

Electronic Supplementary Information for

**CO₂ fixation and transformation on a thiolate-bridged dicobalt
scaffold under oxidising conditions**

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I. X-ray Crystallographic Data

Table S1. Crystallographic data for **1** and **2**

	1	2
Formula	C ₂₄ H ₄₀ Co ₂ S ₂	C ₂₅ H ₄₀ Co ₂ O ₃ S ₂
Formula weight	510.54	570.55
Crystal dimensions (mm ³)	0.30 × 0.24 × 0.20	0.28 × 0.24 × 0.21
Crystal system	Monoclinic	Monoclinic
Space group	P 21/m	C 2/c
a (Å)	8.0243(7)	18.7475(6)
b (Å)	15.9984(13)	12.8019(4)
c (Å)	10.5134(9)	10.6678(3)
α (°)	90.00	90.00
β (°)	111.481(2)	93.1660(10)
γ (°)	90.00	90.00
Volume (Å ³)	1255.92(19)	2556.40(14)
Z	2	4
T (K)	223(2)	173(2)
D _{calcd} (g cm ⁻³)	1.350	1.482
μ (mm ⁻¹)	1.495	1.486
F (000)	540	1200
No. of rflns. collected	15786	24564
No. of indep. rflns. /R _{int}	2300 / 0.0543	2253 / 0.0262
No. of obsd. rflns. [I ₀ > 2σ(I ₀)]	2031	2112
Data / restraints / parameters	2300 / 66 / 165	2253 / 0 / 146
R ₁ / wR ₂ [I ₀ > 2σ(I ₀)]	0.0486 / 0.1310	0.0315 / 0.0945
R ₁ / wR ₂ (all data)	0.0559 / 0.1338	0.0334 / 0.0957
GOF (on F ²)	1.190	1.064
Largest diff. peak and hole (e Å ⁻³)	0.562 / -0.715	0.499 / -0.469
CCDC No.	1904424	1904422

Table S2. Crystallographic data for **3[BF₄]** and **4[BF₄]**

	3[BF₄]	4[BF₄]
Formula	C ₂₅ H ₄₁ BCo ₂ F ₄ O ₃ S ₂	C ₂₆ H ₄₃ BCo ₂ F ₄ O ₃ S ₂
Formula weight	658.37	672.39
Crystal dimensions (mm ³)	0.32 × 0.26 × 0.24	0.32 × 0.26 × 0.20
Crystal system	Triclinic	Monoclinic
Space group	P $\bar{1}$	P 2 ₁ /n
a (Å)	8.5950(5)	8.4551(6)
b (Å)	11.9610(7)	13.8353(11)
c (Å)	14.2279(8)	25.1199(19)
α (°)	76.4760(10)	90.00
β (°)	86.4300(10)	96.7756(16)
γ (°)	88.2390(10)	90.00
Volume (Å ³)	1419.18(14)	2918.0(4)
Z	2	4
T (K)	173(2)	173(2)
D_{calcd} (g cm ⁻³)	1.541	1.531
μ (mm ⁻¹)	1.368	1.332
$F(000)$	684	1400
No. of rflns. collected	31858	63745
No. of indep. rflns. / R_{int}	4965 / 0.0203	5127 / 0.0410
No. of obsd. rflns. [$I_0 > 2\sigma(I_0)$]	4763	4733
Data / restraints / parameters	4965 / 0 / 334	5127 / 1249 / 343
R_1 / wR_2 [$I_0 > 2\sigma(I_0)$]	0.0330 / 0.0959	0.0536 / 0.1375
R_1 / wR_2 (all data)	0.0344 / 0.0968	0.0578 / 0.1401
GOF (on F^2)	1.040	1.026
Largest diff. peak and hole (e Å ⁻³)	0.618 / -0.593	1.369 / -1.552
CCDC No.	1904423	1904417

Table S3. Crystallographic data for **5a[PF₆]** and **5b[PF₆]**

	5a[PF₆]	5b[PF₆]
Formula	C ₂₆ H ₄₃ Co ₂ F ₆ O ₃ PS ₂	C ₂₇ H ₄₅ Co ₂ F ₆ O ₃ PS ₂
Formula weight	730.55	744.58
Crystal dimensions (mm ³)	0.28 × 0.24 × 0.20	0.29 × 0.27 × 0.22
Crystal system	Monoclinic	Monoclinic
Space group	P 21/c	P 21/c
a (Å)	8.5889(4)	8.6079(6)
b (Å)	14.0677(7)	14.3633(10)
c (Å)	25.9890(12)	25.9474(19)
α (°)	90.00	90.00
β (°)	98.0250(10)	96.4298(14)
γ (°)	90.00	90.00
Volume (Å ³)	3109.4(3)	3187.9(4)
Z	4	4
T (K)	243(2)	190(2)
D _{calcd} (g cm ⁻³)	1.561	1.551
μ (mm ⁻¹)	1.316	1.285
F (000)	1512	1544
No. of rflns. collected	31638	31471
No. of indep. rflns. /R _{int}	5478 / 0.1048	5577 / 0.0240
No. of obsd. rflns. [I ₀ > 2σ(I ₀)]	3941	5321
Data / restraints / parameters	5478 / 0 / 361	5577 / 0 / 370
R ₁ / wR ₂ [I ₀ > 2σ(I ₀)]	0.0623 / 0.1328	0.0417 / 0.1128
R ₁ / wR ₂ (all data)	0.0971 / 0.1441	0.0434 / 0.1137
GOF (on F ²)	1.043	1.063
Largest diff. peak and hole (e Å ⁻³)	0.782 / -0.392	0.609 / -0.448
CCDC No.	1904418	1904425

Table S4. Crystallographic data for **6a[PF₆]**, **6b[PF₆]** and **6c[PF₆]**

	6a[PF₆]	6b[PF₆]	6c[PF₆]
Formula	C ₂₉ H ₅₀ Co ₂ F ₆ NO ₂ PS ₂	C ₂₈ H ₄₇ Co ₂ F ₆ NO ₂ PS ₂	C ₂₉ H ₅₀ Co ₂ F ₆ NO ₂ PS ₂
Formula weight	771.65	756.61	771.65
Crystal dimensions (mm ³)	0.32 × 0.25 × 0.21	0.32 × 0.28 × 0.26	0.33 × 0.27 × 0.23
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	P 21/c	P 21/n	C 2/c
a (Å)	16.2996(10)	15.3556(9)	20.2675(9)
b (Å)	8.8769(5)	11.3607(6)	11.2385(5)
c (Å)	24.0942(14)	20.2644(11)	15.4741(7)
α (°)	90.00	90.00	90.00
β (°)	94.1780(10)	108.5488(10)	104.8430(10)
γ (°)	90.00	90.00	90.00
Volume (Å ³)	3476.9(4)	3351.5(3)	3407.0(3)
Z	4	4	4
T (K)	173(2)	173(2)	173(2)
D _{calcd} (g cm ⁻³)	1.474	1.499	1.504
μ (mm ⁻¹)	1.179	1.222	1.204
F (000)	1608	1572	1608
No. of rflns. collected	32446	43759	21048
No. of indep. rflns. /R _{int}	6039/ 0.0178	5864 / 0.0246	2996 / 0.0245
No. of obsd. rflns. [I ₀ > 2σ(I ₀)]	5851	5469	2864
Data / restraints / parameters	6039 / 0 / 388	5864 / 0 / 379	2996 / 0 / 197
R ₁ / wR ₂ [I ₀ > 2σ(I ₀)]	0.0450 / 0.1228	0.0379 / 0.1083	0.0414 / 0.1195
R ₁ / wR ₂ (all data)	0.0460 / 0.1235	0.0405 / 0.1102	0.0431 / 0.1210
GOF (on F ²)	1.057	0.988	1.035
Largest diff. peak and hole (e Å ⁻³)	1.139 / -0.689	0.649 / -0.578	1.087 / -0.585
CCDC No.	1904421	1904420	1904419

Fig. S1. ORTEP diagram of **1**

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms are omitted for clarity.

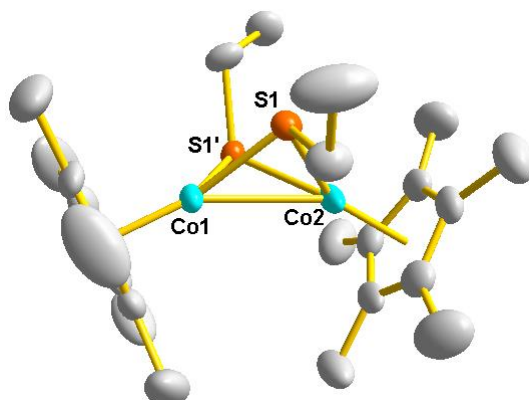


Table S5. Selected bond distances (Å) and angles (°) for **1**

Distances (Å)			
Co1–Co2	2.5472(10)		
Co1–S1	2.140(2)	Co1–S1'	2.303(2)
Co2–S1	2.090(2)	Co2–S1'	2.364(2)
Co1–Cp*1	1.7117(8)	Co2–Cp*2	1.6913(7)
Angles (°)			
Co1–S1–Co2	74.04(7)	Co1–S1'–Co2	66.14(6)
Torsion angles (°)			
S1–Co1Co2–S1'	109.91(10)	S1'–Co1Co2–S1	109.91(10)
Dihedral angles (°)			
Cp*1–Cp*2	54.04(18)		

Fig. S2. ORTEP diagram of **2**

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms are omitted for clarity.

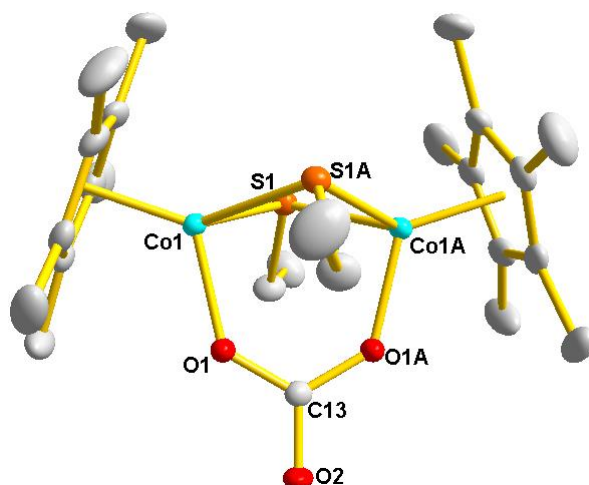


Table S6. Selected bond distances (Å) and angles (°) for **2**

Distances (Å)			
Co1···Co1A	3.1535(1)		
Co1–S1	2.2293(7)	Co1–O1	1.9167(17)
O1–C13	1.311(2)	C13–O2	1.243(4)
Co1–Cp*1	1.6954(3)		
Angles (°)			
Co1–S1–Co1A	89.83(2)	Co1–O1–C13	132.50(17)
O1–C13–O2	119.44(15)	O1–C13–O1A	121.1(3)
Torsion angles (°)			
S1–Co1Co2–S1A	142.044(35)		
Dihedral angles (°)			
Cp*1–Cp*2	37.95(12)		

Fig. S3. ORTEP diagram of **3[BF₄]**

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms on carbons and counter anion BF₄[−] are omitted for clarity.

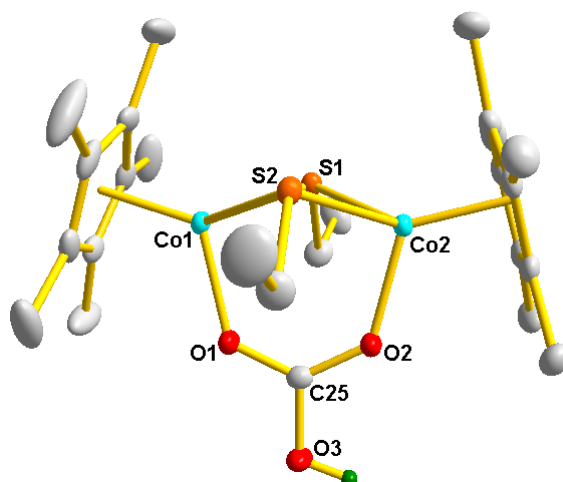


Table S7. Selected bond distances (Å) and angles (°) for **3[BF₄]**

Distances (Å)			
Co1···Co2	3.1966(5)	Co1–S1	2.2530(7)
Co1–S2	2.2543(7)	Co2–S1	2.2521(7)
Co2–S2	2.2497(7)	Co1–O1	1.9624(18)
Co2–O2	1.9448(17)	O1–C25	1.257(3)
O2–C25	1.254(3)	O3–C25	1.340(3)
Co1–Cp*1	1.6873(4)	Co2–Cp*2	1.6831(3)
Angles (°)			
Co1–S1–Co2	90.40(2)	Co1–S2–Co2	90.43(2)
Co1–O1–C25	128.55(16)	Co2–O2–C25	130.69(16)
O1–C25–O2	128.4(2)	O1–C25–O3	115.0(2)
O2–C25–O3	116.5(2)		
Torsion angles (°)			
S1–Co1Co2–S2	137.544(34)		
Dihedral angles (°)			
Cp*1–Cp*2	33.213(97)		

Fig. S4. ORTEP diagram of **4[BF₄]**

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms and counter anion BF₄[−] are omitted for clarity.

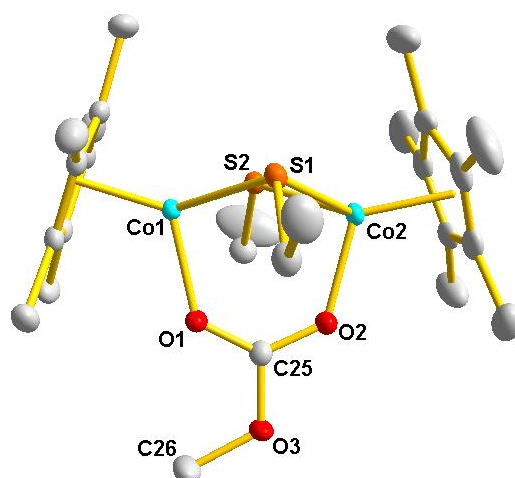


Table S8. Selected bond distances (Å) and angles (°) for **4[BF₄]**

Distances (Å)			
Co1···Co2	3.1942(8)		
Co1–S1	2.2499(11)	Co1–S2	2.2419(11)
Co2–S1	2.2564(11)	Co2–S2	2.2463(12)
Co1–O1	1.956(3)	Co2–O2	1.958(3)
O1–C25	1.253(5)	O2–C25	1.250(5)
O3–C25	1.342(5)	O3–C26	1.453(6)
Co1–Cp*1	1.6807(6)	Co2–Cp*2	1.7147(5)
Angles (°)			
Co1–S1–Co2	90.28(4)	Co1–S2–Co2	90.75(4)
Co2–O2–C25	129.1(3)	Co1–O1–C25	129.7(3)
O1–C25–O2	128.8(4)	O2–C25–O3	113.6(4)
O1–C25–O3	117.6(4)	C25–O3–C26	117.5(3)
Torsion angles (°)			
S3–Co1Co2–S4	138.738(59)		
Dihedral angles (°)			
Cp*1–Cp*2	35.08(20)		

Fig. S5. ORTEP diagram of **5a**[PF₆]

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms and counter anion PF₆[−] are omitted for clarity.

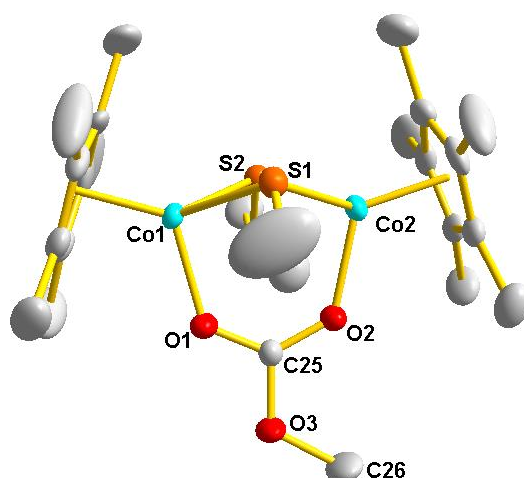


Table S9. Selected bond distances (Å) and angles (°) for **5a**[PF₆]

Distances (Å)			
Co1···Co2	3.1814(10)		
Co1–S1	2.2410(15)	Co1–S2	2.2535(15)
Co2–S1	2.2428(15)	Co2–S2	2.2443(15)
Co1–O1	1.972(3)	Co2–O2	1.966(3)
O1–C25	1.258(6)	O2–C25	1.247(6)
O3–C25	1.338(6)	O3–C26	1.449(7)
Co1–Cp*1	1.6932(7)	Co2–Cp*2	1.6819(7)
Angles (°)			
Co1–S1–Co2	90.38(5)	Co1–S2–Co2	90.02(5)
Co2–O2–C25	129.7(3)	Co1–O1–C25	128.6(3)
O1–C25–O2	128.8(5)	O2–C25–O3	118.5(4)
O1–C25–O3	112.7(4)	C25–O3–C26	117.4(4)
Torsion angles (°)			
S1–Co1Co2–S2	139.143(75)		
Dihedral angles (°)			
Cp*1–Cp*2	35.64(20)		

Fig. S6. ORTEP diagram of **5b**[PF₆]

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms and counter anion PF₆[−] are omitted for clarity.

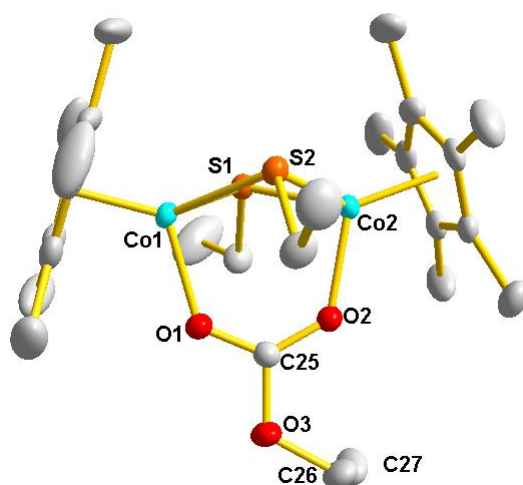


Table S10. Selected bond distances (Å) and angles (°) for **5b**[PF₆]

Distances (Å)			
Co1···Co2	3.1851(7)	Co1–S1	2.2517(9)
Co1–S2	2.2419(9)	Co2–S1	2.2611(9)
Co2–S2	2.2459(9)	Co1–O1	1.969(2)
Co2–O2	1.957(2)	O1–C25	1.257(4)
O2–C25	1.258(4)	O3–C25	1.341(4)
O3–C26	1.458(4)	C26–C27	1.490(6)
Co1–Cp*1	1.6866(4)	Co2–Cp*2	1.6875(5)
Angles (°)			
Co1–S1–Co2	89.78(3)	Co1–S2–Co2	90.42(3)
Co1–O1–C25	129.1(2)	Co2–O2–C25	129.4(2)
O1–C25–O2	128.5(3)	O1–C25–O3	119.1(3)
O2–C25–O3	112.3(3)	C25–O3–C26	118.4(3)
Torsion angles (°)			
S3–Co1Co2–S4	139.869(45)		
Dihedral angles (°)			
Cp*1–Cp*2	36.42(12)		

Fig. S7. ORTEP diagram of **6a**[PF₆]

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms on carbons and counter anion PF₆[−] are omitted for clarity.

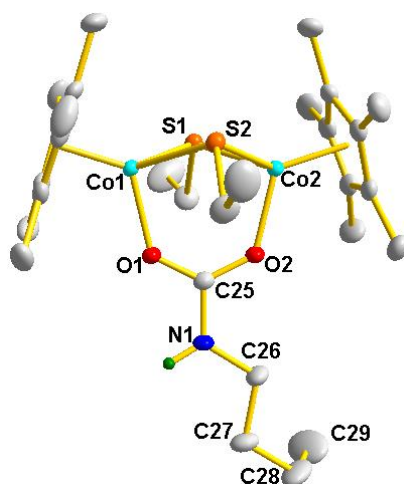


Table S11. Selected bond distances (Å) and angles (°) for **6a**[PF₆]

Distances (Å)			
Co1···Co2	3.1725(6)	Co1–S1	2.2424(8)
Co1–S2	2.2470(8)	Co2–S1	2.2458(8)
Co2–S2	2.2476(8)	Co1–O1	1.940(2)
Co2–O2	1.946(2)	O1–C25	1.272(4)
O2–C25	1.266(4)	N1–C25	1.356(4)
N1–C26	1.456(4)	C26–C27	1.515(5)
Co1–Cp*1	1.6871(4)	Co2–Cp*2	1.6835(5)
Angles (°)			
Co1–S1–Co2	89.96(3)	Co1–S2–Co2	89.80(3)
Co1–O1–C25	129.8(2)	Co2–O2–C25	129.6(2)
O1–C25–O2	127.3(3)	O1–C25–N1	116.3(3)
O2–C25–N1	116.4(3)	C25–N1–C26	122.0(3)
Torsion angles (°)			
S1–Co1Co2–S2	139.144(40)		
Dihedral angles (°)			
Cp*1–Cp*2	36.614(115)		

Fig. S8. ORTEP diagram of **6b**[PF₆]

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms on carbons and counter anion PF₆[−] are omitted for clarity.

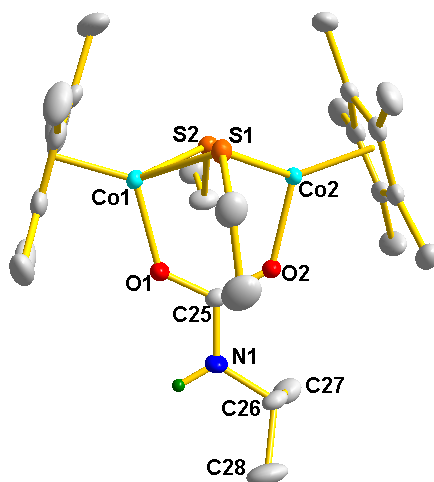


Table S12. Selected bond distances (Å) and angles (°) for **6b**[PF₆]

Distances (Å)			
Co1···Co2	3.1939(5)	Co1–S1	2.2264(7)
Co1–S2	2.2602(7)	Co2–S1	2.2327(7)
Co2–S2	2.2576(7)	Co1–O1	1.9417(19)
Co2–O2	1.9460(18)	O1–C25	1.274(3)
O2–C25	1.270(3)	N1–C25	1.356(3)
N1–C26	1.461(4)	C26–C27	1.518(4)
Co1–Cp*1	1.6853(3)	Co2–Cp*2	1.6875(3)
Angles (°)			
Co1–S1–Co2	91.49(3)	Co1–S2–Co2	89.97(3)
Co1–O1–C25	130.23(17)	Co2–O2–C25	129.75(16)
O1–C25–O2	127.1(2)	O1–C25–N1	115.7(2)
O2–C25–N1	117.2(2)	C25–N1–C26	123.2(2)
Torsion angles (°)			
S1–Co1Co2–S2	135.73(4)		
Dihedral angles (°)			
Cp*1–Cp*2	38.1(1)		

Fig. S9. ORTEP diagram of **6c[PF₆]**

The thermal ellipsoids are shown at 50% probability level. Hydrogen atoms on carbons and counter anion PF₆[−] are omitted for clarity.

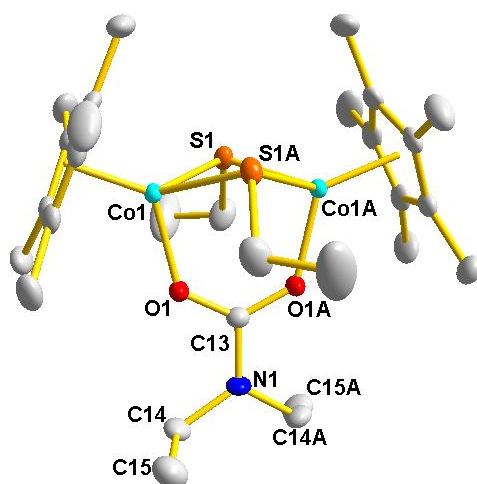


Table S13. Selected bond distances (Å) and angles (°) for **6c[PF₆]**

Distances (Å)			
Co1···Co1A	3.1842(1)		
Co1–S1	2.2397(8)	Co1–O1	1.939(2)
O1–C13	1.274(3)	C13–N1	1.354(5)
N1–C14	1.447(4)	C14–C15	1.510(5)
Co1–Cp*1	1.6956(1)		
Angles (°)			
Co1–S1–Co1A	90.50(3)	Co1–O1–C13	130.0(2)
O1–C13–N1	116.54(19)	C13–N1–C14	120.99(18)
O1–C13–O1A	126.9(4)	C14–N1–C14A	118.0(4)
Torsion angles (°)			
S1–Co1Co1A–S1A	140.379(2)		
Dihedral angles (°)			
Cp*1–Cp*2	44.234		

II. NMR Spectra

Fig. S10. ^1H NMR spectrum of **1** in CDCl_3

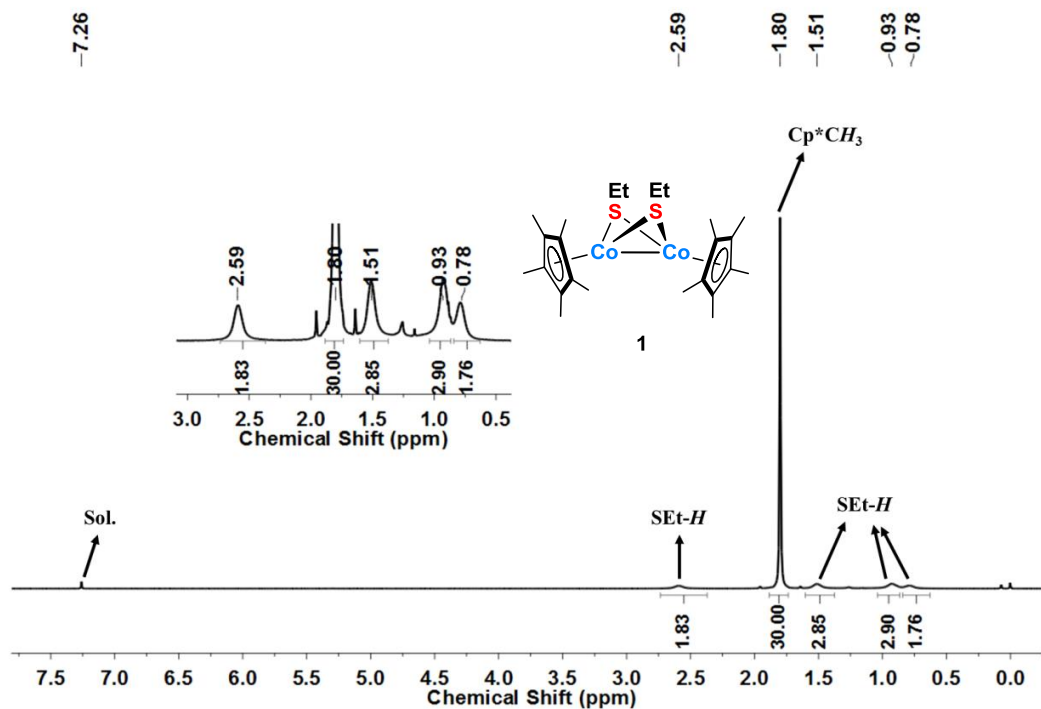


Fig. S11. ^{13}C NMR spectrum of **1** in CDCl_3

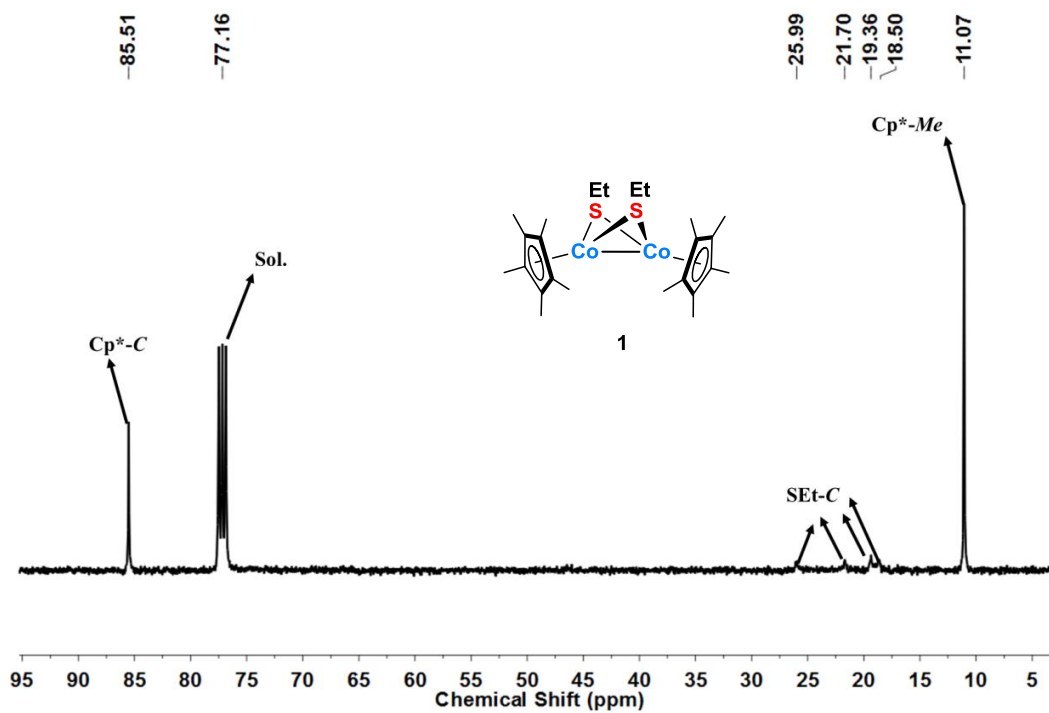


Fig. S12. ^1H NMR spectrum of **2** in CDCl_3

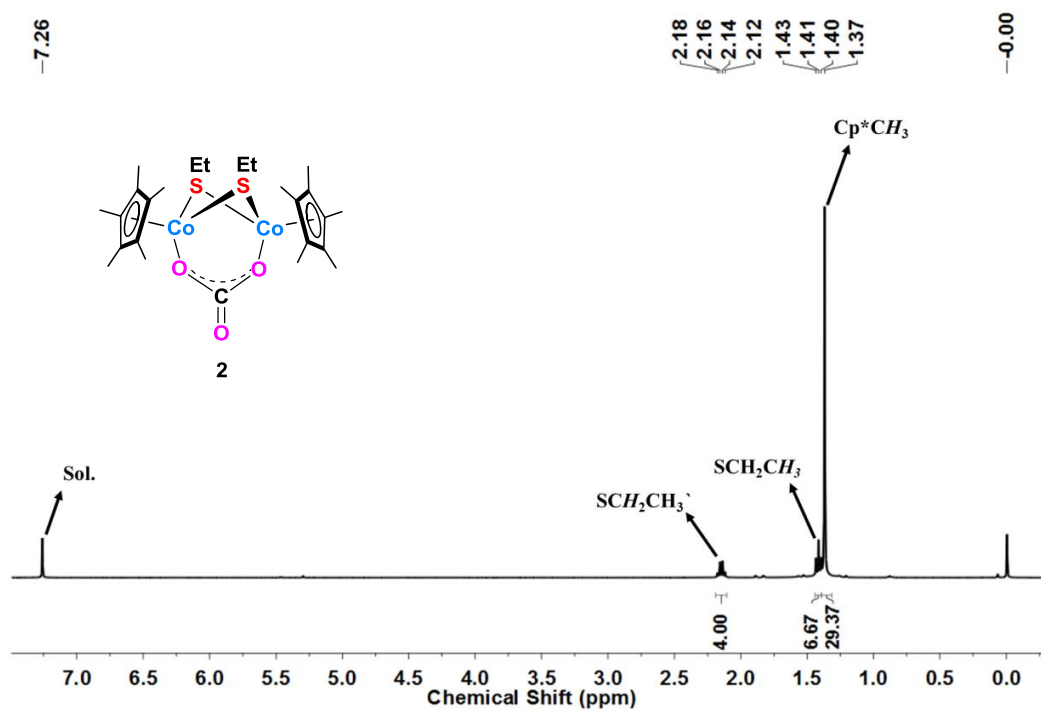


Fig. S13. ^{13}C NMR spectrum of **2** in CDCl_3

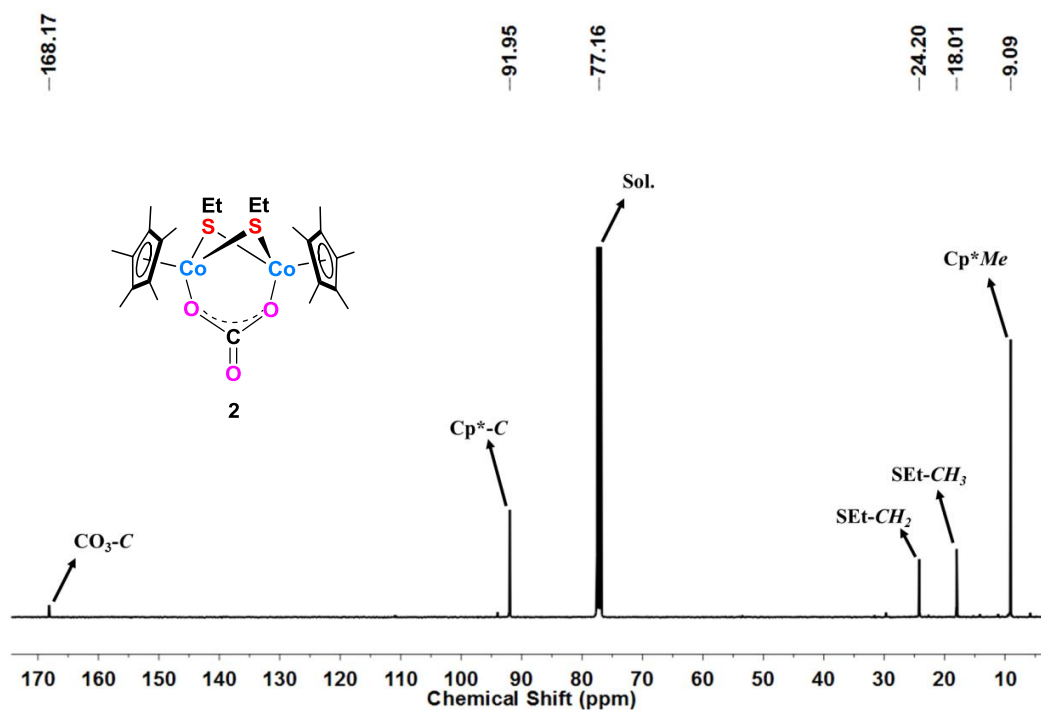


Fig. S14. ^1H NMR spectrum of **3**[BF₄] in CD₂Cl₂

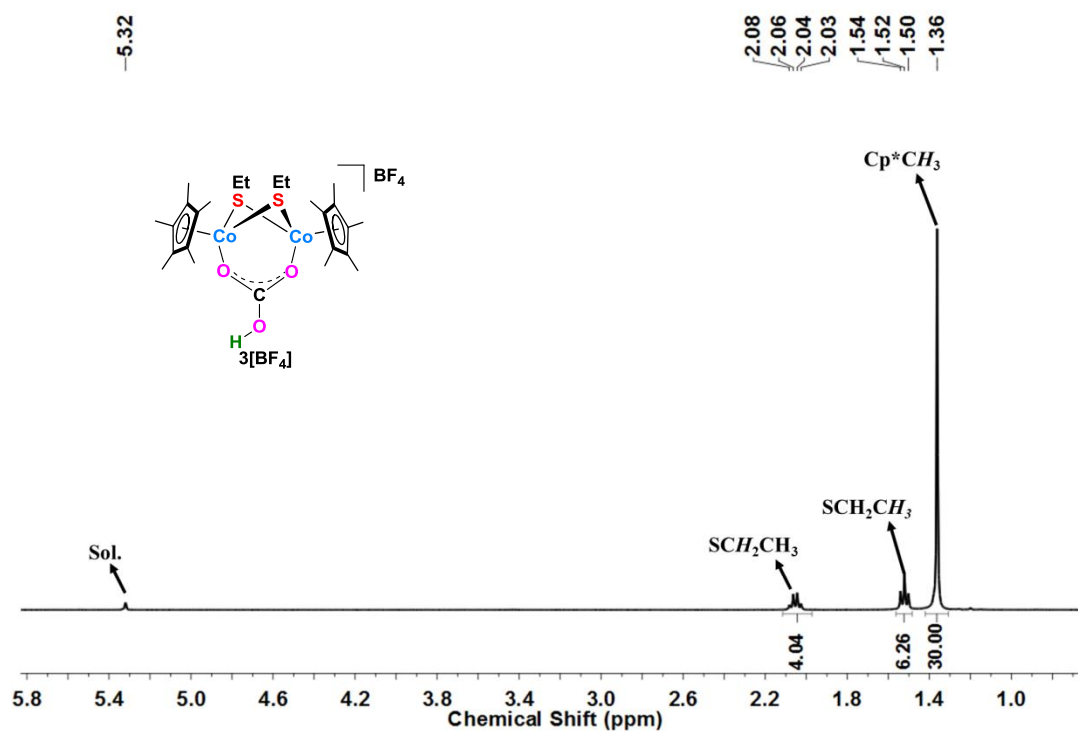


Fig. S15. ^{13}C NMR spectrum of **3**[BF₄] in CD₂Cl₂

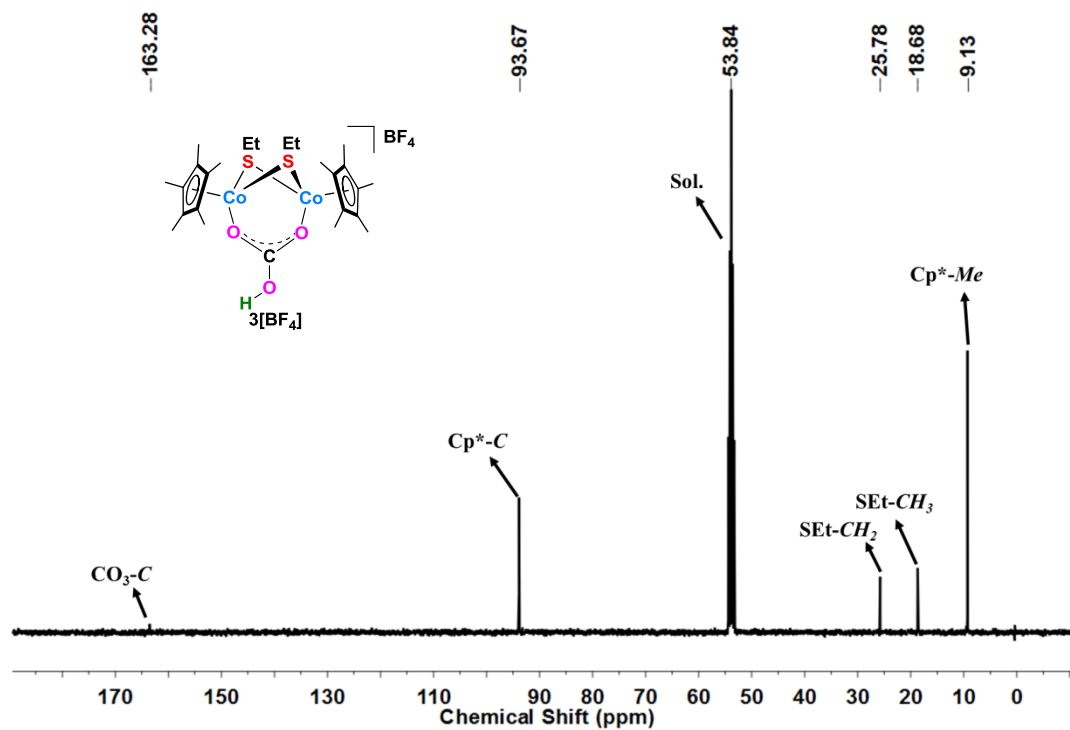


Fig. S16. ^{19}F NMR spectrum of **3**[BF_4] in CD_2Cl_2

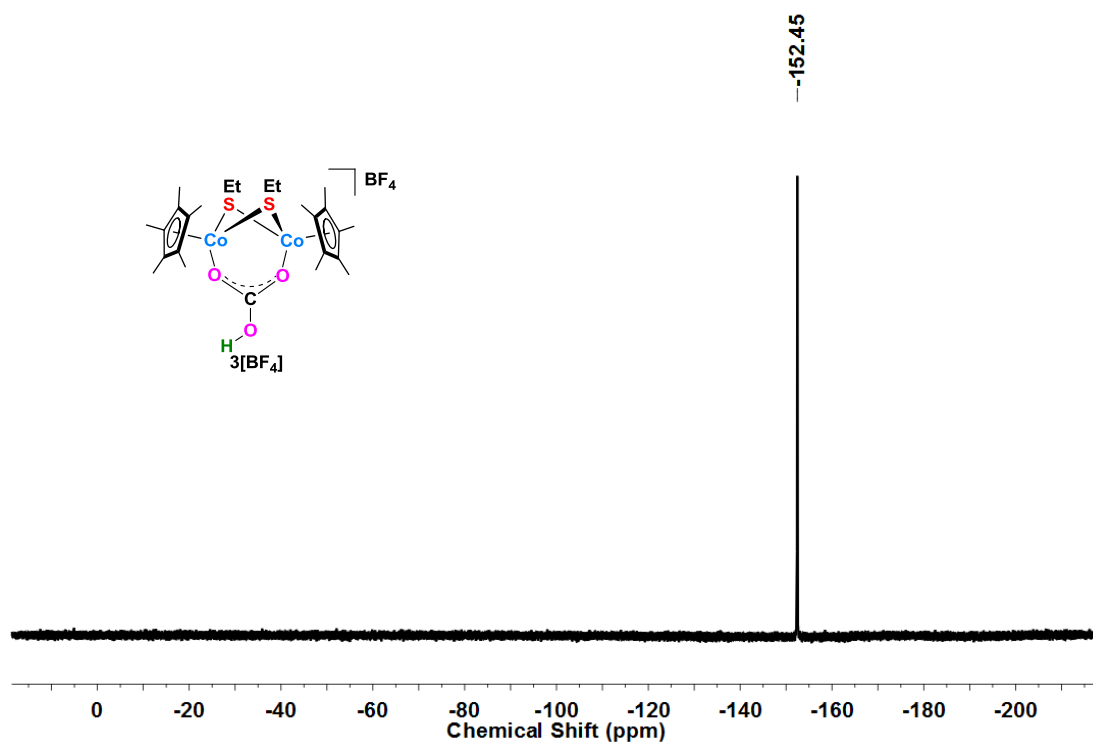


Fig. S17. ^1H NMR spectrum of **4**[BF_4] in CD_2Cl_2

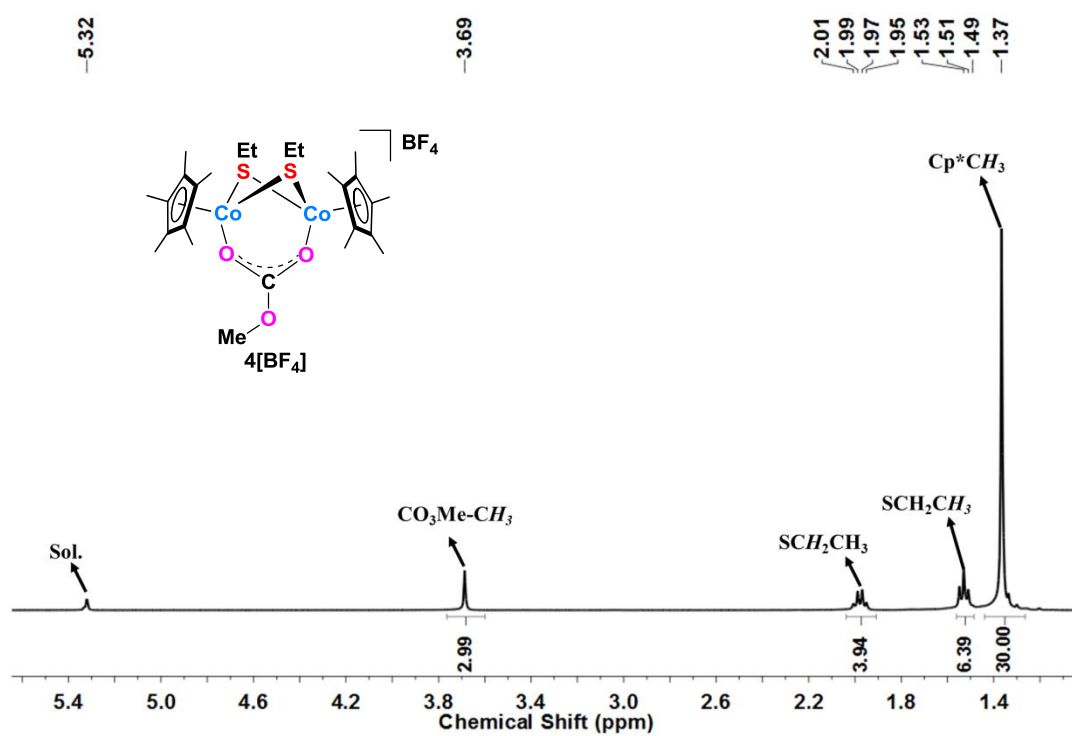


Fig. S18. ^{13}C NMR spectrum of **4**[BF_4] in CD_2Cl_2

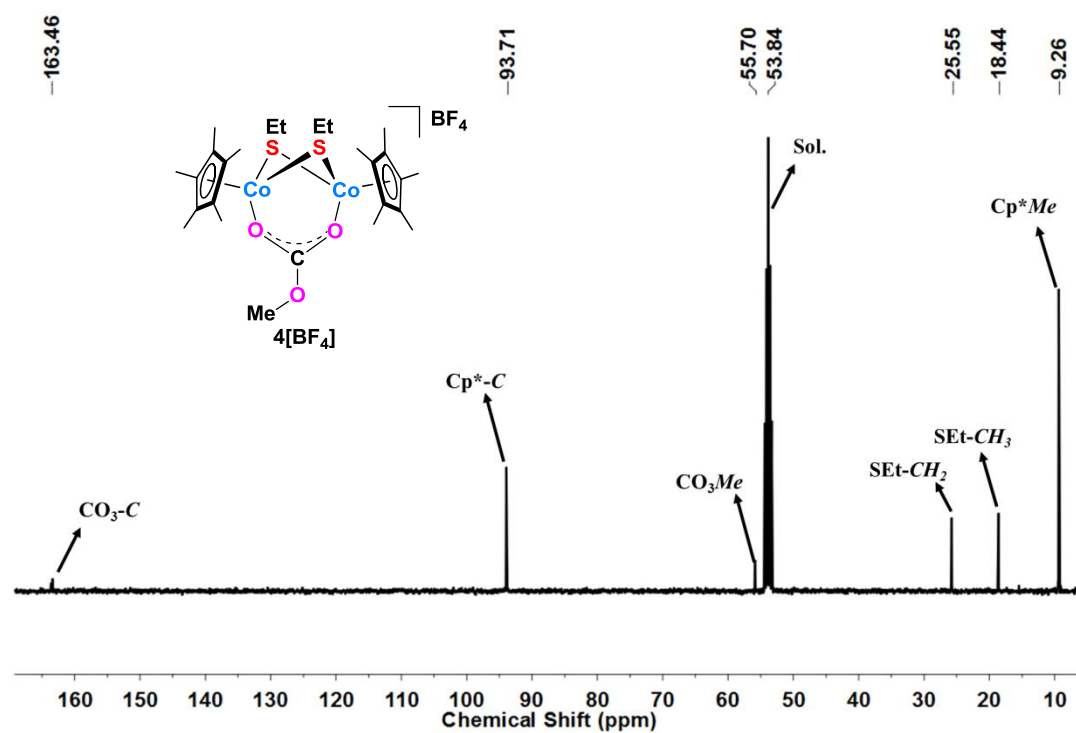


Fig. S19. ^{19}F NMR spectrum of **4**[BF_4] in CD_2Cl_2

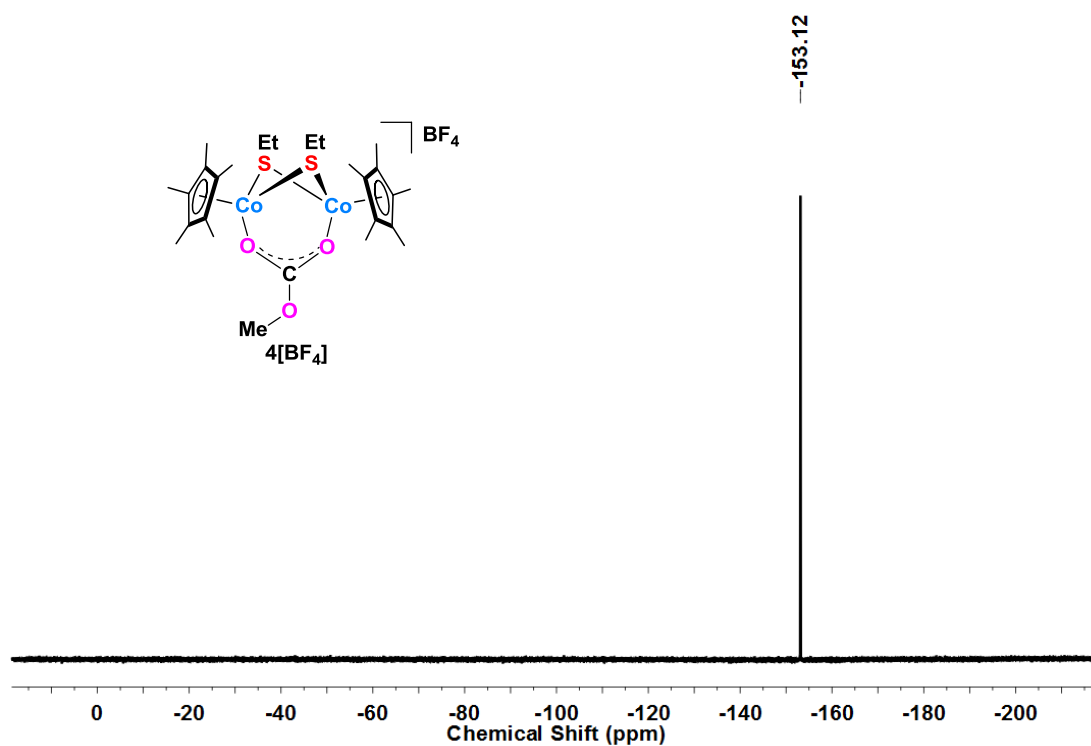


Fig. S20. ^1H NMR spectrum of **5a**[PF₆] in CD₂Cl₂

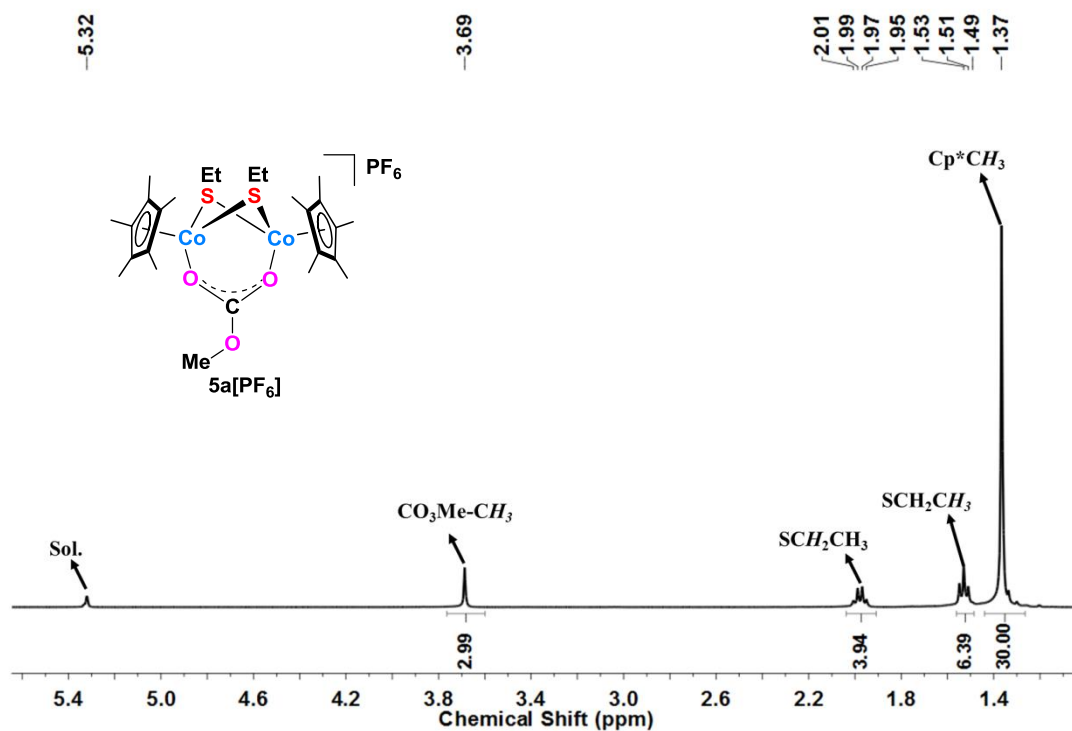


Fig. S21. ^{13}C NMR spectrum of **5a**[PF₆] in CD₂Cl₂

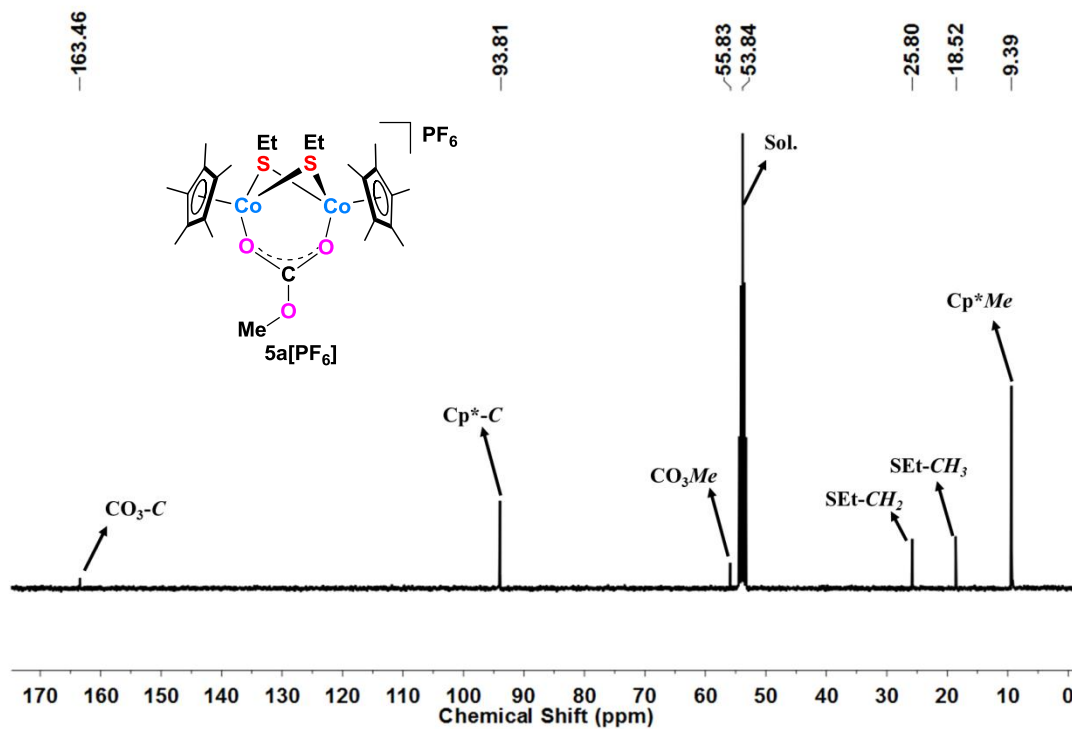


Fig. S22. ^{31}P NMR spectrum of **5a**[PF₆] in CD₂Cl₂

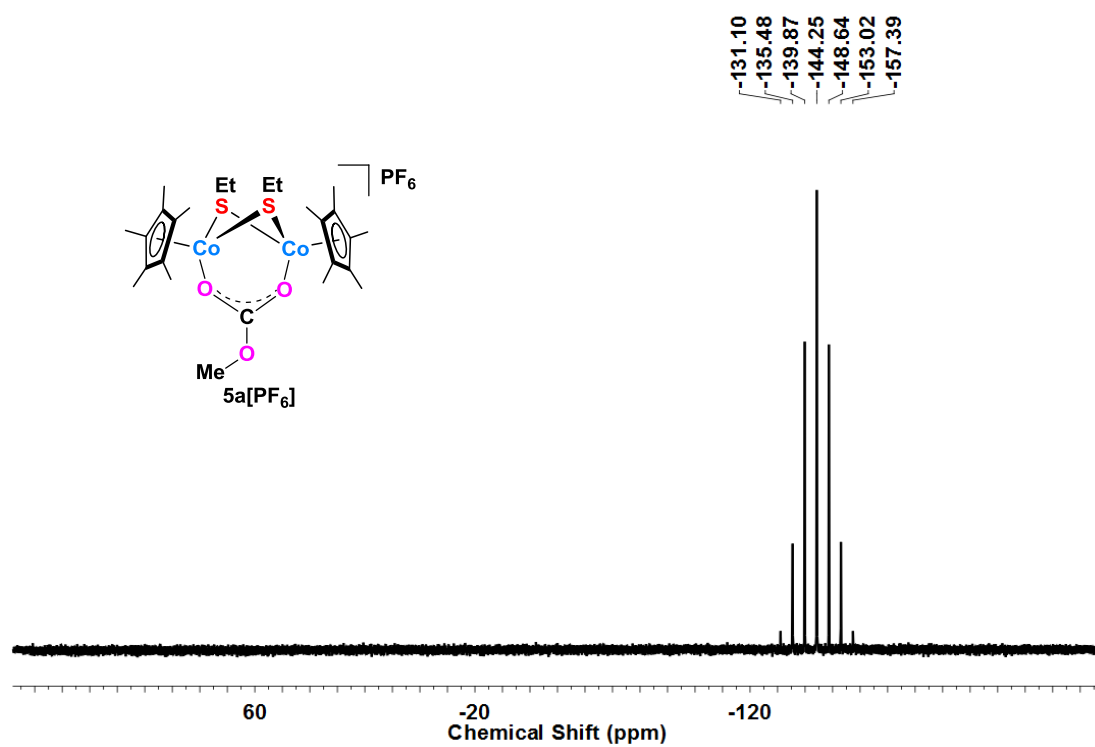
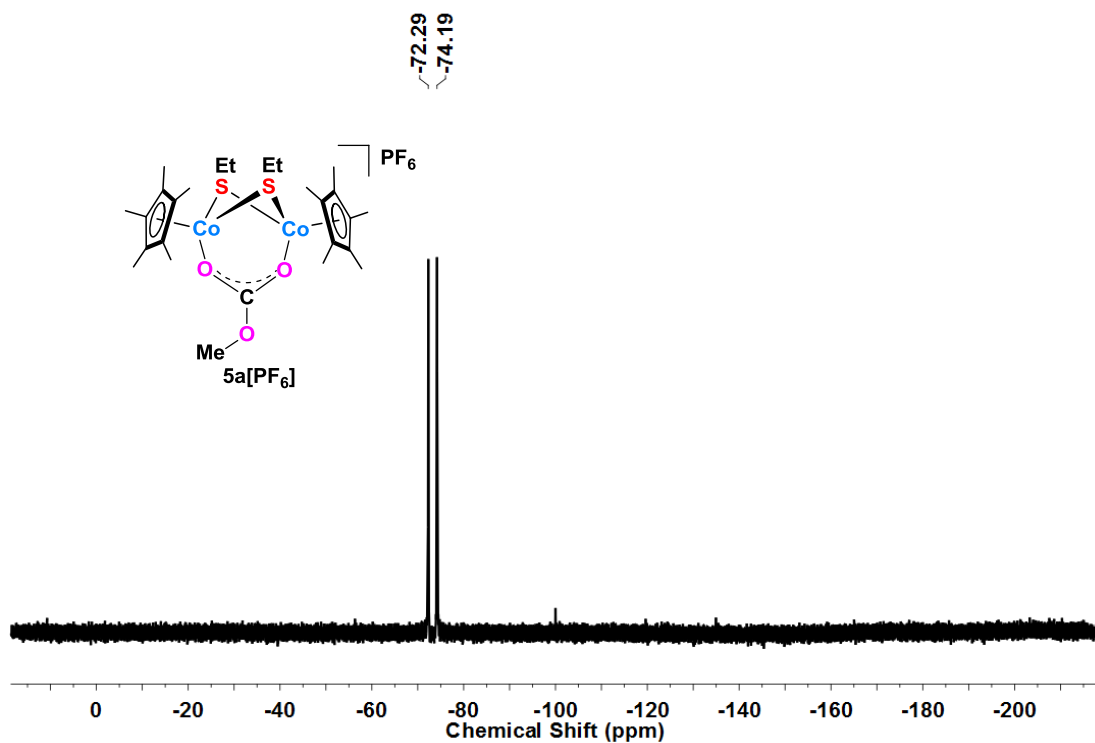


Fig. S23. ^{19}F NMR spectrum of **5a**[PF₆] in CD₂Cl₂



Chemical structure of **5b**[PF₆]:

The structure shows a cobalt complex with two cyclopentadienyl rings, two ethylthio groups (Et-S), and a central carbon atom bonded to two ethoxy groups (Et-O). The counterion is PF₆⁻.

¹H NMR spectrum (CDCl₃) showing chemical shifts (ppm) and integration values:

- Sol. (Solvent): ~7.26 ppm
- CO₃Et-CH₂: ~4.10 ppm (Integration: 2.01)
- SCH₂CH₃: ~2.00 ppm (Integration: 4.04)
- Cp*CH₃: ~1.36 ppm (Integration: 30.00)
- CO₃Et-CH₃: ~1.25 ppm (Integration: 3.26)

Integration values for the Cp*CH₃ peak are also listed at the top right: 2.00, 1.98, 1.96, 1.94, 1.55, 1.53, 1.51, 1.36, 1.25, 1.24, 1.22.

Chemical structure of **5b**[PF₆]:

5b[PF₆]

¹³C NMR spectrum (CDCl₃) of **5b**[PF₆]:

Chemical Shift (ppm)	Assignment
163.00	CO ₃ -C
93.55	Cp*-C
64.46	CO ₃ Et-CH ₂
53.84	Sol.
25.85	SEt-CH ₂
18.61	SEt-CH ₃
14.54	CO ₃ Et-CH ₃
9.27	Cp*Me

Fig. S26. ^{31}P NMR spectrum of **5b**[PF₆] in CD₂Cl₂

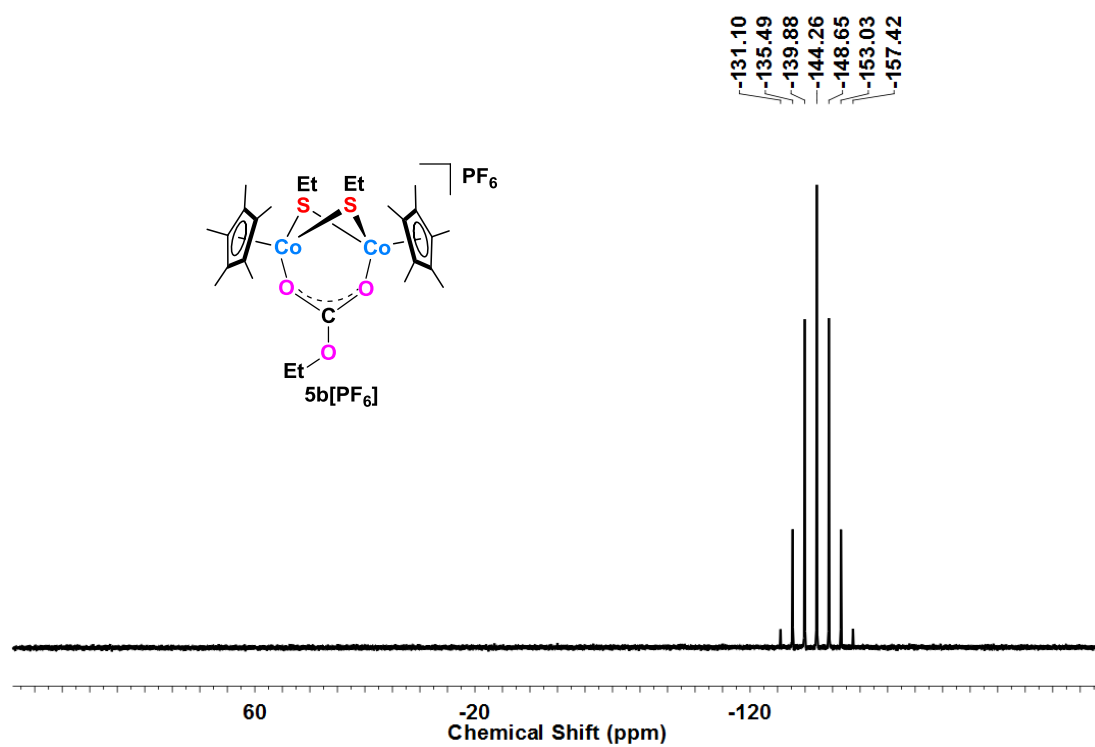
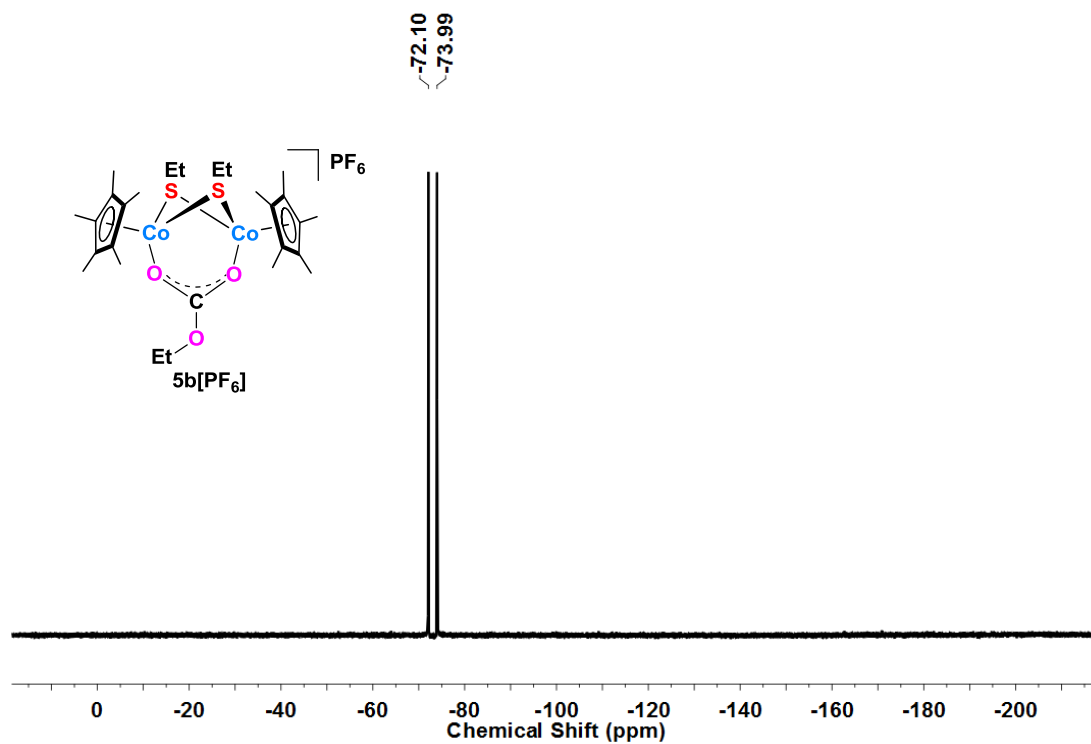


Fig. S27. ^{19}F NMR spectrum of **5b**[PF₆] in CD₂Cl₂



Chemical structure of **6a** is shown, featuring a cobalt dimer complexed with two ferrocene units, two ethyl groups, and a 6a[PF₆] ligand. The ¹H NMR spectrum (CDCl₃) displays peaks corresponding to the complex, with chemical shifts (ppm) labeled above the peaks: 5.32, 4.89, 3.17, 3.16, 3.14, 3.12, 1.90, 1.89, 1.88, 1.88, 1.86, 1.86, 1.85, 1.84, 1.42, 1.41, 1.37, 1.34, 0.94, 0.92, and 0.90. The spectrum shows peaks for the solvent (Sol.), NH (1.09 ppm), NⁿBu-CH₂ (2.09 ppm), SCH₂CH₃ (4.16 ppm), Cp*CH₃ (15.00 ppm), and NⁿBu-CH₃ (2.93 ppm). The inset shows the aromatic region with peaks at 3.17, 3.16, 3.14, 3.12, 1.90, 1.89, 1.88, 1.88, 1.86, 1.86, 1.85, 1.84, 1.42, 1.41, 1.37, 1.34, 0.94, 0.92, and 0.90 ppm.

Chemical structure of **6a** is shown in the inset. The structure features a cobalt dimer core with two Cp* rings, two S-Et groups, and a central C-NH group. The counterion is PF₆⁻.

The ¹³C NMR spectrum (CDCl₃) shows the following chemical shifts (ppm):

- 165.55 (NCO₂-C)
- 93.18 (Cp*-C)
- 53.59 (Sol.)
- 42.37 (N"Bu-C)
- 32.57 (SEt-CH₂)
- 25.28 (SEt-CH₃)
- 17.92 (Cp*Me)
- 13.56
- 9.01

Fig. S30. ^{31}P NMR spectrum of **6a**[PF₆] in CD₂Cl₂

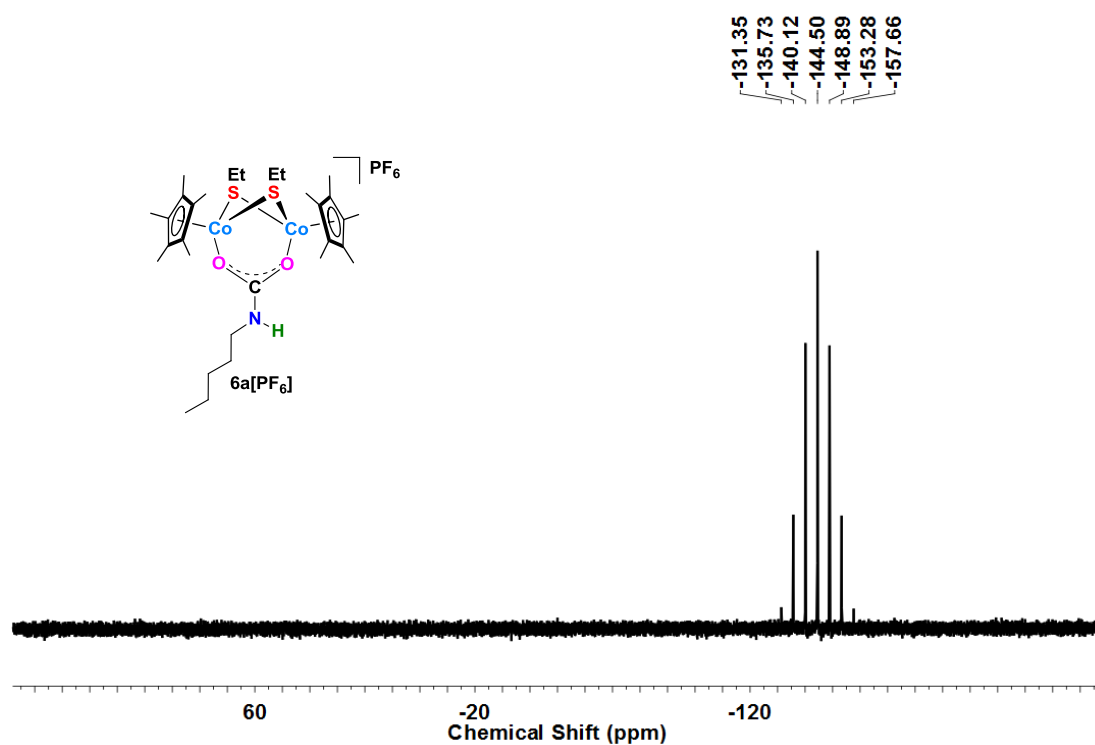


Fig. S31. ^{19}F NMR spectrum of **6a**[PF₆] in CD₂Cl₂

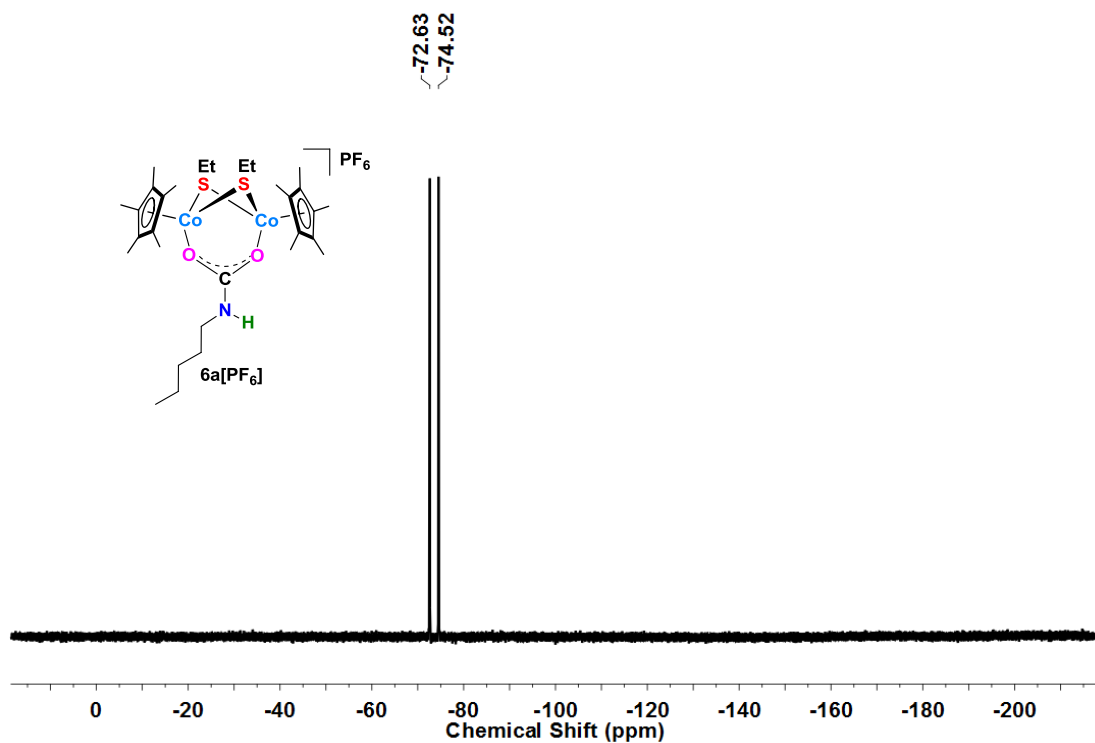


Fig. S32. ^1H NMR spectrum of **6b**[PF₆] in CD₂Cl₂

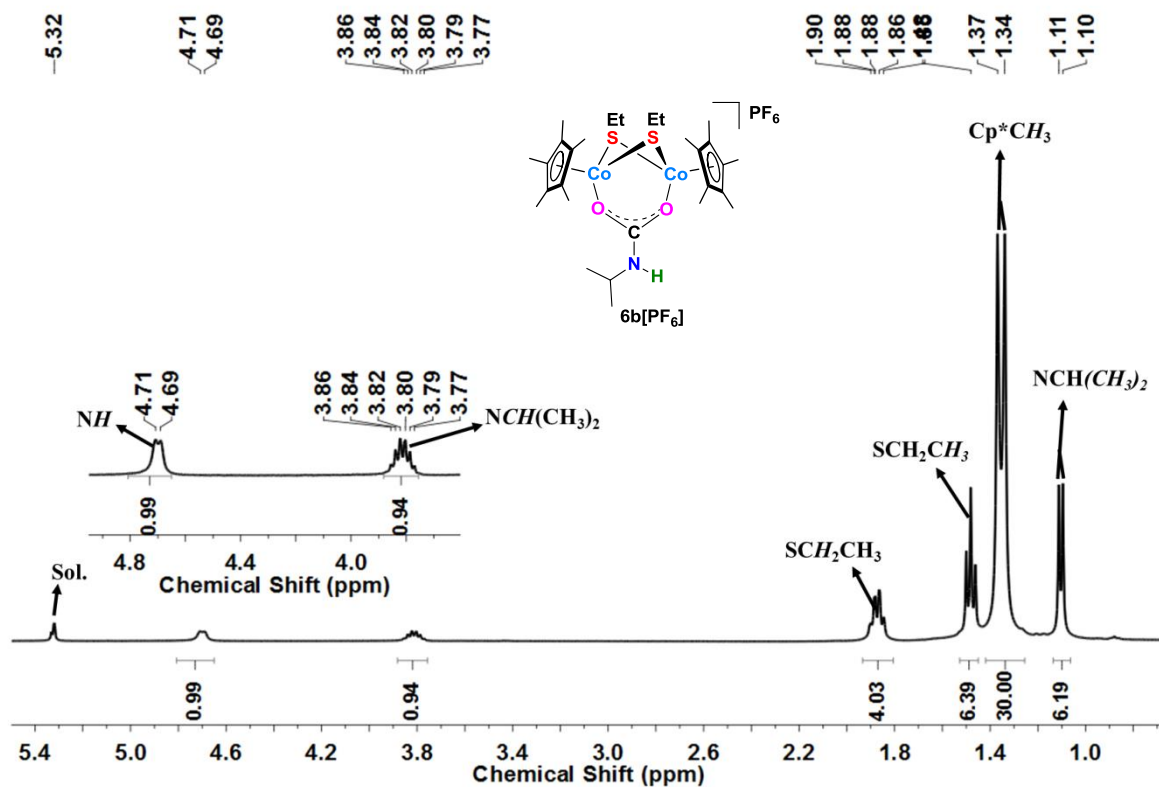


Fig. S33. ^{13}C NMR spectrum of **6b**[PF₆] in CD₂Cl₂

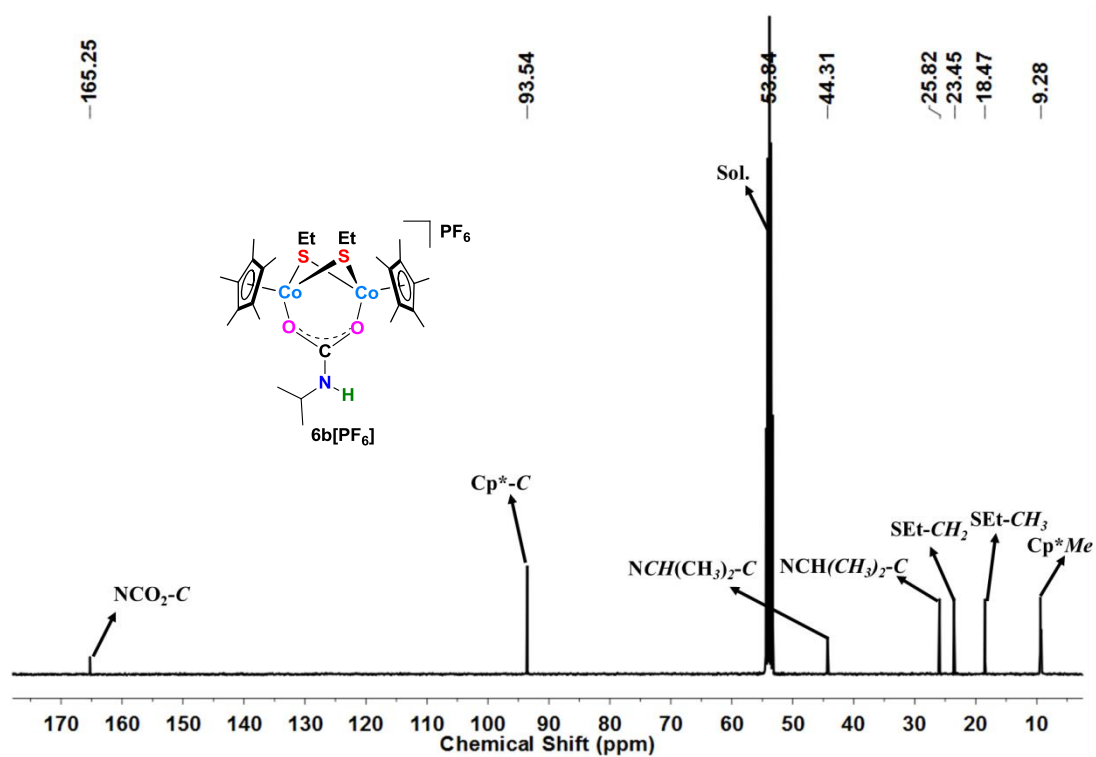


Fig. S34. ^{31}P NMR spectrum of **6b**[PF₆] in CD₂Cl₂

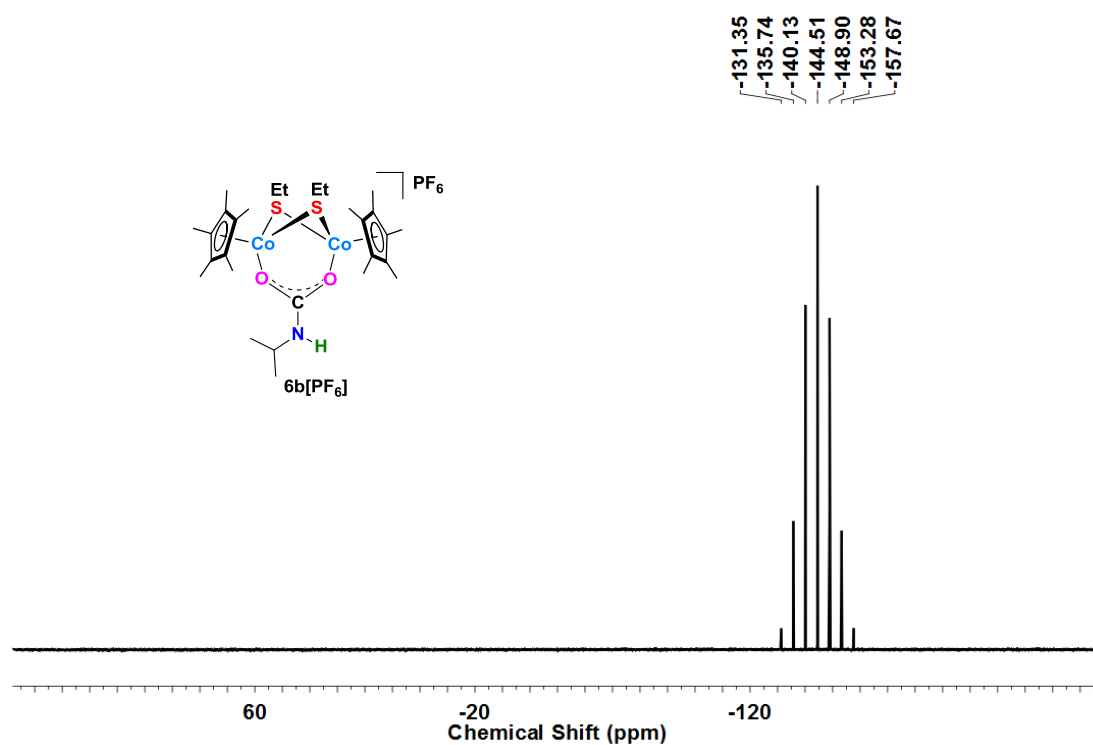


Fig. S35. ^{19}F NMR spectrum of **6b**[PF₆] in CD₂Cl₂

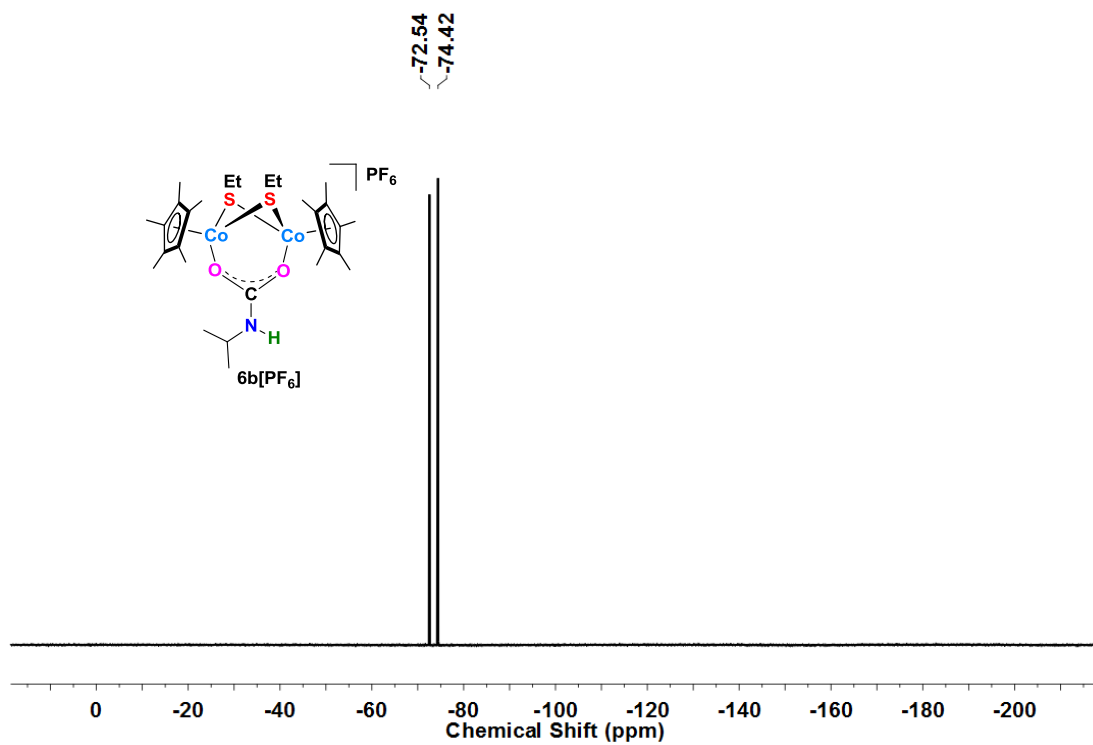


Fig. S36. ^1H NMR spectrum of **6c**[PF₆] in CD₂Cl₂

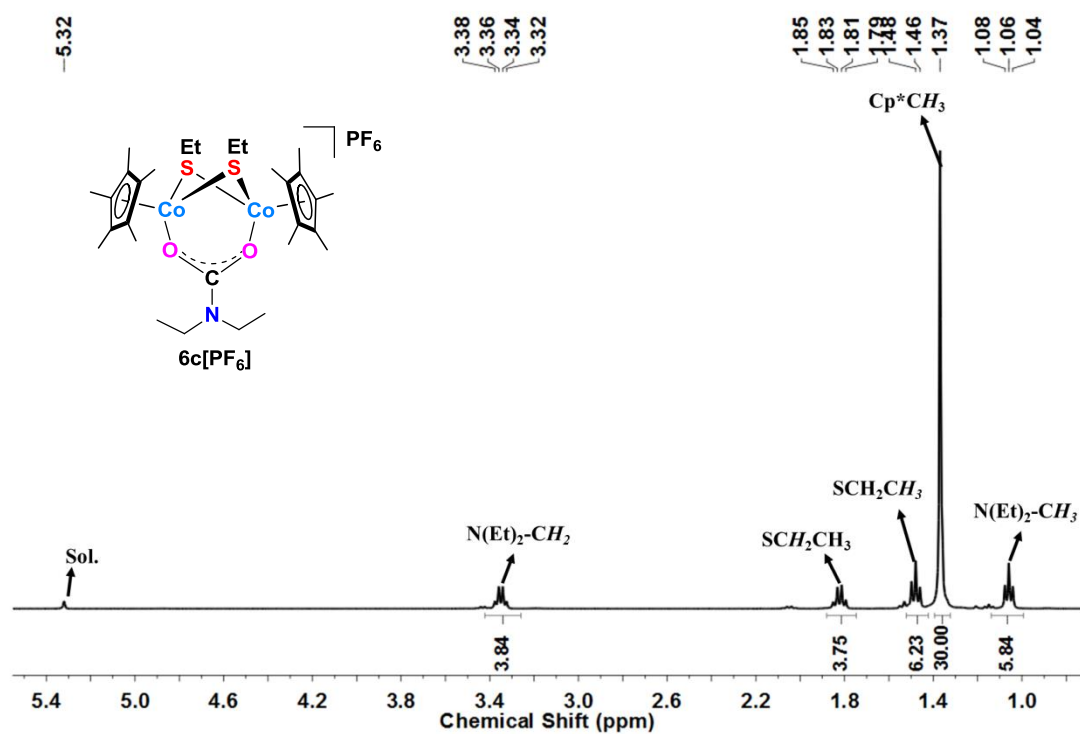
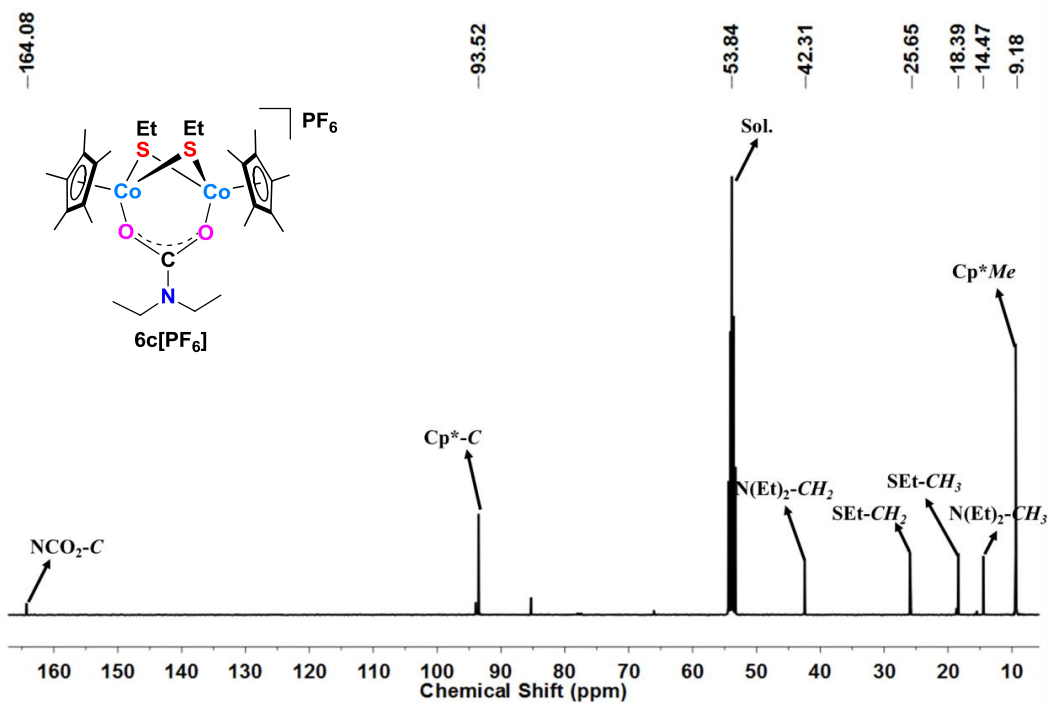


Fig. S37. ^{13}C NMR spectrum of **6c**[PF₆] in CD₂Cl₂



Chemical structure of **6c[PF₆]** is shown, featuring a cobalt complex with two phenyl rings, two ethyl groups, and a central carbon atom bonded to a nitrogen atom.

The ¹³C NMR spectrum (CDCl₃) displays several peaks in the aromatic region (131.25 to 157.58 ppm) and a cluster of peaks in the aliphatic region (140.03 to 148.81 ppm).

Chemical Shift (ppm):

- 131.25
- 135.64
- 140.03
- 144.42
- 148.81
- 153.19
- 157.58

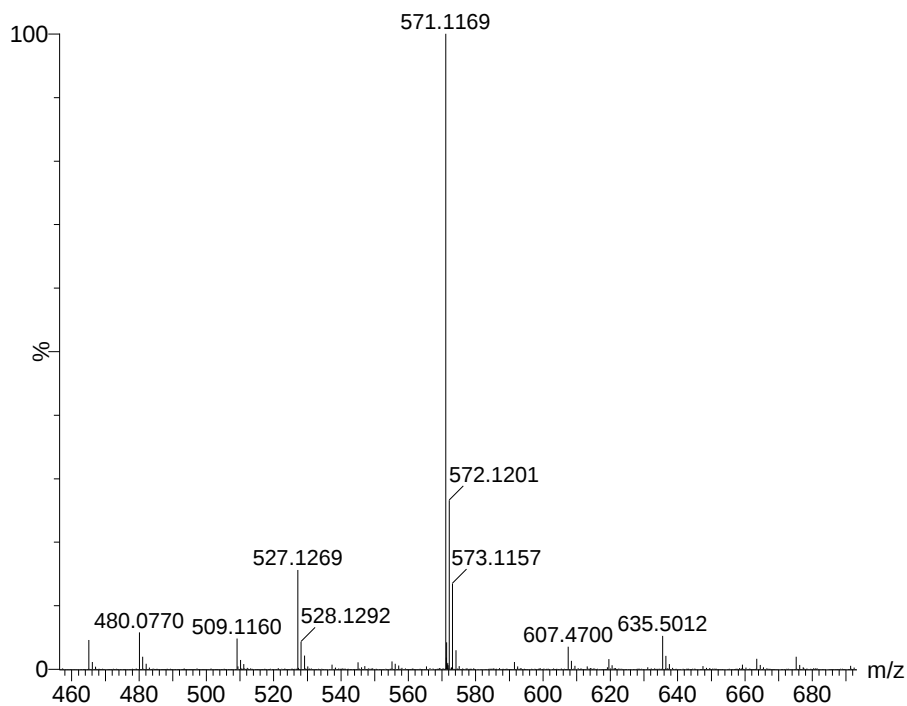
Chemical structure of the cobalt complex **6c** is shown, which is a dimeric complex consisting of two cobalt centers bridged by two ethylthio groups (Et-S) and two phenyl groups. The complex is shown as **6c**[PF₆].

III. ESI-HRMS Spectra

Fig. S40. ESI-HRMS of **3**[BF₄] in CH₂Cl₂

(a) The signal at an $m/z = 571.1169$ corresponds to **[3]**⁺. (b) Calculated isotopic distribution for **[3]**⁺ (upper) and the amplifying diagram for **[3]**⁺ (bottom).

(a)



(b)

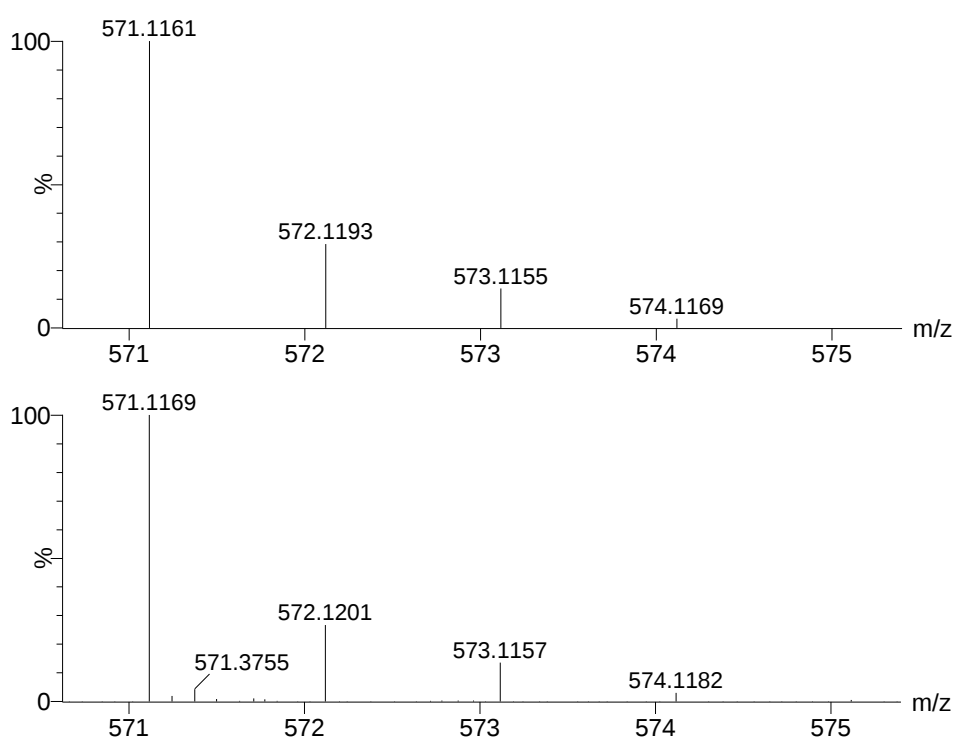
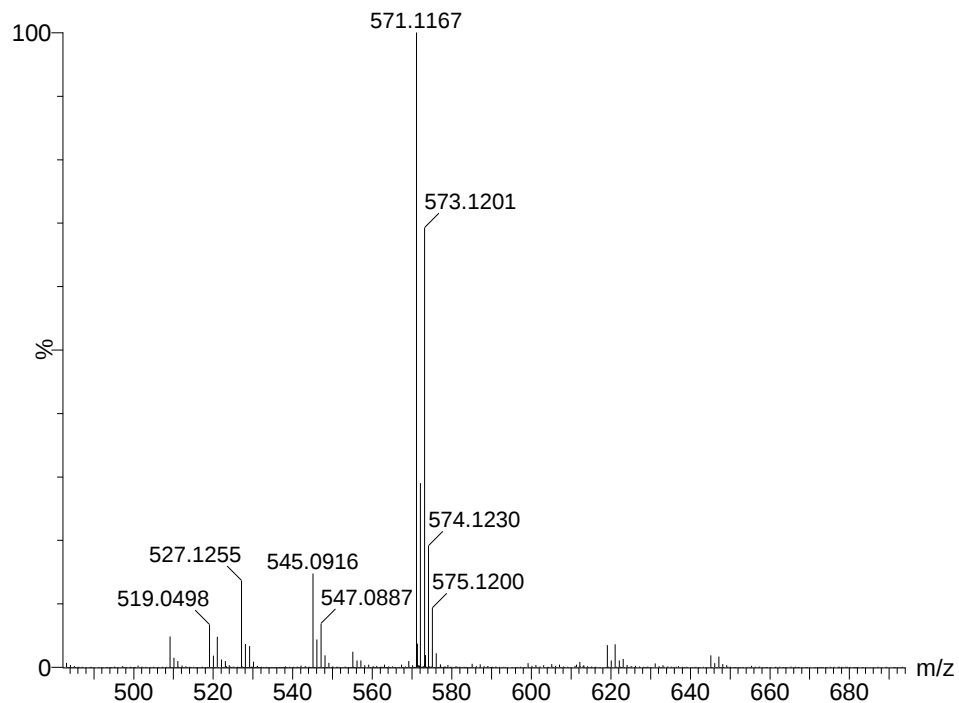


Fig. S41. ESI-HRMS of ^{18}O -**3**[**BF**₄] in CH_2Cl_2

(a) The signal at an $m/z = 573.1201$ corresponds to ^{18}O -[**3**]⁺. (b) Calculated isotopic distribution for ^{18}O -[**3**]⁺ (upper) and the amplifying diagram for ^{18}O -[**3**]⁺ (bottom).

(a)



(b)

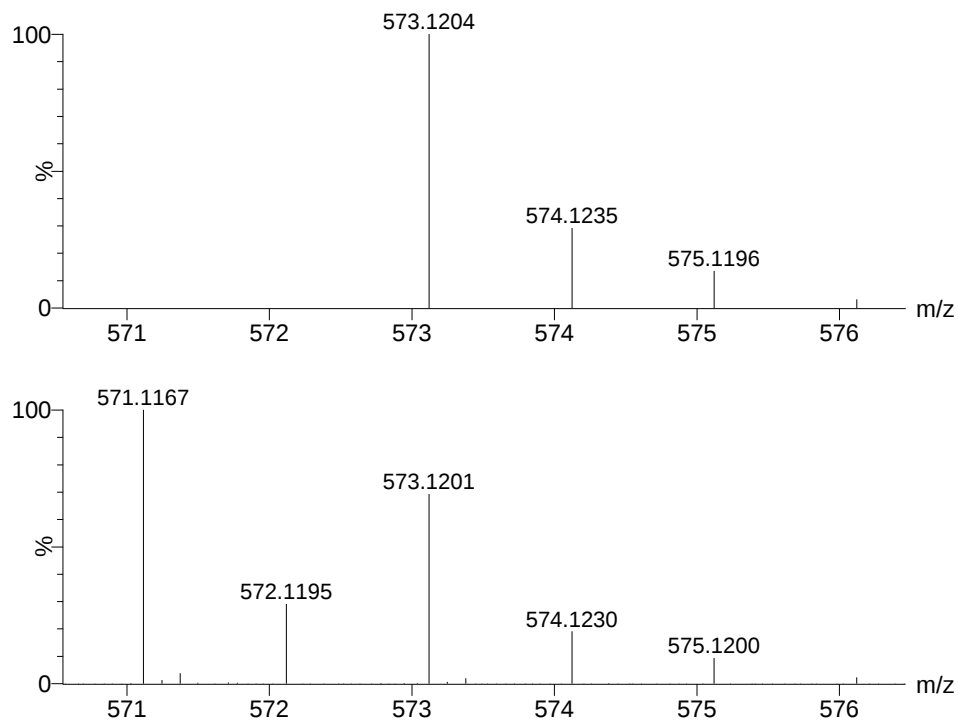
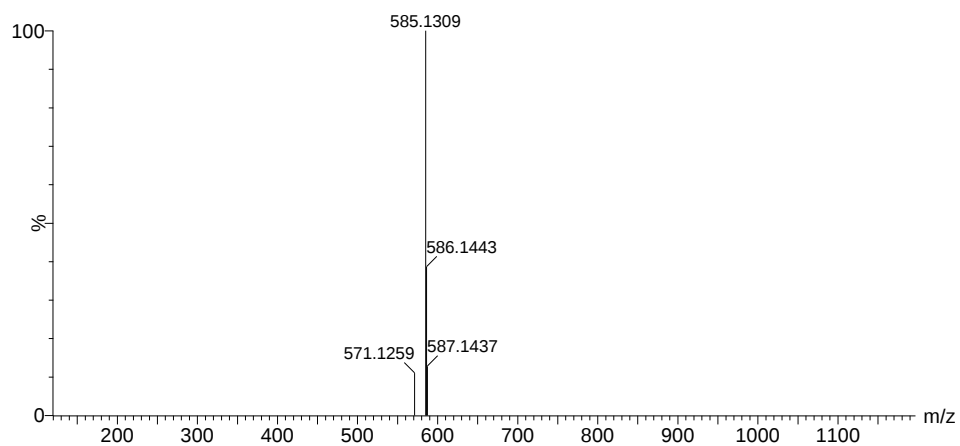


Fig. S42. ESI-HRMS of **4**[BF₄] in CH₂Cl₂

(a) The signal at an $m/z = 585.1309$ corresponds to $[\mathbf{4}]^+$. (b) Calculated isotopic distribution for $[\mathbf{4}]^+$ (upper) and the amplifying diagram for $[\mathbf{4}]^+$ (bottom).

(a)



(b)

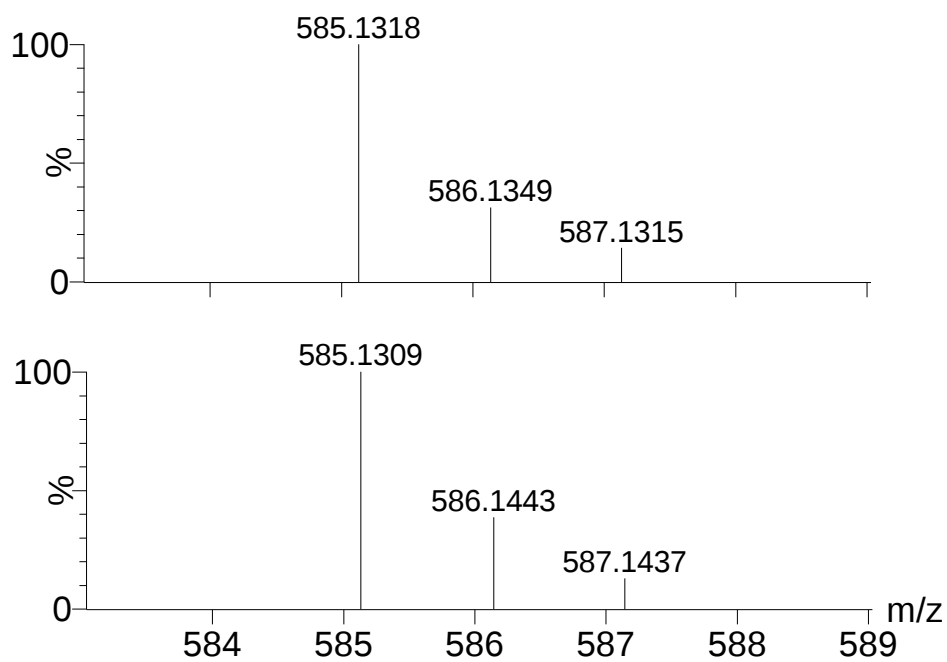
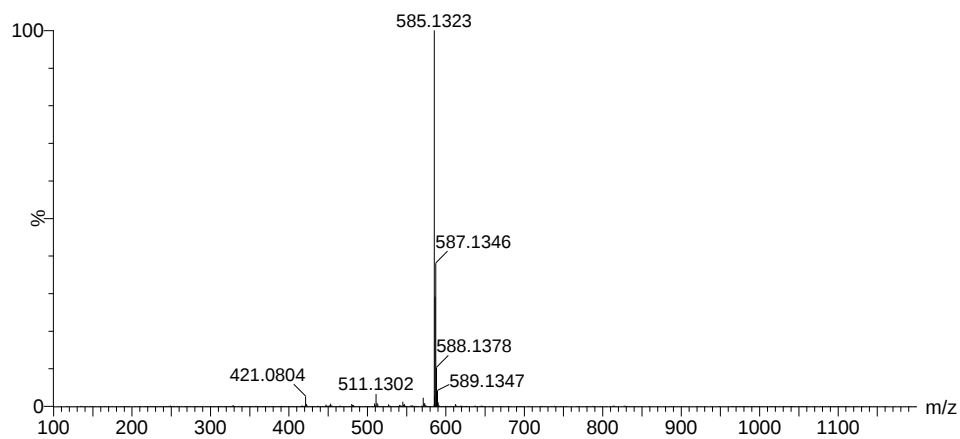


Fig. S43. ESI-HRMS of ^{18}O -**4**[BF₄] in CH₂Cl₂

(a) The signal at an $m/z = 587.1346$ corresponds to ^{18}O -[**4**]⁺. (b) Calculated isotopic distribution for ^{18}O -[**4**]⁺ (upper) and the amplifying diagram for ^{18}O -[**4**]⁺ (bottom).

(a)



(b)

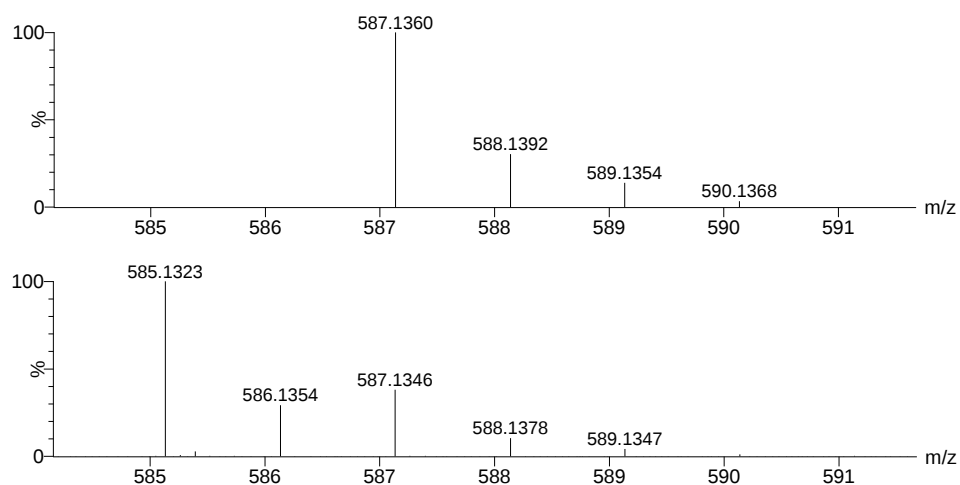
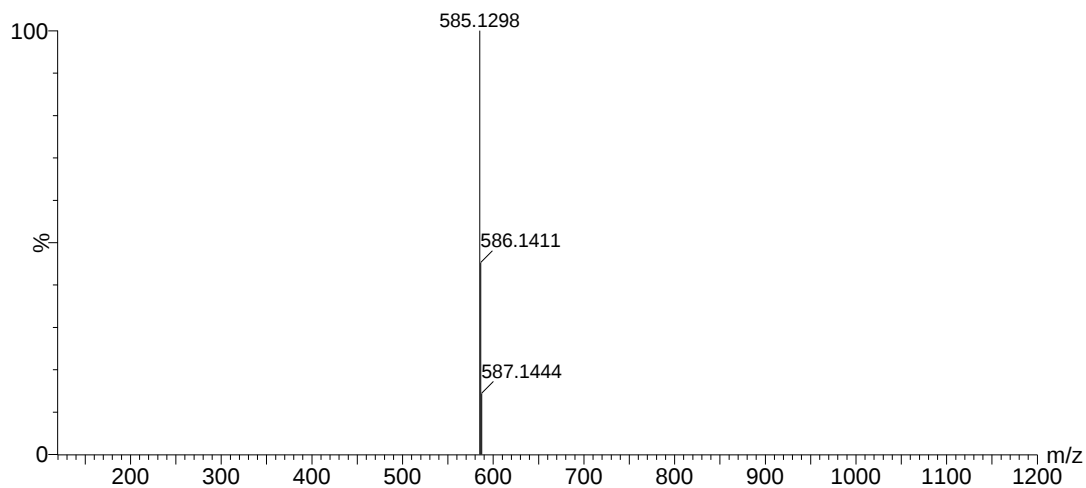


Fig. S44. ESI-HRMS of **5a**[PF₆] in CH₂Cl₂

(a) The signal at an $m/z = 585.1298$ corresponds to **[5a]⁺**. (b) Calculated isotopic distribution for **[5a]⁺** (upper) and the amplifying diagram for **[5a]⁺** (bottom).

(a)



(b)

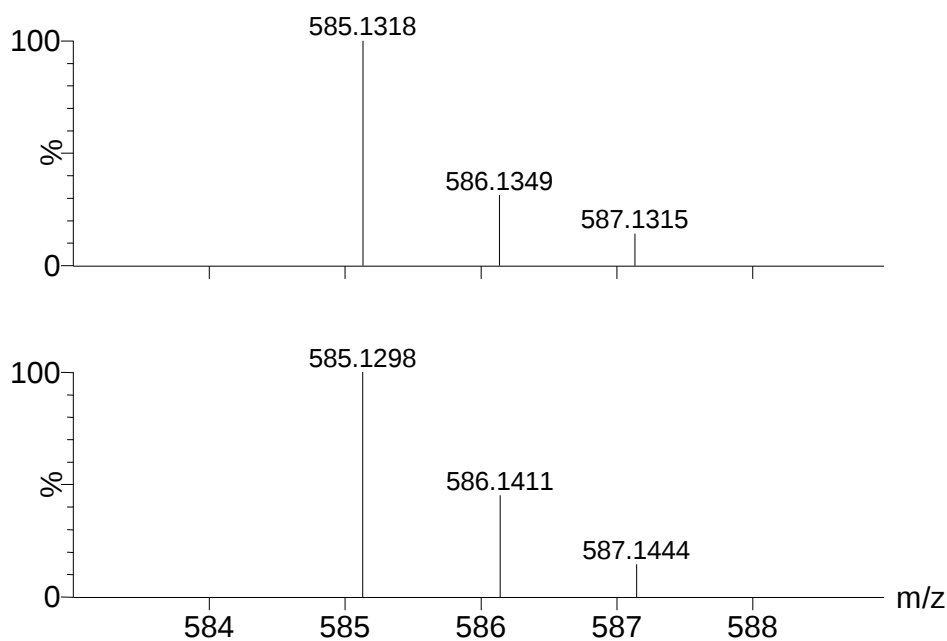
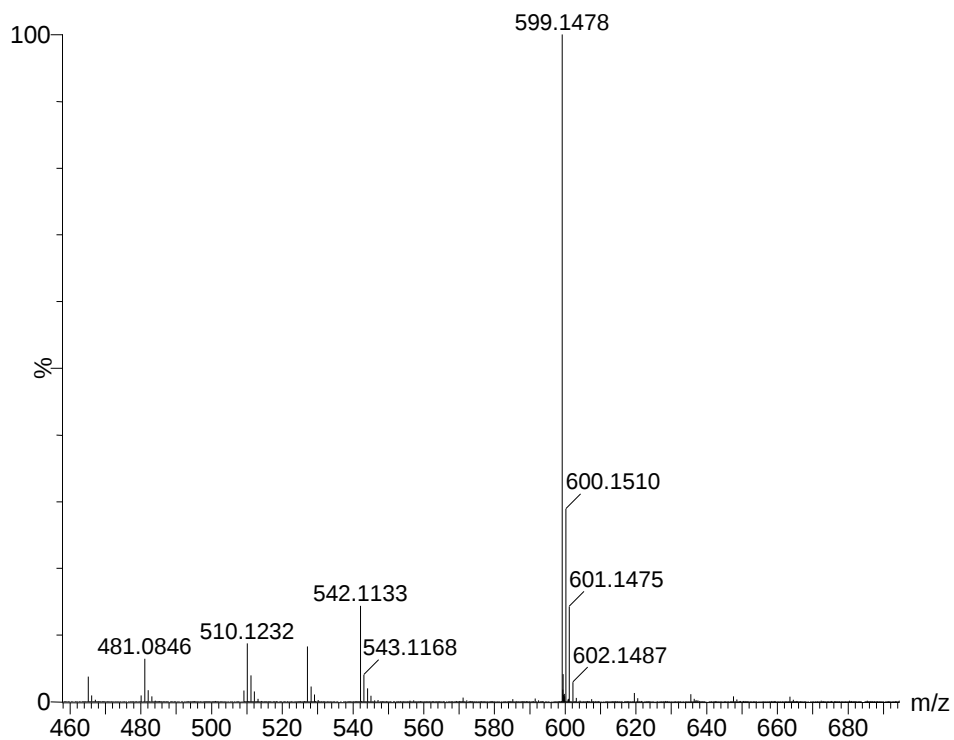


Fig. S45. ESI-HRMS of **5b**[PF₆] in CH₂Cl₂

(a) The signal at an $m/z = 599.1478$ corresponds to **[5b]⁺**. (b) Calculated isotopic distribution for **[5b]⁺** (upper) and the amplifying diagram for **[5b]⁺** (bottom).

(a)



(b)

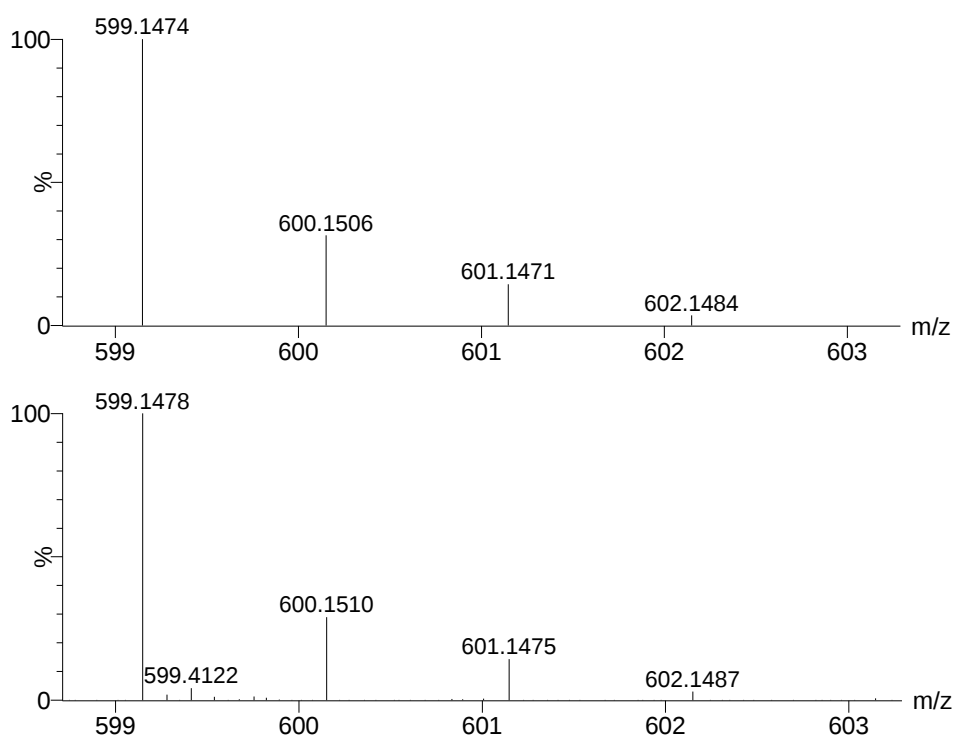
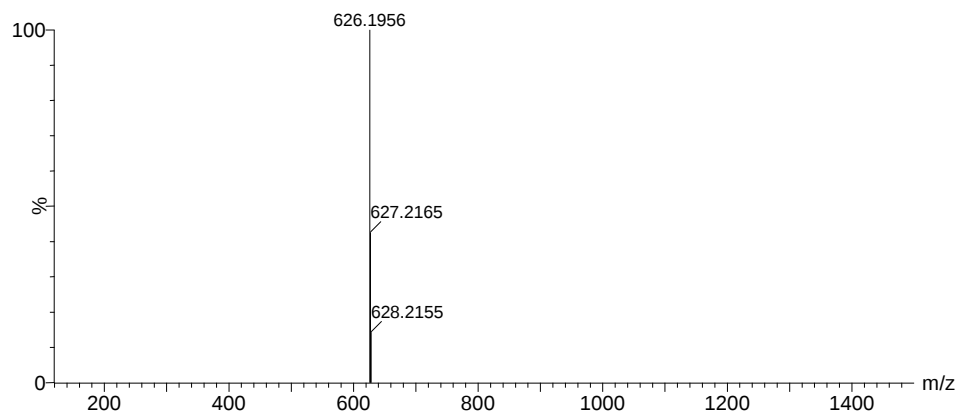


Fig. S46. ESI-HRMS of **6a**[PF₆] in CH₂Cl₂

(a) The signal at an $m/z = 626.1956$ corresponds to **[6a]**⁺. (b) Calculated isotopic distribution for **[6a]**⁺ (upper) and the amplifying diagram for **[6a]**⁺ (bottom).

(a)



(b)

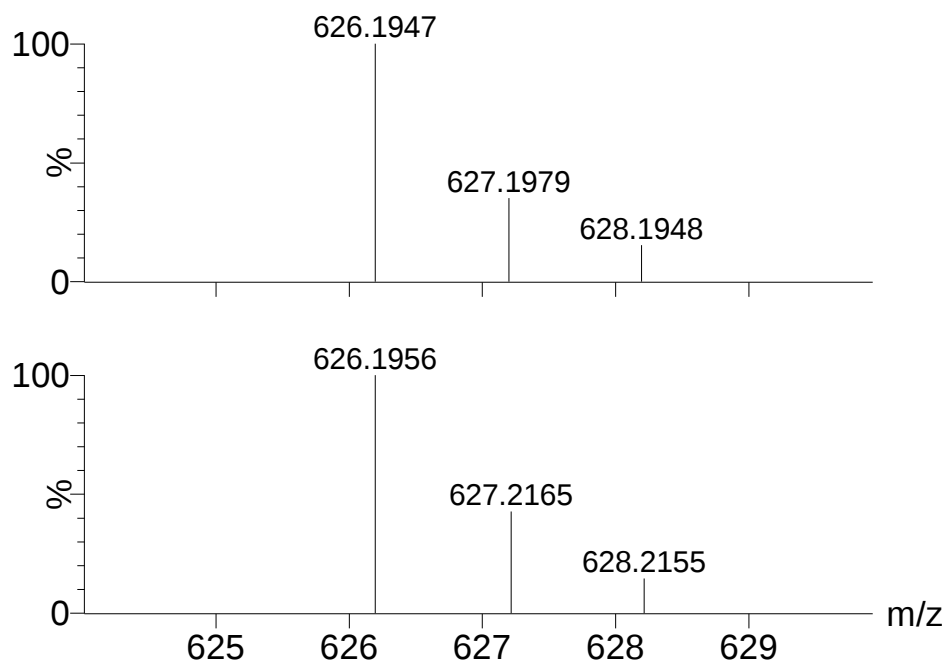
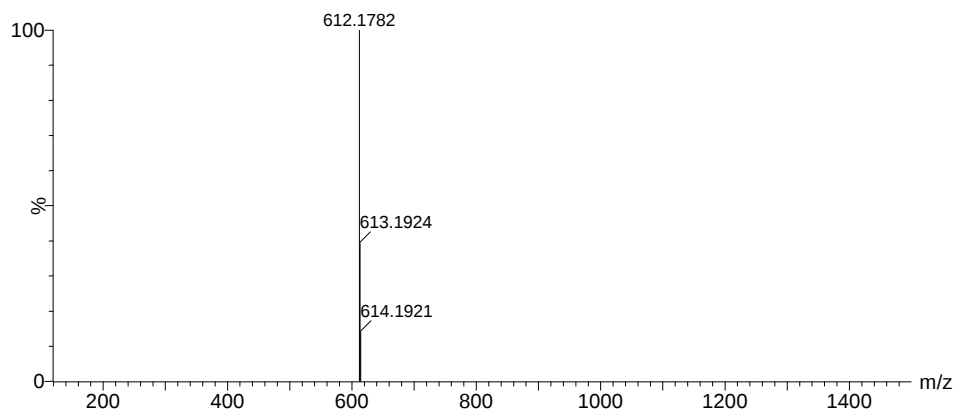


Fig. S47. ESI-HRMS of **6b**[PF₆] in CH₂Cl₂

(a) The signal at an $m/z = 612.1782$ corresponds to **[6b]⁺**. (b) Calculated isotopic distribution for **[6b]⁺** (upper) and the amplifying diagram for **[6b]⁺** (bottom).

(a)



(b)

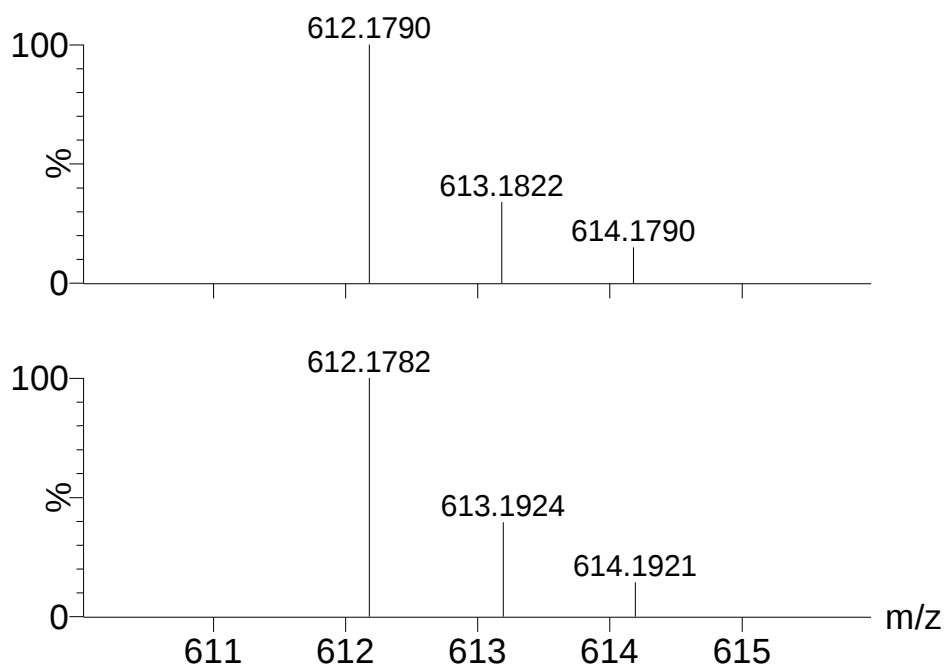
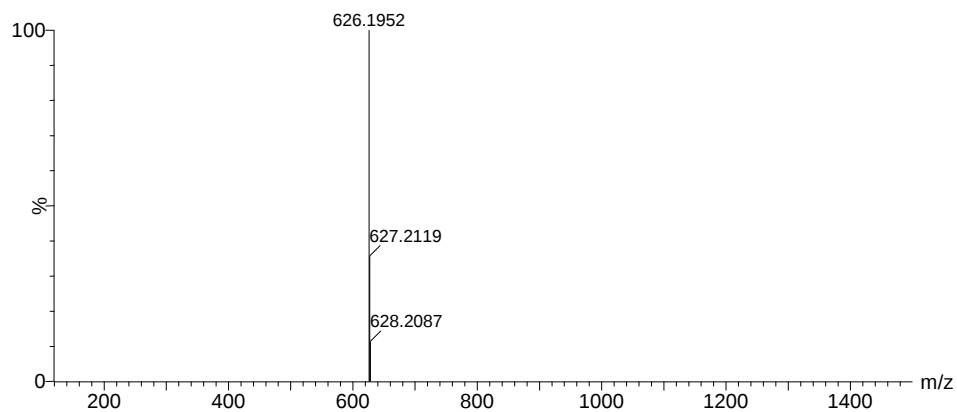


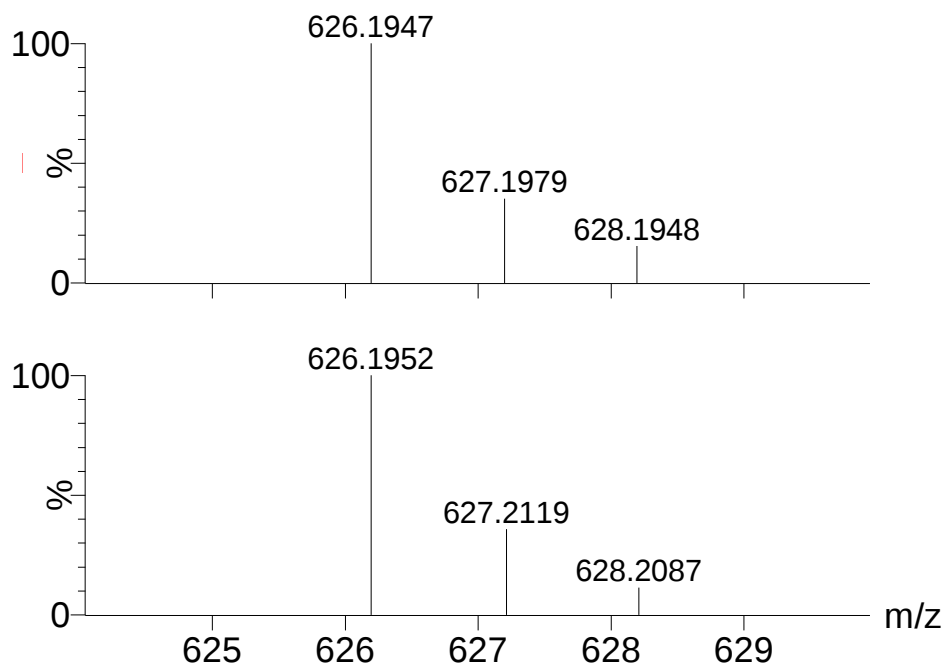
Fig. S48. ESI-HRMS of **6c**[PF₆] in CH₂Cl₂

(a) The signal at an $m/z = 612.1952$ corresponds to $[\mathbf{6c}]^+$. (b) Calculated isotopic distribution for $[\mathbf{6c}]^+$ (upper) and the amplifying diagram for $[\mathbf{6c}]^+$ (bottom).

(a)



(b)



IV. IR(film) Spectra

Fig. S49. IR (film) spectrum of **1**

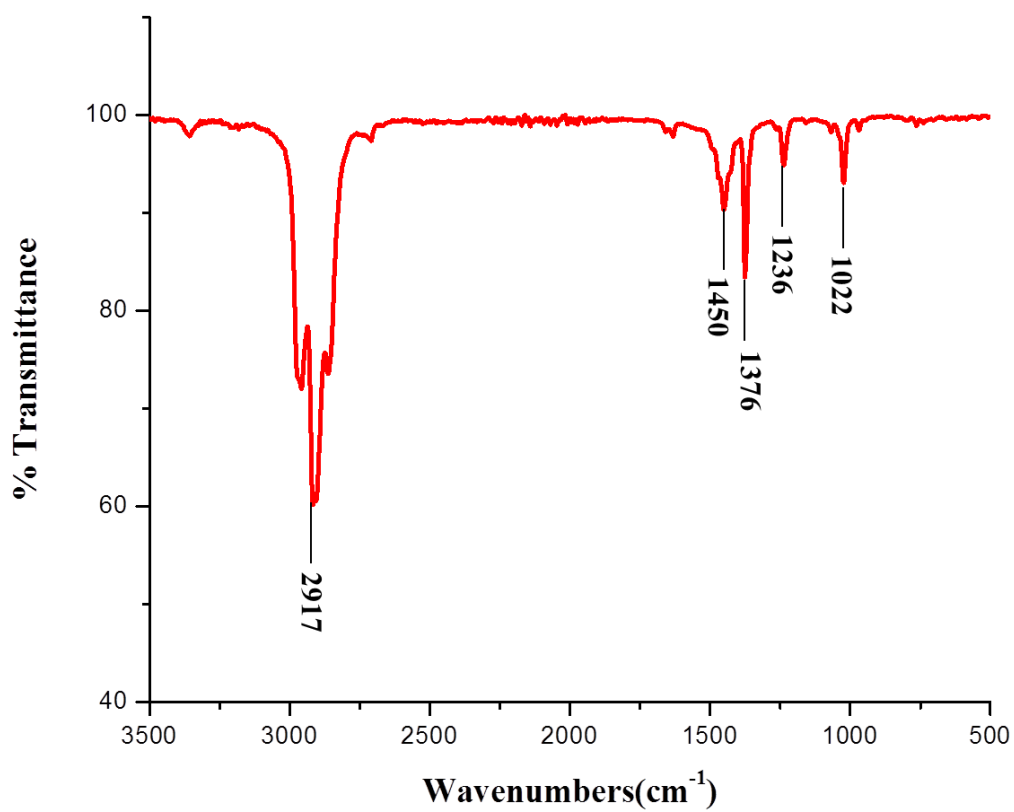


Fig. S50. IR (film) spectrum of **2**

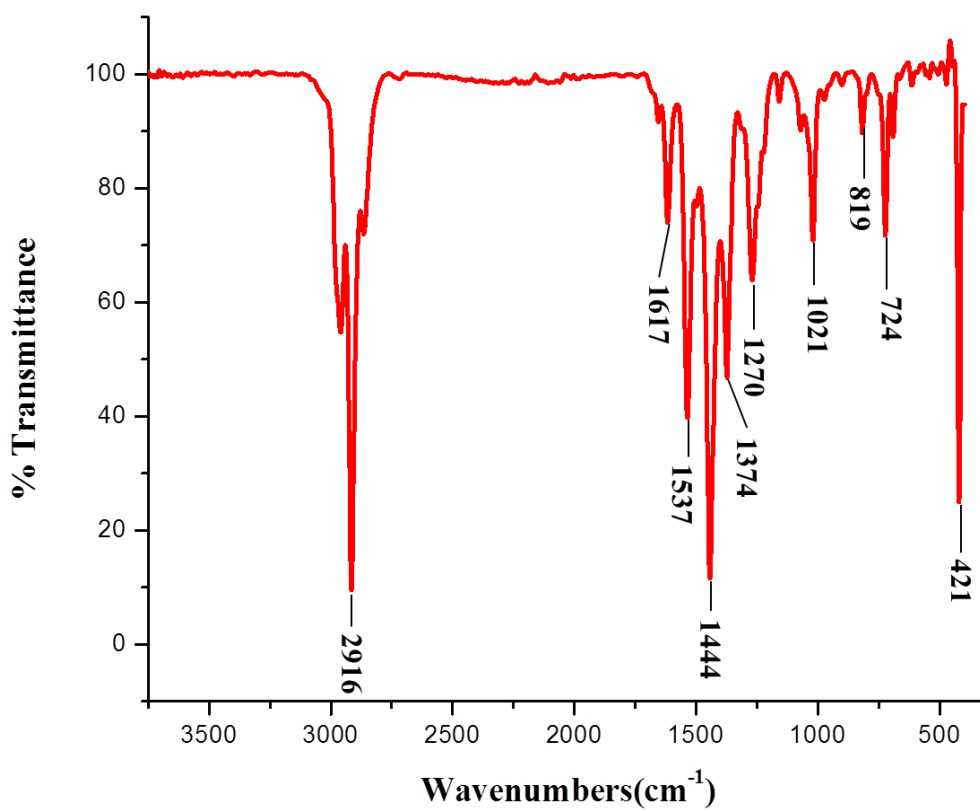


Fig. S51. IR (film) spectrum of ¹⁸O-**2**

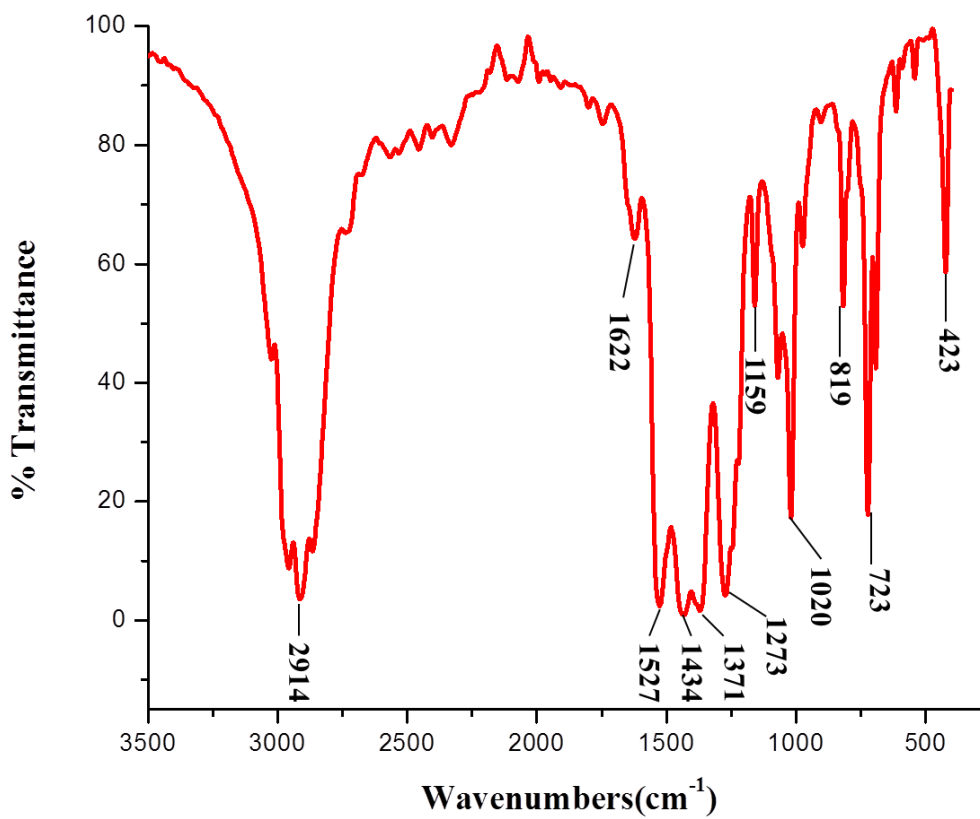


Fig. S52. IR (film) spectrum of 3[BF₄]

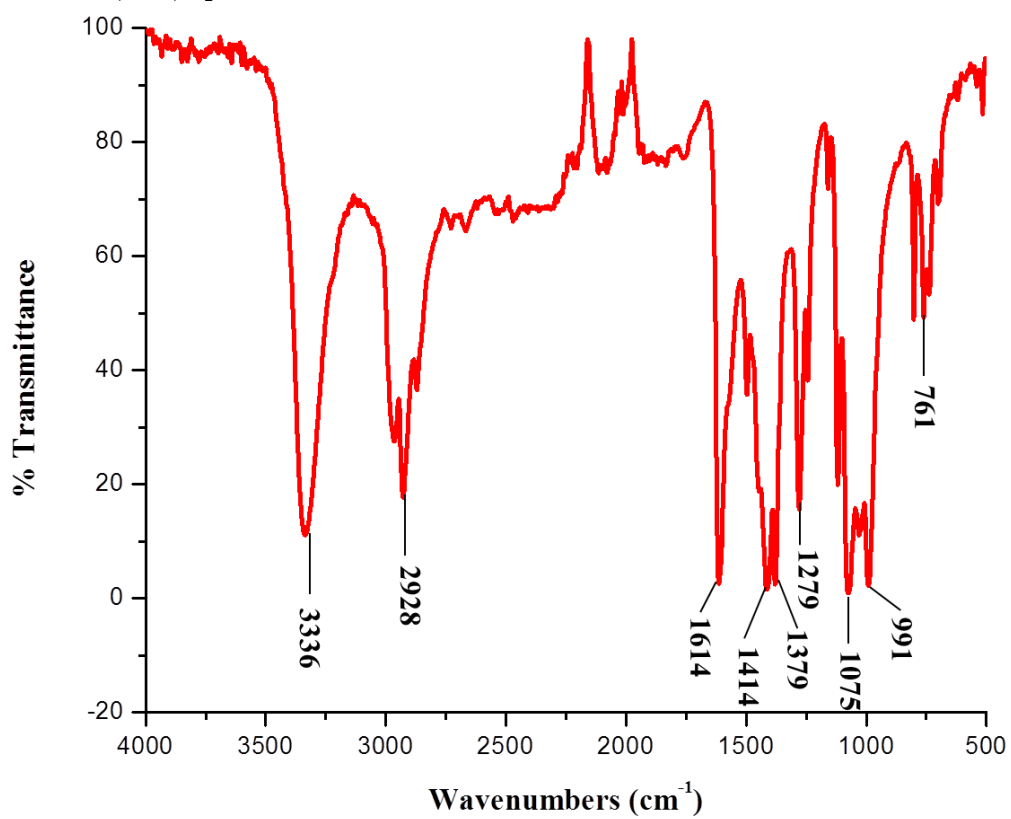


Fig. S53. IR (film) spectrum of 4[BF₄]

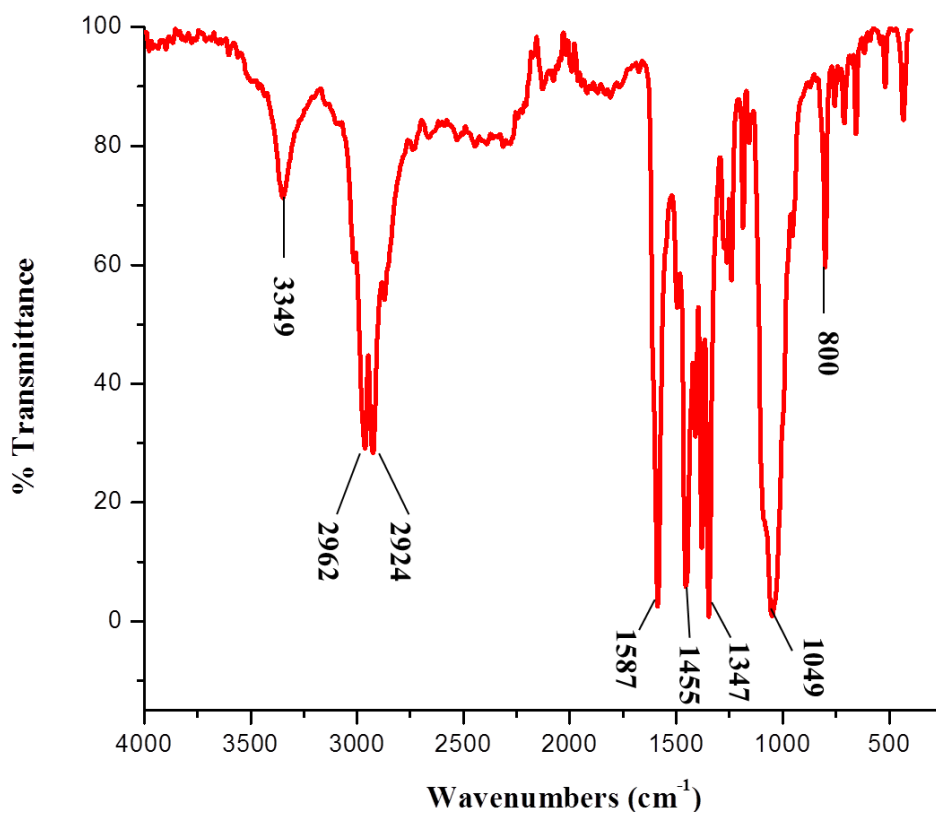


Fig. S54. IR (film) spectrum of **5a**[PF₆]

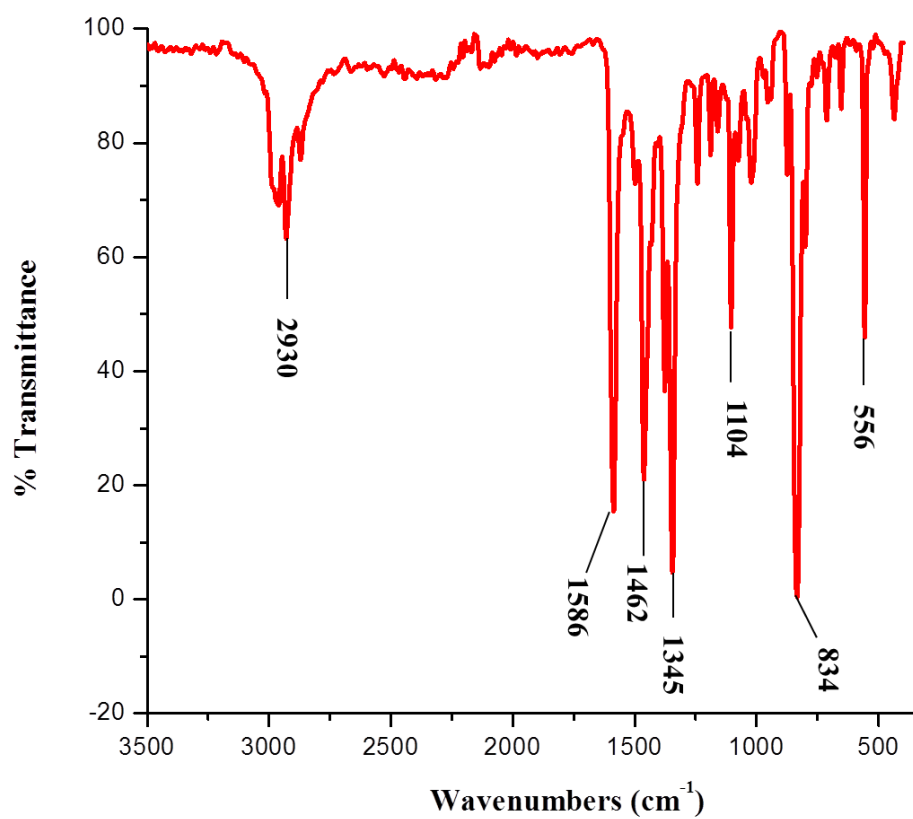


Fig. S55. IR (film) spectrum of **5b**[PF₆]

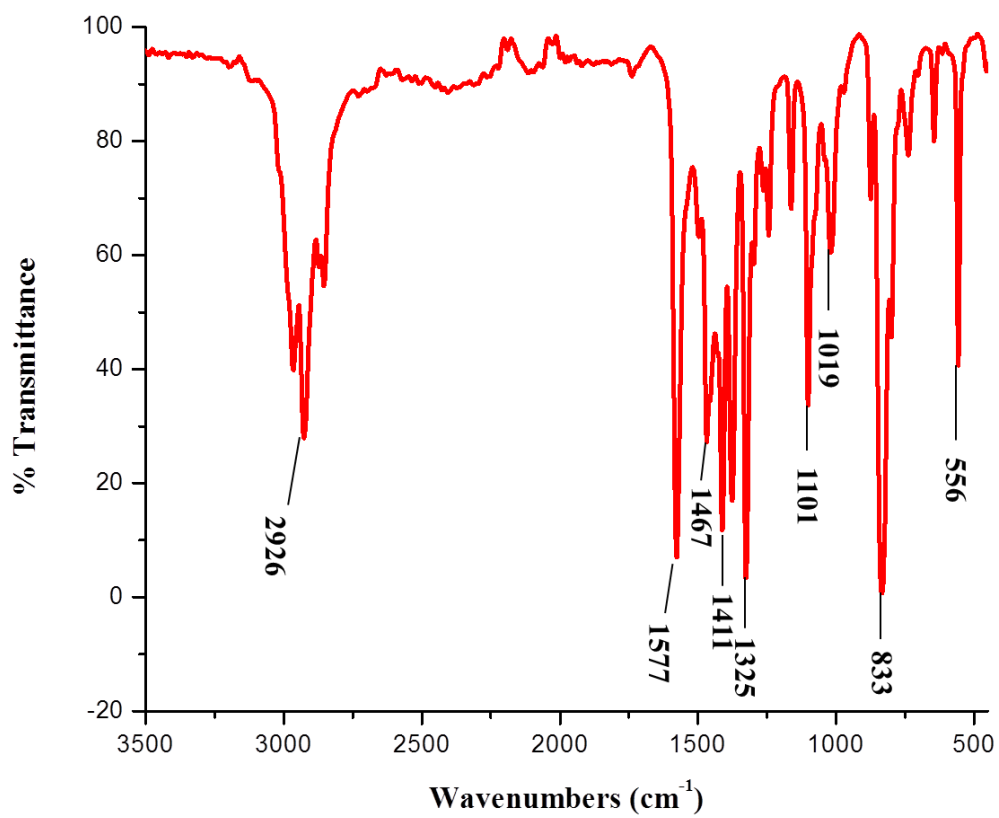


Fig. S56. IR (film) spectrum of **6a**[PF₆]

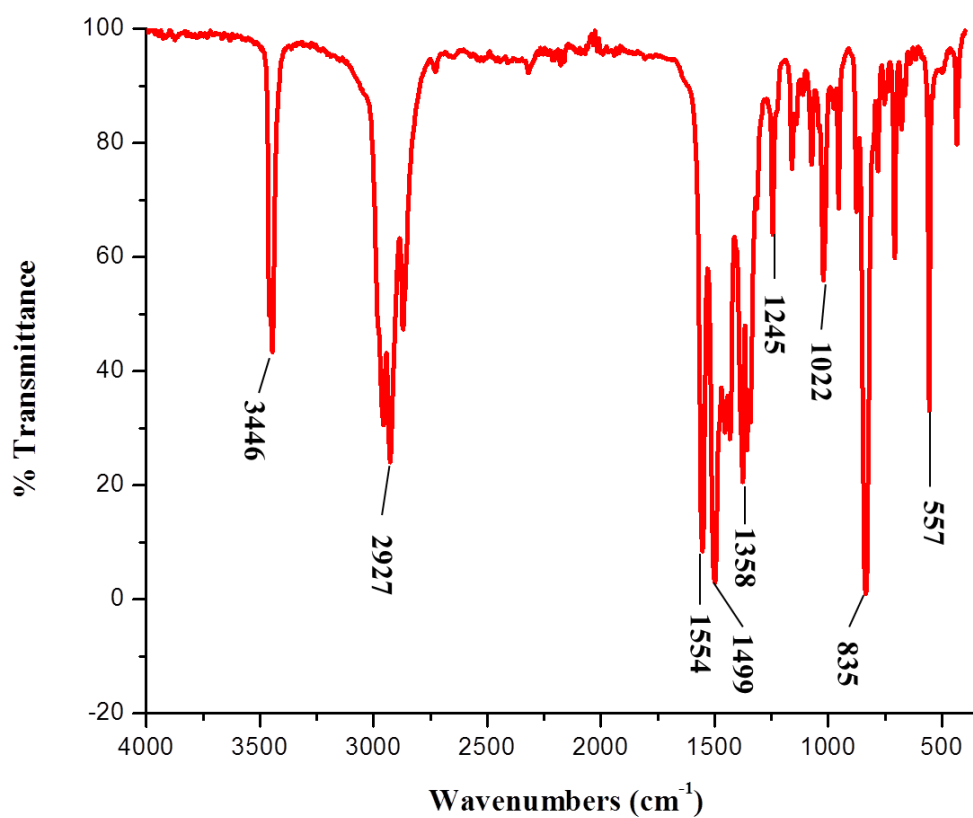


Fig. S57. IR (film) spectrum of **6b**[PF₆]

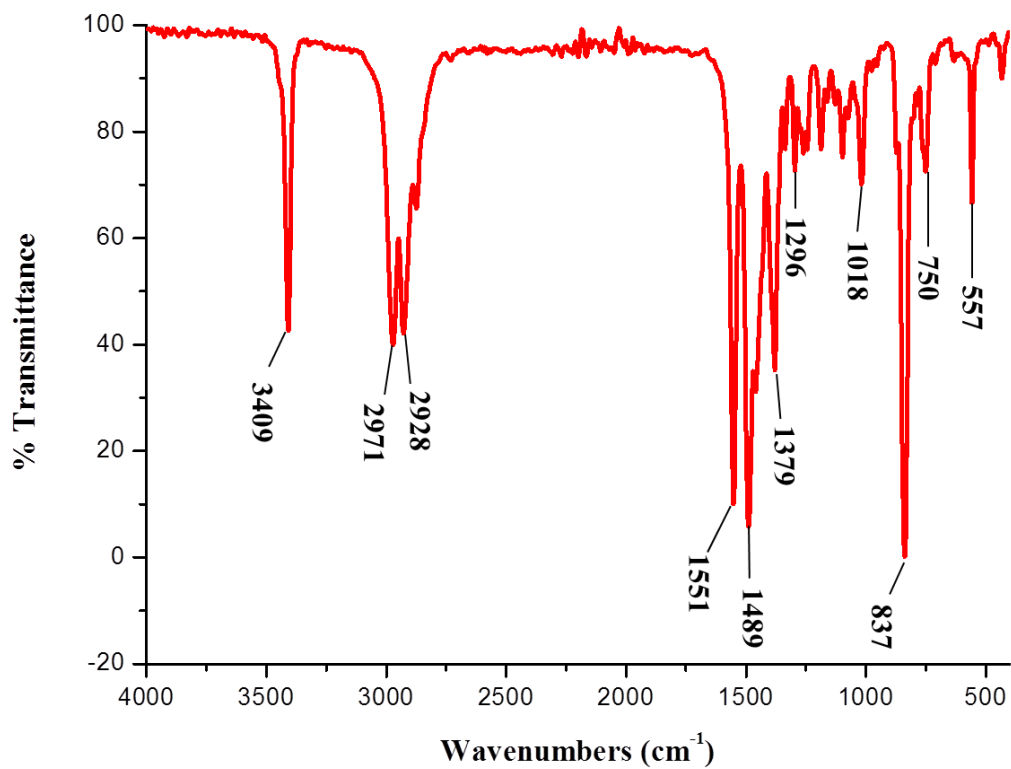
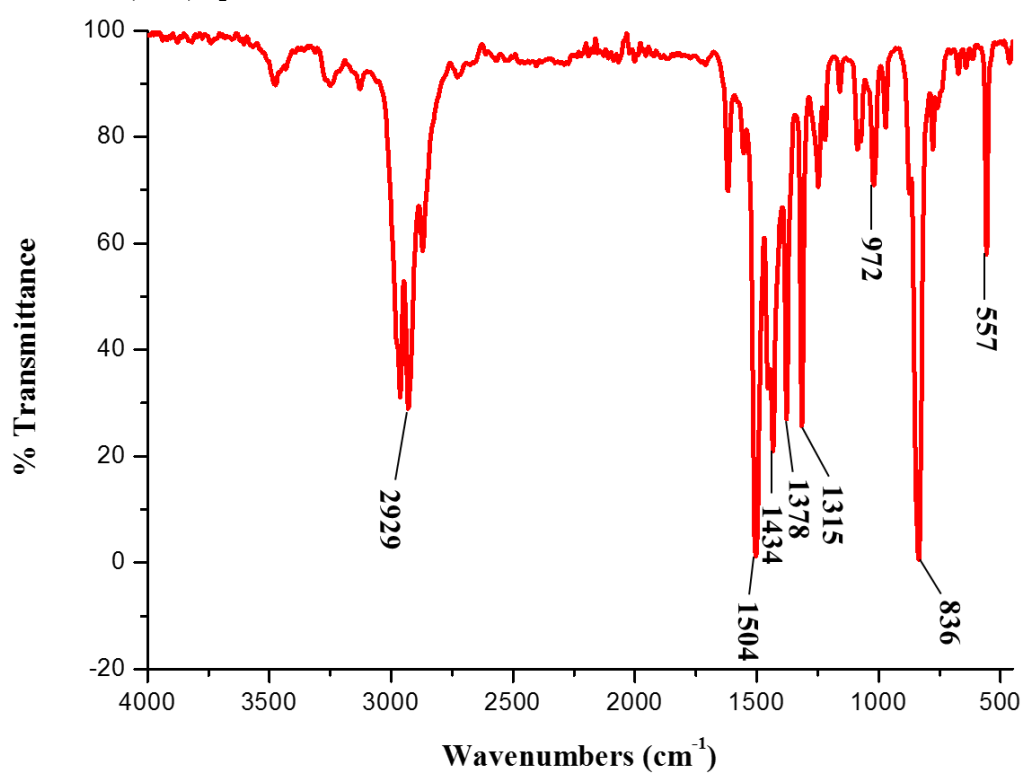


Fig. S58. IR (film) spectrum of **6c[PF₆]**



V. UV-Vis Spectra

Fig. S59. UV-Vis spectra of **1** and **2**

The red line corresponds to **1** and the black line corresponds to **2**.

