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Supplementary Information

A Novel Strategy for the Synthesis of Hard Carbon Spheres Encapsulated with Graphene Networks as a Low-Cost and Large-Scalable Anode Material for Fast Sodium Storage with an Ultralong Cycle Life

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Fig. S1. SEM image of hard carbon spheres.



Fig. S2. SEM image of hard carbon spheres wrapped with flexible conductive networks of graphene nanosheets.



Fig. S3. SEM image (a) and (b-d) SEM-EDS elemental mapping of G-HCS.



Fig. S4. EDS spectra for elemental composition of G-HCS.



Fig. S5. XPS analysis of G-HCS sample (a) XPS survey, High-resolution XPS spectra of (b) C 1s and (c) O 1s.



Fig. S6. Galvanostatic discharge/charge curves of HCS at current rate of 100 mA g⁻¹.

Carbon Anode	Initial reversible capacity (mA h g ⁻¹)	Current rate for ultra-long cycling (A g ⁻¹)	Cycle number	Capacity after cycles (mA h g ⁻¹)	References
G-HCS	421	10	4000	122	This work
	100 mA g ⁻¹				
Hollow Carbon	223	0.1	100	160	[1]
Nanospheres	50 mA g ⁻¹				
Hard Carbon	322	0.128	40	73	[2]
Microspheres	28 mA g ⁻¹				
Hard carbon	220	1	1000	117	[3]
from orange peel	500 mA g ⁻¹				
Rice husk-derived	372	0.025	100	346	[4]
hard carbons	25 mA g ⁻¹				
Sulfurized	295	10	10000	126.5	[5]
polyacrylonitrile	500 mA g ⁻¹				
derived carbon					
Micro-nano	323	0.020	100	286	[6]
structure carbon	20 mA g-1				
Carbon from	385	0.5	600	210	[7]
Banana Peels	50 mA g ⁻¹				
S-doped N-rich	419	1	1000	211	[8]
carbon	50 mA g ⁻¹				
Nitrogen-doped	315.2	1.69	2000	ca. 70	[9]
carbon sheets	56 mA g ⁻¹				
Amorphous	280	0.5	2500	142	[10]
carbon/graphene	100 mA g ⁻¹				
composite					
N/S codoped	280	0.5	3400	ca. 150	[11]
carbon	30 mA g ⁻¹				
microspheres					

Table S1. Performance comparison of G-HCS with other reported state-of-the-artcarbon anode materials for SIBs.

Table S2. Performance comparison of G-HCS with other reported state-of-the-artanode materials for SIBs except carbon materials.

	Initial	Current rate	Cycle	Capacity	
Carbon Anode	reversible	for ultra-long	number	after cycles	References
	capacity	cycling (A g ⁻¹)		(mA h g ⁻¹)	
	(mA h g ⁻¹)				
G-HCS	421	10	4000	122	This work
	100 mA g ⁻¹				
Graphene-TiO ₂	265	0.5	4300	ca. 120	[12]
hybrid	50 mA g ⁻¹				
Amorphous	495	1	800	220	[13]
SnO ₂	50 mA g ⁻¹				
Na ₂ Ti ₃ O ₇ @NDope	297	8.8	1000	68	
d Carbon Hollow	177 mA g ⁻¹				[14]
Spheres					
Graphene/Ni ₂ P	516	0.3	500	161	[15]
Hybrid	100 mA g ⁻¹				
WS ₂ /CNT-rGO	305	0.2	100	252.9	[16]
Aerogel	200 mA g ⁻¹				
Sb-C nanofibers	663	0.2	400	446	[17]
	40 mA g ⁻¹				
SnSe/carbon	447.7	0.5	200	324.6	[18]
nanocomposite	500 mA g ⁻¹				
Sb@C yolk-shell	637	0.2	200	598	[19]
microspheres	50 mA g ⁻¹				
Layered nickel					
sulfide-reduced	512.7	0.1	50	391.6	[20]
graphene oxide	100 mA g ⁻¹				
composites					

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