## **Supplementary Information**

## Visible light photocatalytic water oxidation over complex perovskite $Sr_3BNb_2O_9$ (B = Mg, Ca and Sr) doped with nitrogen

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Figure S1. Thermogravimetric analysis (TGA) curves of  $Sr_3BNb_2O_{9-x}N_y$  (B = Mg, Ca and Sr) powders in air with heating rate of 10 K min<sup>-1</sup> from 30 °C to 1000 °C.

Figure S2. FT-IR spectra of  $Sr_3BNb_2O_{9-x}N_y$  (B = Mg, Ca and Sr) powders.

Figure S3. Transmission electron microscopic (TEM) images of  $Sr_3BNb_2O_{9-x}N_y$  (B =

Mg, Ca and Sr) powders: (a)  $Sr_3MgNb_2O_{9-x}N_y$ , (b)  $Sr_3CaNb_2O_{9-x}N_y$  and (c)  $Sr_3SrNb_2O_{9-x}N_y$ .

Figure S4. Repeated cycles of photocatalytic oxygen evolution for  $Sr_3SrNb_2O_{9-x}N_y$ loaded with 1 wt% Rh<sub>2</sub>O<sub>3</sub> under visible light illumination ( $\lambda \ge 400$  nm), sodium persulfate (0.05 M) was used as a sacrificial agent. Evacuation was performed every 2.5 hours.

Figure S5. XRD patterns of  $Sr_3SrNb_2O_{9-x}N_y$  before and after photocatalytic experiment.

Figure S6. High resolution TEM image of  $Sr_3SrNb_2O_{9-x}N_y$  loaded with 1 wt%  $Rh_2O_3$ , lattice fringe marked corresponds to (110) planes of  $Rh_2O_3$ .

Table S1 Effective chemical formula of  $Sr_3BNb_2O_{9-x}N_y$  (B = Mg, Ca and Sr) determined by thermogravimetric analysis (TGA)

Table S2 Cation percentage of  $Sr_3BNb_2O_{9-x}N_y$  (B = Mg, Ca and Sr) determined by energy dispersive X-ray spectroscopy (EDS), standard deviation is included in the parenthesis.



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Figure S4. Repeated cycles of photocatalytic oxygen evolution for Sr<sub>3</sub>SrNb<sub>2</sub>O<sub>9-x</sub>N<sub>y</sub>

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Compound	Effective chemical formula		
Sr <sub>3</sub> MgNb <sub>2</sub> O <sub>9-x</sub> N <sub>y</sub>	Sr <sub>3</sub> MgNb <sub>2</sub> O <sub>8.927</sub> N <sub>0.049</sub>		
Sr <sub>3</sub> CaNb <sub>2</sub> O <sub>9-x</sub> N <sub>y</sub>	Sr <sub>3</sub> CaNb <sub>2</sub> O <sub>8.912</sub> N <sub>0.059</sub>		
Sr <sub>3</sub> SrNb <sub>2</sub> O <sub>9-x</sub> N <sub>y</sub>	$Sr_{3}SrNb_{2}O_{8.891}N_{0.073}$		

Table S2 Cation percentage of  $Sr_3BNb_2O_{9-x}N_y$  (B = Mg, Ca and Sr) determined by energy dispersive X-ray spectroscopy (EDS), standard deviation is included in the parenthesis.

	Mg (at. %)	Ca (at. %)	Sr (at. %)	Nb (at. %)
Sr <sub>3</sub> MgNb <sub>2</sub> O <sub>9-x</sub> N <sub>y</sub>	16(1)		49(1)	35(1)
Sr <sub>3</sub> CaNb <sub>2</sub> O <sub>9-x</sub> N <sub>y</sub>		16(1)	50(1)	34(1)
Sr <sub>3</sub> SrNb <sub>2</sub> O <sub>9-x</sub> N <sub>y</sub>			67(1)	33(1)