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Supporting information

Fabrication of redox-active polyoxometalates-based ionic crystals onto singlewalled carbon nanotubes for high-performance anode materials of lithium-ion batteries

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Figure S1. FT-IR spectra of PMo₁₂O₄₀, Co₃ cluster, and Composite 1, respectively.



Fig. S2. Raman spectra of $PMo_{12}O_{40}$.



Figure S3. XRD spectra of Composite 1, SWNTs, and Composite 1/SWNTs nanocomposite.



Figure S4. HRTEM-EDX pattern of Composite 1/SWNTs nanocomposite.



Figure S5. (a) The discharge/charge curves of the Composite 1 at 100 mA g⁻¹, (b) cyclic performance of Composite 1, PMo₁₂O₄₀ and Co₃ cluster, (c) rate capability of Composite 1, and (d) CV of Composite 1 at a scan rate of 0.1 mV s⁻¹ within 0-3 V.



Figure S6. Nyquist plots of Composite 1, PMo₁₂O₄₀, and Co₃ cluster. (Inserted) The simulated equivalent circuit model of the electrode/electrolyte interface.

Electrode Materials	Current density	Reversible capacity (cycle times) /mAh g ⁻¹	Ref.
PMo ₁₀ V ₂ /PDA	100 mA g ⁻¹	915.3 (63)	1
TBA-PMo ₁₁ V/CNTs	0.5 mA cm ⁻²	850 (100)	2
CNTs–SiW ₁₁	0.2 mA cm ⁻²	650 (100)	3
SWNT-sFe ₃ O ₄ /CMC	450 mAg ⁻¹	687 (100)	4
p-SWNT/GNS	200 mAg ⁻¹		5
N-carbon/rGO	0.1 mAg ⁻¹	669 (200)	6
Py-Anderson-CNTs	0.5 mAcm ⁻²	665.3 (100)	7
GO-IL-P ₂ Mo ₁₈	100 mAg ⁻¹	973 (100)	8
GQD/metal oxide composites	100 mAg ⁻¹	970 (100)	9
TBA ₄ [Py-SiW ₁₁]-SWNTs	0.5 mAcm ⁻²	580 (100)	10
PMo ₁₂ -PPy/RGO	100 mAg ⁻¹	1000 (50)	11
Composite 1/SWNTs	100 mAg ⁻¹	1012 (100)	This work

Table S1. Comparison of different anode materials and their LIBs performance

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