	1.0013
С	14.1496	-8.4874	0.9918	H	0.7831
	10 0575	0 5507	0 0100		1 0465
н	13.85/5	-9.002/	0.9120	н	1.0405
Н	15.2416	-8.3960	0.8004	C	-0.5671
0	10 7664	7 0044	0 0000		1 0075
C	13./664	-/.8844	2.3360	н	-1.09/5
Н	13.3840	-6.8518	2.2507	H	0.2235
	14 6007	7 0005	2 05 47		1 5 2 0 0
Н	14.608/	-/.8995	3.054/	C	-1.5300
C	12 0957	-8 2921	4 0646	н	-2 3174
C	12.0007	0.2021	4.0040	11	2.31/4
H	12.8188	-7.7391	4.6923	H	-2.0046
11	11 2727	7 61 20	2 7000	C	1 2026
11	11.2/3/	1.0120	5.7000	C	4.2930
C	11.6186	-9.5744	4.7507	H	4.4397
	10 0707	10 2000	1 6060		4 2240
н	12.3/8/	-10.3808	4.6863	н	4.2240
Н	10.6657	-9.9547	4.3445	C	-0.0766
0	10 0107	0 0004	6 6774		0 2577
Ç	10.212/	-9.0024	0.0//4	н	0.3377
н	9 4655	-9 7128	6 2753	н	-0 7570
		5.7120			
Н	10.3480	-9.1940	/./648	C	0.9/98
C	9 8338	-7 5522	6 4114	C	3 0889
		6.00022	6 4 6 4 9	~	0.0000
Н	10./039	-0.9183	6.1618	C	0.8185
н	9 2967	-7 1034	7 2699	C	1 8558
	5.2507			Ŭ	1.0000
C	8.3245	-6.3265	4.9424	H	-0.0895
TT	0 0707	5 7765	E 0704	C	2 0002
11	0.0151	5.7705	3.0724	C	5.0005
H	9.0541	-5.7349	4.3603	H	3.8197
C	7 0700	-6 6764	1 15/2	N	2 1022
C	1.0700	0.0704	4.1342	14	2.1022
H	6.2650	-5.9253	4.3167	0	1.6193
ц	6 6903	-7 6859	1 1011	C	2 4663
п.	0.0903			Č	2.3000
С	12.2339	-7.9954 -	-0.4581	Н	2.4972
11	12 2000	-7 7840	-1 5505	11	3 1700
	12.2009	/./040	1.0020	п	5.4/00
H	11.9869	-9.0645 -	-0.3050	С	1.7564
<u> </u>	8 1506	-7 5284	2 1 2 5 7	C	2 1367
C C	0.1300	1.0204	2.123/	U U	2.430/
H	8.4490	-8.3193	2.8431	С	0.2900
	7 5170	-8 0033	1 3196	C	1 2952
11	1.5470	0.0000	1.0100	C	1.2002
С	9.3457	-6.8105	1.5466	H	3.0783
C	11 3132	-7 0537	0 28/3	C	-0 0502
Č,	11.0102	1.0331	0.2045	0	0.0302
C	9.5320	-5.4331	1.6522	Н	0.1612
C	10.6670	-4.8701	1.0302	H	-0.3764
TT	0 0 0 2 1	1 7001	2 1 9 7 0	TT	1 2072
п	0.0234	-4./554	2.10/0	п	1.32/3
C	11.5702	-5.6755	0.3263	H	1.3860
ц	12 4444	-5 26/1	-0 1781	U	-0 /130
**	12.1111	5.2041	0.1/01	11	0.4150
N	10.2290	-/.6258	0.8811	C	3.2642
0	10.7685	-3.5244	1.2093	C	-1.1006
, G	11 0140	0 7757	0 4050	0	0 0050
C	11.0140	-2.1131	0.4050	0	0.9039
H	11.5602	-2.8139 -	-0.5876	C	8.7058
н	12 7941	-3 2535	0 6651	н	9 6408
	11 7100	1 2000	1 1100		0.0570
C	11./180	-1.3869	1.1120	н	8.3572
С	12.8233	-0.4159	0.5996	C	7.6414
0	10 2020	0 (747	0 0252		7 0 / 1 7
C	10.3020	-0.0/4/	0.0352	C	/.001/
С	12.2024	0.9991	0.7176	C	6.2095
н	13 7104	-0 4937	1 2727	н	7 6964
**	10.7104	0.4007	1.2/2/	11	1.0004
С	10.6894	0.8326	1.0007	C	6.4541
н	10 0117	-0 8997 -	-0 1812	н	8 2899
	10.0117	1 0055	1 5064		C.2000
н	9.5896	-1.0055	1.5264	C	5.4115
H	12.3865	1.5947 -	-0.1940	H	5.7612
11	12 6056	1 5640	1 5 2 2 0	TT	6 1062
11	12.0030	1.3042	1.0000	11	0.1905
H	10.4972	1.1217	2.0639	H	6.3092
C	13 2274	-0 6030 -	-0 8229	U	6 3292
C	13.22/4	0.0555	0.0225	11	0.5252
С	9.8286	1.6520	0.0805	H	4.7574
0	10 00/2	0 5546	2 0056	C	0 7642
0	10.0045	0.3340	5.0050	C	0.7042
С	18.0674	-0.5501 -	-4.5339	C	4.5800
ц	18 7501	-1 2065	-1 0527	U	2 7065
**	10.7551	1.2000	4.5527	11	2.7005
Н	17.4517	-0.1078 -	-5.3365	H	5.1387
C	17 2525	-1 0884 -	-3 3578	н	8 4756
	1 6 2020	0 5075	2 6460		10 1107
C	10./3/3	-2.32/3 -	-3.0409	н	TO'TTA/
С	15.9955	-0.2502 -	-3.0545	0	-3.2400
	17 0056	-1 1037	-2 4/02	0	-2 6750
		1.100/		0	2.0/09
С	15.4653	-2.6634 -	-2.7832	0	-6.6980
н	16.4514	-2.6030 -	-4.7250	0	-7.3517
	14 0001	1 0000	2 6004	Š	C 05017
C	14.0004	-1.72AT -	-2.0084	0	-0.0324
H	16.2046	0.5135 -	-2.2854	С	-2.1478
	15 6507	0 3150	-3 9/27	U	-1 2002
п.	10.005/	0.0100	5.5741	п	1.2990
Н	15.6973	-3.1243 -	-⊥.8058	Н	-1.7565
н	14.7308	-3.3367 -	-3.2554	С	-3.2320
	14 0000				
н	1 /1 1 1 1 1 1 1 1	1 1100	J . J . J	- н	1 0000
С	14.0092	-1.1188 -	-3.28/1	11	-4.2300
C	14.0092	-1.1188 - -3.5975 -	-3.28/1 -3.3367	Н	-4.2300 -2.9995
	14.0092 17.7751 14 4794	-1.1188 - -3.5975 -	-3.28/1 -3.3367 -1 1825	Н	-4.2300 -2.9995 -4 2878
н	14.0092 17.7751 14.4794	-1.1188 - -3.5975 - -0.9875 -	-3.2871 -3.3367 -1.1825	H C	-4.2300 -2.9995 -4.2878
Н	14.0092 17.7751 14.4794 12.4174	-1.1188 - -3.5975 - -0.9875 - -0.6222 -	-3.2871 -3.3367 -1.1825 -1.5518	H C H	-4.2300 -2.9995 -4.2878 -3.9541
	14.0092 17.7751 14.4794 12.4174 15.2895	-1.1188 - -3.5975 - -0.9875 - -0.6222 - -1.0598 -	-3.28/1 -3.3367 -1.1825 -1.5518 -0.4532	H C H H	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087
<b>.</b>	14.0092 17.7751 14.4794 12.4174 15.2895 18 1206	-1.1188 - -3.5975 - -0.9875 - -0.6222 - -1.0598 - -3.5658	-3.28/1 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958	H C H H	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367
0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206	-1.1188 - -3.5975 - -0.9875 - -0.6222 - -1.0598 - -3.5658 -	-3.28/1 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958	H C H C	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367
	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004	-1.1188 - -3.5975 - -0.9875 - -0.6222 - -1.0598 - -3.5658 - -5.3792	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124	H C H C H	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386
0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084	-1.1188 - -3.5975 - -0.9875 - -0.6222 - -1.0598 - -3.5658 - -5.3792 - 3.3169 -	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955	н С Н С Н	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211
0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5160	-1.1188 - -3.5975 - -0.9875 - -0.6222 - -1.0598 - -3.5658 - -5.3792 - 3.3169 - -7.5000	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7127	н СннСнн С	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641
0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168	-1.1188 -3.5975 -0.9875 -0.6222 -1.0598 -3.5658 -5.3792 -3.3169 -7.5989	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137	н С H H С H H С	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641
0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305	-1.1188 -3.5975 -0.9875 -0.6222 -1.0598 -3.5658 -5.3792 -3.3169 -7.5989 -7.8153	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376	н С н н С н н С н	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717
0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305 -0.8747	-1.1188 -3.5975 -0.9875 -0.6222 -1.0598 -3.5658 -5.3792 -3.3169 -7.5989 -7.8153 -4.8312	-3.28/1 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376 0.9321	н С H Н С H Н С H Н С H H	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717 -7.4919
00000	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305 -0.8747	-1.1188 -3.5975 -0.9875 -1.0598 -3.5658 -3.35658 -5.3792 -3.3169 -7.5989 -7.8153 -4.8312	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376 0.9321	н сн н сн н сн н с н н с н н с н н с н н с н с н с н	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717 -7.4919
0 0 0 0 0 0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305 -0.8747 5.9044	-1.1188 -3.5975 -0.9875 -0.6222 -1.0598 -3.5658 -5.3792 -3.3169 -7.5989 -7.5989 -7.8153 -4.8312 -4.8312	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376 0.9321 -0.2883	н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н С Н С Н С Н С С Н С Н С С Н С Н С С Н С	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717 -7.4919 -8.1066
0 0 0 0 C H	14.4794 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.3305 -0.8747 5.9044 5.5566	-1.1188 - -3.5975 - -0.9875 - -0.6222 - -1.0598 - -3.5658 - -5.3792 - -3.3169 - -7.5989 - -7.5989 - -7.8153 - -4.8312 - -5.4243 -	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376 0.9321 -0.2883 -0.8146	н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н Н С Н С Н С Н С Н С Н С Н С Н С С Н С С Н С С Н С С Н С С Н С С Н С С Н С С Н С С Н С С Н С С Н С С Н С С Н С С С Н С С С С С Н С	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717 -7.4919 -8.1066 -8.6770
0 0 0 С Н Н	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305 -0.8747 5.9044 5.5566 7.0127	-1.1188 - -3.5975 - 0.6222 - -1.0598 - -3.5658 - -5.3792 - -3.3169 - -7.5989 - -7.5989 - -7.8153 - -4.8122 - 4.5157 - -5.4243 - -4.4612 -	-3.28/1 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376 0.9321 -0.2883 -0.8146 -0.3546	н С н н С н н С н н Н Н Н Н Н Н Н Н Н Н	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717 -7.4919 -8.1066 -8.6770 -8.8138
0 0 0 0 0 0 0 0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305 -0.8747 5.9044 5.5566 7.0127	-1.1188 - -3.5975 - -0.9875 - -0.6222 - -1.0598 - -3.5658 - -5.3792 - -3.3169 - -7.5989 - -7.8153 - -4.8312 - -4.5157 - -5.4243 - -4.4612 - -4.4612 -	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376 0.9321 -0.2883 -0.8146 -0.3546	н С н н С н н С н н С н н С	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717 -7.4919 -8.1066 -8.6770 -8.8138 -8.1000
0 0 0 С Н Н С	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305 -0.8747 5.9044 5.5566 7.0127 5.4294	-1.1188 - -3.5975 - 0.6222 - -1.0598 - -3.5658 - -5.3792 - -3.3169 - -7.5989 - -7.5989 - -7.8153 - 4.8312 - 4.5157 - -4.4612 - -4.4629 -	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376 0.9321 -0.2883 -0.8146 -0.3546 1.1575	н С н н С н н С н н С	$\begin{array}{c} -4.2300\\ -2.9995\\ -4.2878\\ -3.9541\\ -4.3087\\ -5.6367\\ -5.7386\\ -5.8211\\ -7.0641\\ -6.1717\\ -7.4919\\ -8.1066\\ -8.6770\\ -8.8138\\ -8.1920 \end{array}$
0 0 0 С н н С н	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305 -0.8747 5.9044 5.5566 7.0127 5.4294 5.1008	-1.1188 - -3.5975 - 0.9875 - 0.6222 - -1.0598 - -3.5658 - -5.3792 - -3.3169 - -7.5989 - -7.5989 - -7.8153 - -4.8312 - -4.45157 - -5.4243 - -4.4612 - -4.4629 - -3.4513	-3.2871 -3.3367 -1.1825 -1.5518 -0.4532 -2.2958 1.2124 -0.9955 2.7137 0.8376 0.9321 -0.2883 -0.8146 -0.3546 1.1575 1.4577	н С н н С н н С н н С н Н С н	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717 -7.4919 -8.1066 -8.6770 -8.8138 -8.1920 -8.9679
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0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	14.0092 17.7751 14.4794 12.4174 15.2895 18.1206 4.3004 5.5084 2.5168 0.0305 -0.8747 5.9044 5.5566 7.0127 5.9044 5.5566 7.0127 5.4294 5.1008 6.2030 3.6143 4.0374	-1.1188 - -3.5975 - 0.8272 - -1.0598 - -3.5658 - -5.3792 - -3.3169 - -7.5989 - -7.5989 - -7.5989 - -7.5989 - -4.8312 - -4.4612 - -4.4622 - -4.4623 - -3.4513 - 4.8164 - 5.4202 - -4.870	-3.387 -3.387 -1.1825 -1.5518 -0.4532 -2.2558 1.2124 -0.9955 2.7137 0.8376 0.9321 -0.2883 -0.8146 -0.3546 1.1577 1.8686 2.4896 2.4896	и н С н н С н н С н н С н н С н н С н н С н н С н н С н н С н н С н н С н н С н н С н н С н	-4.2300 -2.9995 -4.2878 -3.9541 -4.3087 -5.6367 -5.7386 -5.8211 -7.0641 -6.1717 -7.4919 -8.1066 -8.6770 -8.8138 -8.1920 -8.9679 -8.9679 -8.9679 -7.2512 -7.7813
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Structur	re: all- <i>trans</i>	H	-((7-12	complex)-1	l0)4-H c	cooligo	mer, c	CHARGE:4	SPIN	MULTI	PLICITY	X: 1	
0 -8.8270	1.1468 -0.9367	Н	12.1447	-2.4141 -4.1519	C 5.1600	-2.0246	1.6966	C -10.0333	4.4449	5.1547	C 5.1591	9.5232	-2.6910
С -9.7116 н -10 7211	0.0194 -0.5758	С	10.7645	-0.0541 -3.2486	C 7.6195	-2.4569 -0.8139	2 9346	H -10.7285 H -10 6101	5.2928	4.9685	C 4.3742 N 5.7196	9.3521	-0.4863
н -9.3773	-0.4356 0.3712	н	12.4511	0.9416 -4.2367	F 4.9482	-2.8268	2.7414	C -8.9494	4.7995	6.1617	н 6.1271	12.0350	-1.6023
C -9.5972	-0.9093 -1.7851	Η	10.6636	-1.5175 -1.6092	F 4.2351	-1.0647	1.7901	н -8.0792	4.1175	6.1216	н 6.4920	13.2085	-0.2983
C -8.4797	-1.9873 -1.6726	H	9.9957	-2.1178 -3.1241	F 4.7948	-2.7409	0.6285	H -9.3448	4.8144	7.1970	H 5.0450	12.3337	-0.3107
H -9.4150	-0.2796 -2.6901	С	12.6334	-3.1569 -2.1906	F 7.4795	-3.6909	1.1023	H -7.6497	6.5087	7.6790	C 5.0395	9.9596	-1.6295
C -9.1415	-3.2120 -0.9998	Ĉ	9.8863	0.8111 -2.3883	F 7.4917	-2.6067	-0.3232	н -6.5135	6.0118	6.3672	н 4.6801	14.7498	-0.8695
н -8.2006	-2.2730 -2.7196	Н	12.4873	-2.9735 -1.1235	F 8.2933	-0.2459	2.9154	C -7.2433	8.0811	6.2142	н 5.7959	14.5572	-2.2493
C -10.6708 H -11 7851	-3.0766 -1.2133	0 C	2.7260	6.8630 -2.6535	F 7.1575 F 6.2742	0 -1.7116	3.9255	H -6.7965 H -8.1948	8.6777	7.0342 5.8974	C 3.8960	13.5229	-2.5154
н -11.0850	-1.9161 -3.0235	Н	1.0841	6.9016 -1.3496	C 11.9757	-11.0298	-2.4370	C -6.0664	9.3845	4.5305	H 4.2714	12.7825	-3.2593
н -8.8913	-3.2666 0.0745	Η	2.3791	5.7127 -0.9208	C 10.7614	-10.9256	-3.4337	н -6.9361	9.6609	3.9055	C 2.7240	12.9775	-1.7510
н -8.7635	-4.1524 -1.4352	С	1.1244	5.1395 -2.6669	C 13.1161	-11.9305	-3.0219	н -5.9550	10.1268	5.3458	C 2.0995	13.7453	-0.7567
C -7.2580	-1.5241 -0.9347	c	-0.1759	4.6225 -2.0200	F 9.9547	-11.9866	-3.4928	H -4.0126	8.6599	4.2609	C 2.2438 C 1.0129	13.2417	-0.0369
C -11.3902	-3.0995 0.1082	Н	0.8711	5.6633 -3.6257	F 9.9706	-9.8938	-3.1241	н -4.3548	10.2545	3.4952	н 2.4620	14.7530	-0.5359
н -7.3648	-1.4726 0.1518	С	1.0078	2.6665 -2.9303	F 11.1330	-10.6976	-4.6972	C -4.3833	7.4592	2.1260	C 1.1579	11.1851	-1.3347
C -18.9849	-5.5012 1.5402	С	-0.4024	3.9952 -4.1098	F 14.0118 F 12.6859	-12.3874	-2.1397	H -3.3047	6.7171	2.0550	н 2.7079 с 0.5424	11.0836	-2.8297
н -19.9184	-5.2519 1.0016	Н	-0.0869	4.5826 -0.9173	F 13.8469	-11.2823	-3.9352	C -4.9573	6.9985	0.7949	н 0.5343	13.8406	0.7430
н -18.7140	-4.6811 2.2250	Н	-1.0277	5.2925 -2.2252	F 12.4744	-11.5996	-0.1033	н -4.1697	6.6458	0.1006	0 -0.5215	11.3618	0.3188
C -19.02/9	-6.8831 2.1936 -7 1042 3 3755	н н	1.3445	2 0829 -3 8666	F 10.9456 F 10 5918	-12./301	-1.023/	H -5.5641 C -8 7093	3 0104	0.2941	H -0.8836	9 9203	-1 6683
C -20.4423	-7.1560 2.7550	Н	-0.9710	3.3188 -3.5708	0 -18.0000	1.4960	-3.8273	н -7.8159	3.0700	4.3422	н -0.0491	9.6265	-1.0985
н -18.7891	-7.6375 1.4011	С	3.1753	3.6606 -2.1802	0 -18.8424	-0.7518	-2.2369	н -9.3195	2.1388	3.9961	C 15.9641	-1.2419	0.1969
C -18.8934	-7.2105 4.6518	0	6.3080	-0.4276 0.6107	0 - 15.5194	1.3486	-5.4996	C -6.2619	5.1217	-0.0074	C 16.0164	-1.8028	-1.0890
C -20.2634	-7.7368 4.1758	Н	8.3000	0.1412 0.0902	0 -14.0123 0 -14.6726	-2.8164	-3.6245	H -5.3635	4.7273	-0.5316	C 17.0305	-2.7186	-1.4341
н -21.0526	-6.2342 2.7837	Н	7.3205	1.3605 1.0035	C -19.5819	0.4801	-2.4048	C -7.1474	4.0031	0.4769	н 17.0759	-3.1415	-2.4393
н -20.9987	-7.8535 2.1061	С	6.7127	1.0841 -1.1084	H -20.1444	0.2658	-3.3400	C -8.3346	2.9791	2.2240	0 19.8978	-1.1039	4.8155
H -18.9976 H -18 4338	-6.2305 5.1503	C	7.7485 5.4444	1.9809 -1.8530	H -20.2935 C -18 6703	0.6124	-1.5661	N =7.9927	6.6805	3.0251	0 18.2133	-0.3523	2.6012
н -20.2175	-8.8511 4.1055	Н	6.4635	0.1963 -1.7415	н -17.9132	1.7660	-1.7481	н -7.4206	6.9957	3.8450	0 22.1936	-5.0681	2.1426
C -16.9694	-6.0499 3.5047	С	6.8925	2.9894 -2.6454	н -19.2506	2.6328	-2.5866	н -8.5456	5.8430	3.3300	0 19.7206	-4.4199	0.4347
C -21.4009	-7.3563 5.1162	Н	8.3576	2.5440 -1.1016	C -17.0248 u _17 /889	2.5341 2.5314	-4.1263	N = 7.5072	3.9761	2 5/29	C 18.3918	0.4115	3.8169
н -21.2503	-7.7740 6.1200	Н	4.5464	1.3616 -1.2080	H -16.1743	2.4317	-3.4261	H -9.5814	8.0967	3.3766	H 17.5311	1.0937	3.9643
н -21.5082	-6.2714 5.2351	Η	5.2876	2.2345 0.1214	C -16.6288	2.2901	-5.5739	н -8.2799	8.6969	2.3102	C 18.6067	-0.4718	5.0367
0 -10.3420	-5.8696 4.0246	H	6.6610	2.6063 -3.6569	H -16.2762	2 3.2181	-6.0662	C -9.6846	7.3758	1.3024	H 17.8322	-1.2526	5.1537
н -10.3750	-4.0136 2.9751	н Н	5.7523	4.0903 -1.1524	H -17.4466 C -15.0110	0.9532	-6.8072	H -10.1940 H -8.9842	0.2030 7.0747	0.4886	C 20.2482	-2.0672	5.8487
н -11.3360	-4.0905 4.5089	С	8.6423	1.1553 -2.7330	н -15.7546	0.2872	-7.2843	C -10.7166	6.3137	1.5506	н 20.1516	-1.5933	6.8455
C -12.2118	-5.2209 2.8153	С	4.3979	3.4580 -2.6801	H -14.8668	1.8519	-7.4385	C -11.6576	6.4535	2.5822	H 19.5522	-2.9252	5.7845
C -12.8150	-4.1462 1.8626	н	2.9818	3.6434 =1.1049	С =13.6884 н =13.0983	0.2558	-6.5260	C = 10.7533 C = 12.6261	5.1841	2 8091	С 21.68/6 н 22.2191	-2.4546	5.5493
н -11.8396	-6.0875 2.2120	н	8.1992	0.8420 -3.6812	H -13.0751	0.1682	-7.4465	H -11.6402	7.3407	3.2215	H 22.2637	-1.6378	5.0752
C -14.3071	-4.5230 1.7491	Н	10.3327	1.1305 -1.4432	C -13.2186	-1.5936	-5.0122	C -11.7282	4.2086	0.9308	C 22.8752	-4.1024	4.2061
H -12.7393	-3.1413 2.3500	0 C	12.3516	-9.6761 -2.4521	H -12.1536 H -13 3480	) -1.6051 ) -0.9168	-5.3163	H = 10.0371	5.0623 4 3464	-0.0995	H 23.3598 H 23.5123	-3.3654	5 0978
н -13.3907	-6.8042 3.7603	Н	14.2515	-9.8524 -1.4944	C -13.7513	-2.9922	-4.7429	H -13.3511	5.5812	3.6208	C 22.5573	-5.4154	3.5060
н -13.2472	-5.3715 4.7617	Η	12.9519	-9.0012 -0.5662	н -12.9458	-3.6956	-4.4553	0 -13.5631	3.3119	2.1208	H 21.7440	-5.9718	4.0079
H -14.4746	-5.2490 0.9328	С	13.7725	-7.8470 -2.2719	H -14.3144 C -18 1760	-3.4117	-5.5986	H -14.2188	3.4917	2.8684	H 23.4527	-6.0663	3.4363
н -15.0392	-4.3024 3.7881	C	12.7647	-6.7082 -2.0132	H -17.3754	-0.1304	-0.8627	H -12.5299	2.5404	0.2147	H 21.5116	-6.9433	1.5203
C -12.1202	-4.1344 0.5320	Н	13.8215	-8.0423 -3.3726	н -18.9117	-0.7335	-0.1381	C 3.5113	8.2994	-0.8164	н 20.2551	-5.8713	2.2508
C -15.7435	-6.1820 2.9900	С	15.0695	-5.7934 -1.9429	C -15.0568	-4.0752	-3.0139	C 3.5052	7.8621	-2.1506	C 20.8573	-5.3092	0.2150
н -17.2005 н -15.4420	-7.0730 2.4351	С	13.5834	-5.4217 -1.7355	H -15.4440 H -14.1527	-4.7648	-2.5535	C 4.3349	8.4720	-3.1100	H 20.5837 H 21.6994	-4.7105	-0.1808
н -12.2462	-5.0391 -0.0662	Н	12.0866	-6.5856 -2.8764	C -16.1159	-3.7945	-1.9804	н 4.3253	8.1321	-4.1477	C 16.9585	-1.0836	2.5188
н -11.2650	-2.1964 0.7116	Н	12.1065	-6.9289 -1.1544	C -17.6200	-2.2972	-0.9783	0 5.4759	13.4425	1.9842	н 16.9484	-1.8888	3.2827
C -18.1112 C -17 5579	-3.2821 -0.1128	н	15.4264	-5.4821 -2.9414	N -16.9274 H -16 2029	-1.1004 -1.7738	-4.5318	0 3.9264	14 2829	0.8230	H 16.11/2	-0.3913	-0 8025
н -18.8898	-3.0486 0.6129	Н	13.4348	-5.1319 -0.6663	н -16.4588	-0.1777	-4.7043	0 8.4085	13.6439	-2.4290	н 19.8382	-3.5037	-1.4527
C -16.5458	-4.8423 -1.1582	С	13.1613	-4.2983 -2.6410	н -17.6304	-0.9623	-3.7677	0 6.9565	11.0764	-3.0066	н 18.7067	-4.9072	-1.3274
H -16.1231	-5.8454 -1.2405	H C	13.315/ -1 1439	-4.483/ -3./06/ 2 3160 -1 6654	N -16.63/7 C -17 5847	-2.5336 -1 6094	-1.8956	C 3.6981	12 7713	2.0915	C 17.9695	-3.054/	-0.4525
C -1.2401	-1.2497 -0.4759	Н	-0.8075	2.3636 -0.6260	н -18.3230	-0.8503	-6.1577	н 3.0930	11.2439	2.7517	N 20.5761	-2.1764	2.1868
н -0.4373	-0.5020 -0.5618	С	-9.0177	-5.8462 4.4943	H -16.8089	-1.6880	-6.6036	C 4.9963	12.3133	2.7659	н 20.2531	-2.9045	1.5013
H -1.4782	-1.4417 0.5851	C	-8.9271	-7.3454 4.9647	C -18.2422 u -18 5553	2 -2.9699	-5.5812	H 5.7647	11.5176	2.7636	H 20.7593	-2.6507	3.1049
C -2.8555	0.5946 -1.0972	c	-8.8090	-4.8629 5.6910	H -17.4867	-3.7084	-5.2253	C 6.7307	13.9883	2.4806	N 17.9297	-2.5318	0.8107
C -3.7129	-1.7321 -0.9807	F	-7.7498	-4.2748 3.0798	C -19.4372	-2.9372	-4.6718	н 6.5520	14.4393	3.4766	C 21.8251	-1.4678	1.6988
н -2.2070	-1.0554 -2.4026	F	-6.8249	-6.1344 3.4469	C -20.4956	-2.0465	-4.9075	н 7.4749	13.1745	2.5780	H 22.1410	-0.7179	2.4653
H -2.6259	0.8761 -0.0380	.म म	-8.8664	-3.5645 5.3738	C -19.4989 C -21.6056	-3.8284	-3.58/9	H 7.7753	15.0295	1.4507	H 22.6610 C 21.6026	-2.2103	0.3369
C -4.8921	-0.7449 -0.7925	F	-9.7215	-5.0099 6.6581	H -20.4613	-1.3704	-5.7662	н 6.2734	15.4978	0.9445	H 22.5883	-0.4342	-0.0357
н -3.9174	-2.4650 -1.7815	F	-7.8293	-7.6845 5.6430	C -20.6113	-3.8040	-2.7457	C 8.4774	15.1161	-0.5839	н 21.3130	-1.5809	-0.4171
H -3.5700	-2.3331 -0.0652	F	-9.9515	-7.6726 5.7574	H -18.6951	4.5439	-3.4004	H 7.6502	15.5763	-1.1573	C 20.6182	0.3217	0.3605
н -4.8463	1.4597 -0.7294	F	-8.4991	-6.0221 2.1499	H -22.4235	-1.3070	-4.2406	C 9.3159	14.1593	-1.4162	C 19.5187	0.2954	-0.5126
н -5.1158	-0.6675 0.3006	С	0.4562	-3.0606 -0.8778	0 -22.6958	-2.9267	-2.0614	н 9.7290	13.3281	-0.8123	C 19.8765	2.4590	1.2653
C -2.1336	1.5046 -2.0481	С	0.2273	-4.4660 -1.5507	H -23.4065	-2.2439	-2.2830	H 10.1464	14.6784	-1.9339	H 21.6648	1.4389	1.8955
с -6.1140 Н -5.9987	-1.2462 -2.6270	C C	1.5158	-2.2193 -1.6666 -3.2094 0.6331	∪ =20.6223 H =21.4982	-4./08// -4.6951	-1.2079	с 8.8862 Н 9.7632	12.6855	-3.6964	с 18.6014 Н 19.3744	-0.5308	-0.48/0
н -2.4681	1.4499 -3.0867	F	1.2105	-5.3552 -1.3997	C -8.8272	1.9795	1.3765	н 9.1929	11.6957	-2.2990	C 18.7716	2.4246	0.4130
0 15.1600	-1.5215 -2.1080	F	-0.8679	-5.0624 -1.0704	C -8.4353	2.0219	0.0283	C 7.7159	11.9391	-3.9068	H 20.0138	3.2982	1.9528
C 14.0039 H 14 4092	-U.638U -1.8484 0.3846 -1 7346	Р Т	0.0210 2.0556	-4.3925 -2.8685 -1.1925 -0 9996	н -9.4853 С -7 5861	3 0425		н 8.0586 н 7.0510	12 7590	-4.2415	U 17.7884 H 17 9639	3.3908 4 1227	U.3729 1 0464
н 13.5007	-0.9520 -0.9190	F	2.5736	-2.9196 -2.0944	H -7.2870	3.0753	-1.4909	C 4.4402	9.8823	0.9263	0 17.5430	1.2657	-1.3635
C 13.1460	-0.8046 -3.1042	F	0.9982	-1.6674 -2.7686	0 -8.5242	6.1435	5.8015	н 5.4808	9.9109	1.3103	н 16.9922	2.1116	-1.3646
C 12.1981	-2.0416 -3.0958	F	1.0662	-2.0606 1.2768	0 -9.5047 0 -6 3147	4.2207	3.8259	H 3.8126	9.3047	1.6337	С 16.3184 н 16 3737	-7.9350	-2.5482
н 13.8363	-0.8972 -3.9801	с F	-0.1219	-3.8059 1.3580	0 -5.1079	8.6809	2.4395	H 5.3981	10.8280	-4.4012	H 16.2580	-7.7498	-3.6282
C 10.8067	-1.5004 -2.7044	С	6.6038	-1.4103 1.5708	0 -5.8316	5.8851	1.1491	н 6.6075	9.4885	-4.3130	н 17.2799	-7.5250	-2.2116

## 8. Determination of the logK for the formation of complex 7-12 via NMR titration

Solutions of pyridino-18-crown-6 ether (7) and dopamine.HCl (12) having the same concentration ( $c_0 = 0.0143$ M) were mixed in various volume ratios and NMR spectra of the mixtures were measured. Complex stability constants were determined using the method described in ref. <sup>8</sup> following the chemical shift of the methylene protons next to the pyridine-ring of the pyridino-18-crown-6 ether.

Chemical equation of complex formation (see Table 1 for notations of concentrations):

$$7 + 12 \rightleftharpoons 7-12$$

**Table S1.** Concentration balances in the equilibrium of complexation between dopamine.HCl (**12**) and pyridino-18-crown-6 (**7**) solutions ( $c_0 = 0.0050$ M) during NMR titration.

	pyridino-18-crown-6 ether (7)	dopamine.HCl (12)	complex 7-12
initial	$c_{7,0} = \frac{c_0 V_7}{V_7 + V_{12}}$	$c_{12,0} = \frac{c_0 V_{12}}{V_7 + V_{12}}$	0
change	-x	- <i>x</i>	+ <i>x</i>
equilibrium	$c_{7,0} - x$	$c_{12,0} - x$	x

The equilibrium constant is calculated as:

$$K = \frac{c_{7-12}}{c_7 c_{12}} = \frac{x}{(c_{7,0} - x)(c_{12,0} - x)}$$

After rearranging, one arrives at the following quadratic equation:

$$Kx^2 - [K(c_{7,0} + c_{12,0}) + 1]x + Kc_{7,0}c_{12,0} = 0.$$

Dividing the previous equation by K and taking into account that  $c_{7,0} + c_{12,0} = c_0$ :

$$x^{2} - [c_{0} + K^{-1}]x + c_{7,0}c_{12,0} = 0.$$

The appropriate root of the quadratic equation, where  $x < \min(c_{7,0}, c_{12,0})$ , is:

$$x = \frac{[c_0 + K^{-1}] - \sqrt{[c_0 + K^{-1}]^2 - 4c_{7,0}c_{12,0}}}{2}$$

The  $\delta_x$  chemical shift corresponding to the *x* complex concentration can be calculated by linear interpolation from the mole fractions and NMR chemical shifts of the crown ether in the free  $(x_7; \delta_7)$  and complex forms  $(x_{7-12} = 1 - x_7; \delta_{7-12})$ :

$$\delta_x^{\text{calc}} = x_7 \delta_7 + x_{7-12} \delta_{7-12} = \frac{c_7 \delta_7 + c_{7-12} \delta_{7-12}}{c_7 + c_{7-12}}$$

The sum of square deviation of the experimental NMR shift ( $\delta_x^{exp}$ ) and calculated  $\delta_x^{calc}$  NMR shift along the titration curve (i.e. for all  $x_i$  values) was minimized by optimizing the value of K,  $\delta_7$  and  $\delta_{7-12}$ :

$$\min_{\boldsymbol{K},\delta_{7},\delta_{7-12}}\sum_{i} \left(\delta_{x_{i}}^{\text{calc}}(\boldsymbol{K},\delta_{7},\delta_{7-12})-\delta_{x_{i}}^{\exp}\right)^{2}$$

The measured and fitted chemical shift values are shown on Figure S53 and in Table S1.



## Figure S53.

The experimental values and fitted theoretical curve of chemical shifts for methylene protons next to the pyridine-ring within the crown ether as a function of the mixing ratios of the dopamine. HCl (12) and pyridino-18-crown-6 ether (7). The sum of concetrations in the mixture was constant:  $c_{12,0} + c_{7,0} = 0.0143$ M.

**Table S2.** The experimental and fitted chemical shift values for methylene protons next to the pyridine-ring within the crown ether as a function of the mixing ratios of the dopamine. HCl (**12**) and pyridino-18-crown-6 ether (**7**). The sum of concetrations in the mixture was constant:  $c_{12,0} + c_{7,0} = 0.0143$ M.

$\frac{c_{12,0}}{c_{7,0}} = \frac{V_{12,0}}{V_{7,0}}$	$\delta_{x_i}^{\exp}/\operatorname{ppm}$	$\delta_{x_i}^{ ext{calc}}/ ext{ppm}$		
0	4.638	4.636		
0.11	4.629	4.628		
0.24	4.620	4.620		
0.36	4.612	4.612		
0.48	4.604	4.604		
0.59	4.595	4.597		
0.69	4.587	4.591		
0.81	4.580	4.584		
0.93	4.576	4.577		
1.06	4.575	4.573		
1.21	4.574	4.571		
1.55	4.572	4.570		
1.86	4.571	4.570		
2.54	4.571	4.570		

The optimized log *K* value and its 95% confidence interval ( $\pm 2.2\sigma$  for 14-3=11 degrees of freedom from Student's t-distribution) are:

$$\log K = 4.3 \pm 0.6$$

Izatt et al. found log *K* values of 3.62 and 3.29 for similar complexes<sup>9</sup> shown on Figure S54 [R=Me; (R) and (S) enantiomers of PhEt]. Their values are somewhat lower which can be explained by the presence of the extra methyl groups which hinder complexation both sterically and by reducing the conformational flexibility of the crown ether ring.



Figure S54. Complexation of molecules investigated by Izatt et al.9

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