

$\text{Pb}_3\text{Ba}_3\text{Zn}_6(\text{BO}_3)_8$ and $\alpha\text{-BaZn}_2(\text{BO}_3)_2$: New Members of the
Zincoborates Containing Two Different Dimensional Zn-O Units

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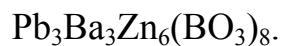
Table S1. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Pb}_3\text{Ba}_3\text{Zn}_6(\text{BO}_3)_8$.

Atoms	x	y	z	$U(\text{eq})^{[a]}$	BVS ^[b, c]
Pb(1)	3555(1)	2309(2)	7170(1)	18(1)	2.062
Pb(2)	5000	1585(2)	7500	20(1)	1.994
Ba(1)	4202(1)	2295(2)	5411(1)	14(1)	2.153
Ba(2)	2500	2500	5000	10(1)	2.055
Zn(1)	2451(1)	7793(4)	3031(1)	11(1)	2.105
Zn(2)	3419(1)	7440(4)	4416(1)	12(1)	2.098
Zn(3)	4671(1)	-2907(4)	8597(1)	16(1)	1.979
B(1)	3012(8)	12570(40)	3592(11)	7(4)	3.089
B(2)	4327(8)	6890(40)	6939(10)	9(4)	2.951
B(3)	3190(9)	7460(40)	5948(12)	17(5)	3.007
B(4)	4477(9)	2100(40)	9215(12)	18(5)	3.093
O(1)	3227(5)	11230(20)	4255(6)	13(3)	1.970
O(2)	4250(5)	-1710(30)	7554(6)	16(3)	1.818
O(3)	4145(5)	6570(30)	4584(7)	21(3)	1.926
O(4)	5375(5)	-1780(30)	8469(7)	17(3)	1.919
O(5)	1981(5)	9080(30)	3565(7)	17(3)	2.102
O(6)	2175(5)	6100(20)	2060(6)	13(3)	2.023
O(7)	4153(5)	4420(30)	6742(7)	24(3)	2.349
O(8)	4743(6)	-6670(30)	8829(9)	30(4)	2.171
O(9)	2977(5)	5230(30)	3598(6)	18(3)	2.114
O(10)	3337(6)	140(30)	6073(7)	27(3)	2.047
O(11)	4495(6)	-630(30)	9311(7)	25(3)	2.115
O(12)	3223(6)	6310(30)	5299(7)	29(3)	2.007

[a] $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

[b] Bond valences calculated with the program Bond Valence Calculator Version 2.00, C. Hormillosa, S. Healy, T. Stephen, McMaster University, 1993;

[c] Valence sums calculated with the formula: $\text{Si} = \exp[(\text{R0}-\text{Ri})/\text{B}]$, where Si = valence of bond “i” and B = 0.37.

Table S2. Selected bond distances (Å) and angles (deg) for

Pb(1)-O(10)	2.231(12)	O(3)-Ba(1)-O(1)#5	79.8(4)
Pb(1)-O(7)	2.232(13)	O(7)-Ba(1)-O(1)#5	118.6(4)
Pb(1)-O(5)#1	2.450(12)	O(8)#4-Ba(1)-O(1)#5	163.6(4)
Pb(1)-O(2)	2.677(13)	O(4)#2-Ba(1)-O(1)#5	119.0(4)
Pb(2)-O(4)	2.454(13)	O(11)#3-Ba(1)-O(10)	130.1(4)
Pb(2)-O(4)#2	2.454(14)	O(3)-Ba(1)-O(10)	126.4(4)
Pb(2)-O(2)	2.602(13)	O(7)-Ba(1)-O(10)	62.6(4)
Pb(2)-O(2)#2	2.602(13)	O(8)#4-Ba(1)-O(10)	127.7(4)
Pb(2)-O(7)#2	2.686(14)	O(4)#2-Ba(1)-O(10)	69.6(4)
Pb(2)-O(7)	2.686(14)	O(1)#5-Ba(1)-O(10)	67.9(3)
Ba(1)-O(11)#3	2.525(12)	O(11)#3-Ba(1)-O(3)#5	46.1(4)
Ba(1)-O(3)	2.600(13)	O(3)-Ba(1)-O(3)#5	117.0(5)
Ba(1)-O(7)	2.727(13)	O(7)-Ba(1)-O(3)#5	140.1(4)
Ba(1)-O(8)#4	2.800(16)	O(8)#4-Ba(1)-O(3)#5	107.9(4)
Ba(1)-O(4)#2	2.894(12)	O(4)#2-Ba(1)-O(3)#5	71.3(4)
Ba(1)-O(1)#5	2.902(11)	O(1)#5-Ba(1)-O(3)#5	64.8(3)
Ba(1)-O(10)	3.091(14)	O(10)-Ba(1)-O(3)#5	86.9(3)
Ba(1)-O(3)#5	3.220(14)	O(11)#3-Ba(1)-O(12)	125.3(4)
Ba(1)-O(12)	3.237(16)	O(3)-Ba(1)-O(12)	63.1(4)

Ba(2)-O(12)	2.641(14)	O(7)-Ba(1)-O(12)	64.8(4)
Ba(2)-O(12)#6	2.641(14)	O(8)#4-Ba(1)-O(12)	126.0(4)
Ba(2)-O(1)#5	2.751(12)	O(4)#2-Ba(1)-O(12)	128.2(3)
Ba(2)-O(1)#1	2.751(12)	O(1)#5-Ba(1)-O(12)	62.3(3)
Ba(2)-O(10)	2.785(14)	O(10)-Ba(1)-O(12)	64.2(4)
Ba(2)-O(10)#6	2.785(14)	O(3)#5-Ba(1)-O(12)	126.0(3)
Ba(2)-O(5)#5	3.130(13)	O(12)-Ba(2)-O(12)#6	179.998(1)
Ba(2)-O(5)#1	3.130(13)	O(12)-Ba(2)-O(1)#5	72.5(4)
Zn(1)-O(5)	1.914(12)	O(12)#6-Ba(2)-O(1)#5	107.5(4)
Zn(1)-O(6)	1.936(11)	O(12)-Ba(2)-O(1)#1	107.5(4)
Zn(1)-O(6)#7	1.953(12)	O(12)#6-Ba(2)-O(1)#1	72.5(4)
Zn(1)-O(9)	1.960(12)	O(1)#5-Ba(2)-O(1)#1	180.0(6)
Zn(2)-O(3)	1.914(13)	O(12)-Ba(2)-O(10)	76.6(4)
Zn(2)-O(12)	1.940(13)	O(12)#6-Ba(2)-O(10)	103.4(4)
Zn(2)-O(1)	1.951(12)	O(1)#5-Ba(2)-O(10)	74.5(4)
Zn(2)-O(9)	1.967(12)	O(1)#1-Ba(2)-O(10)	105.5(4)
Zn(3)-O(11)	1.903(12)	O(12)-Ba(2)-O(10)#6	103.4(4)
Zn(3)-O(8)	1.919(14)	O(12)#6-Ba(2)-O(10)#6	76.6(4)
Zn(3)-O(4)	2.029(13)	O(1)#5-Ba(2)-O(10)#6	105.5(4)
Zn(3)-O(2)	2.031(12)	O(1)#1-Ba(2)-O(10)#6	74.5(4)
B(1)-O(9)#8	1.33(2)	O(10)-Ba(2)-O(10)#6	179.998(2)

B(1)-O(1)	1.37(2)	O(12)-Ba(2)-O(5)#5	133.3(4)
B(1)-O(6)#7	1.38(2)	O(12)#6-Ba(2)-O(5)#5	46.7(4)
B(2)-O(7)	1.33(2)	O(1)#5-Ba(2)-O(5)#5	69.4(3)
B(2)-O(2)#8	1.40(2)	O(1)#1-Ba(2)-O(5)#5	110.6(3)
B(2)-O(4)#4	1.41(2)	O(10)-Ba(2)-O(5)#5	116.7(3)
B(3)-O(5)#1	1.36(2)	O(10)#6-Ba(2)-O(5)#5	63.3(3)
B(3)-O(12)	1.36(3)	O(12)-Ba(2)-O(5)#1	46.7(4)
B(3)-O(10)#8	1.39(2)	O(12)#6-Ba(2)-O(5)#1	133.3(4)
B(4)-O(8)#8	1.30(3)	O(1)#5-Ba(2)-O(5)#1	110.6(3)
B(4)-O(11)	1.37(3)	O(1)#1-Ba(2)-O(5)#1	69.4(3)
B(4)-O(3)#11	1.42(3)	O(10)-Ba(2)-O(5)#1	63.3(3)
O(10)-Pb(1)-O(7)	85.7(5)	O(10)#6-Ba(2)-O(5)#1	116.7(3)
O(10)-Pb(1)-O(5)#1	83.4(5)	O(5)#5-Ba(2)-O(5)#1	180.0(4)
O(7)-Pb(1)-O(5)#1	79.4(5)	O(5)-Zn(1)-O(6)	120.0(5)
O(10)-Pb(1)-O(2)	82.1(4)	O(5)-Zn(1)-O(6)#7	100.7(5)
O(7)-Pb(1)-O(2)	86.5(5)	O(6)-Zn(1)-O(6)#7	110.6(3)
O(5)#1-Pb(1)-O(2)	160.4(4)	O(5)-Zn(1)-O(9)	114.1(5)
O(4)-Pb(2)-O(4)#2	93.9(6)	O(6)-Zn(1)-O(9)	104.0(5)
O(4)-Pb(2)-O(2)	72.3(4)	O(6)#7-Zn(1)-O(9)	106.9(5)
O(4)#2-Pb(2)-O(2)	56.2(4)	O(3)-Zn(2)-O(12)	107.3(6)
O(4)-Pb(2)-O(2)#2	56.2(4)	O(3)-Zn(2)-O(1)	116.8(5)

O(4)#2-Pb(2)-O(2)#2	72.3(4)	O(12)-Zn(2)-O(1)	106.7(6)
O(2)-Pb(2)-O(2)#2	101.9(6)	O(3)-Zn(2)-O(9)	110.5(6)
O(4)-Pb(2)-O(7)#2	82.5(4)	O(12)-Zn(2)-O(9)	104.4(6)
O(4)#2-Pb(2)-O(7)#2	147.9(4)	O(1)-Zn(2)-O(9)	110.4(5)
O(2)-Pb(2)-O(7)#2	147.7(4)	O(11)-Zn(3)-O(8)	117.0(6)
O(2)#2-Pb(2)-O(7)#2	79.4(4)	O(11)-Zn(3)-O(4)	109.6(6)
O(4)-Pb(2)-O(7)	147.9(4)	O(8)-Zn(3)-O(4)	104.9(6)
O(4)#2-Pb(2)-O(7)	82.5(4)	O(11)-Zn(3)-O(2)	108.3(5)
O(2)-Pb(2)-O(7)	79.4(4)	O(8)-Zn(3)-O(2)	119.5(6)
O(2)#2-Pb(2)-O(7)	147.7(4)	O(4)-Zn(3)-O(2)	94.7(5)
O(7)#2-Pb(2)-O(7)	116.7(6)	O(9)#8-B(1)-O(1)	119.2(15)
O(11)#3-Ba(1)-O(3)	76.7(4)	O(9)#8-B(1)-O(6)#7	121.7(16)
O(11)#3-Ba(1)-O(7)	165.2(4)	O(1)-B(1)-O(6)#7	119.0(14)
O(3)-Ba(1)-O(7)	102.1(4)	O(7)-B(2)-O(2)#8	124.4(17)
O(11)#3-Ba(1)-O(8)#4	88.5(4)	O(7)-B(2)-O(4)#4	119.3(16)
O(3)-Ba(1)-O(8)#4	91.7(4)	O(2)#8-B(2)-O(4)#4	116.3(16)
O(7)-Ba(1)-O(8)#4	76.7(4)	O(5)#1-B(3)-O(12)	118.0(17)
O(11)#3-Ba(1)-O(4)#2	101.9(4)	O(5)#1-B(3)-O(10)#8	124.3(18)
O(3)-Ba(1)-O(4)#2	160.5(4)	O(12)-B(3)-O(10)#8	117.8(18)
O(7)-Ba(1)-O(4)#2	74.2(4)	O(8)#8-B(4)-O(11)	122(2)
O(8)#4-Ba(1)-O(4)#2	68.8(4)	O(8)#8-B(4)-O(3)#11	123.9(19)

O(11)#3-Ba(1)-O(1)#5	75.9(4)	O(11)-B(4)-O(3)#11	113.7(18)
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Symmetry transformations used to generate equivalent atoms:

#1 $-x+1/2, -y+3/2, -z+1$ #2 $-x+1, y, -z+3/2$

#3 $x, -y, z-1/2$ #4 $-x+1, y+1, -z+3/2$

#5 $x, y-1, z$ #6 $-x+1/2, -y+1/2, -z+1$

#7 $-x+1/2, y+1/2, -z+1/2$ #8 $x, y+1, z$

#9 $-x+1, y-1, -z+3/2$ #10 $x, -y, z+1/2$

#11 $x, -y+1, z+1/2$ #12 $x, -y+1, z-1/2$

#13 $-x+1/2, y-1/2, -z+1/2$

Table S3. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for α -BaZn₂(BO₃)₂.

Atom	x	y	z	<i>U</i> (eq)	BVS
Ba(1)	-628(1)	10398(1)	1591(1)	12(1)	1.917
Zn(1)	2904(1)	10528(1)	1790(1)	12(1)	2.084
Zn(2)	-1398(1)	11036(1)	4490(1)	13(1)	1.999
B(1)	1814(5)	5737(11)	678(4)	14(1)	2.944
B(2)	-590(5)	5811(11)	3587(4)	13(1)	2.964
O(1)	2260(4)	4203(6)	1702(3)	20(1)	1.997
O(2)	-297(3)	8484(6)	3954(2)	14(1)	2.002
O(3)	1832(3)	4691(6)	-366(3)	14(1)	1.908
O(4)	1377(3)	8331(6)	726(2)	13(1)	2.068
O(5)	4762(3)	9611(6)	1932(3)	20(1)	2.013
O(6)	-1710(3)	4483(6)	3699(3)	18(1)	1.922

Table S4. Selected bond distances (Å) and angles (deg) for α - $\text{BaZn}_2(\text{BO}_3)_2$.

Ba(1)-O(6)#1	2.651(3)	O(4)#3-Ba(1)-O(2)	169.65(9)
Ba(1)-O(5)#2	2.686(3)	O(4)-Ba(1)-O(2)	114.19(8)
Ba(1)-O(4)#3	2.711(3)	O(3)#3-Ba(1)-O(2)	130.96(8)
Ba(1)-O(4)	2.814(3)	O(6)#1-Ba(1)-O(3)#4	64.48(9)
Ba(1)-O(3)#3	2.885(3)	O(5)#2-Ba(1)-O(3)#4	167.04(9)
Ba(1)-O(2)	2.927(3)	O(4)#3-Ba(1)-O(3)#4	76.50(9)
Ba(1)-O(3)#4	2.959(3)	O(4)-Ba(1)-O(3)#4	74.16(9)
Ba(1)-O(5)#5	3.328(3)	O(3)#3-Ba(1)-O(3)#4	115.99(10)
Zn(1)-O(5)	1.898(3)	O(2)-Ba(1)-O(3)#4	95.53(8)
Zn(1)-O(1)#7	1.926(3)	O(6)#1-Ba(1)-O(5)#5	88.98(9)
Zn(1)-O(4)	1.956(3)	O(5)#2-Ba(1)-O(5)#5	110.53(11)
Zn(1)-O(1)#2	2.011(3)	O(4)#3-Ba(1)-O(5)#5	132.86(9)
Zn(2)-O(3)#8	1.914(3)	O(4)-Ba(1)-O(5)#5	78.58(8)
Zn(2)-O(6)#7	1.929(3)	O(3)#3-Ba(1)-O(5)#5	170.15(8)
Zn(2)-O(2)	1.955(3)	O(2)-Ba(1)-O(5)#5	43.75(8)
Zn(2)-O(2)#9	2.060(3)	O(3)#4-Ba(1)-O(5)#5	61.15(8)
B(1)-O(4)	1.369(6)	O(5)-Zn(1)-O(1)#7	122.79(15)
B(1)-O(3)	1.377(6)	O(5)-Zn(1)-O(4)	117.20(14)

B(1)-O(1)	1.387(6)	O(1)#7-Zn(1)-O(4)	108.39(13)
B(2)-O(5)#5	1.361(6)	O(5)-Zn(1)-O(1)#2	104.59(14)
B(2)-O(6)	1.370(6)	O(1)#7-Zn(1)-O(1)#2	102.68(9)
B(2)-O(2)	1.395(6)	O(4)-Zn(1)-O(1)#2	96.35(12)
O(6)#1-Ba(1)-O(5)#2	107.07(10)	O(3)#8-Zn(2)-O(6)#7	102.81(13)
O(6)#1-Ba(1)-O(4)#3	90.80(9)	O(3)#8-Zn(2)-O(2)	125.92(13)
O(5)#2-Ba(1)-O(4)#3	114.47(10)	O(6)#7-Zn(2)-O(2)	114.68(14)
O(6)#1-Ba(1)-O(4)	137.64(9)	O(3)#8-Zn(2)-O(2)#9	116.31(12)
O(5)#2-Ba(1)-O(4)	115.24(9)	O(6)#7-Zn(2)-O(2)#9	106.65(13)
O(4)#3-Ba(1)-O(4)	70.38(10)	O(2)-Zn(2)-O(2)#9	89.31(12)
O(6)#1-Ba(1)-O(3)#3	81.48(9)	O(4)-B(1)-O(3)	120.5(4)
O(5)#2-Ba(1)-O(3)#3	70.37(10)	O(4)-B(1)-O(1)	118.8(4)
O(4)#3-Ba(1)-O(3)#3	50.31(8)	O(3)-B(1)-O(1)	120.7(4)
O(4)-Ba(1)-O(3)#3	110.21(9)	O(5)#5-B(2)-O(6)	121.2(4)
O(6)#1-Ba(1)-O(2)	79.70(9)	O(5)#5-B(2)-O(2)	117.8(4)
O(5)#2-Ba(1)-O(2)	72.71(10)	O(6)-B(2)-O(2)	120.9(4)

Symmetry transformations used to generate equivalent atoms:

#1 $-x-1/2, y+1/2, -z+1/2$ #2 $-x+1/2, y+1/2, -z+1/2$

#3 $-x, -y+2, -z$ #4 $-x, -y+1, -z$ #5 $-x+1/2, y-1/2, -z+1/2$

#6 $-x-1/2, y-1/2, -z+1/2$ #7 $x, y+1, z$

#8 $x-1/2, -y+3/2, z+1/2$ #9 $-x, -y+2, -z+1$

#10 $x, y-1, z$ #11 $x+1/2, -y+3/2, z-1/2$

Table S5. The basic information of the existed anhydrous-zincoborates.^[d]

No.	Chemical formula	Coordination of the Zn atom	No .	Chemical formula	Coordination of the Zn atom
1	Ba ₂ Zn(BO ₃) ₂	isolated ZnO ₄	31	NdZnB ₅ O ₁₀	isolated Zn ₂ O ₁₀
2	Ba ₂ ZnSc(BO ₃) ₃	isolated ZnO ₄	32	TbZnB ₅ O ₁₀	isolated Zn ₂ O ₁₀
3	Ba ₃ Zn(BO ₃)(B ₂ O ₅)F	isolated ZnO ₄	33	EuZn(BO ₂) ₅	isolated Zn ₂ O ₁₀
4	Ba ₃ Zn ₂ B ₃ O ₉ F	isolated ZnO ₄	34	Zn ₃ B ₇ O ₁₃ Cl	isolated Zn ₃ O ₁₂ Cl
5	Ba ₅ Zn ₂ B ₄ O ₁₂ F ₂	isolated ZnO ₄	35	Zn ₃ B ₇ O ₁₃ Br	isolated Zn ₃ O ₁₂ Br
6	Bi ₂ ZnB ₂ O ₇	isolated ZnO ₄	36	Zn ₄ (B ₆ O ₁₂)O	isolated Zn ₄ O ₁₃
7	K ₃ ZnB ₅ O ₁₀	isolated ZnO ₄	37	Zn ₈ (BO ₂) ₁₂ Se ₂	isolated Zn ₄ O ₁₂ Se
8	K ₂ NaZnB ₅ O ₁₀	isolated ZnO ₄	38	KBa₂Zn₃(B₃O₆)(B₆O₁₃)	isolated Zn₂O₇ and isolated Zn₂O₆
9	Li ₆ Zn ₃ (BO ₃) ₄	isolated ZnO ₄	39	K₂Ba₄Zn₅(B₃O₆)₃(B₉O₁₉)	isolated Zn₂O₇ and isolated Zn₃O₁₀
10	Na ₃ ZnB ₅ O ₁₀ - I	isolated ZnO ₄	40	<i>α</i>-Pb₂Ba₄Zn₄B₁₄O₃₁	isolated ZnO₄ and isolated Zn₂O₆
11	Na ₃ ZnB ₅ O ₁₀ - II	isolated ZnO ₄	41	<i>β</i>-Pb₂Ba₄Zn₄B₁₄O₃₁	isolated ZnO₄ and isolated Zn₂O₆
12	Pb ₄ Zn ₂ B ₁₀ O ₂₁	isolated ZnO ₄	42	<i>γ</i>-Pb₂Ba₄Zn₄B₁₄O₃₁	isolated ZnO₄ and isolated Zn₂O₆

13	$\text{Rb}_3\text{ZnB}_5\text{O}_{10}$	isolated ZnO_4	43	LiZnBO_3 - II	1D Zn_2O_6 chain
14	$\text{Cs}_{12}\text{Zn}_4(\text{B}_5\text{O}_{10})_4$	isolated ZnO_4	44	LiZnBO_3 -III	1D Zn_2O_6 chain
15	$\text{ZnV}_{12}\text{B}_{18}\text{O}_{63}$	isolated ZnO_4	45	$\text{Ba}_4\text{Zn}_2(\text{BO}_3)_2(\text{B}_2\text{O}_5)\text{F}_2$	1D Zn_2O_7 chain
16	$\text{Zn}(\text{B}_4\text{O}_7)$ - I	isolated ZnO_4	46	$\text{PbZn}_2(\text{BO}_3)_2$	1D Zn_2O_7 chain
17	BaZnBO_3F	isolated ZnO_3F_2	47	$\text{Ba}_4\text{Zn}_5\text{Sc}_2(\text{BO}_3)_8$	1D Zn_3O_{11} chain
18	$\text{K}_7\text{ZnSc}_2\text{B}_{15}\text{O}_{30}$	isolated ZnO_6	48	$\text{Ba}_2\text{ZnSc}(\text{BO}_3)_3$	isolated ZnO_4
19	$\text{Zn}(\text{B}_4\text{O}_7)$ - II	isolated ZnO_5	49	$\text{BaZn}_2(\text{BO}_3)_2$	1D Zn_4O_{12} chain
20	$\text{Ba}_2\text{Zn}(\text{B}_3\text{O}_6)_2$	isolated Zn_2O_6	50	$\text{CsZn}_2\text{B}_3\text{O}_7$	2D layer
21	KZnB_3O_6	isolated Zn_2O_6	51	$\text{CsZn}_4(\text{BO}_3)_3$	2D layer
22	$\text{KCdZn}_2\text{B}_2\text{O}_6\text{F}$	isolated $\text{Zn}_2\text{O}_6\text{F}$	52	$\text{Cs}_3\text{Zn}_6\text{B}_9\text{O}_{21}$	2D layer
23	$\text{Ba}_5\text{Zn}_4(\text{BO}_3)_6$	isolated Zn_2O_7	53	$\text{KZn}_4\text{B}_3\text{O}_9$	2D layer
24	LiZnBO_3 - I	isolated Zn_2O_7	54	$\text{KZn}_2\text{BO}_3\text{Cl}_2$	2D layer
25	$\text{Na}_2\text{Ba}_4\text{Zn}_4(\text{B}_3\text{O}_6)_2(\text{B}_{12}\text{O}_{24})$	isolated Zn_2O_7	55	$\text{RbZn}_4(\text{BO}_3)_3$	2D layer
26	$\alpha\text{-Pb}_5\text{Zn}_4\text{B}_6\text{O}_{18}$	isolated Zn_2O_7	56	$\text{RbZn}_2\text{BO}_3\text{Cl}_2$	2D layer

27	β -Pb ₅ Zn ₄ B ₆ O ₁₈	isolated Zn ₂ O ₇	57	RbZn ₂ BO ₃ Br ₂	2D layer
28	γ -Pb ₅ Zn ₄ B ₆ O ₁₈	isolated Zn ₂ O ₇	58	Zn ₃ (BO ₃) ₂ - I	3D
29	CeZnB ₅ O ₁₀	isolated Zn ₂ O ₁₀	59	Zn ₃ (BO ₃) ₂ - II	3D
30	LaZnB ₅ O ₁₀	isolated Zn ₂ O ₁₀	60	Zn ₃ (BO ₃) ₂ -III	3D

[d]. Inorganic Crystal Structure Database (ICSD-3.7.0, the latest release of ICSD-2019/08/26)

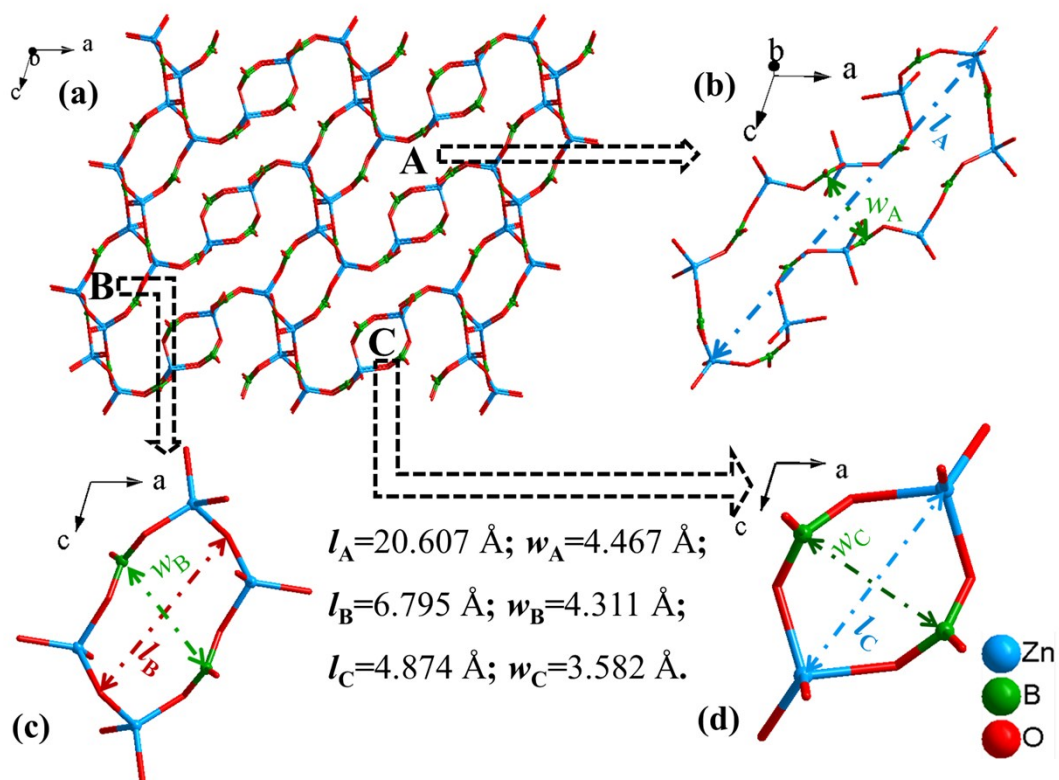


Figure S1. (a). Three types of tunnels in $\text{Pb}_3\text{Ba}_3\text{Zn}_6(\text{BO}_3)_8$; (b) A-type tunnel; (c) B-type tunnel; (d) C-type tunnel.

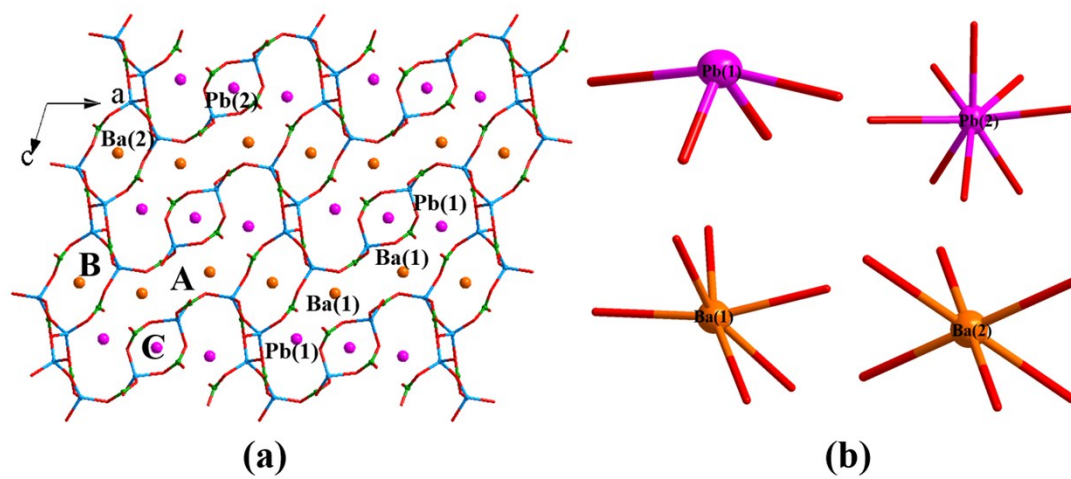


Figure S2. (a). Crystal structure of $\text{Pb}_3\text{Ba}_3\text{Zn}_6(\text{BO}_3)_8$ viewed along the b -axis; (b) Coordination environments of the Pb and the Ba atoms.

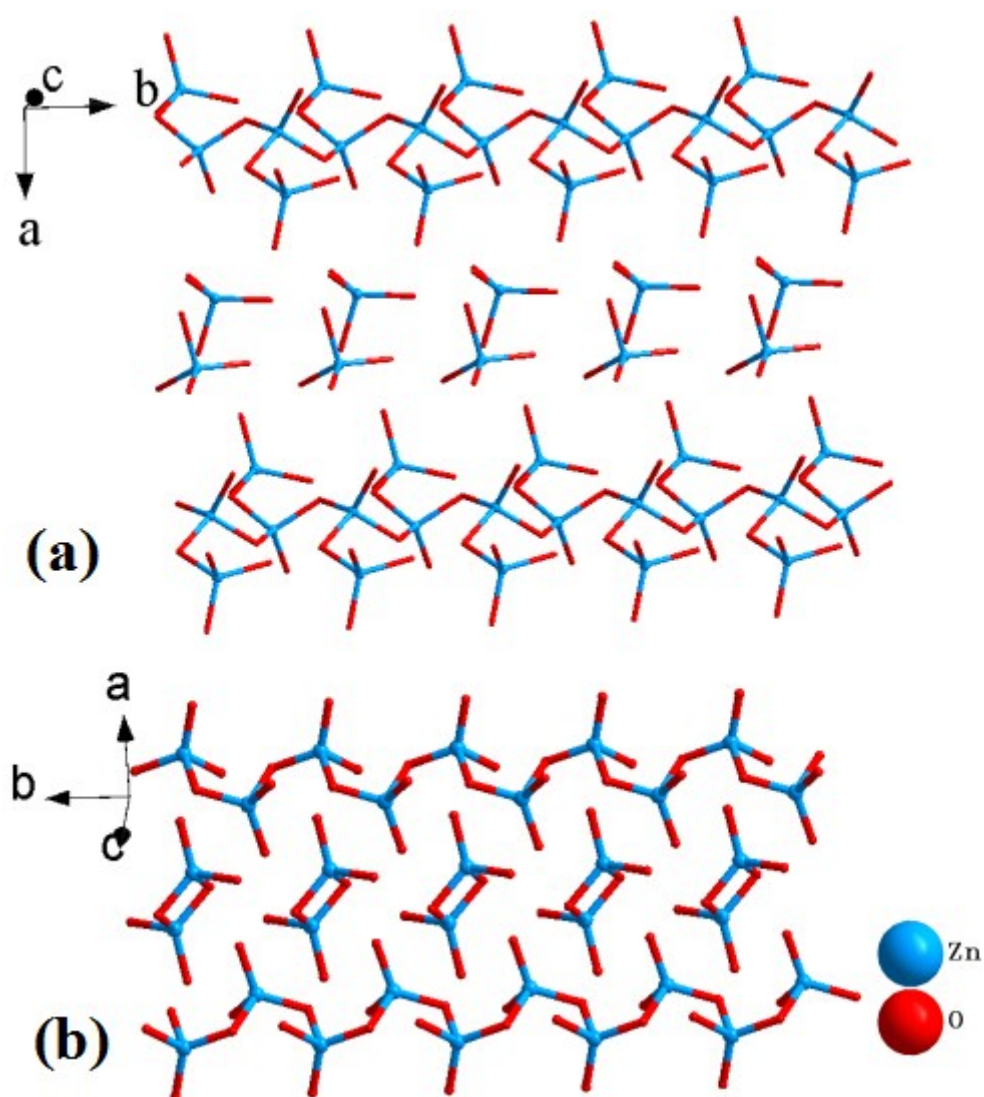


Figure S3. Coordination of the Zn atom: 0D isolated ZnO_4 and 1D $[\text{Zn}_4\text{O}_{12}]$ chain in $\text{Pb}_3\text{Ba}_3\text{Zn}_6(\text{BO}_3)_8$, (a); 0D isolated Zn_2O_6 dipolymer and 1D $[\text{Zn}_2\text{O}_7]$ chain in $\alpha\text{-BaZn}_2(\text{BO}_3)_2$, (b).

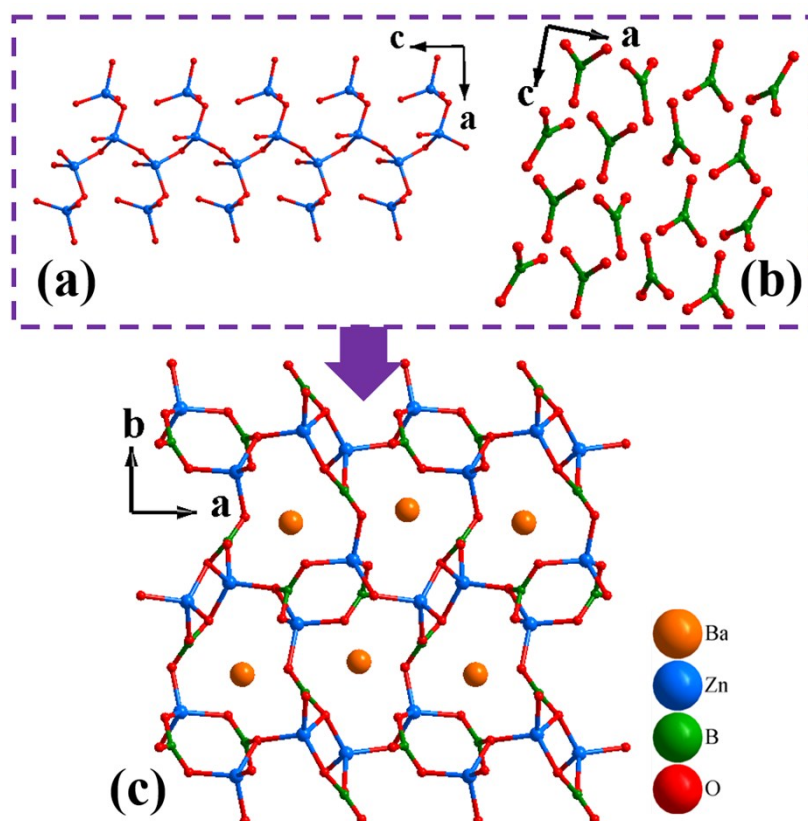


Figure S4 (a) [Zn₄O₁₂] chain; (b) isolated BO₃ triangles; (c) crystal structure of β -BaZn₂(BO₃)₂ viewing along the *c*-axis; All the Ba-O bonds are removed for clarity.

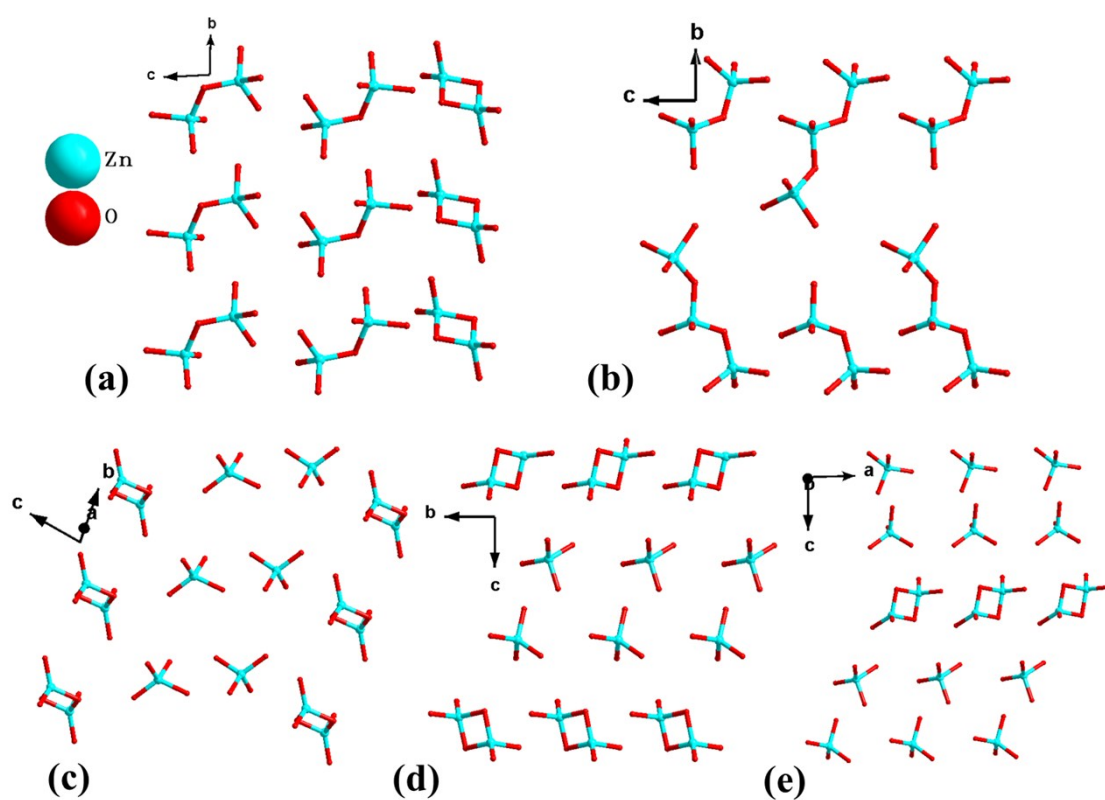


Figure S5. Coordination of the Zn atom: isolated Zn₂O₇ and Zn₂O₆ in KBa₂Zn₃(B₃O₆)(B₆O₁₃), (a); isolated Zn₂O₇ and Zn₃O₁₀ in K₂Ba₄Zn₅(B₃O₆)₃(B₉O₁₉), (b); isolated ZnO₄ and Zn₂O₆ in α -, β - and γ - Pb₂Ba₄Zn₄B₁₄O₃₁, (c)(d)&(e).

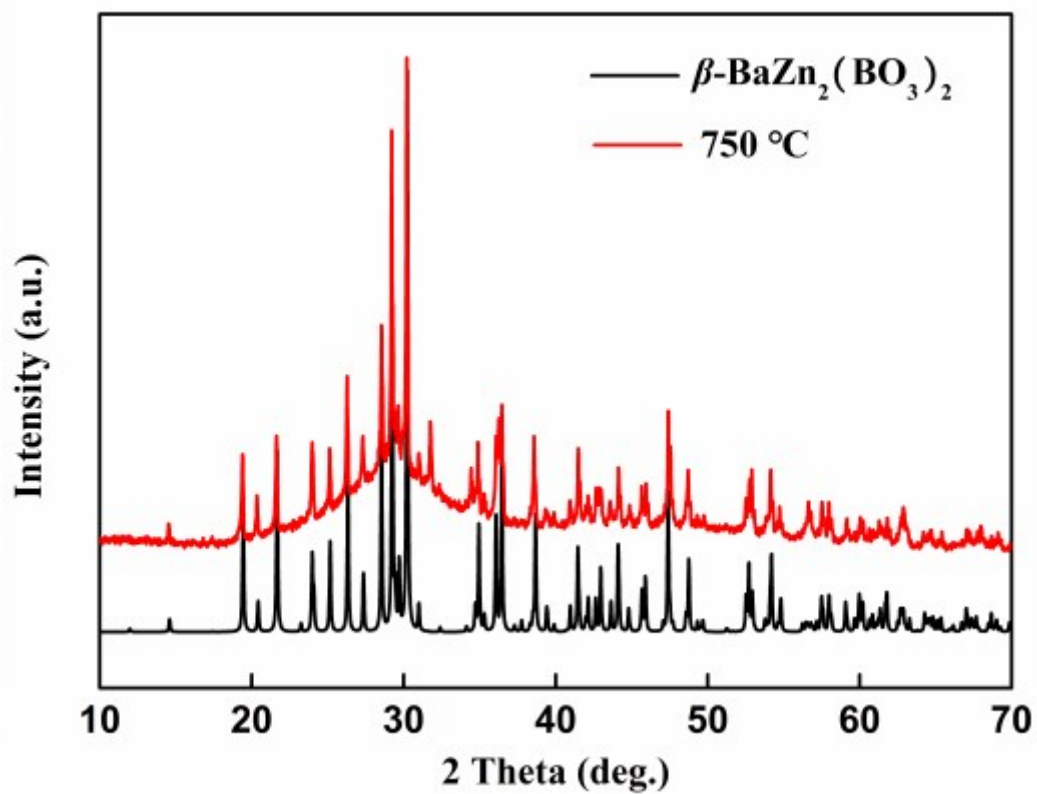


Figure S6. The powder XRD patterns of $\text{Pb}_3\text{Ba}_3\text{Zn}_6(\text{BO}_3)_8$ after holding at 750 °C.