Supporting Information

IrO₂-incorporated La_{0.8}Sr_{0.2}MnO₃ as bifunctional oxygen catalysts with

enhanced activities

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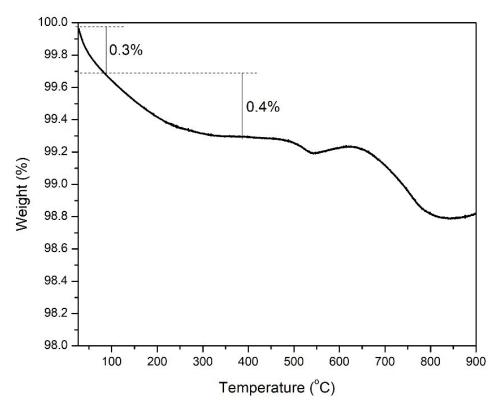


Figure S1. TGA test of La_{0.8}Sr_{0.2}MnO₃ under O₂ atmosphere.

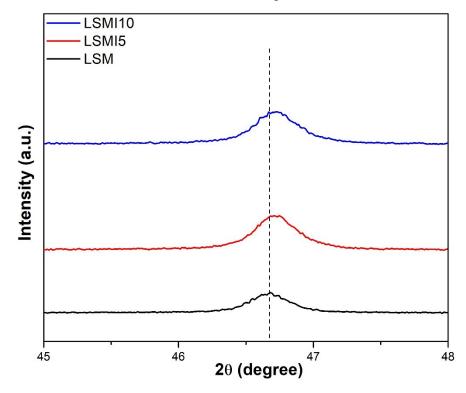


Figure S2. The enlarged XRD of the pristine $La_{0.8}Sr_{0.2}MnO_3$ (LSM), 5 wt% IrO₂-La_{0.8}Sr_{0.2}MnO₃ (LSMI) and 10 wt% IrO₂-La_{0.8}Sr_{0.2}MnO₃ (LSMI10) in the range of 20 from 45-48°, respectively.

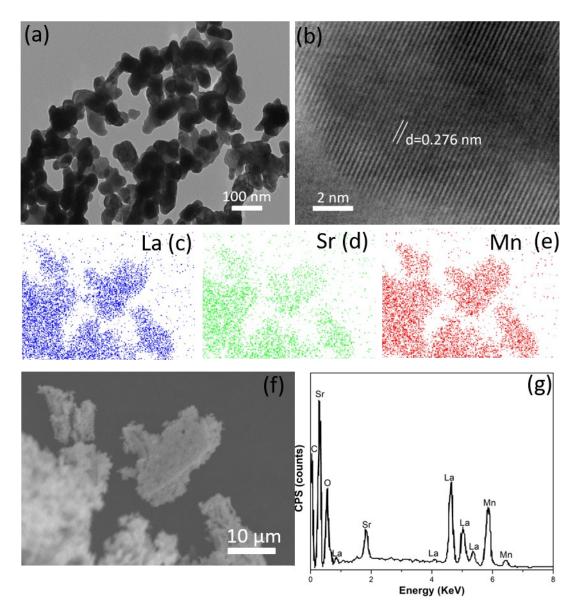


Figure S3. TEM (a), HRTEM (b) and SEM (f) image of the pristine La_{0.8}Sr_{0.2}MnO₃; EDX mapping of (c) La, (d) Sr, (e) Mn and (g) EDS spectra of pristine La_{0.8}Sr_{0.2}MnO₃, respectively.

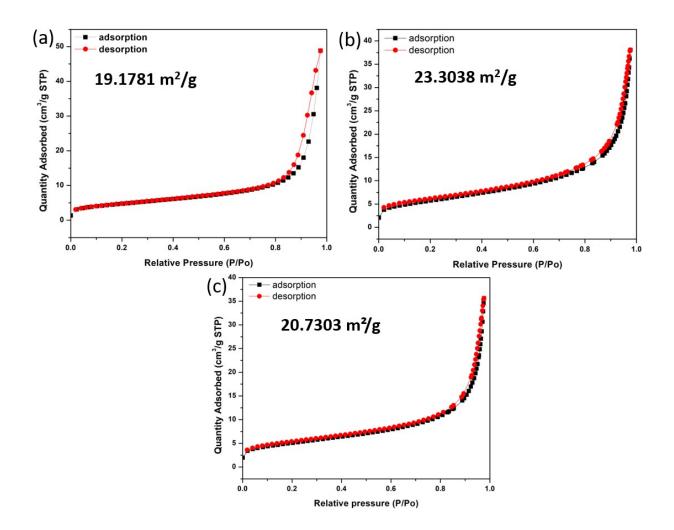


Figure S4. BET test for the pristine $La_{0.8}Sr_{0.2}MnO_3$ (a), 5 wt% $IrO_2-La_{0.8}Sr_{0.2}MnO_3$ (b) and 10 wt% $IrO_2-La_{0.8}Sr_{0.2}MnO_3$ (c), respectively.

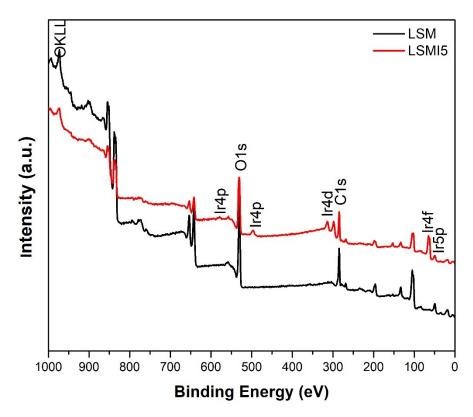


Figure S5. The XPS entire survey scan for the pristine $La_{0.8}Sr_{0.2}MnO_3$ (LSM) and 5 wt% IrO₂- $La_{0.8}Sr_{0.2}MnO_3$ (LSMI5), respectively.

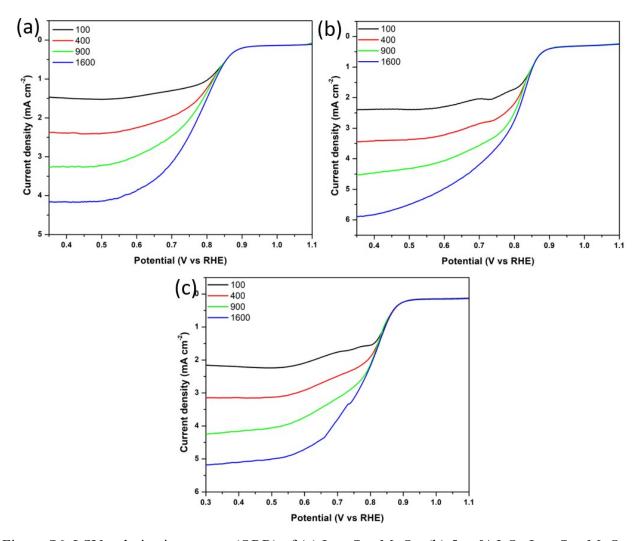


Figure S6. LSV polarization curves (ORR) of (a) $La_{0.8}Sr_{0.2}MnO_3$, (b) 5 wt% $IrO_2-La_{0.8}Sr_{0.2}MnO_3$ and (c) 10 wt% $IrO_2-La_{0.8}Sr_{0.2}MnO_3$ in O_2 -saturated 0.1 mol L⁻¹ KOH with a scan rate of 10 mV s⁻¹ under four (100, 400, 900 and 1600 rpm) different rotation rates.

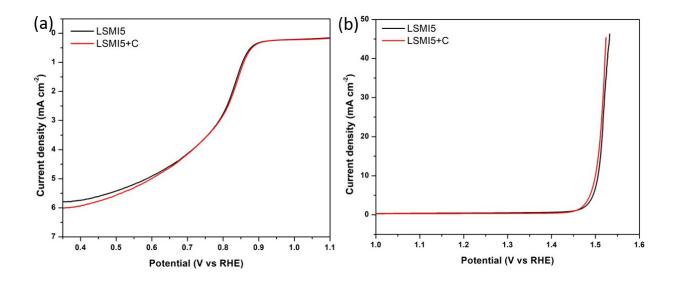


Figure S7. LSV curves of ORR (a) and OER (b) performance comparison between LSMI5 with/without carbon black in 0.1 mol L^{-1} KOH solution with a scan rate of 10 mV s⁻¹ at a rotation speed of 1600 rpm.

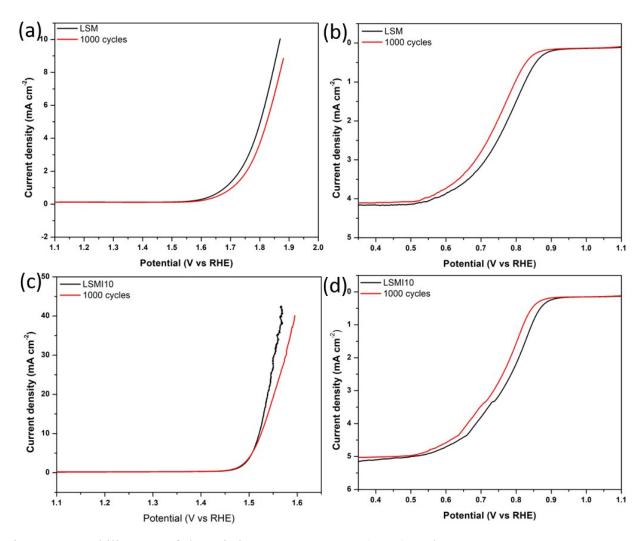


Figure S8. Stability test of the pristine $La_{0.8}Sr_{0.2}MnO_3$ (LSM) and 10 wt% IrO₂-La_{0.8}Sr_{0.2}MnO₃ (LSMI10). LSVs of OER (a, c) and ORR (b, d) tested in O₂-saturated 0.1 mol L⁻¹ KOH solution before and after 1000 cycles at a sweep rate of 50 mV s⁻¹ with a rotation speed of 1600 rpm.