

Semi-sacrificial template-assisted synthesis of Ni single atoms on nitrogen-doped hollow carbon spheres as efficient and stable catalysts for CO₂ electroreduction

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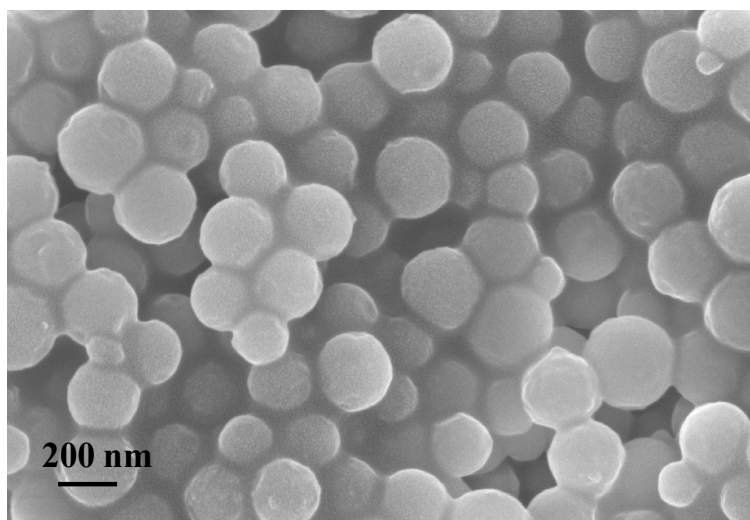


Fig. S1 SEM image of prepared SiO₂/polydopamine spheres templates.

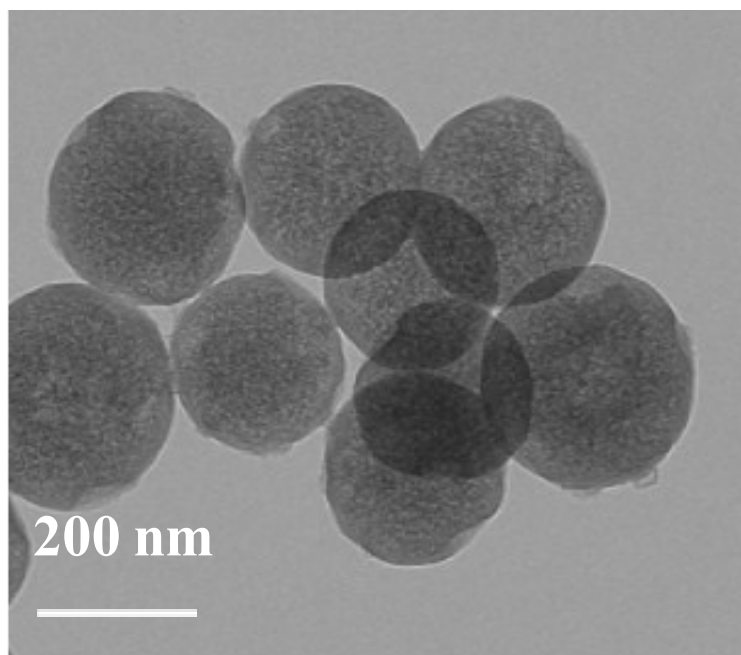


Fig. S2 TEM image of prepared SA-Ni/N-CS single atoms catalysts.

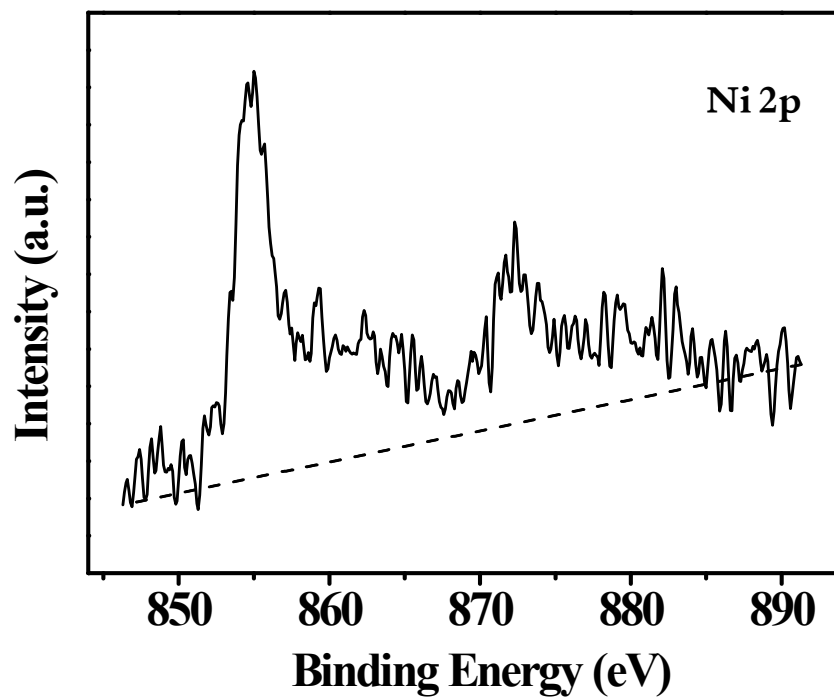


Fig. S3 The high-resolution Ni 2p XPS spectrum of the obtained SA-Ni/N-CS catalysts.

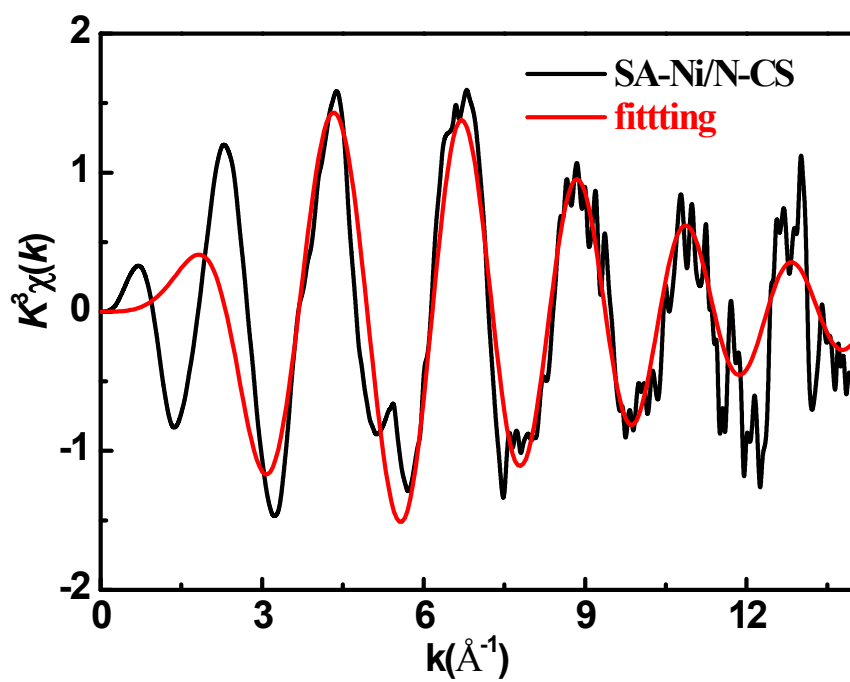


Fig. S4 Ni K-edge EXAFS spectrum and fitting result in k-space for the SA-Ni/N-CS sample.

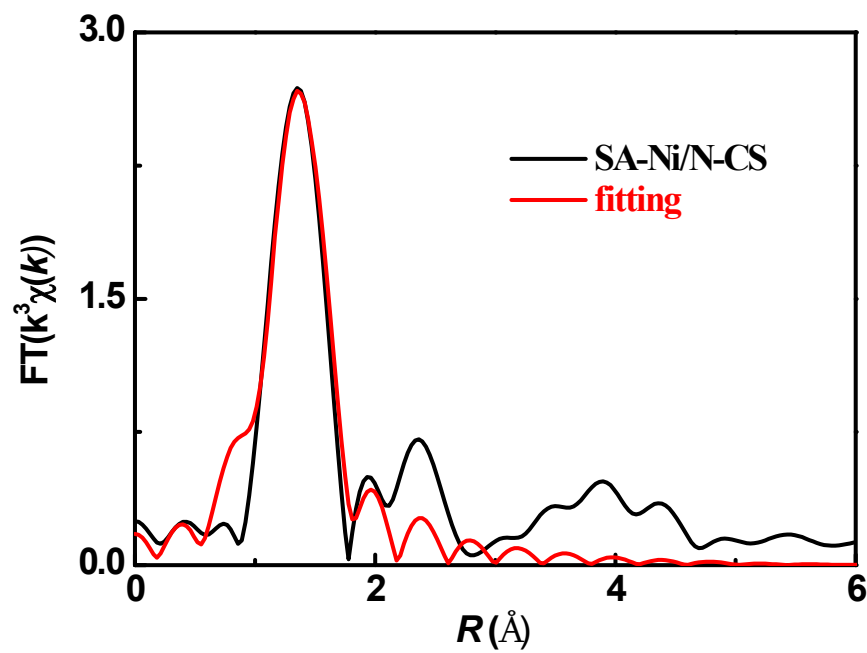


Fig. S5 Ni K-edge EXAFS spectrum and fitting result in q-space for the SA-Ni/N-CS sample.

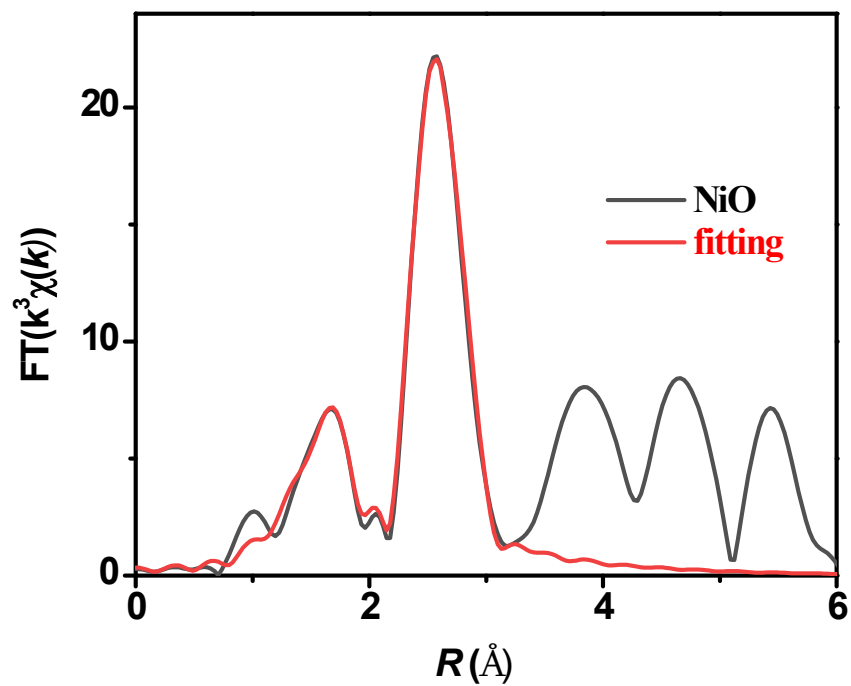


Fig. S6 Corresponding EXAFS fitting curves for NiO sample.

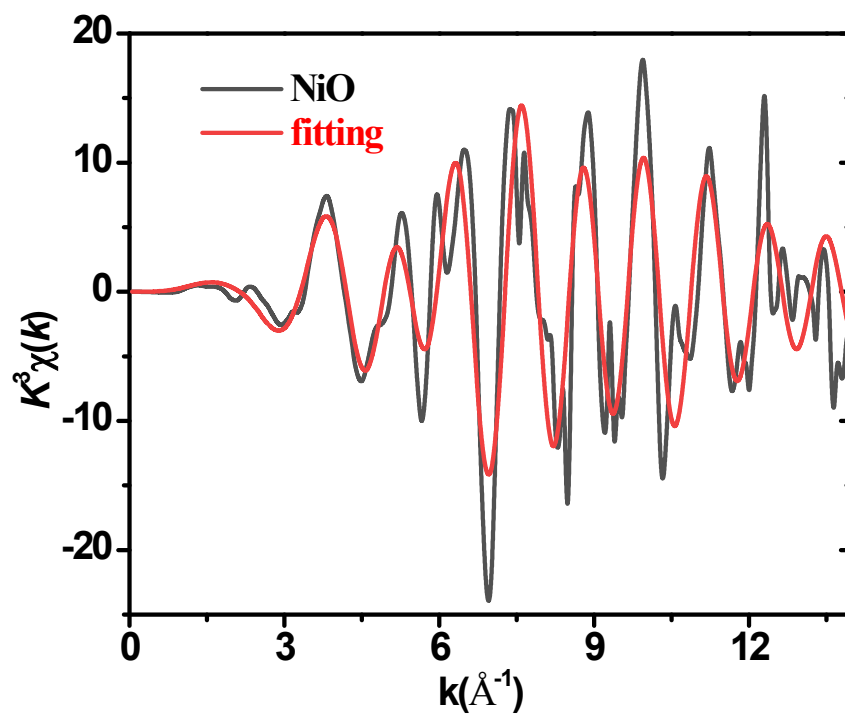


Fig. S7 Ni K-edge EXAFS spectrum and fitting result in k-space for the NiO sample.

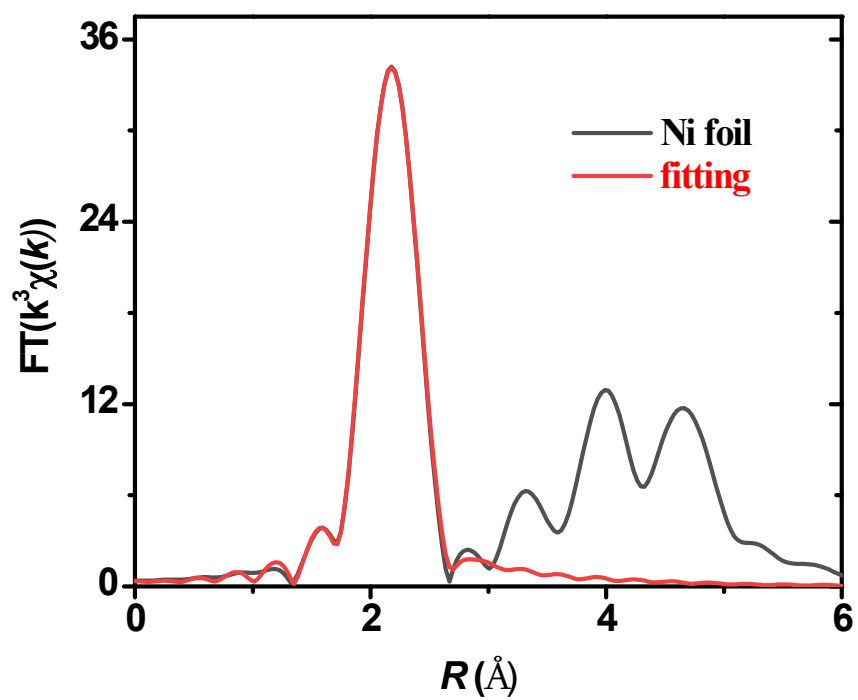


Fig. S8 Corresponding EXAFS fitting curves for Ni foil sample.

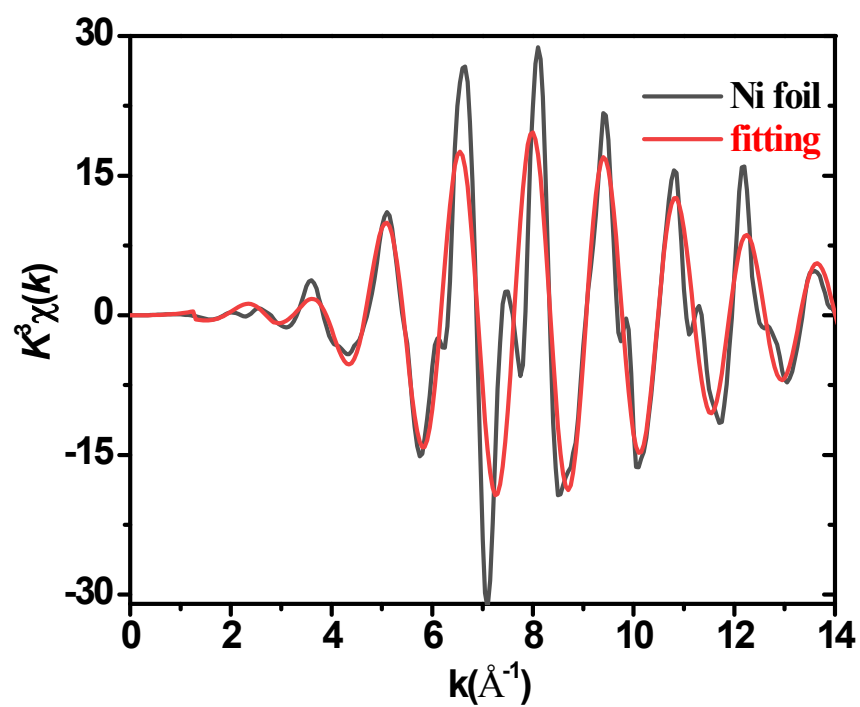


Fig. S9 Ni K-edge EXAFS spectrum and fitting result in k-space for the Ni foil sample.

Table S1 Structural parameters of catalysts extracted from the EXAFS fitting. ($S_0^2=0.85$)

Sample	Scattering pair	CN	R(Å)	$\sigma^2(10^{-3}\text{Å}^2)$	$\Delta E_0(\text{eV})$	R factor
Ni foil	Ni-Ni	12.0±0.1	2.49±0.01	5.8±0.3	4.5±0.5	0.01
NiO	Ni-O	6.0±0.1	2.09±0.01	4.1±0.5	5.8±0.4	0.01
	Ni-Ni	12.0±0.2	2.95±0.01	6.0±0.3	8.0±0.05	
Ni	Ni-N	3.7±0.6	1.89±0.02	6.7±1.4	-4.9±0.6	0.02

S_0^2 is the amplitude reduction factor; CN is the coordination number; R is interatomic distance (the bond length between central atoms and surrounding coordination atoms); σ^2 is Debye-Waller factor (a measure of thermal and static disorder in absorber-scatterer distances); ΔE_0 is edge-energy shift (the difference between the zero kinetic energy value of the sample and that of the theoretical model). R factor is used to value the goodness of the fitting. p is the fraction of Ni atoms.

Table S2 The zero point energy corrections and entropic contributions used in the related free energies calculations.

Species	States	ZPE/eV	TS/eV
CO ₂		0.267	0.661
COOH *	N-C	0.573	8.839×10 ⁻³
	Ni/N-C	0.528	0.013
CO *	N-C	0.129	1.23×10 ⁻⁵
	Ni/N-C	0.123	1.93×10 ⁻⁵
H ₂ O (l)		0.568	0.67
CO		0.132	0.611
H*	N-C	0.307	2.12×10 ⁻³
	Ni/N-C	0.311	2.557×10 ⁻³
H ₂		0.269	0.404