Supporting Information

Metal-organic framework templated fabrication of Cu₇S₄@Ni(OH)₂ core-shell nanoarrays for high-performance supercapacitors

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Fig. S1. SEM image of Ni(OH)₂.



Fig. S2. FT-IR spectra of TCNQ, Cu-TCNQ and Cu₇S₄.



Fig. S3. EDS pattern of Cu₇S₄/CF.



Fig. S4. EDS pattern of Cu_7S_4 @Ni(OH)₂/CF.



Fig. S5. PXRD pattern of Cu-TCNQ scratched from CF.



Fig. S6. GCD curves of Ni(OH)₂/CF at different current densities.



Fig. S7. GCD curves of Cu_7S_4/CF at different current densities.



Fig. S8. Comparison of CV curves of $Cu_7S_4@Ni(OH)_2/CF$ electrode before and after 10000 cycles at scan rate of 30 mV s⁻¹.



Fig. S9. SEM image of the Cu_7S_4 @Ni(OH)₂/CF electrode after 10000 cycles.



Fig. S10. Magnified Nyquist plot.



Fig. S11. (a) N₂ adsorption-desorption isotherms; (b) Pore size distribution of Cu₇S₄@Ni(OH)₂.



Fig. S12. (a) CV curves of AC at different scan rates; (b) GCD curves of AC at different current densities; (c) the magnified EIS plot of AC



Fig. S13. Comparison EIS plots of the Cu₇S₄@Ni(OH)₂//AC ASC before and after 10000 cycles.



Fig. S14. Comparison CV curves of the Cu₇S₄@Ni(OH)₂//AC ASC before and after 10000 cycles.



Fig. S15. Schematic image of assembly of the Cu₇S₄@Ni(OH)₂//AC ASC device and its LED illuminated time.

Table S1 Content of Cu and Ni obtained from ICP-AES

	Cu	Ni
Cu ₇ S ₄	74.23%	/
Ni(OH) ₂	/	61.14%
Cu ₇ S ₄ @Ni(OH) ₂	54.28%	15.62%