

Electronic Supplementary Information

Effect of the addition of salt to Pickering emulsion polymerizations using polymeric nanogels as stabilizer

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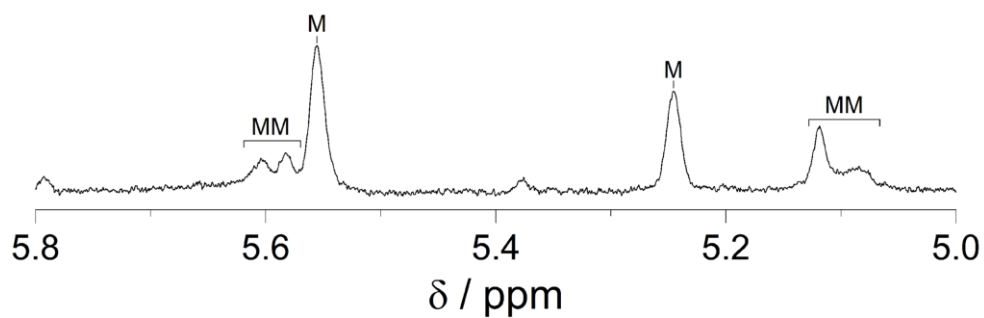


Figure S1. Vinyl protons region of the ^1H -NMR spectrum of a 1.0 wt% dispersion of nanogels in D_2O showing residual macromonomer (MM) and crosslinker (M), *i.e.* trimethylolpropane methacrylate, vinyl groups.

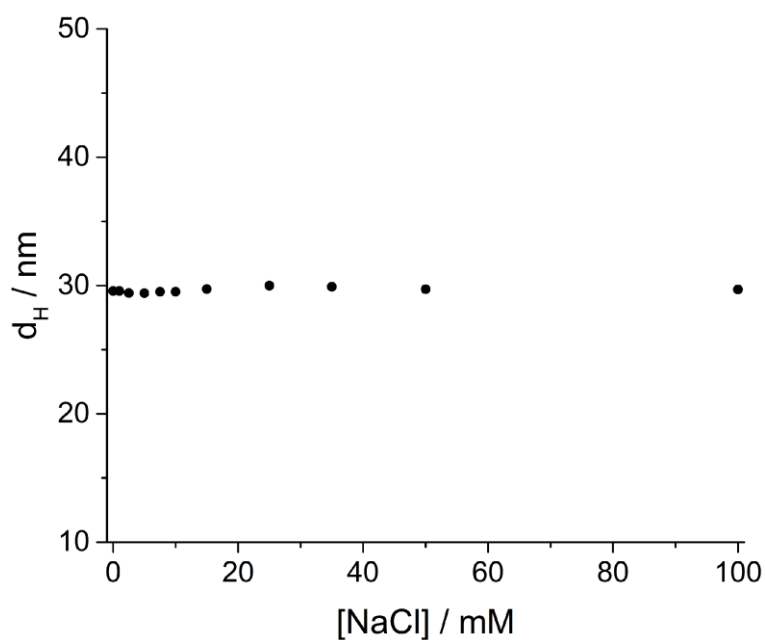


Figure S2. Hydrodynamic diameter (d_H) of 1.0 wt% of N1 nanogels in water at pH 8.8 and in presence of varying [NaCl].

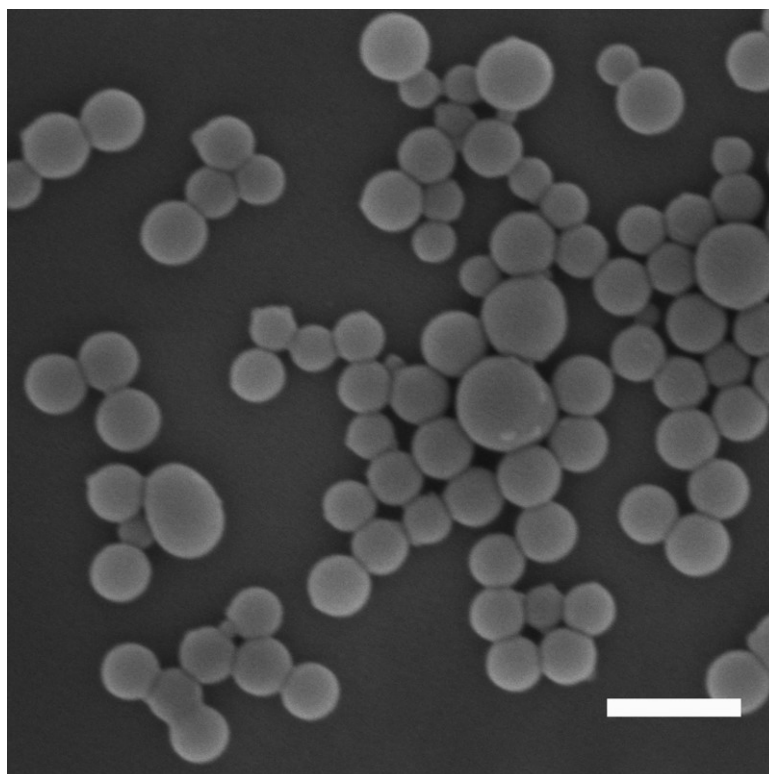


Figure S3. Emulsion polymerizations of styrene performed in the presence of nanogels as stabilizers at $[\text{NaCl}] = 2.5 \text{ mM}$. Scale bar: 300 nm.

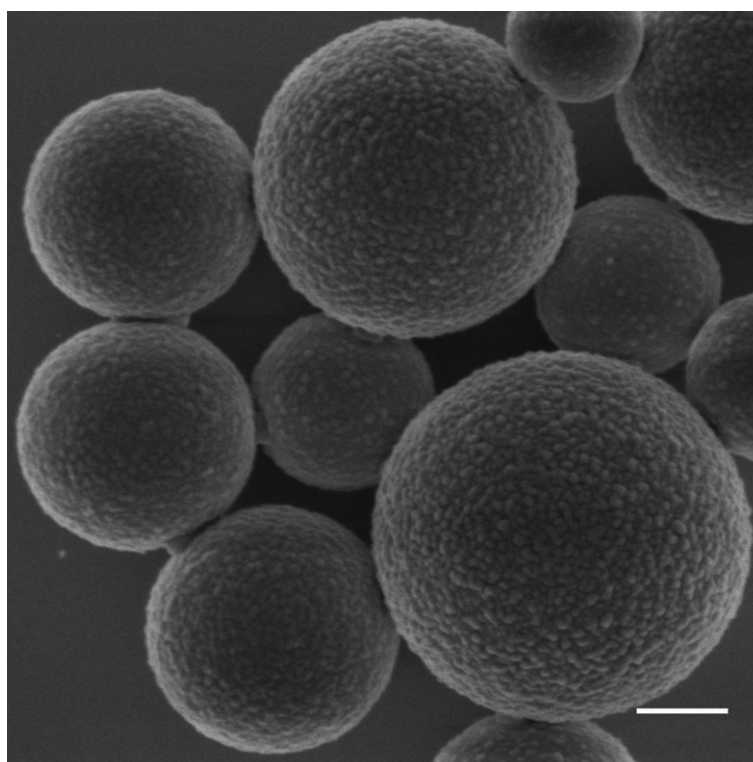


Figure S4. Emulsion polymerizations of styrene performed in the presence of nanogels as stabilizers at $[\text{NaCl}] = 25.0 \text{ mM}$ and $[\text{KPS}] = 0.25 \text{ mM}$. Scale bar: 300 nm.

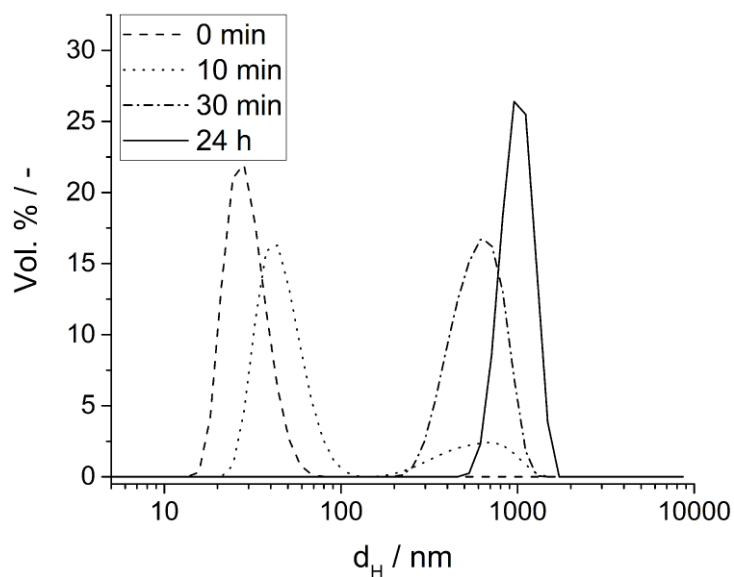


Figure S5. Particle size distribution at different stages of the emulsion polymerization of styrene carried out in presence of 25.0 mM [NaCl] and [KPS] = 0.25 mM. The time 0 sample was recorded before initiator injection.

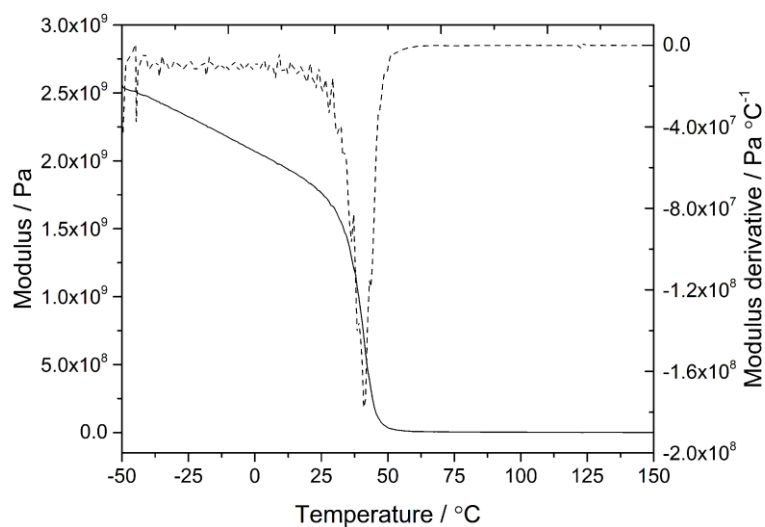


Figure S6. Dynamic mechanical analysis carried out on a film (thickness 1.1 mm) of the styrene:*n*-butyl acrylate latex particles made in presence of [NaCl] = 25.0 mM. The solid line represents the storage modulus, whereas the dotted line shows its first derivative with respect to temperature. The T_g measured as the minimum of the derivative = 40.9°C.

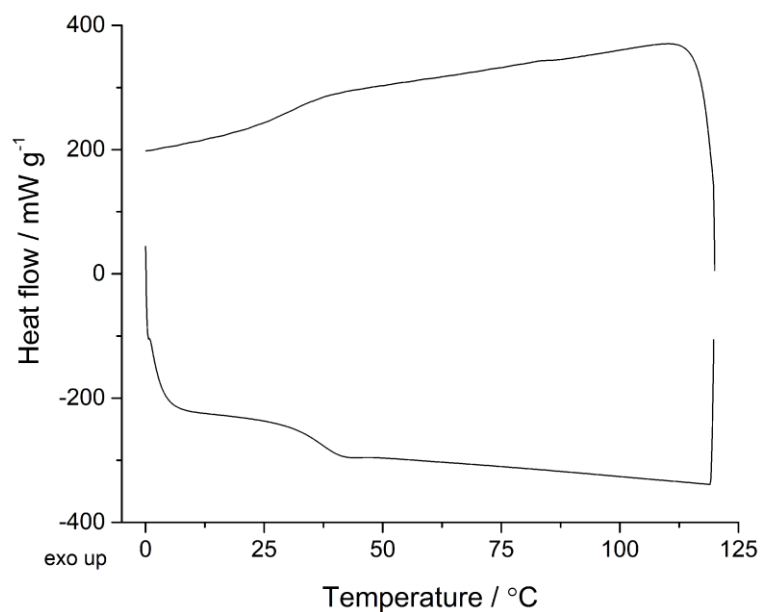


Figure S7. Differential scanning calorimetry (DSC) analysis carried out on a film (thickness 1.1 mm) of the styrene:*n*-butyl acrylate latex particles made in presence of [NaCl] = 25.0 mM. The T_g measured as the midpoint of the flex was 36.2°C and 30.7°C respectively for the heating (bottom) and cooling (top) cycle.



Figure S8. Film formation of the Sty/BA = 1.45:1 w:w latex made in presence of [NaCl] = 25.0 mM and dried at 40°C.