Electronic Supplementary Information (ESI) for:

Centrosymmetric K₂SO₄·(SbF₃)₂ and Noncentrosymmetric Rb₂SO₄·(SbF₃)₂ Resulting from Cooperative Effects of Lone Pair and Cation Size

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Atom	X	у	Z	U(ea)	BVS
	0.000(0.(0)				DVS
Sb1	0.82363 (3)	0.26103 (4)	0.72136 (2)	0.00819 (11)	2.87
Sb2	0.31451 (3)	0.78202 (5)	0.50037 (2)	0.00918 (11)	2.95
K1	0.63949 (11)	0.74910 (13)	0.82655 (5)	0.0126 (2)	1.01
K2	1.01873 (11)	0.24770 (14)	0.92575 (5)	0.0135 (2)	0.93
S 1	0.63048 (11)	0.73935 (15)	0.62219 (5)	0.0085 (2)	5.97
F1	0.8624 (3)	0.4739 (4)	0.80384 (11)	0.0165 (5)	1.09
F2	0.9294 (3)	0.0301 (4)	0.79144 (12)	0.0186 (5)	0.91
F3	0.6470 (2)	0.1549 (4)	0.75095 (12)	0.0149 (5)	1.13
F4	0.2645 (2)	0.5580 (4)	0.56661 (11)	0.0147 (5)	1.14
F5	0.1080 (3)	0.7131 (4)	0.44596 (13)	0.0180 (5)	0.90
F6	0.2082 (2)	1.0234 (4)	0.54291 (12)	0.0148 (5)	1.12
01	0.7634 (3)	0.8914 (6)	0.64463 (15)	0.0206 (7)	1.88
O2	0.4936 (3)	0.8891 (5)	0.60789 (13)	0.0121 (5)	1.90
O3	0.6391 (3)	0.6206 (6)	0.55520 (14)	0.0225 (7)	1.76
O4	0.6236 (3)	0.5680 (6)	0.67873 (16)	0.0237 (7)	1.89

Table S1. Atomic coordinates and equivalent isotropic displacement parameters (Å²), and calculated Bond Valence Sum for K₂SO₄·(SbF₃)₂. U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Atom	х	у	Z	U(eq)	BVS
Sb1	0.17363 (6)	0.5010 (5)	0.55869 (6)	0.00731 (14)	2.99
Sb2	0.31898 (7)	0.5032 (4)	0.99959 (6)	0.02175 (19)	3.06
Rb1	0.02396 (9)	0.9986 (7)	0.85431 (9)	0.01102 (19)	1.12
Rb2	0.35956 (9)	0.0020 (6)	0.34806 (9)	0.01104 (18)	1.26
S1	0.3706 (2)	-0.0018 (15)	0.7562 (2)	0.0118 (5)	5.97
F1	0.2525 (12)	0.2838 (11)	1.1222 (10)	0.034 (3)*	1.24
F2	0.2244 (10)	0.7570 (16)	1.0886 (9)	0.021 (2)*	1.14
F3	0.1159 (6)	0.501 (3)	0.8857 (6)	0.0162 (11)*	1.11
F4	0.1300 (10)	0.7163 (15)	0.4003 (9)	0.0203 (17)*	1.16
F5	0.0734 (10)	0.2715 (15)	0.4201 (8)	0.0201 (17)*	1.00
F6	0.3456 (8)	0.4126 (15)	0.4977 (9)	0.0233 (18)	1.27
01	0.2453 (10)	0.1428 (18)	0.7095 (10)	0.020 (2)	2.12
O2	0.5121 (10)	0.1092 (15)	0.7865 (9)	0.0129 (17)*	2.10
O3	0.3556 (11)	-0.1376 (19)	0.8843 (11)	0.025 (2)	1.72
O4	0.3721 (12)	-0.1924 (19)	0.6474 (11)	0.026 (2)	1.65

Table S2. Atomic coordinates and equivalent isotropic displacement parameters (Å²), and calculated Bond Valence Sum for Rb₂SO₄·(SbF₃)₂. U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Sb1—F3	1.935 (2)	K2—O1 ^{viii}	2.783 (3)
Sb1—F1	1.958 (2)	K2—F2	2.809 (2)
Sb1—F2	1.967 (2)	K2—F6 ^{ix}	2.827 (2)
Sb1—O4	2.525 (3)	K2—F4 ⁱⁱ	2.852 (2)
Sb1—O1 ⁱ	2.545 (3)	K2—F5 ⁱⁱ	2.966 (3)
Sb1—K1 ⁱ	4.1111 (9)	K2—F5 ^{ix}	3.157 (3)
Sb1—K1	4.0173 (9)	K2—O3 ^{viii}	3.167 (3)
Sb1—K1 ⁱⁱ	4.1472 (10)	O1—K2 ^x	2.783 (3)
Sb2—F4	1.925 (2)	S1—O4	1.468 (3)
Sb2—F6	1.963 (2)	S1—O3	1.471 (3)
Sb2—F5	1.987 (2)	S1—O1	1.476 (3)
Sb2—O2	2.408 (3)	S1—O2	1.490 (3)
K1—F1	2.687 (2)	F3—K1 ⁱⁱ	2.766 (2)
K1—F4 ⁱⁱⁱ	2.687 (2)	F4—K1 ⁱⁱ	2.687 (2)
K1—F3 ^{vi}	2.725 (2)	F4—K2 ⁱⁱⁱ	2.852 (2)
K1—F3 ⁱⁱⁱ	2.766 (2)	F5—K2 ^v	2.732 (3)
K1—O2 ⁱⁱ	2.817 (3)	F5—K2 ⁱⁱⁱ	2.966 (3)
K1—F6 ⁱⁱ	2.879 (2)	F5—K2 ^{iv}	3.157 (3)
K1—O4	2.994 (3)	F6—K2 ⁱⁱⁱ	2.779 (2)
K1—O4 ⁱⁱⁱ	3.000 (3)	F6—K2 ^{iv}	2.827 (2)
K2—F1	2.768 (2)	O3—K2 ^x	3.167 (3)
K2—F6 ⁱⁱ	2.779 (2)	O4—K1 ⁱⁱ	3.000 (3)
S1—O3—K2 ^x	92.63 (13)	Sb2—F6—K2 ^{iv}	105.11 (9)
S1—O4—Sb1	122.84 (16)	K2 ⁱⁱⁱ —F6—K2 ^{iv}	94.19 (7)
S1—O4—K1 ⁱⁱ	122.82 (16)	S1—O1—Sb1 ^{vi}	134.71 (16)
Sb1—O4—K1 ⁱⁱ	96.90 (10)	S1—O1—K2 ^x	109.16 (15)
K1—O4—K1 ⁱⁱ	95.83 (8)	Sb1 ^{vi} —O1—K2 ^x	115.86 (10)
O2 ⁱⁱ —K1—F6 ⁱⁱ	57.59 (7)	04—S1—O3	111.58 (19)

Table S3. Selected Bond lengths (Å) and angles (deg) for $K_2SO_4 \cdot (SbF_3)_2$.

-	F1—K1—O4	61.39 (7)	O4—S1—O1	109.90 (18)
	F4 ⁱⁱⁱ —K1—O4	152.20 (8)	O3—S1—O1	108.37 (18)
	F3 ^{vi} —K1—O4	77.45 (8)	O4—S1—O2	109.52 (16)
	F3 ⁱⁱⁱ —K1—O4	65.71 (7)	O3—S1—O2	108.05 (16)
	O2 ⁱⁱ —K1—O4	104.26 (8)	O1—S1—O2	109.38 (17)
	F6 ⁱⁱ —K1—O4	125.82 (8)	O4—S1—K2 ^x	113.34 (12)
	S1—O4—K1	118.48 (18)	S1—O2—Sb2	113.52 (14)
	Sb1—O4—K1	93.02 (9)	S1—O2—K1 ⁱⁱⁱ	139.75 (15)

Symmetry codes: (i) *x*, *y*-1, *z*; (ii) -*x*+1, *y*-1/2, -*z*+3/2; (iii) -*x*+1, *y*+1/2, -*z*+3/2; (iv) *x*-1, -*y*+3/2, *z*-1/2; (v) *x*-1, -*y*+1/2, *z*-1/2; (vi) *x*, *y*+1, *z*; (vii) *x*+1, -*y*+1/2, *z*+1/2; (viii) -*x*+2, *y*-1/2, -*z*+3/2; (ix) *x*+1, -*y*+3/2, *z*+1/2; (x) -*x*+2, *y*+1/2, -*z*+3/2.

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Table S4.	Selected E	Sond lengths	s (Å) and	angles (deg)) for Rb_2SO_4 ·	$(SbF_3)_2$.

Sb1—F6	1.923 (8)	Rb1—F5 ^x	2.947 (8)
Sb1—F4	1.957 (8)	Rb1—F2	2.955 (10)
Sb1—F5	1.966 (8)	Rb1—F3	2.969 (17)
Sb1—O1	2.528 (9)	Rb1—F3 ⁱ	2.996 (17)
Sb1—O4 ⁱ	2.558 (11)	Rb1—F3 ^{xi}	3.153 (5)
Sb1—Rb1 ⁱⁱ	4.0584 (11)	Rb1—O3 ⁱ	3.160 (11)
Sb2—F1	1.9482 (10)	S1—O2	1.440 (10)
Sb2—F3	1.971 (6)	S1—O3	1.520 (10)
Sb2—F2	2.012 (8)	S1—O4	1.534 (11)
Sb2—O2 ^{iv}	2.398 (9)	S1—O1	1.423 (10)
Sb2—O3 ⁱ	2.414 (10)	Rb2—F4 ^{vi}	2.852 (9)
Rb2—F5	3.323 (9)	Rb2—F6 ^{viii}	2.872 (8)
Rb1—F2 ^{xi}	2.931 (9)	Rb2—F2 ^{ix}	2.931 (9)
Rb1—F1 ^{xi}	2.935 (11)	Rb2—O2 ^{viii}	3.007 (9)
O4—Sb1 ^{vi}	2.558 (10)	O2—Sb2 ^{xiv}	2.398 (9)
O2—Rb2 ⁱⁱⁱ	3.007 (9)		

F6—Sb1—F4	86.9 (4)	F5—Rb2—Sb1 ^{vi}	75.76 (15)
F6—Sb1—F5	84.6 (3)	S1—Rb2—Sb1 ^{vi}	53.60 (10)
F4—Sb1—F5	83.7 (3)	Sb1—Rb2—Sb1 ^{vi}	86.86 (2)
F6—Sb1—O1	81.1 (3)	F4 ^x —Rb1—O1 ⁱ	73.5 (3)
F4—Sb1—O1	163.1 (3)	$F4^x$ —Rb1— $F2^{xi}$	72.0 (2)
F5—Sb1—O1	83.4 (3)	O1 ⁱ —Rb1—F2 ^{xi}	130.4 (3)
F6—Sb1—O4 ⁱ	71.8 (3)	$F4^x$ — $Rb1$ — $F1^{xi}$	88.9 (3)
F4—Sb1—O4 ⁱ	80.7 (4)	$O1^i$ —Rb1—F1 ^{xi}	155.3 (3)
F6—Sb1—Rb2	34.6 (2)	O1—Sb1—Rb2 ⁱⁱⁱ	76.0 (2)
F6—Sb1—Rb2 ⁱ	62.6 (3)	O1 ⁱ —Rb1—F3 ^{xi}	152.3 (3)
F1—Sb2—O2 ^{iv}	81.9 (4)	F4 ^x —Rb1—O3 ⁱ	118.0 (3)
F3—Sb2—O2 ^{iv}	148.4 (4)	O1 ⁱ —Rb1—O3 ⁱ	46.4 (3)
F2—Sb2—O2 ^{iv}	72.6 (4)	F2 ^{xi} —Rb1—O3 ⁱ	157.0 (3)
F1—Sb2—O3 ⁱ	160.9 (4)	$F1^{xi}$ —Rb1—O3 ⁱ	139.7 (3)
F3—Sb2—O3 ⁱ	88.0 (5)	F5 ^x —Rb1—O3 ⁱ	93.4 (3)
F4 ^{vi} —Rb2—O4 ⁱⁱⁱ	169.0 (3)	O3—S1—Rb2	148.8 (6)
F6 ^{viii} —Rb2—O4 ⁱⁱⁱ	53.6 (2)	O4—S1—Rb2	45.7 (5)
O1—S1—O2	117.6 (8)	O1—S1—O4	109.4 (6)
01—S1—O3	109.4 (6)	O2—S1—O4	107.2 (6)
02—S1—O3	108.5 (5)	O3—S1—O4	103.8 (8)

Symmetry codes: (i) x, y+1, z; (ii) -x, y-1/2, -z+1; (iii) -x+1, y+1/2, -z+1; (iv) -x+1, y+1/2, -z+2; (v) -x, y-1/2, -z+2; (vi) x, y-1, z; (vii) x, y, z-1; (viii) -x+1, y-1/2, -z+1; (ix) x, y-1, z-1; (x) -x, y+1/2, -z+1; (xi) -x, y+1/2, -z+2; (xii) x, y, z+1; (xiii) x, y+1, z+1; (xiv) -x+1, y-1/2, -z+2.



Fig. S1. Crystal photographs of compounds (a) $K_2SO_4 \cdot (SbF_3)_2$ and (b) $Rb_2SO_4 \cdot (SbF_3)_2$.



Fig. S2. TGA curves of compounds (a) K_2SO_4 · (SbF₃)₂ and (b) Rb_2SO_4 · (SbF₃)₂.



Fig. S3. Experimental XRD patterns for compounds (a) $K_2SO_4 \cdot (SbF_3)_2$ and (b) $Rb_2SO_4 \cdot (SbF_3)_2$ before and after melting.



Fig. S4. The IR spectra of compounds (a) K_2SO_4 · $(SbF_3)_2$ and (b) Rb_2SO_4 · $(SbF_3)_2$.



Fig. S5. Calculated band structures of (a) $K_2SO_4 \cdot (SbF_3)_2$ and (b) $Rb_2SO_4 \cdot (SbF_3)_2$.