

Electronic Supplementary Information (ESI)

1. General

All reagents and starting materials were obtained from commercial suppliers and used without further purification. Anhydrous dichloromethane (DCM) were distilled under a nitrogen atmosphere over sodium and calcium hydride, respectively. Compound 3,10-dibromo-1,1-bis(4-tetradecylphenyl)-1H-cyclopenta[ghi]perylene (**1**) was synthesized according to the literature.^[1] Column chromatography was performed on silica gel 60 (Merck 40-60 nm, 230-400 mesh). All NMR spectra were recorded on the Bruker AMX500 spectrometer. All chemical shifts are quoted in ppm, relative to tetramethylsilane, using the residual solvent peak as a reference standard. Atmospheric Pressure Chemical Ionization Mass Spectrometry (APCI MS) measurements were performed on a Finnigan TSQ 7000 triple stage quadrupole mass spectrometer. UV-vis-NIR absorption spectra was recorded on a Shimadzu UV-1700/UV-3600 spectrophotometer. Continuous wave X-band ESR spectra of **DA-Per** in solution were obtained with a JEOL (FA200) spectrometer using a variable temperature liquid nitrogen cryostat. The ESR data recorded in powder were fitted by the following equation:

$$IT = \frac{C}{K_B[3 + \exp(-2J/K_BT)]}$$

Where, I is the ESR intensity, C is the constant, K_B is Boltzmann constant, and $-2J$ is correlated to the excitation energy from the singlet state to the triplet state.

2. Additional spectra

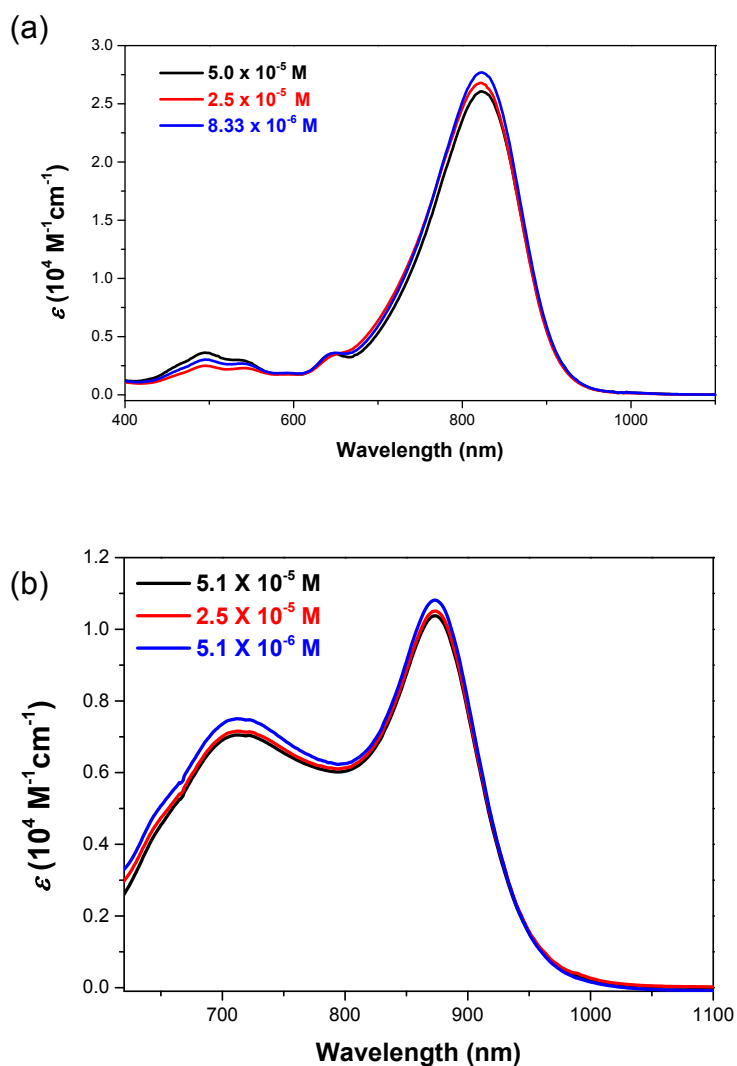


Figure S1. Concentration-dependent absorption spectra of DA-Per in (a) DCM and (b) CHCl_3 .

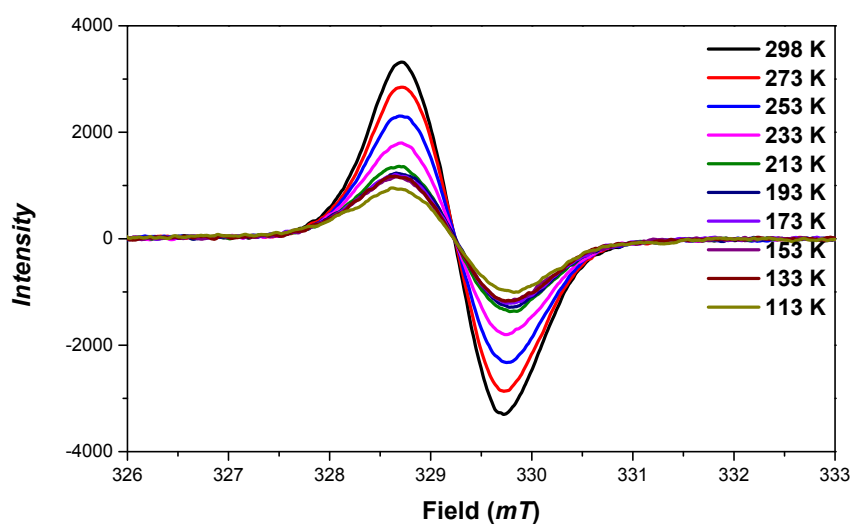


Figure S2. VT ESR spectra of the DA-Per recorded in powder.

Electronic Supplementary Information (ESI)

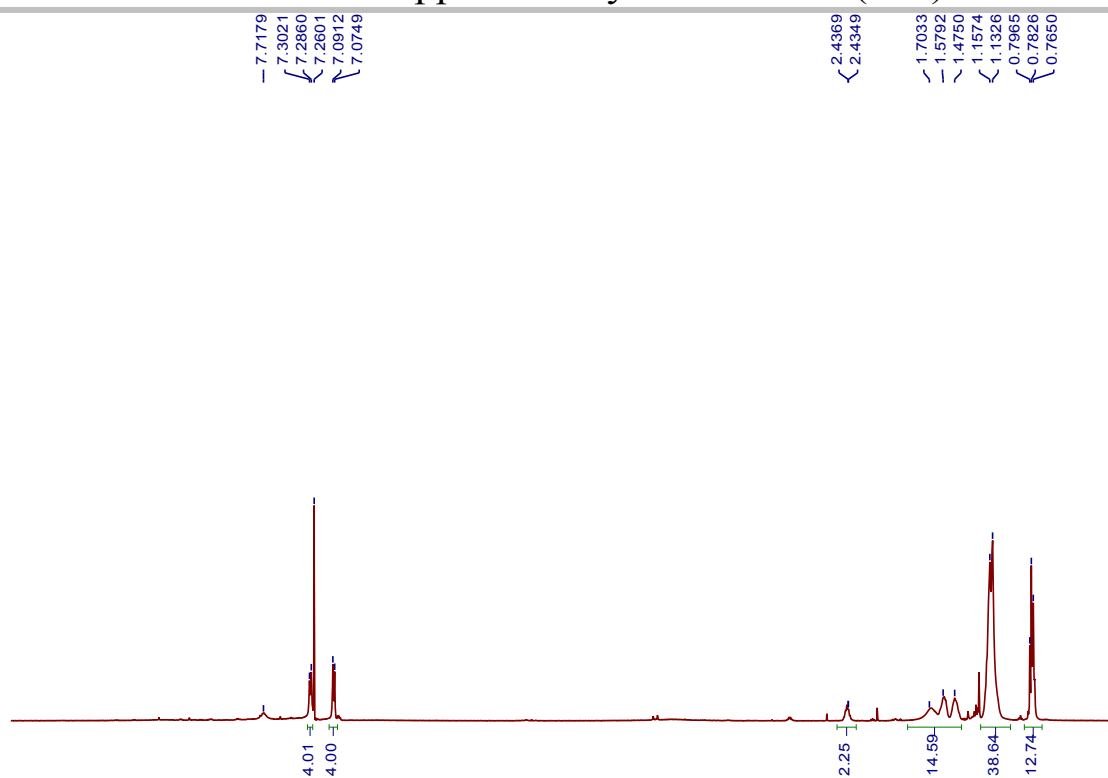


Figure S3. ¹H NMR (500 MHz) spectrum of compound **DA-Per** in CDCl₃ at room temperature.

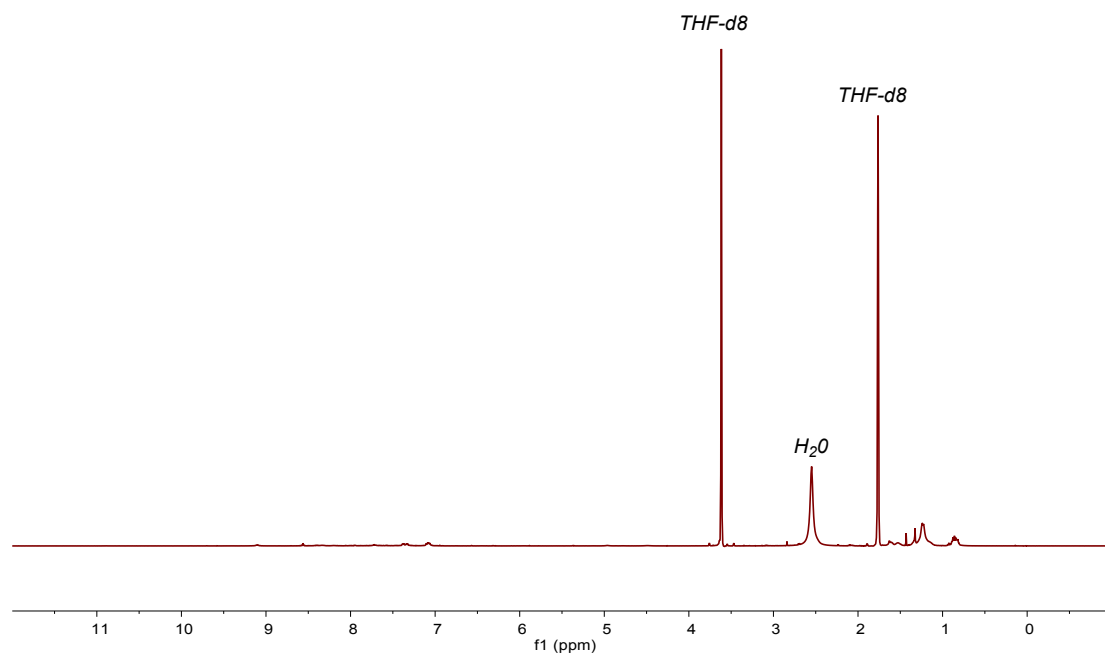


Figure S4. ¹H NMR (500 MHz) spectrum of compound **DA-Per** in THF-*d*₈ at room temperature.

Electronic Supplementary Information (ESI)

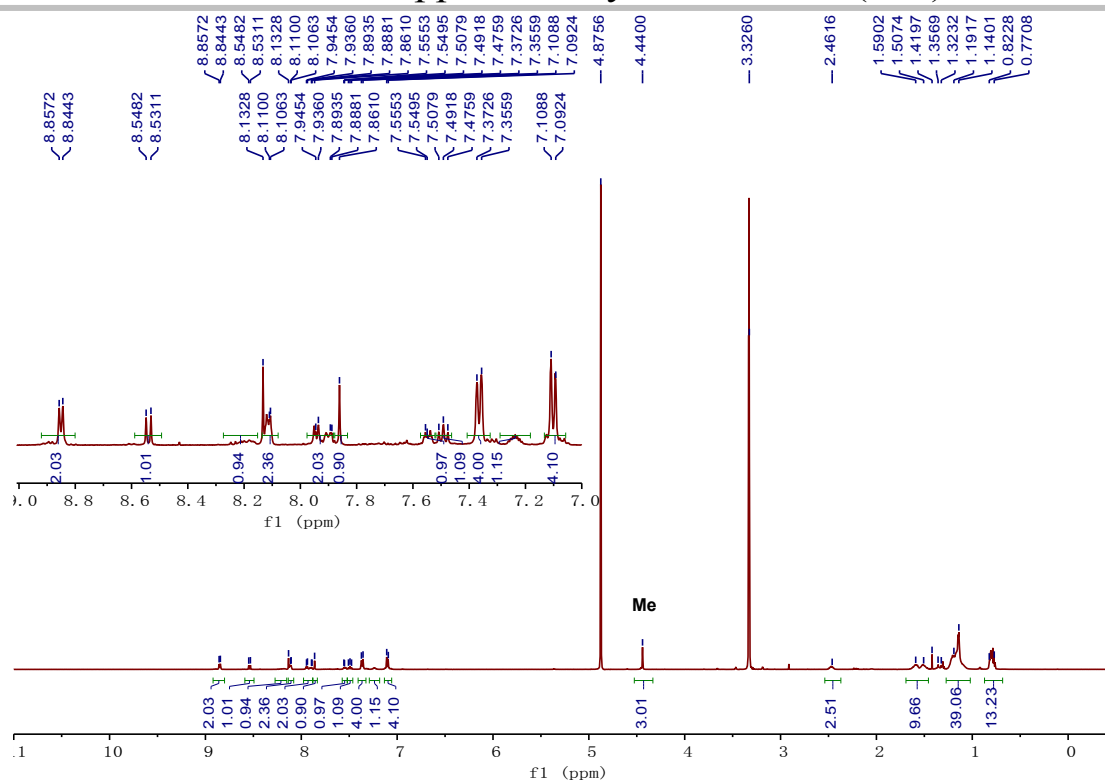


Figure S5. ^1H NMR (500 MHz) spectrum of compound **DA-Per** in $\text{CD}_3\text{OD}-d_4$ at room temperature.

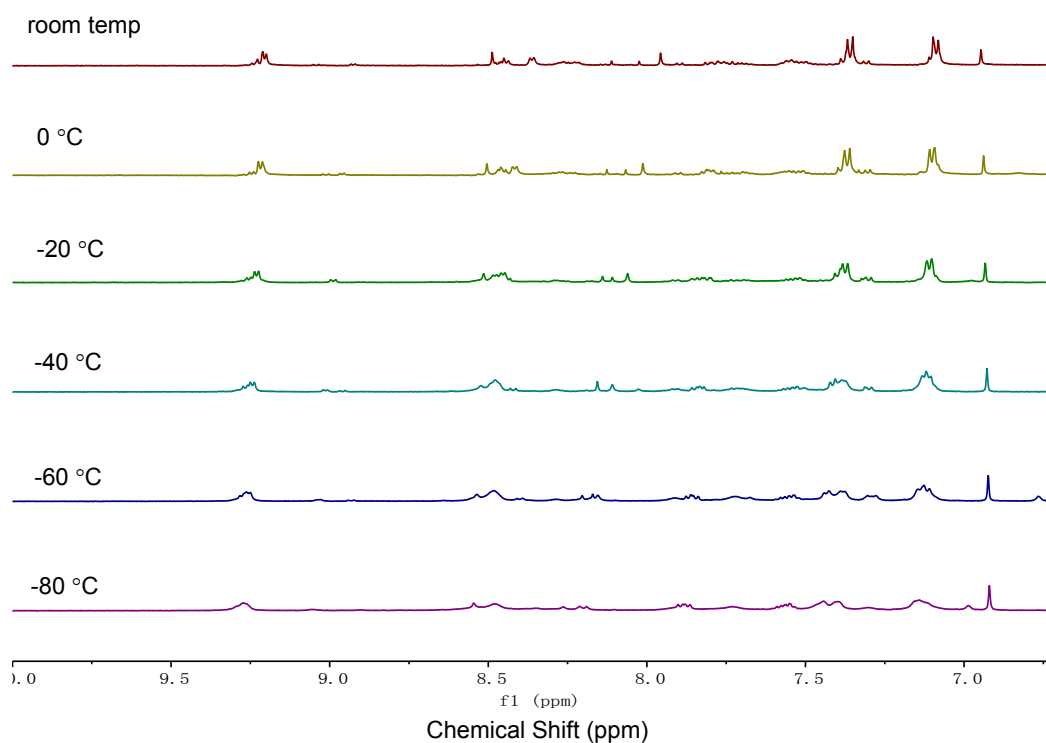


Figure S6. VT ^1H NMR spectra of **DA-Per** (500 MHz, aromatic region) in $\text{acetone}-d_6$.

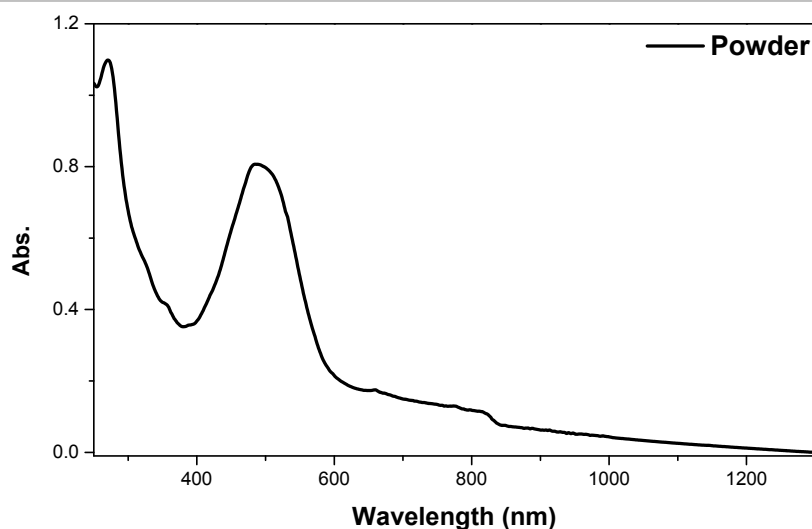


Figure S7. UV-vis absorption spectrum of the **DA-Per** recorded in powder.

3. Theoretical calculations

All results presented in this report have been obtained for a molecular model where the branched alkyl chains are replaced by methyl groups in **DA-Per**. Molecular geometries of molecule in the ground state and the singlet excited state have been obtained at both the B3LYP/6-311G(d,p) level with Gaussian 09 package^[2] utilizing a high performance computing cluster facility of NUS. The environment effects geometries were included by using the polarizable continuum model (PCM). Therefore, the diradical radical character and vertical excitation energies of **DA-Per** were calculated using the restricted active space spin flip method (RAS-SF)^[3] with Q-Chem 4.3 package based on the optimised triplet state geometries in different solvents.^[4] The diradical character index (y_0) is defined as the occupation number of the lowest unoccupied natural orbital (LUNO). Time-dependent DFT calculations were conducted at B3LYP/6-31G(d,p) level based on the ground geometries optimised at the B3LYP/6-31G(d,p) level in different solvents.

Electronic Supplementary Information (ESI)

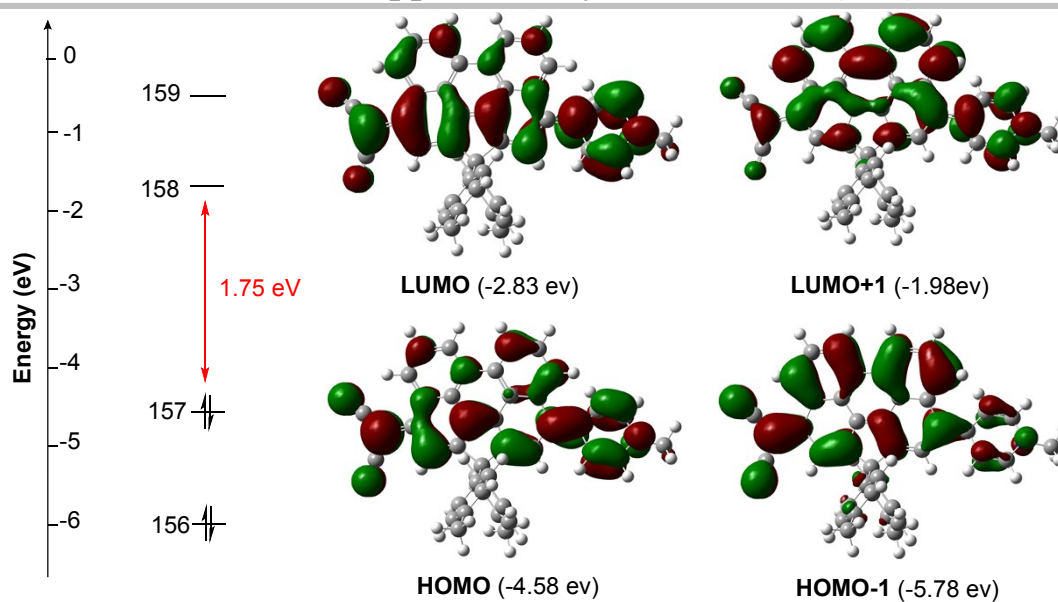


Figure S8. Frontier molecular orbital profiles and energy diagram of **DA-Per** obtained by B3LYP/6-31G(d,p) level calculation under gas phase.

Table S1. Selected TD-DFT (B3LYP/6-31G(d,p)) calculated energies, oscillator strength and compositions of major electronic transitions of **DA-Per** (in DMSO).

Calcd. (nm)	f	Composition (H=HOMO, L= LUMO, L+1 = LUMO+1, etc.)
700.4	1.068	HOMO->LUMO (101 %)
543.0	0.026	HOMO->LUMO+1 (97%)
448.5	0.0026	HOMO-1>LUMO (73%)

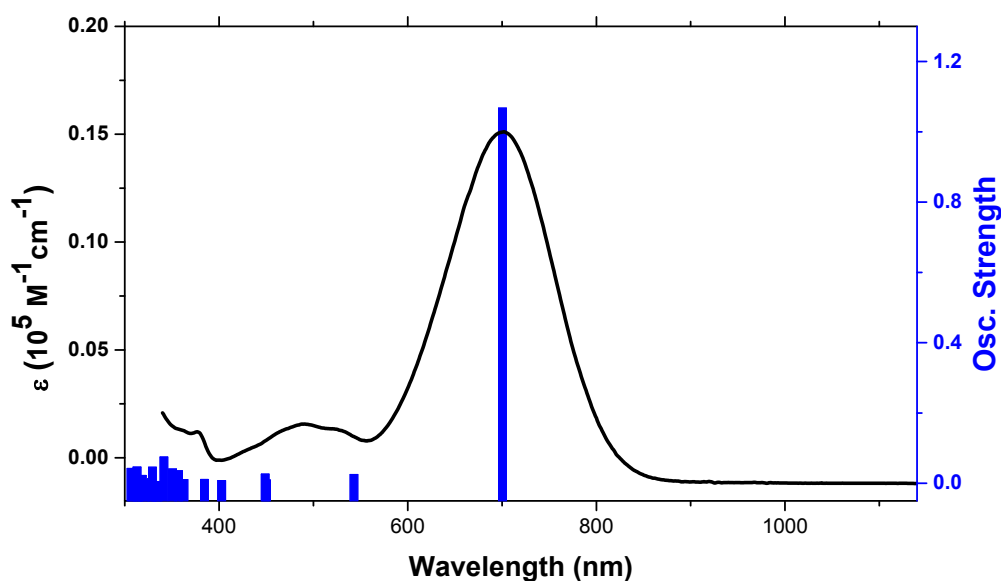


Figure S9. Calculated stick spectrum (B3LYP/6-31G(d,p)) of **DA-Per** along with the experimental spectrum in DMSO.

Electronic Supplementary Information (ESI)

4. Additional NMR spectra and HR-Mass spectra

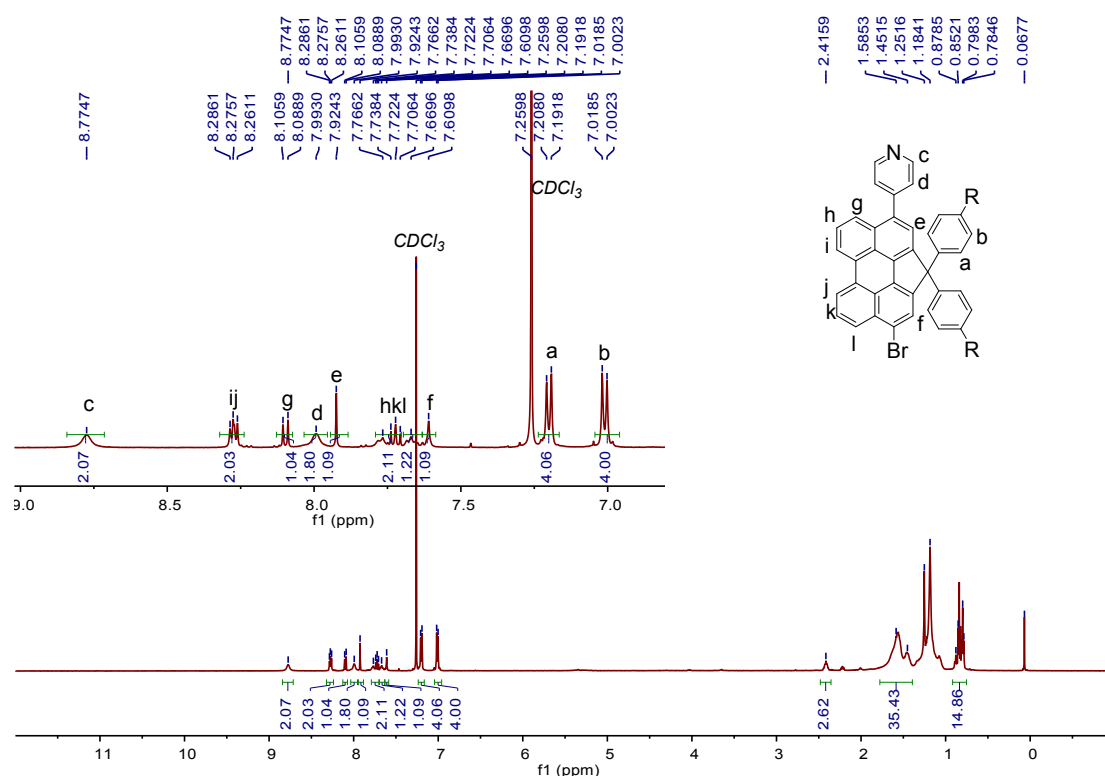


Figure S10. ¹H NMR spectrum of compound **2** (500 MHz, CDCl₃, rt).

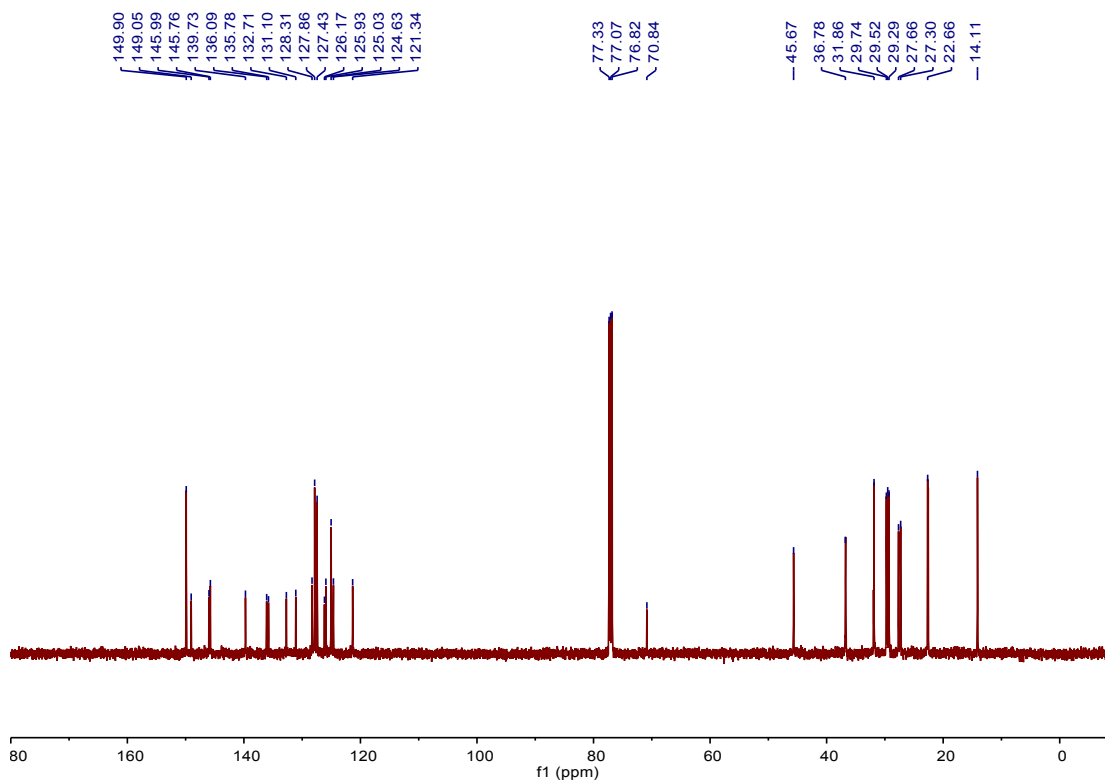


Figure S11. ¹³C NMR spectrum of compound **2** (125 MHz, CDCl₃, rt).

Electronic Supplementary Information (ESI)

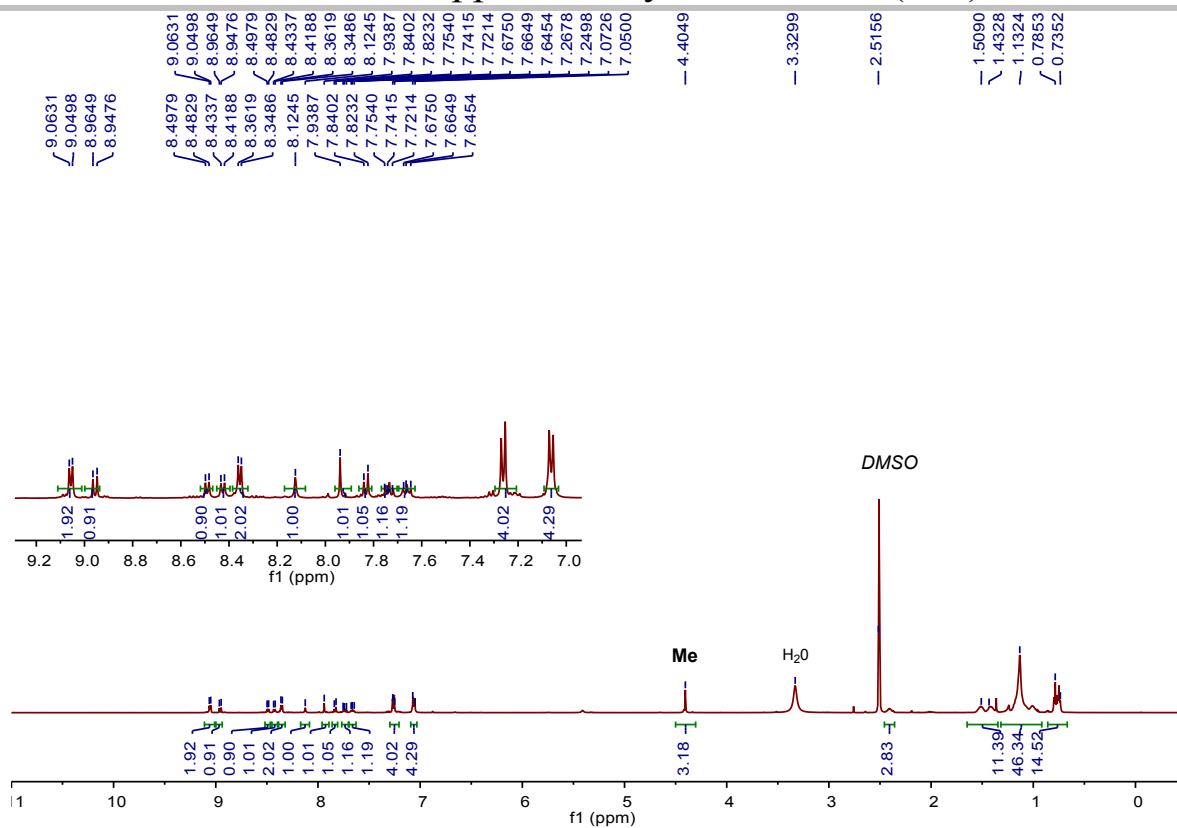


Figure S12. ^1H NMR spectrum of compound **DA-Per** (500 MHz, $\text{DMSO}-d_6$, rt).

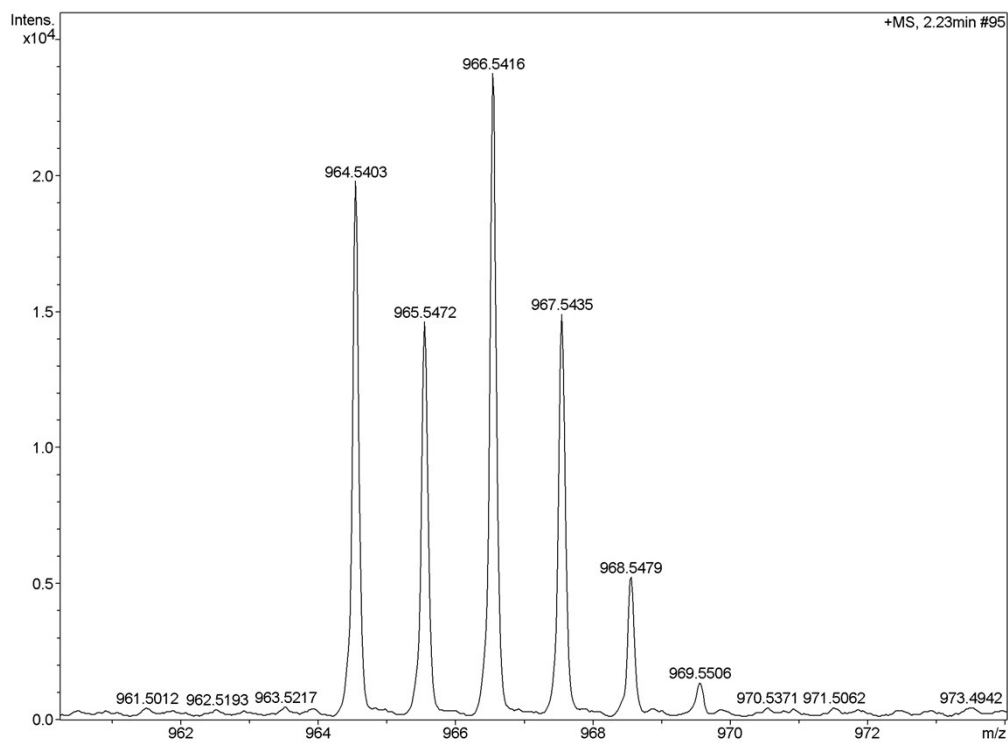


Figure S13. HR mass spectrum (APCI) of the compound **2**.

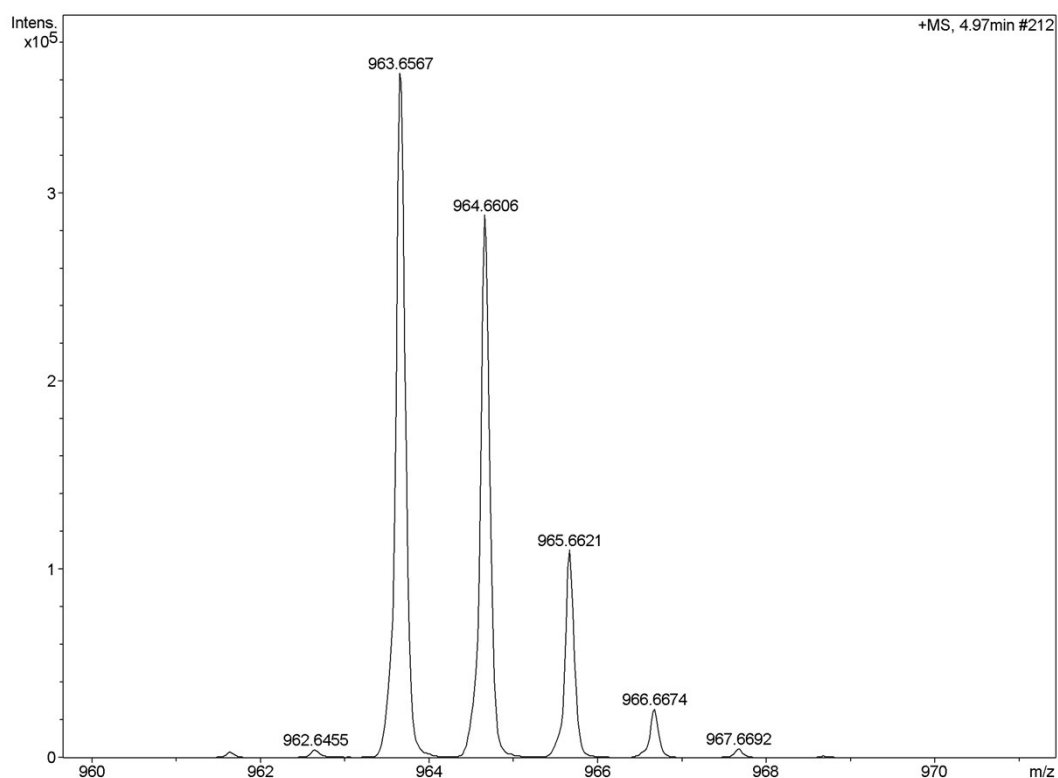


Figure S14. HR mass spectrum (APCI) of the compound **DA-Per**.

5. References

- [1] Zeng, W.; Phan, H.; Heng, T. S.; Gopalakrishna, T. Y.; Aratani, N.; Zeng, Z.; Yamada, H.; Ding, J.; Wu, J. *Chem* **2017**, 2, 81.
- [2] Gaussian 09, Revision D.01, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian, Inc., Wallingford CT, 2013.
- [3] (a) Casanova, D.; Head-Gordon, M. *Phys. Chem. Chem. Phys.* **2009**, 11, 9779. (b) Head-Gordon, M. *Chem. Phys. Lett.* **2003**, 372, 508.

Electronic Supplementary Information (ESI)

[4] Shao, Y.; Gan, Z.; Epifanovsky, E.; Gilbert, A.T.B.; Wormit, M.; Kussmann, J.; Lange, A.W.; Behn, A.; Deng, J.; Feng, X.; Ghosh, D.; Goldey, M.; Horn, P.R.; Jacobson, L.D.; Kaliman, I.; Khaliullin, R.Z.; Kus, T.; Landau, A.; Liu, J.; Proynov, E.I.; Rhee, Y.M.; Richard, R.M.; Rohrdanz, M.A.; Steele, R.P.; Sundstrom, E.J.; Woodcock III, H.L.; Zimmerman, P.M.; Zuev, D.; Albrecht, B.; Alguires, E.; Austin, B.; Beran, G.J.O.; Bernard, Y.A.; Berquist, E.; Brandhorst, K.; Bravaya, K.B.; Brown, S.T.; Casanova, D.; Chang, C.-M.; Chen, Y.; Chien, S.H.; Closser, K.D.; Crittenden, D.L.; Diedenhofen, M.; DiStasio Jr., R.A.; Do, H.; Dutoi, A.D.; Edgar, R.G.; Fatehi, S.; Fusti-Molnar, L.; Ghysels, A.; Golubeva-Zadorozhnaya, A.; Gomes, J.; Hanson-Heine, M.W.D.; Harbach, P.H.P.; Hauser, A.W.; Hohenstein, E.G.; Holden, Z.C.; Jagau, T.-C.; Ji, H.; Kaduk, B.; Khistyayev, K.; Kim, J.; Kim, J.; King, R.A.; Klunzinger, P.; Kosenkov, D.; Kowalczyk, T.; Krauter, C.M.; Laog, K.U.; Laurent, A.; Lawler, K.V.; Levchenko, S.V.; Lin, C.Y.; Liu, F.; Livshits, E.; Lochan, R.C.; Luenser, A.; Manohar, P.; Manzer, S.F.; Mao, S.-P.; Mardirossian, N.; Marenich, A.V.; Maurer, S.A.; Mayhall, N.J.; Oana, C.M.; Olivares-Amaya, R.; O'Neill, D.P.; Parkhill, J.A.; Perrine, T.M.; Peverati, R.; Pieniazek, P.A.; Prociuk, A.; Rehn, D.R.; Rosta, E.; Russ, N.J.; Sergueev, N.; Sharada, S.M.; Sharma, S.; Small, D.W.; Sodt, A.; Stein, T.; Stuck, D.; Su, Y.-C.; Thom, A.J.W.; Tsuchimochi, T.; Vogt, L.; Vydrov, O.; Wang, T.; Watson, M.A.; Wenzel, J.; White, A.; Williams, C.F.; Vanovschi, V.; Yeganeh, S.; Yost, S.R.; You, Z.-Q.; Zhang, I.Y.; Zhang, X.; Zhou, Y.; Brooks, B.R.; Chan, G.K.L.; Chipman, D.M.; Cramer, C.J.; Goddard III, W.A.; Gordon, M.S.; Hehre, W.J.; Klamt, A.; Schaefer III, H.F.; Schmidt, M.W.; Sherrill, C.D.; Truhlar, D.G.; Warshel, A.; Xu, X.; Aspuru-Guzik, A.; Baer, R.; Bell, A.T.; Besley, N.A.; Chai, J.-D.; Dreuw, A.; Dunietz, B.D.; Furlani, T.R.; Gwaltney, S.R.; Hsu, C.-P.; Jung, Y.; Kong, J.; Lambrecht, D.S.; Liang, W.Z.; Ochsenfeld, C.; Rassolov, V.A.; Slipchenko, L.V.; Subotnik, J.E.; Van Voorhis, T.; Herbert, J.M.; Krylov, A.I.; Gill, P.M.W.; Head-Gordon, M. *Mol. Phys.*, **2015**, *113*, 184.