

## Supporting Information

### Synthesis and Self-Assembly of Photoacid-Containing Block Copolymers based on 1-Naphthol

*Felix Wendler,<sup>a,b</sup> Jessica C. Tom<sup>a,b</sup> and Felix H. Schacher<sup>a,b\*</sup>*

<sup>a</sup> Institute of Organic and Macromolecular Chemistry (IOMC), Friedrich Schiller University, Jena, Humboldtstrasse 10, 07743 Jena, Germany

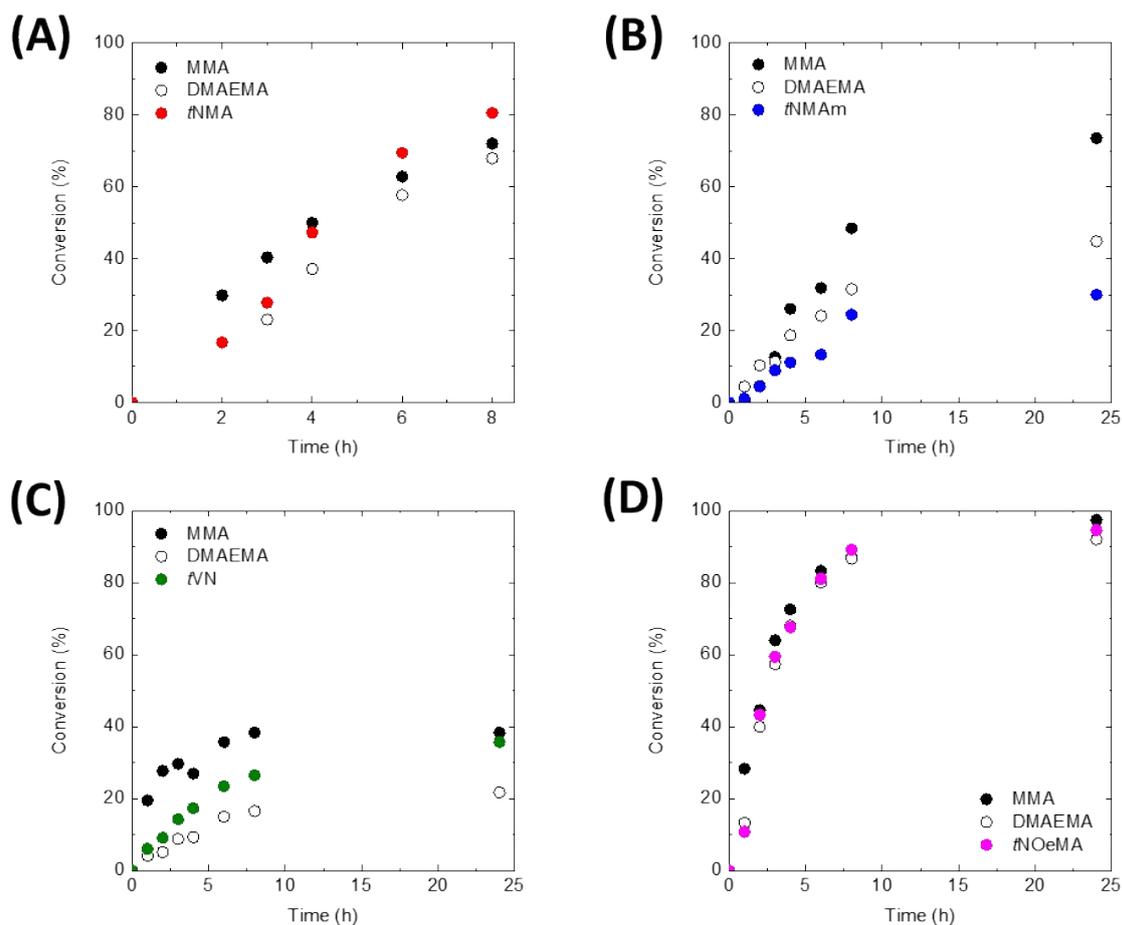
E-Mail: felix.schacher@uni-jena.de

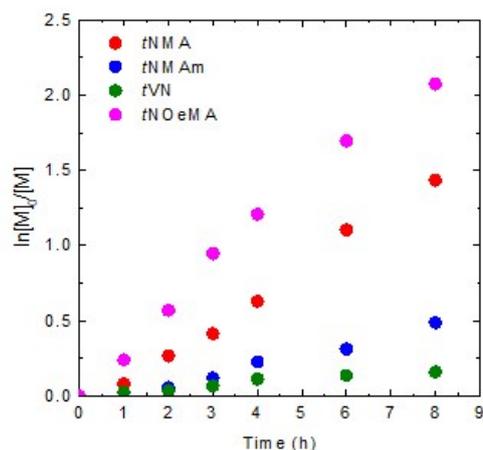
<sup>b</sup> Jena Center for Soft Matter (JCSM), Friedrich Schiller University Jena, Philosophenweg 7, 07743 Jena, Germany

**Table S1: Optimization of RAFT terpolymerization with *t*NMAm or *t*VN.**

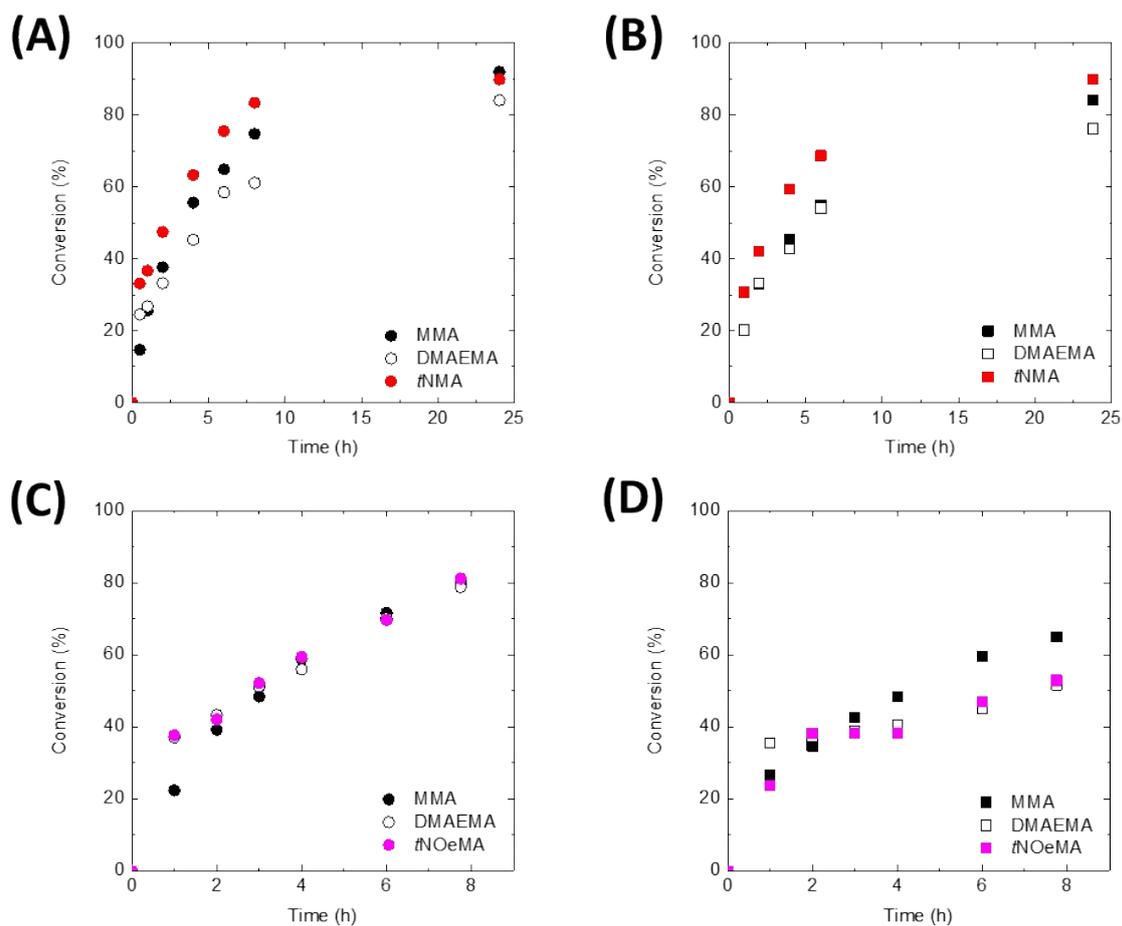
Entry	Monomer	Solvent	CTA	T (°C)	I	$M_n^{a,SEC}$ (kg mol <sup>-1</sup> )	$\bar{D}^a$	Overall Conversion <sup>b</sup> (%)
1	<i>t</i> NMAm	1,4-dioxane	CPADB	70	AIBN	9	1.24	25
2				90	ACHN	9.5	1.20	27
3				ACHN	6	1.10	4	
4		bulk	CPADB	70	AIBN	5	1.14	14
5				70	AIBN	11	1.61	44
6				70	AIBN	5	1.09	16
7	<i>t</i> VN	1,4-dioxane	CPADB	70	AIBN	5.5	1.23	24
8				90	AIBN	10	1.33	46
9				90	ACHN	7	1.28	41
10		bulk	CPADB	70	AIBN	5.5	1.31	24
11				70	AIBN	8	1.22	46
12				Toluene	70	AIBN	2	1.02

<sup>a</sup> SEC (DMAc/LiCl) (PMMA calibration), <sup>b</sup> <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>).

**Figure S1: Monomer conversion vs time plot for (A) *t*NMA, (B) *t*NMAm, (C) *t*VN, and (D) *t*NOeMA, copolymerized with MMA and DMAEMA with a [monomer]:[CTA] = 125:1. TBS was used as an internal standard (IS) for (B) and (C).**



**Figure S2:** Semilogarithmic plot for *t*NMA, *t*NMAm, *t*VN, and *t*NOeMA, copolymerized with MMA and DMAEMA with a [monomer]:[CTA] = 125:1.



**Figure S3:** Monomer conversion vs time plot for *t*NMA with a [monomer]:[CTA] ratio of (A) 200:1 or (B) 500:1; and *t*NOeMA with a [monomer]:[CTA] ratio of (C) 200:1 or (D) 500:1, copolymerized with MMA and DMAEMA.

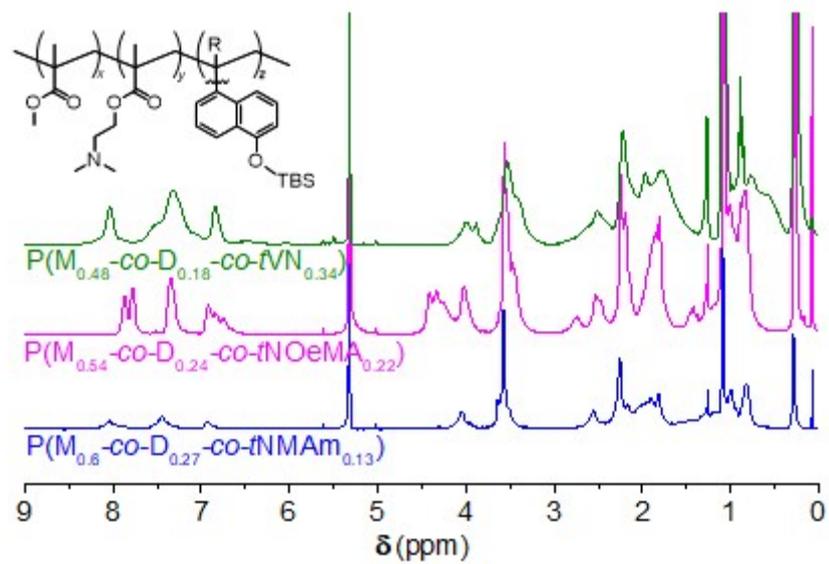


Figure S4:  $^1\text{H-NMR}$  spectra of  $\text{P}(\text{M}_{0.6}\text{-co-D}_{0.27}\text{-co-tNMAM}_{0.13})$ ,  $\text{P}(\text{M}_{0.54}\text{-co-D}_{0.24}\text{-co-tNOeMA}_{0.22})$  and  $\text{P}(\text{M}_{0.48}\text{-co-D}_{0.18}\text{-co-tVN}_{0.34})$ .

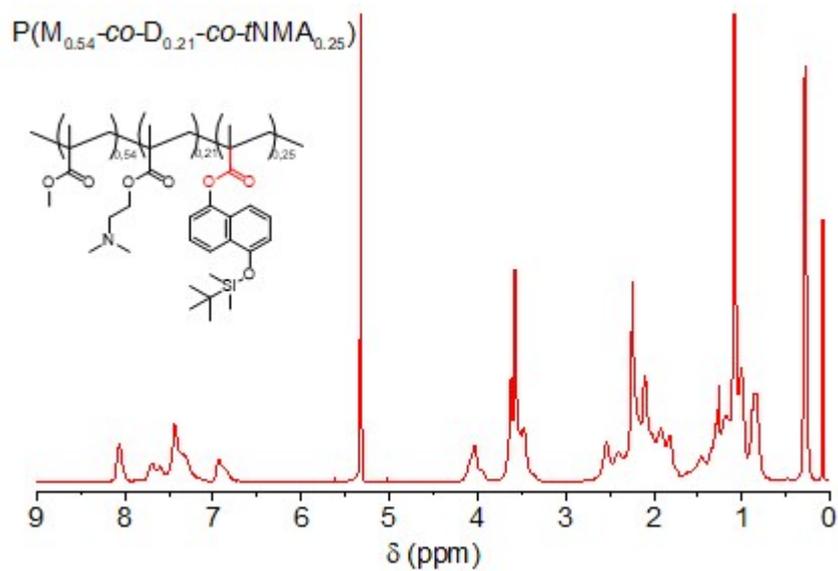


Figure S5:  $^1\text{H-NMR}$  spectrum of  $\text{P}(\text{M}_{0.54}\text{-co-D}_{0.21}\text{-co-tNMA}_{0.25})$ .

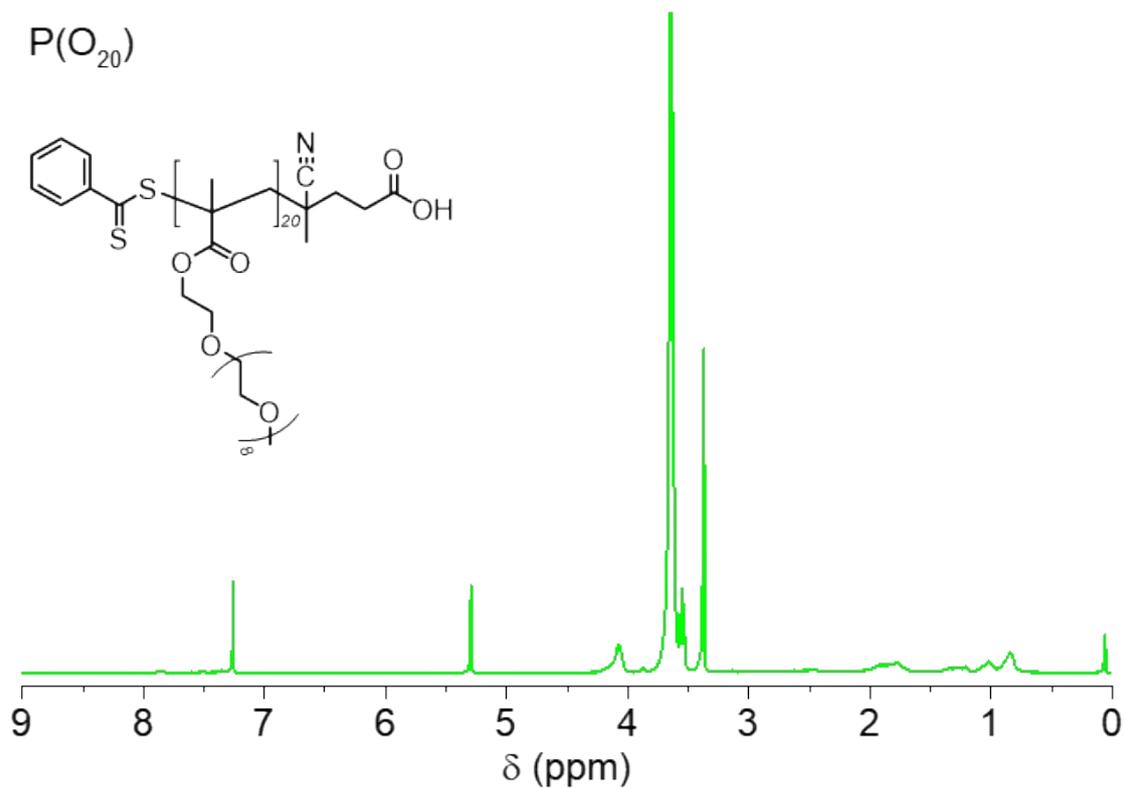


Figure S6: <sup>1</sup>H-NMR spectrum of P(O<sub>20</sub>).

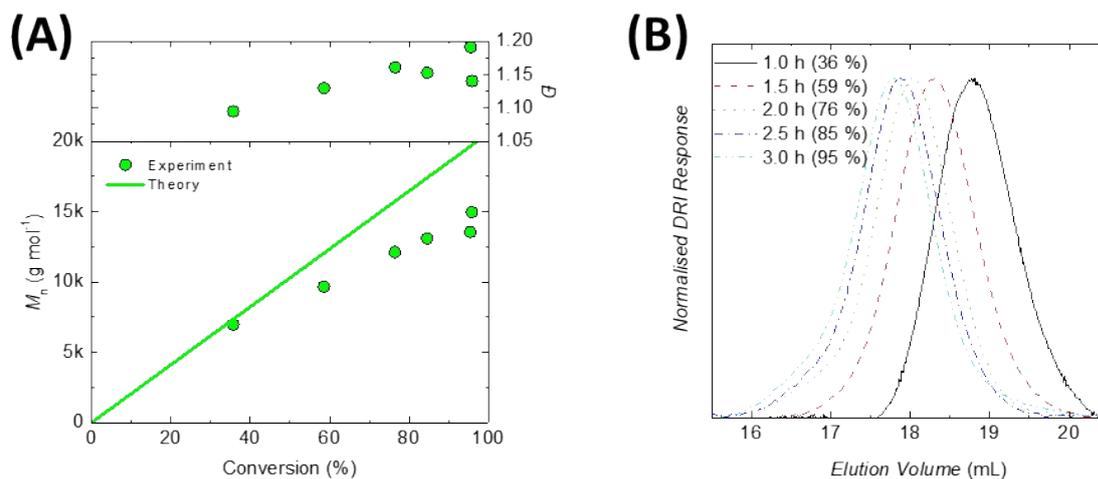


Figure S7: (A) Overall monomer conversion vs time plot for MEO<sub>9</sub>MA homopolymerization with a [monomer]:[CTA] = 40:1; and the corresponding (B) SEC traces after different reaction times.

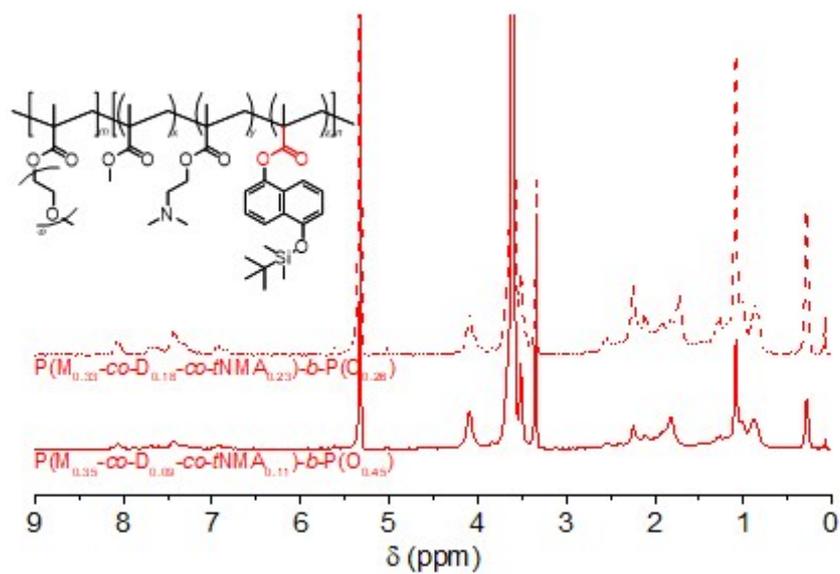


Figure S8:  $^1\text{H-NMR}$  spectra of  $\text{P}(\text{M}_{0.33}\text{-co-D}_{0.18}\text{-co-tNMA}_{0.23})\text{-b-P}(\text{O})_{0.26}$  and  $\text{P}(\text{M}_{0.35}\text{-co-D}_{0.09}\text{-co-tNMA}_{0.11})\text{-b-P}(\text{O})_{0.45}$ .

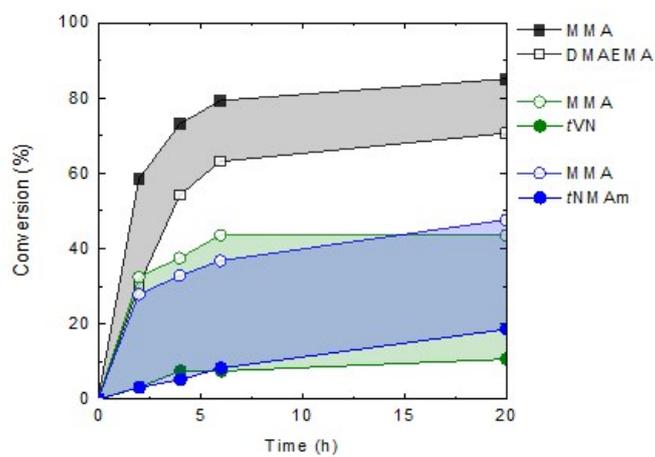


Figure S9: Monomer conversion vs time plot for copolymerization of various monomers (DMAEMA,  $t\text{VN}$ , and  $t\text{NMAm}$ ) with MMA (reaction conditions: AIBN, 1,4-dioxane  $70\text{ }^\circ\text{C}$ ).

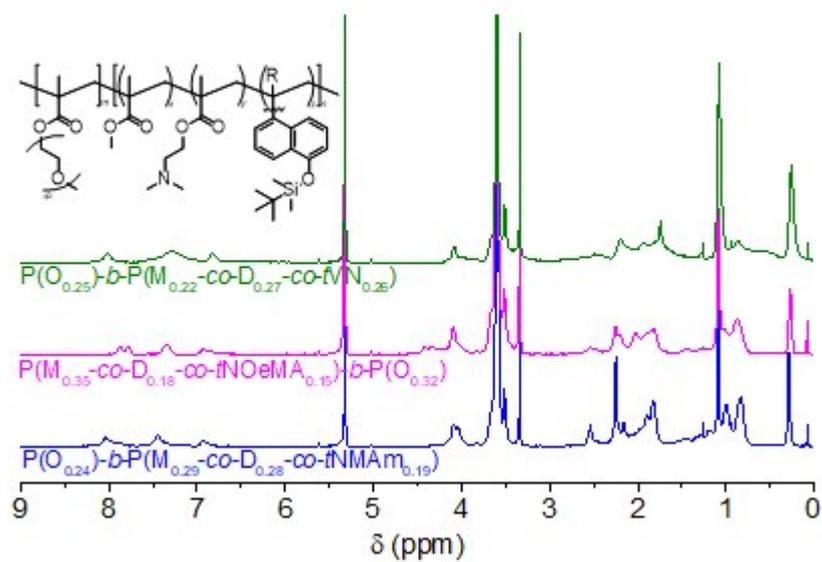


Figure S10:  $^1\text{H-NMR}$  spectra of  $\text{P}(\text{O}_{0.24})\text{-}b\text{-P}(\text{M}_{0.29}\text{-}co\text{-D}_{0.28}\text{-}co\text{-}t\text{NMAM}_{0.19})$ ,  $\text{P}(\text{M}_{0.40}\text{-}co\text{-D}_9\text{-}co\text{-}t\text{NOeMA}_{13})\text{-}b\text{-P}(\text{O}_{0.32})$  and  $\text{P}(\text{O}_{0.25})\text{-}b\text{-P}(\text{M}_{0.22}\text{-}co\text{-D}_{0.27}\text{-}co\text{-}t\text{VN}_{0.26})$ .

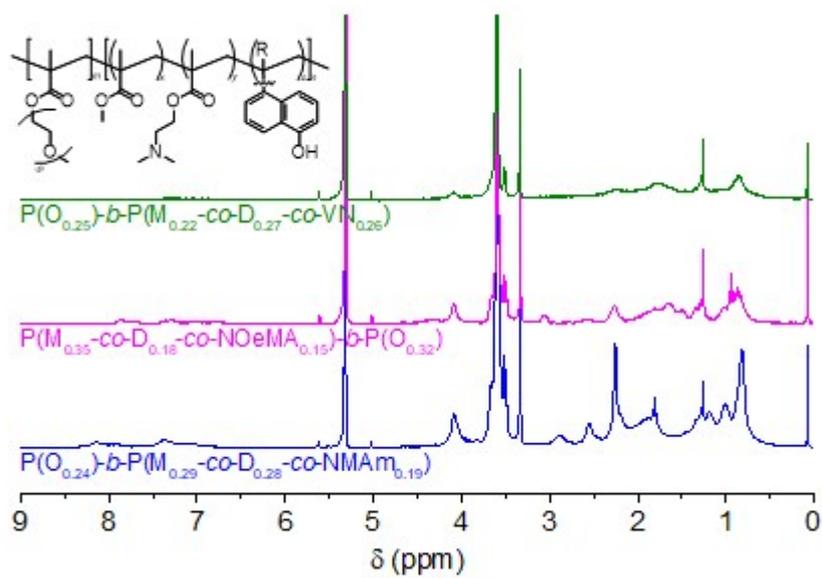


Figure S11:  $^1\text{H-NMR}$  spectra of  $\text{P}(\text{O}_{0.24})\text{-}b\text{-P}(\text{M}_{0.29}\text{-}co\text{-D}_{0.28}\text{-}co\text{-NMAM}_{0.19})$ ,  $\text{P}(\text{M}_{0.40}\text{-}co\text{-D}_9\text{-}co\text{-NOeMA}_{13})\text{-}b\text{-P}(\text{O}_{0.32})$  and  $\text{P}(\text{O}_{0.25})\text{-}b\text{-P}(\text{M}_{0.22}\text{-}co\text{-D}_{0.27}\text{-}co\text{-VN}_{0.26})$ .

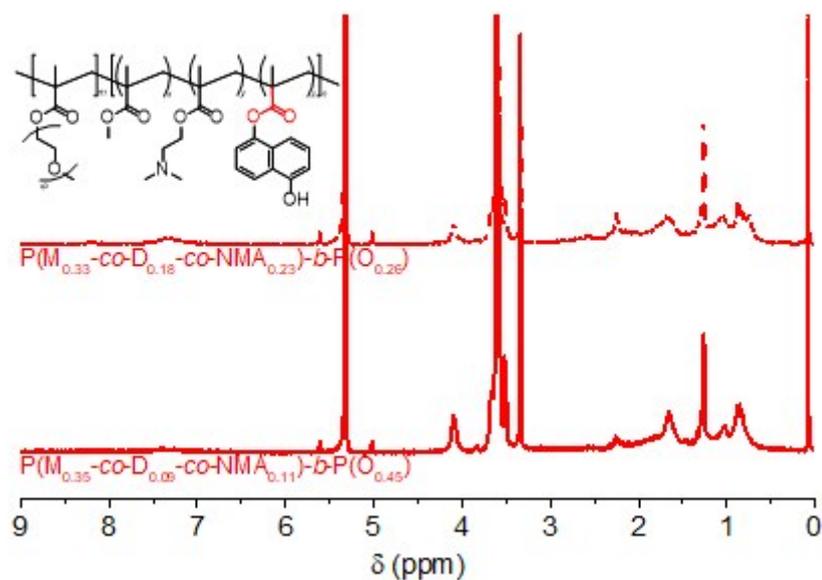


Figure S12:  $^1\text{H-NMR}$  spectra of  $\text{P}(\text{M}_{0.35}\text{-co-D}_{0.09}\text{-co-NMA}_{0.11})\text{-b-P}(\text{O}_{0.45})$  and  $\text{P}(\text{M}_{0.33}\text{-co-D}_{0.18}\text{-co-NMA}_{0.23})\text{-b-P}(\text{O}_{0.26})$ .

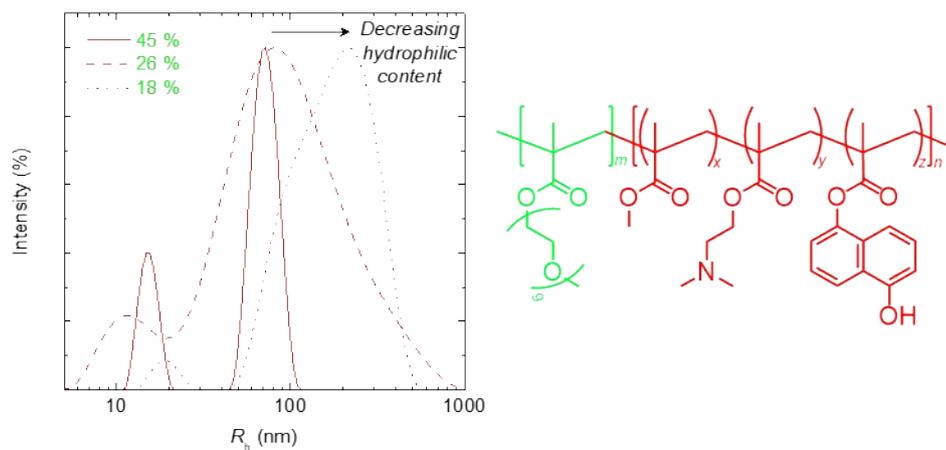


Figure S13: Intensity-weighted hydrodynamic radius of block tetrapolymers with varying hydrophilic contents: (A)  $\text{P}(\text{M}_{0.35}\text{-co-D}_{0.09}\text{-co-NMA}_{0.11})\text{-b-P}(\text{O}_{0.45})$ , (B)  $\text{P}(\text{M}_{0.33}\text{-co-D}_{0.18}\text{-co-NMA}_{0.23})\text{-b-P}(\text{O}_{0.26})$ , and (C)  $\text{P}(\text{M}_{0.31}\text{-co-D}_{0.20}\text{-co-NMA}_{0.31})\text{-b-P}(\text{O}_{0.18})$  corresponding to 45, 26, and 18 %, respectively.

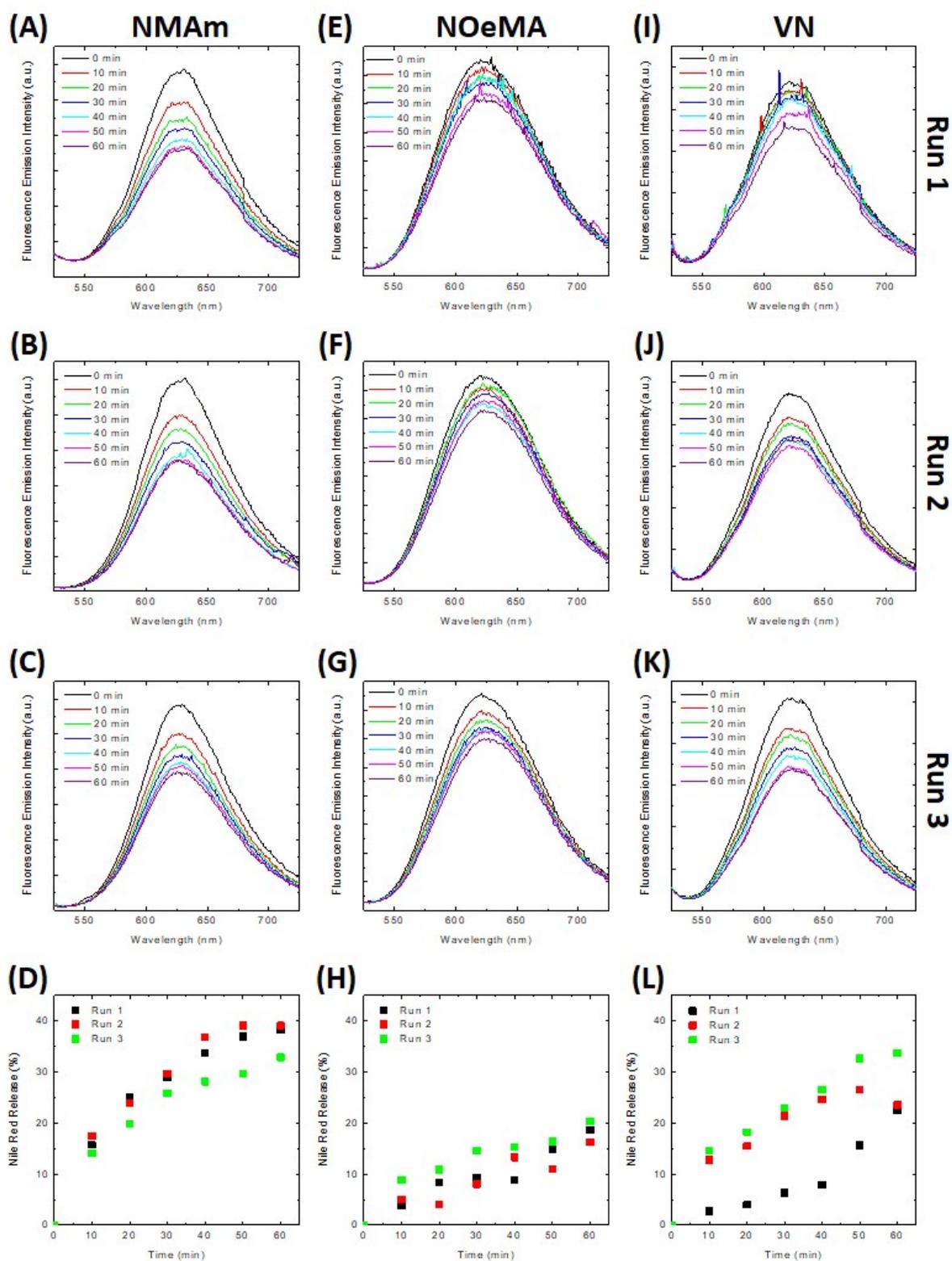


Figure S14: Fluorescence emission spectra of NR-loaded micelles of  $P[(M_x-co-D_y-co-'N')_n-b-(O)_m]$  upon 365 nm light irradiation and corresponding NR release where “N” is NMAm (A)–(D), NOeMA (E)–(H), and VN (I)–(L).