## Supporting Information for

## Ligandless Nickel-Catalyzed Transfer Hydrogenation of Alkenes and Alkynes Using Water as the Hydrogen Donor

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General Methods. Unless noted, all reactions were carried out using standard Schlenk technique under an argon atmosphere or a dry box technique under a nitrogen atmosphere. Tetrahydrofuran was distilled from sodium and benzophenone. 1,4-Dioxane was distilled from sodium. Acetonitrile was dried using Innovative Technology Solvent Purifier. Ni(COD) ${ }_{2}$, $\mathrm{NiI}_{2}, \mathrm{Ni}(\mathrm{acac})_{2}$ and $\mathrm{NiF}_{2}$ were purchased from Strem Chemicals Inc. $\mathrm{NiCl}_{2}$ (DME) was purchased from Sigma-Aldrich. $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ and $\mathrm{NiCl}_{2}$ were purchased from Alfa Aesar. Zinc powder ( $98 \%$, -325 mesh) was purchased from Adamas. Zinc flake ( $99.8 \%,-325 \mathrm{mesh}$ ) and Mn powder (APS $<10$ micron, $99.6 \%$ metals basis) were purchased from Alfa Aesar. Al powder (200 mesh) and Mg turning (99.9+\%) were purchased from Acros Organics. Before using, zinc powder and zinc flake were stirred with 1 M HCl aqueous solution, filtered and washed thoroughly with water, acetone and diethyl ether and dried under vacuum. The deionized water was deoxygenated by argon gas bubbling method prior to use. Unless noted, all commercial reagents were used without further purification. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were recorded at room temperature in $\mathrm{CDCl}_{3}$ (containing $0.03 \% \mathrm{TMS}$ ) on Varian or Agilent XL400 MHz spectrometer. ${ }^{1} \mathrm{H}$ NMR spectra was recorded with tetramethylsilane ( $\delta=0.00 \mathrm{ppm}$ ); ${ }^{13} \mathrm{C}$ NMR spectra was recorded with $\mathrm{CDCl}_{3}(\delta=77.00 \mathrm{ppm})$ as internal reference. Highresolution mass spectra were obtained by using Waters Micromass GCT Premier mass spectrometer, Agilent Technologies 6224 TOF LC/MS or Thermo Fisher Scientific LTQ FT Ultra mass spectrometers. The IR spectra were measured on a ThermoFisher Nicolet FT-IR spectrometer. $\mathrm{H}_{2}$ gas analysis was acquired on Agilent 6890Plus gas chromatograph equipped with a Thermal Conductivity Detector.

Alkenes were synthesized according to the published methods. The spectral data of known compounds were in agreement with the literature. For the synthesis of new alkenes $\mathbf{1 0}$ and $\mathbf{5 d}$, see the following:


To a solution of 7-hydroxy-3-(4-methoxyphenyl)-4H-chromen-4-one ( $2.68 \mathrm{~g}, 10 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}(2.02 \mathrm{~g}, 20 \mathrm{mmol})$ in $\mathrm{DCM}(50 \mathrm{~mL})$ was added $\mathrm{Tf}_{2} \mathrm{O}(3.10 \mathrm{~g}, 11 \mathrm{mmol})$ at $0{ }^{\circ} \mathrm{C}$ under Argon, then the mixture was warmed up to room temperature and stirred for another 2 h . The reaction was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ and extracted with DCM. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated in vacuum. The residue was purified by column chromatography on silica gel (eluent: dichloromethane/ petroleum ether $=$ 2:1) to give 3-(4-methoxyphenyl)-4-oxo-4H-chromen-7-yl trifluoromethanesulfonate (s-10) in $69 \%$ yield $(2.76 \mathrm{~g})$ as a white solid. M.p.: $125.0-126.3^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{M}, \mathrm{CDCl}_{3}$ ): $\delta 8.40$ (d, $J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.01(\mathrm{~s}, 1 \mathrm{H}), 7.49(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.45(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.33(\mathrm{dd}, J$ $=8.8,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.97(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.84(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{M}, \mathrm{CDCl}_{3}\right): \delta 175.11$, $159.86,156.27,152.80,152.02,130.00,129.12,125.62,124.14,123.08,118.63\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=\right.$ 321.3 Hz ), 118.47, 114.04, 111.39, 55.26. IR (neat): 3097, 3076, 3058, 2954, 2907, 2836, $1644,1612,1568,1516,1424,1400,1292,1250,1212,1182,1168,1132,1095,1029,952$, 883, 864, 824, 801, 788, 782, 689, $659 \mathrm{~cm}^{-1}$. HRMS (ESI) calcd for $\mathrm{C}_{17} \mathrm{H}_{12} \mathrm{~F}_{3} \mathrm{O}_{6} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 401.0301, found 401.0298.


To a solution of $\mathbf{s - 1 0}(1.20 \mathrm{~g}, 3 \mathrm{mmol}), \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(210.6 \mathrm{mg}, 0.3 \mathrm{mmol})$ and LiCl ( $127.2 \mathrm{mg}, 3 \mathrm{~mol}$ ) in DMF ( 18 mL ) was added tributyl(vinyl)stannane ( $1.05 \mathrm{~g}, 3.3 \mathrm{mmol}$ ) dropwise under Argon. The mixture was heated at $90^{\circ} \mathrm{C}$ for 4 h . After completion the reaction mixture was cooled down to room temperature and diluted with $\mathrm{Et}_{2} \mathrm{O}$ and washed with 1 M HCl (aq.), then washed with saturated $\mathrm{NaHCO}_{3}$ (aq.) and brine. The combined aqueous layers were extracted once with $\mathrm{Et}_{2} \mathrm{O}$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated in vacuum. The residue was purified by column chromatography on silica gel (eluent: petroleum ether/dichloromethane/ethyl acetate $=6: 1: 1$ ) to give 3-(4-methoxyphenyl)-7-vinyl-4 H -chromen-4-one (10) in $76 \%$ yield ( 637.7 mg ) as a white solid. M.p.: $154.0-155.1^{\circ} \mathrm{C} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{M}, \underset{\mathrm{s} 3}{ } \mathrm{CDCl}_{3}\right): \delta 8.24(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.96(\mathrm{~s}, 1 \mathrm{H})$,
7.52-7.46 (m, 3H), 7.41 (s, 1H), 6.98-6.96 (m, 2H), 6.79 (dd, $J=17.6,11.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.94$ (d, $J=17.6, \mathrm{~Hz}, 1 \mathrm{H}), 5.48(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{M}, \mathrm{CDCl}_{3}$ ): $\delta$ $176.06,159.55,156.44,152.49,142.89,135.32,130.02,126.49,124.97,124.03,123.62$, $122.83,117.81,115.37,113.91,55.26$. IR (neat): 3092, 3076, 3040, 3011, 2972, 2930, 2838, $1624,1609,1557,1513,1439,1407,1358,1291,1253,1233,1176,1106,1024,985,927$, 901, 886, 878, 830, 821, 802, 791, 767, $717 \mathrm{~cm}^{-1}$. HRMS (ESI) calcd for $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}$: 279.1016, found 279.1007.


To a solution of 1-tosylpiperidin-4-ol ( $1.79 \mathrm{~g}, 7 \mathrm{mmol}$ ) in DCM ( 50 mL ) was added pyridine $(12.5 \mathrm{~mL})$. Then the mixture was cooled to $0^{\circ} \mathrm{C}$ and $\mathrm{Tf}_{2} \mathrm{O}(4.48 \mathrm{~mL}, 26.6 \mathrm{mmol})$ was slowly added. Then the mixture was stirred for 3 h at $0^{\circ} \mathrm{C}$. The reaction mixture was quenched by $\mathrm{H}_{2} \mathrm{O}$ and extracted with DCM. The combined organic layers were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The mixture was filtered and the solvent was evaporated under the reduced pressure. The residue was purified by column chromatography on silica gel (eluent: petroleum ether/DCM = 1/1 to DCM ) to afford 1-tosyl-1,2,3,6-tetrahydropyridine (5d) in 46\% yield ( 762.3 mg ) as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.68(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.32(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 5.76-5.73(\mathrm{~m}, 1 \mathrm{H}), 5.63-5.60(\mathrm{~m}, 1 \mathrm{H}), 3.59-3.56(\mathrm{~m}, 2 \mathrm{H}), 3.17(\mathrm{t}, J=$ $5.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), $2.43(\mathrm{~s}, 3 \mathrm{H}), 2.24-2.18(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 143.44,133.18$, $129.55,127.59,124.97,122.65,44.70,42.56,25.17,21.43$. The spectroscopic data are in agreement with that previously reported. ${ }^{1}$

## Optimization studies for transfer hydrogenation of aryl alkenes.

## General procedure for optimization studies.

The reaction was conducted in an oven-dried screw-cap vial ( 4 mL ) equipped with a magnetic stir bar. In a nitrogen-filled glove box, $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(3.6 \mathrm{mg}, 0.015 \mathrm{mmol})$ [or other $\mathrm{Ni}(\mathrm{II})$ salts], Zinc powder ( $39.2 \mathrm{mg}, 0.6 \mathrm{mmol}$ ) [or other reductants], 2-vinylnaphthalene (46.3
$\mathrm{mg}, 0.3 \mathrm{mmol}$ ), dioxane or other solvents $(1.5 \mathrm{~mL})$ were added sequentially to a screw-cap vial. The vial cap was then securely fitted and taken outside the glove box. Then $\mathrm{H}_{2} \mathrm{O}(0.3 \mathrm{~mL}$ or x mL ) was added to the vial. After the reaction mixture was stirred at $80^{\circ} \mathrm{C}$, the mixture was filtered through a pad of silica gel and washed with petroleum ether. The solvent was evaporated under the reduced pressure and the residue was dissolved in $\mathrm{CDCl}_{3}$. The NMR yields were obtained by ${ }^{1} \mathrm{H}$ NMR analysis of the crude mixture using 1,3,5trimethoxybenzene ( $50.5 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) as an internal standard.

Note: $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ used in the optimization study was stored under air (outside of glove box), which was brought into the glove box and used immediately for every experiment.

Table 1. Optimization of the reaction conditions

|  | $5 \mathrm{~mol} \% \mathrm{Ni}$ catalyst 2.0 equiv reductant solvent, $80^{\circ} \mathrm{C}, \mathrm{t}(\mathrm{h})$ <br> 2a |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| entry | catalyst | reductant | solvent | time (h) | yield (\%) ${ }^{\text {a }}$ |
| 1 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 88 |
| 2 | $\mathrm{NiCl}_{2}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 89 |
| 3 | $\mathrm{NiBr}_{2}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 89 |
| 4 | $\mathrm{Nil}_{2}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 79 (7) |
| 5 | $\mathrm{NiF}_{2}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 2 (85) |
| 6 | $\mathrm{Ni}(\mathrm{acac})_{2}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | - (89) |
| 7 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Al | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 24 (68) |
| 8 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Mn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 13 (81) |
| 9 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Mg | dioxane/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | - (96) |
| 10 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | DMF/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 90 (1) |
| 11 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | THF/ $\mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 95 |
| 12 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}(5 / 1)$ | 10 | 1 (58) |
| 13 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(25 / 1)$ | 10 | 33 (59) |
| 14 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(3 / 1)$ | 10 | 96 |
| 15 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(3 / 1)$ | 5 | 95 |
| $16^{b}$ | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(3 / 1)$ | 5 | 48 (48) |
| $17^{\text {c }}$ | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(3 / 1)$ | 5 | 32 (63) |
| 18 | - | Zn | dioxane/ $\mathrm{H}_{2} \mathrm{O}(3 / 1)$ | 5 | - (94) |
| 19 | $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ | - | dioxane/ $\mathrm{H}_{2} \mathrm{O}(3 / 1)$ | 5 | - (94) |

${ }^{\text {a }}$ Determined by ${ }^{1} \mathrm{H}$ NMR using 1,3,5-trimethoxybenzene as an internal standard. The yields of the unreacted 1a are shown in parentheses. ${ }^{b} 1.0$ equiv Zn was used. ${ }^{c} 50^{\circ} \mathrm{C}$.

## Ni-Catalyzed transfer hydrogenation of aryl/heteroaryl alkenes, alkyl alkenes and heterocycles.

## Typical procedure for the synthesis of 2a.



To an oven dried Schlenk tube ( 25 mL ) were added $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), and 2-vinylnaphthalene ( $77.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) under air. The Schlenk tube was capped with a rubber septum, evacuated and back filled with argon for three times. Then dioxane $(2.5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were added sequentially under argon. (If the starting material is a liquild, after the $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ and Zinc were added, the Schlenk tube was capped with a rubber septum, evacuated and back filled with argon for three times. Then dioxane, the alkene and $\mathrm{H}_{2} \mathrm{O}$ were added sequentially under argon). The tube cap was then securely fitted and sealed with electrical tape, and the stopcock valve on the sidearm of the Schlenk tube was closed. After the reaction mixture was stirred at $80^{\circ} \mathrm{C}$ for 10 h , it was filtered through a pad of silica gel and washed with petroleum ether. Note: a stirring speed above 600 rpm is highly important for reproducibility. The solvent was evaporated under the reduced pressure and the residue was purified by column chromatography on silica gel (eluent: petroleum ether) to give the desired product 2a in $92 \%$ yield $(72.0 \mathrm{mg})$ as a colorless oil.


2a
2-Ethylnaphthalene (2a). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.77-7.71(\mathrm{~m}, 3 \mathrm{H}), 7.58(\mathrm{~s}, 1 \mathrm{H})$, $7.42-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.31-7.29(\mathrm{~m}, 1 \mathrm{H}), 2.77(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.29(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 141.69,133.67,131.90,127.77,127.56,127.38,127.04,125.79$, $125.50,124.97,29.01,15.51$. The spectroscopic data are in agreement with that previously reported. ${ }^{2}$


2b
1-Ethyl-4-methylbenzene (2b). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( 65.4 mg ,
1.0 mmol ), dioxane ( 2.5 mL ), 1-methyl-4-vinylbenzene ( $59.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8$ mL ) were stirred at $80^{\circ} \mathrm{C}$ for 11.5 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{2 b}$ in $44 \%$ yield ( 26.7 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.09(\mathrm{~s}, 4 \mathrm{H}), 2.61(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.31(\mathrm{~s}, 3 \mathrm{H}), 1.22(\mathrm{t}, J=7.6 \mathrm{~Hz}$, 3 H ); ${ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta$ 141.18, 134.96, 128.97, 127.71, 28.42, 20.96, 15.78. The spectroscopic data are in agreement with that previously reported. ${ }^{3}$

Due to the volatile nature of this compound, the NMR yield was determined. $86 \%$ NMR yield of the desired product $\mathbf{2 b}$ was obtained.


2c
1-Ethyl-2-methylbenzene(2c). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( 65.4 mg , 1.0 mmol ), dioxane ( 2.5 mL ), 1-methyl-2-vinylbenzene ( $59.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8$ mL ) were stirred at $80^{\circ} \mathrm{C}$ for 24 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{2 c}$ in $21 \%$ yield ( 12.5 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( 400 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.15-7.07(\mathrm{~m}, 4 \mathrm{H}), 2.63(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.31(\mathrm{~s}, 3 \mathrm{H}), 1.21(\mathrm{t}, J=7.6 \mathrm{~Hz}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 142.29,135.73,129.99,127.86,125.98,125.70,26.15$, $19.15,14.36$. The spectroscopic data are in agreement with that previously reported. ${ }^{4}$

Due to the volatile nature of this compound, the NMR yield was determined. $89 \%$ NMR yield of the desired product $\mathbf{2 c}$ was obtained.


2d
1-Ethyl-4-methoxybenzene(2d). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol}$ ), Zinc powder ( 65.4 mg , 1.0 mmol ), dioxane ( 2.5 mL ), 1-methoxy-4-vinylbenzene ( $67.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8$ mL ) were stirred at $80^{\circ} \mathrm{C}$ for 10.5 h . Column chromatography on silica gel (eluent: petroleum ether to petroleum ether/ethyl ether $=50: 1$ ) afforded the desired product $\mathbf{2 d}$ in $63 \%$ yield $(42.6 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.11(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.82(\mathrm{~d}$,
$J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 2.58(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.20(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 157.56,136.33,128.65,113.67,55.18,27.93,15.87$. The spectroscopic data are in agreement with that previously reported. ${ }^{4}$


2e
4-Ethylaniline(2e). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), 4-vinylaniline ( $59.6 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80{ }^{\circ} \mathrm{C}$ for 7 h . The mixture was filtered through a pad of silica gel and washed with ethyl ether. Column chromatography on silica gel (eluent: petroleum ether/ethyl ether $=1: 1$ ) afforded the desired product 2e in $90 \%$ yield ( 54.8 mg ) as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ $6.98(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.61(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.46(\mathrm{bs}, 2 \mathrm{H}), 2.53(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.18$ ( $\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 143.97,134.31,128.48,115.17,27.88$, 15.87. The spectroscopic data are in agreement with that previously reported. ${ }^{5}$

$2 f$
4-Ethylphenol (2f). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), 4vinylphenol ( $60.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 7 h , the mixture was filtered through a pad of silica gel and washed with ethyl ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl ether = 3:1) afforded the desired product $\mathbf{2 f}$ in $86 \%$ yield $(52.5 \mathrm{mg})$ as a colorless soild. ${ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.05(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.75(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.39(\mathrm{~s}, 1 \mathrm{H}), 2.57(\mathrm{q}, J=$ $8.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.19(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 153.19,136.56,128.88$, $115.13,27.92,15.83$. The spectroscopic data are in agreement with that previously reported. ${ }^{6}$

$2 g$

1-Chloro-4-ethylbenzene (2g). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol}$ ), Zinc powder ( 65.4 mg , 1.0 mmol ), dioxane ( 2.5 mL ), 1-chloro-4-vinylbenzene ( $69.3 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL}$ ) were stirred at $80^{\circ} \mathrm{C}$ for 20 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{2 g}$ in $35 \%$ yield ( 24.7 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 7.24(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.11(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.61(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.21(\mathrm{t}$, $J=7.2 \mathrm{~Hz}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 142.60,131.22,129.18,128.34,28.23,15.52$. The spectroscopic data are in agreement with that previously reported. ${ }^{7}$

Due to the volatile nature of this compound, the NMR yield was determined. $62 \%$ NMR yield of the desired product $\mathbf{2 g}$ was obtained.


2h
Methyl 4-ethylbenzoate (2h). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0$ mmol ), methyl 4-vinylbenzoate ( $81.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 23.5 h , the mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=15: 1$ ) afforded the desired product $\mathbf{2 h}$ in $76 \%$ yield ( 62.8 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.95$ (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.25(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), $3.89(\mathrm{~s}, 3 \mathrm{H}), 2.69(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.25(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 167.10,149.66,129.63,127.80,127.55,51.86,28.86,15.14$. The spectroscopic data are in agreement with that previously reported. ${ }^{8}$


2-(4-Ethylphenyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (2i). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ (5.9 mg, 0.025 mmol ), Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), 4,4,5,5-tetramethyl-2-(4-vinylphenyl)-1,3,2dioxaborolane ( $115.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane $(2.5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at 80 ${ }^{\circ} \mathrm{C}$ for 11 h , the mixture was filtered through a pad of silica gel and washed with ethyl acetate.

Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=20: 1$ ) afforded the desired product $\mathbf{2 i}$ in $72 \%$ yield ( 83.4 mg ) as a colorless solid. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 7.74(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.21(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.65(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.33(\mathrm{~s}$, 12 H ), $1.23(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 147.66,134.87,127.30,83.54$, $29.06,24.80,15.44$. The spectroscopic data are in agreement with that previously reported. ${ }^{9}$


2j
4-Ethyl-1,1'-biphenyl (2j). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0$ mmol), 4-vinyl-1, 1'-biphenyl ( $90.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 7.5 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{2} \mathbf{j}$ in $94 \%$ yield ( 85.3 mg ) as a white solid. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 7.58(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.52(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.43(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.34-$ $7.27(\mathrm{~m}, 3 \mathrm{H}), 2.70(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.28(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 143.32,141.14,138.57,128.67,128.25,127.04,126.96,126.92,28.48,15.57$. The spectroscopic data are in agreement with that previously reported. ${ }^{10}$

(4-Ethylphenyl)methanol (2k). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol}$ ), Zinc powder ( 65.4 mg , 1.0 mmol ), dioxane ( 2.5 mL ), (4-vinylphenyl)methanol ( $67.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8$ mL ) were stirred at $80^{\circ} \mathrm{C}$ for 12 h . Column chromatography on silica gel (eluent: petroleum ether : ether acetate $=5: 1$ ) afforded the desired product $\mathbf{2 k}$ in $81 \%$ yield $(55.2 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.24(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.17(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H})$, 4.57 (s, 2H), 2.63 ( $\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 2.27 (s, 1H), 1.22 (t, $J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 143.63,138.10,127.93,127.10,64.99,28.49,15.58$. The spectroscopic data are in agreement with that previously reported. ${ }^{11}$


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1-Ethylferrocene (21). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), 1-vinylferrocene ( $106.0 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 17 h . The mixture was filtered through a pad of silica gel and washed with petroleum ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{2 l}$ in $97 \%$ yield $(103.4 \mathrm{mg})$ as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( 400 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 4.09-4.03(\mathrm{~m}, 9 \mathrm{H}), 2.33(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.16(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 91.03,68.31,67.36,66.88,22.17,14.61$. The spectroscopic data are in agreement with that previously reported. ${ }^{12}$


3-Ethyl-1H-indole (2m). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol}$ ), Zinc powder ( $65.4 \mathrm{mg}, 1.0$ mmol ), 3-vinyl-1 $H$-indole ( $71.6 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 8 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=$ 5:1) afforded the desired product $\mathbf{2 m}$ in $98 \%$ yield $(71.0 \mathrm{mg})$ as a colorless soild. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.60(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.25-7.23(\mathrm{~m}, 1 \mathrm{H}), 7.17(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H})$, $7.12-7.08(\mathrm{~m}, 1 \mathrm{H}), 6.84(\mathrm{~s}, 1 \mathrm{H}), 2.76(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.31(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 136.28,127.30,121.78,120.44,118.98,118.88,118.60,111.02,18.27$, 14.40. The spectroscopic data are in agreement with that previously reported. ${ }^{13}$


2n
2-Ethylpyridine (2n). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), 2-vinylpyridine ( $52.6 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at 80
${ }^{\circ} \mathrm{C}$ for 9 h . The mixture was filtered through a pad of silica gel and washed with ethyl ether. Column chromatography on silica gel (eluent: petroleum ether to petroleum ether/ethyl ether $=3: 1$ ) afforded the desired product $\mathbf{2 n}$ in $25 \%$ yield $(13.3 \mathrm{mg})$ as a colorless oil. The product contains small amount of stabilizer and solvent. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 8.53(\mathrm{~d}, J=$ $4.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.59$ (td, $J=8.0,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.16(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.11-7.08(\mathrm{~m}, 1 \mathrm{H}), 2.83$ (q, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), $1.31(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 163.47, 149.10, $136.34,122.00,120.85,31.33,13.87$. The spectroscopic data are in agreement with that previously reported. ${ }^{14}$

Due to the volatile nature of this compound, the NMR yield was determined. $92 \%$ NMR yield of the desired product $\mathbf{2 n}$ was obtained.


7-Ethyl-3-(4-methoxyphenyl)-4 $\boldsymbol{H}$-chromen-4-one (20). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(7.1 \mathrm{mg}, 0.03 \mathrm{mmol}$ ), Zinc powder ( $39.2 \mathrm{mg}, 0.6 \mathrm{mmol}$ ), 3-(4-methoxyphenyl)-7-vinyl-4H-chromen-4-one ( 83.5 mg , $0.3 \mathrm{mmol})$, dioxane $(1.5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 20 h . the mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=5: 1$ ) afforded the desired product $\mathbf{2 o}$ in $86 \%$ yield $(72.4 \mathrm{mg})$ as a white soild. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 8.20(\mathrm{~d}, J$ $=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.93(\mathrm{~s}, 1 \mathrm{H}), 7.50(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.25-7.23(\mathrm{~m}, 2 \mathrm{H}), 6.96(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, 2 H ), 3.82 (s, 3H), 2.77 (q, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), $1.29(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 176.21,159.43,156.33,152.21,150.87,129.98,126.08,125.44,124.67,124.17$, $122.32,116.33,113.82,55.19,28.86,14.88$. IR (neat): 3082, 3063, 3032, 2961, 2928, 2870, $2852,1726,1633,1622,1607,1575,1559,1511,1461,1439,1372,1358,1289,1246,1225$, 1194, 1178, 1105, 1027, 906, 885, 873, 836, 817, 803, 792, 761, $697 \mathrm{~cm}^{-1}$. HRMS (ESI) calcd for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}:$281.1172, found 281.1179.

( $8 R, 9 S, 13 S, 14 S$ )-3-Ethyl-13-methyl-6,7,8,9,11,12,13,14,15,16-decahydro-17H-
cyclopenta $[\boldsymbol{a}]$ phenanthren-17-one (2p). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(7.1 \mathrm{mg}, 0.03 \mathrm{mmol})$, Zinc powder $(39.2 \mathrm{mg}, \quad 0.6 \mathrm{mmol}), \quad(8 R, 9 S, 13 S, 14 S)$-13-methyl-3-vinyl-6,7,8,9,11,12,13,14,15,16-decahydro- 17 H -cyclopenta[ $a$ ]phenanthren-17-one ( $84.1 \mathrm{mg}, 0.3 \mathrm{mmol}$ ), dioxane ( 1.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 13 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=$ $10: 1$ ) afforded the desired product $\mathbf{2 p}$ in $80 \%$ yield ( 68.2 mg ) as a white soild. ${ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.21(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.00(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~s}, 1 \mathrm{H}), 2.90-2.88(\mathrm{~m}$, $2 \mathrm{H}), 2.59(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.53-2.40(\mathrm{~m}, 2 \mathrm{H}), 2.30-2.25(\mathrm{~m}, 1 \mathrm{H}), 2.18-1.93(\mathrm{~m}, 4 \mathrm{H}), 1.67-$ $1.38(\mathrm{~m}, 6 \mathrm{H}), 1.23(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.89(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 220.80$, $141.60,136.88,136.21,128.40,125.26,125.23,50.40,47.91,44.21,38.16,35.77,31.53$, 29.33, 28.21, 26.51, 25.66, 21.50, 15.53, 13.75. IR (neat): 2968, 2943, 2925, 2860, 1735, $1499,1459,1437,1403,1374,1281,1256,1211,1085,1048,1006,910,880,823,783,711$ $\mathrm{cm}^{-1}$. HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{27} \mathrm{O}[\mathrm{M}+\mathrm{H}]^{+}: 283.2056$, found 283.2060.


2q
Ethane-1,1-diyldibenzene (2q). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc flake ( $65.4 \mathrm{mg}, 1.0$ mmol ), dioxane ( 2.5 mL ), ethene-1,1-diyldibenzene ( $90.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 24 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product 2q in $90 \%$ yield ( 81.8 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 7.27-7.11(\mathrm{~m}, 10 \mathrm{H}), 4.13(\mathrm{q}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.62(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 146.30,128.32,127.58,125.98,44.70,21.83$. The spectroscopic data are in agreement with that previously reported. ${ }^{15}$


2r
2-Propylnaphthalene (2r). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0$ $\mathrm{mmol})$, dioxane ( 2.5 mL ), 2-allylnaphthalene ( $84.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 10 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{2 r}$ in $93 \%$ yield ( 79.1 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 7.79-7.73 (m, 3H), $7.59(\mathrm{~s}, 1 \mathrm{H}), 7.44-7.37(\mathrm{~m}, 2 \mathrm{H}), 7.31(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.73(\mathrm{t}, J=7.6$ $\mathrm{Hz}, 2 \mathrm{H}), 1.77-1.67(\mathrm{~m}, 2 \mathrm{H}), 0.96(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 140.14$, 133.61, 131.94, 127.69, 127.57, 127.43, 127.38, 126.36, 125.76, 124.96, 38.17, 24.44, 13.85. The spectroscopic data are in agreement with that previously reported. ${ }^{16}$


2s
1-Methoxy-4-propylbenzene (2s). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( 65.4 mg , 1.0 mmol ), dioxane ( 2.5 mL ), 1-allyl-4-methoxybenzene ( $74.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8$ mL ) were stirred at $80^{\circ} \mathrm{C}$ for 8 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=50: 1$ ) afforded the desired product 2 s in $93 \%$ yield ( 70.2 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.08(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.81(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, $3.76(\mathrm{~s}, 3 \mathrm{H}), 2.52(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.64-1.55(\mathrm{~m}, 2 \mathrm{H}), 0.92(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 157.55,134.72,129.25,113.53,55.12,37.09,24.79,13.74$. The spectroscopic data are in agreement with that previously reported. ${ }^{17}$


Butoxy(tert-butyl)diphenylsilane (2t). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder
( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), (but-3-en-1-yloxy)(tert-butyl)diphenylsilane ( 155.3 $\mathrm{mg}, 0.5 \mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 20 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=50: 1$ ) afforded the desired product $\mathbf{2 t}$ in $94 \%$ yield $(147.0 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.68(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.42-$ $7.35(\mathrm{~m}, 6 \mathrm{H}), 3.66(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 1.58-1.50(\mathrm{~m}, 2 \mathrm{H}), 1.43-1.34(\mathrm{~m}, 2 \mathrm{H}), 1.05(\mathrm{~s}, 9 \mathrm{H})$, 0.88 ( $\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 135.57$, 134.17, 129.47, 127.56, $63.69,34.76,26.86,19.23,19.01,13.91$. The spectroscopic data are in agreement with that previously reported. ${ }^{18}$


1-Phenylpentan-3-ol (2u). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0$ mmol ), dioxane ( 2.5 mL ), 5-phenylpent-1-en-3-ol ( $81.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 12 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=8: 1$ ) afforded the desired product $\mathbf{2 u}$ in $86 \%$ yield ( 71.0 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 7.29-7.25(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.15(\mathrm{~m}, 3 \mathrm{H}), 3.57-3.51(\mathrm{~m}, 1 \mathrm{H}), 2.83-$ $2.75(\mathrm{~m}, 1 \mathrm{H}), 2.69-2.62(\mathrm{~m}, 1 \mathrm{H}), 1.83-1.67(\mathrm{~m}, 3 \mathrm{H}), 1.58-1.41(\mathrm{~m}, 2 \mathrm{H}), 0.93(\mathrm{t}, J=7.6 \mathrm{~Hz}$, 3 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 142.18,128.34,128.30,125.69,72.51,38.49,31.99$, $30.18,9.78$. The spectroscopic data are in agreement with that previously reported. ${ }^{19}$


2v
tert-Butyldimethyl((1-phenylpentan-3-yl)oxy)silane (2v). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O} \quad(11.9 \mathrm{mg}, 0.05$ $\mathrm{mmol})$, Zinc powder $(65.4 \mathrm{mg}, 1.0 \mathrm{mmol})$, dioxane $(2.5 \mathrm{~mL})$, tert-butyldimethyl( $(5-$ phenylpent-1-en-3-yl)oxy)silane ( $138.3 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at 80 ${ }^{\circ} \mathrm{C}$ for 24 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate.

Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=50: 1$ ) afforded the desired product $2 \mathbf{2 v}$ in $96 \%$ yield ( 133.3 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 7.28-7.25(\mathrm{~m}, 2 \mathrm{H}), 7.20-7.14(\mathrm{~m}, 3 \mathrm{H}), 3.68-3.62(\mathrm{~m}, 1 \mathrm{H}), 2.73-2.55(\mathrm{~m}, 2 \mathrm{H}), 1.77-$ $1.71(\mathrm{~m}, 2 \mathrm{H}), 1.55-1.48(\mathrm{~m}, 2 \mathrm{H}), 0.92-0.87(\mathrm{~m}, 12 \mathrm{H}), 0.06(\mathrm{~s}, 3 \mathrm{H}), 0.05(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 142.85,128.32,128.31,125.60,73.06,38.49,31.79,29.69,25.94$, 18.17, 9.57, -4.36, -4.46. IR (neat): 3061, 3027, 2956, 2928, 2852, 1602, 1495, 1459, 1360, 1253, 1062, 1046, 1005, 833, 792, 772, 745, $697 \mathrm{~cm}^{-1}$. HRMS (ESI) calcd for $\mathrm{C}_{17} \mathrm{H}_{31} \mathrm{OSi}$ $[\mathrm{M}+\mathrm{H}]^{+}: 279.2139$, found 279.2136.


2w
$\boldsymbol{N}$-Propylaniline (2w). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), $N$-allylaniline ( $66.6 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 24 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=20: 1$ ) afforded the desired product $\mathbf{2 w}$ in $91 \%$ yield ( 61.3 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.16(\mathrm{t}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.67(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.59(\mathrm{bs}, 1 \mathrm{H})$, $3.06(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 1.67-1.58(\mathrm{~m}, 2 \mathrm{H}), 0.98(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 148.47,129.16,117.01,112.63,45.74,22.68,11.59$. The spectroscopic data are in agreement with that previously reported. ${ }^{17 a}$


2x
$N$-(Quinolin-8-yl)butyramide (2x). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( 65.4 $\mathrm{mg}, 1.0 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), $N$-(quinolin-8-yl)but-3-enamide ( $106.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 9 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=5: 1$ ) afforded the desired product $\mathbf{2 x}$ in $92 \%$ yield $(98.8 \mathrm{mg})$ as a
colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 9.80(\mathrm{~s}, 1 \mathrm{H}), 8.80-8.76(\mathrm{~m}, 2 \mathrm{H}), 8.09(\mathrm{~d}, J=8.4$ $\mathrm{Hz}, 1 \mathrm{H}), 7.52-7.38(\mathrm{~m}, 3 \mathrm{H}), 2.53(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.90-1.80(\mathrm{~m}, 2 \mathrm{H}), 1.05(\mathrm{t}, J=7.6 \mathrm{~Hz}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 171.57,147.91,138.08,136.14,134.34,127.70,127.18$, $121.38,121.16,116.15,39.94,18.96,13.67$. The spectroscopic data are in agreement with that previously reported. ${ }^{20}$

$2 y$
4-Methoxyphenyl butyrate (2y). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( 65.4 mg , 1.0 mmol ), dioxane ( 2.5 mL ), 4-methoxyphenyl but-3-enoate ( $96.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}$ $(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 10.5 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=10: 1$ ) afforded the desired product $2 \mathbf{y}$ in $78 \%$ yield $(76.2 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 6.98$ (d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.87 (d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), $3.77(\mathrm{~s}, 3 \mathrm{H}), 2.51(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.81-1.72(\mathrm{~m}, 2 \mathrm{H}), 1.03(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 172.41,157.05,144.12,122.22,114.28,55.42,36.03,18.37,13.53$. The spectroscopic data are in agreement with that previously reported. ${ }^{21}$


Diethyl propylphosphonate (2z). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( 65.4 mg , 1.0 mmol ), dioxane ( 2.5 mL ), diethyl allylphosphonate ( $89.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 21 h . Column chromatography on silica gel (eluent: petroleum ether to ethyl ether) afforded the desired product $\mathbf{2 z}$ in $80 \%$ yield ( 71.7 mg ) as a yellow oil. ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 4.15-4.04(\mathrm{~m}, 4 \mathrm{H}), 1.75-1.60(\mathrm{~m}, 4 \mathrm{H}), 1.33(\mathrm{t}, J=7.2 \mathrm{~Hz}, 6 \mathrm{H})$, 1.04-1.00 (m, 3H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 61.09\left(\mathrm{~d}, J_{\mathrm{C}-\mathrm{P}}=6.4 \mathrm{~Hz}\right), 27.46\left(\mathrm{~d}, J_{\mathrm{C}-\mathrm{P}}=\right.$ $139.5 \mathrm{~Hz}), 16.23\left(\mathrm{~d}, J_{\mathrm{C}-\mathrm{P}}=5.9 \mathrm{~Hz}\right), 15.90\left(\mathrm{~d}, J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}\right), 15.06\left(\mathrm{~d}, J_{\mathrm{C}-\mathrm{P}}=17.7 \mathrm{~Hz}\right)$. The spectroscopic data are in agreement with that previously reported. ${ }^{22}$

$2 z a$

4-Ethyl-4'-pentyl-1,1'-bi(cyclohexane) (2za). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ ( $11.9 \mathrm{mg}, 0.05 \mathrm{mmol}$ ), Zinc powder $(65.4 \mathrm{mg}, 1.0 \mathrm{mmol}),($ trans,trans $)-4-$ pentyl-4'-vinyl-1,1'-bi(cyclohexane) $(131.2 \mathrm{mg}$, $0.5 \mathrm{mmol})$, dioxane $(2.5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80{ }^{\circ} \mathrm{C}$ for 26 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{2 z a}$ in $58 \%$ yield $(77.1 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 1.76-1.68(\mathrm{~m}, 8 \mathrm{H}), 1.32-0.79$ $(\mathrm{m}, 28 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta 43.56,39.77,38.00,37.56,33.73,33.29,32.31$, $30.16,30.13,30.10,26.73,22.76,14.13,11.54$. IR (neat): $2958,2916,2848,1459,1447$, $1378,1219,1213,974,894,754,720 \mathrm{~cm}^{-1} . \operatorname{HRMS}(E I)$ calcd for $\mathrm{C}_{19} \mathrm{H}_{36}[\mathrm{M}]^{+}: 264.2817$, found 264.2812 .


4a

9-Ethyl-9H-carbazole (4a). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder $(65.4 \mathrm{mg}, 1.0$ mmol), 9-vinyl-9H-carbazole ( $96.6 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane $(2.5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 12 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=$ $30: 1)$ afforded the desired product $\mathbf{4 a}$ in $89 \%$ yield $(86.7 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 8.08(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.45-7.41(\mathrm{~m}, 2 \mathrm{H}), 7.35-7.33(\mathrm{~m}, 2 \mathrm{H}), 7.20(\mathrm{t}, J=$ $8.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.26(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.34(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right):$ $\delta 139.82,125.53,122.83,120.35,118.67,108.36,37.36,13.71$. The spectroscopic data are in agreement with that previously reported. ${ }^{23}$


4b
(Ethylsulfonyl)benzene (4b). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder $(65.4 \mathrm{mg}, 1.0$ mmol ), (vinylsulfonyl)benzene ( $84.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL}$ ) were stirred at $80^{\circ} \mathrm{C}$ for 21 h . The mixture was filtered through a pad of silica gel and washed with ethyl ether. The residue was purified by preparative TLC on silica gel (eluent: petroleum ether/ethyl ether $=1: 1$ ) to give the desired product $\mathbf{4 b}$ in $88 \%$ yield $(74.6 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.91(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.69-7.65(\mathrm{~m}, 1 \mathrm{H}), 7.60-7.56(\mathrm{~m}$, $2 \mathrm{H}), 3.13(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.28(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 138.29$, $133.58,129.14,128.03,50.42,7.31$. The spectroscopic data are in agreement with that previously reported. ${ }^{24}$


4c
Ethyldimethyl(phenyl)silane (4c). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol}$ ), Zinc powder ( 65.4 $\mathrm{mg}, 1.0 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), dimethyl(phenyl)(vinyl)silane ( $81.2 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 12 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{4 c}$ in $78 \%$ yield ( 64.3 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 7.52-7.49(\mathrm{~m}, 2 \mathrm{H}), 7.34-7.33(\mathrm{~m}, 3 \mathrm{H}), 0.95(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H})$, 0.73 ( $\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), $0.25(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 139.43,133.58,128.75$, 127.69, 7.41, 7.40, -3.56. IR (neat): 3069, 3048, 2954, 2907, 2873, 1426, 1247, 1112, 1012, 957, 832, 815, 774, 728, $698 \mathrm{~cm}^{-1}$. HRMS (EI) calcd for $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{Si}[\mathrm{M}]^{+}: 164.1021$, found 164.1027.

Due to the volatile nature of this compound, the NMR yield was determined. $93 \%$ NMR yield of the desired product $\mathbf{4 c}$ was obtained.


1,2-Diphenylethane (6a). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0$ $\mathrm{mmol})$, $E$-stilbene $(90.1 \mathrm{mg}, 0.5 \mathrm{mmol})$, dioxane $(2.5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 17 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{6 a}$ in $91 \%$ yield ( 82.8 mg ) as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 7.27-7.24 (m, 4H), 7.18-7.15 (m, 6H), $2.90(\mathrm{~s}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 141.72, 128.41, 128.29, 125.88, 37.90. The spectroscopic data are in agreement with that previously reported. ${ }^{25}$

When $Z$-stilbene was used as the substrate, the desired product $\mathbf{6 a}$ was obtained in $99 \%$ yield ( 90.5 mg ) as a white solid.


1-Methoxy-4-propylbenzene (2s). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( 65.4 mg , 1.0 mmol ), (E)-1-methoxy-4-(prop-1-en-1-yl)benzene ( $74.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 24 h . The mixture was filtered through a pad of silica gel and washed with petroleum ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl ether $=50: 1$ ) afforded the desired product $\mathbf{2 s}$ in $93 \%$ yield $(70.1 \mathrm{mg})$ as a colorless oil. The spectroscopic data are in agreement with that obtained from 1s.


6c
Cyclohexylbenzene (6c). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc flake ( $98.1 \mathrm{mg}, 1.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), 2,3,4,5-tetrahydro-1,1'-biphenyl ( $79.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL}$ ) were stirred at $80^{\circ} \mathrm{C}$ for 24 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{6 c}$ in $89 \%$ yield $(71.1 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 7.29-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.14(\mathrm{~m}, 3 \mathrm{H}), 2.51-2.45(\mathrm{~m}, 1 \mathrm{H}), 1.88-1.82(\mathrm{~m}, 4 \mathrm{H}), 1.76-$ $1.72(\mathrm{~m}, 1 \mathrm{H}), 1.47-1.33(\mathrm{~m}, 4 \mathrm{H}), 1.29-1.22(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta 148.05$, $128.25,126.80,125.74,44.57,34.44,26.90,26.15$. The spectroscopic data are in agreement
with that previously reported. ${ }^{26}$


6d
1-Tosylpiperidine ( $\mathbf{6 d}$ ). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder ( $98.1 \mathrm{mg}, 1.5$ mmol), 1-tosyl-1,2,3,6-tetrahydropyridine ( $118.7 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}$ $(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 23 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl ether $=5: 1$ ) afforded the desired product $\mathbf{6 d}$ in $81 \%$ yield $(96.5 \mathrm{mg})$ as a white soild. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.64(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.32(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.96$ $(\mathrm{t}, J=5.2 \mathrm{~Hz}, 4 \mathrm{H}), 2.43(\mathrm{~s}, 3 \mathrm{H}), 1.66-1.60(\mathrm{~m}, 4 \mathrm{H}), 1.43-1.38(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 143.20,132.96,129.41,127.54,46.80,25.00,23.33,21.37$. The spectroscopic data are in agreement with that previously reported. ${ }^{27}$


8a
Benzyl propionate (8a). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0$ mmol ), dioxane ( 2.5 mL ), benzyl acrylate ( $81.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 8 h . Column chromatography on silica gel (eluent: petroleum ether/ ethyl acetate $=20: 1$ ) afforded the desired product 8a in $91 \%$ yield $(74.5 \mathrm{mg})$ as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.35-7.30(\mathrm{~m}, 5 \mathrm{H}), 5.11(\mathrm{~s}, 2 \mathrm{H}), 2.37(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.15(\mathrm{t}, J=7.2 \mathrm{~Hz}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 174.15,136.02,128.43,128.06,66.00,27.46,8.99$. The spectroscopic data are in agreement with that previously reported. ${ }^{28}$


Ethyl 3-phenylpropanoate ( $\mathbf{8 b}$ ). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder ( 65.4 mg , $1.0 \mathrm{mmol})$, dioxane ( 2.5 mL ), ethyl cinnamate ( $88.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were
stirred at $80^{\circ} \mathrm{C}$ for 12 h . The mixture was filtered through a pad of silica gel and washed with petroleum ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ ethyl acetate $=20: 1$ ) afforded the desired product $\mathbf{8 b}$ in $96 \%$ yield $(85.6 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.29-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.17(\mathrm{~m}, 3 \mathrm{H}), 4.11(\mathrm{q}, J$ $=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.94(\mathrm{t}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.61(\mathrm{t}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 1.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 172.78,140.45,128.35,128.19,126.10,60.27,35.81,30.84$, 14.08. The spectroscopic data are in agreement with that previously reported. ${ }^{29}$


8c
1,3-Diphenylpropan-1-one (8c). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(23.8 \mathrm{mg}, 0.1 \mathrm{mmol})$, Al powder ( $53.9 \mathrm{mg}, 2.0$ $\mathrm{mmol}),(E)$-chalcone ( $104.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane $(2.5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 10 h in Schlenk tube $(10 \mathrm{~mL})$. The mixture was filtered through a pad of silica gel and washed with petroleum ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl ether $=30: 1$ ) afforded the desired product $\mathbf{8 c}$ in $86 \%$ yield ( 89.9 mg ) as a white soild. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.96(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.55(\mathrm{t}, J=7.6$ $\mathrm{Hz}, 1 \mathrm{H}), 7.45(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.32-7.24(\mathrm{~m}, 4 \mathrm{H}), 7.22-7.19(\mathrm{~m}, 1 \mathrm{H}), 3.30(\mathrm{t}, J=7.2 \mathrm{~Hz}$, 2 H ), 3.07 ( $\mathrm{t}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 199.18,141.25,136.79,133.03$, $128.57,128.49,128.39,128.00,126.10,40.42,30.07$. The spectroscopic data are in agreement with that previously reported. ${ }^{30}$


8d
Propiophenone (8d). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(23.8 \mathrm{mg}, 0.1 \mathrm{mmol})$, Al powder ( $53.9 \mathrm{mg}, 2.0 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), 1-phenylprop-2-en-1-one ( $66.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 10 h in Schlenk tube $(10 \mathrm{~mL})$. The mixture was filtered through a pad of silica gel and washed with petroleum ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=30: 1$ ) afforded the desired product $\mathbf{8 d}$ in $72 \%$
yield ( 48.0 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.96(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.55(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.45(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.00(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.23(\mathrm{t}, J=7.2 \mathrm{~Hz}$, 3 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 200.76, 136.82, 132.81, 128.48, 127.90, 31.71, 8.16. The spectroscopic data are in agreement with that previously reported. ${ }^{31}$


10a
$\mathbf{1 , 2 , 3}, 4-$ Tetrahydroquinoline (10a). To an oven dried sealable Schlenk tube ( 10 mL ) were added $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(7.1 \mathrm{mg}, 0.03 \mathrm{mmol})$, Zinc powder ( $58.9 \mathrm{mg}, 0.9 \mathrm{mmol}$ ), dioxane ( 1.5 mL ), quinoline ( $38.7 \mathrm{mg}, 0.3 \mathrm{mmol}$ ), $\mathrm{NEt}_{3}(30.4 \mathrm{mg}, 0.3 \mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.3 \mathrm{~mL})$ were added sequentially under argon. The tube was sealed with a teflon screwcap and the mixture was stirred at $100^{\circ} \mathrm{C}$ for 10 h , then the mixture was filtered through a pad of silica gel and washed with ethyl acetate. Note: the tube should be tightly sealed, which is highly important for reproducibility. The solvent was evaporated under the reduced pressure and the residue was purified by column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=5: 1$ ) to give the desired product $\mathbf{1 0 a}$ in $76 \%$ yield $(30.4 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 6.97-6.93(\mathrm{~m}, 2 \mathrm{H}), 6.59(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{br}, 1 \mathrm{H})$, $3.28(\mathrm{t}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.75(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 1.96-1.90(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 144.71,129.45,126.65,121.36,116.85,114.11,41.90,26.90,22.10$. The spectroscopic data are in agreement with that previously reported. ${ }^{32}$

When the reaction was carried out in a 25 mL sealable Schlenk tube, $55 \%$ yield of $\mathbf{1 0 a}$ was obtained. The results indicated that the yield of the product 10a was decreased by using a larger tube.


10b
$\mathbf{1 , 2 , 3 , 4}$-Tetrahydroquinoxaline (10b). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(7.1 \mathrm{mg}, 0.03 \mathrm{mmol})$, Zinc powder ( 58.9 $\mathrm{mg}, 0.9 \mathrm{mmol}$ ), quinoxaline ( $39.0 \mathrm{mg}, 0.3 \mathrm{mmol}$ ), dioxane $(1.5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$ were
stirred at $100^{\circ} \mathrm{C}$ for 11 h in Schlenk tube $(10 \mathrm{~mL})$. The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl ether $=2: 1$ ) to give the desired product $\mathbf{1 0 b}$ in $52 \%$ yield $(20.8 \mathrm{mg})$ as a white soild. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 6.59-6.57(\mathrm{~m}, 2 \mathrm{H}), 6.50-6.49(\mathrm{~m}, 2 \mathrm{H}), 3.65$ (bs, 2 H ), 3.42 (s, 4H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 133.62,118.69,114.64,41.31$. The spectroscopic data are in agreement with that previously reported. ${ }^{33}$


10c
3,8a-Diphenyl-8,8a-dihydroindolizin-1(7H)-one (10c). To an oven dried sealable Schlenk tube $(10 \mathrm{~mL})$ were added $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(7.1 \mathrm{mg}, 0.03 \mathrm{mmol})$, Zinc powder ( $39.2 \mathrm{mg}, 0.6$ mmol ), 3,8a-diphenylindolizin-1( $8 \mathrm{a} H$ )-one ( $85.6 \mathrm{mg}, 0.3 \mathrm{mmol}$ ), dioxane ( 1.5 mL ), $\mathrm{NEt}_{3}$ ( $30.4 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.3 \mathrm{~mL}$ ) were added sequentially under argon. The tube was sealed with a teflon screwcap and the mixture was stirred at $100^{\circ} \mathrm{C}$ for 10 h , the mixture was filtered through a pad of silica gel and washed with ethyl acetate. Note: the tube should be tightly sealed, which is highly important for reproducibility. The solvent was evaporated under the reduced pressure and the residue was purified by column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=5: 1$ ) to give the desired product $\mathbf{1 0 c}$ in $65 \%$ yield $(55.9 \mathrm{mg})$ as a yellow soild. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.60-7.54(\mathrm{~m}, 7 \mathrm{H}), 7.38(\mathrm{t}, J=7.2$ $\mathrm{Hz}, 2 \mathrm{H}), 7.32-7.28(\mathrm{~m}, 1 \mathrm{H}), 6.73(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.36(\mathrm{~s}, 1 \mathrm{H}), 4.96(\mathrm{t}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H})$, 2.69-2.65 (m, 1H), 2.04-1.73 (m, 3H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 200.86,172.66,136.79$, $130.96,129.81,128.96,128.60,128.03,127.56,125.29,123.71,109.50,101.60,71.02,31.16$, 19.71. IR (neat): $3095,3066,3003,2966,2917,2848,1668,1637,1581,1537,1487,1449$, $1407,1384,1298,1210,1143,1124,1071,1029,1017,994,883,828,809,775,767,734$, 696, 669, $656 \mathrm{~cm}^{-1}$. HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{NO}[\mathrm{M}+\mathrm{H}]^{+}: 288.1383$, found 288.1386 .

## Ni-Catalyzed transfer hydrogenation of alkynes.

## Typical procedure for the synthesis of 2d from alkyne 11a.



To an oven dried Schlenk tube ( 25 mL ) were added $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $98.1 \mathrm{mg}, 1.5 \mathrm{mmol}$ ). The Schlenk tube was capped with a rubber septum, evacuated and back filled with argon for three times. Then dioxane ( 2.5 mL ), 1-ethynyl-4methoxybenzene ( $66.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were added sequentially under argon. The tube cap was then securely fitted and sealed with electrical tape, and the stopcock valve on the sidearm of the Schlenk tube was closed. After the mixture was stirred at $80^{\circ} \mathrm{C}$ for 16.5 h , it was filtered through a pad of silica gel and washed with petroleum ether and ethyl ether. Note: a stirring speed above $\mathbf{6 0 0} \mathbf{~ r p m}$ is highly important for reproducibility. The solvent was evaporated under the reduced pressure and the residue was purified by column chromatography on silica gel (eluent: petroleum ether to petroleum ether/ethyl ether = $50: 1)$ to give the desired product $\mathbf{2 d}$ in $71 \%$ yield ( 48.1 mg ) as a colorless oil. The spectroscopic data are in agreement with that obtained from 1d.


4-Ethylaniline (2e). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ ( $5.9 \mathrm{mg}, 0.025 \mathrm{mmol}$ ), Zinc powder ( $98.1 \mathrm{mg}, 1.5 \mathrm{mmol}$ ), 4ethynylaniline ( $58.6 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 22 h . The mixture was filtered through a pad of silica gel and washed with ethyl ether. Column chromatography on silica gel (eluent: petroleum ether/ethyl ether $=1: 1$ ) afforded the desired product 2 e in $93 \%$ yield ( 56.4 mg ) as a yellow oil. The spectroscopic data are in agreement with that obtained from $\mathbf{1 e}$.


12c
Ethyl 4-ethylbenzoate (12c). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder $(98.1 \mathrm{mg}, 1.5$
mmol ), ethyl 4-ethynylbenzoate ( $87.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 24 h . The mixture was filtered through a pad of silica gel and washed with petroleum ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=20: 1$ ) afforded the desired product $\mathbf{1 2 c}$ in $90 \%$ yield $(80.5 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.96(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.24(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $2 \mathrm{H}), 4.36(\mathrm{q}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.69(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.38(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 1.24(\mathrm{t}, J=$ $7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 166.58,149.50,129.57,127.90,127.73,60.62$, $28.84,15.15,14.25$. The spectroscopic data are in agreement with that previously reported. ${ }^{34}$


4-Ethyl-1,1'-biphenyl (2j). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder ( $98.1 \mathrm{mg}, 1.5$ mmol ), 4-ethynyl-1,1'-biphenyl ( $89.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL}$ ) were stirred at $80^{\circ} \mathrm{C}$ for 24 h . The mixture was filtered through a pad of silica gel and washed with petroleum ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{2} \mathbf{j}$ in $89 \%$ yield ( 80.7 mg ) as a white soild. The spectroscopic data are in agreement with that obtained from $\mathbf{1} \mathbf{j}$.


2a

2-Ethylnaphthalene (2a). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $98.1 \mathrm{mg}, 1.5$ mmol), 2-ethynylnaphthalene ( $76.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 10 h . The mixture was filtered through a pad of silica gel and washed with petroleum ether. Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product 2a in $91 \%$ yield ( 71.3 mg ) as a colorless oil. The spectroscopic data are in agreement with that obtained from 1a.

$12 f$

1-Methoxy-4-propoxybenzene (12f). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( 98.1 $\mathrm{mg}, 1.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), 1-methoxy-4-(prop-2-yn-1-yloxy)benzene ( $81.1 \mathrm{mg}, 0.5$ $\mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 22.5 h , the mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=10: 1$ ) afforded the desired product $\mathbf{1 2 f}$ in $68 \%$ yield ( 56.1 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 6.82(\mathrm{~s}, 4 \mathrm{H}), 3.85(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.75(\mathrm{~s}$, $3 \mathrm{H}), 1.82-1.73(\mathrm{~m}, 2 \mathrm{H}), 1.02(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 153.57$, $153.21,115.32,114.51,70.05,55.62,22.63,10.48$. The spectroscopic data are in agreement with that previously reported. ${ }^{35}$


2-Propylisoindoline-1,3-dione (12g). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder ( 98.1 $\mathrm{mg}, 1.5 \mathrm{mmol}$ ), 2-(prop-2-yn-1-yl)isoindoline-1,3-dione ( $92.6 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 $\mathrm{mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 23 h . The mixture was filtered through a pad of silica gel and washed with ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether/ethyl acetate $=10: 1$ ) afforded the desired product $\mathbf{1 2 g}$ in $86 \%$ yield ( 81.2 mg ) as a white soild. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.85-7.83(\mathrm{~m}, 2 \mathrm{H}), 7.73-7.71(\mathrm{~m}, 2 \mathrm{H}), 3.66(\mathrm{t}$, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.76-1.67(\mathrm{~m}, 2 \mathrm{H}), 0.96(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ $168.30,133.69,131.99,122.98,39.43,21.77,11.19$. The spectroscopic data are in agreement with that previously reported. ${ }^{36}$


12h

1-Ethylcyclohexan-1-ol (12h). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder $(98.1 \mathrm{mg}$, $1.5 \mathrm{mmol})$, dioxane $(2.5 \mathrm{~mL})$, 1-ethynylcyclohexan-1-ol $(62.0 \mathrm{mg}, 0.5 \mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.8$ mL ) were stirred at $80^{\circ} \mathrm{C}$ for 24 h , the mixture was filtered through a pad of silica gel and washed with petroleum ether and ethyl acetate. Column chromatography on silica gel (eluent: petroleum ether /ethyl acetate $=50: 1)$ afforded the desired product $\mathbf{1 2 h}$ in $67 \%$ yield $(43.2 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 1.63-1.37(\mathrm{~m}, 11 \mathrm{H}), 1.31-1.24(\mathrm{~m}, 2 \mathrm{H}), 0.90$ $(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta 71.38,36.81,34.63,25.82,22.16,7.17$. The spectroscopic data are in agreement with that previously reported. ${ }^{37}$


12i
Butylbenzene (12i). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder $(98.1 \mathrm{mg}, 1.5 \mathrm{mmol})$, dioxane $(2.5 \mathrm{~mL})$, but-3-yn-1-ylbenzene $(65.1 \mathrm{mg}, 0.5 \mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 24 h , the mixture was filtered through a pad of silica gel and washed with petroleum ether. Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{1 2 i}$ in $47 \%$ yield $(31.7 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ $7.27(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.20-7.14(\mathrm{~m}, 3 \mathrm{H}), 2.60(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.63-1.56(\mathrm{~m}, 2 \mathrm{H}), 1.40-$ $1.31(\mathrm{~m}, 2 \mathrm{H}), 0.92(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta 142.88,128.39,128.19$, $125.52,35.66,33.68,22.37,13.96$. The spectroscopic data are in agreement with that previously reported. ${ }^{38}$

Due to the volatile nature of this compound, the NMR yield was determined. $87 \%$ NMR yield of the desired product $\mathbf{1 2 i}$ was obtained.


12j
Ethyltriisopropylsilane (12j). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder $(98.1 \mathrm{mg}, 1.5$
mmol ), dioxane ( 2.5 mL ), ethynyltriisopropylsilane ( $91.2 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 22 h , the mixture was filtered through a pad of silica gel and washed with pentane. Column chromatography on silica gel (eluent: pentane) afforded the desired product $\mathbf{1 2 j}$ in $64 \%$ yield $(59.9 \mathrm{mg})$ as a white soild. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 1.04-0.97$ ( $\mathrm{m}, 24 \mathrm{H}$ ), $0.60(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 18.84,10.79,8.21,0.85$. The spectroscopic data are in agreement with that previously reported. ${ }^{39}$


1,2-Diphenylethane (6a). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc flake ( $98.1 \mathrm{mg}, 1.5 \mathrm{mmol}$ ), 1,2-diphenylethyne ( $89.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 24 h . The mixture was filtered through a pad of silica gel and washed with petroleum ether. Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product 6a in $94 \%$ yield ( 85.4 mg ) as a white soild. The spectroscopic data are in agreement with that obtained from 5a.

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Hexylbenzene (12I). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc powder ( $98.1 \mathrm{mg}, 1.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), hex-1-yn-1-ylbenzene ( $79.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 24 h . The mixture was filtered through a pad of silica gel and washed with petroleum ether. Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{1 2 1}$ in $93 \%$ yield ( 75.3 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 7.28-7.24 (m, 2H), 7.18-7.14 (m, 3H), $2.59(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.64-1.57(\mathrm{~m}, 2 \mathrm{H}), 1.36-1.30$ $(\mathrm{m}, 6 \mathrm{H}), 0.88(\mathrm{t}, J=6.0 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 142.92,128.38$, 128.19, $125.52,36.00,31.75,31.52,29.03,22.63,14.11$. The spectroscopic data are in agreement with that previously reported. ${ }^{10}$


12m

Dodecane (12m). $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ ( $11.9 \mathrm{mg}, 0.05 \mathrm{mmol}$ ), Zinc flake $(98.1 \mathrm{mg}, 1.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ), dodec-6-yne ( $83.2 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), and $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were stirred at $80^{\circ} \mathrm{C}$ for 24 h . Column chromatography on silica gel (eluent: petroleum ether) afforded the desired product $\mathbf{1 2 m}$ in $75 \%$ yield $(63.6 \mathrm{mg})$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 1.26(\mathrm{~s}$, 20H), 0.90-0.86 (m, 6H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 31.99,29.77,29.72,29.43,22.74$, 14.13. The spectroscopic data are in agreement with that previously reported. ${ }^{41}$

## Control experiments.

(1) Deuterium labeling experiments.
$\mathrm{NiCl}_{2}$-catalyzed deuterium labeling experiment of 11 k with $\mathbf{D}_{\mathbf{2}} \mathrm{O}$.


To an oven dried Schlenk tube were added $\mathrm{NiCl}_{2}(6.5 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc flake ( 98.1 mg , 1.5 mmol ), 1,2-diphenylethyne ( $89.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ). The tube was evacuated and back filled with argon for three times, then dioxane $(2.5 \mathrm{~mL})$ and $\mathrm{D}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were added sequentially under argon. The tube was sealed and the mixture was stirred at $80^{\circ} \mathrm{C}$ for 24 h , the mixture was filtered through a pad of silica gel and washed with petroleum ether. The solvent was evaporated under the reduced pressure and the residue was purified by column chromatography on silica gel (eluent: petroleum ether) to give the desired product 6a- $d$ in $86 \%$ yield ( 80.0 mg ) as a white soild. ${ }^{11} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.28-7.24(\mathrm{~m}, 4 \mathrm{H}), 7.19-7.15$ (m, 6H), 2.87 ( $\mathrm{s}, 0.36 \mathrm{H}$ ). ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 141.64, 128.40, 128.29, 125.87, 37.77-36.63 (m). HRMS (EI) calcd for $\mathrm{C}_{14} \mathrm{H}_{10} \mathrm{D}_{4}[\mathrm{M}]^{+}: 186.1347$, found 186.1351.

## (2) KIE experiment.



To an oven dried Schlenk tube were added $\mathrm{NiCl}_{2}(6.5 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zinc flake ( 98.1
$\mathrm{mg}, 1.5 \mathrm{mmol}$ ），1，2－diphenylethyne（ $89.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ）．The tube was evacuated and back filled with argon for three times，then dioxane（ 2.5 mL ）， $\mathrm{H}_{2} \mathrm{O}(396 \mathrm{mg}, 22 \mathrm{mmol}, 44$ equiv） and $\mathrm{D}_{2} \mathrm{O}$（ $440 \mathrm{mg}, 22 \mathrm{mmol}, 44$ equiv）were added sequentially under argon．The tube was sealed and the mixture was stirred at $80^{\circ} \mathrm{C}$ for 24 h ，the mixture was filtered through a pad of silica gel and washed with petroleum ether．The solvent was evaporated under the reduced pressure and the residue was purified by column chromatography on silica gel（eluent： petroleum ether）to give the desired product $\mathbf{6 a -} d^{\prime}$ in $90 \%$ yield（ 82.4 mg ）as a colorless oil．
（目标产物 FW：183．5）${ }^{1} \mathrm{H}$ NMR（ $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ）：$\delta$ 7．27－7．24（m，4H），7．18－7．15（m， 6 H ），2．90－2．87（m，2．77H）；${ }^{13} \mathrm{C}$ NMR（ $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ）：$\delta$ 141．71，141．68，128．40，128．28， 125．87，37．90，37．83，37．76， $37.53\left(\mathrm{t}, J_{C-D}=19.5 \mathrm{~Hz}\right.$ ）， $37.46\left(\mathrm{t}, J_{C-D}=19.5 \mathrm{~Hz}\right)$ ．HRMS（EI） calcd for $\mathrm{C}_{14} \mathrm{H}_{10} \mathrm{D}_{4}[\mathrm{M}]^{+}:$186．1347，found 186．1345．

## Kinetic isotopic effect with individual reaction．



Following the general procedure：（1）The reduction of diphenylacetylene with $\mathrm{H}_{2} \mathrm{O}$ ： $\mathrm{NiCl}_{2}$（ $6.5 \mathrm{mg}, 0.05 \mathrm{mmol}$ ），Zinc flake（ $98.1 \mathrm{mg}, 1.5 \mathrm{mmol}$ ），1，2－diphenylethyne（ $89.1 \mathrm{mg}, 0.5$ mmol ），dioxane（ 2.5 mL ）and $\mathrm{H}_{2} \mathrm{O}\left(792 \mathrm{mg}, 44 \mathrm{mmol}, 88\right.$ equiv）were stirred at $80^{\circ} \mathrm{C}$ for 20 min ．（2）The reduction of diphenylacetylene with $\mathrm{D}_{2} \mathrm{O}: \mathrm{NiCl}_{2}(6.5 \mathrm{mg}, 0.05 \mathrm{mmol})$ ，Zinc flake （ $98.1 \mathrm{mg}, 1.5 \mathrm{mmol}$ ），1，2－diphenylethyne（ $89.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ），dioxane（ 2.5 mL ）and $\mathrm{D}_{2} \mathrm{O}$ （ $881 \mathrm{mg}, 44 \mathrm{mmol}, 88$ equiv）were stirred at $80^{\circ} \mathrm{C}$ for 20 min ．After quenching the reactions， the mixture of the two reactions was combined and purified by preparative TLC on silica gel （eluent：hexane）to give a mixture of $\mathbf{6 a}$ and $\mathbf{6 a -} d$ ，along with a byproduct of $E$－stilbene．The ratio of $\mathbf{6 a}$ and $\mathbf{6 a}-d$ was determined by ${ }^{1} \mathrm{H}$ NMR，and the KIE of $k_{\mathrm{H}} / k_{\mathrm{D}}$ was found to be 3．0．

## （3）Deuterium labeling experiment．

## $\mathrm{NiCl}_{2}$-catalyzed deuterium labeling experiment of 1 a with $\mathrm{D}_{\mathbf{2}} \mathrm{O}$.



To an oven dried Schlenk tube were added $\mathrm{NiCl}_{2}$ ( $3.2 \mathrm{mg}, 0.05 \mathrm{mmol}$ ), Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), 2-vinylnaphthalene ( $77.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ). The tube was evacuated and back filled with argon for three times, then dioxane $(2.5 \mathrm{~mL})$ and $\mathrm{D}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ were added sequentially under argon. The tube was sealed and the mixture was stirred at $80^{\circ} \mathrm{C}$ for 10 h , the mixture was filtered through a pad of silica gel and washed with petroleum ether. The solvent was evaporated under the reduced pressure and the residue was purified by column chromatography on silica gel (eluent: petroleum ether) to give the desired product 2a-d in $92 \%$ yield ( 73.1 mg ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.77-7.71(\mathrm{~m}, 3 \mathrm{H}), 7.58(\mathrm{~s}$, $1 \mathrm{H}), 7.43-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.30(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.80-2.75(\mathrm{~m}, 1.59 \mathrm{H}), 1.34-1.26(\mathrm{~m}, 1.54 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 141.70,133.66,131.90,127.76,127.57,127.38,127.04$, 125.79, 125.50, 124.96, 29.01, $28.84\left(\mathrm{t}, J_{C-D}=8.3 \mathrm{~Hz}\right), 15.52,15.44,15.22\left(\mathrm{t}, J_{C-D}=19.4 \mathrm{~Hz}\right)$. HRMS (EI) calcd for $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{D}_{2}[\mathrm{M}]^{+}: 158.1065$, found 158.1060.

## (4) Mercury poisoning experiment.



To an oven dried Schlenk tube were added $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, $\mathrm{Zn}(65.4$ $\mathrm{mg}, 1.0 \mathrm{mmol})$, $\mathbf{1 a}(77.1 \mathrm{mg}, 0.5 \mathrm{mmol})$. The tube was evacuated and back filled with argon for three times, then dioxane $(2.5 \mathrm{~mL}), \mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ and $\mathrm{Hg}(501.5 \mathrm{mg}, 2.5 \mathrm{mmol}, 100$ equiv respect to Ni ) were added sequentially under argon. The tube was sealed and the mixture was stirred at $80^{\circ} \mathrm{C}$ for 10 h , the mixture was filtered through a pad of silica gel and washed with petroleum ether. The solvent was evaporated under the reduced pressure and the residue was
dissolved in $\mathrm{CDCl}_{3}$. The NMR yields were obtained by ${ }^{1} \mathrm{H}$ NMR analysis of the crude mixture using 1, 3, 5-methoxybenzene ( $84.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) as an internal standard. $14 \%$ NMR yield of the desired product 2a and $80 \%$ NMR yield of the $\mathbf{1 a}$ was observed.

## (5) Gas-chromatographic analysis of $\mathbf{H}_{\mathbf{2}}$ gas:



The reaction was conducted in an oven-dried screw-cap vial ( 12 mL ) equipped with a magnetic stir bar. In a nitrogen-filled glove box, $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(5.9 \mathrm{mg}, 0.025 \mathrm{mmol})$, Zinc powder ( $65.4 \mathrm{mg}, 1.0 \mathrm{mmol}$ ), 2-vinylnaphthalene ( $77.1 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), dioxane ( 2.5 mL ) were added sequentially to a screw-cap vial. The vial cap was then securely fitted and taken outside the glove box, then $\mathrm{H}_{2} \mathrm{O}(0.8 \mathrm{~mL})$ was added to the vial. After the reaction mixture was stirred at $80^{\circ} \mathrm{C}$ for 1 h . The gas in the vial is analyzed by the gas chromatography equipped with TCD detector using $\mathrm{N}_{2}$ as the carrier gas, and $\mathrm{H}_{2}$ was detected. $\mathrm{H}_{2}$ was also detected in the absence of 1a. See Figure S1.


Figure S1. The gas chromatography of the $\mathrm{H}_{2}$ standard (above). The gas chromatography of the gas in the reaction vial containing alkyne 1a (middle). The gas chromatography of the gas under the reaction conditions without addition of alkyne 1a (below). The results indicated that $\mathrm{H}_{2}$ was formed during the reaction. The retention time of $\mathrm{H}_{2}$ is around $1.5 \mathrm{~min}, \mathrm{O}_{2}$ is around 1.7 min , and $\mathrm{N}_{2}$ is around 2.1 min .

## 1 mmol scale reaction



To an oven dried Schlenk tube were added $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(11.9 \mathrm{mg}, 0.05 \mathrm{mmol})$, Zn ( 130.8 $\mathrm{mg}, 2 \mathrm{mmol}$ ), and $\mathbf{1 j}$ ( $180.3 \mathrm{mg}, 1 \mathrm{mmol}$ ) under air. The Schlenk tube was capped with a rubber septum, evacuated and back filled with argon for three times. Then dioxane ( 5 mL ) and $\mathrm{H}_{2} \mathrm{O}(1.6 \mathrm{~mL})$ were added sequentially under argon. The tube cap was then securely fitted and sealed with electrical tape, and the stopcock valve on the sidearm of the Schlenk tube was closed. After the mixture was stirred at $80^{\circ} \mathrm{C}$ for 12 h , it was filtered through a pad of silica gel and washed with petroleum ether. The solvent was evaporated under the reduced pressure and the residue was purified by column chromatography on silica gel (eluent: petroleum ether) to give the desired product $\mathbf{2} \mathbf{j}$ in $94 \%$ yield ( 171.7 mg ) as a white soild.

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## X-ray crystallographic structure and data for compound 10 c .

The ellipsoid contour is $30 \%$.


10c
CCDC 1908616


Figure S2. X-ray crystal structure of compound 10c

Table S2. Crystal data and structure refinement for d8v19160.

Identification code
d8v19160

Empirical formula
Formula weight
C20 H17 N O
-
287.35

Temperature
Wavelength
Crystal system
Space group
Unit cell dimensions
194(2) K
$0.71073 \AA$
Orthorhombic
Pbca

$$
\begin{array}{ll}
a=14.5964(5) \AA & \alpha=90^{\circ} . \\
b=13.9382(5) \AA & \beta=90^{\circ} .
\end{array}
$$

|  | $\mathrm{c}=14.6669(5) \AA \quad \gamma=90^{\circ}$. |
| :---: | :---: |
| Volume | 2983.94(18) $\AA^{3}$ |
| Z | 8 |
| Density (calculated) | $1.279 \mathrm{Mg} / \mathrm{m}^{3}$ |
| Absorption coefficient | $0.078 \mathrm{~mm}^{-1}$ |
| $\mathrm{F}(000)$ | 1216 |
| Crystal size | $0.180 \times 0.160 \times 0.130 \mathrm{~mm}^{3}$ |
| Theta range for data collection | 2.452 to $25.993^{\circ}$. |
| Index ranges | $-18<=\mathrm{h}<=15,-17<=\mathrm{k}<=15,-18<=1<=17$ |
| Reflections collected | 28649 |
| Independent reflections | $2907[\mathrm{R}(\mathrm{int})=0.0400]$ |
| Completeness to theta $=25.242^{\circ}$ | 99.4 \% |
| Absorption correction | Semi-empirical from equivalents |
| Max. and min. transmission | 0.7456 and 0.6536 |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{2}$ |
| Data / restraints / parameters | 2907 / 0 / 200 |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.034 |
| Final R indices [ $\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})$ ] | $\mathrm{R} 1=0.0359, \mathrm{wR} 2=0.0880$ |
| R indices (all data) | $\mathrm{R} 1=0.0422, \mathrm{wR} 2=0.0931$ |
| Extinction coefficient | 0.018(2) |
| Largest diff. peak and hole | 0.254 and $-0.156 \mathrm{e} . \AA^{-3}$ |

Table S3. Atomic coordinates $\left(\mathrm{x} 10^{4}\right)$ and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for d 8 v 19160 . $\mathrm{U}(\mathrm{eq})$ is defined as one third of the trace of the orthogonalized $\mathrm{U}^{\mathrm{ij}}$ tensor.

|  | x | y | z | $\mathrm{U}(\mathrm{eq})$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O}(1)$ | 4758(1) | 7970(1) | 3759(1) | 36(1) |
| $\mathrm{N}(1)$ | 5932(1) | 6078(1) | 2663(1) | 27(1) |
| $\mathrm{C}(1)$ | 5919(1) | 5385(1) | 1967(1) | 32(1) |
| C(2) | 5421(1) | 5499(1) | 1217(1) | 40(1) |
| C(3) | 4781(1) | 6326(1) | 1086(1) | 40(1) |
| C(4) | 4542(1) | 6773(1) | 2009(1) | 33(1) |
| C(5) | 5420(1) | 6976(1) | 2549(1) | 26(1) |
| C(6) | 5216(1) | 7267(1) | 3549(1) | 27(1) |
| C(7) | 5671(1) | 6576(1) | 4113(1) | 28(1) |
| C(8) | 6062(1) | 5893(1) | 3572(1) | 25(1) |
| C(9) | 6593(1) | 5049(1) | 3876(1) | 26(1) |
| C(10) | 6200(1) | 4377(1) | 4456(1) | 29(1) |
| C (11) | 6708(1) | 3607(1) | 4768(1) | 37(1) |
| C(12) | 7618(1) | 3514(1) | 4518(1) | 42(1) |
| C(13) | 8016(1) | 4178(1) | 3945(1) | 42(1) |
| C(14) | 7503(1) | 4938(1) | 3614(1) | 35(1) |
| C(15) | 6012(1) | 7744(1) | 2094(1) | 26(1) |
| C(16) | 6916(1) | 7584(1) | 1839(1) | 32(1) |
| C(17) | 7425(1) | 8301(1) | 1422(1) | 35(1) |
| C(18) | 7037(1) | 9187(1) | 1252(1) | 33(1) |
| C(19) | 6142(1) | 9360(1) | 1518(1) | 32(1) |
| C(20) | 5633(1) | 8648(1) | 1937(1) | 30(1) |

Table S4. Bond lengths $[\AA]$ and angles $\left[{ }^{\circ}\right]$ for d 8 v 19160 .

| $\mathrm{O}(1)-\mathrm{C}(6)$ | 1.2258(14) |
| :---: | :---: |
| $\mathrm{N}(1)-\mathrm{C}(8)$ | $1.3708(15)$ |
| $\mathrm{N}(1)-\mathrm{C}(1)$ | 1.4067(15) |
| $\mathrm{N}(1)-\mathrm{C}(5)$ | 1.4674(14) |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.3276(18)$ |
| $\mathrm{C}(1)-\mathrm{H}(1)$ | 0.9500 |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | 1.496(2) |
| $\mathrm{C}(2)-\mathrm{H}(2)$ | 0.9500 |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.5302(18)$ |
| $\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~A})$ | 0.9900 |
| $\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~B})$ | 0.9900 |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.5337(16)$ |
| $\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~A})$ | 0.9900 |
| $\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~B})$ | 0.9900 |
| $\mathrm{C}(5)-\mathrm{C}(15)$ | $1.5289(15)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | 1.5513(16) |
| $\mathrm{C}(6)-\mathrm{C}(7)$ | $1.4325(16)$ |
| $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.3634(16)$ |
| $\mathrm{C}(7)-\mathrm{H}(7)$ | 0.9500 |
| $\mathrm{C}(8)-\mathrm{C}(9)$ | $1.4782(16)$ |
| $\mathrm{C}(9)-\mathrm{C}(10)$ | 1.3888(16) |
| $\mathrm{C}(9)-\mathrm{C}(14)$ | 1.3920(17) |
| $\mathrm{C}(10)-\mathrm{C}(11)$ | 1.3824(17) |
| $\mathrm{C}(10)-\mathrm{H}(10)$ | 0.9500 |
| $\mathrm{C}(11)-\mathrm{C}(12)$ | 1.383(2) |
| $\mathrm{C}(11)-\mathrm{H}(11)$ | 0.9500 |
| $\mathrm{C}(12)-\mathrm{C}(13)$ | 1.380(2) |
| $\mathrm{C}(12)-\mathrm{H}(12)$ | 0.9500 |
|  |  |


| $\mathrm{C}(13)-\mathrm{C}(14)$ | $1.3839(18)$ |
| :---: | :---: |
| $\mathrm{C}(13)-\mathrm{H}(13)$ | 0.9500 |
| $\mathrm{C}(14)-\mathrm{H}(14)$ | 0.9500 |
| $\mathrm{C}(15)-\mathrm{C}(16)$ | 1.3897 (17) |
| $\mathrm{C}(15)-\mathrm{C}(20)$ | $1.3944(16)$ |
| $\mathrm{C}(16)-\mathrm{C}(17)$ | 1.3880(17) |
| $\mathrm{C}(16)-\mathrm{H}(16)$ | 0.9500 |
| $\mathrm{C}(17)-\mathrm{C}(18)$ | 1.3820 (18) |
| $\mathrm{C}(17)-\mathrm{H}(17)$ | 0.9500 |
| $\mathrm{C}(18)-\mathrm{C}(19)$ | 1.3854(19) |
| $\mathrm{C}(18)-\mathrm{H}(18)$ | 0.9500 |
| C(19)-C(20) | $1.3833(17)$ |
| $\mathrm{C}(19)-\mathrm{H}(19)$ | 0.9500 |
| $\mathrm{C}(20)-\mathrm{H}(20)$ | 0.9500 |
| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{C}(1)$ | 125.40(10) |
| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{C}(5)$ | 109.96(9) |
| $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(5)$ | 119.73(9) |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{N}(1)$ | 121.79(12) |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{H}(1)$ | 119.1 |
| $\mathrm{N}(1)-\mathrm{C}(1)-\mathrm{H}(1)$ | 119.1 |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 122.72(12) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{H}(2)$ | 118.6 |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{H}(2)$ | 118.6 |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 110.05(10) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~A})$ | 109.7 |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~A})$ | 109.7 |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~B})$ | 109.7 |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~B})$ | 109.7 |
| $\mathrm{H}(3 \mathrm{~A})-\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~B})$ | 108.2 |


| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 109.94(10) |
| :---: | :---: |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~A})$ | 109.7 |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~A})$ | 109.7 |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~B})$ | 109.7 |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~B})$ | 109.7 |
| $\mathrm{H}(4 \mathrm{~A})-\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~B})$ | 108.2 |
| $\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(15)$ | 111.06(9) |
| $\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(4)$ | 109.14(9) |
| $\mathrm{C}(15)-\mathrm{C}(5)-\mathrm{C}(4)$ | 112.09(9) |
| $\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(6)$ | 102.29(9) |
| $\mathrm{C}(15)-\mathrm{C}(5)-\mathrm{C}(6)$ | 109.76(9) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 112.10(10) |
| $\mathrm{O}(1)-\mathrm{C}(6)-\mathrm{C}(7)$ | 130.23(11) |
| $\mathrm{O}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | 123.46(10) |
| $\mathrm{C}(7)-\mathrm{C}(6)-\mathrm{C}(5)$ | 106.30(9) |
| $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{C}(6)$ | 109.13(10) |
| $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{H}(7)$ | 125.4 |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{H}(7)$ | 125.4 |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{N}(1)$ | 112.14(10) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | 126.86(10) |
| $\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{C}(9)$ | 120.98(10) |
| $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{C}(14)$ | 119.22(11) |
| $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{C}(8)$ | 120.32(10) |
| $\mathrm{C}(14)-\mathrm{C}(9)-\mathrm{C}(8)$ | 120.41(10) |
| $\mathrm{C}(11)-\mathrm{C}(10)-\mathrm{C}(9)$ | 120.29(11) |
| $\mathrm{C}(11)-\mathrm{C}(10)-\mathrm{H}(10)$ | 119.9 |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{H}(10)$ | 119.9 |
| $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | 120.05(12) |
| $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{H}(11)$ | 120.0 |
| $\mathrm{C}(12)-\mathrm{C}(11)-\mathrm{H}(11)$ | 120.0 |


| $\mathrm{C}(13)-\mathrm{C}(12)-\mathrm{C}(11)$ | 120.15(12) |
| :---: | :---: |
| $\mathrm{C}(13)-\mathrm{C}(12)-\mathrm{H}(12)$ | 119.9 |
| $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{H}(12)$ | 119.9 |
| $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)$ | 119.95(12) |
| $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{H}(13)$ | 120.0 |
| $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{H}(13)$ | 120.0 |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(9)$ | 120.31(12) |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{H}(14)$ | 119.8 |
| $\mathrm{C}(9)-\mathrm{C}(14)-\mathrm{H}(14)$ | 119.8 |
| $\mathrm{C}(16)-\mathrm{C}(15)-\mathrm{C}(20)$ | 118.49(11) |
| $\mathrm{C}(16)-\mathrm{C}(15)-\mathrm{C}(5)$ | 122.75(10) |
| $\mathrm{C}(20)-\mathrm{C}(15)-\mathrm{C}(5)$ | 118.76(10) |
| $\mathrm{C}(17)-\mathrm{C}(16)-\mathrm{C}(15)$ | 120.77(11) |
| $\mathrm{C}(17)-\mathrm{C}(16)-\mathrm{H}(16)$ | 119.6 |
| $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{H}(16)$ | 119.6 |
| $\mathrm{C}(18)-\mathrm{C}(17)-\mathrm{C}(16)$ | 120.26(11) |
| $\mathrm{C}(18)-\mathrm{C}(17)-\mathrm{H}(17)$ | 119.9 |
| $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{H}(17)$ | 119.9 |
| $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)$ | 119.39(11) |
| $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{H}(18)$ | 120.3 |
| $\mathrm{C}(19)-\mathrm{C}(18)-\mathrm{H}(18)$ | 120.3 |
| $\mathrm{C}(20)-\mathrm{C}(19)-\mathrm{C}(18)$ | 120.48(11) |
| $\mathrm{C}(20)-\mathrm{C}(19)-\mathrm{H}(19)$ | 119.8 |
| $\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{H}(19)$ | 119.8 |
| $\mathrm{C}(19)-\mathrm{C}(20)-\mathrm{C}(15)$ | 120.59(11) |
| $\mathrm{C}(19)-\mathrm{C}(20)-\mathrm{H}(20)$ | 119.7 |
| $\mathrm{C}(15)-\mathrm{C}(20)-\mathrm{H}(20)$ | 119.7 |

Symmetry transformations used to generate equivalent atoms:

Table S5. Anisotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for d8v19160. The anisotropic displacement factor exponent takes the form: $\quad-2 \pi^{2}\left[h^{2} a^{* 2} U^{11}+\ldots \quad+2 h k a^{*} b^{*} U^{12}\right]$

|  | $\mathrm{U}^{11}$ | $\mathrm{U}^{22}$ | $\mathrm{U}^{33}$ | $\mathrm{U}^{23}$ | $\mathrm{U}^{13}$ | $\mathrm{U}^{12}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O}(1)$ | 36(1) | 34(1) | 37(1) | $0(1)$ | 7(1) | 9(1) |
| $\mathrm{N}(1)$ | 31(1) | 25(1) | 24(1) | 1(1) | $0(1)$ | 2(1) |
| C(1) | 41(1) | 28(1) | 28(1) | -2(1) | 4(1) | 0 (1) |
| C(2) | 53(1) | 39(1) | 28(1) | -5(1) | -2(1) | -4(1) |
| C(3) | 45(1) | 44(1) | 30(1) | 2(1) | -10(1) | -8(1) |
| C(4) | 29(1) | 36(1) | 35(1) | 5(1) | -5(1) | -3(1) |
| C(5) | 27(1) | 26(1) | 26(1) | 2(1) | 0 (1) | 2(1) |
| C(6) | 23(1) | 28(1) | 29(1) | 0 (1) | 4(1) | -2(1) |
| C(7) | 29(1) | 31(1) | 24(1) | 1(1) | 1(1) | -1(1) |
| C(8) | 22(1) | 27(1) | 25(1) | 3(1) | $0(1)$ | -4(1) |
| C(9) | 27(1) | 27(1) | 23(1) | -2(1) | -2(1) | 0 (1) |
| C(10) | 30(1) | 30(1) | 28(1) | 1(1) | 1(1) | -1(1) |
| C(11) | 47(1) | 30(1) | 35(1) | 6(1) | 3(1) | 3(1) |
| C(12) | 50(1) | 39(1) | 37(1) | 3(1) | -1(1) | 19(1) |
| C(13) | 33(1) | 54(1) | 40(1) | 2(1) | 4(1) | 14(1) |
| C(14) | 31(1) | 40(1) | 35(1) | 6(1) | 6(1) | 2(1) |
| C(15) | 28(1) | 29(1) | 21(1) | -1(1) | -2(1) | -1(1) |
| C(16) | 29(1) | 30(1) | 35(1) | $0(1)$ | 1(1) | 3(1) |
| C(17) | 28(1) | 39(1) | 38(1) | -2(1) | 3(1) | -2(1) |
| C(18) | 35(1) | 35(1) | 28(1) | 2(1) | -2(1) | -9(1) |
| C(19) | 39(1) | 28(1) | 30(1) | 3(1) | -5(1) | 1(1) |
| C(20) | 29(1) | 32(1) | 28(1) | 2(1) | -1(1) | 4(1) |

Table S6. Hydrogen coordinates ( $\mathrm{x} 10^{4}$ ) and isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for d8v19160.

|  | x | y | z | $\mathrm{U}(\mathrm{eq})$ |
| :---: | :---: | :---: | :---: | :---: |
| H(1) | 6277 | 4820 | 2034 | 39 |
| H(2) | 5473 | 5036 | 745 | 48 |
| H(3A) | 4215 | 6102 | 782 | 48 |
| H(3B) | 5073 | 6814 | 691 | 48 |
| H(4A) | 4145 | 6329 | 2358 | 40 |
| H(4B) | 4200 | 7378 | 1913 | 40 |
| H(7) | 5696 | 6590 | 4760 | 33 |
| H(10) | 5579 | 4446 | 4639 | 35 |
| H(11) | 6433 | 3142 | 5154 | 45 |
| H(12) | 7969 | 2991 | 4743 | 50 |
| H(13) | 8642 | 4115 | 3777 | 50 |
| H(14) | 7774 | 5385 | 3206 | 42 |
| H(16) | 7188 | 6976 | 1951 | 38 |
| H(17) | 8043 | 8182 | 1254 | 42 |
| H(18) | 7381 | 9674 | 954 | 40 |
| H(19) | 5875 | 9972 | 1413 | 38 |
| H(20) | 5020 | 8775 | 2119 | 35 |

Table S7. Torsion angles [ ${ }^{\circ}$ ] for d8v19160.

| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | 149.63(13) |
| :---: | :---: |
| $\mathrm{C}(5)-\mathrm{N}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | -3.01(17) |
| $\mathrm{N}(1)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | -5.0(2) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | -19.83(18) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 50.40(14) |
| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(15)$ | 113.91(10) |
| $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(15)$ | -89.57(12) |
| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(4)$ | -122.03(10) |
| $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(4)$ | 34.49(14) |
| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(6)$ | -3.14(12) |
| $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(6)$ | 153.38(10) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(1)$ | -57.70(13) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(15)$ | 65.75(13) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | -170.29(10) |
| $\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{O}(1)$ | -176.88(10) |
| $\mathrm{C}(15)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{O}(1)$ | 65.14(14) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{O}(1)$ | -60.10(14) |
| $\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | 4.09 (11) |
| $\mathrm{C}(15)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | -113.89(10) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | 120.87(10) |
| $\mathrm{O}(1)-\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | 177.32(12) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | -3.74(12) |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{N}(1)$ | 1.84(13) |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | -179.73(10) |
| $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{C}(7)$ | -153.85(11) |
| $\mathrm{C}(5)-\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{C}(7)$ | 1.02(13) |
| $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{C}(9)$ | 27.62(17) |
| $\mathrm{C}(5)-\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{C}(9)$ | -177.51(9) |
|  |  |


| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | 58.47(16) |
| :---: | :---: |
| $\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | -123.23(12) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(14)$ | -118.95(14) |
| $\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(14)$ | 59.35(15) |
| $\mathrm{C}(14)-\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | -0.04(18) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | -177.48(11) |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | 1.36 (19) |
| $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(13)$ | -1.2(2) |
| $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)$ | -0.3(2) |
| $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(9)$ | 1.7(2) |
| $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{C}(14)-\mathrm{C}(13)$ | -1.48(19) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(14)-\mathrm{C}(13)$ | 175.96(12) |
| $\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(15)-\mathrm{C}(16)$ | -0.36(15) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(15)-\mathrm{C}(16)$ | -122.72(12) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(15)-\mathrm{C}(16)$ | 112.03(12) |
| $\mathrm{N}(1)-\mathrm{C}(5)-\mathrm{C}(15)-\mathrm{C}(20)$ | -179.69(10) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(15)-\mathrm{C}(20)$ | 57.95(14) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(15)-\mathrm{C}(20)$ | -67.31(13) |
| $\mathrm{C}(20)-\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | -1.09(17) |
| $\mathrm{C}(5)-\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | 179.58(11) |
| $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | -0.29(19) |
| $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)$ | 1.40(18) |
| $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(20)$ | -1.14(18) |
| $\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(20)-\mathrm{C}(15)$ | -0.25(18) |
| $\mathrm{C}(16)-\mathrm{C}(15)-\mathrm{C}(20)-\mathrm{C}(19)$ | 1.35 (17) |
| $\mathrm{C}(5)-\mathrm{C}(15)-\mathrm{C}(20)-\mathrm{C}(19)$ | -179.29(10) |

Symmetry transformations used to generate equivalent atoms:

Table S8. Hydrogen bonds for d8v19160 [ $\AA$ and $\left.{ }^{\circ}\right]$.
D-H...A $\quad$ d(D-H) $\quad$ d(H...A) $\quad$ d(D...A) (DHA)

















2a

$\sqrt{\text { O2g }}$


2a



2b



|  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |



2c






2d

$\left\|\|^{\text {®. }}\right.$





| $\\|^{\circ}$ | ${ }_{\text {\% }}^{\text {\% }}$ |  |  |
| :---: | :---: | :---: | :---: |


$2 f$















|  |  |  |  |
| :---: | :---: | :---: | :---: |











| - |  |  | $\underbrace{\circ}_{1 / 2}$ |  |
| :---: | :---: | :---: | :---: | :---: |




Due to the volatile nature of this compound, not all the petroleum ether could be removed.



2n




2n





| - |  |  |  | $1 / 1$ | $\stackrel{\text { \% }}{\stackrel{\circ}{6}}$ | $\stackrel{\text { àm }}{\substack{0}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |




 $\mid\| \|\| \|\| \|\| \|$







$2 q$



$2 r$

$\underbrace{\text { 子 }}$
$\sqrt{4}$











TBDPSO


TBDPSO




2u






2v





$\mathrm{PhHN} \longrightarrow$
2w



## $\mathrm{PhHN} \longrightarrow$ <br> 2w





2x


| E | 空 |  | $1$ |  |  |  | $\stackrel{.}{\omega}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |



2x



MeO


2y


$(\mathrm{EtO})_{2}^{\mathrm{O}} \mathrm{P}_{2 \mathrm{P}}^{\mathrm{O}}$


(EtO) $)_{2}^{\text {OU }}$
$2 z$



2za




4a


| \% |  |  |  | ${ }_{\text {g }}^{8}$ |
| :---: | :---: | :---: | :---: | :---: |





4b




4b


$\mathrm{Me}_{2} \mathrm{PhSi} \uparrow$
4c


## $\mathrm{Me}_{2} \mathrm{PhSi} \Omega$ <br> 4c


Ph
$\square$



Ph Ph
6a















8b


Nin
$\left.\right|^{\stackrel{\circ}{\square}}$


8c




8c











10a




10b



10c




10c




| \% |  |  | $\sqrt{\circ}$ | \% | $\stackrel{\text { ¢ }}{\substack{1 \\ \hline}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |




$12 f$





| \% |  | \% | $\\|^{\circ}$ | 嗛 | E |
| :---: | :---: | :---: | :---: | :---: | :---: |



12g



12h



12h



$12 i$

$\left.\right|^{\text {采 }}$




12i




TIPS
12j


 121


| \| |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |

$\mathrm{Ph} \square_{\mathrm{Bu}}$
121









(D)H



(H)DCD(H)

6a-d'




2a-d, 92\%



