ELECTRONIC SUPPLEMENTARY INFORMATION



Fig. S1 Three-dimensional, wired structure of W8 showing strained double bond.



Fig. S2 PESA spectrum of thin film of **W8**. The dashed-lines show the fits to extract ionisation potential (-5.73 eV) which corresponds to the HOMO energy level.

Details of DFT Calculations

The Gaussian 09 *ab initio*/DFT quantum chemical simulation package was employed to get results described in the present work.^{S1} The geometry optimization of **W8** has been carried out at the B3LYP/6-31G(d) level of theory. To ensure the structures to be real, frequency calculations were carried out. Furthermore, the geometries of **W8** obtained at the B3LYP/6-31G(d) level were subjected to the time-dependent density functional theory (TD-DFT) studies using the M062X/6-31G(d).^{S2,S3} TD-DFT results obtained for **W8** are reported in Table S1. From the TD-DFT results it is seen that **W8** shows two major absorption peaks at 445 nm and 321 nm. The frontier molecular orbitals (FMOs) were generated using Avogadro.^{S4,S5}

Material	Excitation	Excitation	Oscillator	D	% contribution for
	Energy	Wavelength	Strength	Excitations	transition
	(eV)	(nm)	(f)		
W8	2.7858	445.05	0.5010	$492 \rightarrow 497$	$H-4 \rightarrow LUMO (13\%)$
				$492 \rightarrow 498$	$H-4 \rightarrow L+1 (3\%)$
				$493 \rightarrow 497$	$H-3 \rightarrow LUMO (21\%)$
				$494 \rightarrow 498$	$H-2 \rightarrow L+1 (4\%)$
				$495 \rightarrow 498$	$\text{H-1} \rightarrow \text{L+1} (2\%)$
				$496 \rightarrow 497$	HOMO \rightarrow LUMO (30%)
				$496 \rightarrow 498$	$HOMO \rightarrow L+1(14\%)$

Table S1 Calculated TD-DFT excitation properties of W8.

3.8528	321.80	0.4864	$482 \rightarrow 497$	$H-14 \rightarrow LUMO(3\%)$
			$482 \rightarrow 498$	$H-14 \rightarrow L+1 (2\%)$
			$483 \rightarrow 497$	$\text{H-13} \rightarrow \text{LUMO} (22\%)$
			$484 \rightarrow 498$	$H-12 \rightarrow L+1 (16\%)$
			$485 \rightarrow 498$	$H-11 \rightarrow L+1 (27\%)$
			$486 \rightarrow 499$	$H-10 \rightarrow L+2 (4\%)$
			$487 \rightarrow 500$	$H-9 \rightarrow L+3 (7\%)$

Frontier molecular orbitals of W8 with energy levels in eV

W8		eV	
L+4	501	-0.67648	
L+3	500	-2.20849	

L+2	499	-2.23026	
L+1	498	-2.32605	
L	497	-2.35163	

Н	496	-6.37949	
H-1	495	-7.02141	
Н-2	494	-7.04345	

Н-3	493	-7.06059	
H-4	492	-7.16998	
H-5	487	-8.05463	

H-10	486	-8.07559	
H-11	485	-8.14117	
H-12	484	-8.15749	

H-13	483	-8.17981	
H-14	482	-8.20675	



Fig. S3 Computed absorption spectrum of W8 showing transition peaks at 445 nm and 321 nm.



Fig. S4 Thermogravimetric analysis curve showing thermal stability of W8.

Acceptor	Donor	Testing conditions (D: A) ^a	V _{oc} (V)	$\begin{bmatrix} J_{\rm sc} \\ ({\rm mA/cm^2}) \end{bmatrix}$	FF	Best PCE (%)	AveragePCE(%)(± std dev)
W8	PTB7	1: 1.2 (no annealing)	0.94	10.21	0.60	5.72	5.63 (± 0.06)
W8	PTB7	1: 1.2 (annealed)	1.04	13.41	0.62	8.58	8.51 (± 0.05)
W8	РЗНТ	1: 1.2 (annealed)	0.95	9.24	0.60	5.26	5.18 (± 0.06)
PC ₆₁ BM	РЗНТ	1: 1.2 ^b	0.57	8.28	0.64	3.03	2.99 (± 0.04)

Table S2. Photovoltaic cell parameters for W8 blends

^a BHJ devices with specified weight ratio. Device structure was ITO/PEDOT: PSS (38 nm)/active layer/Ca (20 nm)/Al (100 nm) with an active layer thickness of ~75 nm; ^b A standard P3HT: $PC_{61}BM$ device afforded 3.03% efficiency when tested under alike annealing conditions.



Fig. S5 TEM (left) and XRD (right) images for the active blend surfaces of **W8**. The TEM image shows an excellent intermixing of donor and acceptor domains (PTB7: **W8** 1: 1.2; scale bar = 200 nm), whereas XRD spectra indicate the blend surfaces to be amorphous.



Fig. S6 Optical microscopic images for the blend surfaces of **W8** (P3HT left and PTB7 right) showing fairly flat surfaces corroborating TEM and XRD analyses.



Active layer	Electron Mobility (cm²/Vm)
W8	1.20 x 10 ⁻³
PTB7: W8	4.61 x 10 ⁻³
P3HT: W8	2.17 x 10 ⁻³

Fig. S7 Current–voltage characteristics of electron only devices which were applied to Mott-Gurney equation to calculate electron mobilities of **W8** and its blends.

Experimental spectra



Fig. S8 FT-IR spectrum of compound 1.



Fig. S9 ¹H NMR spectrum of compound 1.



Fig. S10 ¹³C NMR spectrum of compound 1.







Fig. S12 MALDI-TOF spectrum of compound 1.



Fig. S13 FT-IR spectrum of W8.



Fig. S14 ¹H NMR spectrum of W8.







Fig. S16 MALDI-TOF spectrum of W8.

References:

S1 M. J. Frisch, et al., Gaussian 09, Revision C.01, Gaussian Inc., Wallingford CT, 2009.

S2 Y. Zhao, et al., Chem. Phys. Lett., 2011, 502, 1.

S3 D. Srivani, et al., Dyes Pigm., 2017, 143, 1.

S4 Avogadro: an open-source molecular builder and visualization tool, Version 1.1.0. http://avogadro.openmolecules.net/

S5 M. D. Hanwell, et al., J. Cheminf., 2012, 4, 17.