Supporting Information

Oxygen Vacancies Confined in Co₃O₄ Quantum Dots for Promoted Oxygen Evolution Electrocatalysis

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S1. Crystal structure of pristine Co₃O₄ and oxygen defected Co₃O₄ quantum dots

Figure S1 (a) The crystal structure of $p-Co_3O_4$ sample. (b) The schematic diagram of oxygen defects in the framework of Co_3O_4 material.

S2. TEM image and size distribution of p-Co₃O₄ quantum dots



Figure S2 (a) TEM image of the obtained pristine Co_3O_4 quantum dots. (b) Distribution diagram of particle size of pristine Co_3O_4 quantum dots.

S3. HRTEM image of p-Co₃O₄ quantum dots



Figure S3 The HRTEM image of synthesized pristine Co₃O₄ quantum dots.

S4. The XRD patterns of various Co₃O₄ products at different temperature



Figure S4 a-b) XRD patterns of as-obtained various Co_3O_4 quantum dots that synthesized at different treatment temperature.



S5. X-ray photoelectron spectroscopy investigation of p-Co₃O₄ sample

Figure S5 The XPS survey of as-obtained p-Co₃O₄ products.

S6. X-ray photoelectron spectroscopy investigation of Co₃O₄-200



Figure S6 The XPS survey of as-obtained Co₃O₄-200 products.



S7. The Co 2p XPS spectra of various Co₃O₄ samples

Figure S7 The Co2p XPS survey of as-obtained Co₃O₄-200, Co₃O₄-210 and Co₃O₄-250 products.

S8. The stability test of Co₃O₄-200 for OER process



Figure S8 Chronoamperometric response of Co_3O_4 -200 catalyst at the applied potential of 0.55V vs. Ag/AgCl.

S9. The stability test of Co₃O₄-170 for OER process



Figure S9 Chronoamperometric response of Co_3O_4 -170 catalyst at the applied potential of 0.55V vs. Ag/AgCl.