

Supplemental Information

Multi-stimuli responsive tetra-PPO₆₀-PEO₂₀ ethylene diamine block copolymer enables pH, temperature, and solvent regulation of Au nanoparticle composite plasmonic response

¹Sokhna I-Y. F Diouf, ¹Darrick J. Williams, ²Soenke Seifert, ¹ Alejandra Londoño-Calderon, ¹Michael T. Pettes, ¹Christopher J. Sheehan, and ¹Millicent A. Firestone*

¹Los Alamos National Laboratory, Los Alamos, NM USA 87545

²Argonne National Laboratory, Lemont, IL USA 60439

¹Los Alamos National Laboratory, Los Alamos, NM USA 87545

²Argonne National Laboratory, Lemont, IL USA 60439

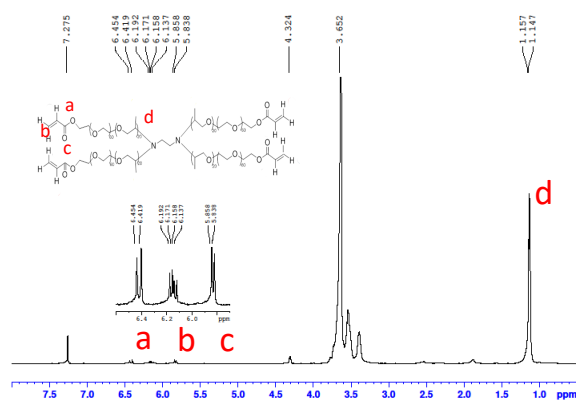
(*) Author to whom correspondence should be addressed:

Millicent A. Firestone
Materials Physics & Applications
Los Alamos National Laboratory
P.O. Box 1663
Los Alamos, NM 87545
Phone: 505-695-8837
E-mail: firestone@lanl.gov

Table of Contents

Table of Contents.....	2
Figure S1: ^1H NMR of tetra-acrylated Tetronic, Tet1107(Acr) ₄	3
Figure S2: ATR/FT-IR spectrum collected on solid Tetronic 1107 (black) and solid tetra-acrylated Tetronic 1107.....	4
Figure S3: Time-dependent Vis-NIR spectra	5
Figure S4: Time-dependent ratio of absorbance post addition of DPPH to complex fluid during Au NP network polymer composite formation	8
Figure S5: TGA and DSC on the photo crosslinked polymer	9
Figure S6: STEM images and EDX composition	8
Figure S7: Temperature dependent SAXS profiles on a Au NP composite prepared with acidic (pH 3.6) buffer	9
Table S1: Small angle X-ray reflections observed for a Au NP composite	10
Figure S8: Normalized temperature dependent Vis-NIR spectra collected on an F98(AcR) ₂ Au NP	11
Figure S9: Temperature dependent SAXS profiles on a Au NP composite prepared with basic pH 3.6 buffer.....	12
Figure S10: Normalized temperature dependent Vis-NIR spectra collected on a Au NP in 3 different pHs.....	13
Figure S11: Temperature dependent SAXS profiles on a Au NP composite prepared with basic pH 9.03 buffer	14
Figure S12: Diffuse reflectance optical spectrum	15

Figure S1. ^1H NMR spectrum of tetra-acrylated Tetronic, Tet1107(Acr) $_4$. Functionalization determined to be 95%.



$$\% \text{ function} = \frac{\text{number of repeats PPO} \times 3 (^1\text{H}(-\text{CH}_3))}{\text{actual number of } ^1\text{H}(-\text{CH}_3)} \times 100$$

Figure S2. Comparison of ATR/FT-IR spectrum collected on solid Tetronic 1107 (black) and solid tetra-acrylated Tetronic 1107, Tet1107(Acr)₄ (blue). (A) High wavenumber region and (B) low wavenumber region.

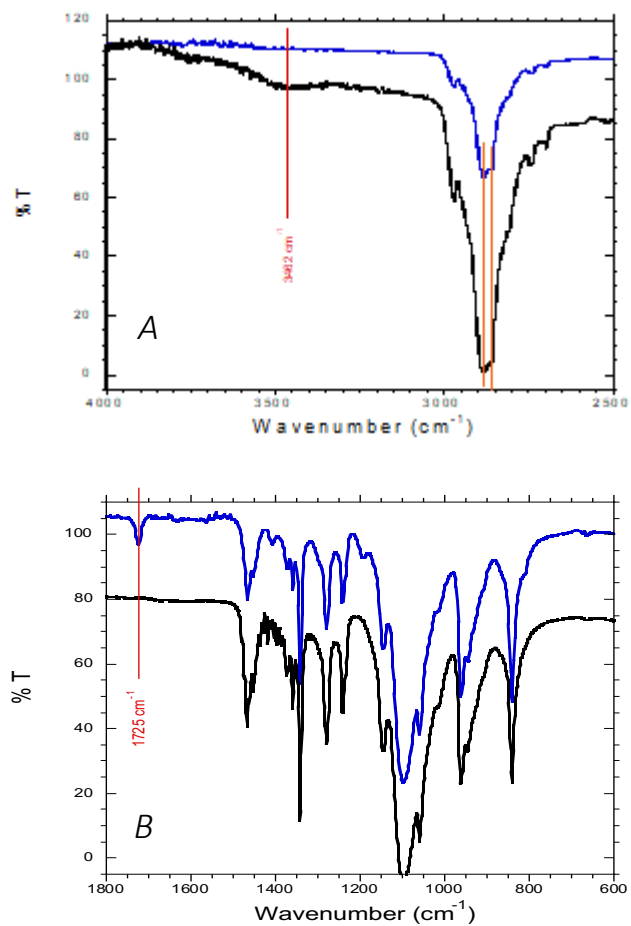


Figure S3.

(A) Time-dependent Vis-NIR spectra collected on a mesophase mixture prepared with 0.733 ± 0.001 weight fraction (Φ_w) water, $\Phi_s = 0.0371 \pm 0.0005$ Triton-X 100 co-surfactant, $\Phi_l = 0.0852 \pm 0.0005$ DMPC lipid, and $\Phi_p = 0.114 \pm 0.005$ Tetronic 1107(Acr)₄, with Au molar concentration of 6.41mM. (B) a mesophase mixture prepared with 0.730 ± 0.001 weight fraction (Φ_w) water, $\Phi_s = 0.0370 \pm 0.0005$ Triton-X 100 co-surfactant, $\Phi_l = 0.0866 \pm 0.0005$ DMPC lipid, and $\Phi_p = 0.147 \pm 0.005$ F98(Acr)₂, with Au molar concentration of 7.09 mM.

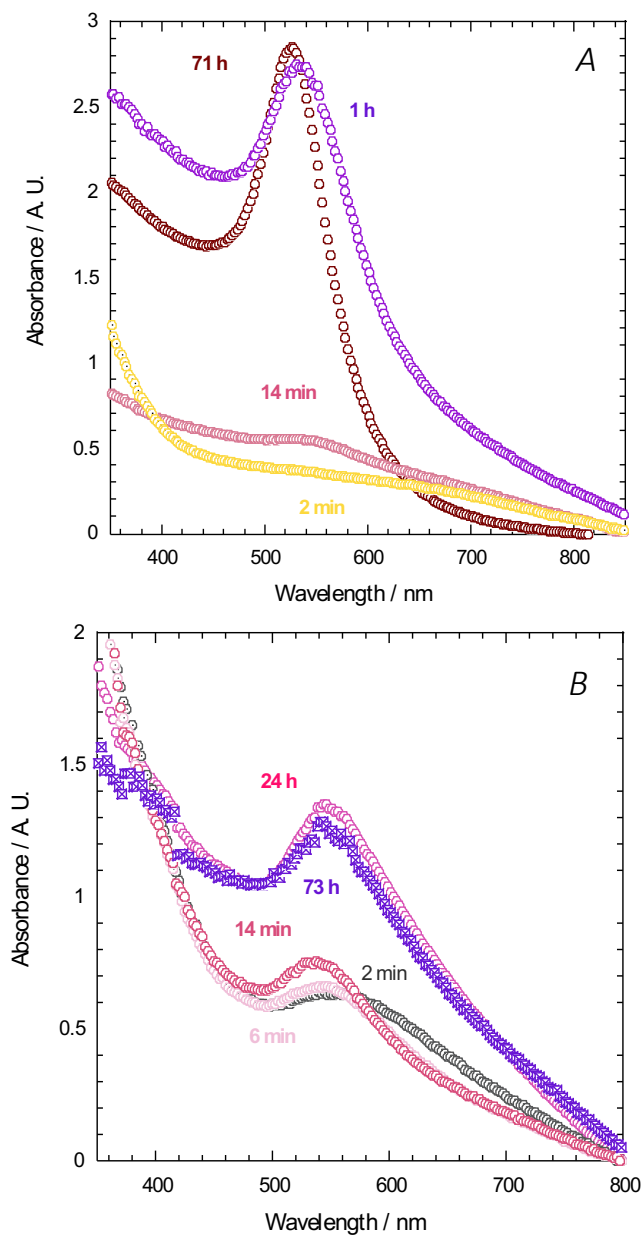


Figure S4. Time-dependent ratio of absorbance post addition of DPPH to complex fluid during Au NP network polymer composite formation. Red curve is time course for a composition prepared with 0.733 ± 0.001 weight fraction (Φ_w) water, $\Phi_s = 0.0371 \pm 0.0005$ Triton-X 100 co-surfactant, $\Phi_l = 0.0852 \pm 0.0005$ DMPC lipid, and $\Phi_p = 0.114 \pm 0.005$ Tetronic 1107(Acr)₄, with Au molar concentration of 6.41mM. Purple curve is composition prepared with 0.730 ± 0.001 weight fraction (Φ_w) water, $\Phi_s = 0.0370 \pm 0.0005$ Triton-X 100 co-surfactant, $\Phi_l = 0.0862 \pm 0.0005$ DMPC lipid, and $\Phi_p = 0.147 \pm 0.005$ F98(Acr)₂, with Au molar concentration of 7.09 mM

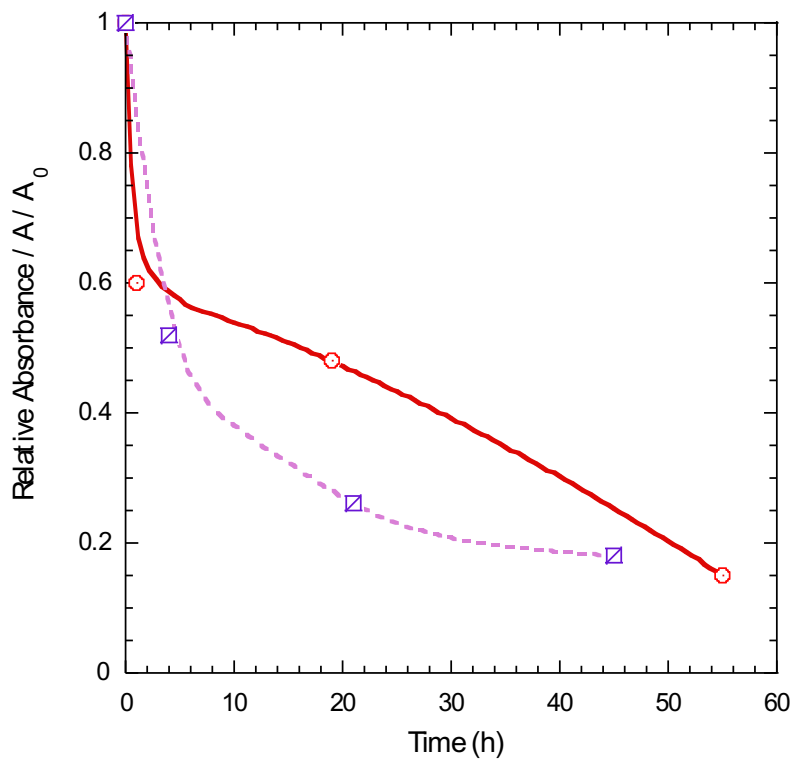


Figure S5. (A) Fast scan ($10\text{ }^{\circ}\text{C min}^{-1}$) N_2 atmosphere TGA collected on a photo-crosslinked complex fluid. Weight % (black) and derivative weight loss (blue). (B) Differential scanning calorimetry (DSC) taken on the photo-crosslinked network polymer.

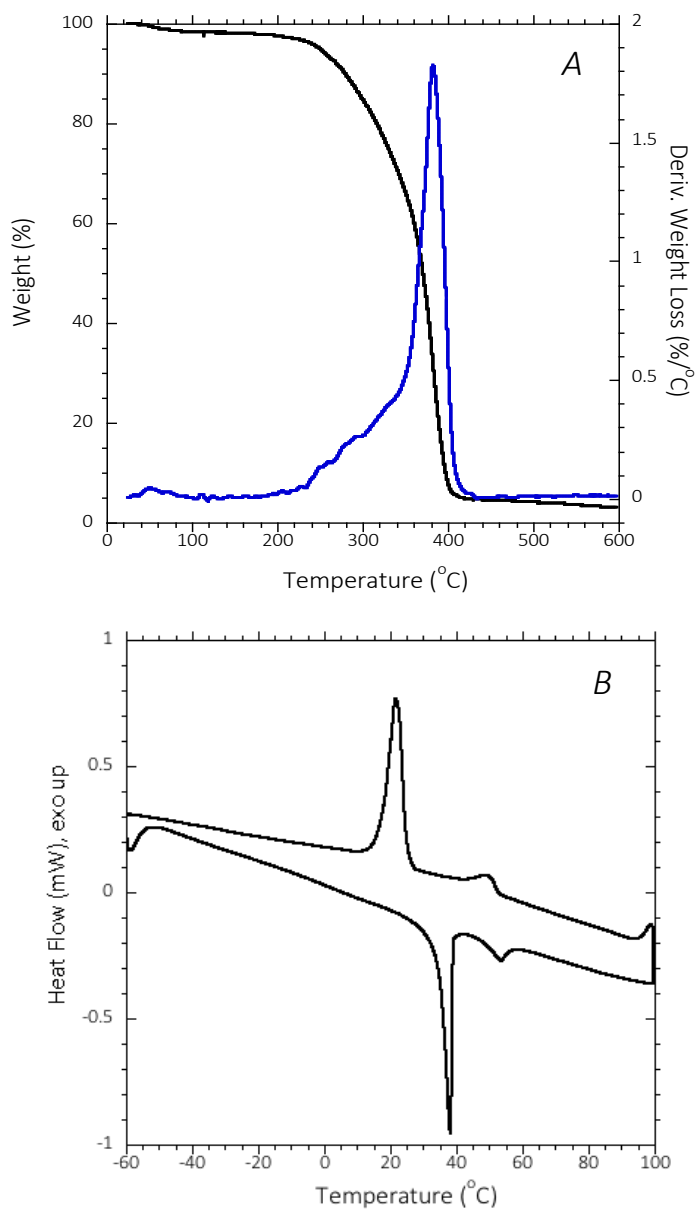


Figure S6.

(A) Representative STEM image collected on grid cast particles released from a network composite using a 17 (v/v) % aqueous hydrazine solution. (B) Histogram of nanoparticle size distribution. (C), (D) Representative STEM images collected on Au NP Tet(Acr)₄-based polymer composites directly formed on the TEM grid. The bright contrast of Au NP in comparison with the dark background of the polymeric matrix is related to the Z-contrast of the HAADF image in which the intensity escalates with the atomic number $I \sim Z$ (1.8-2). The size distribution histogram in Figure E was calculated for N=200 particles and fitted with a Gaussian distribution to obtain the mean \pm 1-sigma diameter. Elemental analysis of the sample is shown in Figure F where the characteristic peaks for Au are present. Additional peaks for C and O are related to the polymer; a small Cl peak can be associated with residual materials from HAuCl₄ precursor, and Cu from the copper support grid. (G) Representative STEM images collected on Au NP F98(Acr)₂-based polymer composites directly formed on the TEM grid. (H) The size distribution histogram N=200 particles.

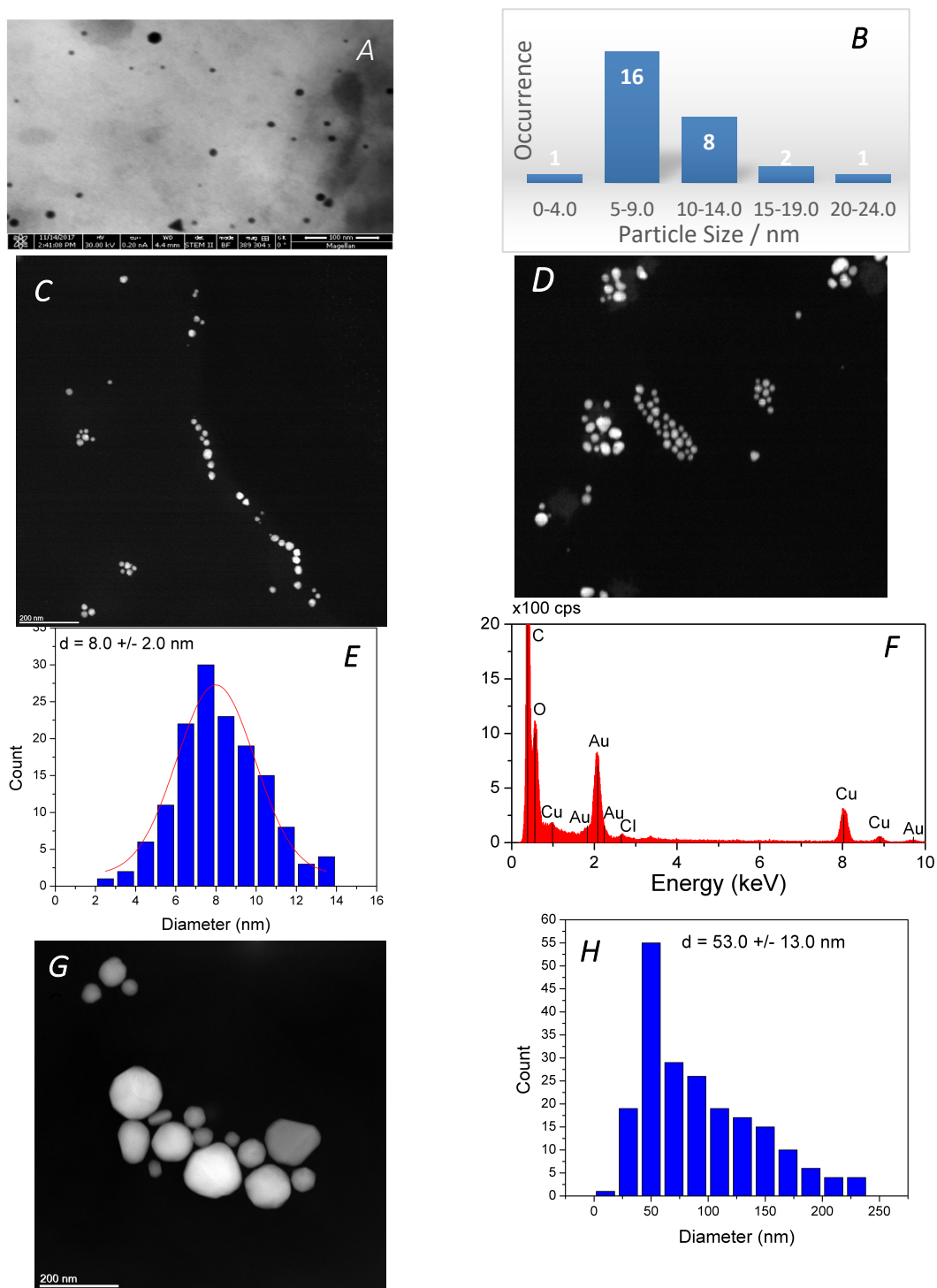


Figure S7. Temperature dependent circularly averaged small-angle X-ray scattering (SAXS) data collected on a complex fluid, lacking $[\text{AuCl}_4^-]$, (A) prepared in 100 mM acetate buffer (pH 3.66); (B) prepared in 10 mM carbonate buffer (pH 9.02).

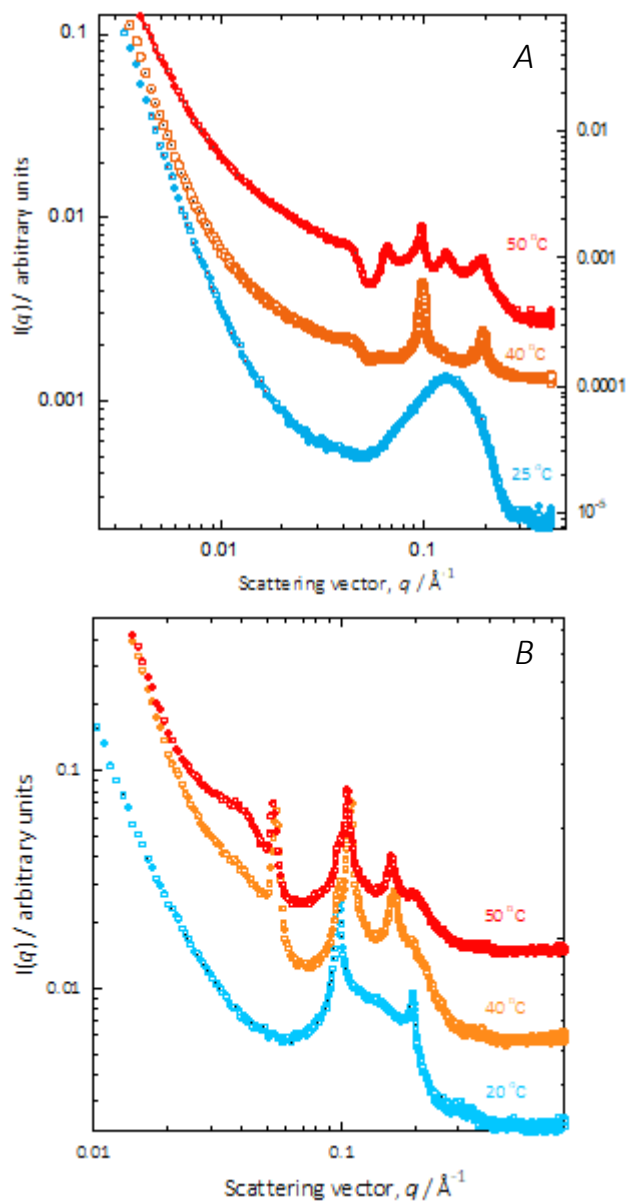


Table S1: Small angle X-ray reflections observed for a Au NP composite prepared in water at 35 °C. Reflections index to a mixture of a cubic Im3m and lamellar structure

H	K	L	q (observed) \AA^{-1}	Im3m	Lamellar
1	1	0	0.116	$\sqrt{2}$	
0	0	1	0.126		1
2	0	0	0.164	$\sqrt{4}$	
2	1	1	0.200	$\sqrt{6}$	
2	2	0	0.231	$\sqrt{8}$	
3	1	0	0.254	$\sqrt{10}$	2
2	2	2	0.284	$\sqrt{12}$	
3	2	1	0.307	$\sqrt{14}$	
4	0	0	0.328	$\sqrt{16}$	
4	1	1	0.347	$\sqrt{18}$	
3	3	2	0.382	$\sqrt{22}$	3
4	3	1	0.410	$\sqrt{26}$	
4	4	0	0.468	$\sqrt{32}$	

Figure S8

(A) Normalized temperature dependent Vis-NIR spectra collected on Au NP F98(Acr)₂/LDAO/DMPC/water composite, 0.739 ± 0.01 weight fraction (Φ_w) of water, $\Phi_s = 0.0116 \pm 0.0005$ LDAO co-surfactant, and $\Phi_l = 0.0951 \pm 0.005$ DMPC lipid, and $\Phi_p = 0.161 \pm 0.001$ F98(Acr)₂, mM Au³⁺ and (B) Au NP F68(Acr)₂/LDAO/DMPC/water 0.673 ± 0.01 weight fraction (Φ_w) of water, $\Phi_s = 0.0110 \pm 0.0005$ LDAO co-surfactant, and $\Phi_l = 0.0951 \pm 0.005$ DMPC lipid, and $\Phi_p = 0.159 \pm 0.001$ F68(Acr)₂, 6.66 mM Au³⁺.

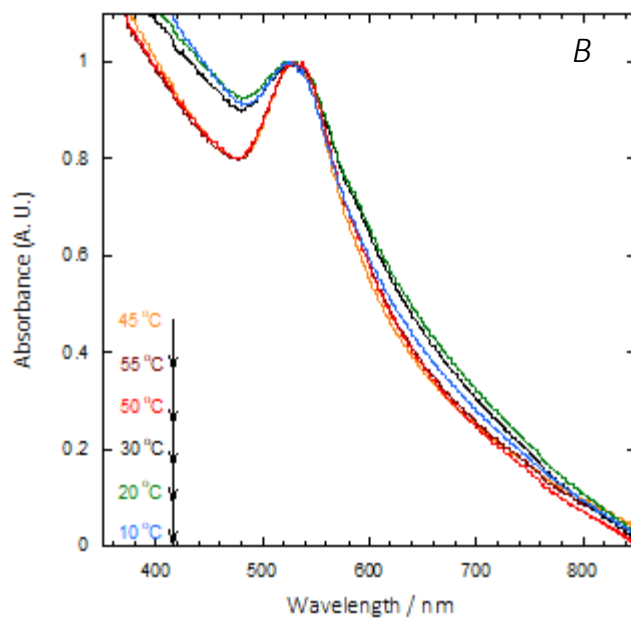
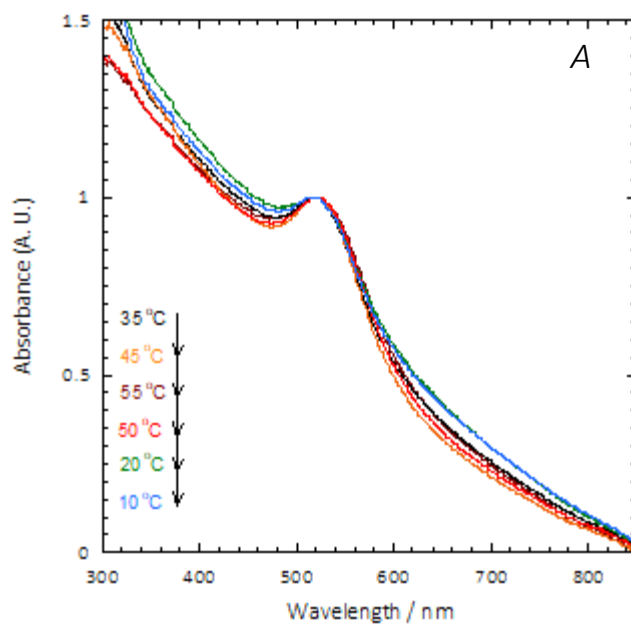


Figure S9. Temperature dependent SAXS profiles on a Au NP composite prepared with acidic (pH 3.6) buffer. The composite contains 0.733 ± 0.001 weight fraction (Φ_w) water, $\Phi_s = 0.0371 \pm 0.0005$ Triton-X 100 co-surfactant, $\Phi_l = 0.0852 \pm 0.0005$ DMPC lipid, and $\Phi_p = 0.114 \pm 0.005$ Tetronic 1107(Acr)₄, $[\text{Au}^{3+}] = 6.41\text{mM}$

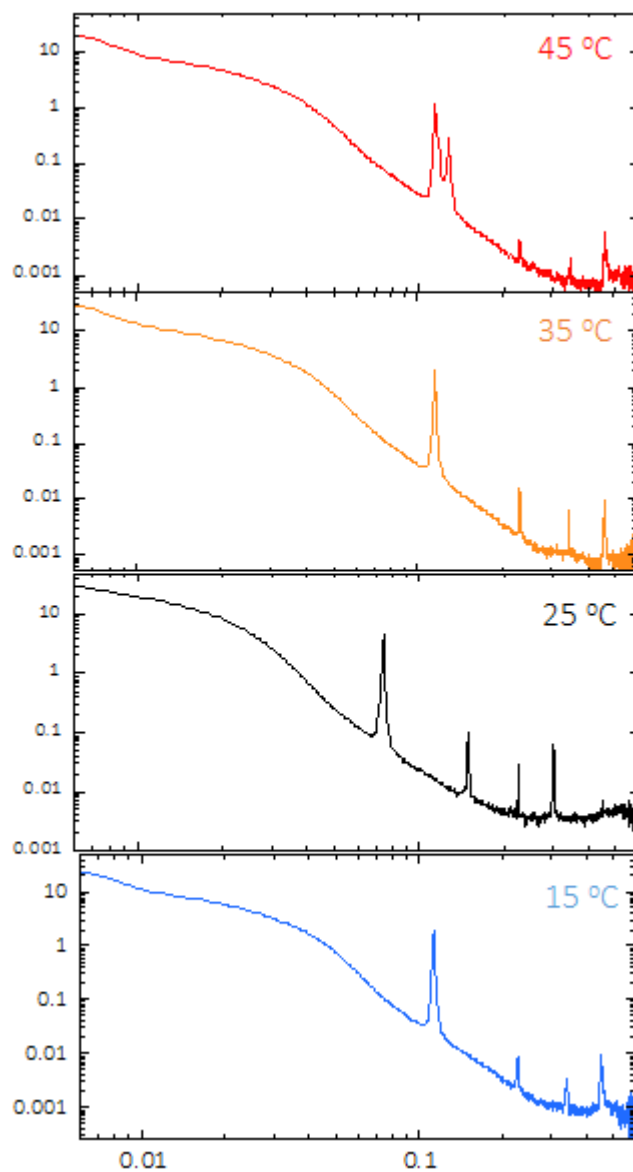


Figure S10

(A) Normalized temperature dependent Vis-NIR spectra collected on Au NP composite prepared with un-buffered water composed of 0.735 ± 0.001 weight fraction (Φ_w) water, $\Phi_s = 0.0372 \pm 0.0005$ Triton-X 100 co-surfactant, $\Phi_l = 0.0837 \pm 0.0005$ DMPC lipid, and $\Phi_p = 0.143 \pm 0.005$ Tetronic 1107(Acr)₄, $[\text{Au}^{3+}] = 7.13$ mM. (B) Normalized temperature dependent Vis-NIR collected on an acidic buffered composite. (C) Normalized temperature dependent Vis-NIR collected on a pH 9 carbonate buffered composite.

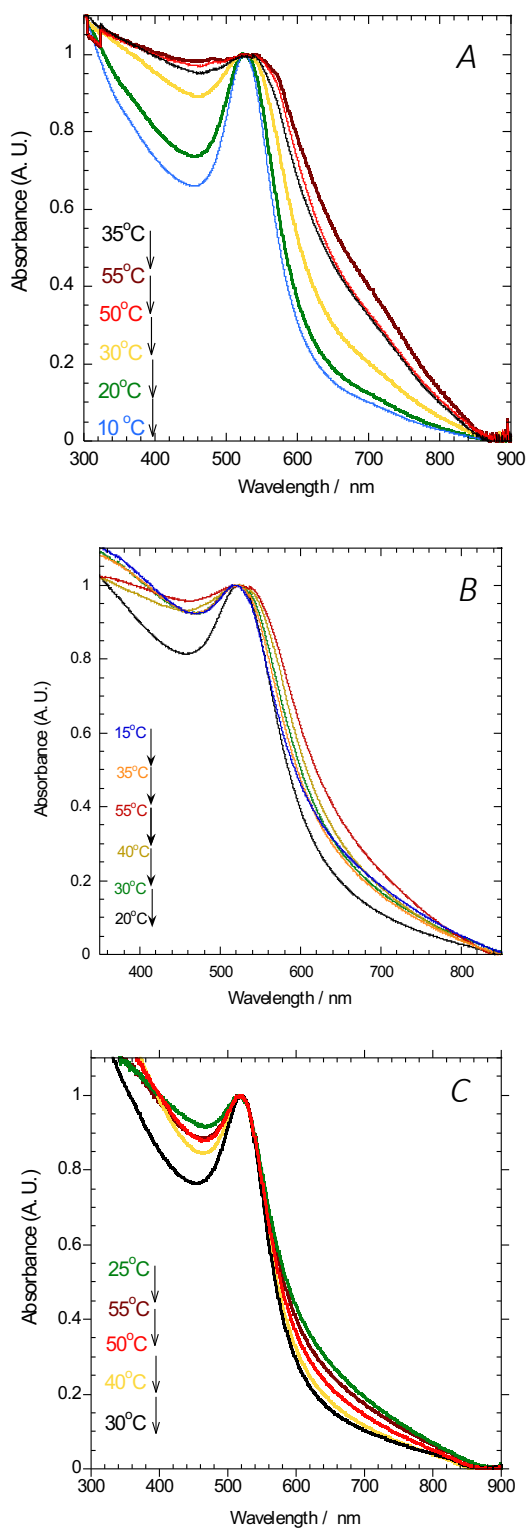


Figure S11 Temperature dependent SAXS profiles on Au NP hydrogel composite prepared in basic pH 9.03 buffer. The composite contains 0.733 ± 0.001 weight fraction (Φ_w) water, $\Phi_s = 0.0371 \pm 0.0005$ Triton-X 100 co-surfactant, $\Phi_l = 0.0852 \pm 0.0005$ DMPC lipid, and $\Phi_p = 0.114 \pm 0.005$ Tetronic 1107(Acr)₄, $[\text{Au}^{3+}] = 6.41\text{mM}$

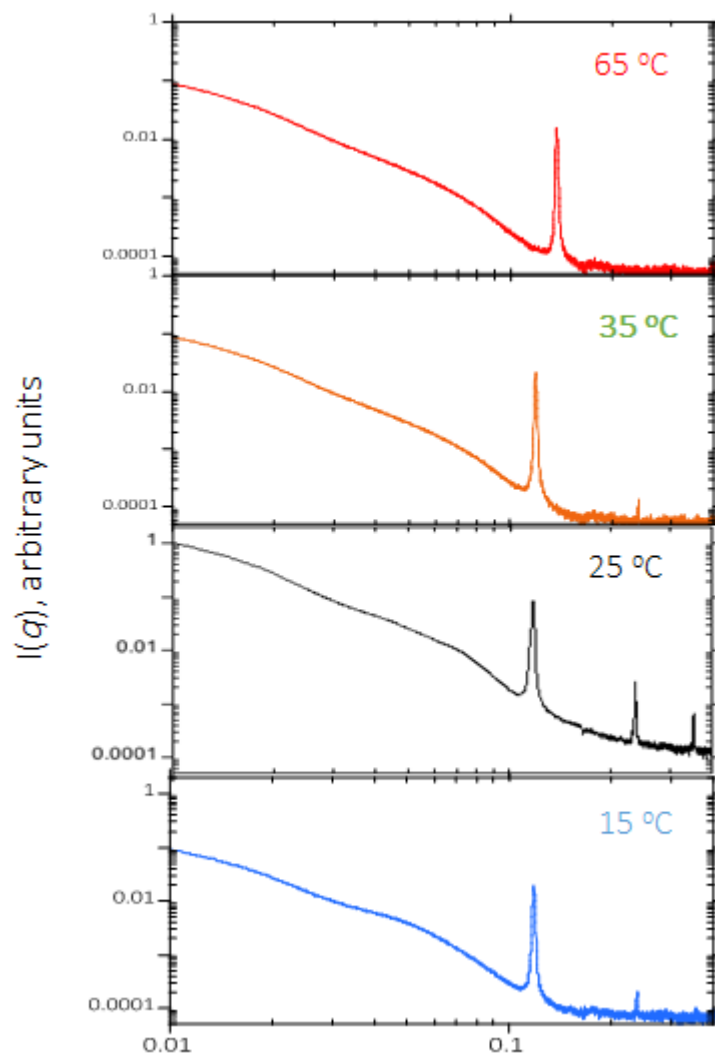


Figure S12 Diffuse reflectance optical spectrum recorded on a dehydrated (de-swollen) Au NP composite. Composite prepared with un-buffered water composed of 0.730 ± 0.001 weight fraction (Φ_w) water, $\Phi_s = 0.0370 \pm 0.0005$ Triton-X 100 co-surfactant, $\Phi_l = 0.0881 \pm 0.0005$ DMPC lipid, and $\Phi_p = 0.145 \pm 0.005$ Tetronic 1107(Acr)₄, $[\text{Au}^{3+}] = 7.01$ mM.

